

Double- σ -Hole XB-Donor Catalysts

2025.4.26 Literature Seminar

M2 Yo Matsumoto

Contents

1. Introduction

2. Asymmetric Counteranion-Directed Halogen Bonding Catalysis (*J. Am. Chem. Soc.* 2025, 147, 8107.)

Prof. Benjamin List & Prof. Stefan M. Huber



Prof. Benjamin List

1993: B.S. @ Free University of Berlin

1997: Ph.D @ University of Frankfurt (Prof. J. Mulzer)

1997~1998: postdoctoral research @ Scripps Research Institute

1999~2003: Assistant Professor @ Scripps Research Institute (Tenure Track)

2003~2005: Group Leader @ Max-Planck-Institut für Kohlenforschung

2004~: Honorary Professor @ University of Cologne

2005~: Director @ Max-Planck-Institut für Kohlenforschung

2012~2014: Managing Director @ Max-Planck-Institut für Kohlenforschung

2018~: Specially Appointed Professor @ Hokkaido University Institute for
Chemical Reaction Design and Discovery

Topic: Aminocatalysis, Brønsted-Acid-Catalysis, Organic Lewis Acid Catalysis



Prof. Stefan M. Huber

2003: B.S. @ Friedrich-Alexander Universität Erlangen-Nuremberg

(Prof. Robert Weiss)

2007: Ph.D @ Friedrich-Alexander Universität Erlangen-Nuremberg

(Prof. Robert Weiss)

2007~2008: postdoctoral research @ the University of Minnesota

(Prof. Christopher J. Cramer, Prof. William B. Tolman)

2008: postdoctoral research @ the Université de Genève (Prof. Laura Gagliardi)

2009: postdoctoral research @ Friedrich-Alexander Universität Erlangen-Nuremberg

(Prof. Harald Gröger)

2009~2013: independent research @ Technische Universität München

2014~2021: Associate Professor @ Ruhr-Universität Bochum

2022~ Full Professor @ Ruhr-Universität Bochum

Topic: Biomolecule, Organocatalysis

(Hydrogen bond, Halogen bond, Chalcogen bond)

Hypervalent Iodine

127

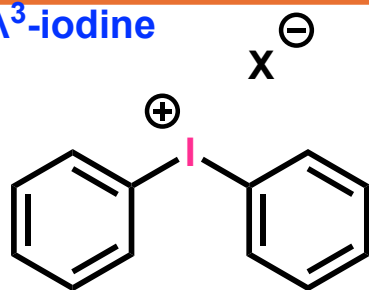
53

Electron configuration: $[\text{Kr}]4d^{10}5s^25p^5$

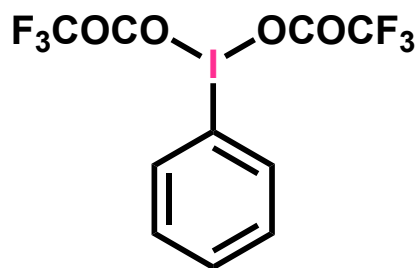
van der Waals radius: 2.2 Å

Role in total synthesis: Radical precursor, Metal coupling precursor
Oxidation (hypervalent iodine) ... etc.

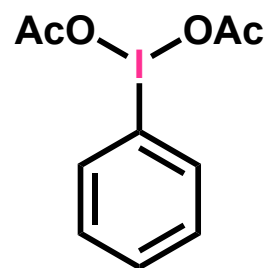
λ^3 -iodine



Diaryliodonium Salts

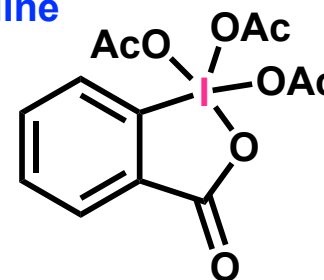


PIFA

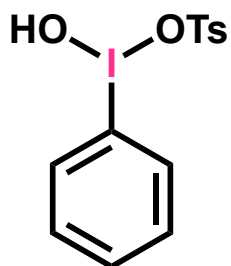


PIDA

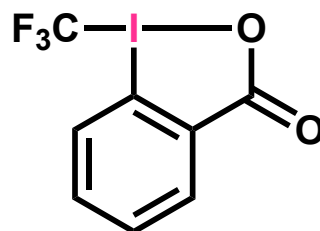
λ^5 -iodine



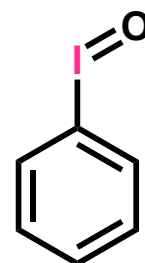
Dess-Martin periodinane
 I(VII) type



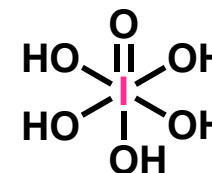
Koser's reagent



Togni II reagent



Iodosylbenzene

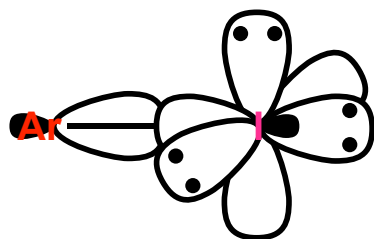
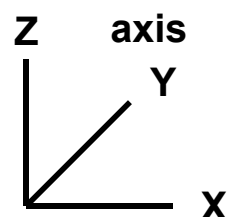


Orthoperiodic Acid

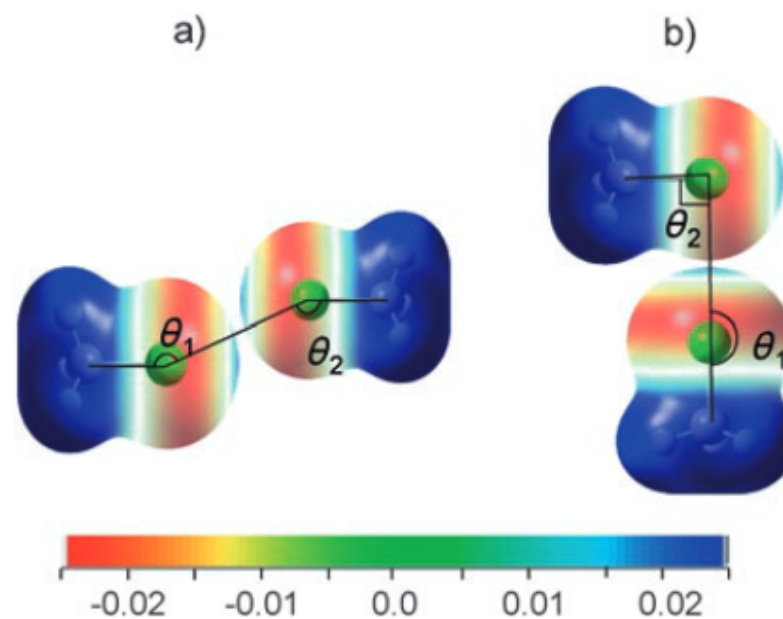
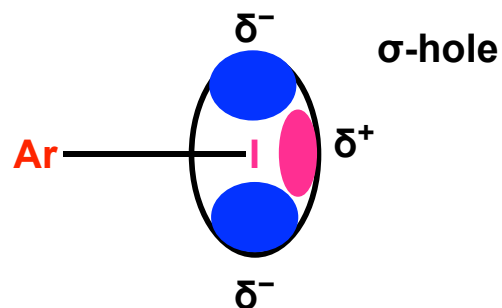
σ Hole: Halogen Halogen Interaction

See 241122_LS_Hibiki_Asai_Halogen_Halogen_Interaction

σ -Hole of Halogen

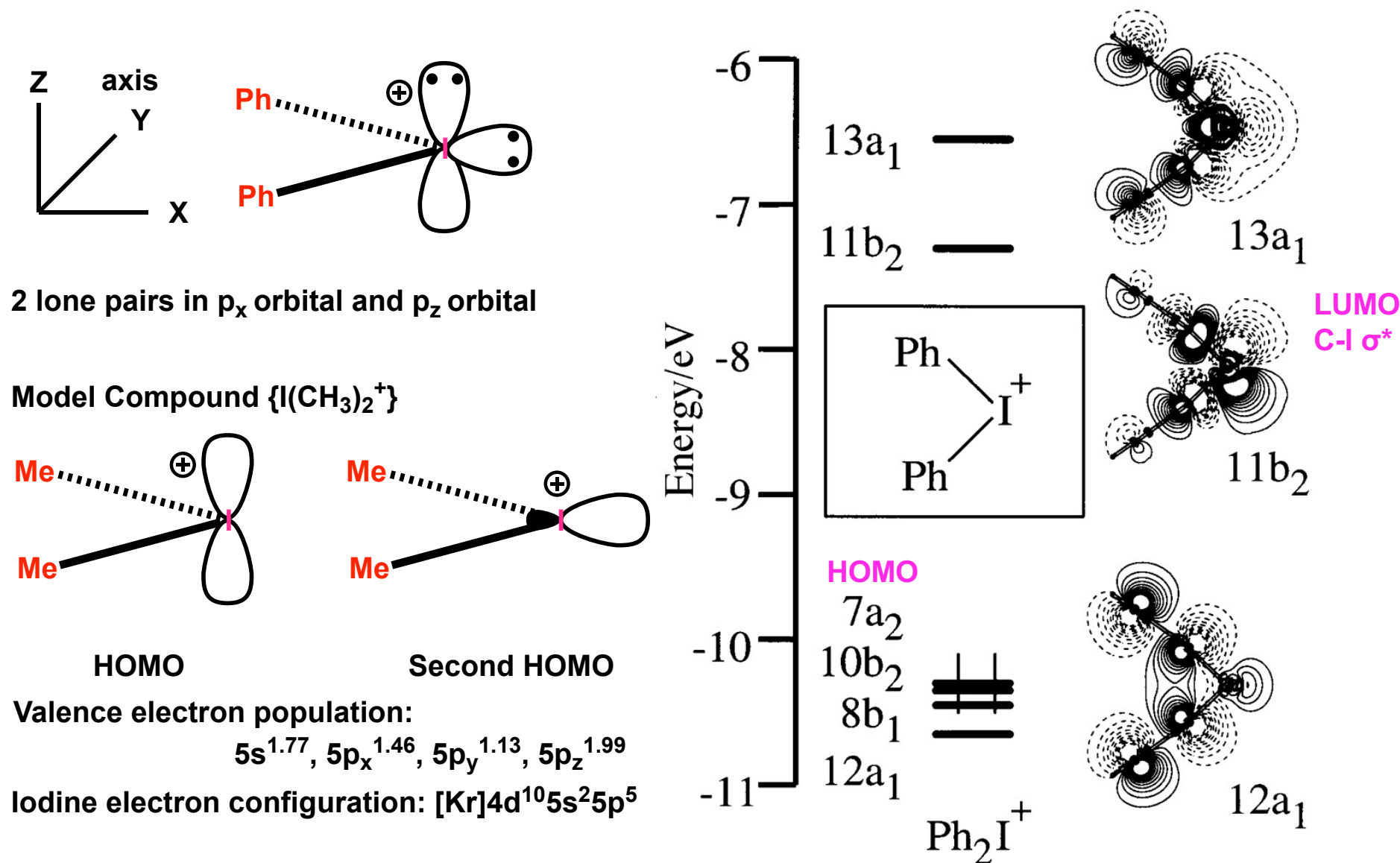


σ -hole:
C-I σ^* > lone pair of p_x orbital



Two Types of Halogen-Halogen Interaction

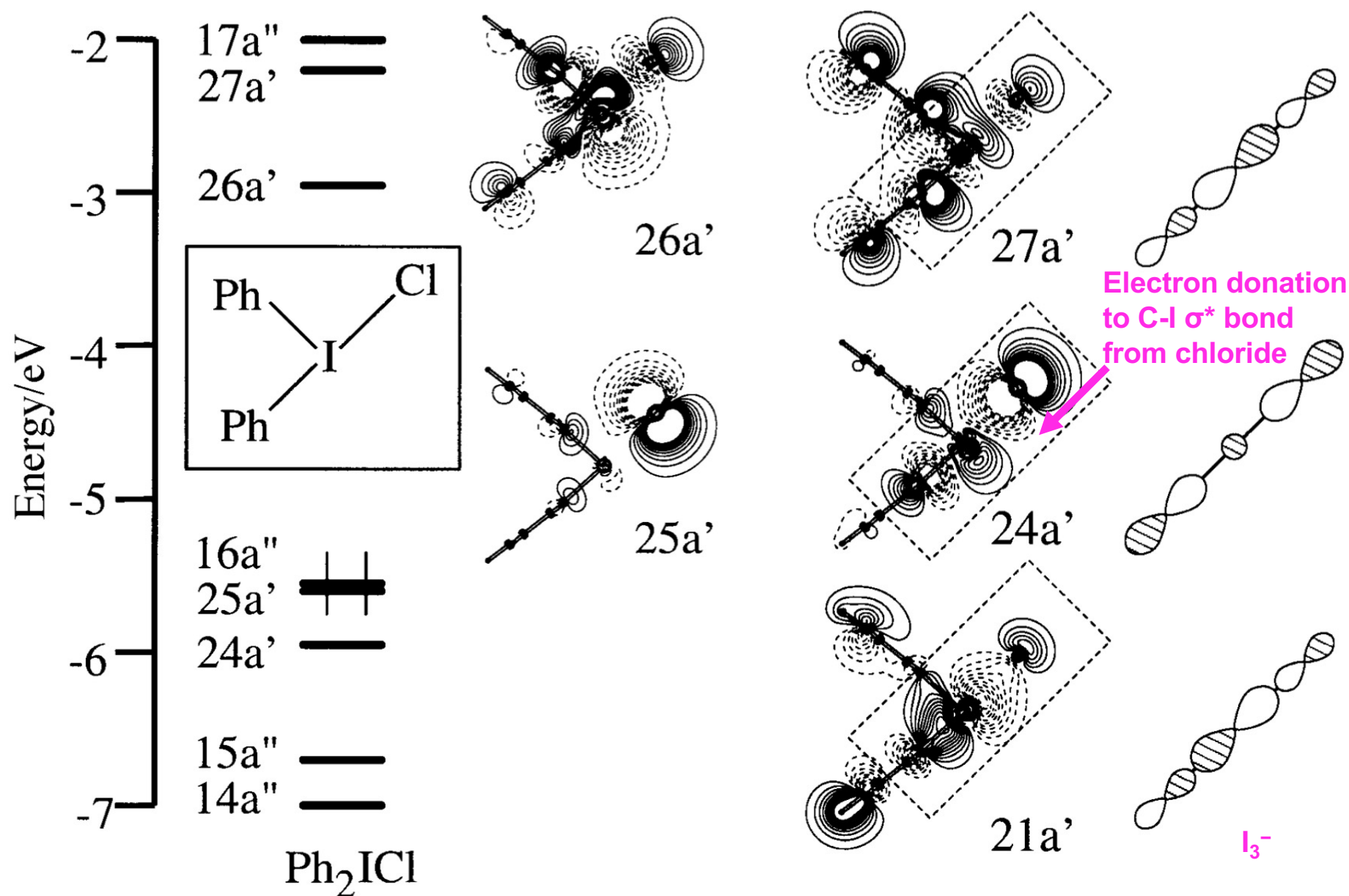
Double σ -Hole: Iodine(III) (Ph_2I^+ Species) I



1. Ikezawa, H.; Takahashi, M.; Takeda, M.; Ito, Y. *Bull. Chem. Soc. Jpn.* **1993**, 66, 1959.

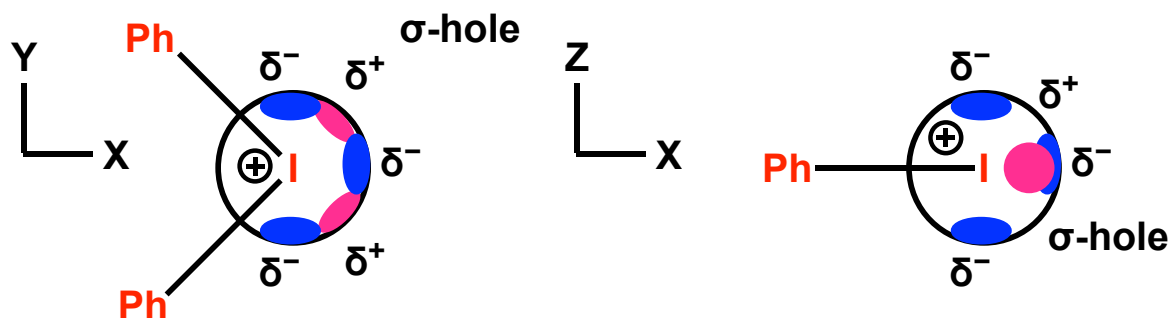
2. Landrum, G. A.; Goldberg, N.; Hoffmann, R.; Minyaev, R. M. *New. J. Chem.* **1998**, 22, 883.

