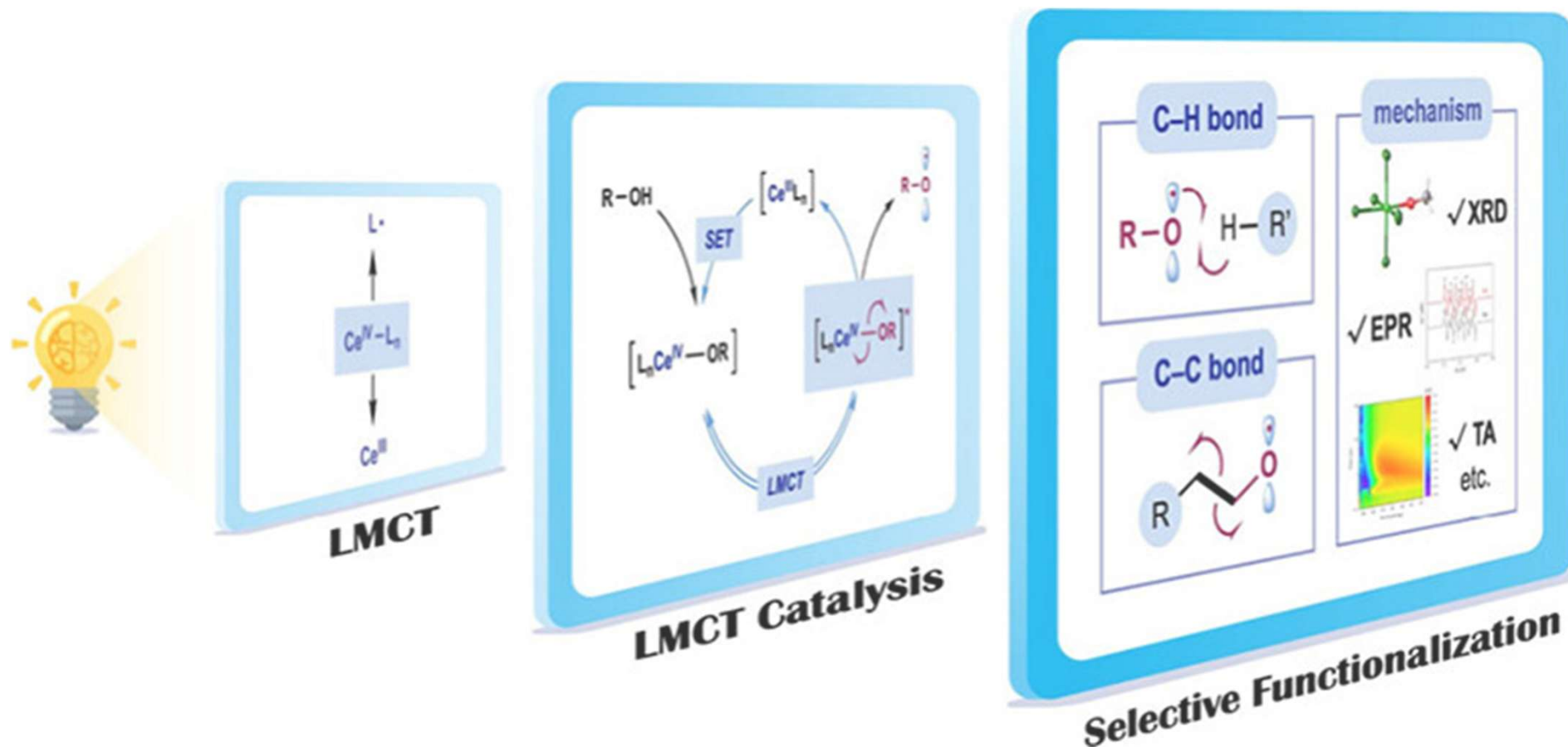




Methylative Cross-Coupling

2025.6.7. Literature Seminar
D3 Kyohei Oga

Contents

1. Introduction

2. ORGANIC CHEMISTRY

Alcohol-alcohol cross-coupling enabled by S_H2 radical sorting

Ruizhe Chen^{1†}, Nicholas E. Intermaggio^{1†}, Jiaxin Xie^{1†}, James A. Rossi-Ashton^{1†}, Colin A. Gould¹, Robert T. Martin¹, Jesús Alcázar², David W. C. MacMillan^{1*}

3.



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Article

Synergistic LMCT and Ni Catalysis for Methylative Cross-Coupling Using *tert*-Butanol: Modulating Radical Pathways via Selective Bond Homolysis

Lingfei Duan, Yunzhi Lin, Qing An, and Zhiwei Zuo*

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J|A|C|S
JOURNAL OF THE AMERICAN CHEMICAL SOCIETY

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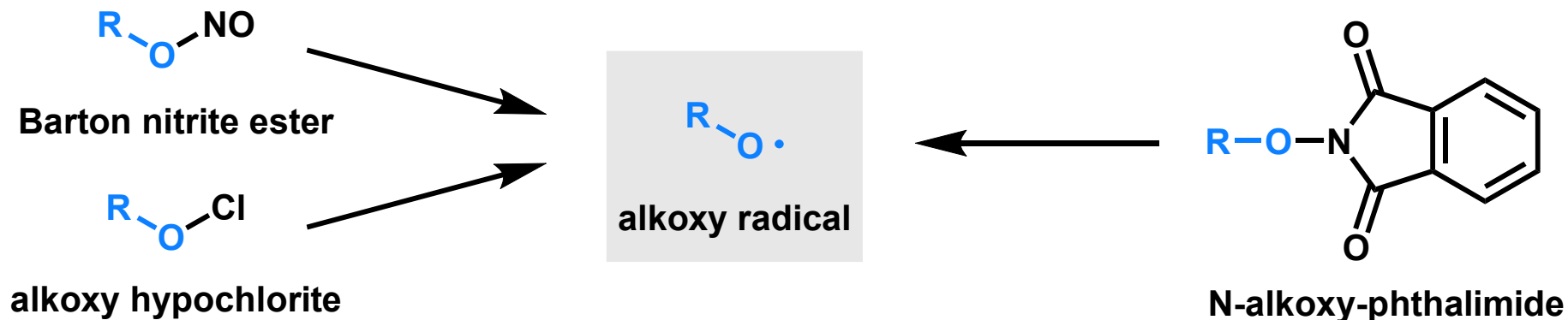
Article

Synergistic LMCT and Ni Catalysis for Methylative Cross-Coupling Using *tert*-Butanol: Modulating Radical Pathways via Selective Bond Homolysis

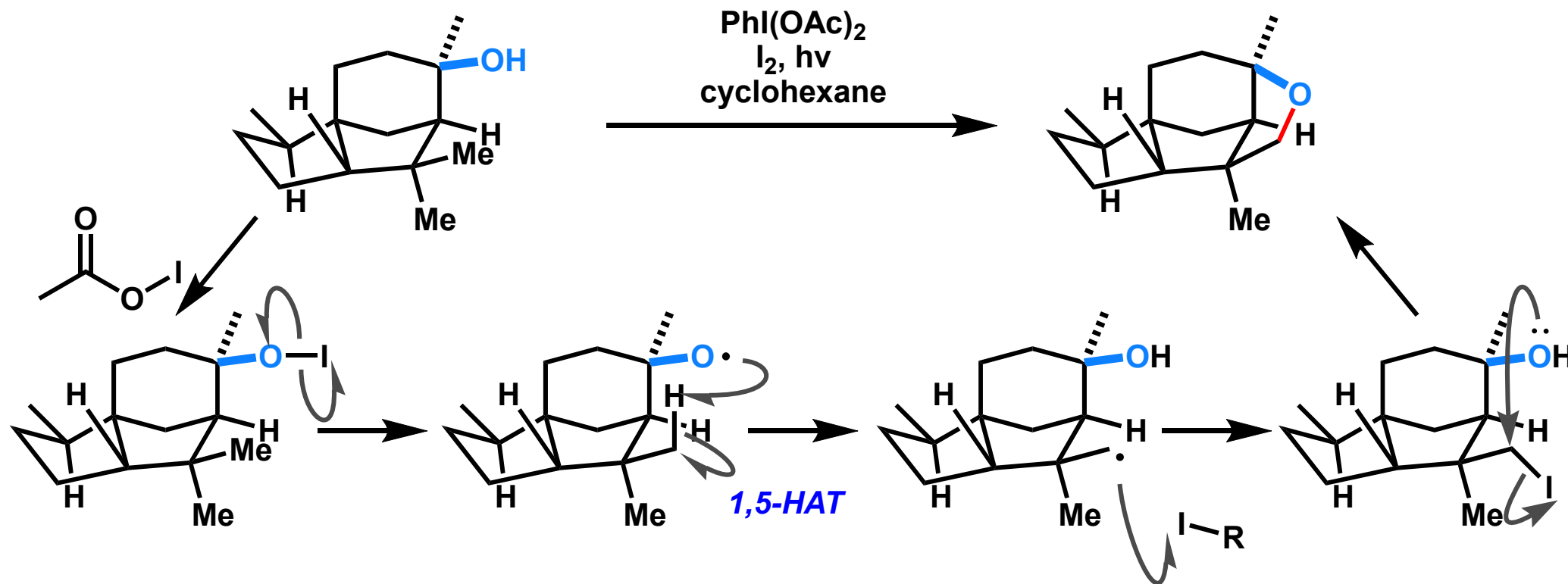
Lingfei Duan, Yunzhi Lin, Qing An, and Zhiwei Zuo*

Alkoxy Radical

Classic precursors for alkoxy radical generation¹⁾



Total synthesis of Pseudoanisatin²⁾



1) Chang, L.; An, Q.; Duan, L.; Feng, K.; Zuo, Z. *Chem. Rev.*, **2022**, 122, 2429.

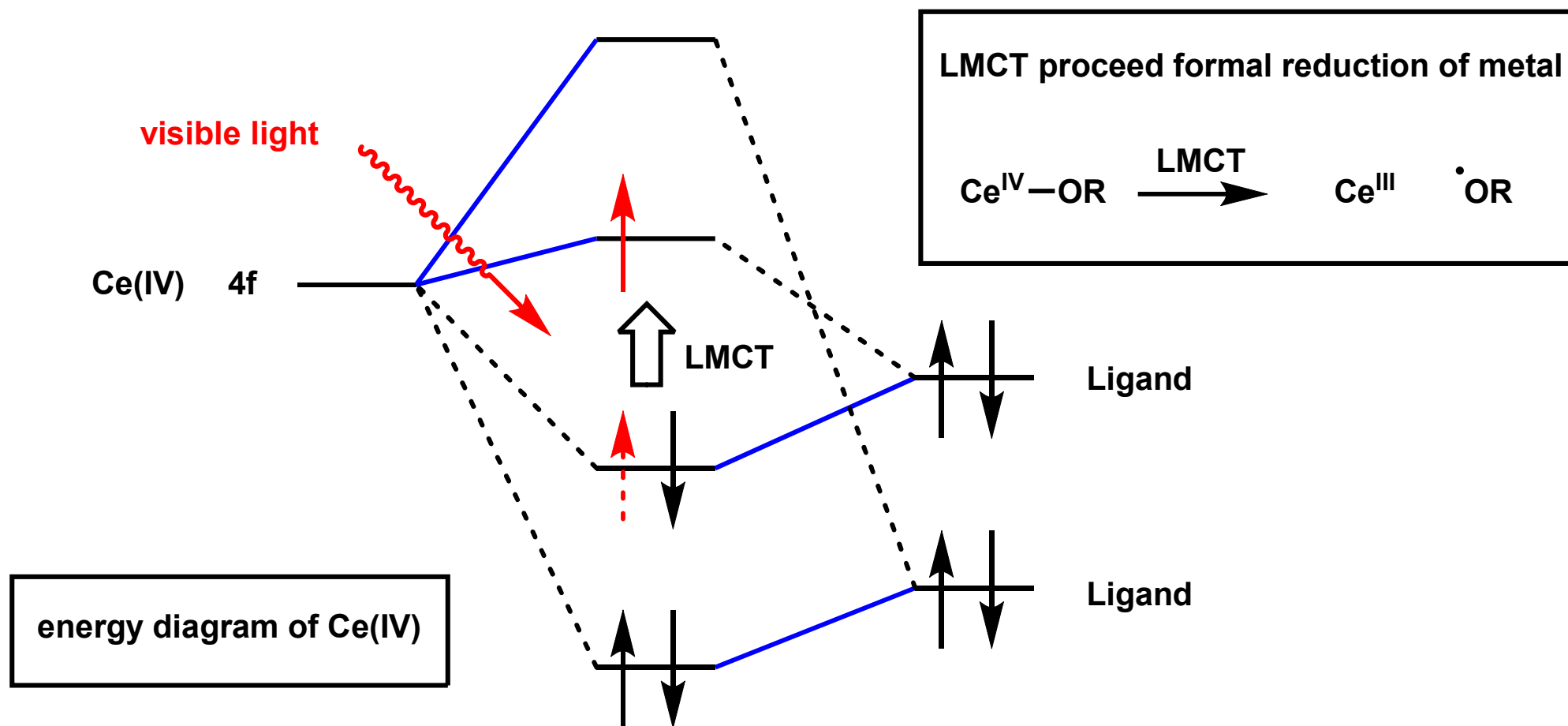
2) Hung, K.; Condakes, M.-L.; Morikawa, T.; Maimone, T.-J. *J. Am. Chem. Soc.* **2016**, 138, 16616.

LMCT (Ligand to Metal Charge Transfer) ^{1) 2)}

LMCT complexes arise from transfer of electrons from MO with ligand-like character to those with metal-like character.

