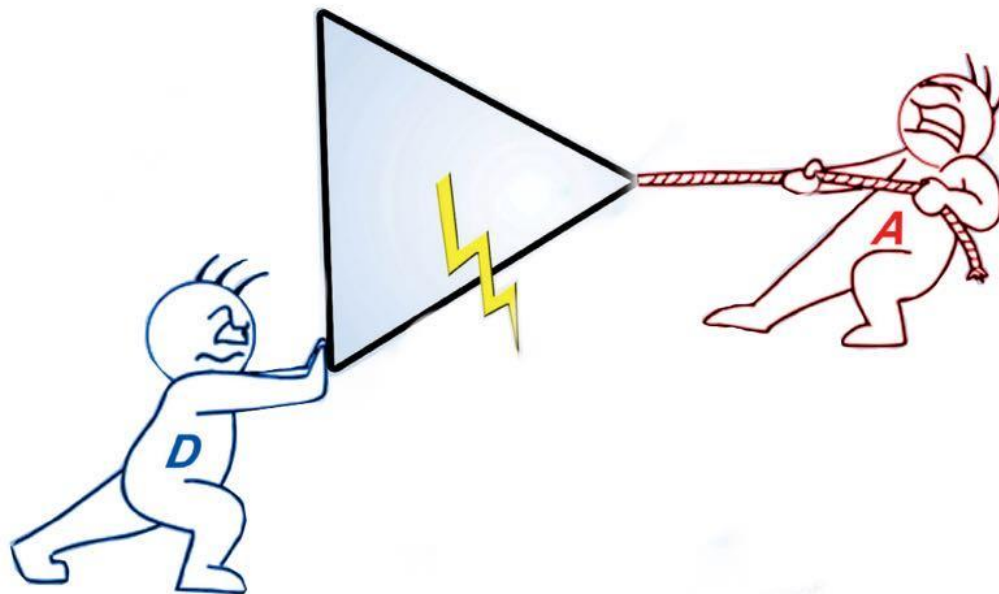


Construction of carbocycles with donor-acceptor cyclopropane



2024.11.14 Literature Seminar
D3 Yuyan Liang

Contents

0. Introduction

1. Lewis acid catalyzed intramolecular [3+2] cross-cycloaddition of cyclopropane 1,1-diesters with alkenes.

(*Angew. Chem. Int. Ed.* **2013**, *52*, 2032.)

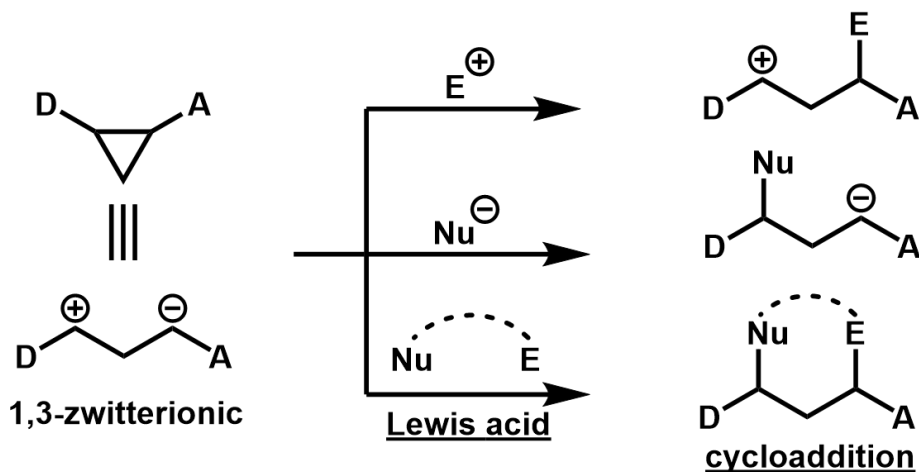
2. Lewis acid catalyzed dearomative [3+2] parallel-cycloaddition of cyclopropane with benzene.

(*Nat. Commun.* **2024**, *15*, 9206.)