Double σ -Hole: Iodine(III) (Ph_2I^+ Species) II

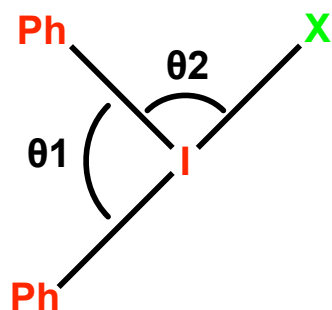


Double σ -Hole: Iodine(III) (Ph_2I^+ Species) III

Double σ -Hole of Iodonium (III) Salts

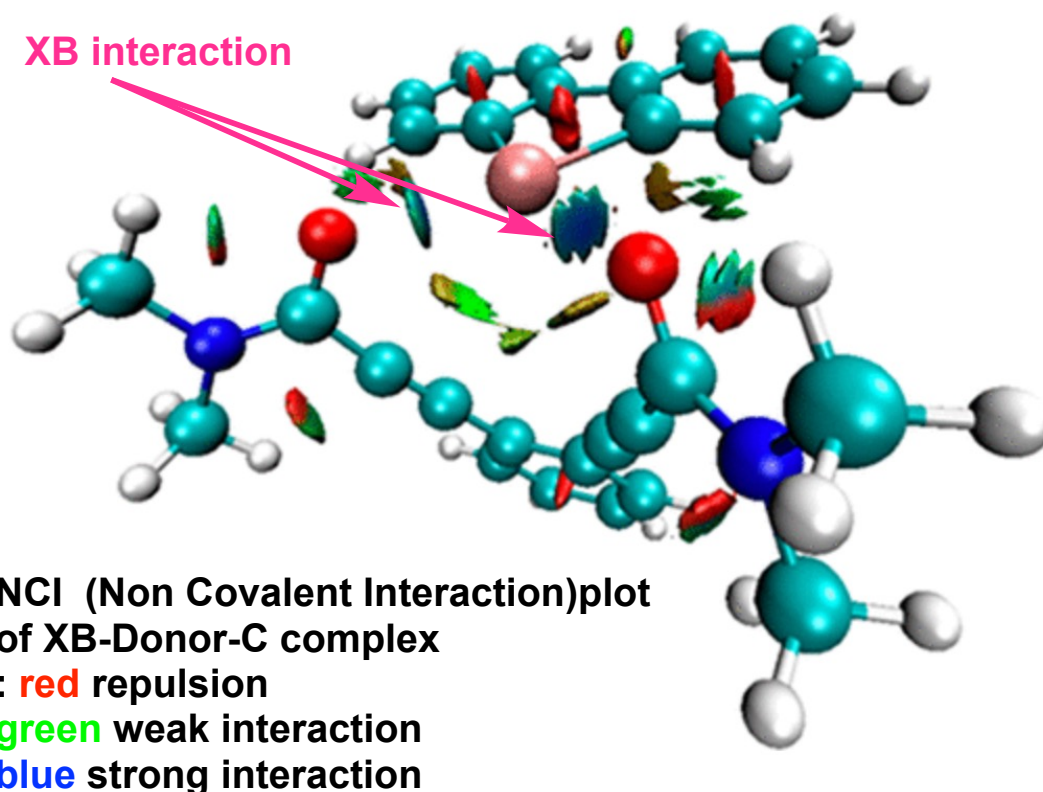
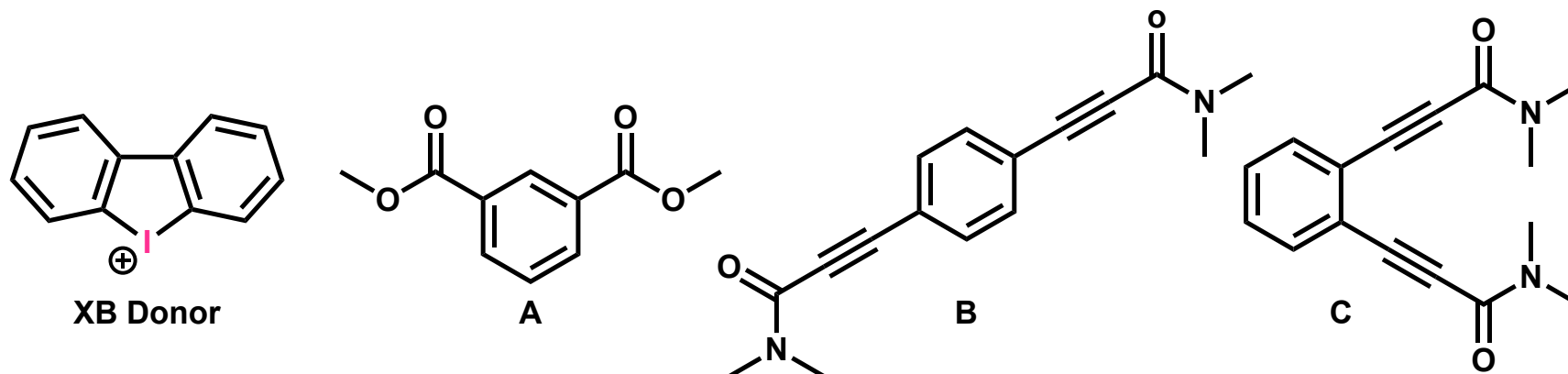


Angle of Ph_2IX species



X	Cl	Br	I
$\theta_1/^\circ$	80	82	82
$\theta_2/^\circ$	87	86	88

Double σ -Hole Donor as Lewis Acid



Binding Energy of XB Donor's Adduct Formation

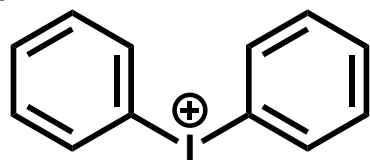
	A	B	C
Binding constants (M^{-1})	1.6×10^1	1.5×10^3	8.3×10^4
Experimental ΔG	-1.7	-4.4	-6.3
Calculated ΔG (SMD, PCM)	0.0, 0.4	-1.3, -2.8	-9.1, -6.6

Double σ -Hole: Diaryliodonium (III) Salt

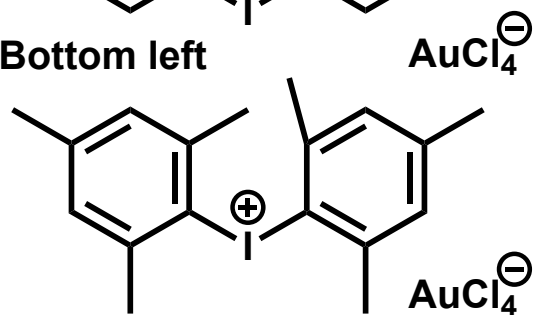
Hirshfeld surfaces:

a method that reveals the hydrophilic nature of intermolecular interactions in a molecular system.

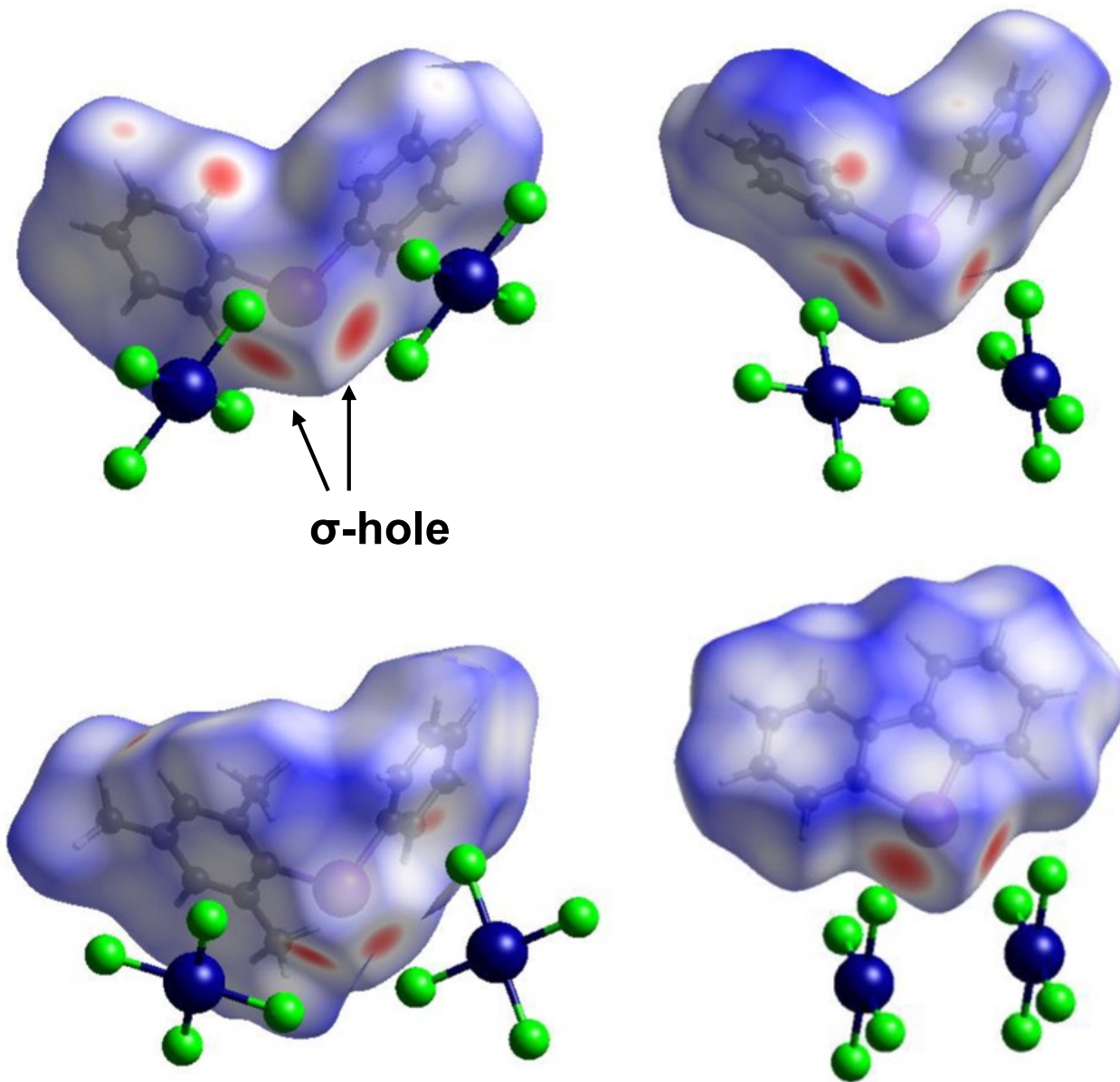
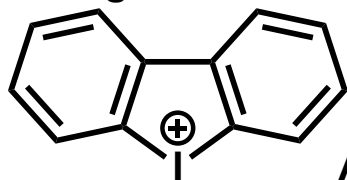
Top 2



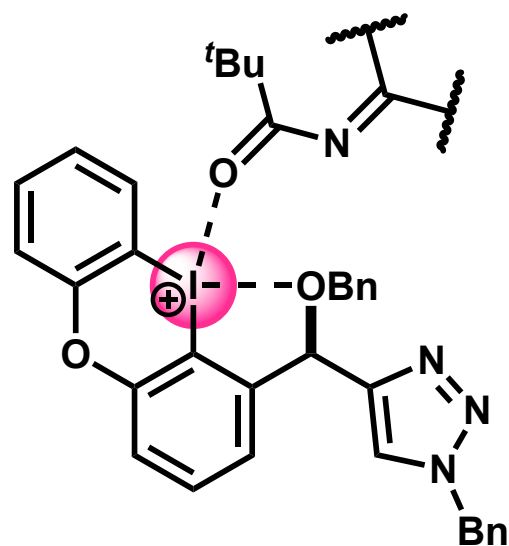
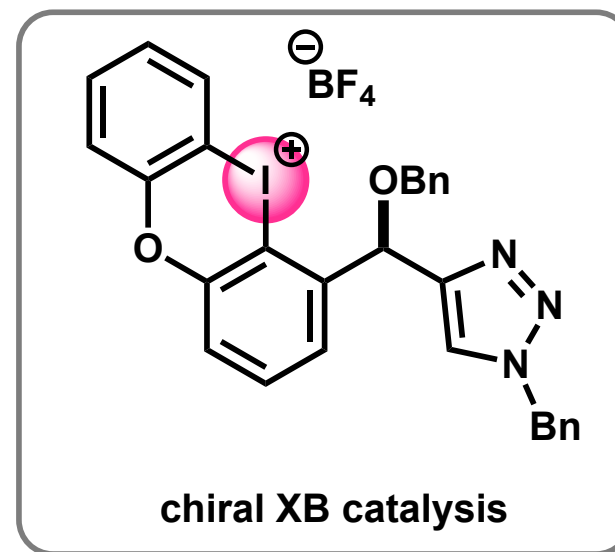
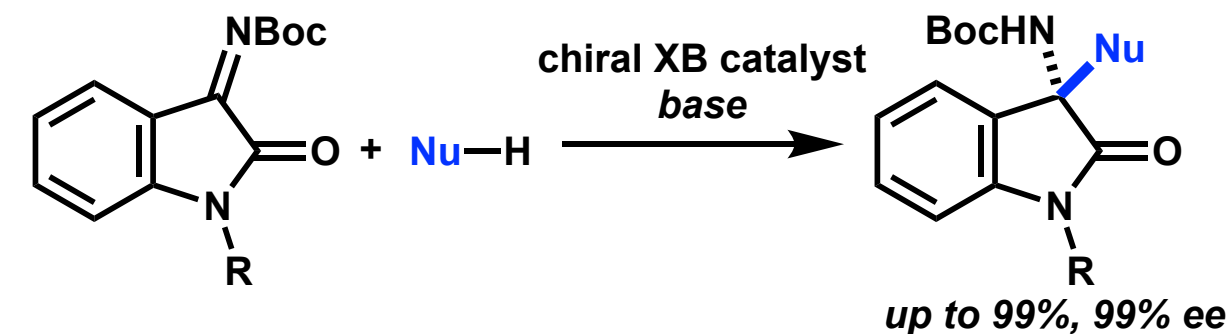
Bottom left