(opposite relationship to MLCT)

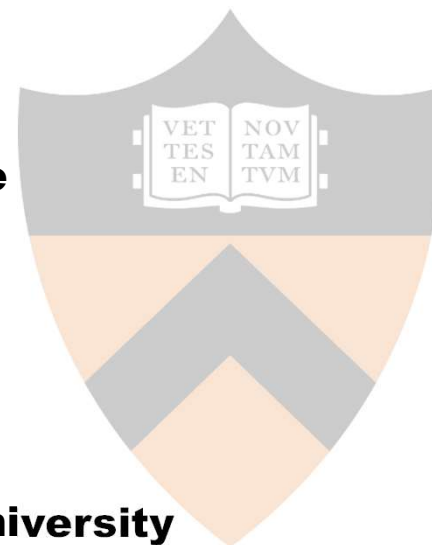
* This type of transfer could occur if complexes have ligands with relatively high-energy lone pairs or if the metal has low-lying empty orbitals.





Prof. David W. C. MacMillan

- 1991 B.S. @The University of Glasgow
- 1996 Ph.D. @The University of California, Irvine (Prof. Larry E. Overman)
- 1998 Postdoctoral Fellow @Harvard University (Prof. David A. Evans)
- 2000 (PI) @University of California, Berkeley
- 2004 (PI) @California Institute of Technology
- 2011 Professor @Princeton University
- Present Distinguished Professor @Princeton University



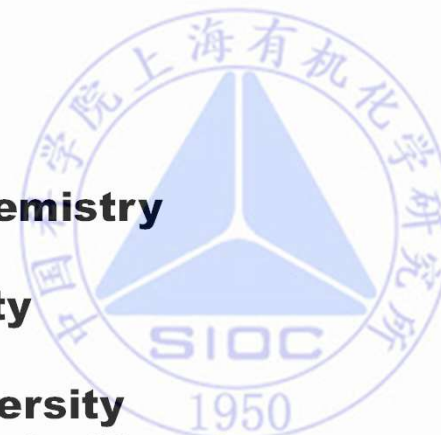
Research topics:

Photoredox catalysis, Organocatalysis, Labeling biomolecules, Total synthesis



Prof. Zhiwei Zuo

- 2007 B.S. @Nanjing University (Prof. Yixiang Chen)
- 2012 Ph.D. @ Shanghai Institute of Organic Chemistry (Prof. Dawei Ma)
- 2015 Postdoctoral Fellow @Princeton University (Prof. David W. C. MacMillan)
- 2020 Assistant Professor @ShanghaiTech University
- Present Professor @Shanghai Institute of Organic Chemistry



Research topics:

Photoinduced LMCT catalysis for C-H functionalization, C-C bond Formation/cleavage, and efficient synthetic process

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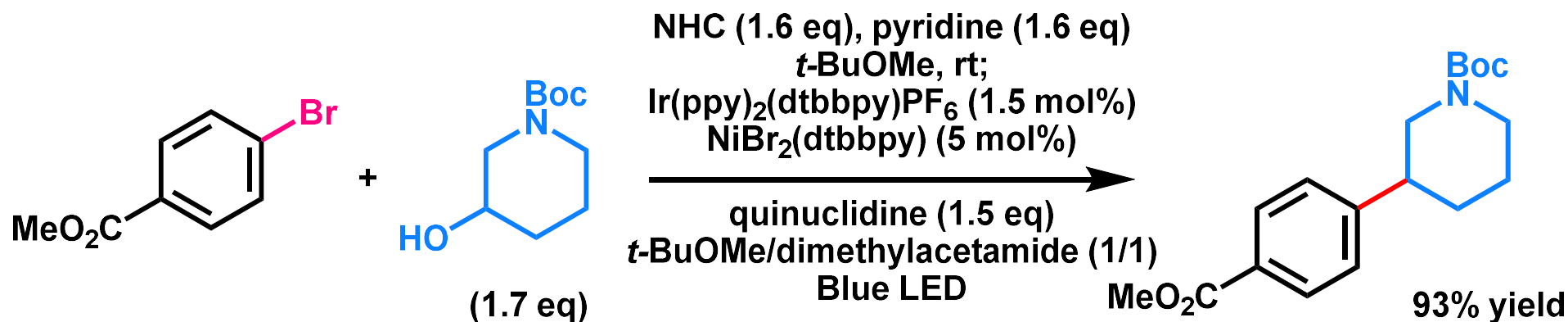
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Article

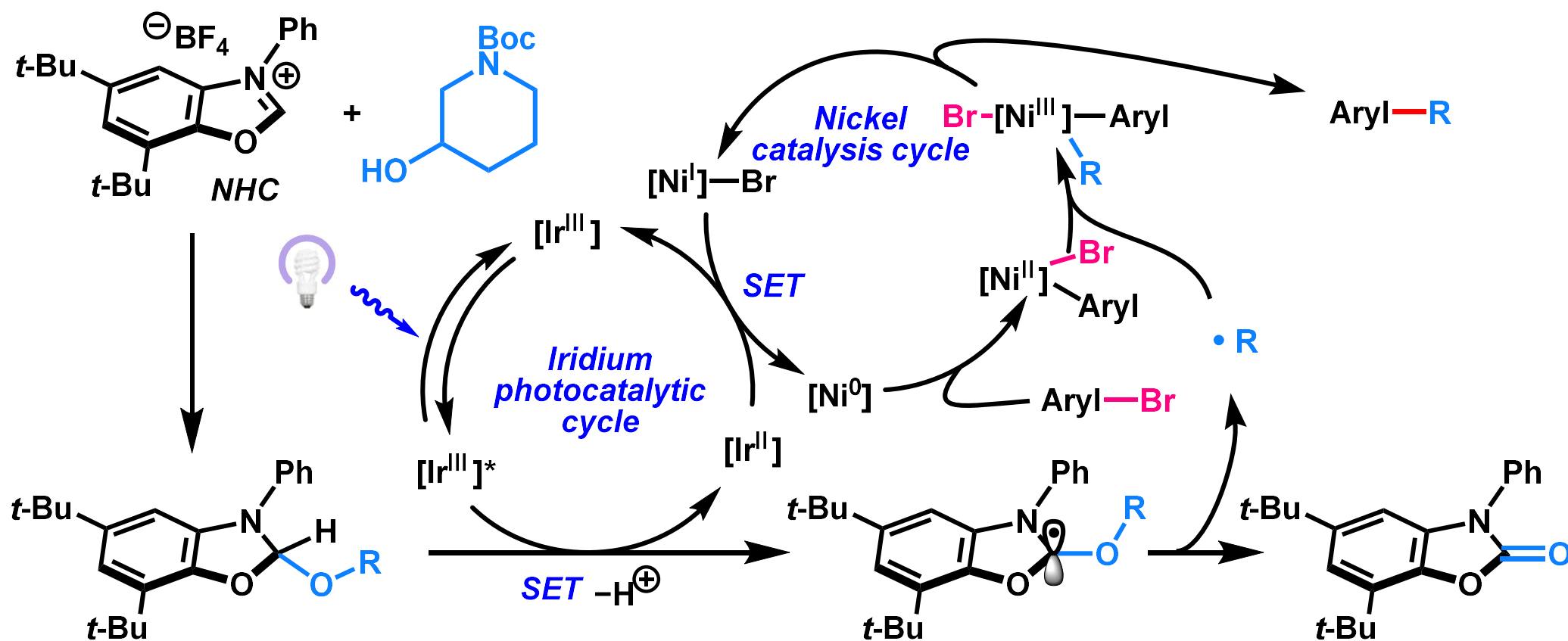
Synergistic LMCT and Ni Catalysis for Methylative Cross-Coupling Using *tert*-Butanol: Modulating Radical Pathways via Selective Bond Homolysis

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Previous Work in MacMillan Lab



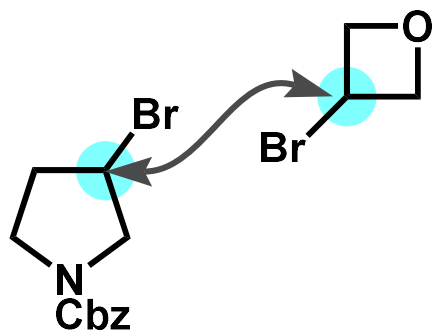
Dong, Z.; MacMillan, D. W. C. *Nature* 2021, 598, 451.



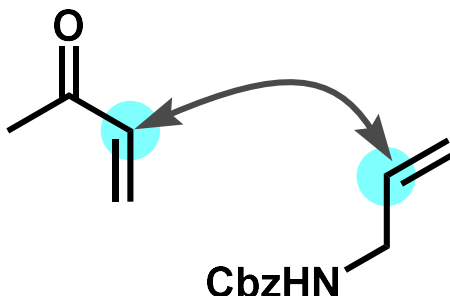
1) Dong, Z.; MacMillan, D. W. C. *Nature* 2021, 598, 451.

2) Nicholas, A.; Seokjoon Oh, T.; MacMillan, D. W. C.; Bird, M. J. *J. Am. Chem. Soc.* 2021, 143, 9332.