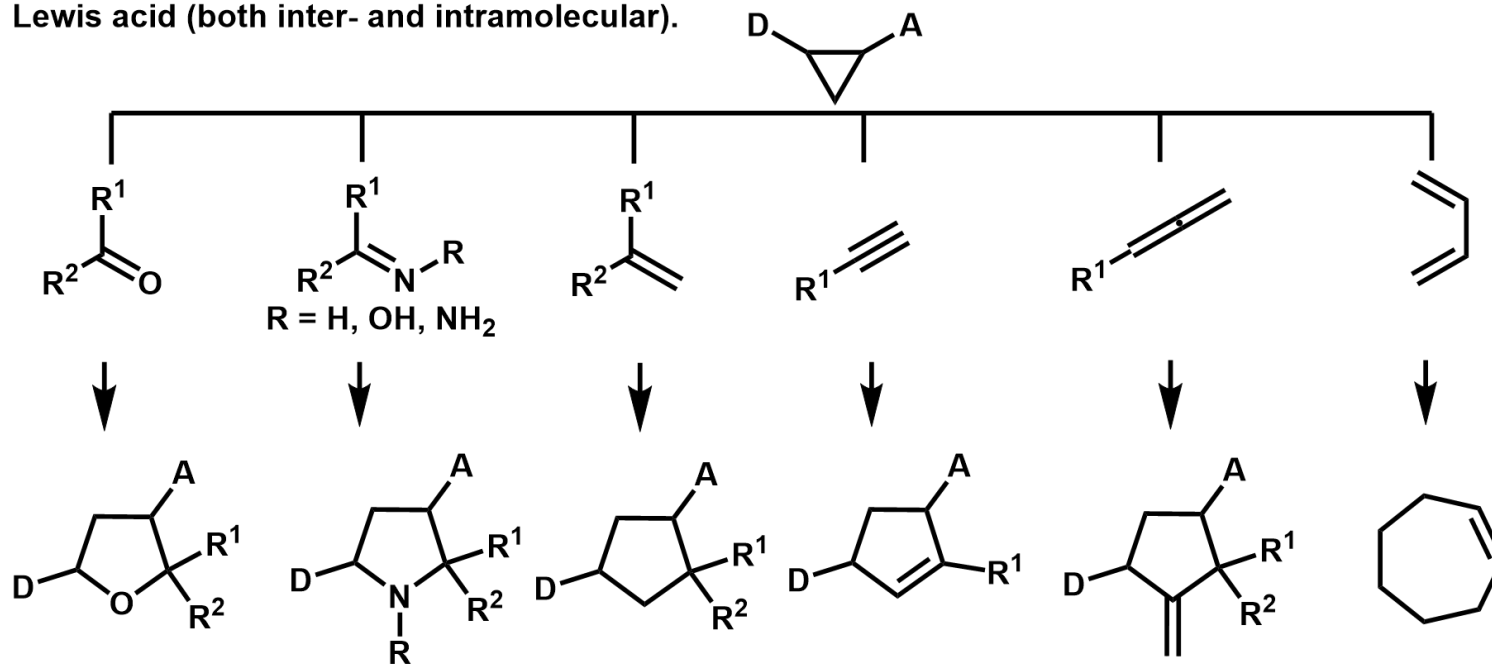
Donor-Acceptor Cyclopropanes

1. Activated cyclopropane with donor and acceptor substituents

→ electronic density distribution (significant polarization)



2. Cycloadditions of activated Donor-Acceptor cyclopropanes with diverse dipolarophiles catalyzed by Lewis acid (both inter- and intramolecular).



Prof. Zhongwen Wang



career:

-1990: B.S. @ Nankai University

1990-1996: Ph.D. @ Nankai University

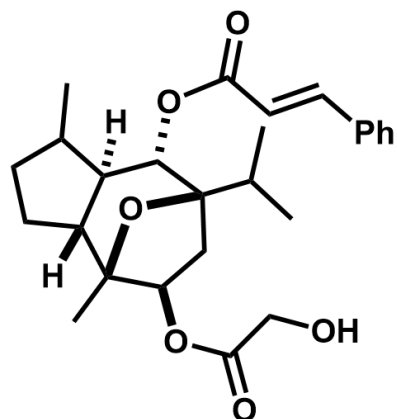
1999-2001: Postdoctoral : @ Nagoya University, Japan (Prof. H. Shinohara)

