

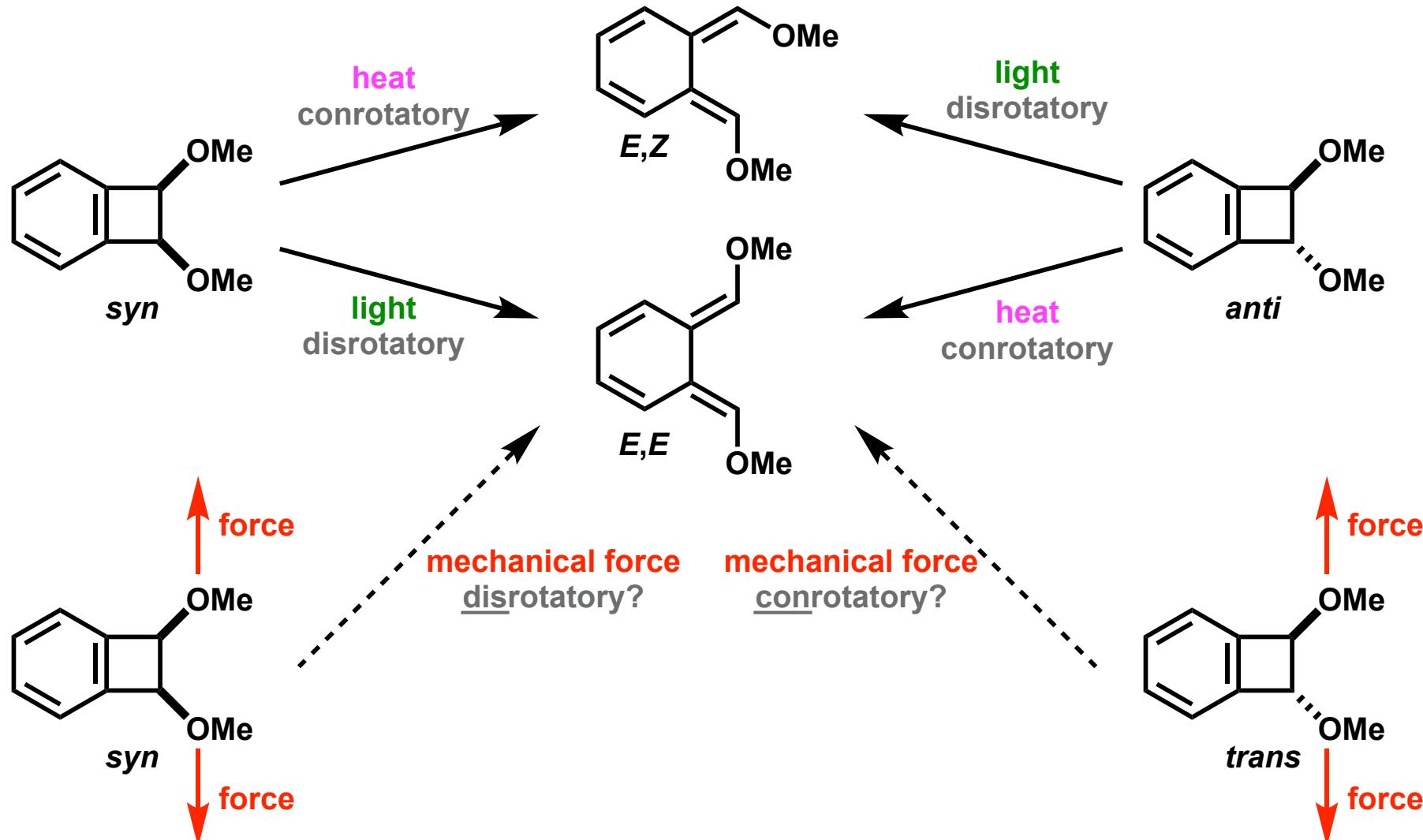
# **Mechanical Force Cleaves Specific Covalent Bond**

**2025.5.31. Literature Seminar  
D3 Hibiki Asai**

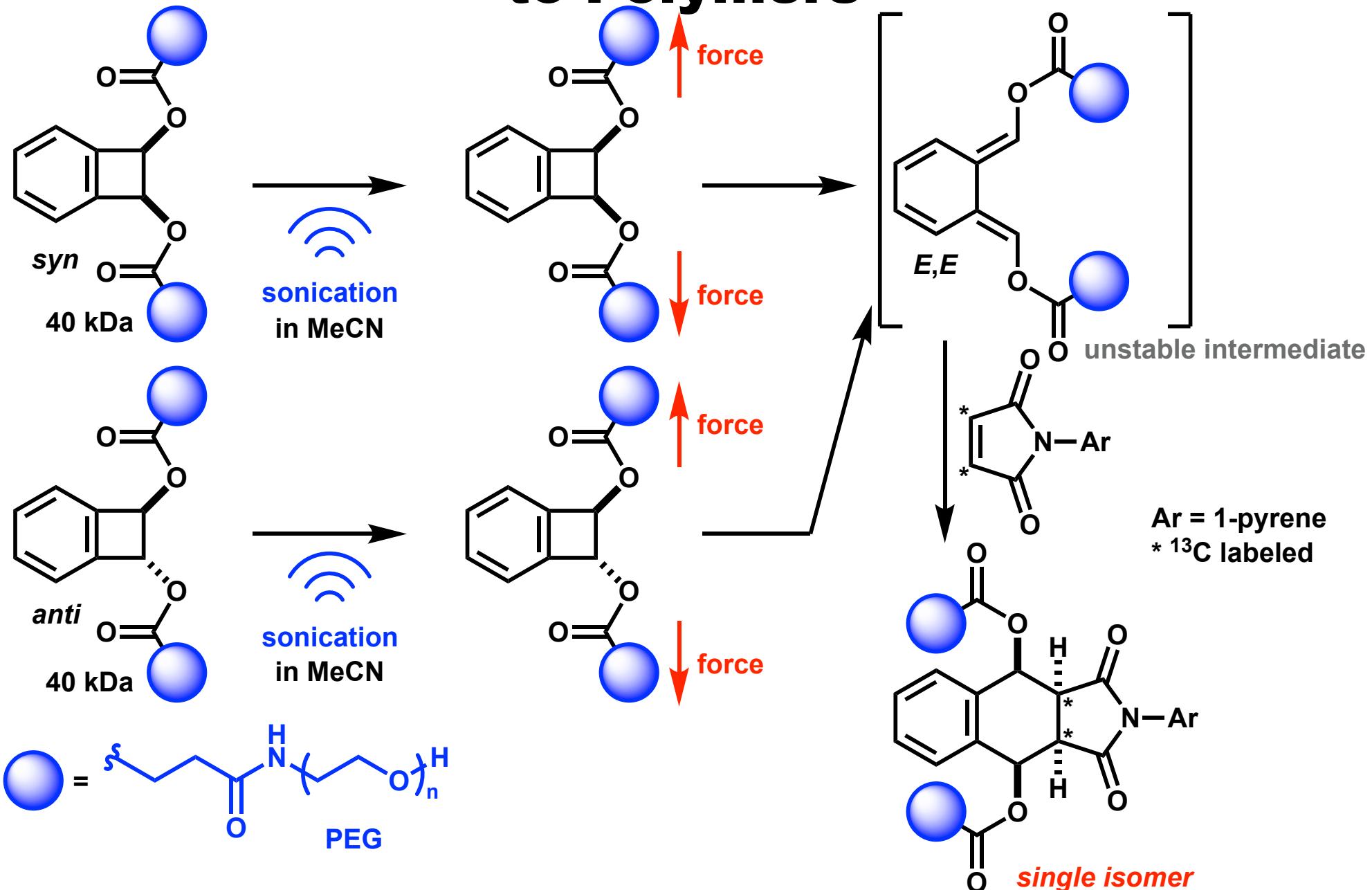
# **Contents**

- 1. Introduction (Mechanochemistry in Polymer)**
- 2. Mechanically Triggered Chemiluminescence  
(Main paper)**

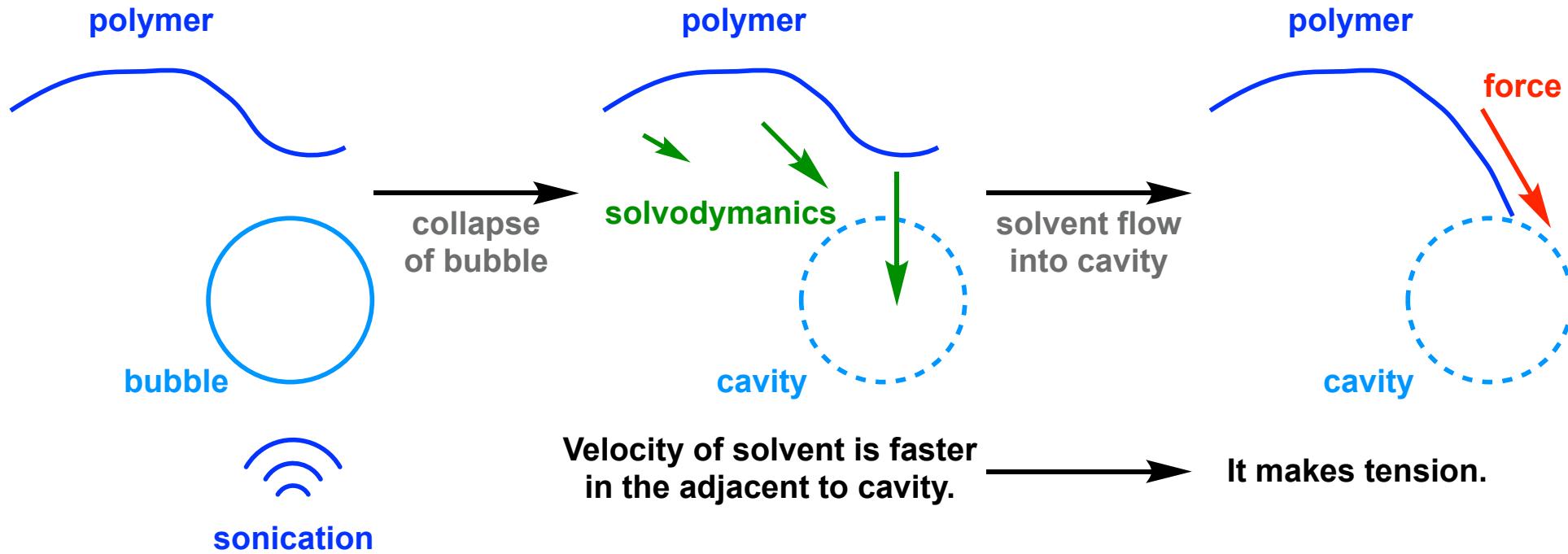
# Can Mechanical Force Overcome Woodward–Hoffmann Rules?