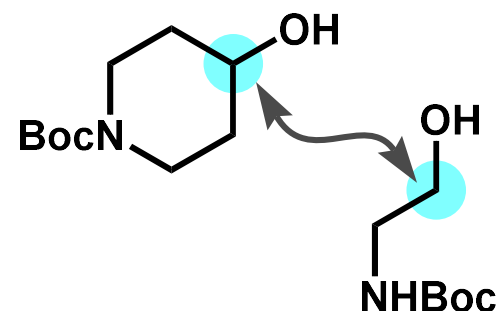
Single Functional Group Coupling



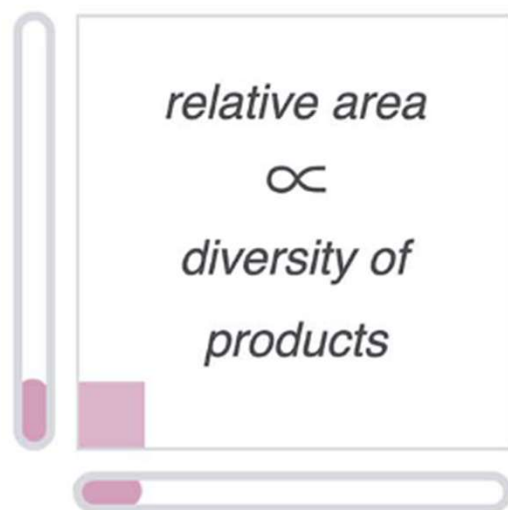
cross-bromide coupling



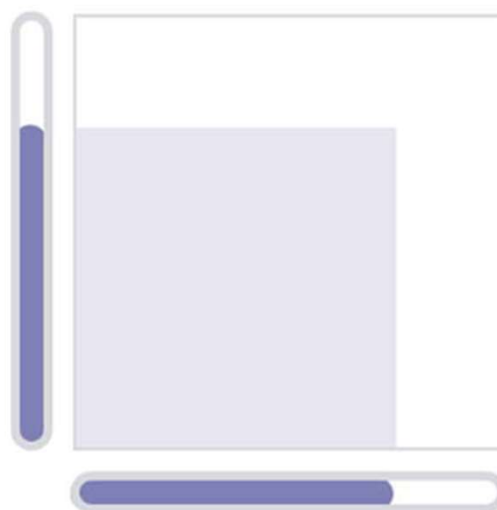
cross metathesis



cross-alcohol coupling



commercial bromides

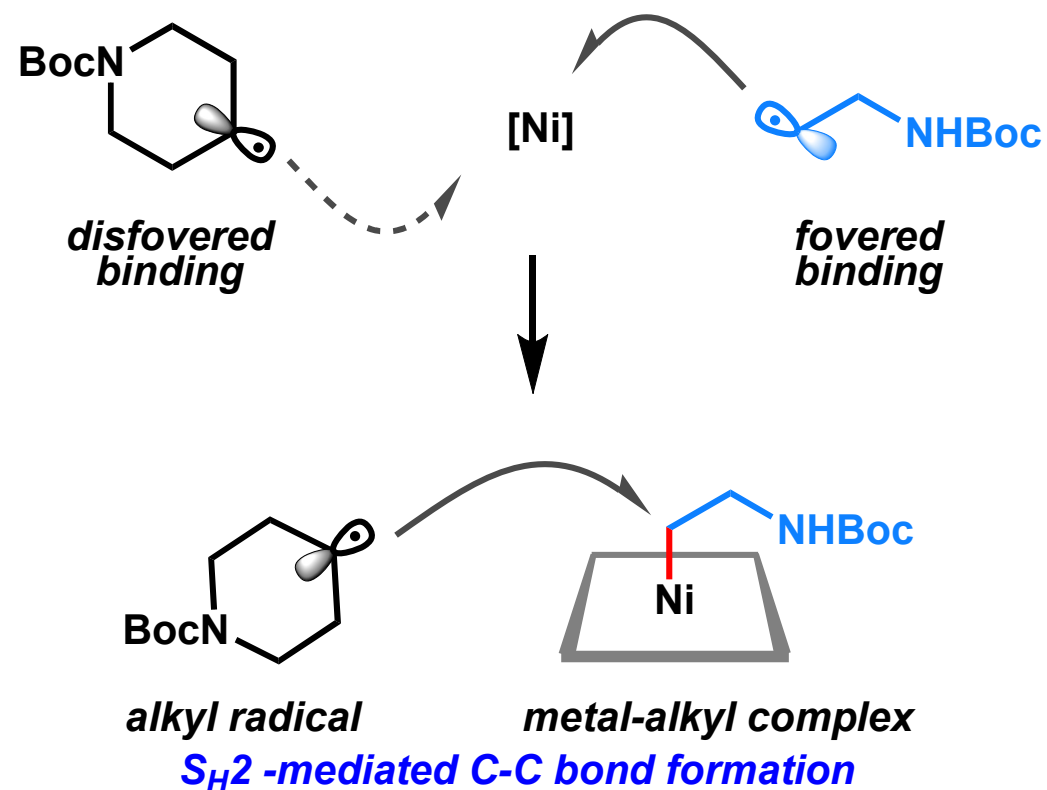
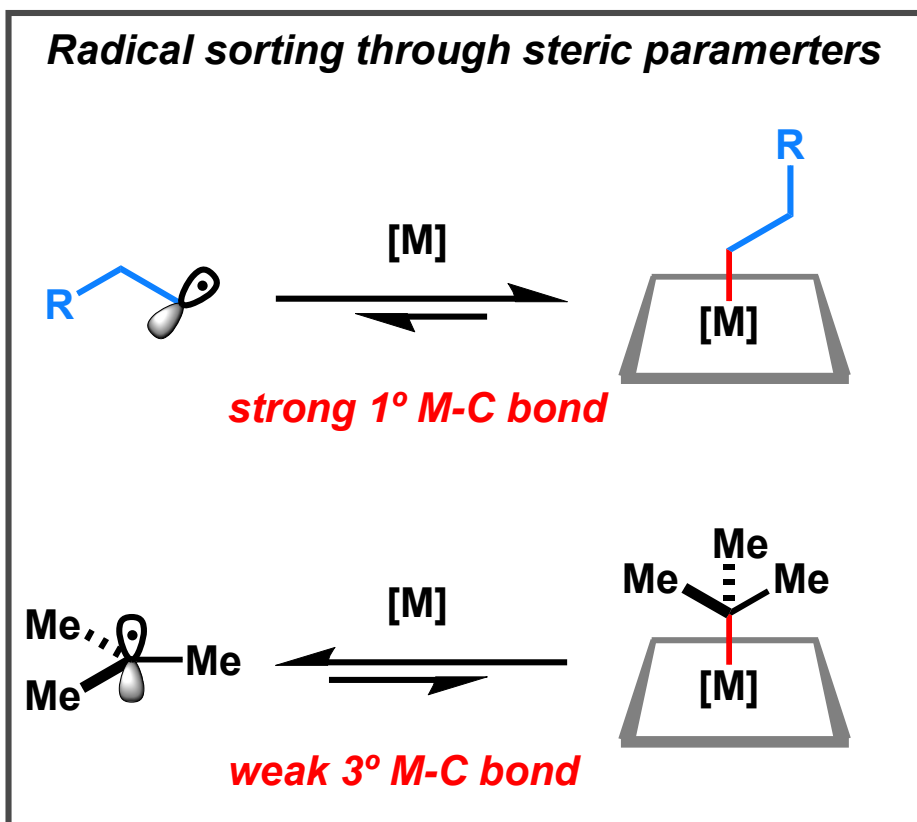
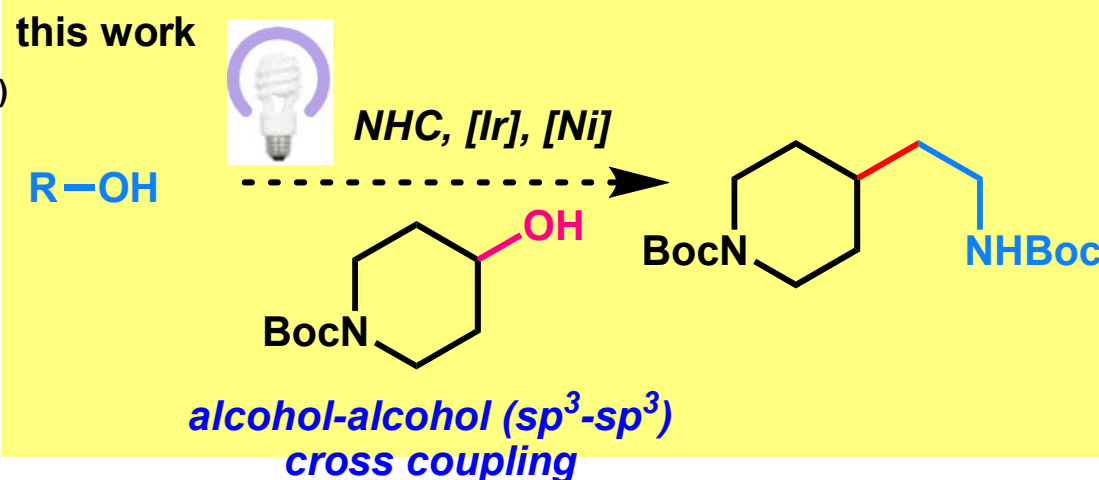
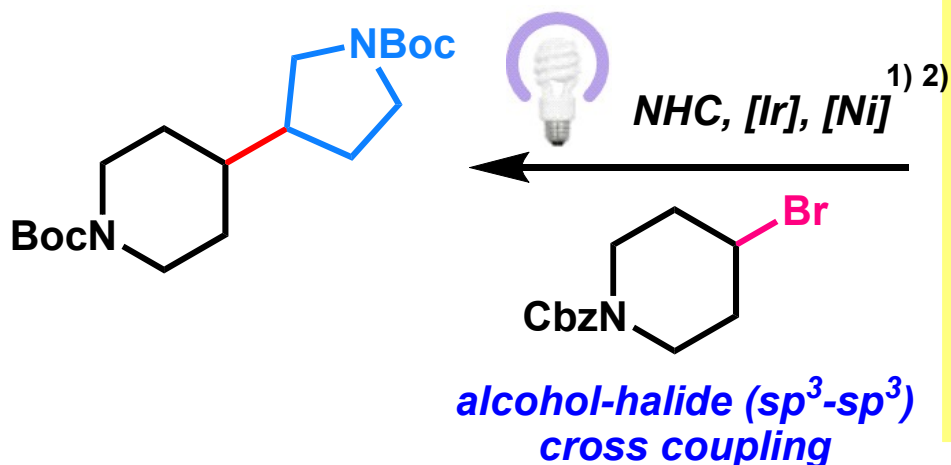


commercial olefins



commercial alcohols

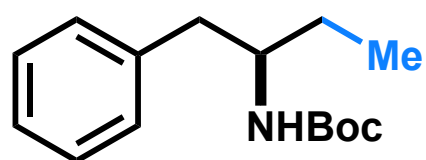
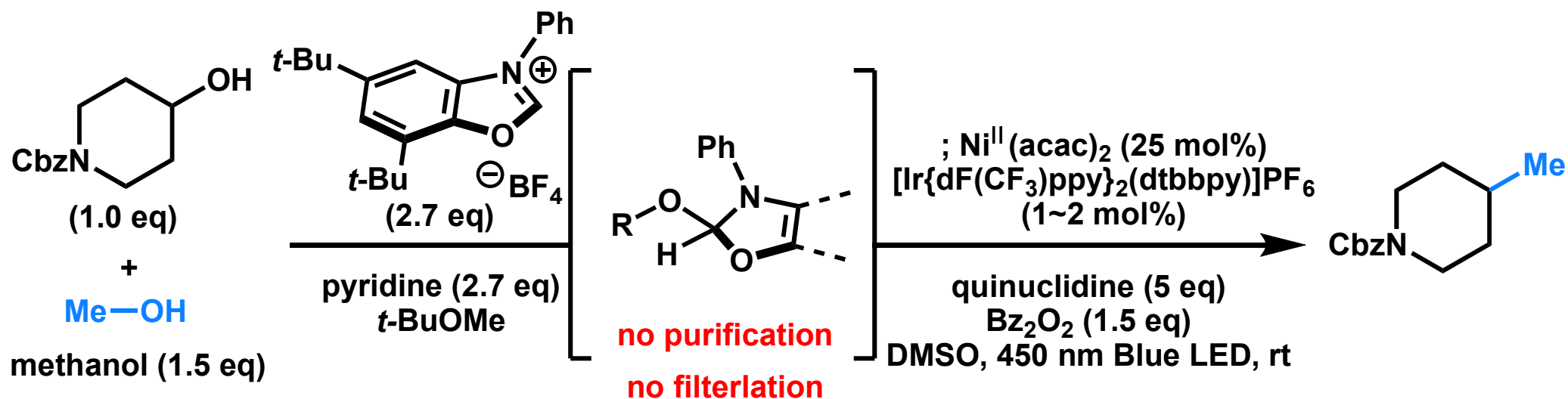
Working Hypothesis



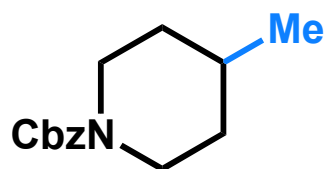
1) Lyon, W. L.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2023**, 145, 7736.

2) 230415_LS_Yuki_Komori

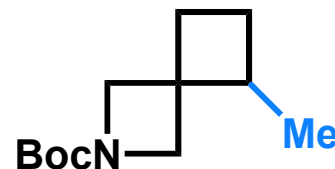
Substrate Scope



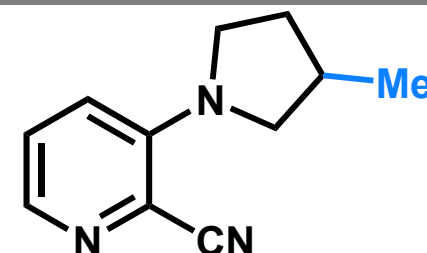
51% yield



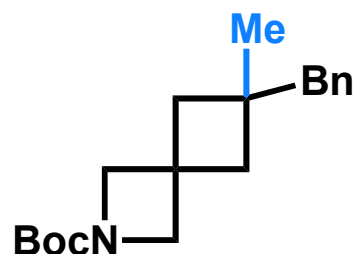
63% yield



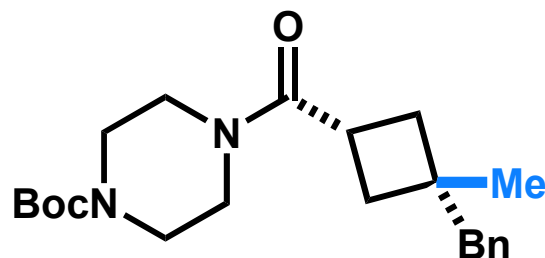
51% yield



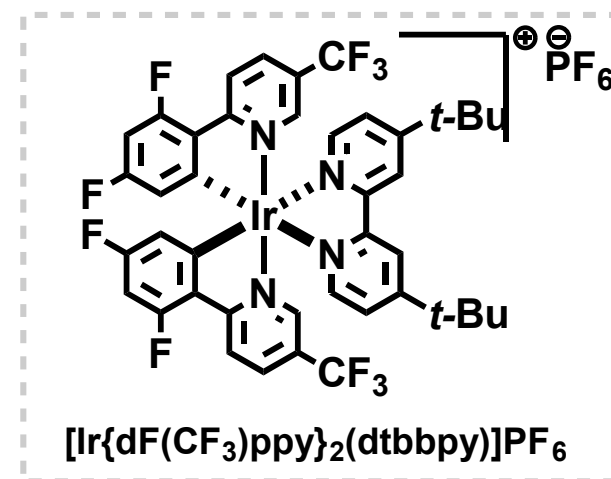
68% yield



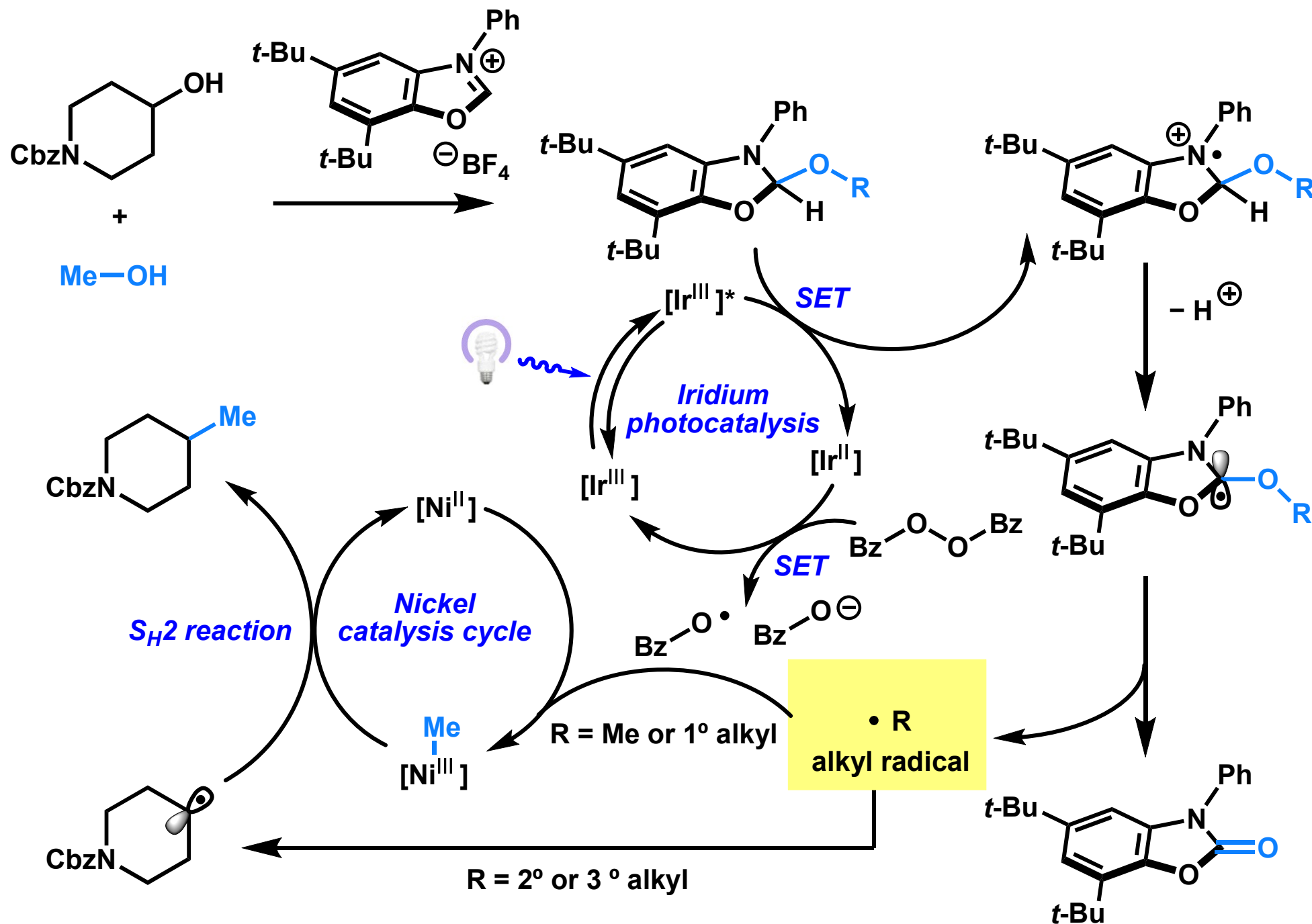
62% yield



60% yield (dr = 1.5:1)



Proposed Reaction Mechanism



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3.