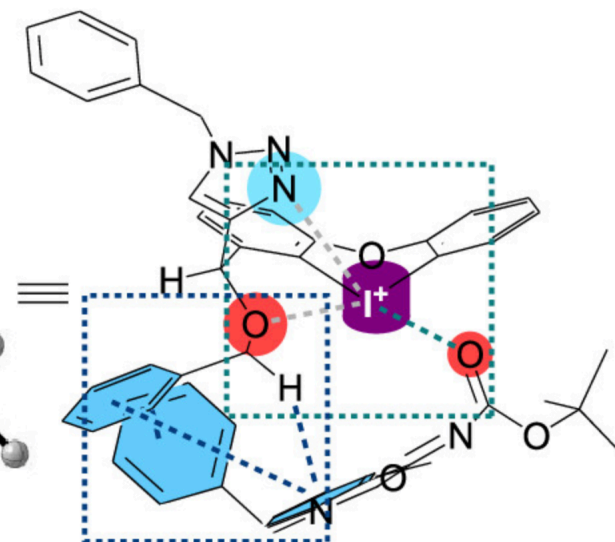
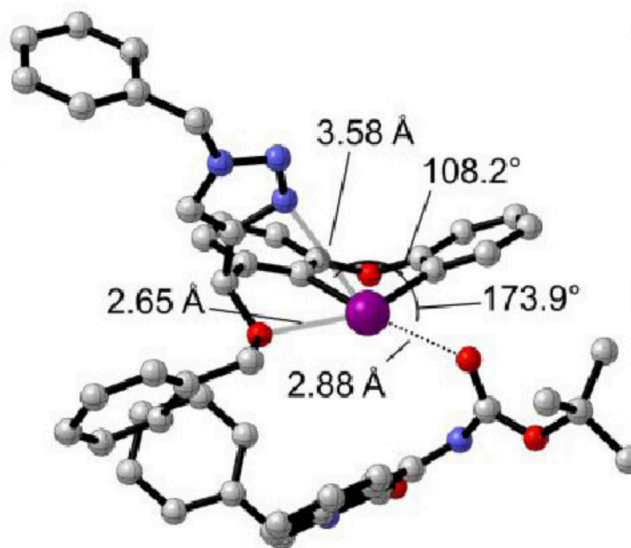
Bottom right



Previous Work: Chiral Iodonium (III) Salts Catalyst

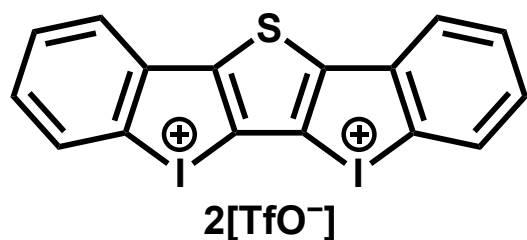


Activation formation



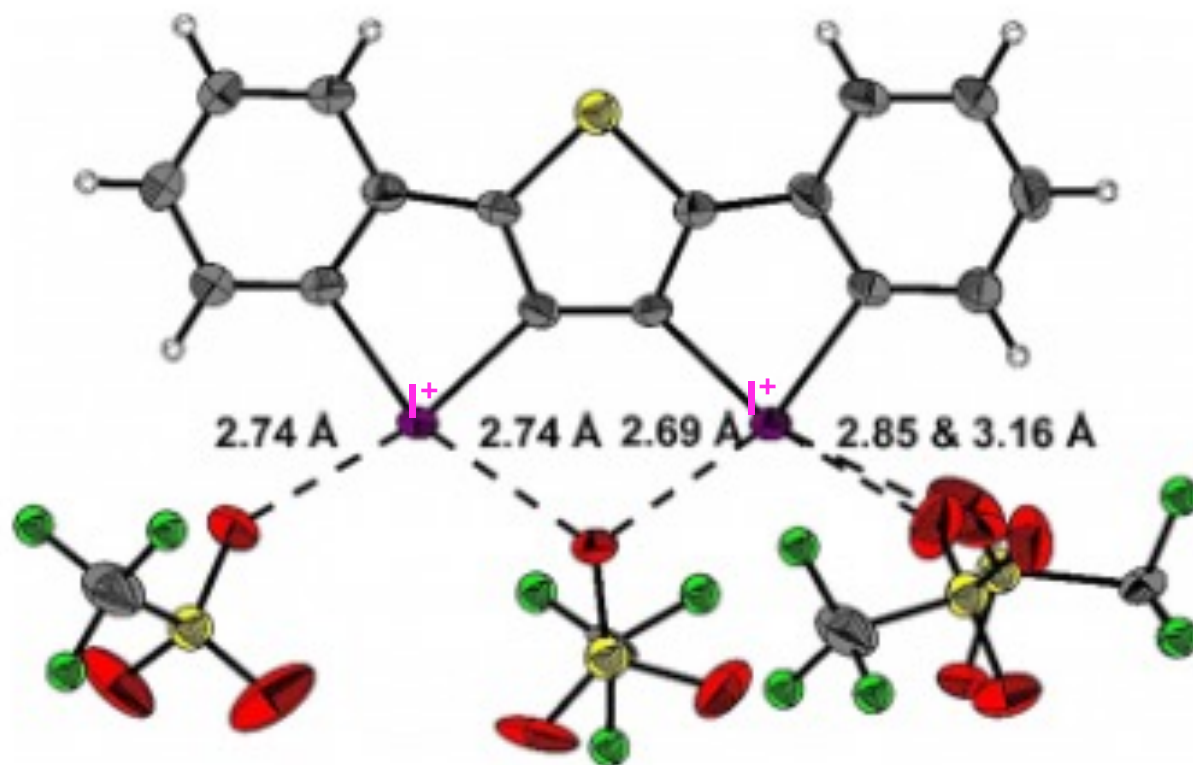
DFT Calculation: B3LYP-D3/def2-TZVP

Bidentate XB (Halogen Bonding) Donor I



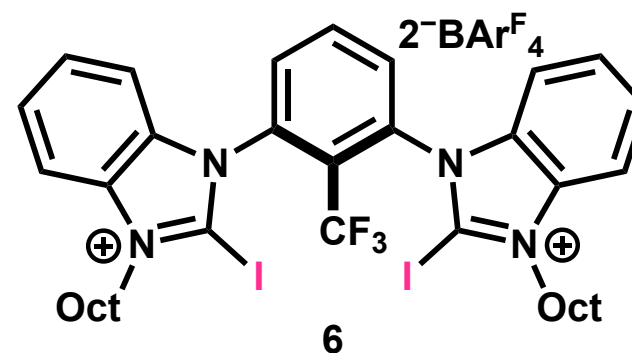
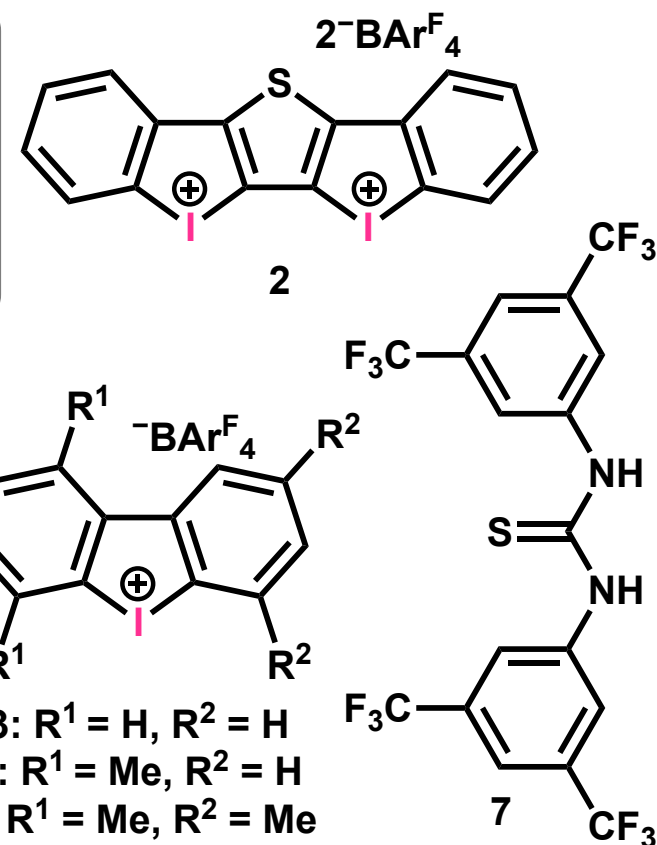
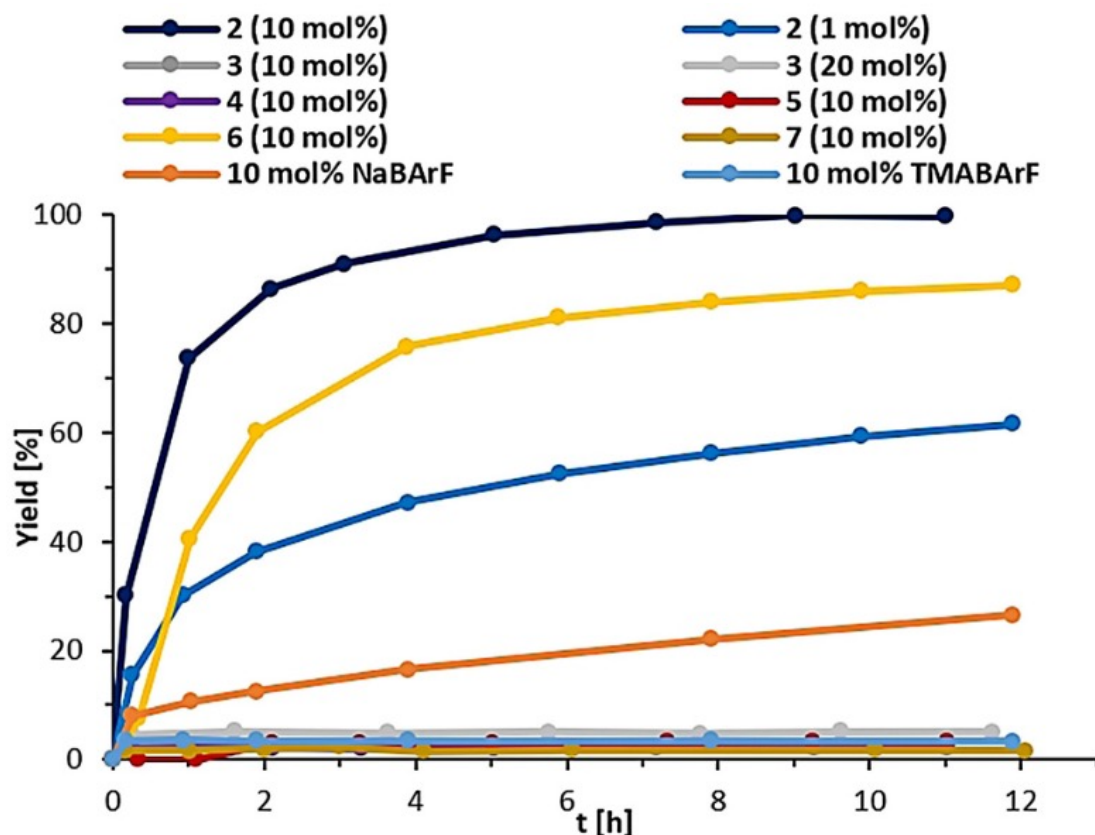
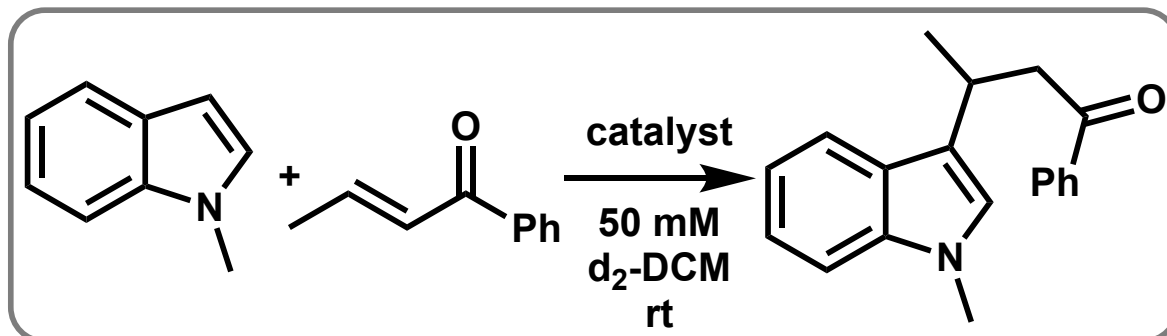
Bidentate XB Donor

Angle of C-I $\cdots$ O: 159~160 °
I $\cdots$ O distances: 2.69-2.73 Å
(van der Waals radii: 3.50 Å)



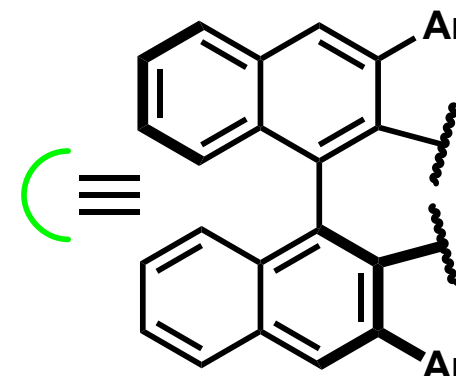
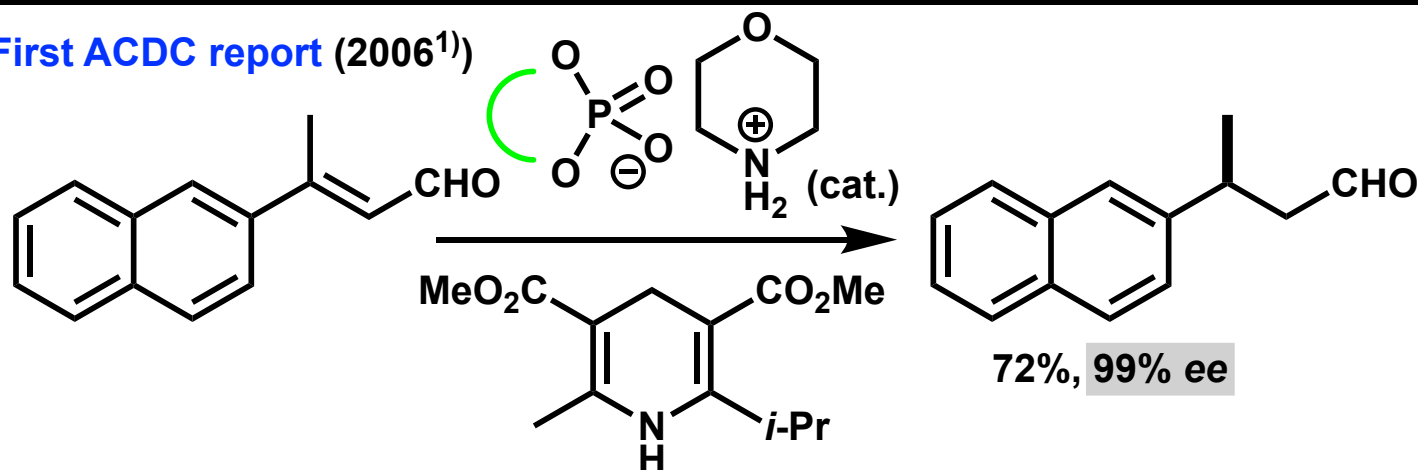
X-ray structural analysis