# Ultrasound Transmits Mechanical Force to Polymers

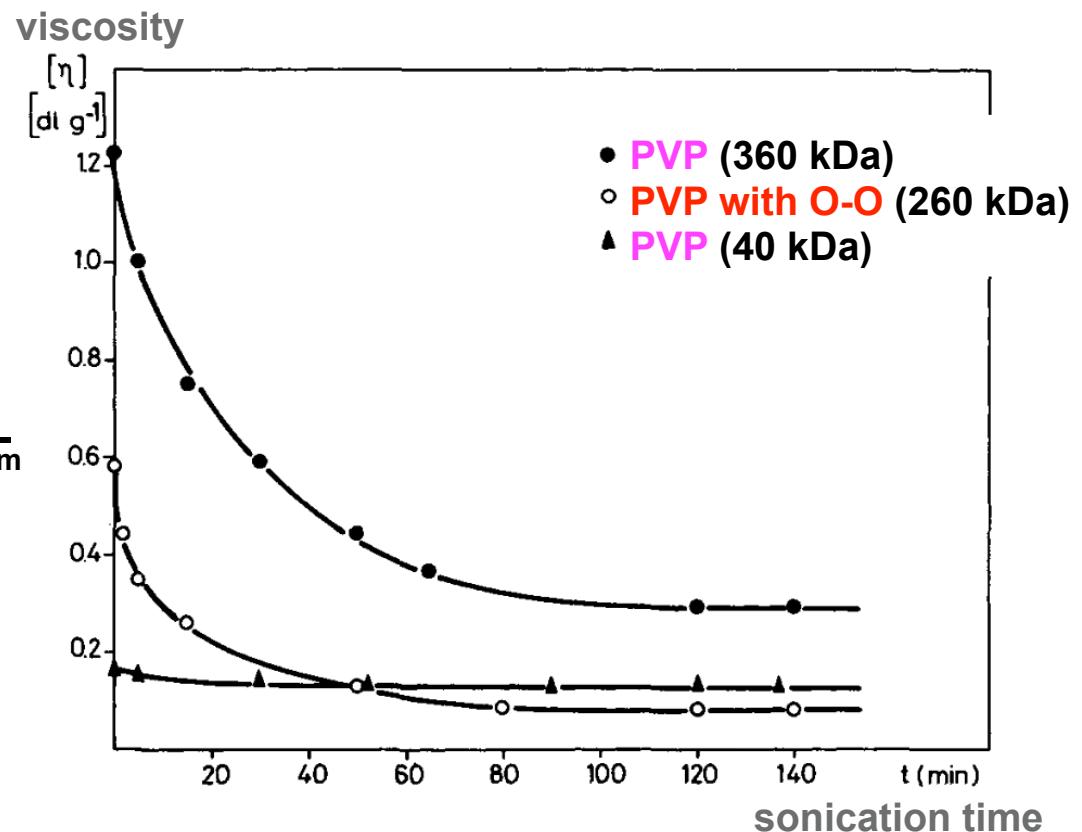
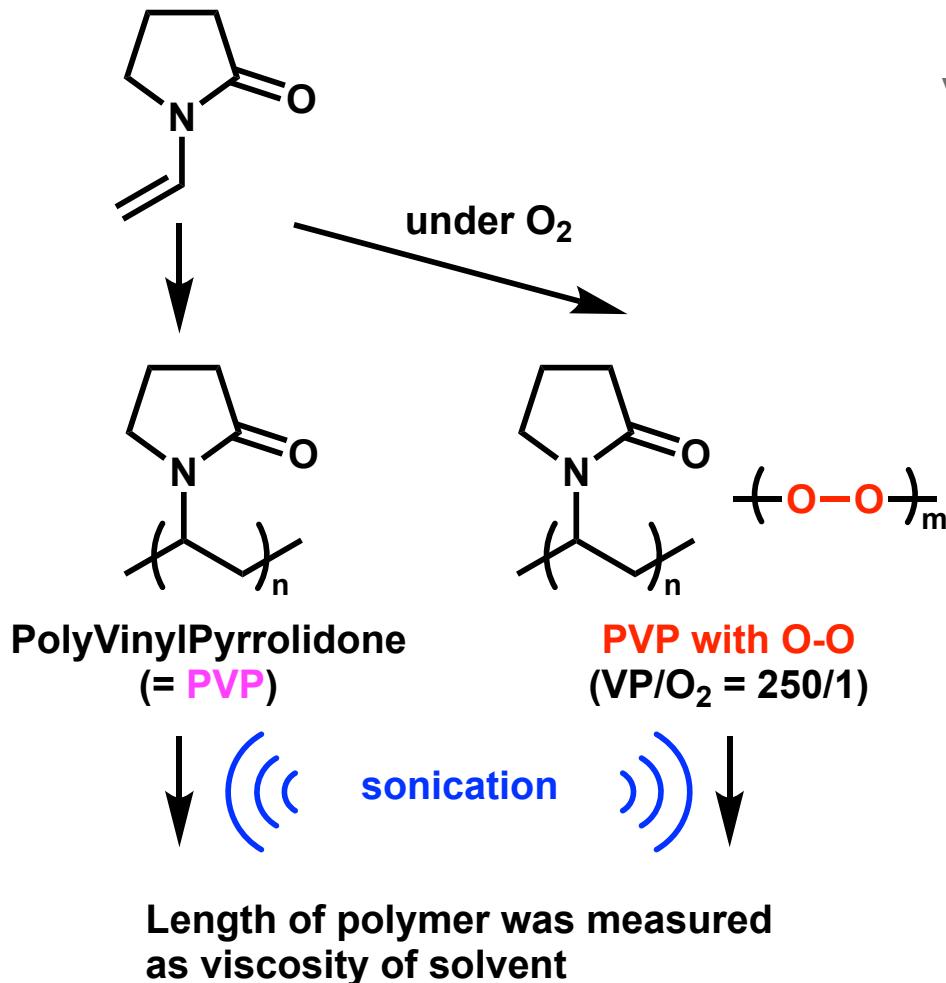


# Accepted Mechanism of Polymer Cleavage by Sonication



- Threshold of polymer length: ~30 kDa
- Cleavage point: Center of polymer
- Efficacy of polymer cleavage: ↑sound intensity, ↓solvent temperature, ↓polymer concentration

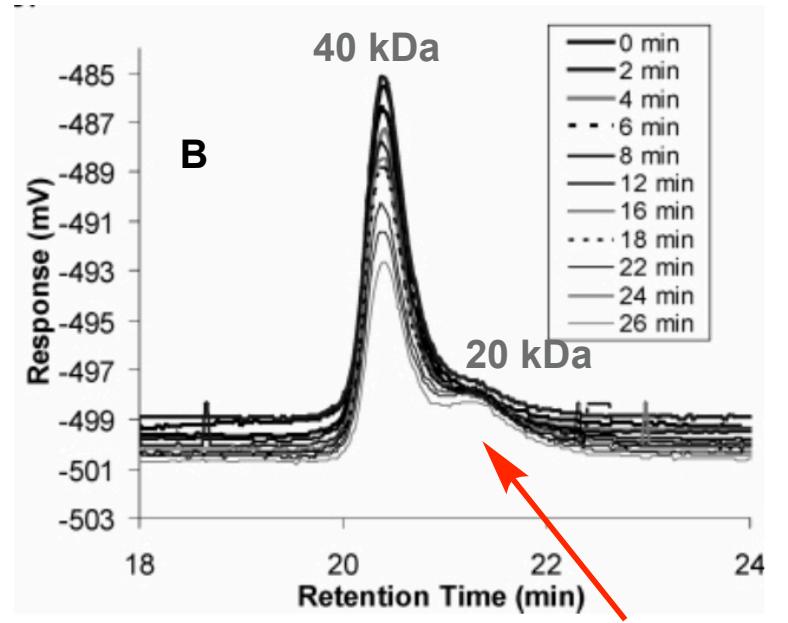
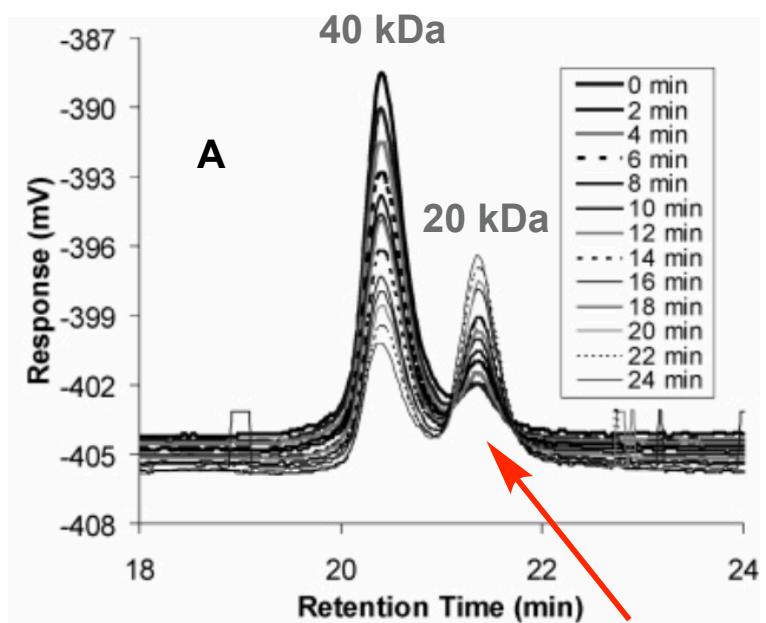
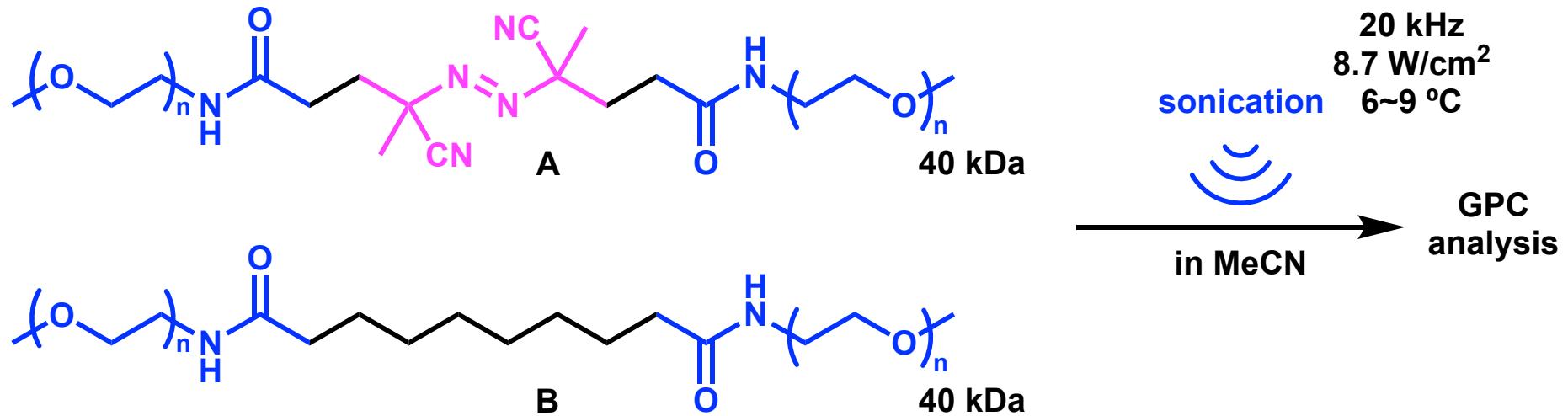
# Ultrasound Cleaves Weak Bond