2001-2004: Postdoctoral : @ University of Kentucky, USA
(Prof. Mark S. Meier, Prof. H.Peter Spielmann)

2004- : Professor, Nankai University

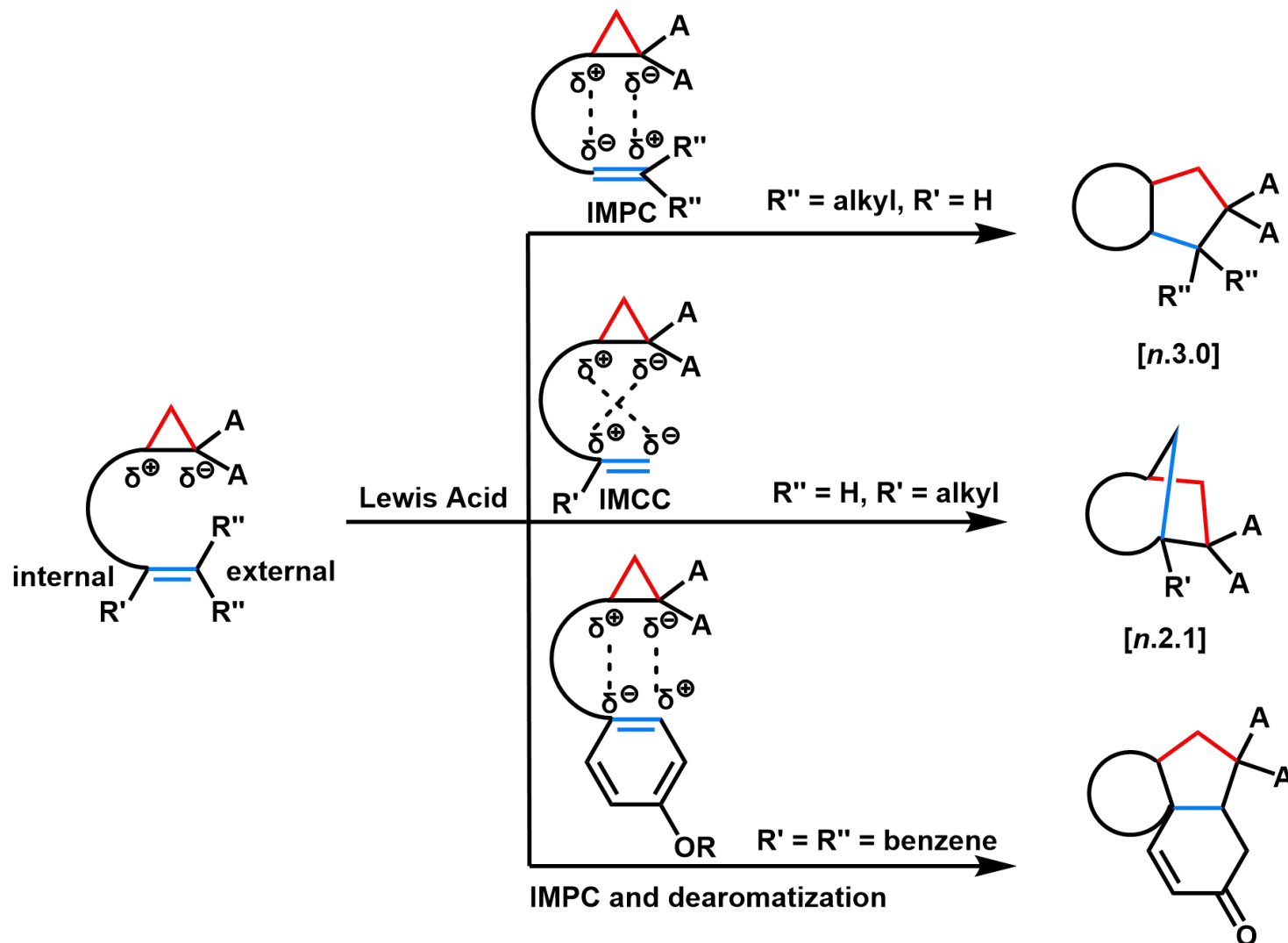
Research interests:

1. Synthetic methodology (Highly efficient and general strategy for construction of polycyclic skeletons);
Novel cycloaddition and tandem reactions of strained rings
2. Natural products total synthesis;
3. New pharmaceutical and agrochemical leads discovery



(-)-englerin A

Intramolecular Cycloaddition of cyclopropane



IMPC: intramolecular parallel cycloaddition
 IMCC: intramolecular cross cycloaddition

Regioselectivity could be finely tuned by changing substituents of alkene.