Bidentate XB (Halogen Bonding) Donor II

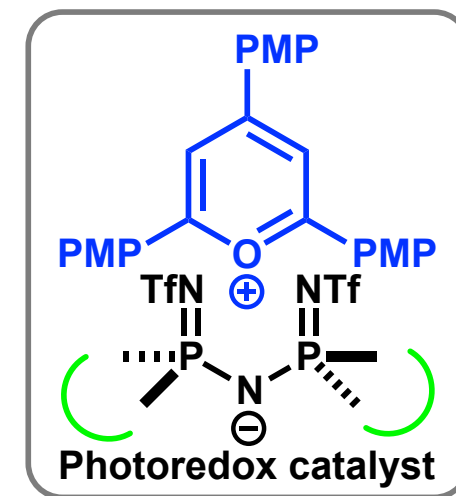
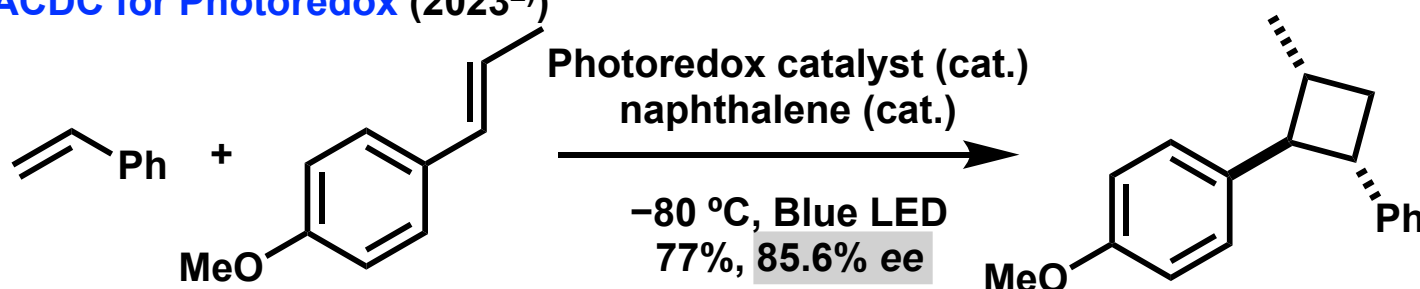


ACDC: Asymmetric Counteranion Directed Catalysis

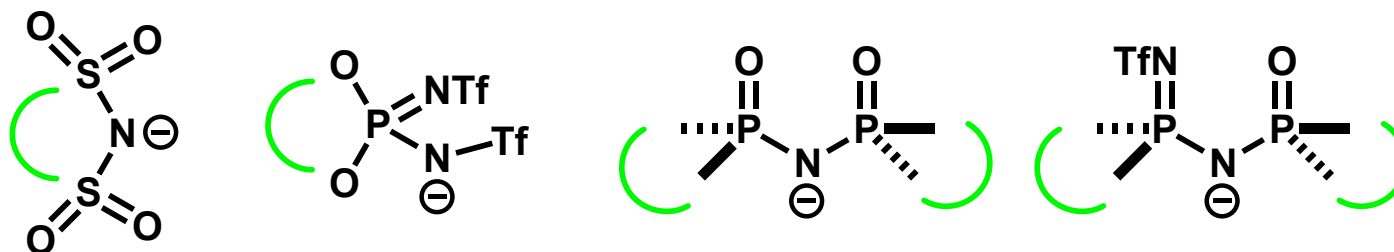
First ACDC report (2006¹⁾)



ACDC for Photoredox (2023²⁾)



Other ACDC (aryl group has a wide variety, so does pKa)

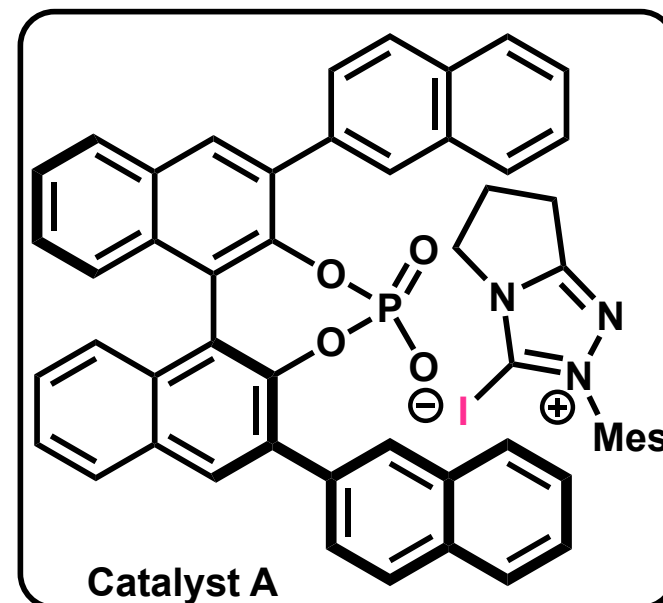
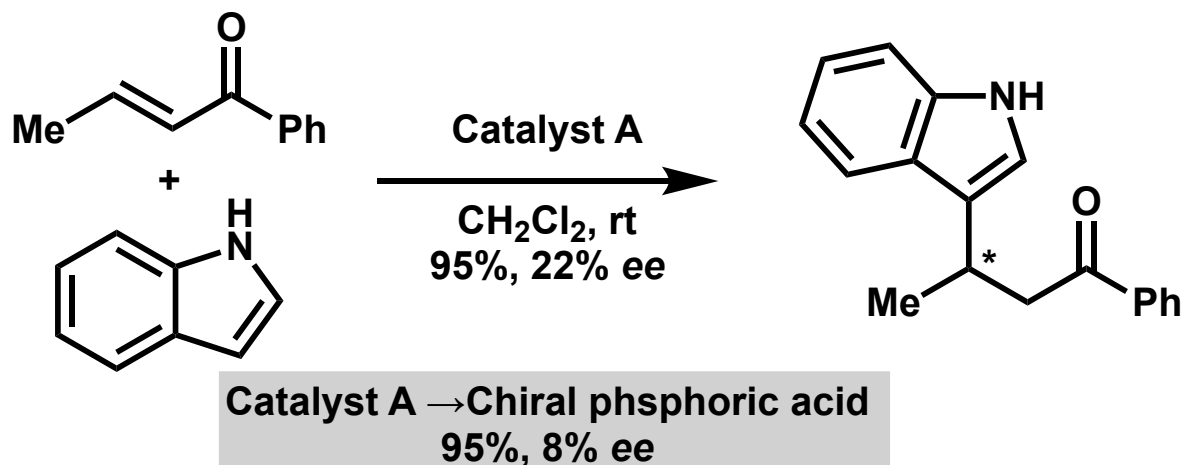


See 180630_LS_Tsukasa_Shimakawa_Asymmetric_Brøsted_Acid_Catalysis_Developed_By_Benjamin_List's Group. 1. Mayer, S.; List, B. *Angew. Chem. Int. Ed.* **2006**, 45, 4193. 2. Das, S.; Zhu, C.; Demirbas, D.; Bill, E.; De, C. K.; List, B. *Science* **2023**, 379, 494.

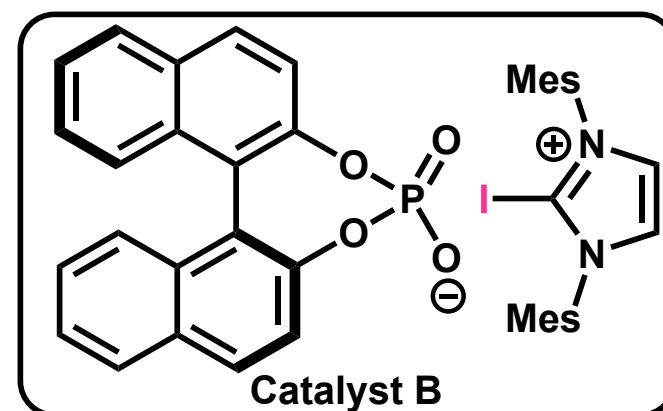
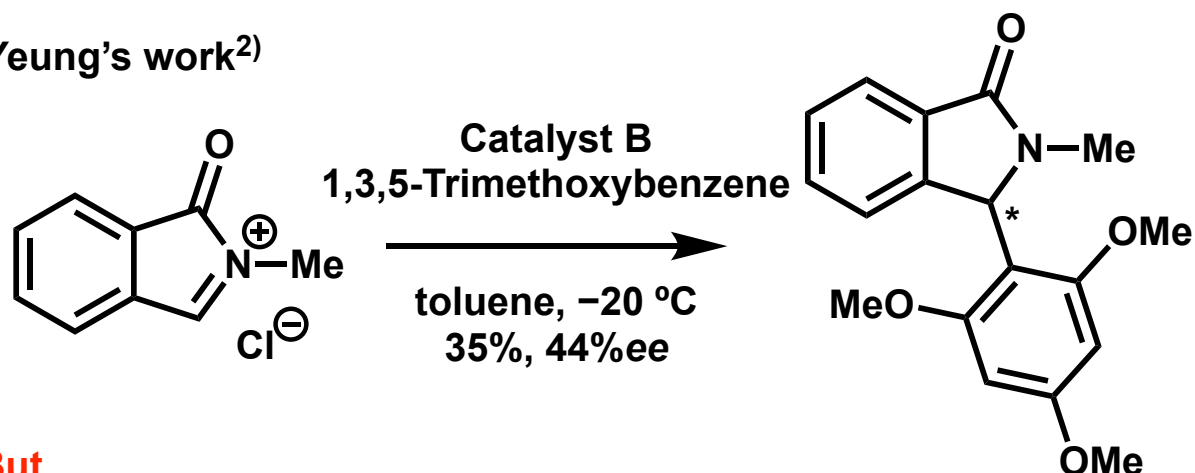
Previous Work: Asymmetric Reaction by XB & ACDC

ACDC: Asymmetric Counteranion Directed Catalysis

Scheidt's work¹⁾



Yeung's work²⁾



But....

No control experiment was conducted to eliminate the possibility of steric control by
「hidden」 Brønsted acid.

1. Squitieri, R. A.; Fitzpatrick, K. P.; Jaworski, A. A.; Scheidt, K. A. *Chem. Eur. J.* **2019**, 25, 10069.

2. Chan, Y.-C.; Yeung, Y.-Y. *Org. Lett.* **2019**, 21, 5665.

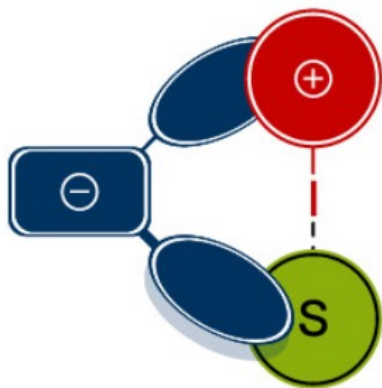
Contents

1. Introduction

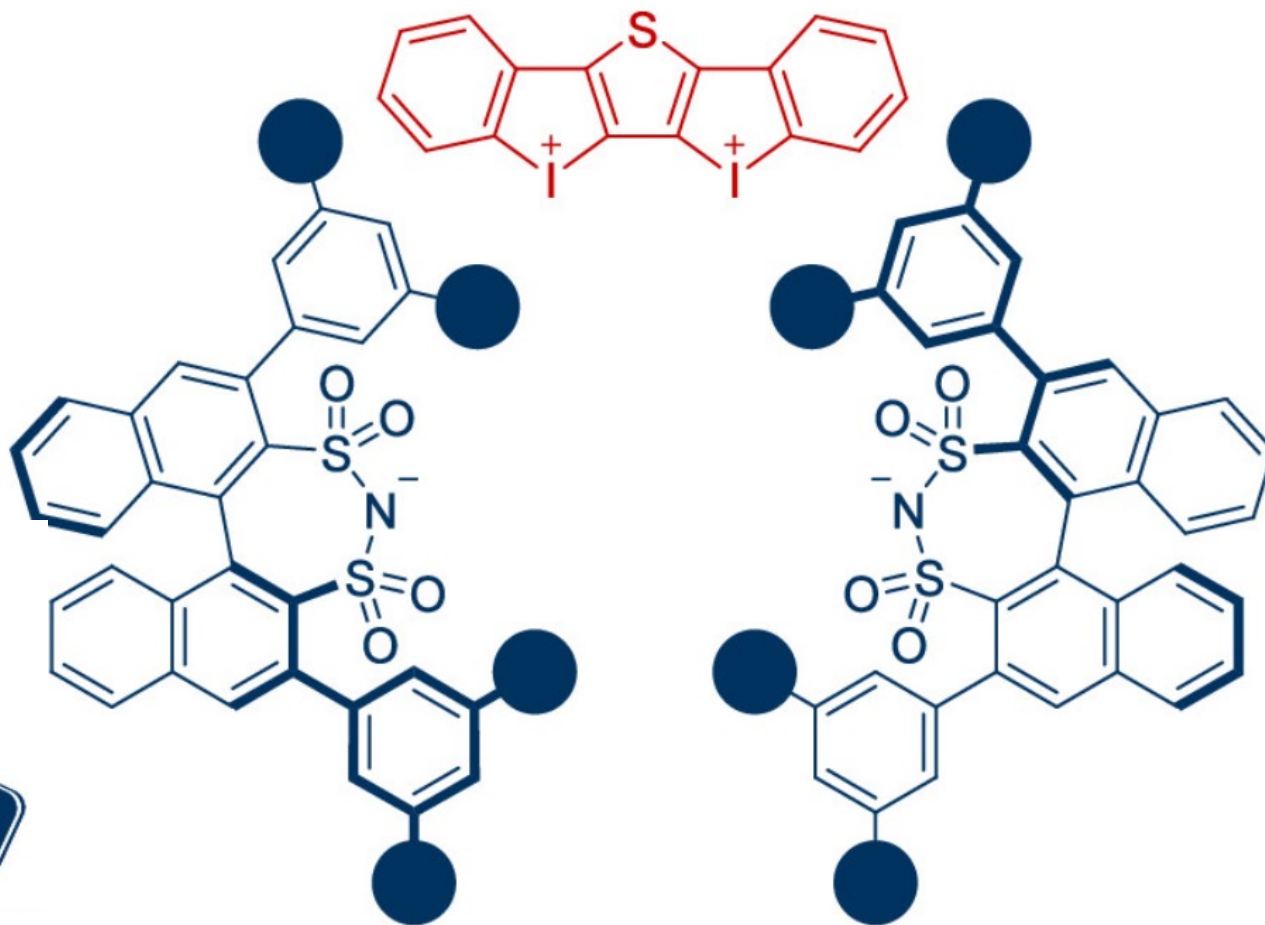
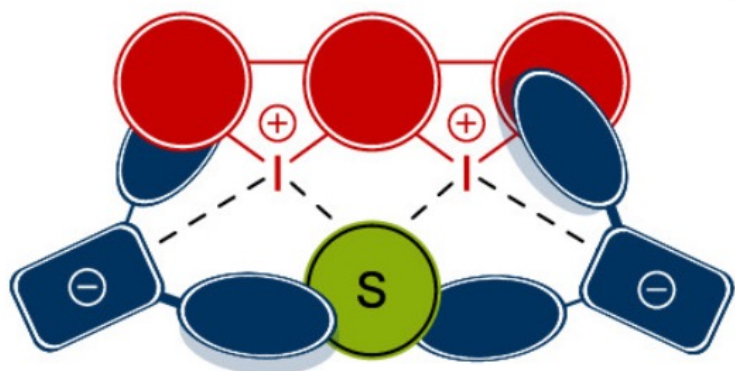
2. Asymmetric Counteranion-Directed Halogen Bonding Catalysis (*J. Am. Chem. Soc.* 2025, 147, 8107.)