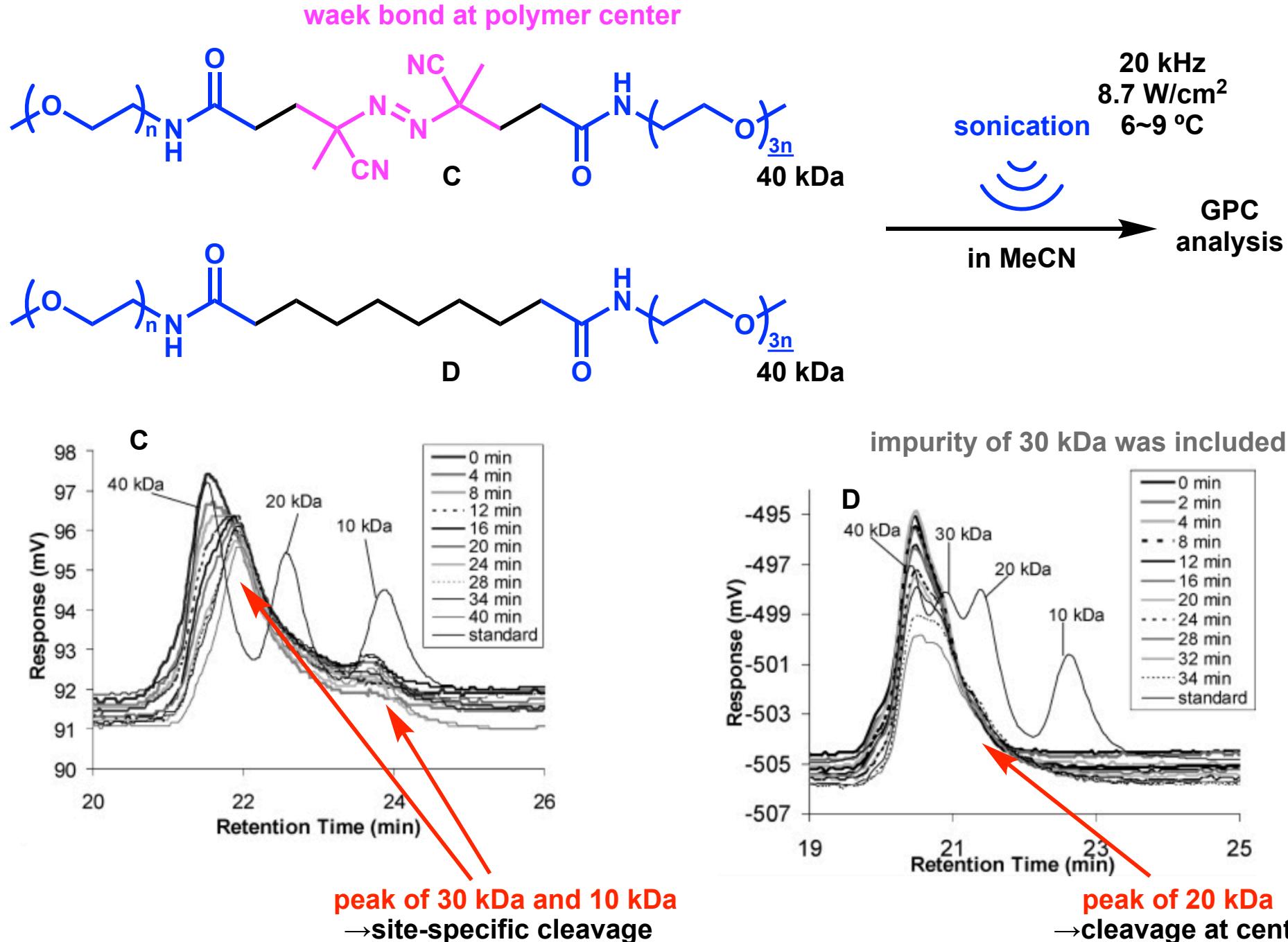
- PVP with O-O is cleaved faster.
- PVP with O-O has smaller threshold for cleavage.

# Site-Specific Cleavage by Ultrasound (I)

weak bond at polymer center

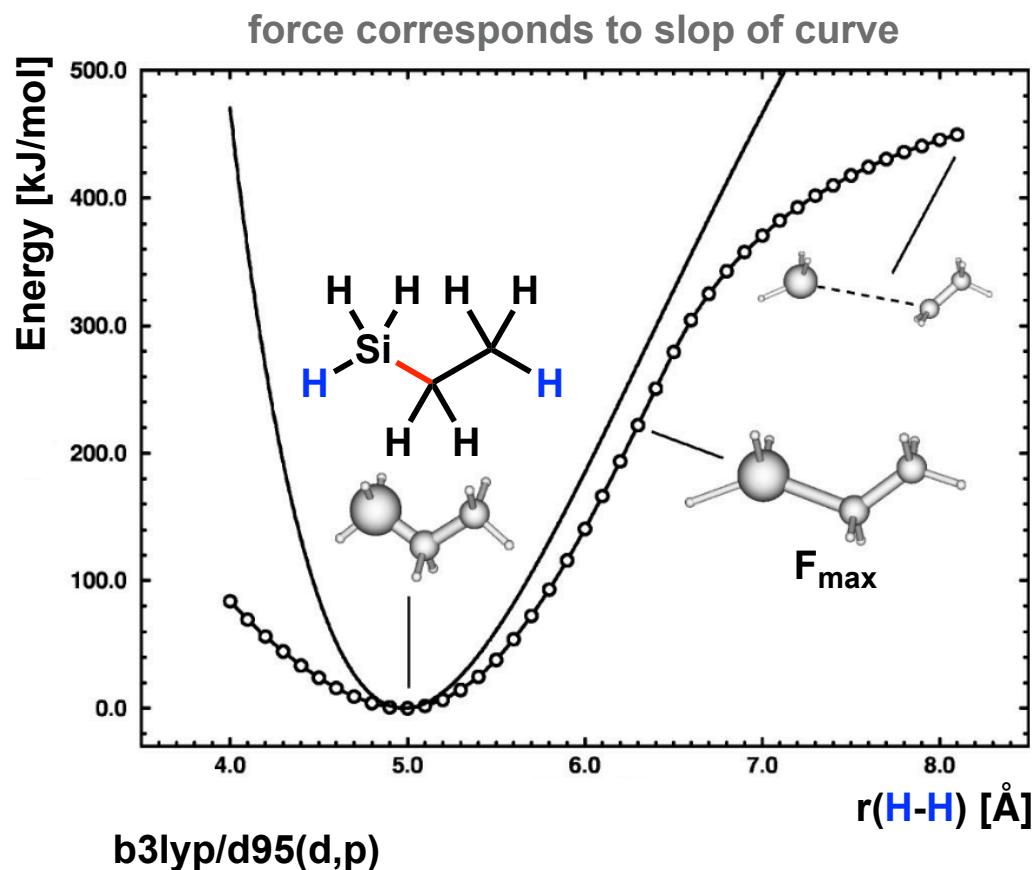


# Site-Specific Cleavage by Ultrasound (II)



# Bond “Strength” in Bond Cleavage by Force

	$r_0$ [Å]	BDE [kcal/mol]	$F_{max}$ [nN]
H—H	0.743	112.7	8.31
Cl—Cl	2.045	50.8	4.60
O=O	1.221	121.6	14.89



	BDE [kcal/mol]	$F_{max}$ [nN]
$\text{H}_3\text{C}-\text{H}$	88.6	6.92
$\text{H}_3\text{C}-\text{O}-\text{H}$	82.3	7.56
$\text{H}_3\text{Si}-\text{H}$	80.6	4.78
$\text{H}_3\text{Si}-\text{O}-\text{H}$	101.7	5.20
$\text{H}_3\text{Si}-\text{SiH}_3$	70.8	3.28

# **Contents**

- 1. Introduction (Mechanochemistry)**
- 2. Mechanically Triggered Chemiluminescence (Main paper)**