Contents

0. Introduction

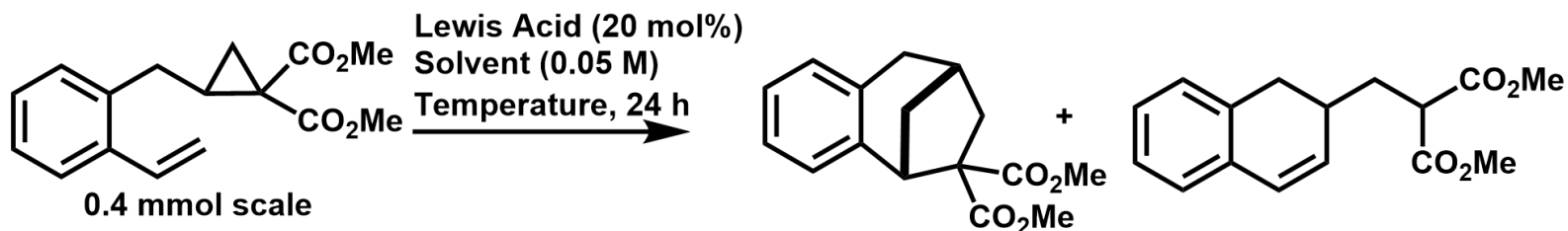
1. Lewis acid catalyzed intramolecular [3+2] cross-cycloaddition of cyclopropane 1,1-diesters with alkenes.

(*Angew. Chem. Int. Ed.* **2013**, *52*, 2032.)

2. Lewis acid catalyzed dearomative [3+2] parallel-cycloaddition of cyclopropane with benzene.

(*Nat. Commun.* **2024**, *15*, 9206.)