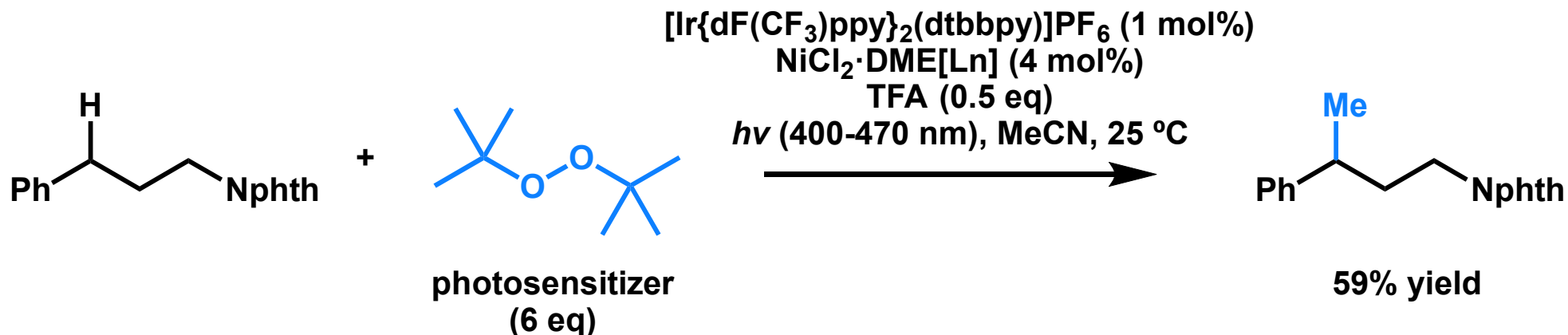
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Article

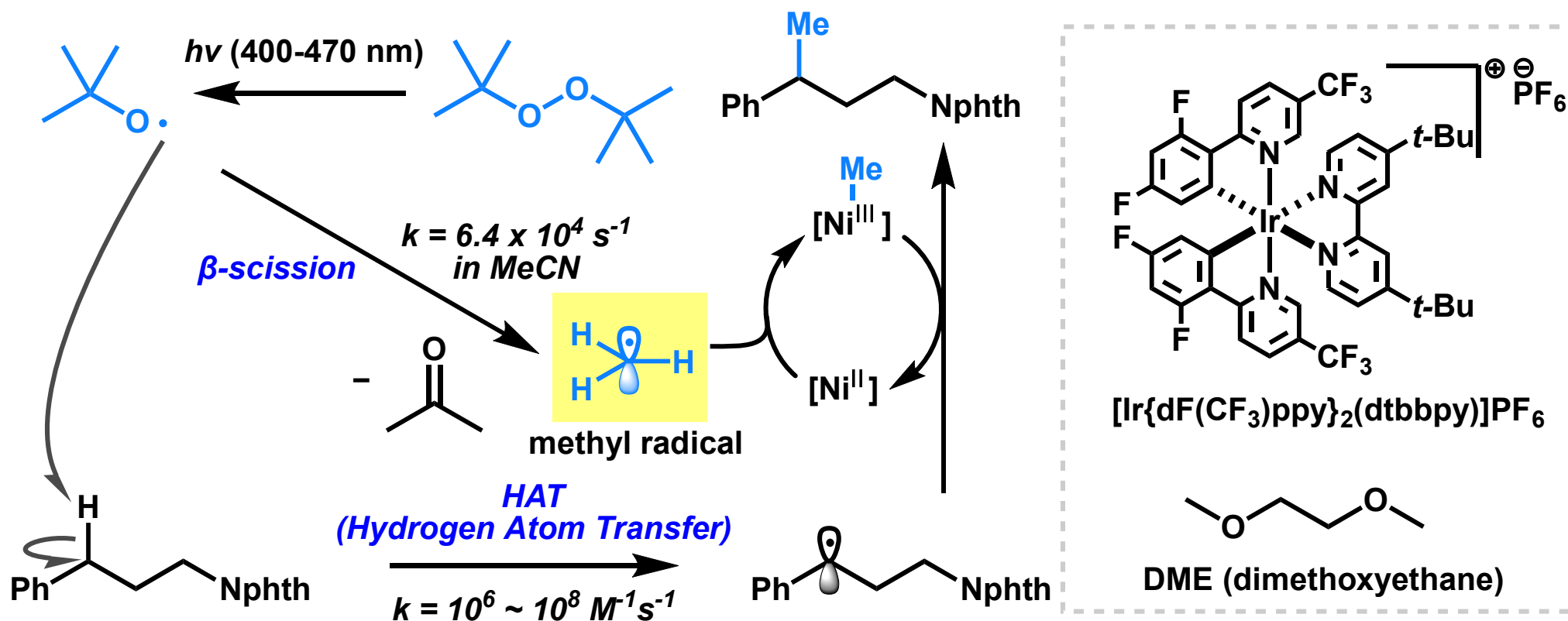
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Use of *tert*-Butoxy Radical as Methyl Source

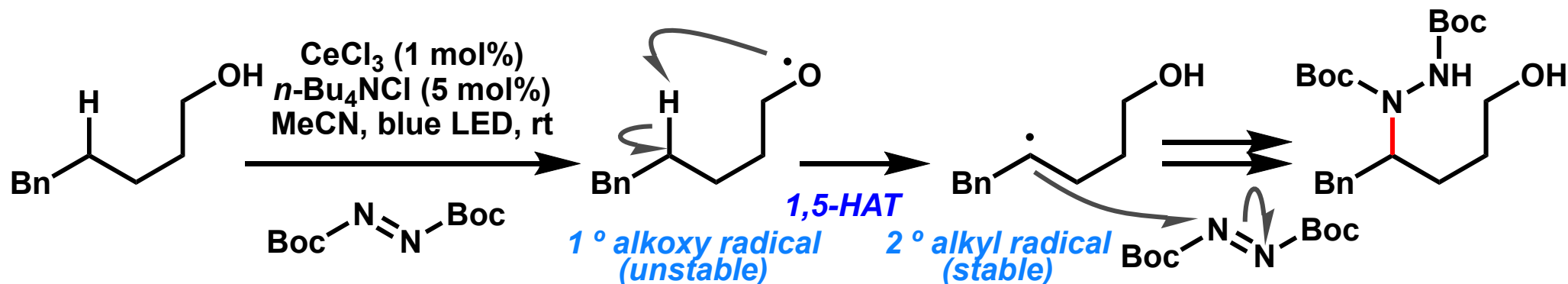


Vasilopoulos, A.; Krska, S. W.; Stahl, S. S. *Science* 2021, 372, 398.

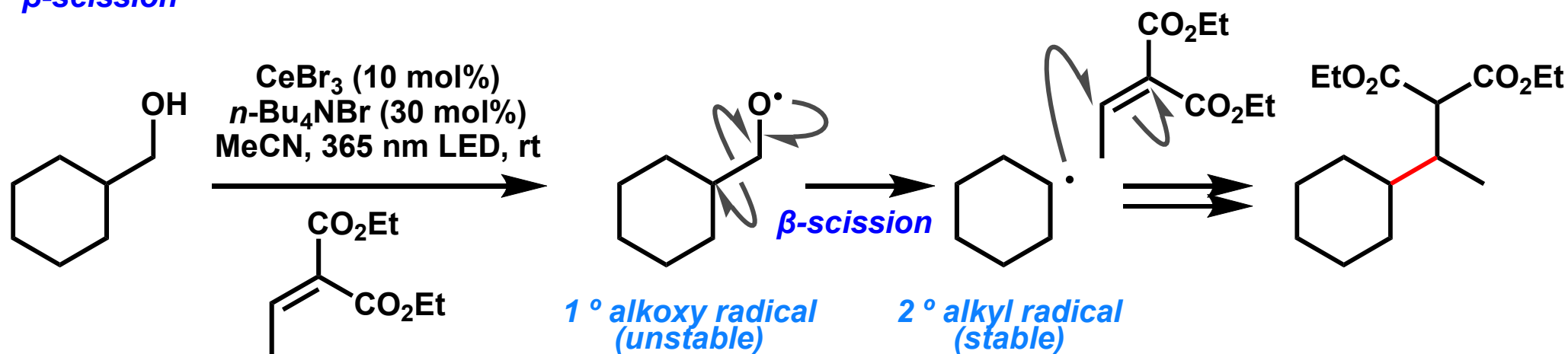


Previous Work by Zuo group

Hydrogen Atom Transfer (HAT)¹⁾



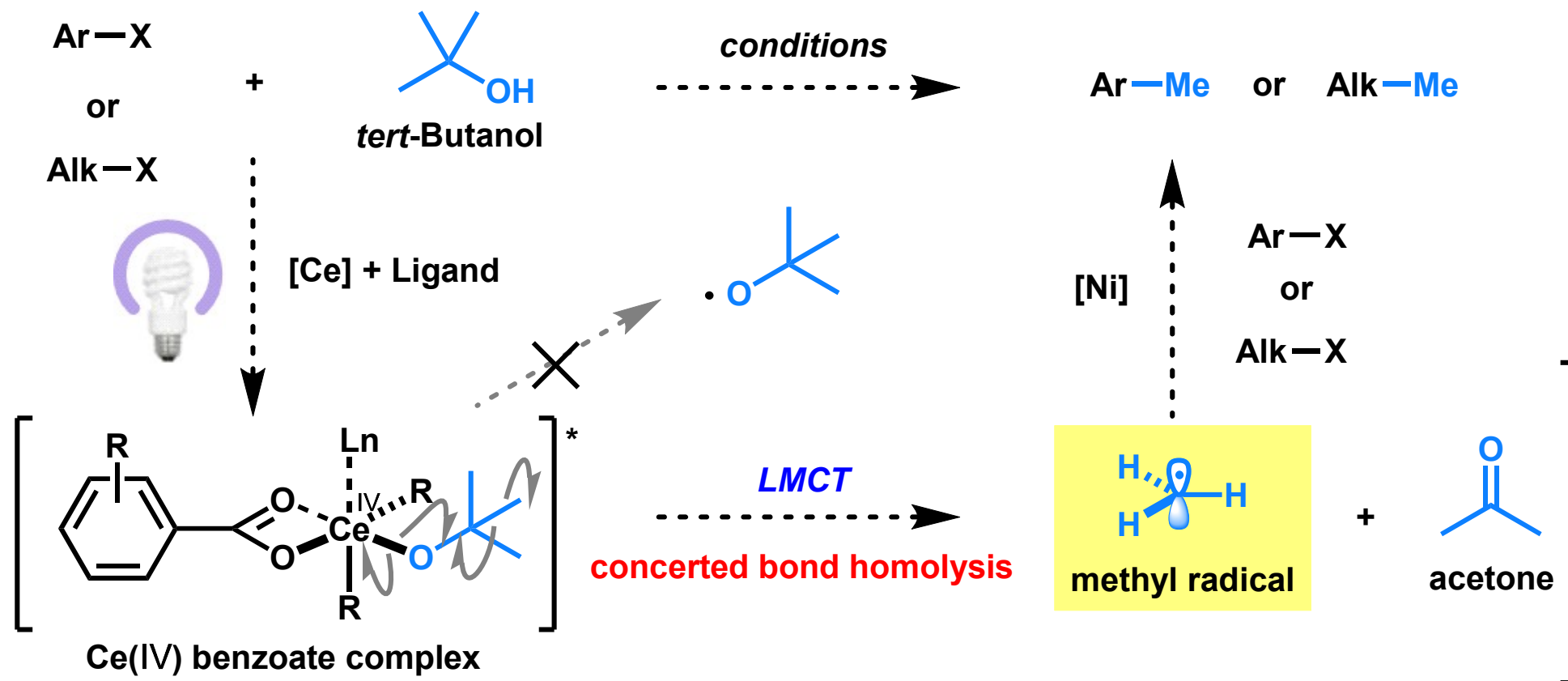
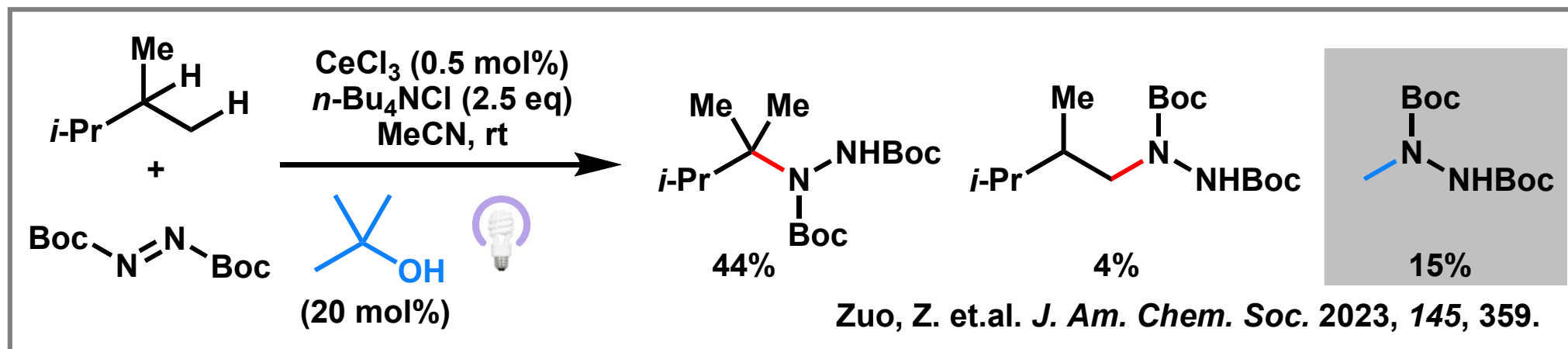
β -scission²⁾