# Introduction of Prof. Maxwell J. Robb



**Prof. Maxwell J. Robb**

**2009 B.S. @ Colorado School of Mines (Prof.)**

**2014 Ph.D @ University of California (Prof. Craig J. Hawker)**

**2014~2017 postdoctoral research@ University of Illinois (Prof. Jeffy S. Moore)**

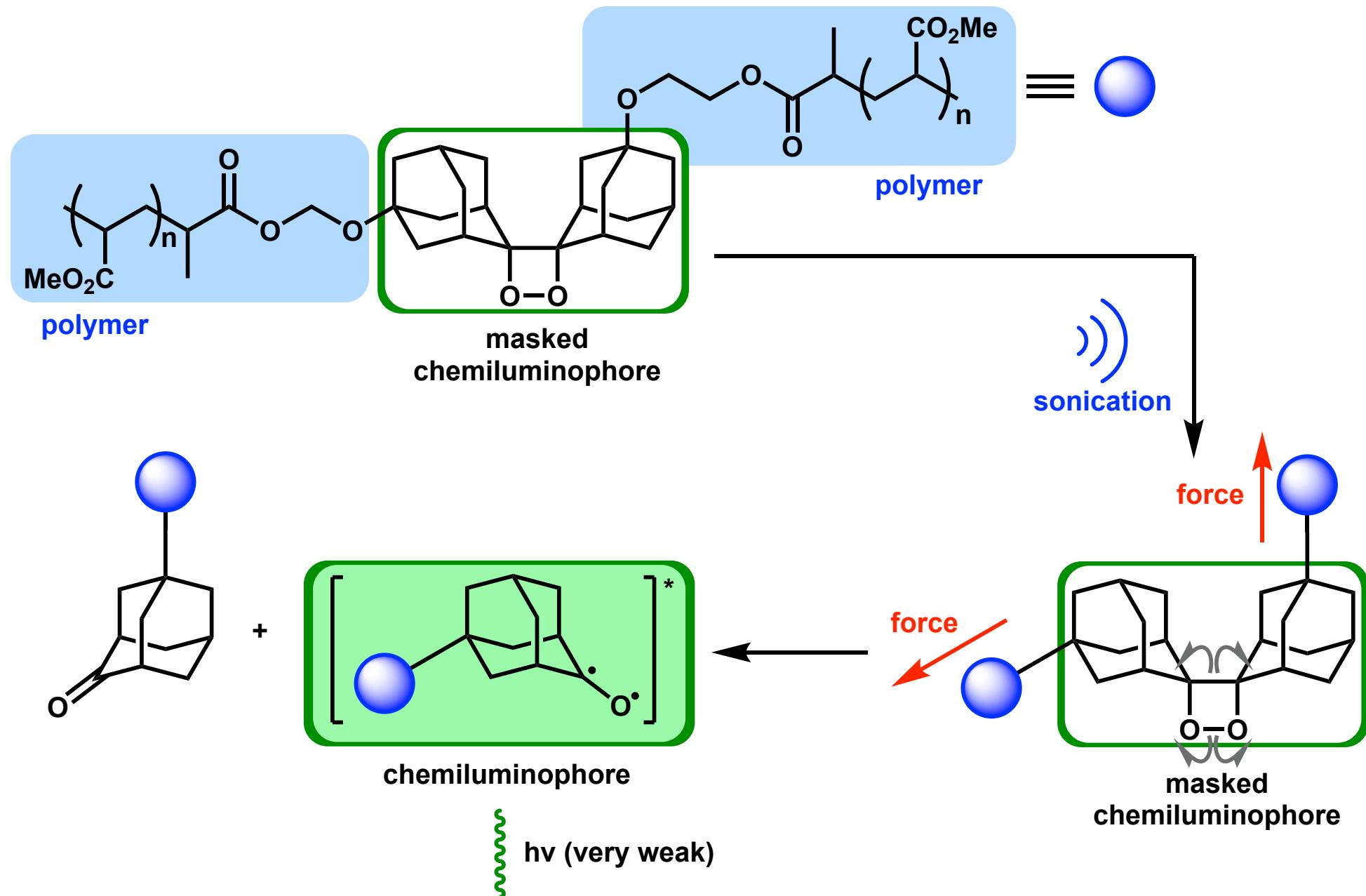
**2017~2024 Assistant Professor @ California Institute of Technology**

**2024~ Full Professor @ California Institute of Technology**

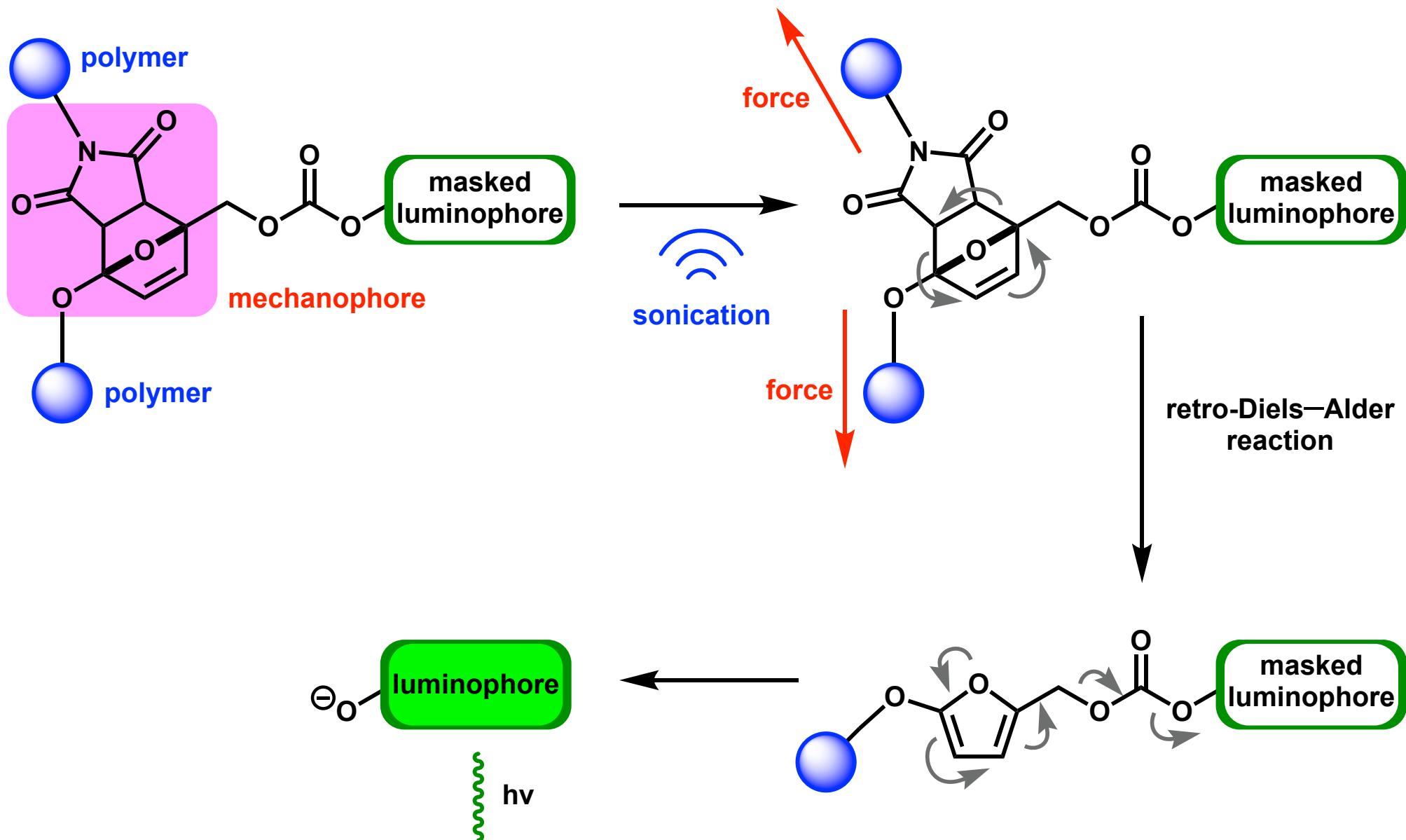
**Research topic: Polymer Mechanochemistry**