Optimization



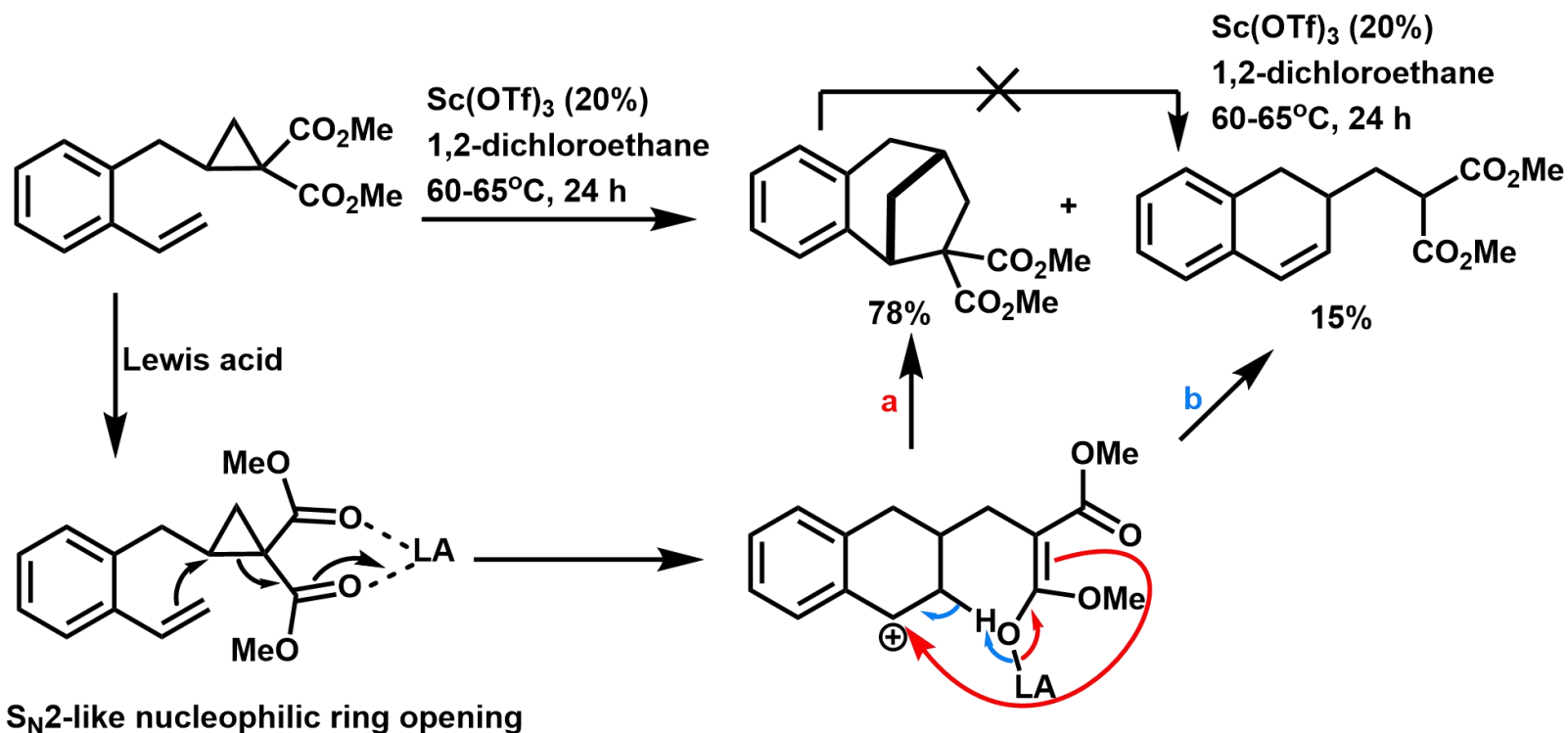
Entry	Lewis Acid	Solvent	Temperature	Yield [%]
1	Sc(OTf) ₃	1,2-dichloroethane	40-45	67
2	Sc(OTf) ₃ (10 mol%)	1,2-dichloroethane	60-65	80
3	Sc(OTf) ₃	1,2-dichloroethane	60-65	78*
4	Sc(OTf) ₃	toluene	60-65	67
5	Sc(OTf) ₃	THF	60-65	29
6	Sc(OTf) ₃	DMF	60-65	NR
7	Sc(OTf) ₃	CH ₃ NO ₂	60-65	79
8	Yb(OTf) ₃	1,2-dichloroethane	60-65	NR
9	Cu(OTf) ₃	1,2-dichloroethane	60-65	43
10	Zn(OTf) ₃	1,2-dichloroethane	60-65	NR
11	In(OTf) ₃	toluene	60-65	72
12	ZnCl	THF	60-65	NR
13	BF ₃ ·Et ₂ O (100 mol%)	CH ₂ Cl ₂	-78 °C	NR
14	SnCl ₄ ·Et ₃ N	1,2-dichloroethane	60-65	NR
15	trifluoroacetic acid	1,2-dichloroethane	60-65	0
16	triflic acid	1,2-dichloroethane	60-65	complex

* Byproduct was obtained in 15%, other entries did not mention.

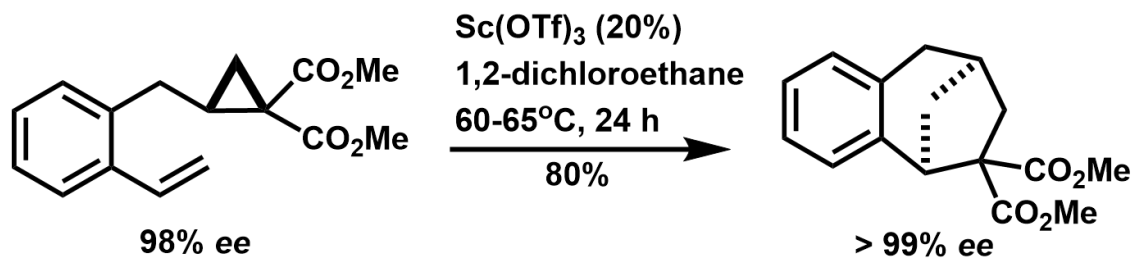
Mechanism

The S_N2-like stepwise mechanism is proposed based on 2 evidences.