Catalyst Concept

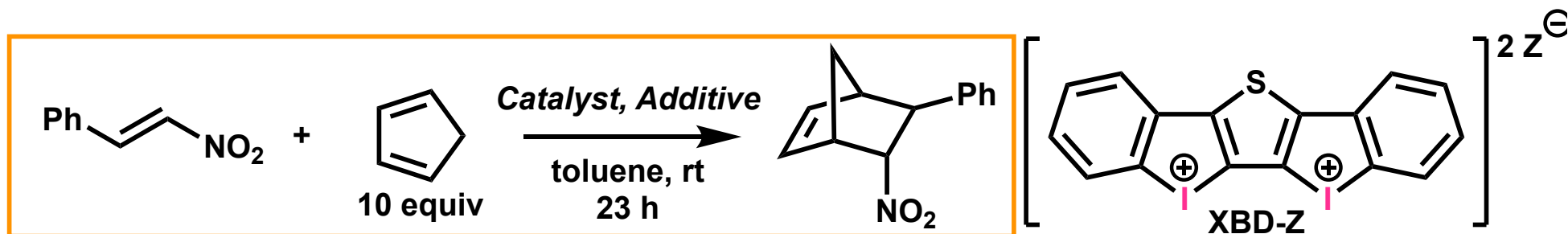
Previous type



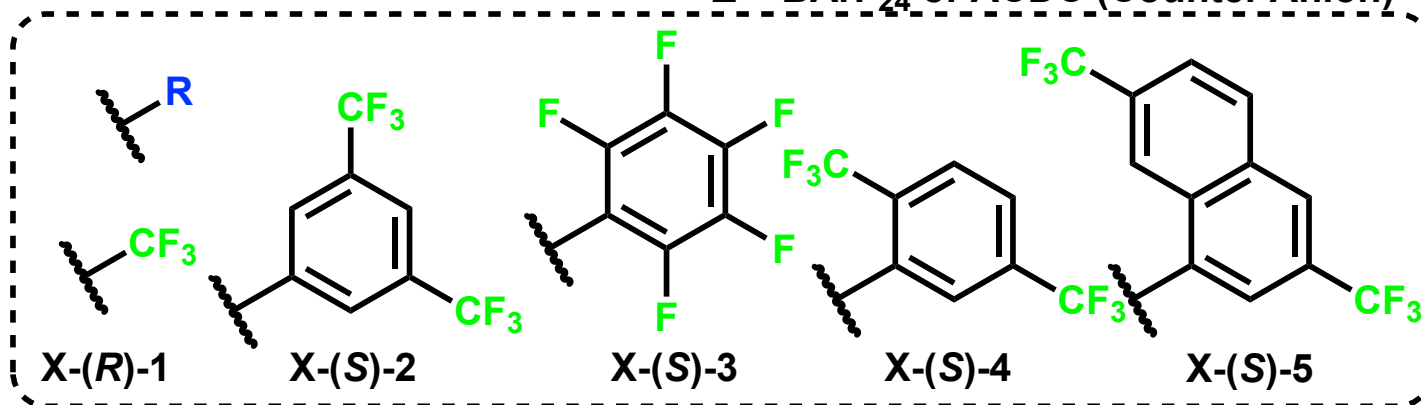
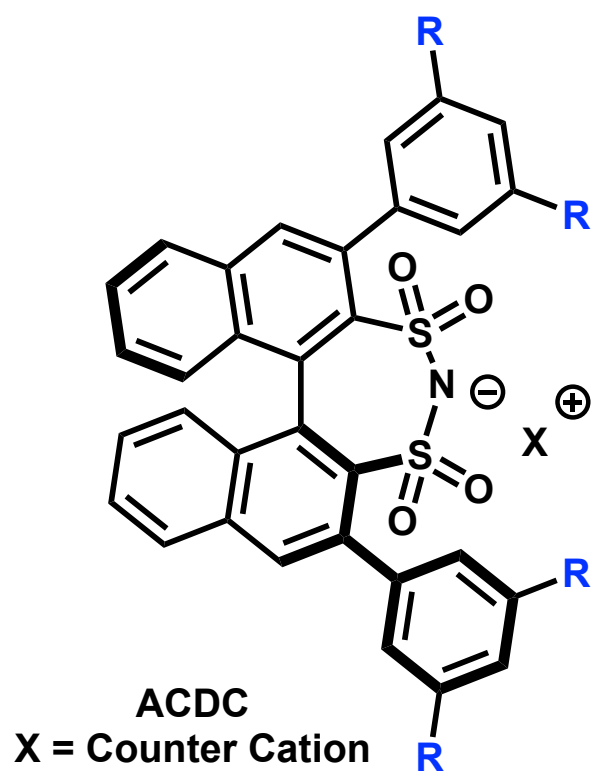
This work



Optimization I

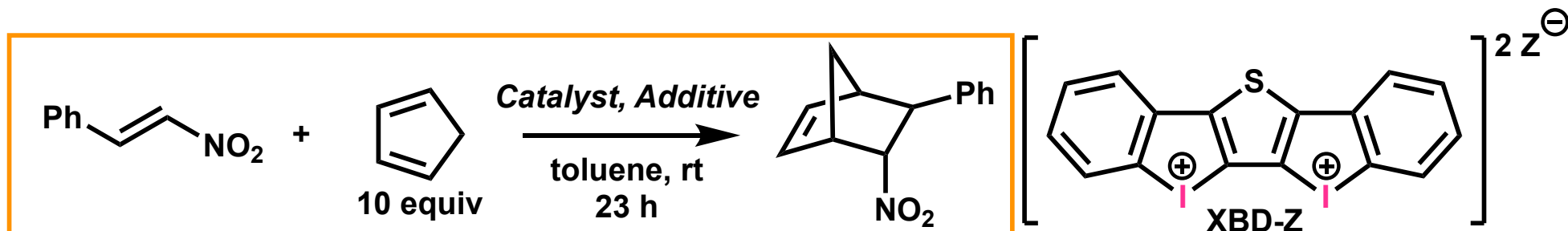


Z = BArF₂₄ or ACDC (Counter Anion)

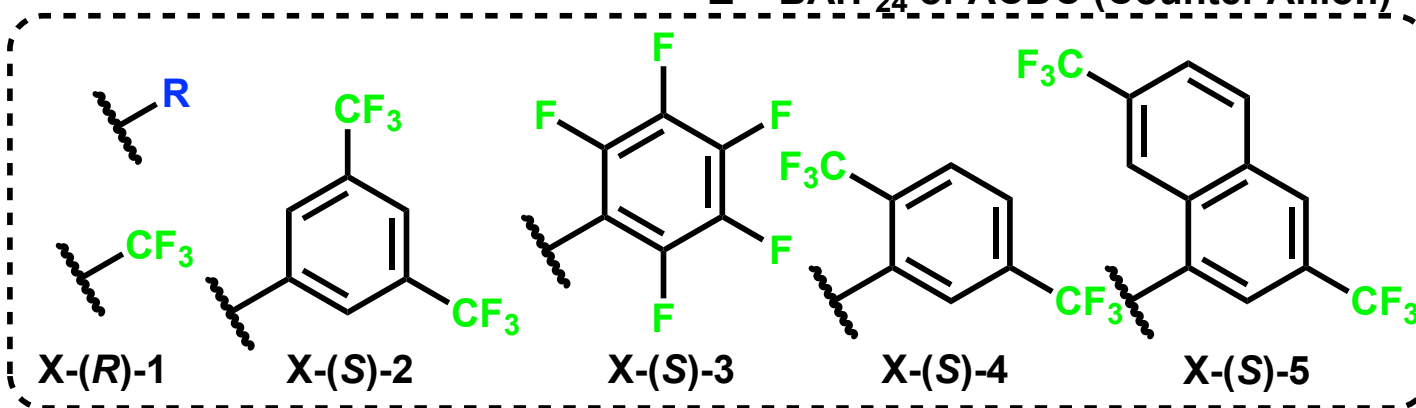
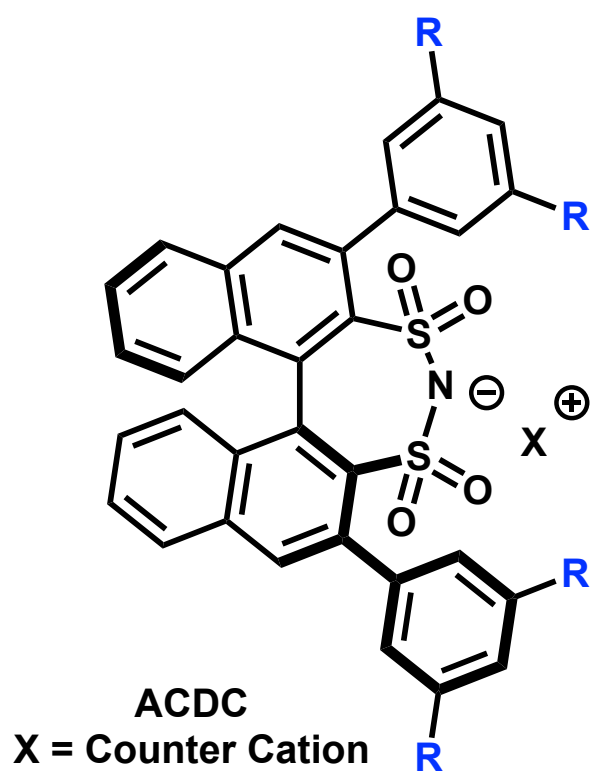


Entry	Catalyst (mol%)	Additive(mol%)	Yield(%)	ee(%)
0	-	-	12	-
1	XBD-BArF ₂₄ (10)	-	60	-
2	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(<i>R</i>)-1 (20)	37	50
3	-	N(<i>n</i> Bu) ₄ -(<i>R</i>)-1 (20)	8	2
4	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(<i>S</i>)-2 (20)	23	38
5	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(<i>S</i>)-3 (20)	35	52
6	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(<i>S</i>)-4 (20)	46	72
7	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(<i>S</i>)-5 (20)	69	70

Optimization II

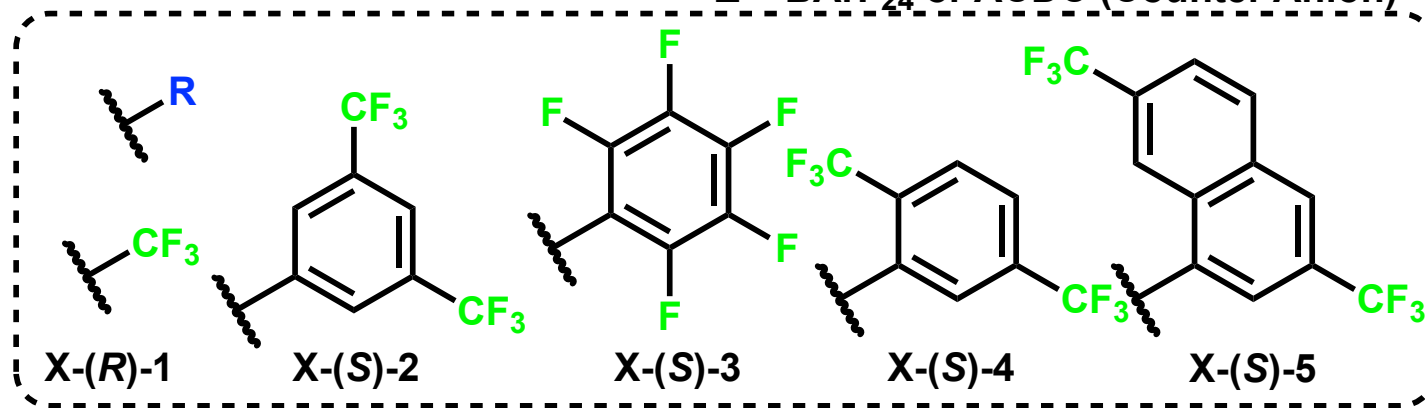
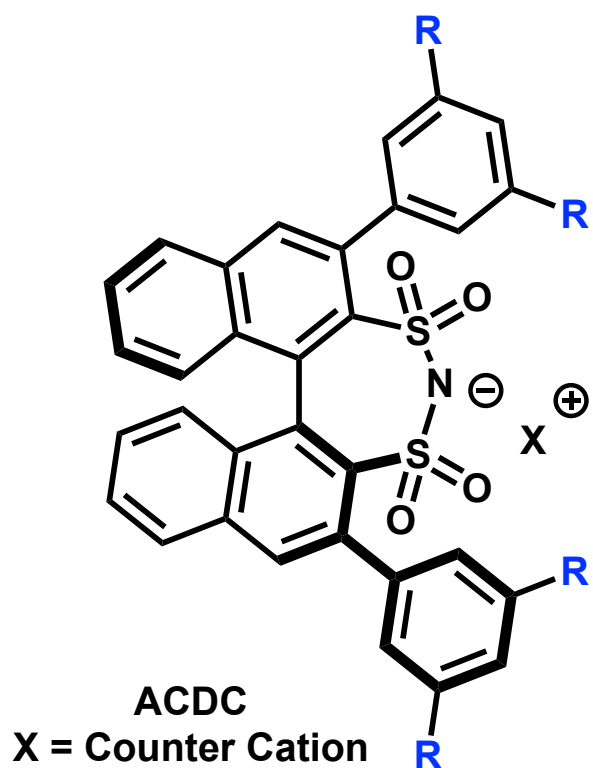
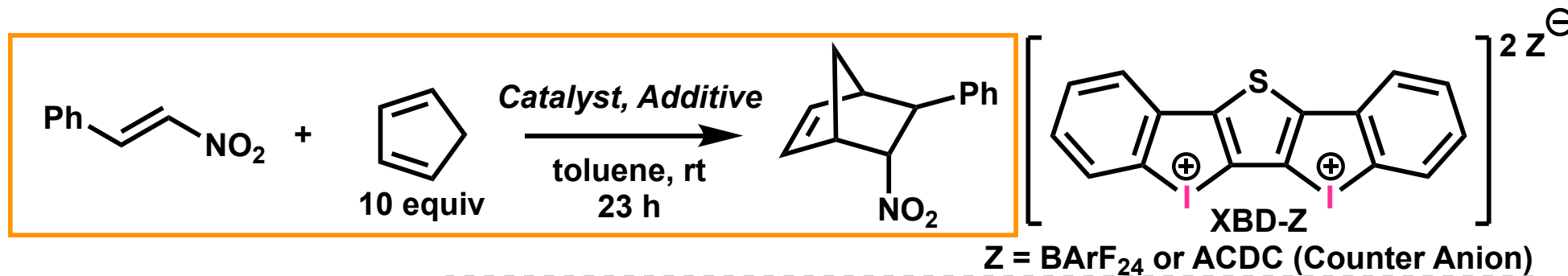