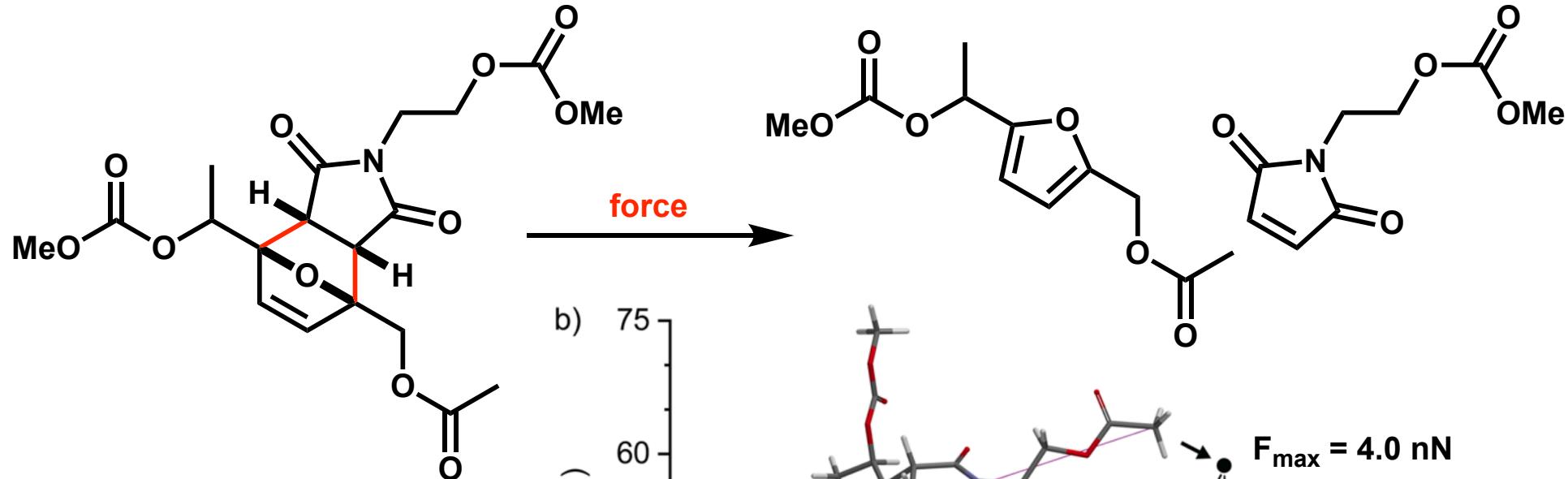
# Landmark Report



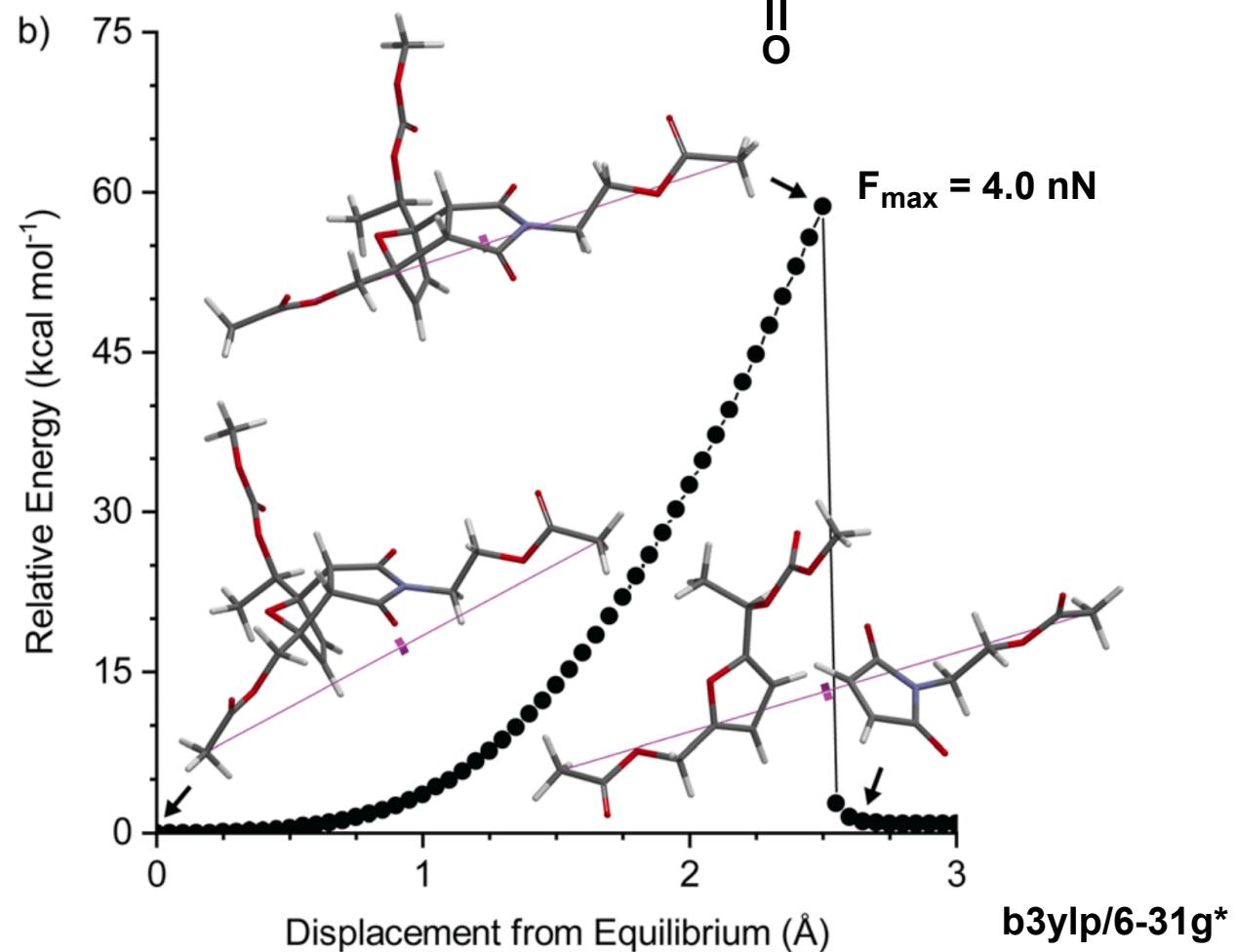
# Concept of This Study



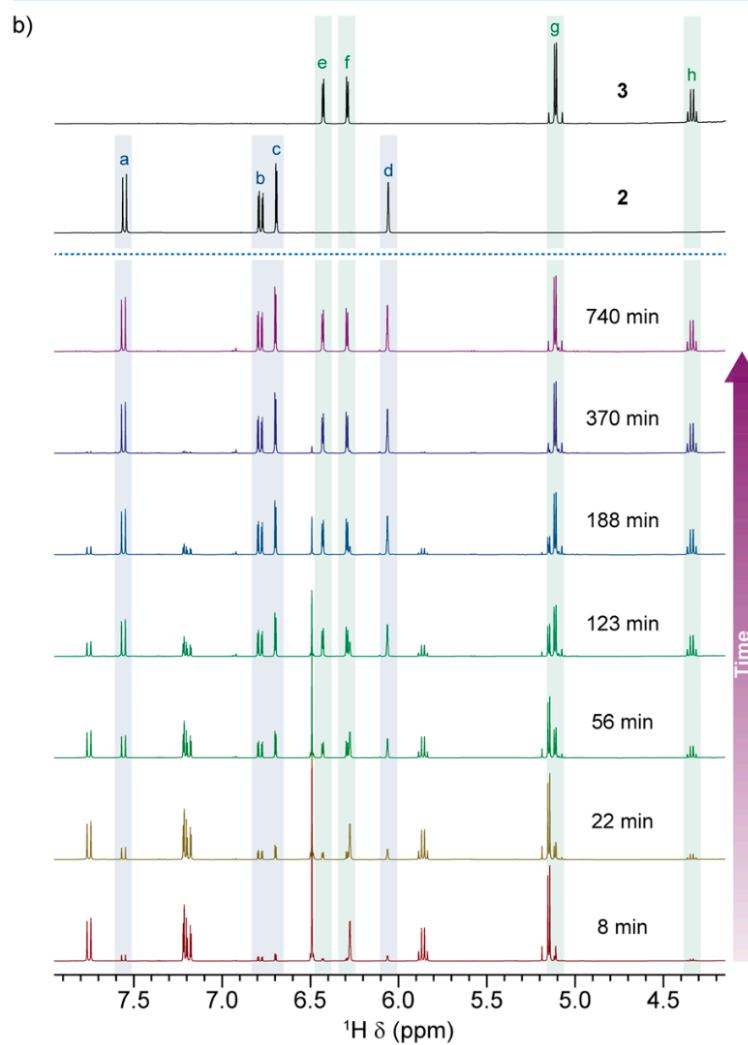
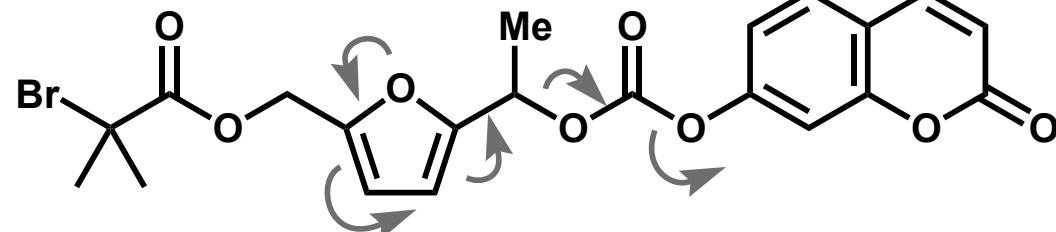
# Mechanophore Unit



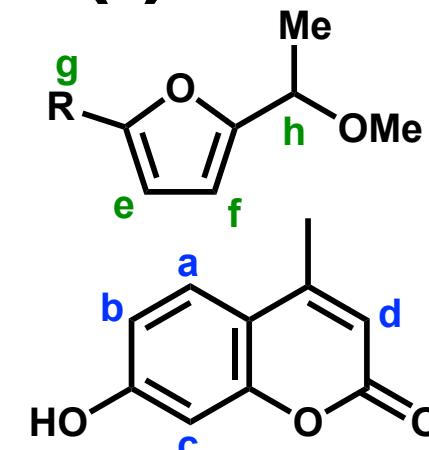
CoGEF method  
calculation



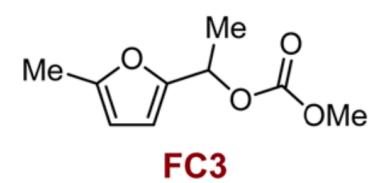
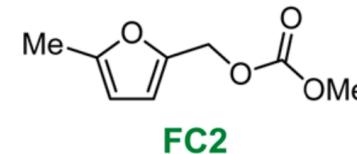
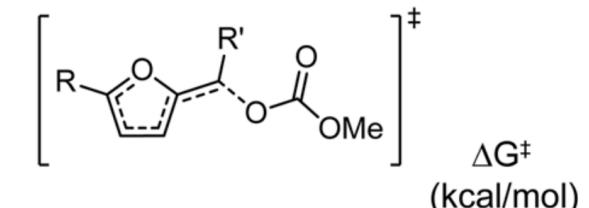
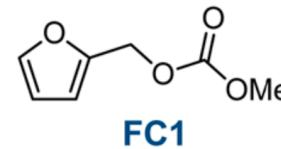
# Chemiluminophore Unit (I)



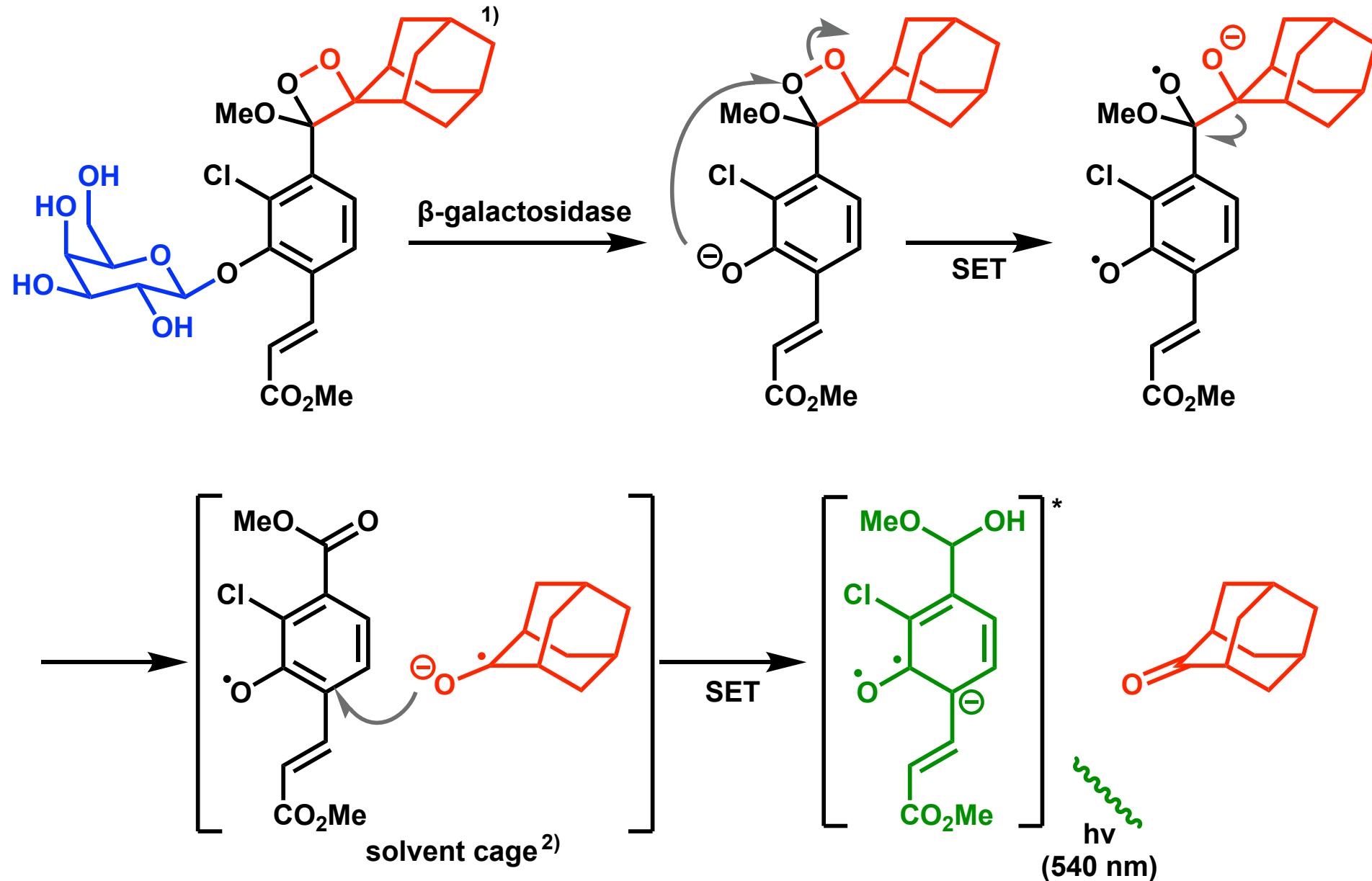
MeCN/MeOH



a)