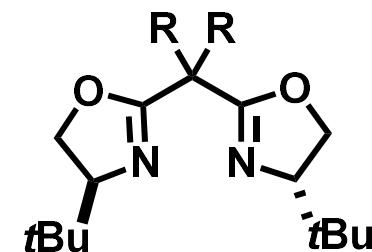
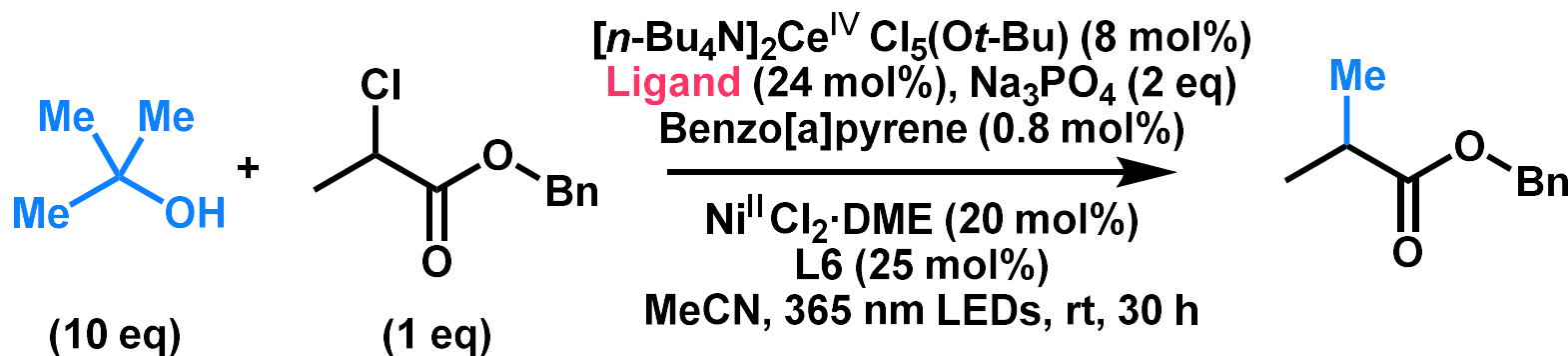
1) Hu, Anhua.; Guo, J.-J.; Pan, H.; Tang, H.; Gao, Z.; Zuo, Z. *J. Am. Chem. Soc.* **2018**, *140*, 1612.

2) Zhang, K.; Chang, L.; An, Q.; Wang, X.; Zuo, Z. *J. Am. Chem. Soc.* **2019**, *141*, 10556.

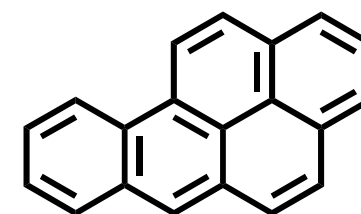
Working Hypothesis



Screening of Benzoate Ligand

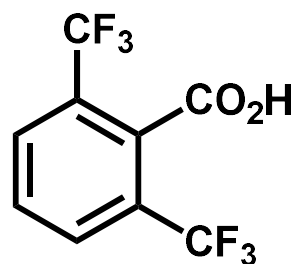


L6 (R = 4-Ph-Bn)

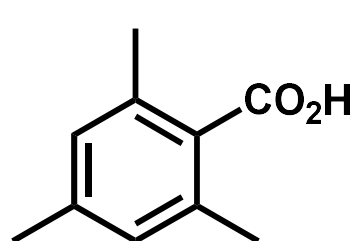


benzo(a)pyrene

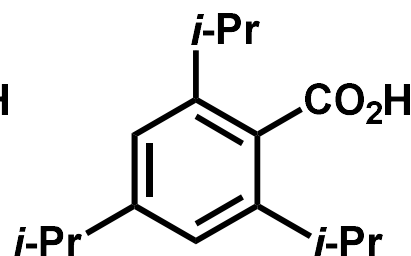
Ligand



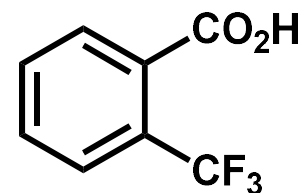
L1: 76%



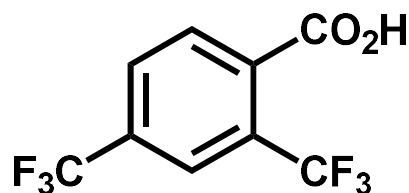
L2: 16%



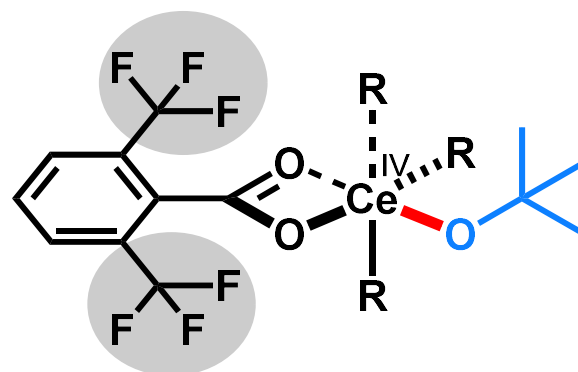
L3: 46%



L4: 43%



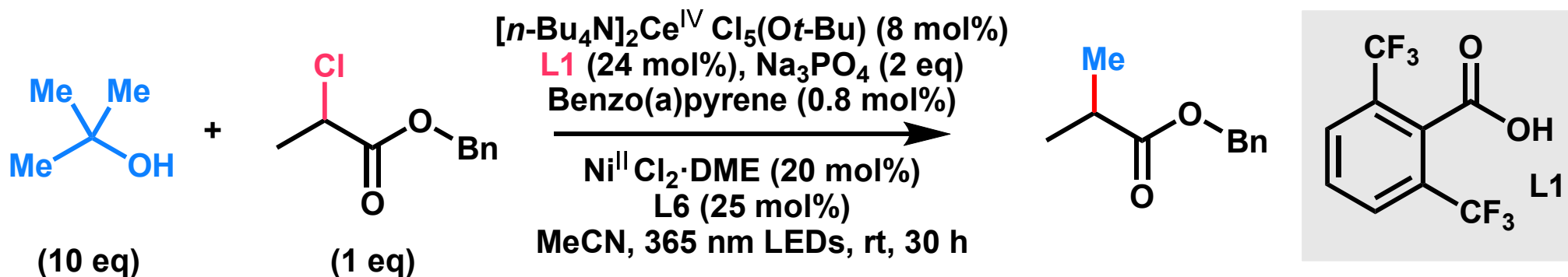
L5: 47%



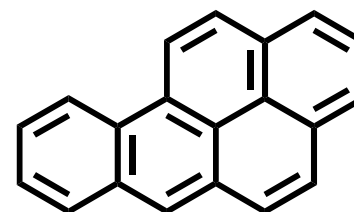
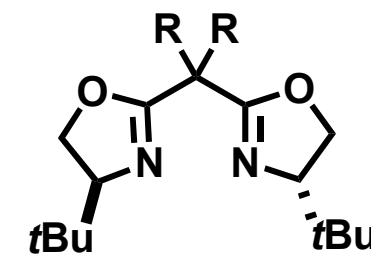
- Steric hindrance of the ligand fixes the coordination to the Ce center.
- Electron-withdrawing substituents on the aromatic ring enhance the **binding affinity** of tBuO to Ce.

➔ Inhibition of generation of *t*-BuO·

Deviation of Reaction Conditions



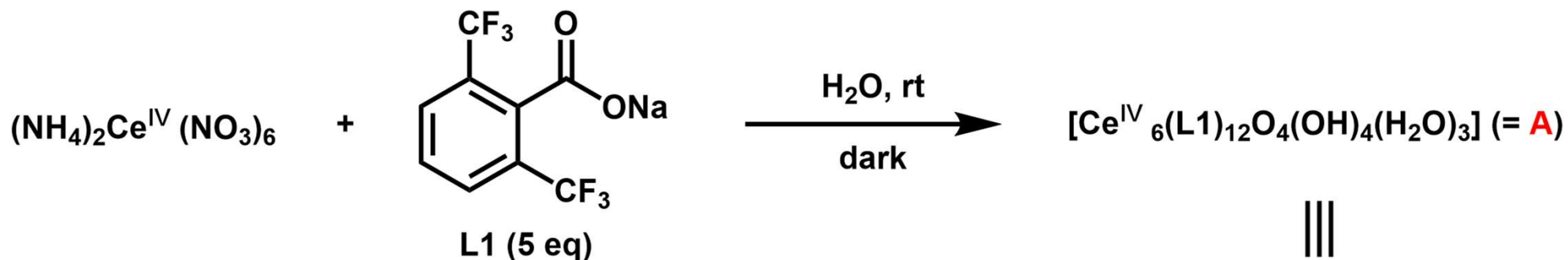
entry	variation from the standard conditions	Yield (%)
1	none	76%
2	$[n\text{-Bu}_4\text{N}]_2\text{Ce}^{\text{IV}}\text{Cl}_5(\text{Ot-Bu})$ or $\text{Ni}^{\text{II}}\text{Cl}_2\cdot\text{DME}$	0%
3	benzo[a]pyrene	2%
4	no Light	0%



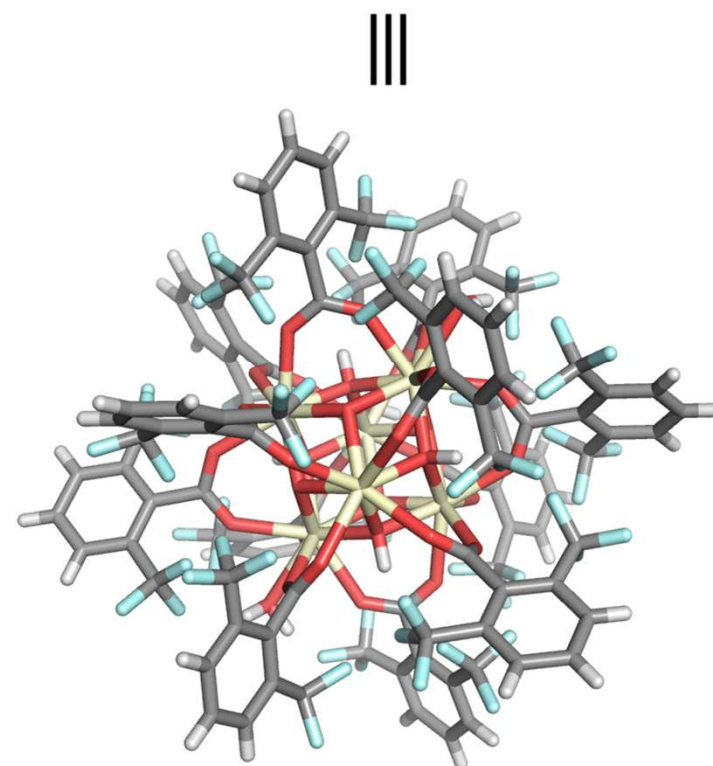
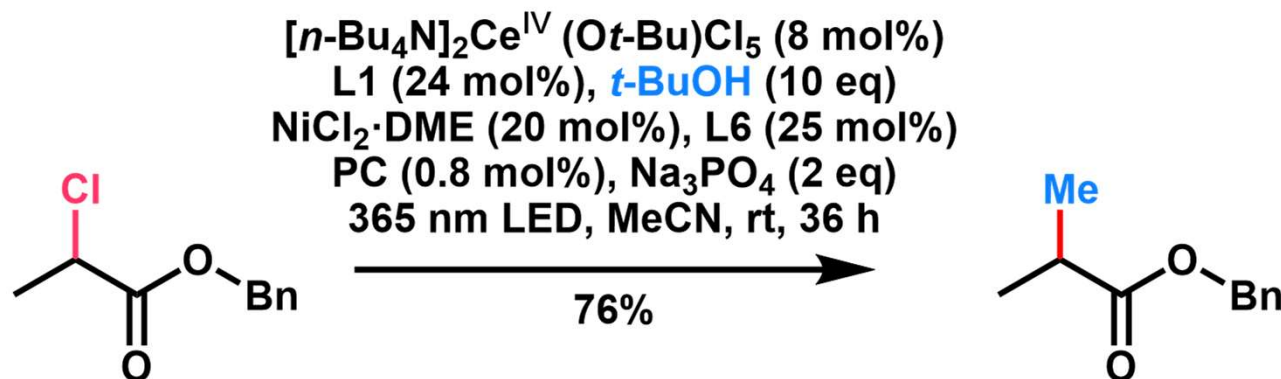
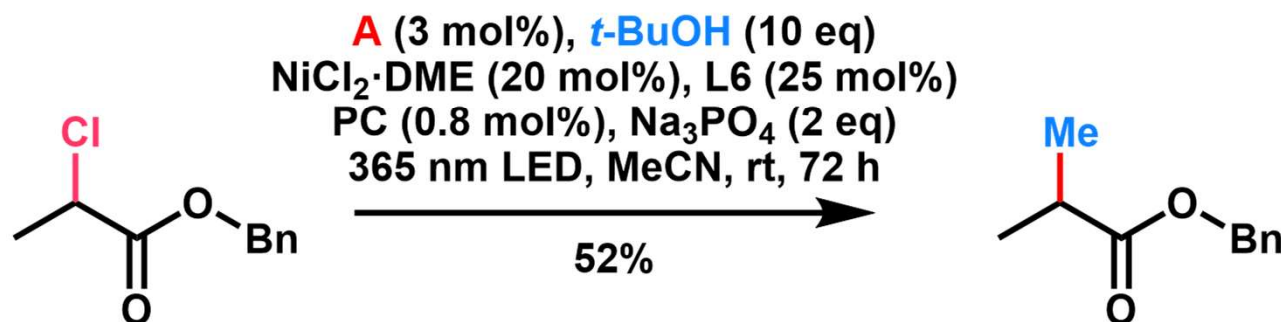
benzo[a]pyrene