Z = BArF₂₄ or ACDC (Counter Anion)



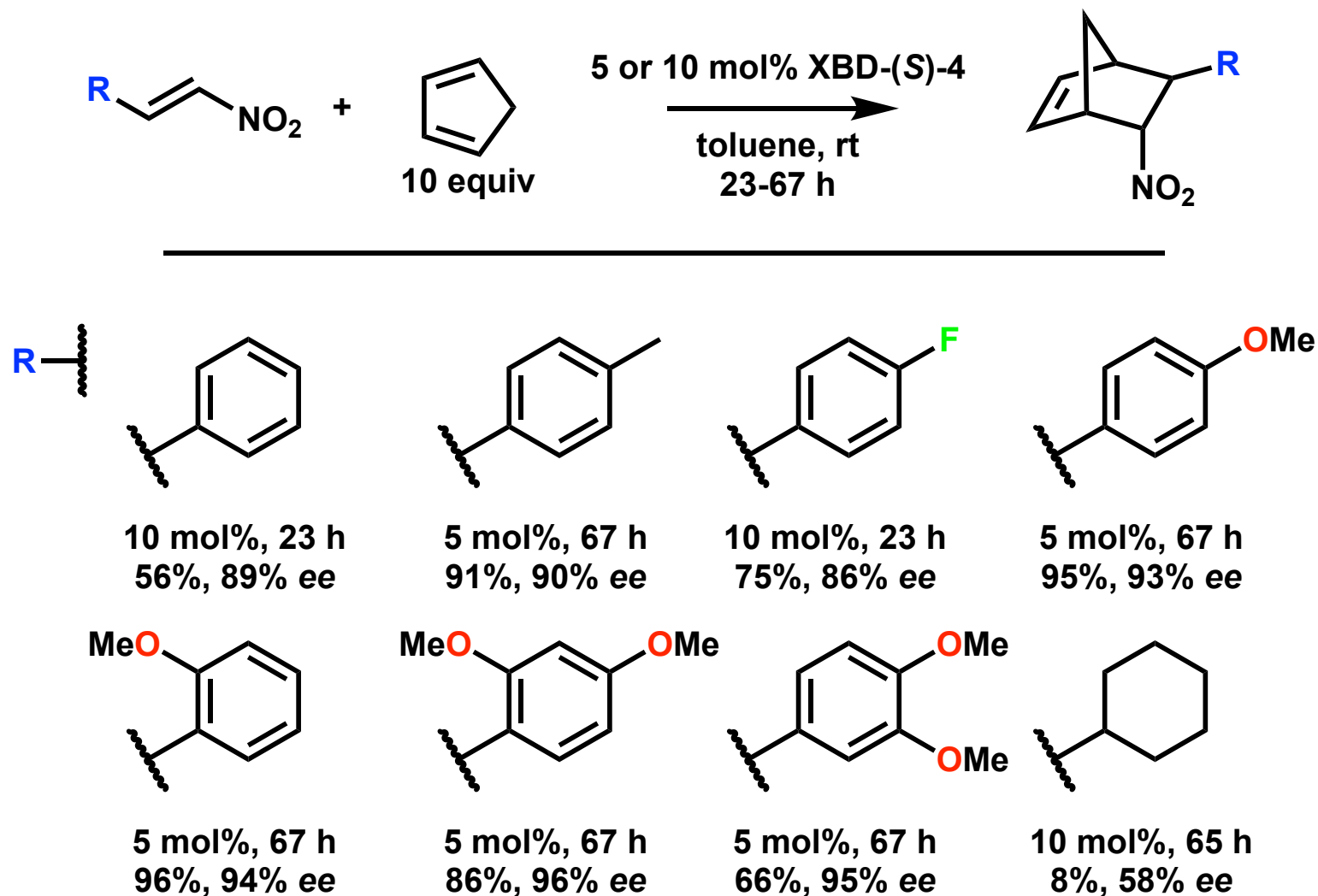
Entry	Catalyst (mol%)	Additive (mol%)	Yield(%)	ee(%)
0	-	-	12	-
6	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(S)-4 (20)	46	72
7	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(S)-5 (20)	69	70
8	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(S)-5 (10)	34	4*
9	XBD-BArF ₂₄ (10)	N(<i>n</i> Bu) ₄ -(S)-5 (30)	54	64
10	XBD-(S)-4 (10)	-	94	86
11	XBD-(S)-5 (10)	-	>95	76

Optimization III



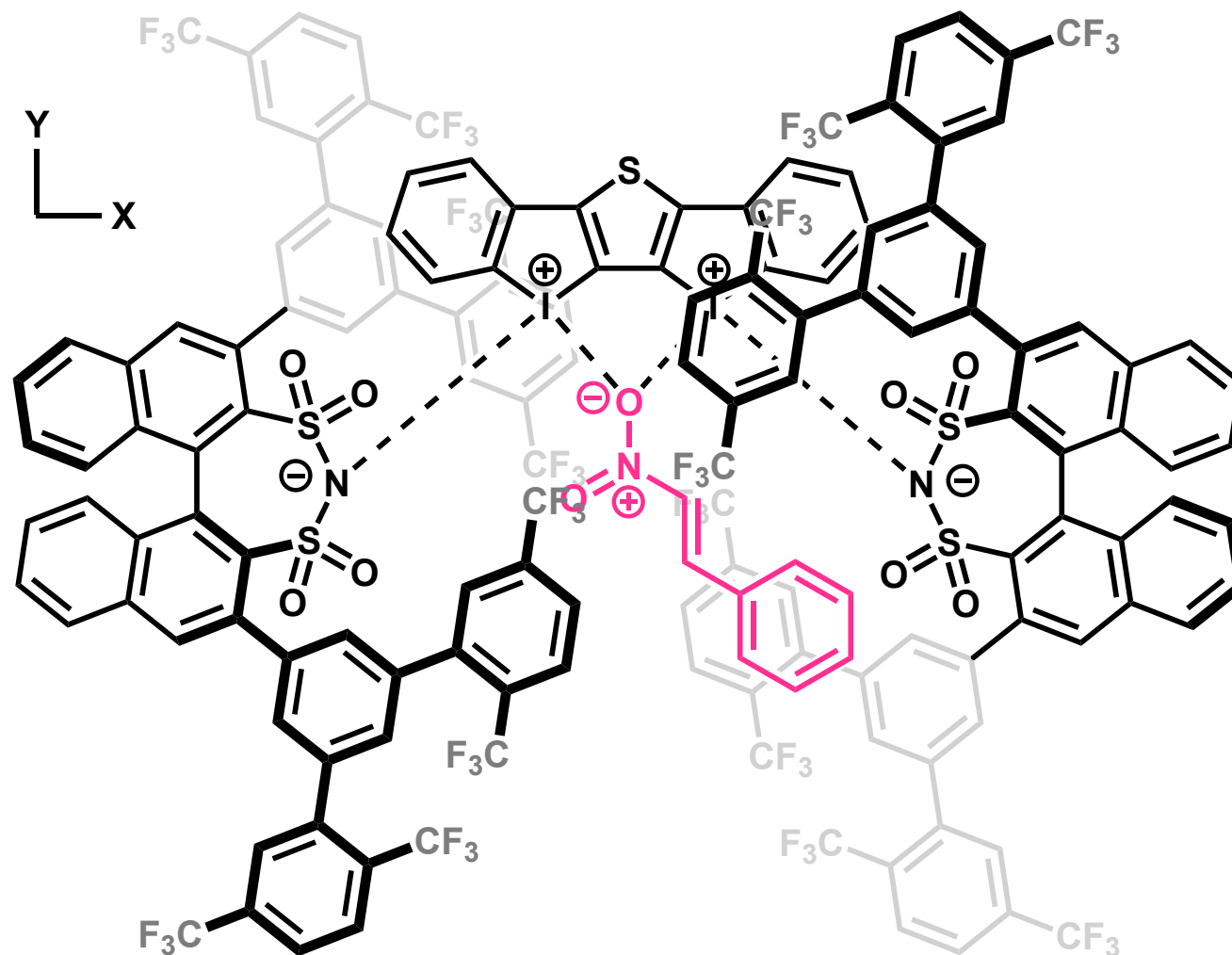
Entry	Catalyst (mol%)	Concentration (mM)	Yield(%)	ee(%)
0	-	100	12	-
10	XBD-(S)-4 (10)	50	94	86
11	XBD-(S)-4 (10)	12.5	56	89
12	XBD-(S)-4 (5)	12.5	32	86
13	XBD-(S)-4 (2)	12.5	10	74
14	H-(S)-4 (20)	50	9	2