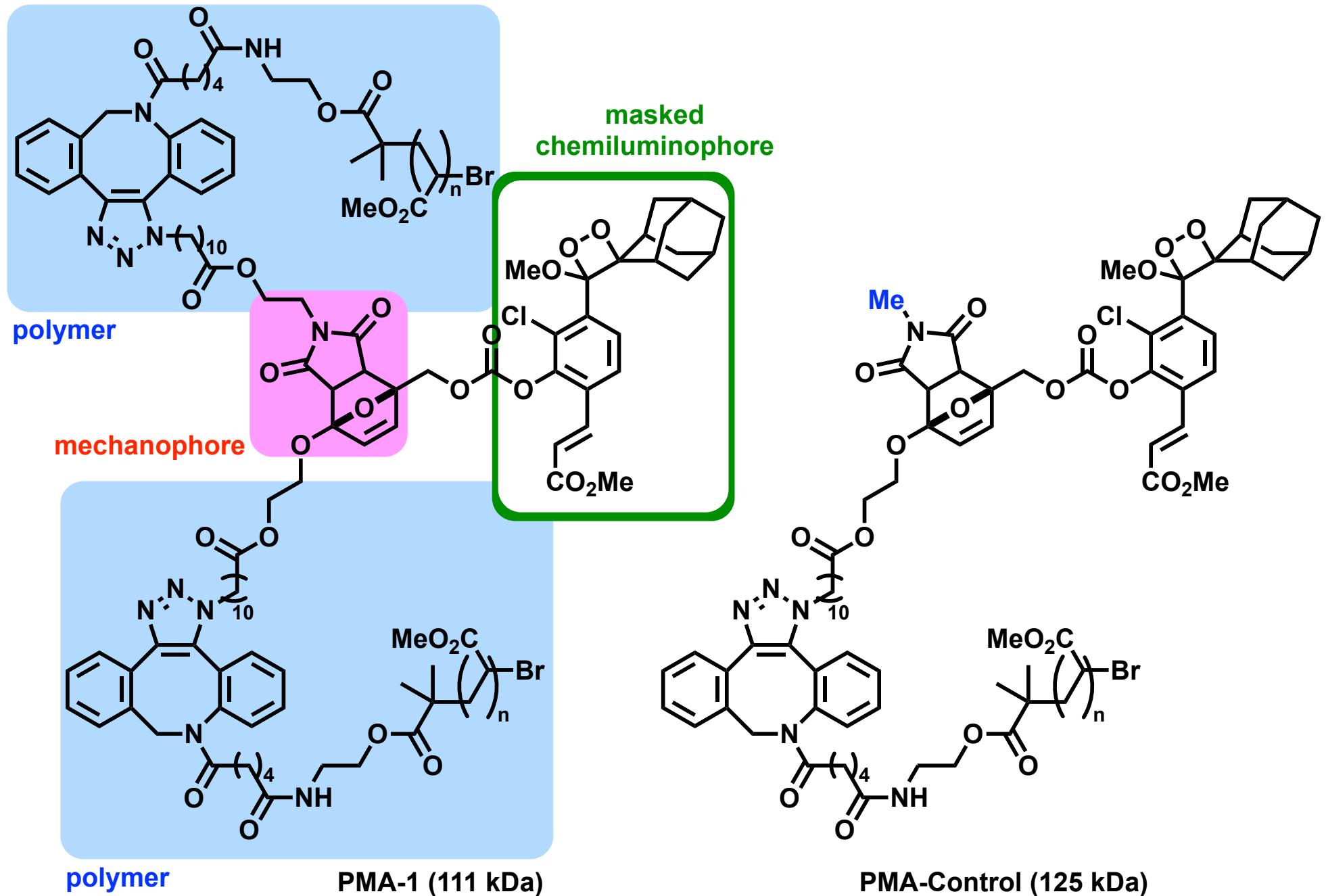
# Chemiluminophore Unit (II)



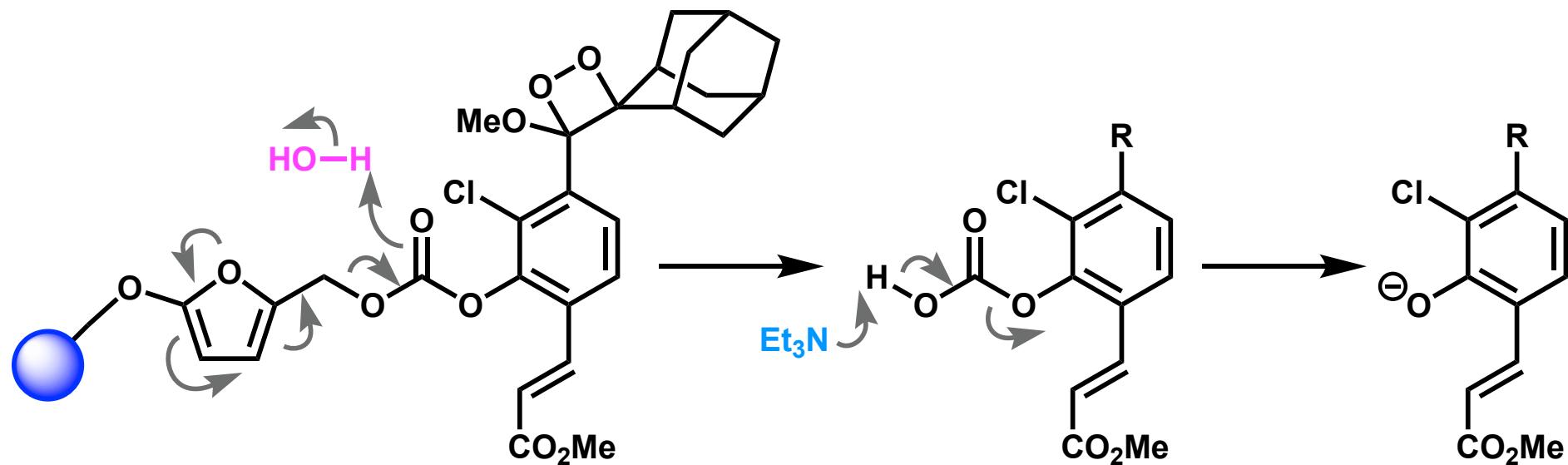
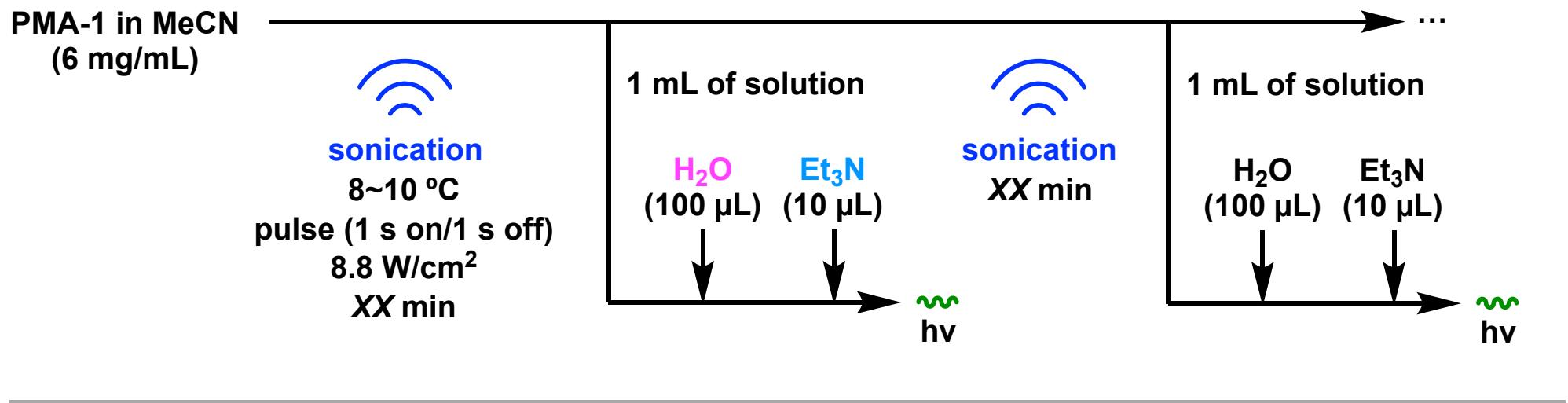
1) Green, O.; Eilon, T.; Hananya, N.; Gutkin, S.; Bauer, C. R.; Shabat, D. *ASC Cent. Sci.* **2017**, 3, 349.

2) Adam, W.; Bronstein, I.; Trofimov, A. V.; Vasil'ev, R. F. *J. Am. Chem. Soc.* **1999**, 121, 958.