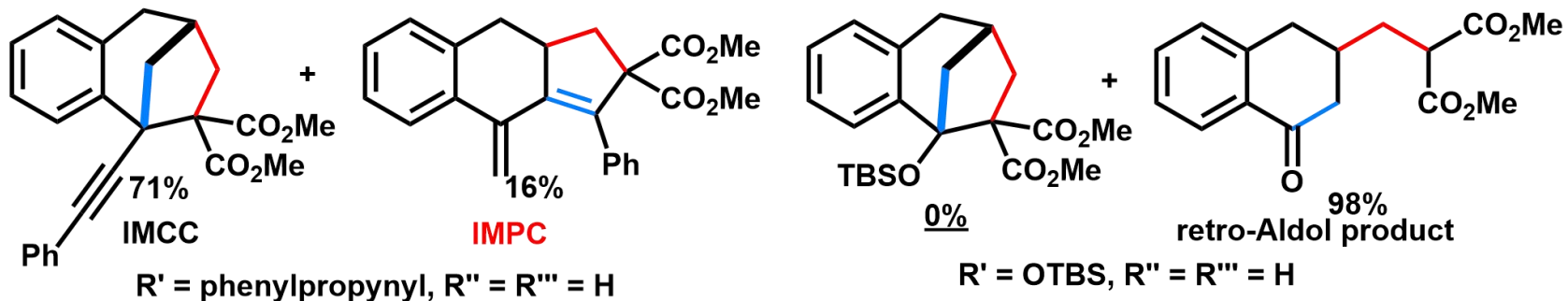
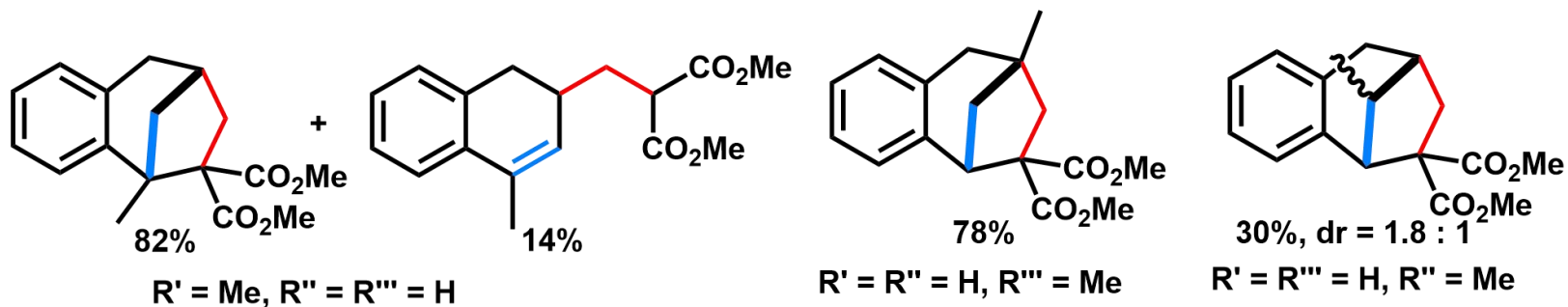
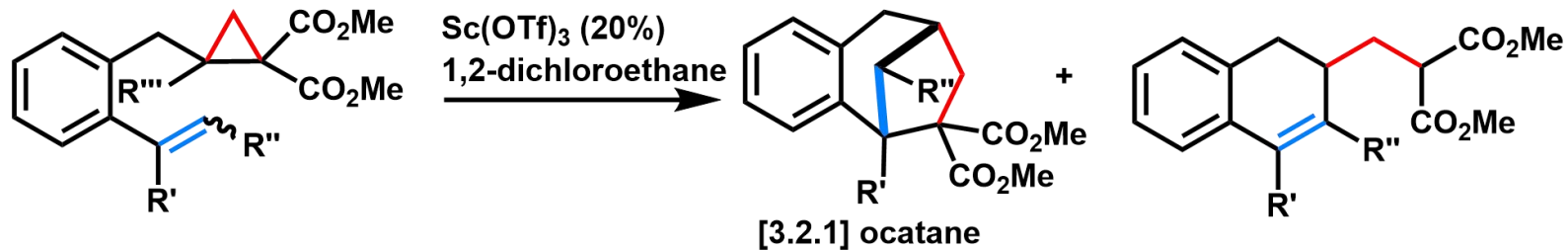
1. Irreversible transformation