Substrate Scope

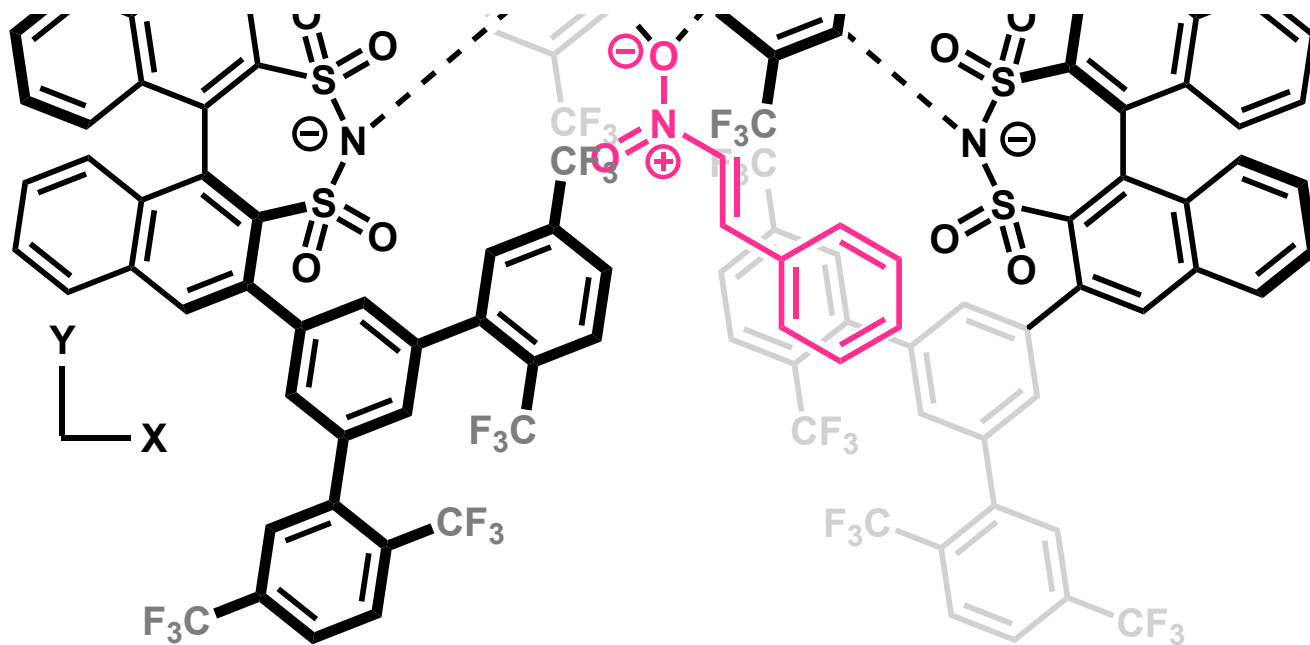


Enantio-Selectivity: My Proposal I

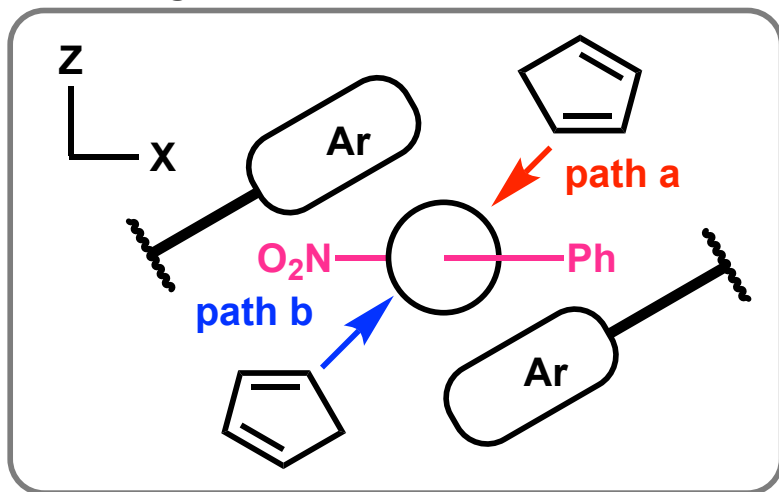
If catalyst concept worked well



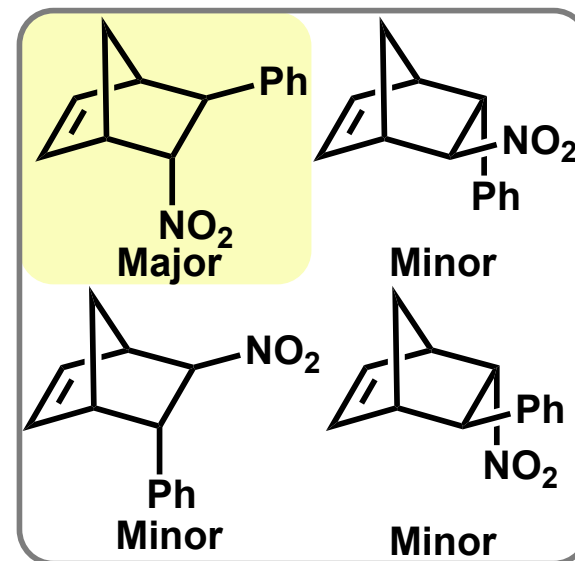
Enantio-Selectivity: My Proposal II



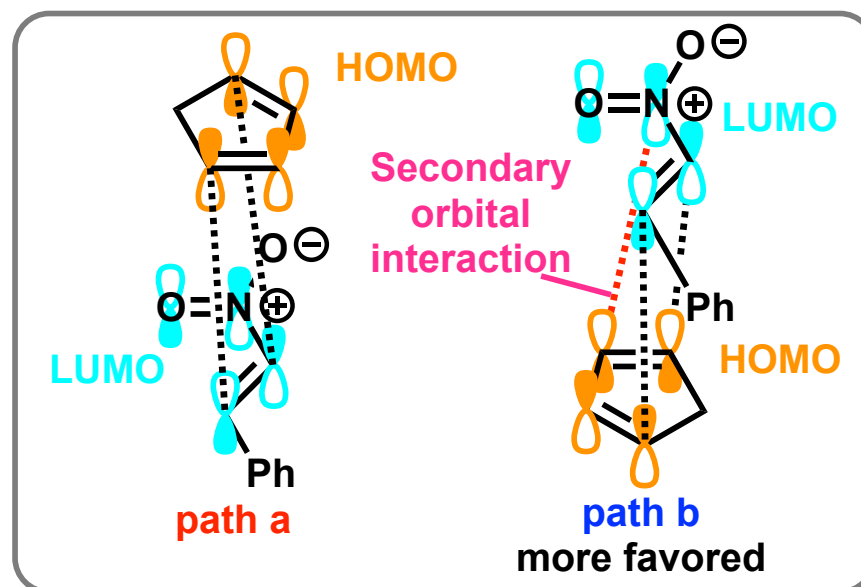
See along the direction of the Y axis



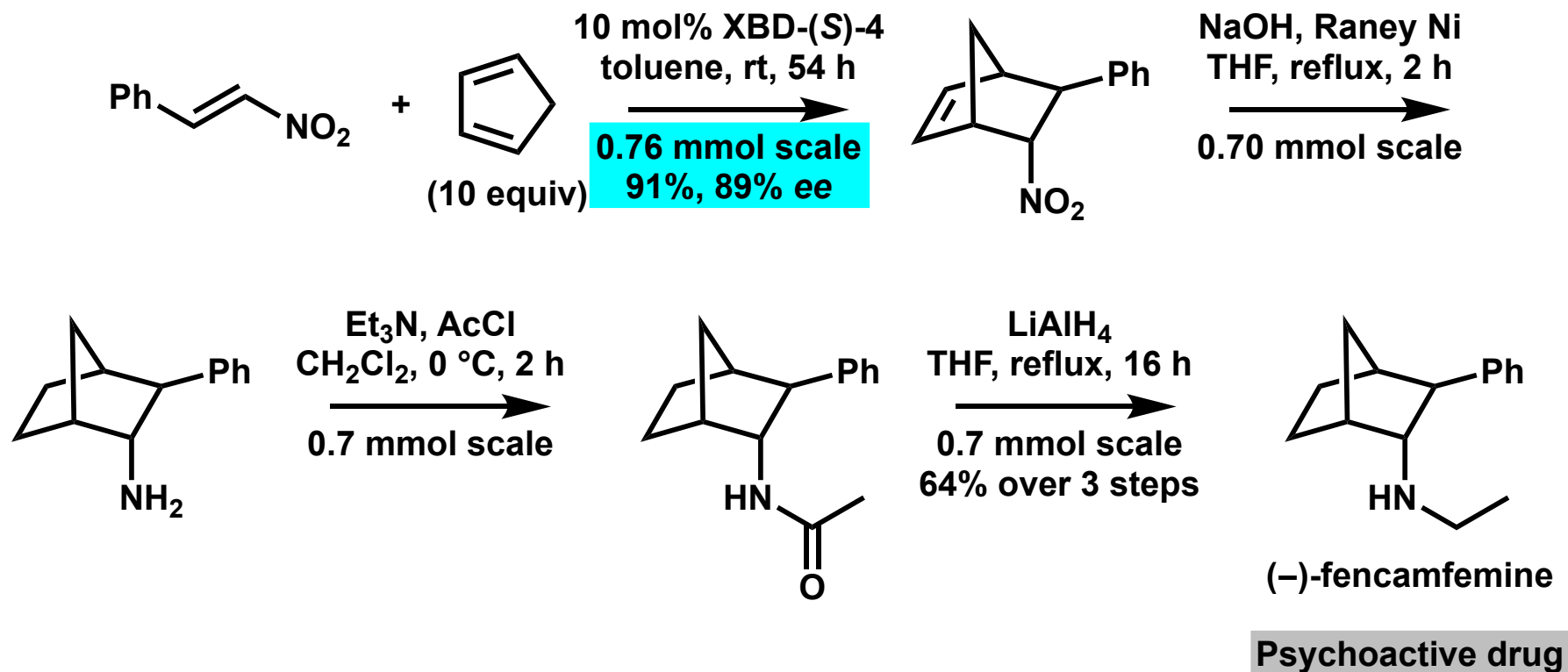
Products



Transition state of Diels-Alder reaction

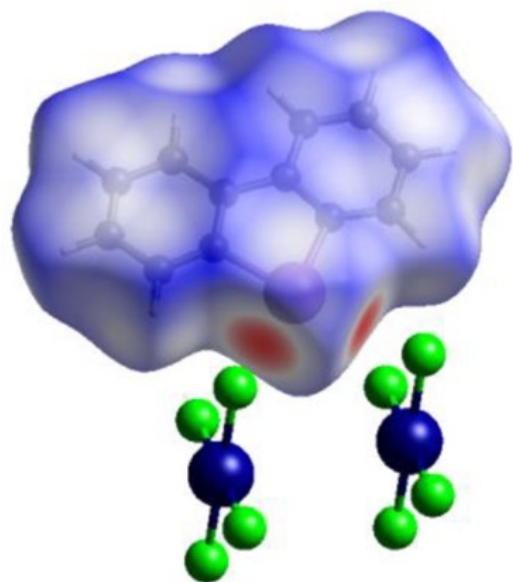
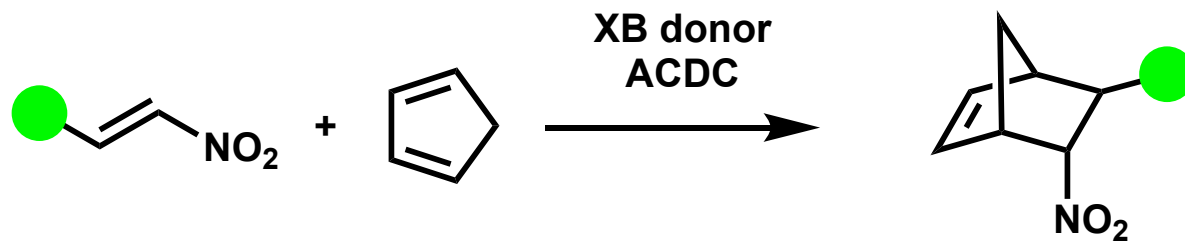


Application to Synthesis of (–)-fencamfamine

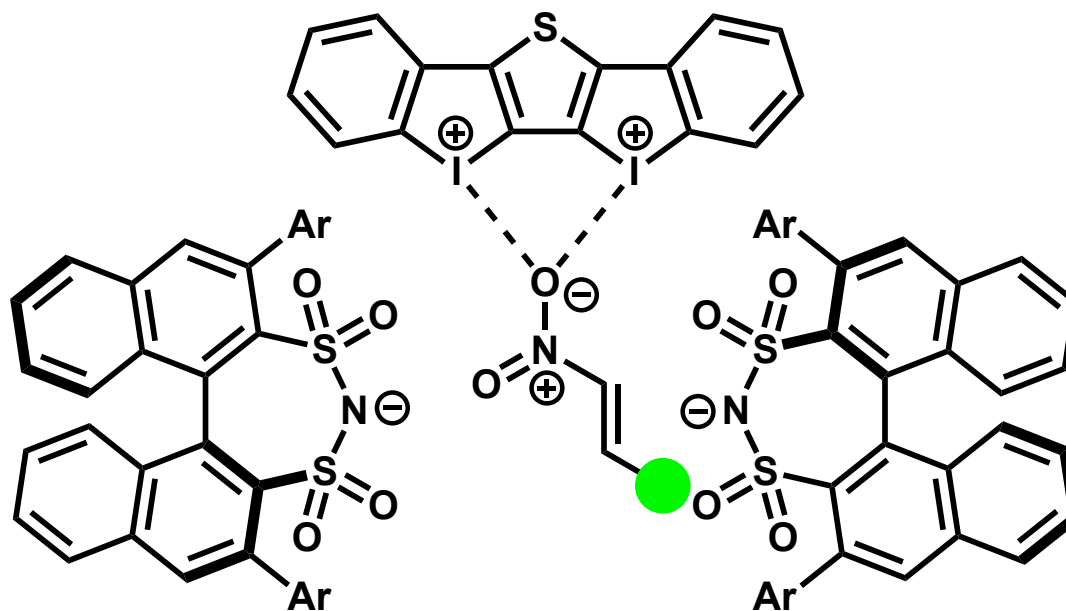


Summary

Highly enantioselective Diels-Alder reaction

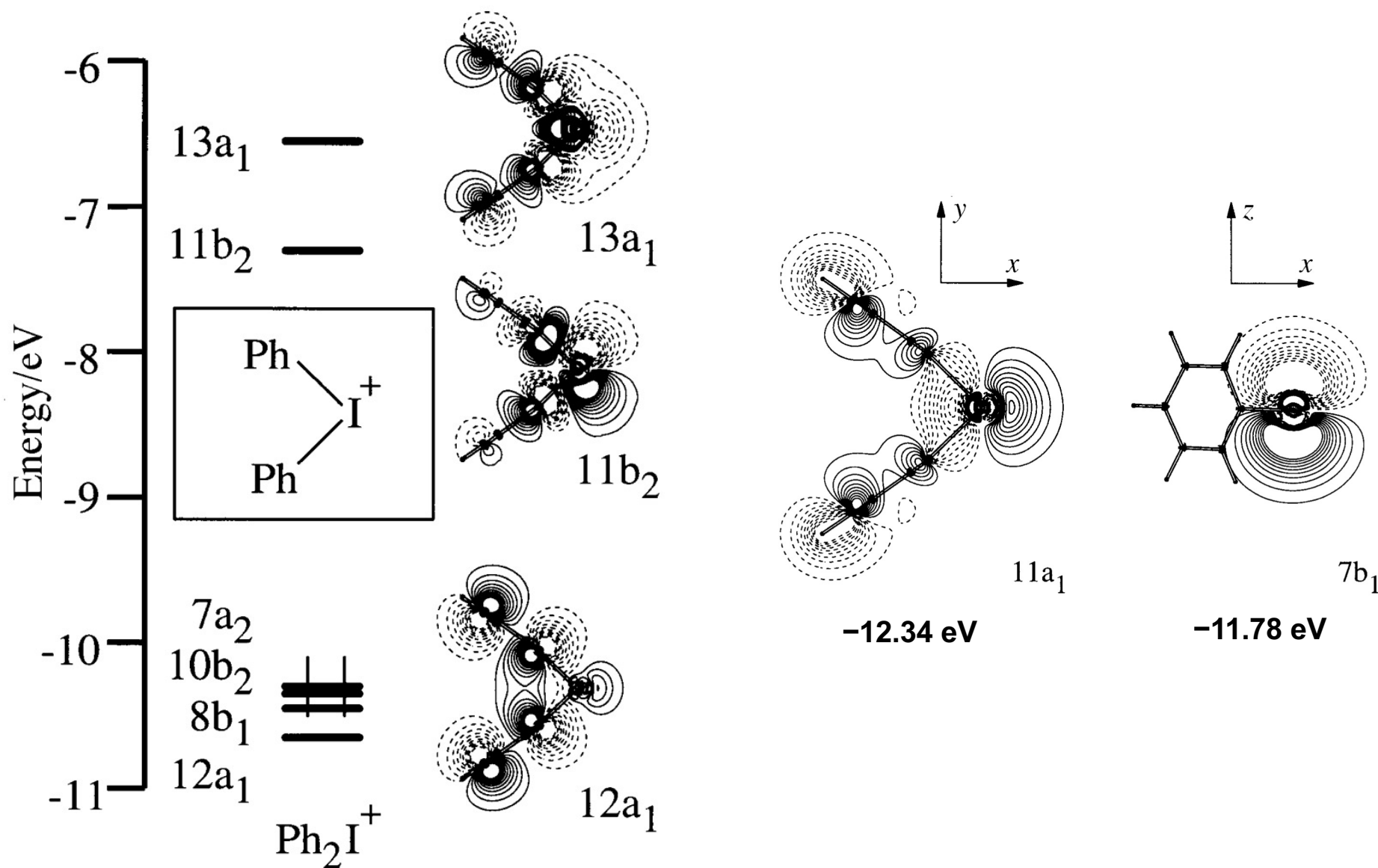


Double σ -hole of iodine (III)