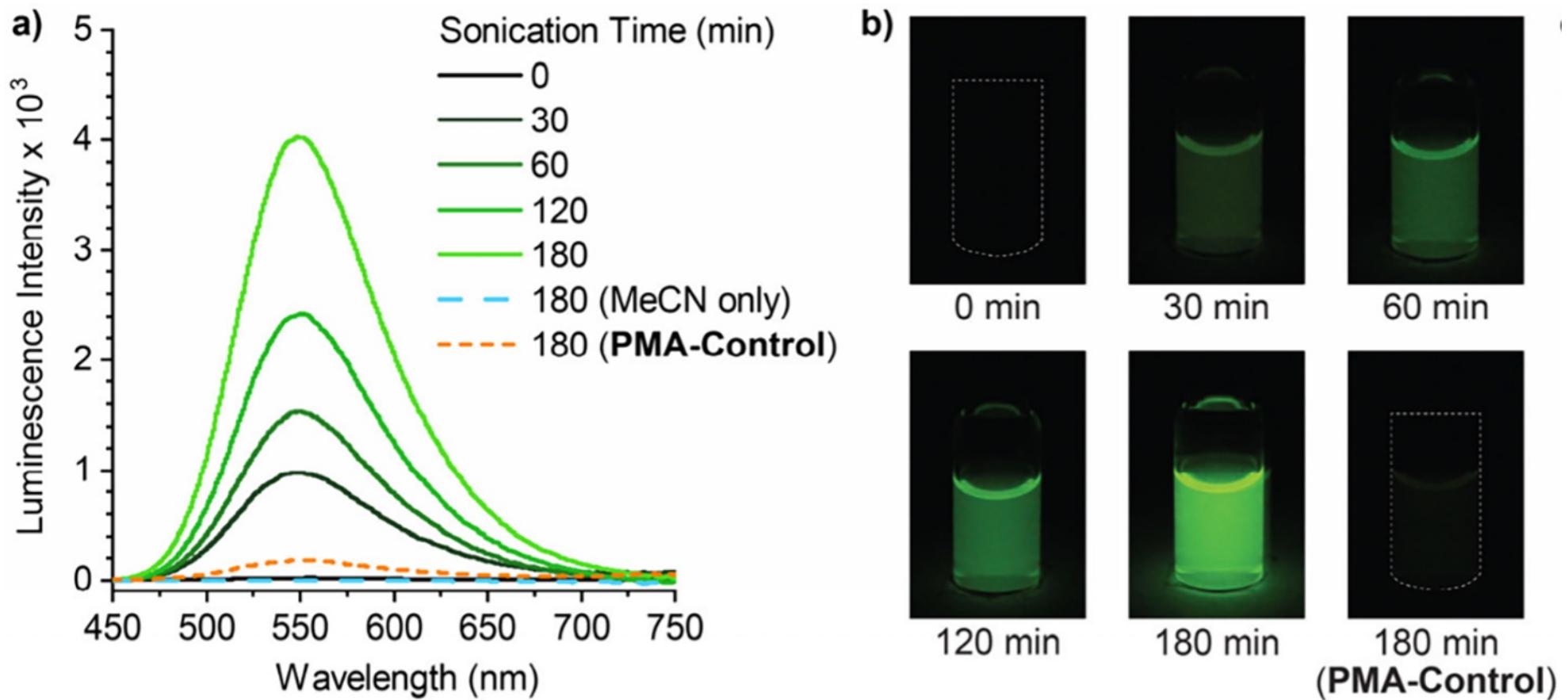
# Design of Mechanoluminophore



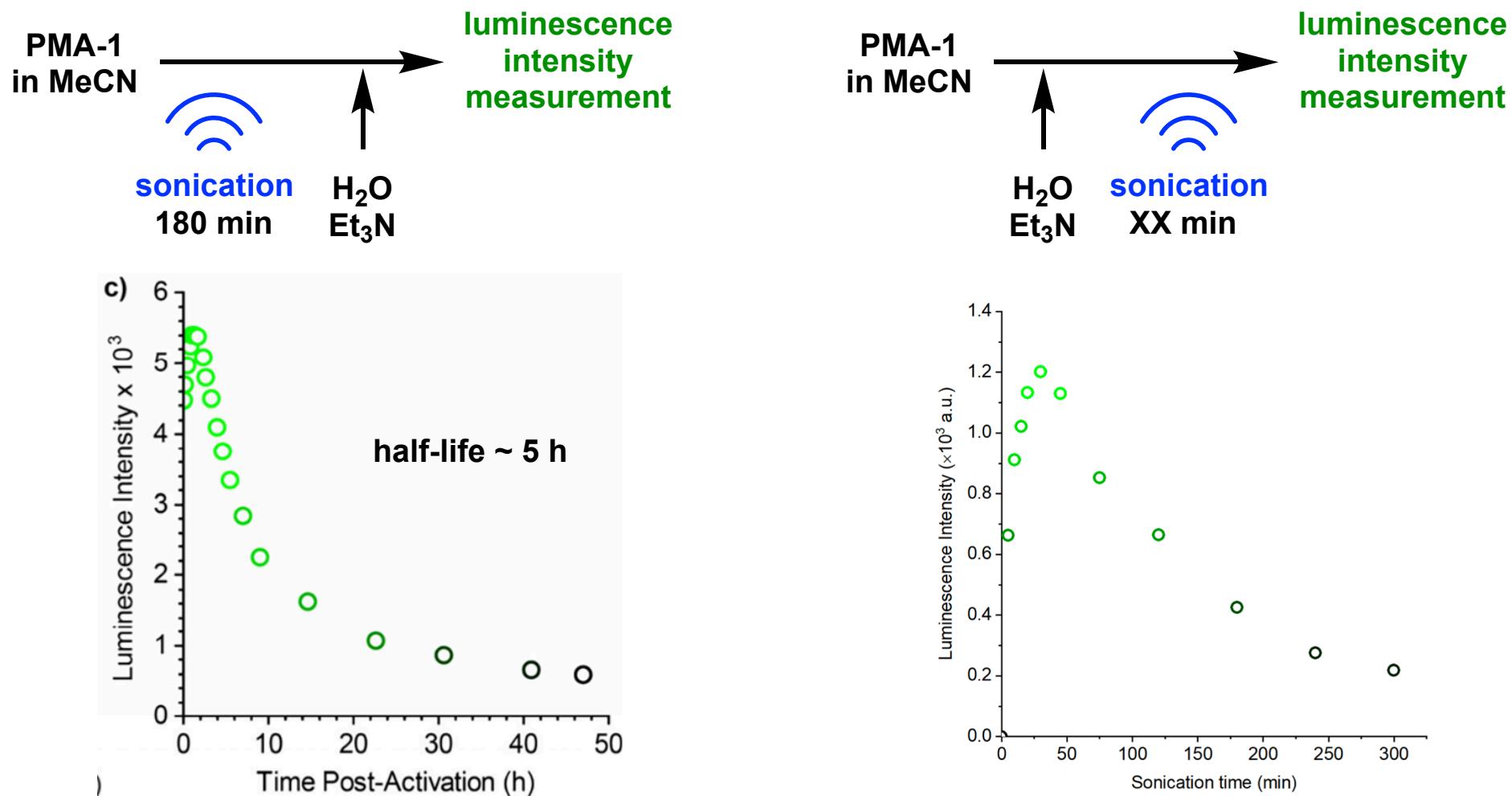
# Experiment Procedure



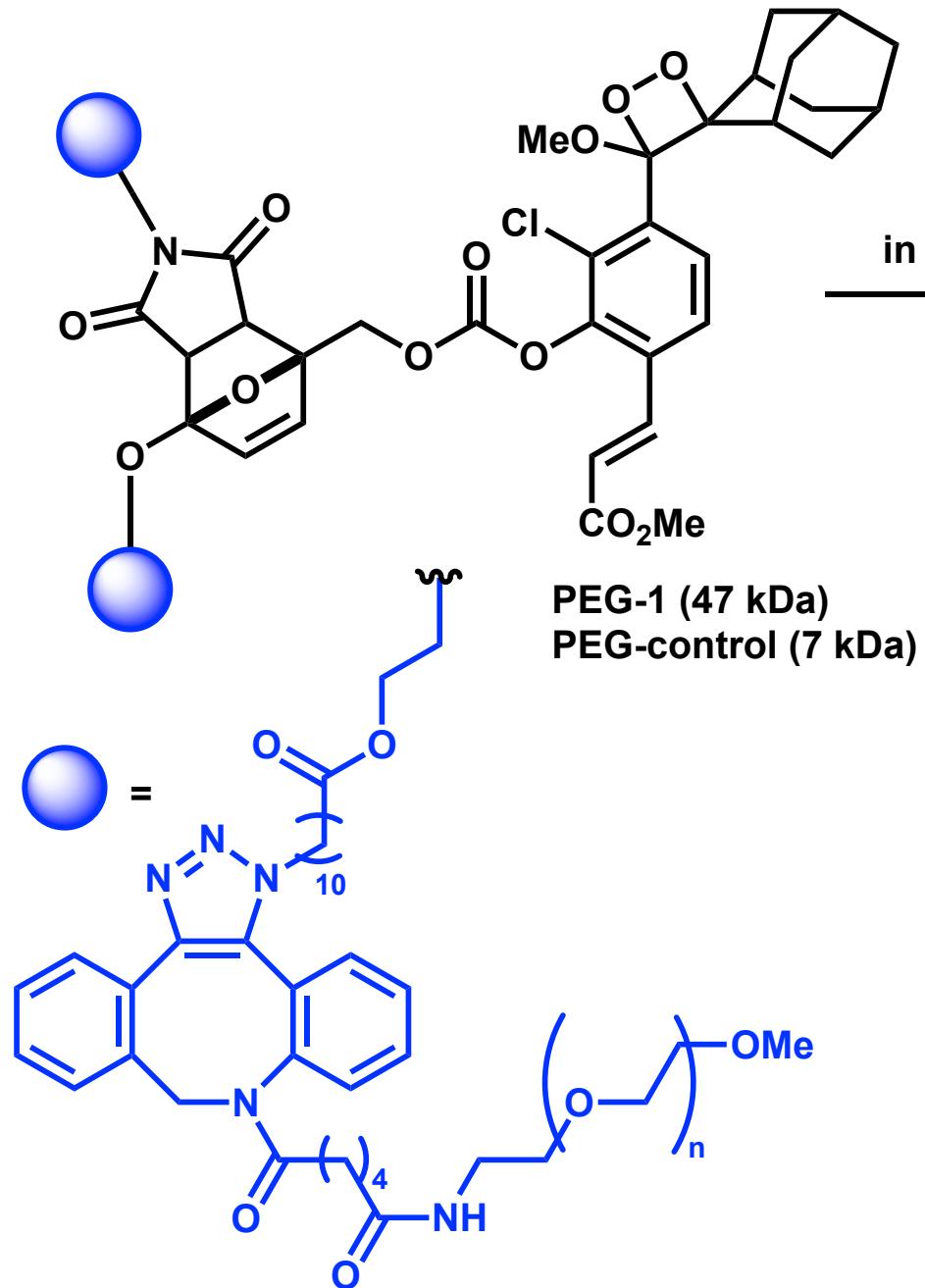
# Luminescence Intensity of Each Sonication Time