Synthesis and Characterization of Ce(IV) Complex

(A) Synthesis of Ce(IV) benzoate complex



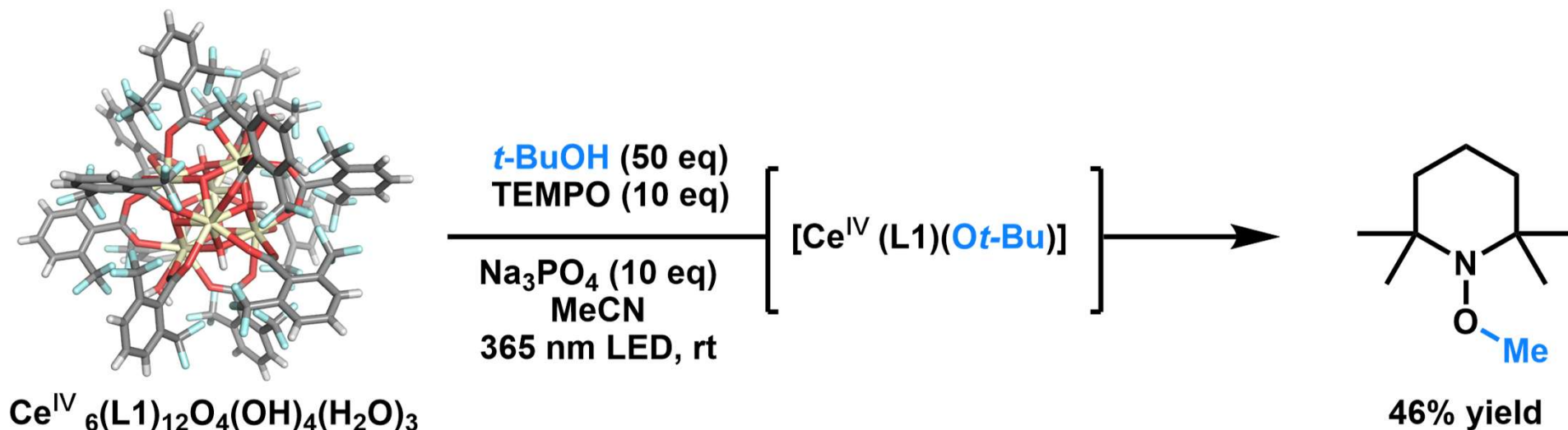
(B) Characterization of Ce(IV) benzoate complex



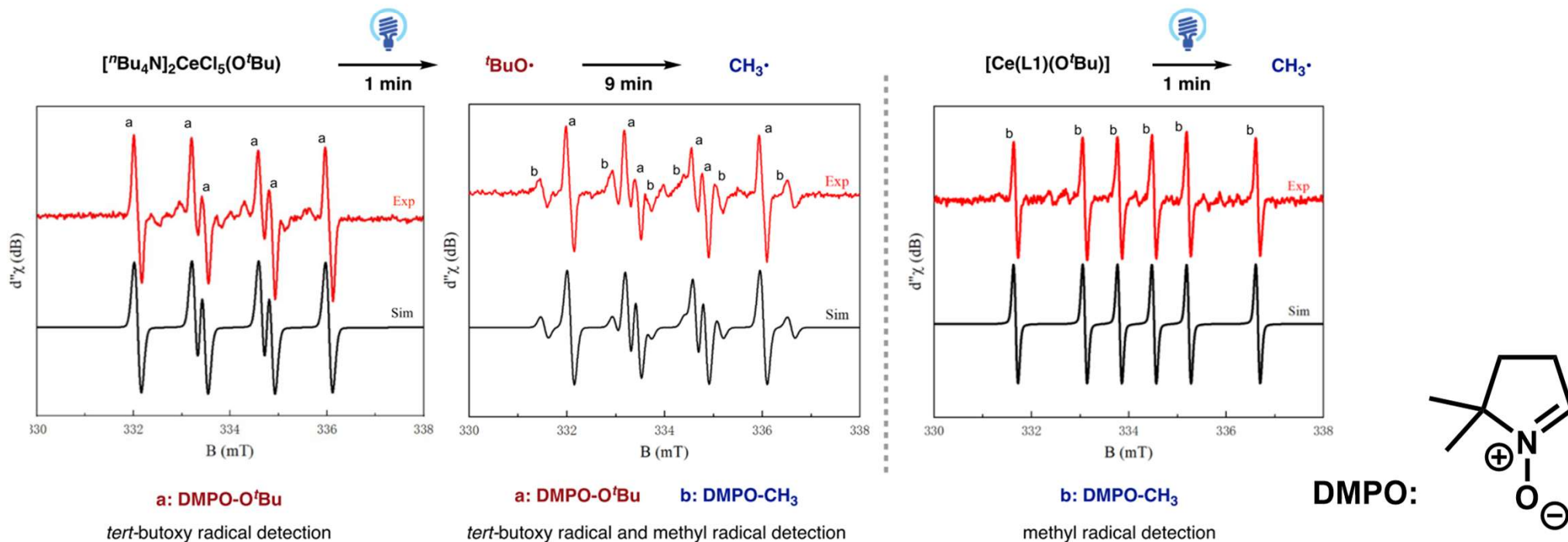
PC = Benzo[a]pyrene

Methyl Radical Trapping and EPR Experiments

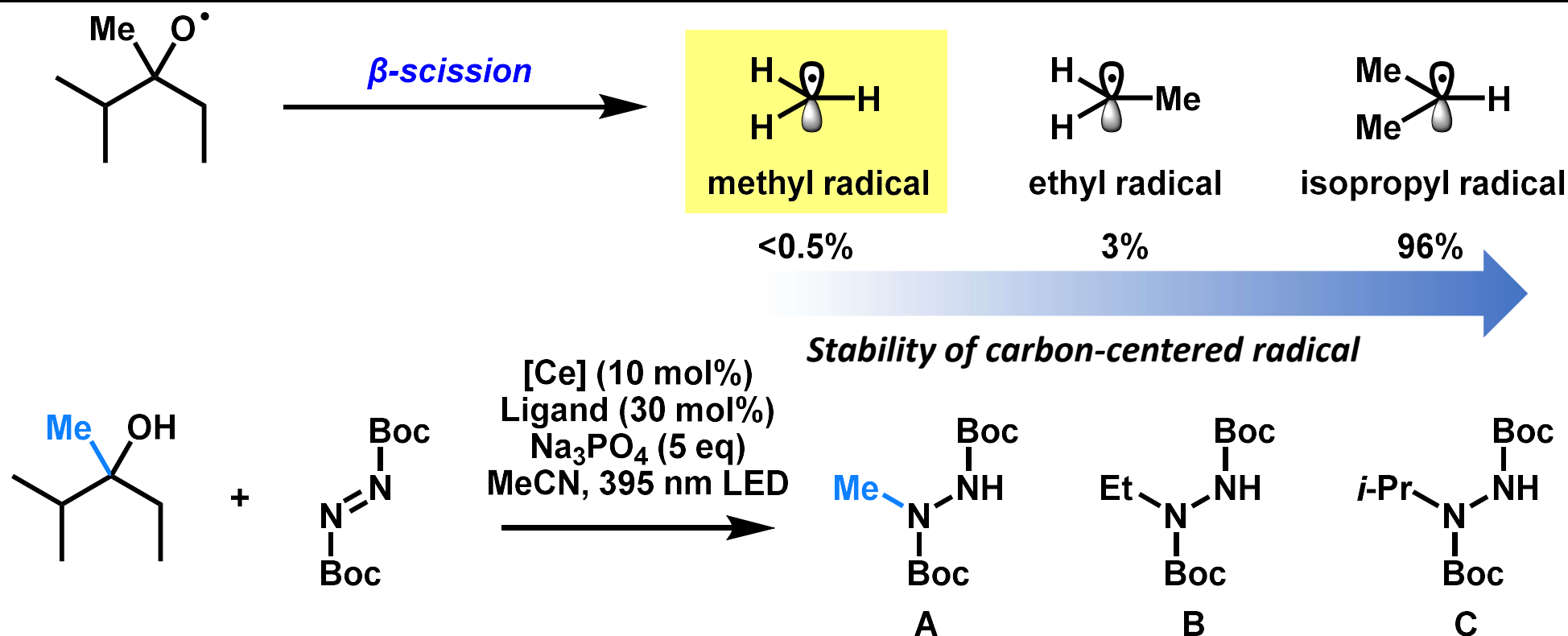
(A) Methyl radical trapping by TEMPO

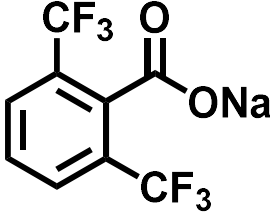


(B) Operando EPR experiments

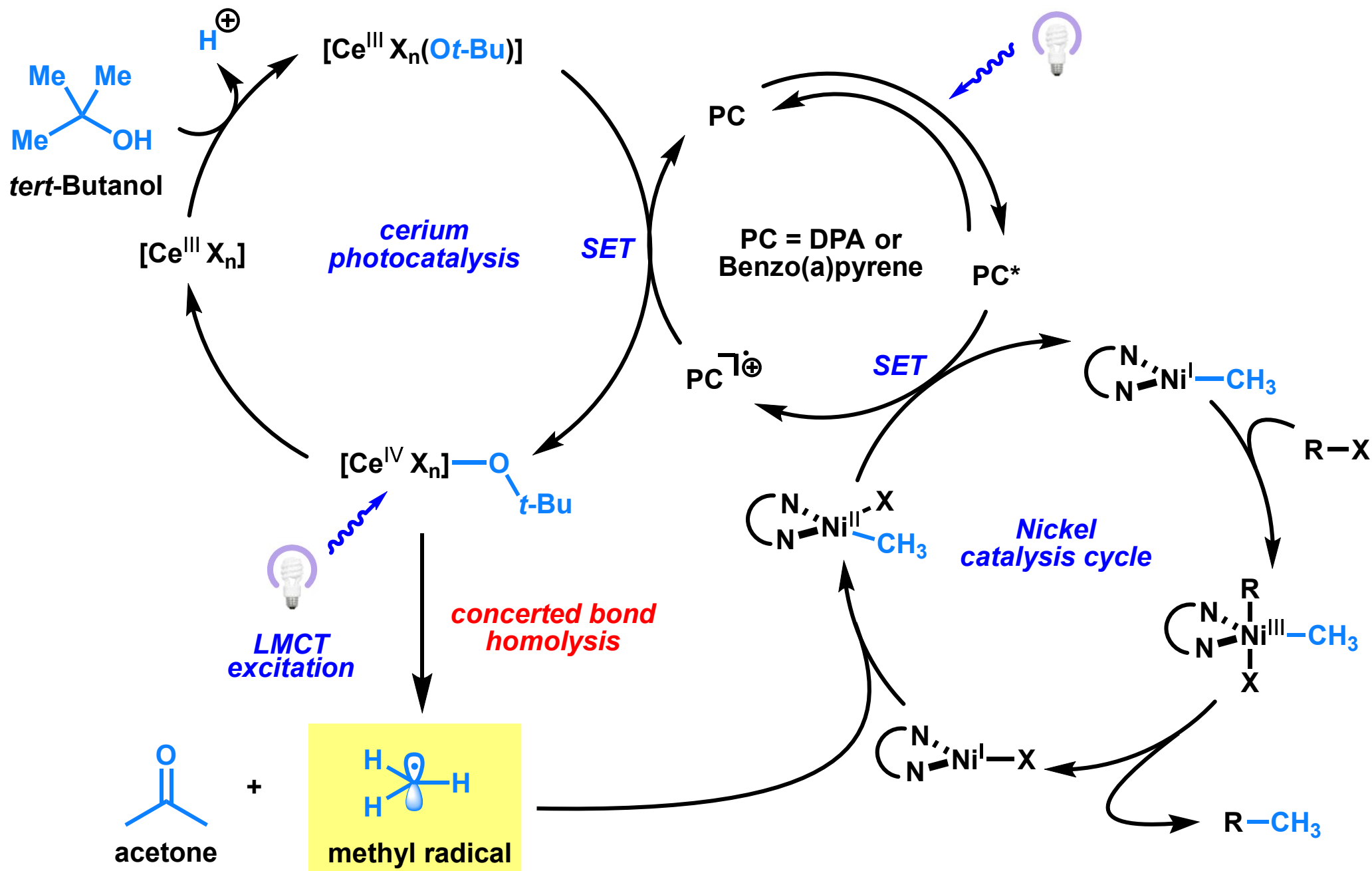


Investigation of Bond Homolysis Selectivity

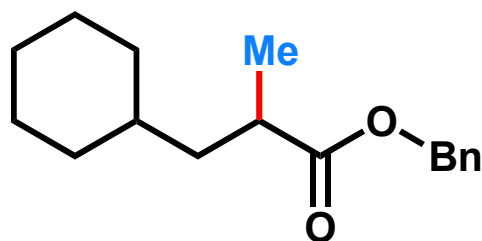
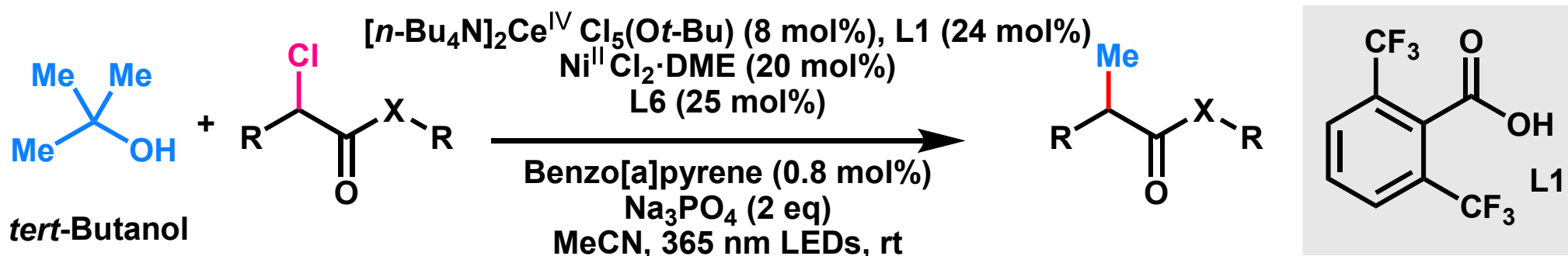