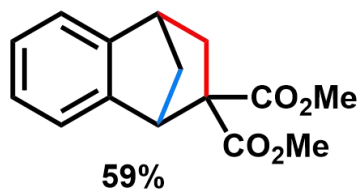
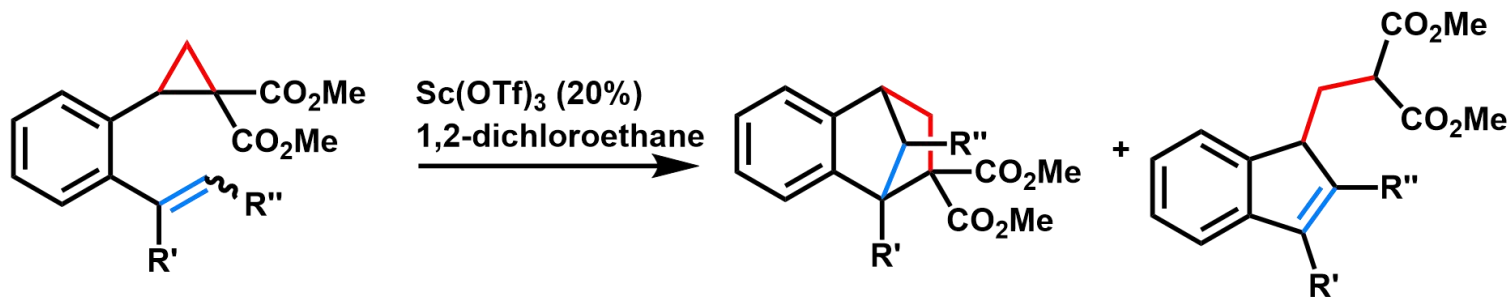
2. Asymmetric experiment: no erosion of enantiomeric excess



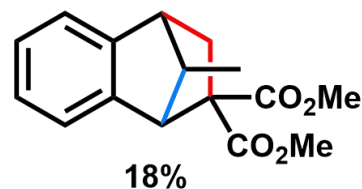
Substrate scope (1)



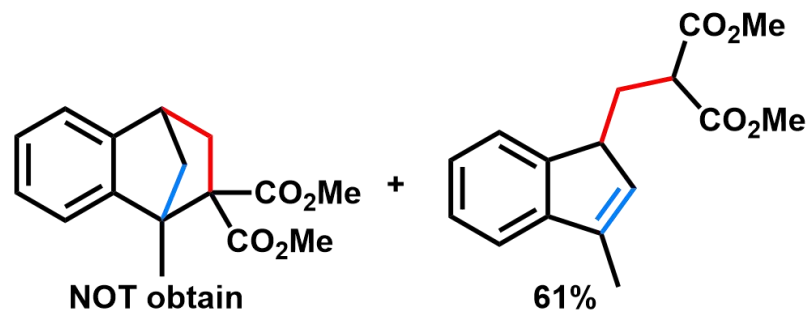
Substrate scope (1)