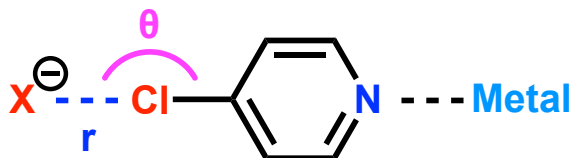


Appendix

Lone-Pair Orbital of Diphenyl Iodonium Salt

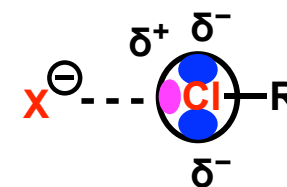


Halogen Halide Interaction



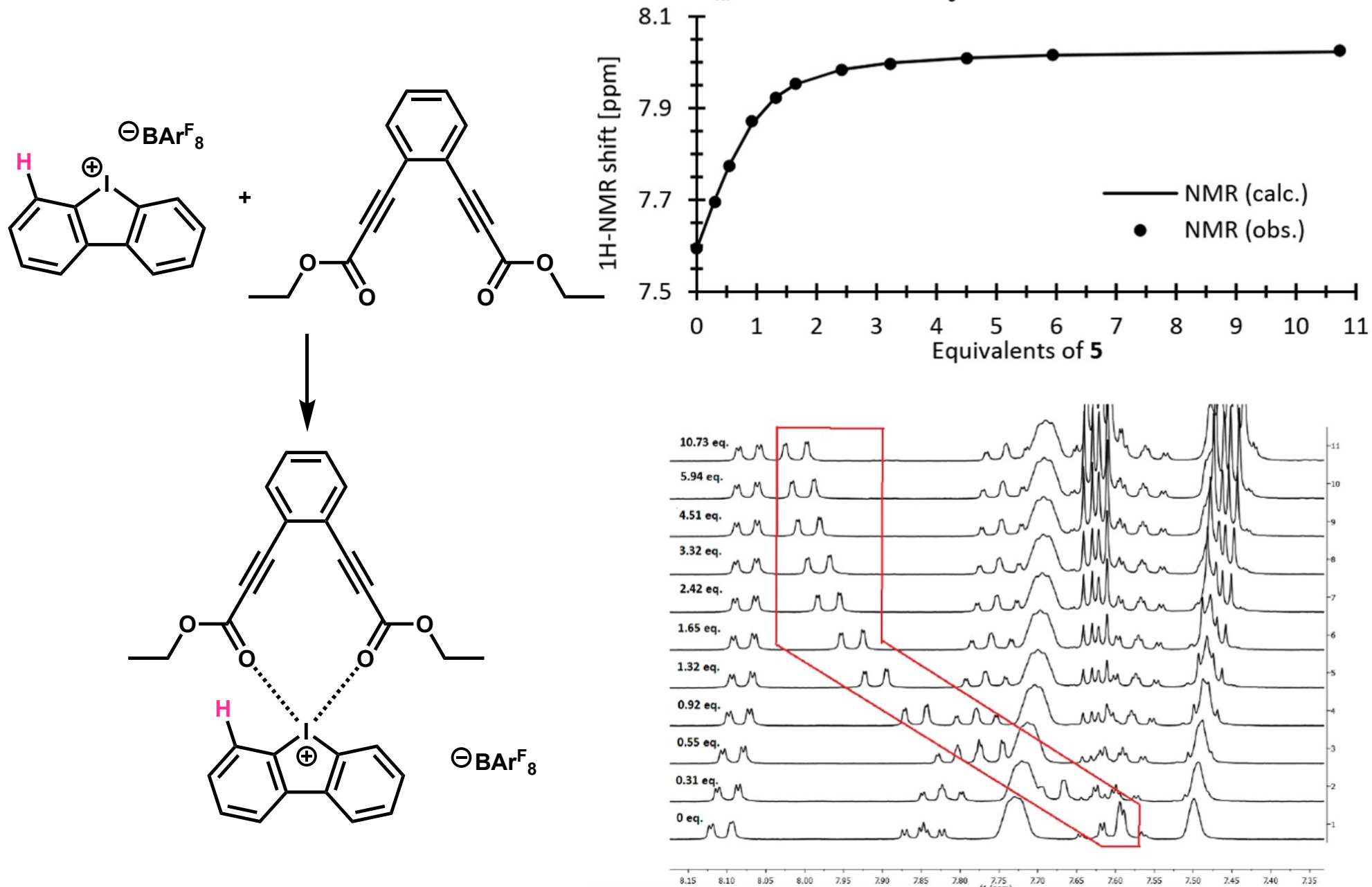
		r	θ	ΔE
no metal	$\text{F}^{\ominus}$	2.344 Å	179.7 °	-14.28 kcal/mol
	$\text{Cl}^{\ominus}$	3.093 Å	179.8 °	-6.28 kcal/mol
	$\text{Br}^{\ominus}$	3.309 Å	175.6 °	-5.22 kcal/mol
$\text{Cu}^{\oplus}$	$\text{F}^{\ominus}$	2.243 Å	180.0 °	-27.99 kcal/mol
	$\text{Cl}^{\ominus}$	2.930 Å	180.0 °	-16.87 kcal/mol
	$\text{Br}^{\ominus}$	3.119 Å	180.0 °	-15.12 kcal/mol

b97-1/lanl2dz(pp) (for Cu)
b97-1/aug-cc-pvdz (for others)

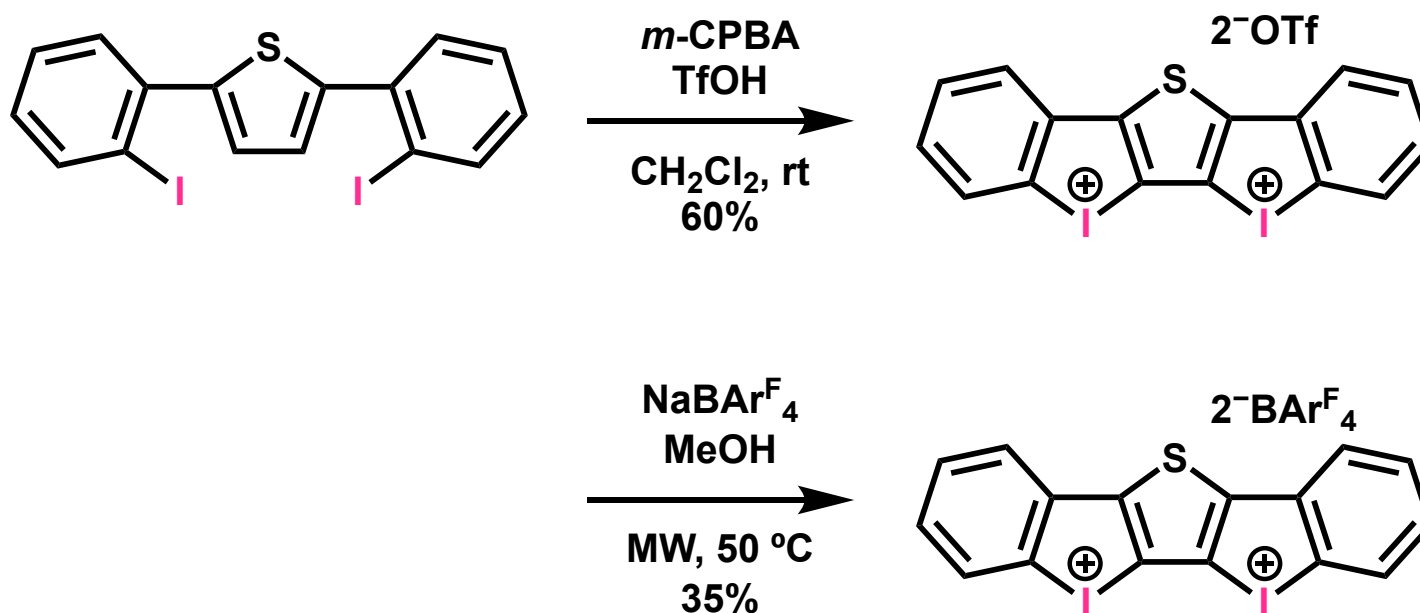


Halide interacts with σ -hole of halogen.

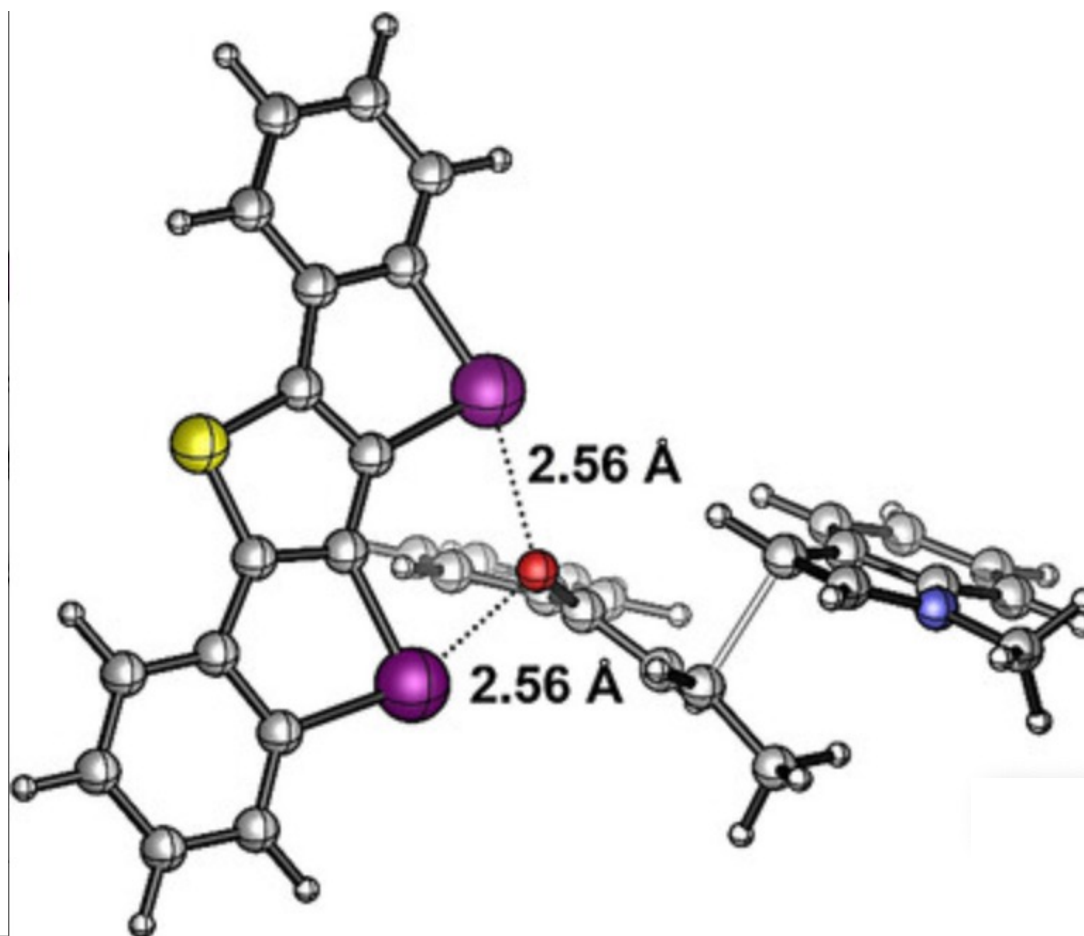
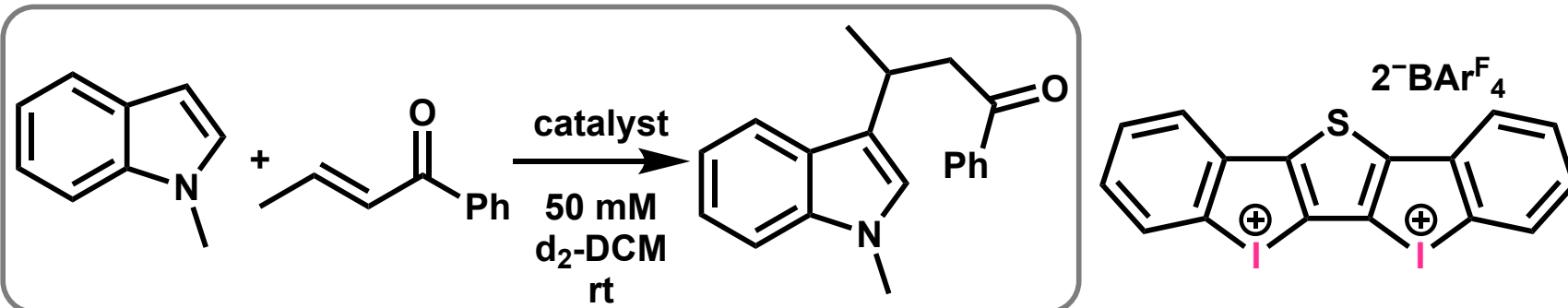
^1H NMR & ITC Experiment



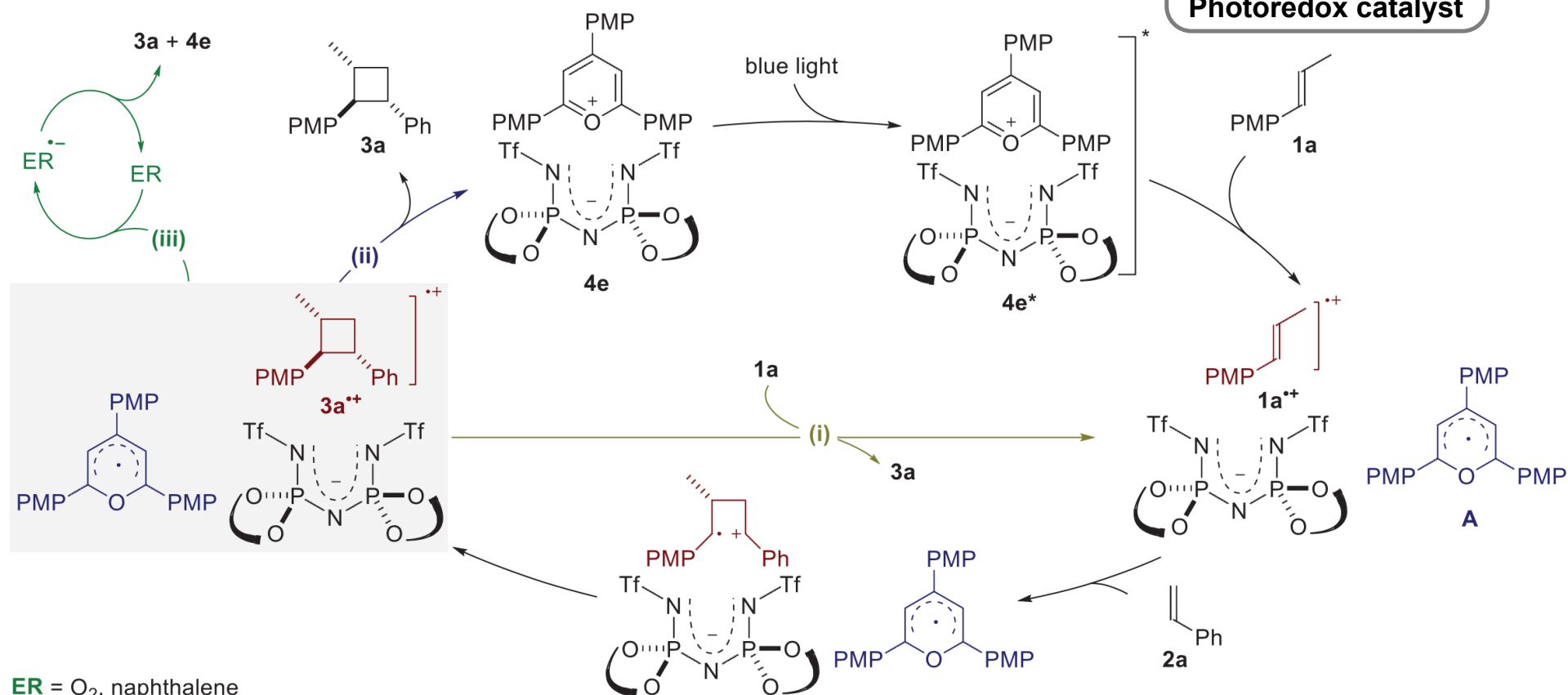
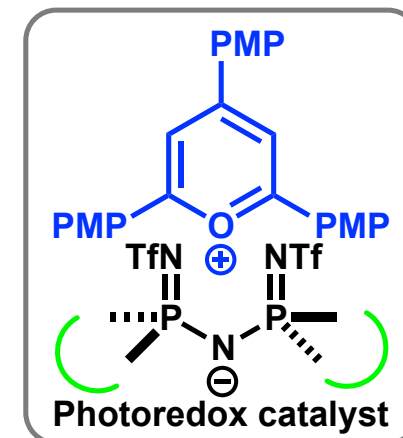
Synthesis of XB Donor



DFT Calculation: Michael Addition



2023³⁾: ACDC for Photoredox



ER = O₂, naphthalene

Catalyst Synthesis