Ce catalyst	ligand	product distribution scission		
		A	B	C
Ce ^{III} Cl ₃	<i>n</i> -Bu ₄ NCl	0.6%	11%	62%
Ce ^{III} Cl ₃	 L1	8%	19%	28%
Ce ^{IV} ₆ (L1) ₁₂ O ₄ (OH) ₄ (H ₂ O) ₃		9%	23%	32%

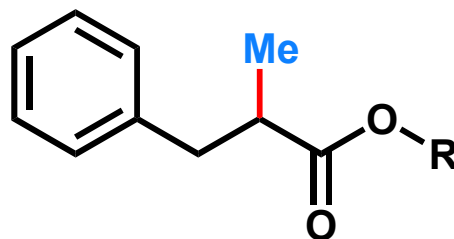
Proposed Reaction Mechanism



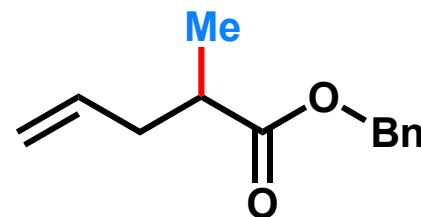
Substrate Scope (1) : Sp^3 - Sp^3 coupling



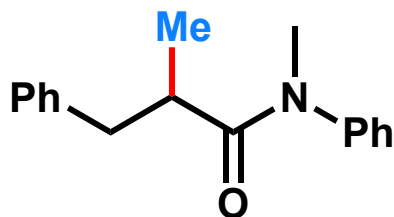
69% yield



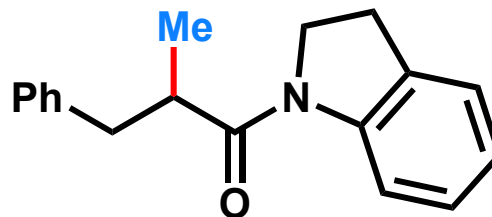
R = Bn: 64% yield
R = Me: 66% yield



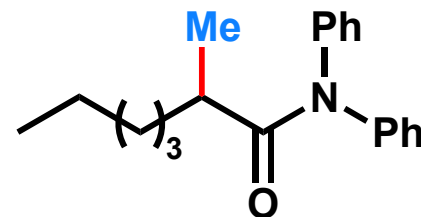
76% yield



55% yield

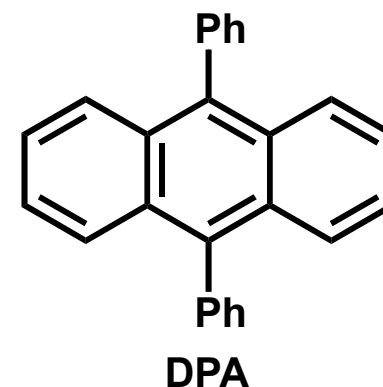
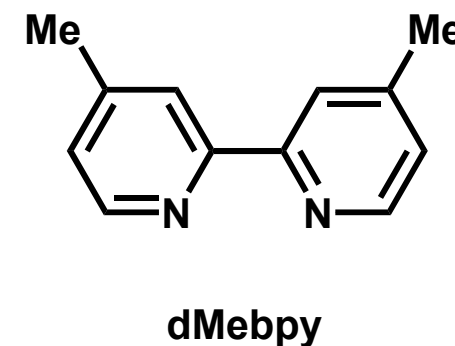
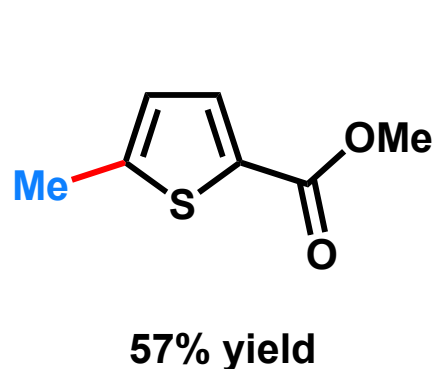
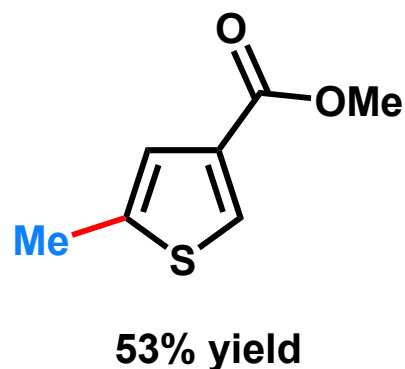
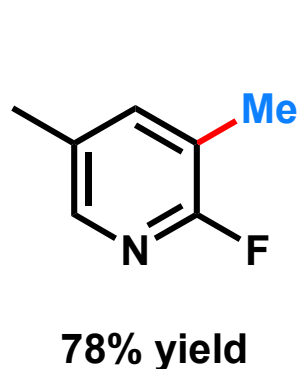
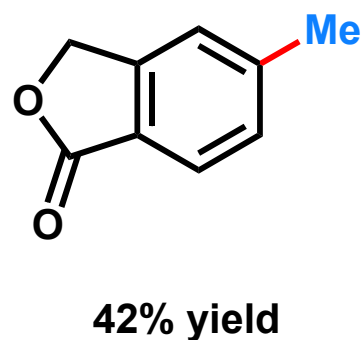
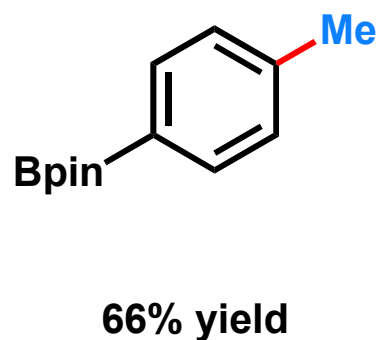
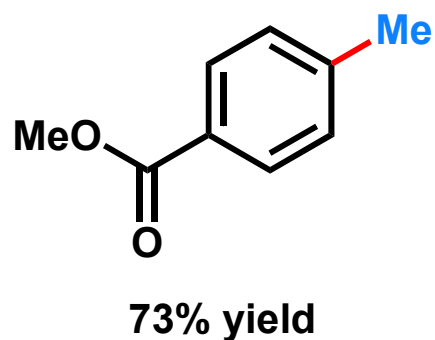
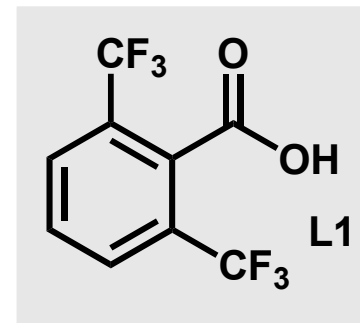
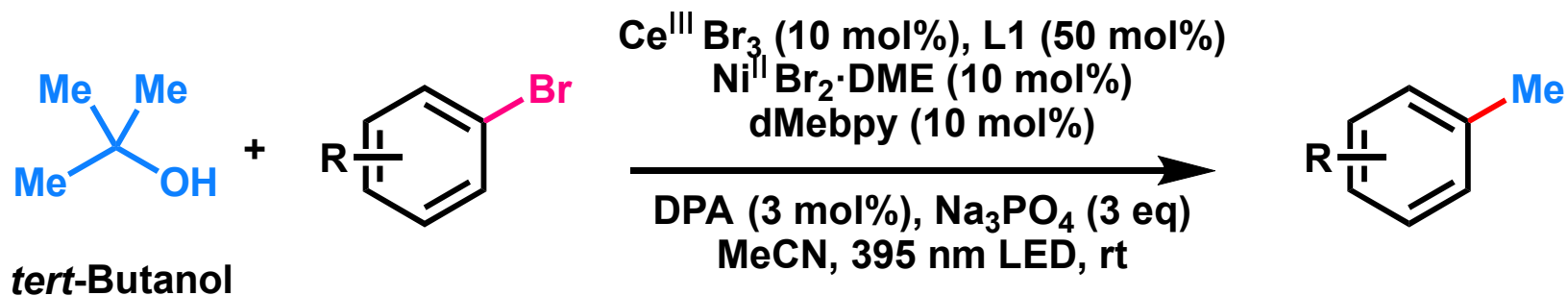


47% yield

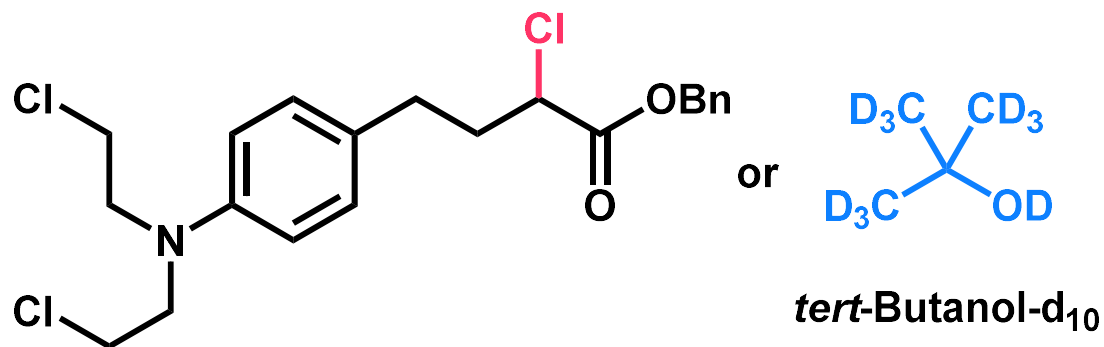


51% yield

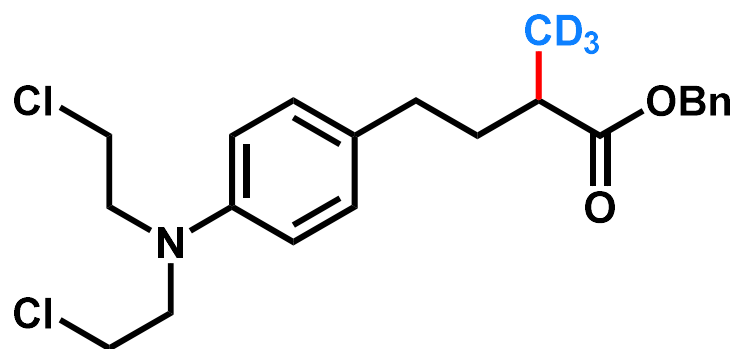
Substrate Scope (2) : Sp^3 - Sp^2 coupling



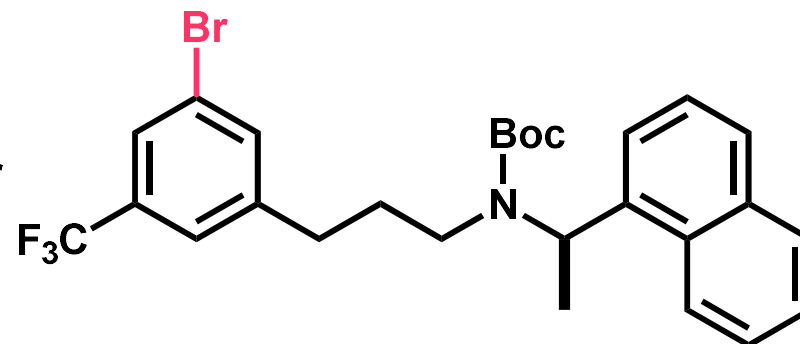
Application for Trideuteromethylation



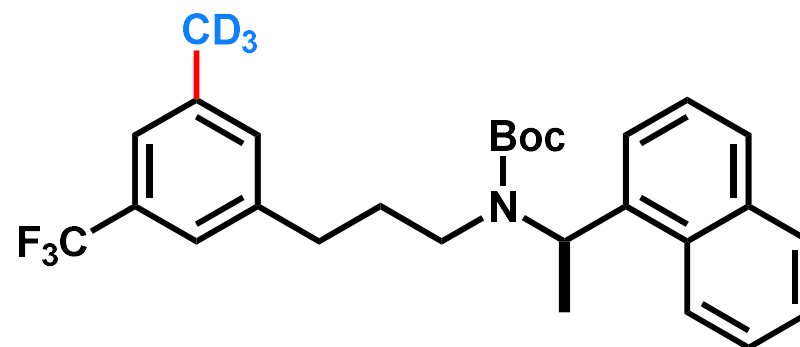
$[n\text{-Bu}_4\text{N}]_2\text{Ce}^{\text{IV}}\text{Cl}_6$ (8 mol%)
L1 (24 mol%)
 $\text{NiCl}_2\cdot\text{DME}$ (20 mol%)
L6 (25 mol%)
Benzo[a]pyrene (0.8 mol%)
 Na_3PO_4 (3 eq)
MeCN, 365 nm LED, rt



43% yield, >99% D-inc



$\text{Ce}^{\text{III}}\text{Br}_3$ (10 mol%)
L1 (50 mol%)
 $\text{NiBr}_2\cdot\text{DME}$ (10 mol%)
dMebpy (10 mol%)
DPA (3 mol%)
 Na_3PO_4 (3 eq)
MeCN, 395 nm LED, rt