# Luminescence Intensity After or During Sonication



# Mechanoluminescence in Biological Condition

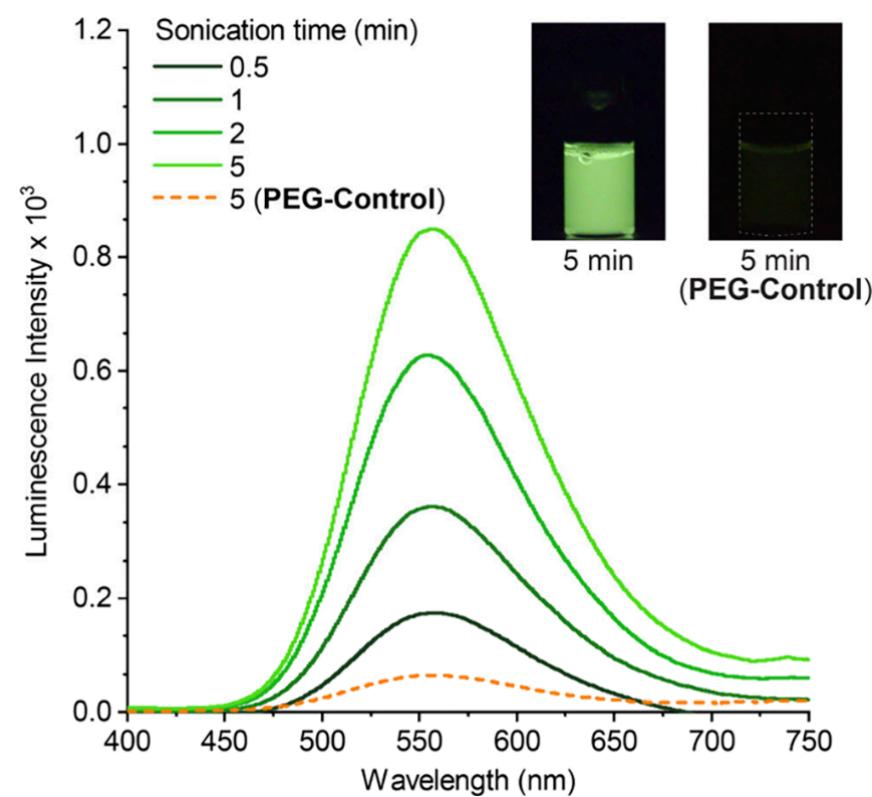


in PBS (pH = 7.4)

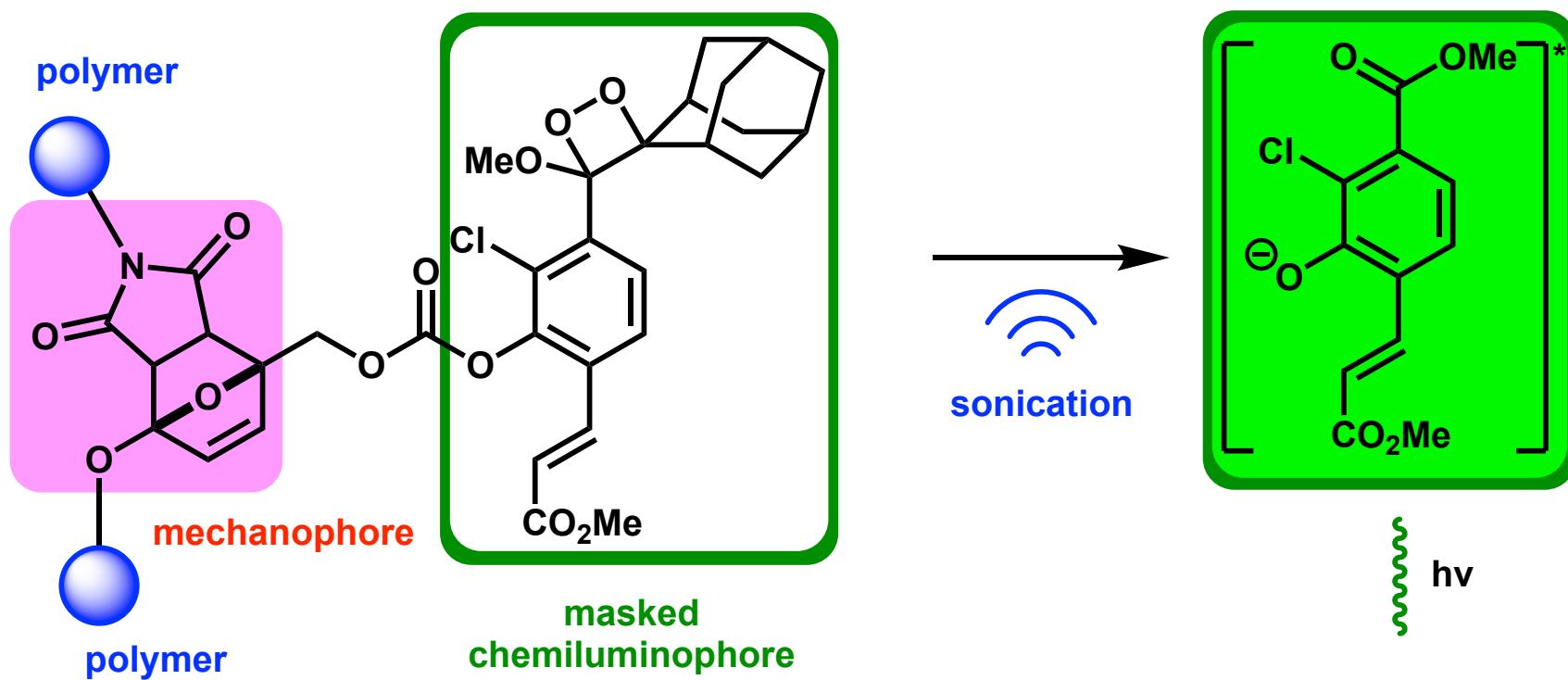
luminescence intensity measurement

sonication XX min

PEG-1 (47 kDa)  
PEG-control (7 kDa)

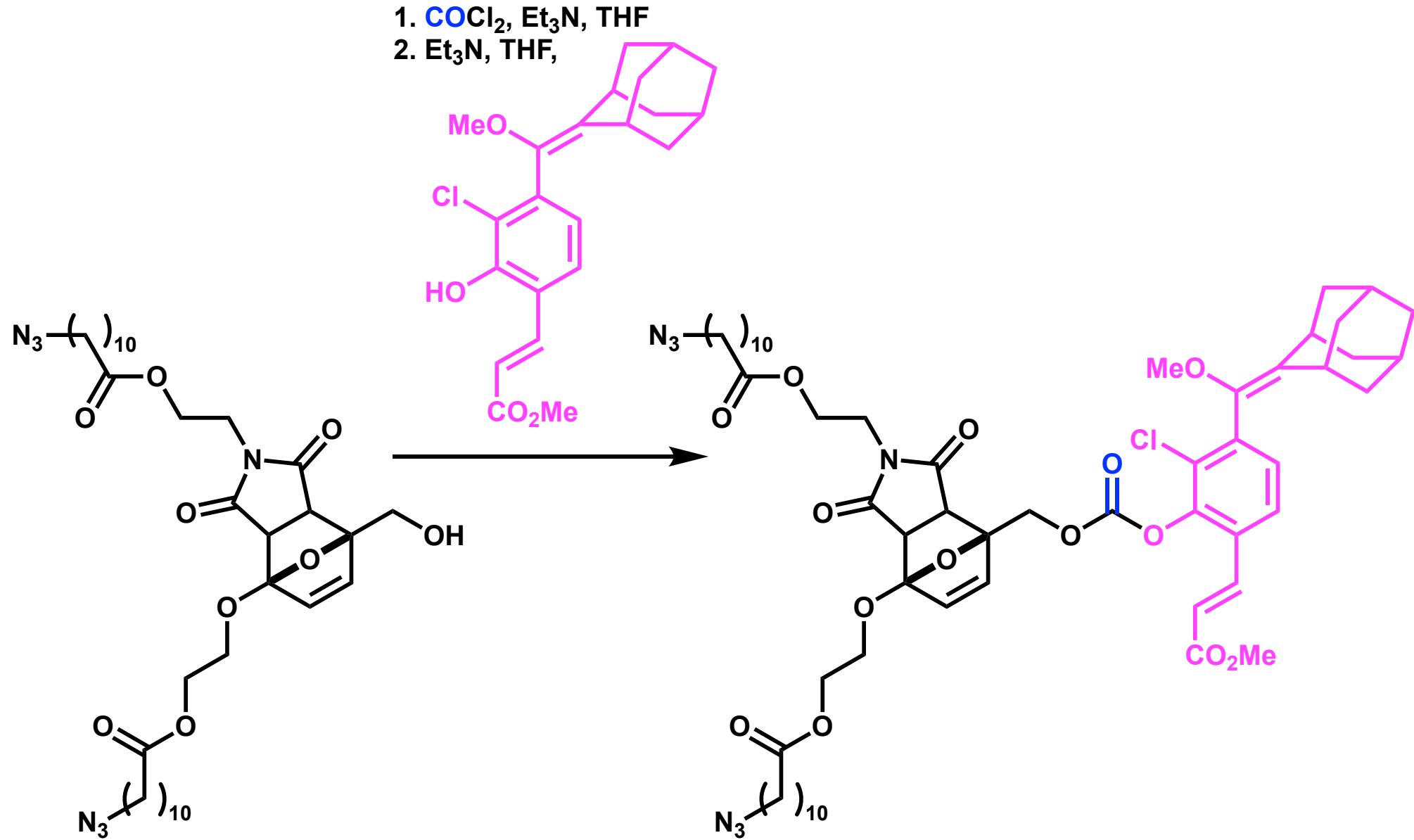


# Summary



# **Appendix**

# Appendix: Synthesis of Mechanoluminophore



# Appendix: Synthesis of Mechanoluminophore