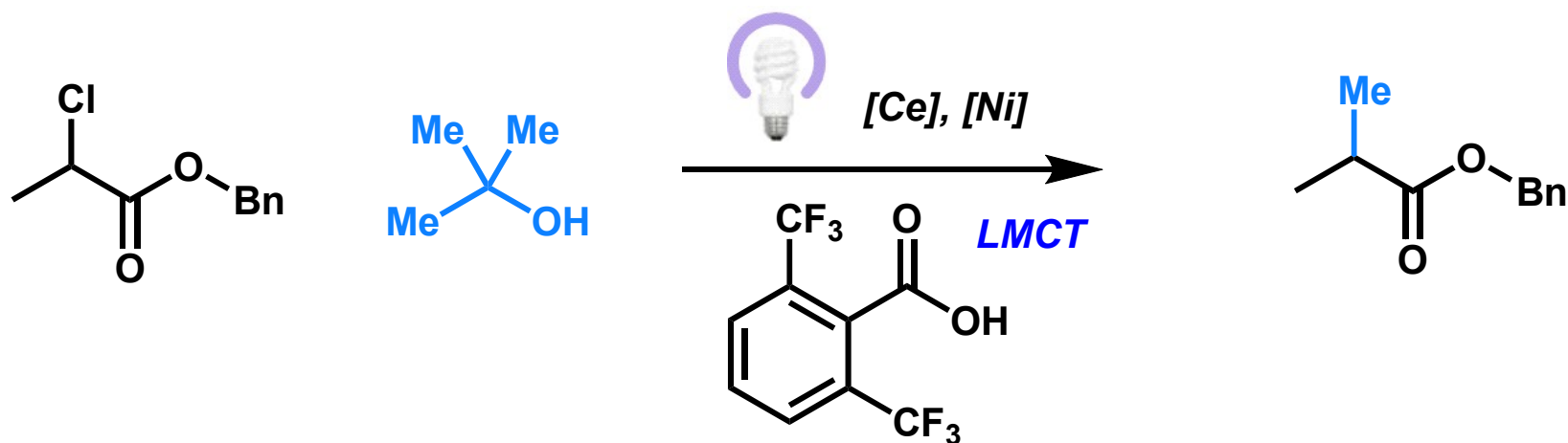
56% yield, >99% D-inc

Summary

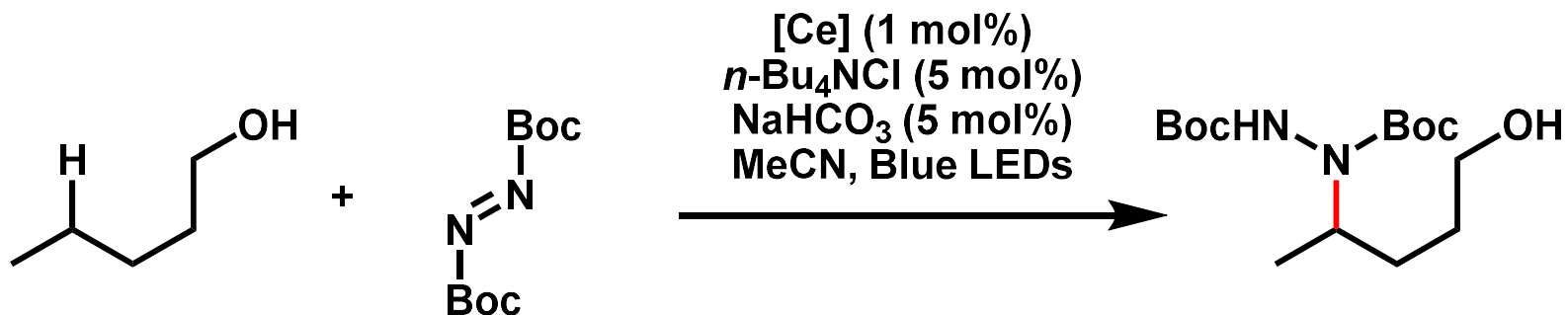
1. Alcohol-Alcohol ($\text{sp}_3\text{-sp}_3$) Cross-Coupling (Me-Donor: **Methanol**)



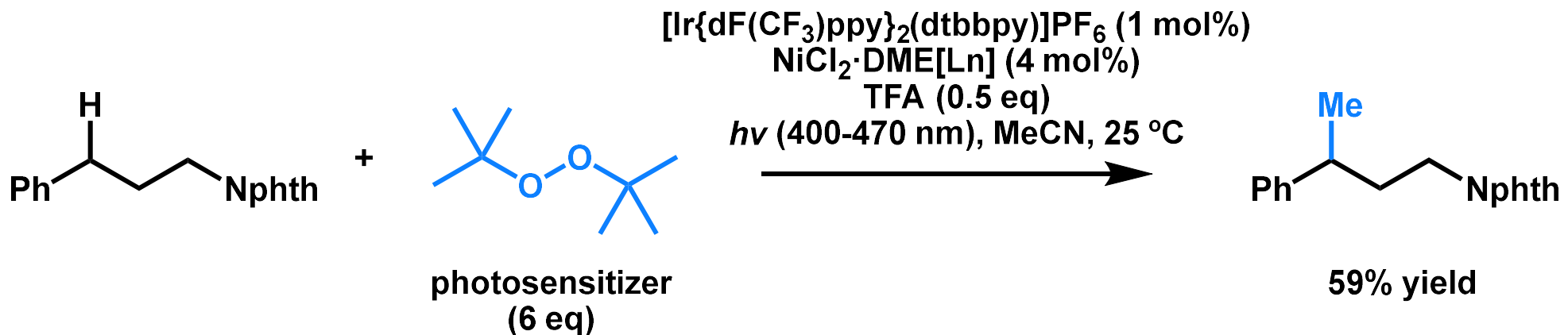
2. Methylation Cross-Coupling (Me-Donor: **tert-Butanol**)



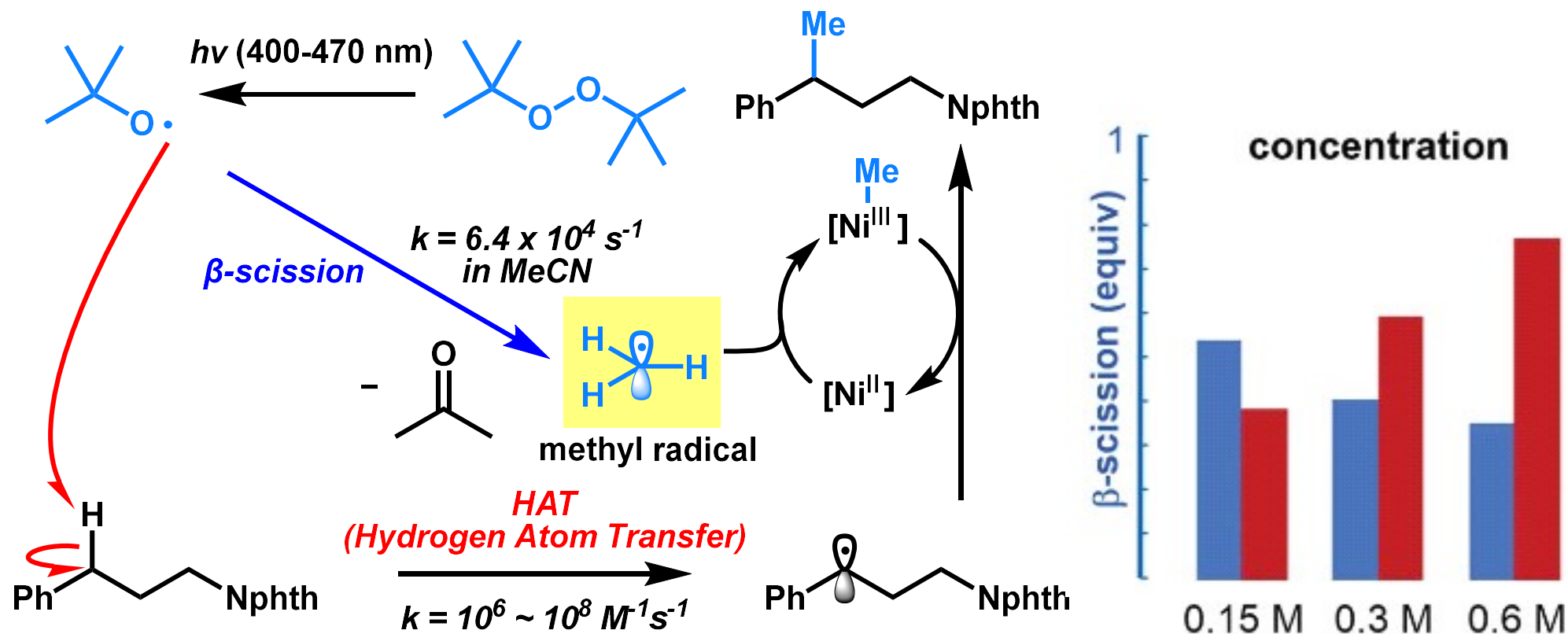
Appendix. Ce(III) and Ce(IV)



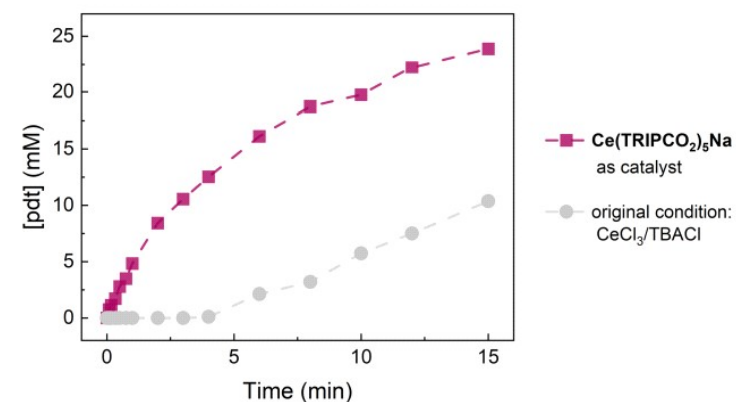
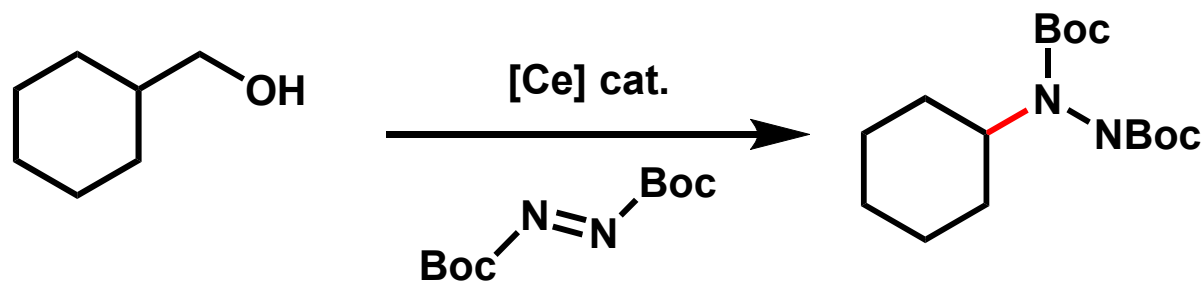
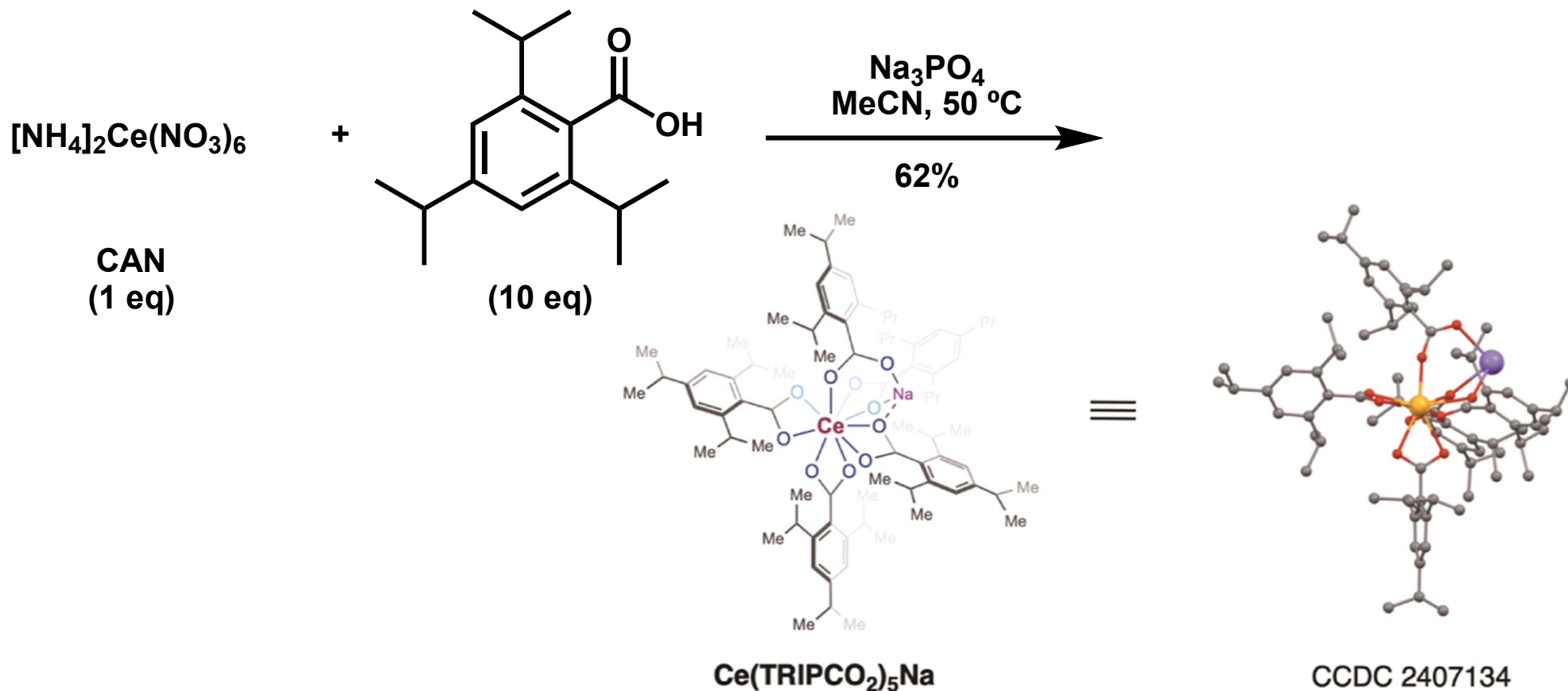
entry	[Ce]	Light	Yield (%)
1	(<i>n</i> -Bu ₄ N) ₂ [Ce ^{IV} Cl ₆]	blue LED	89%
2	(<i>n</i> -Bu ₄ N) ₂ [Ce ^{IV} Cl ₆]	off (60 °C)	0%
3	Ce ^{III} Cl ₃	blue LED	92%



Vasilopoulos, A.; Krska, S. W.; Stahl, S. S. *Science* 2021, 372, 398.



Kinetic Investigation of Ce(IV)-Complex



UV-vis absorption spectra of in situ formed [Ce]