R' = R'' = H



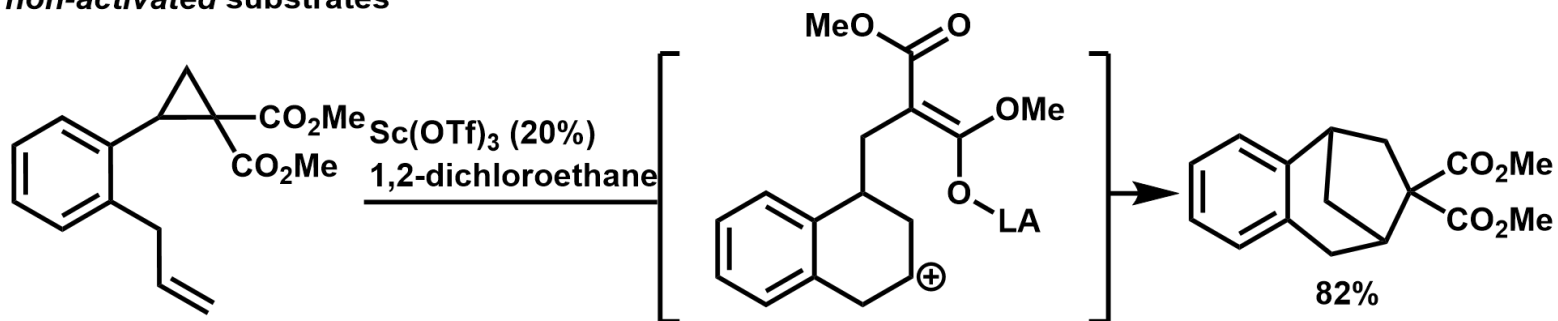
R' = H, R'' = Me



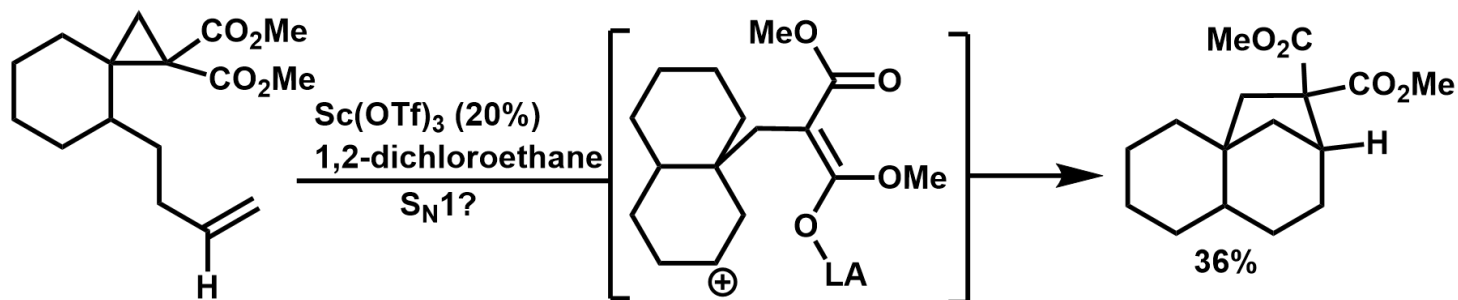
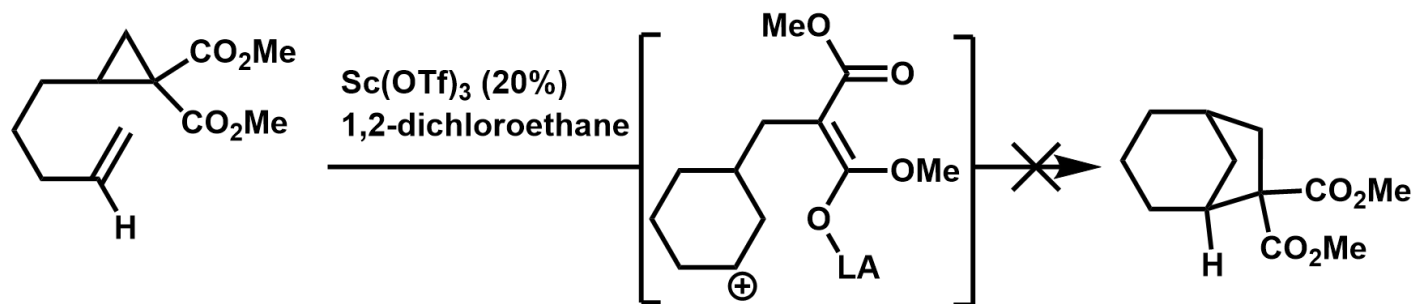
R' = Me, R'' = H

Substrate scope (2)

non-activated substrates



The first example of cycloaddition with non-activated alkene.

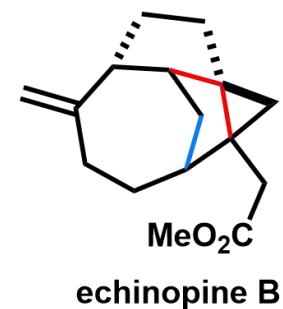
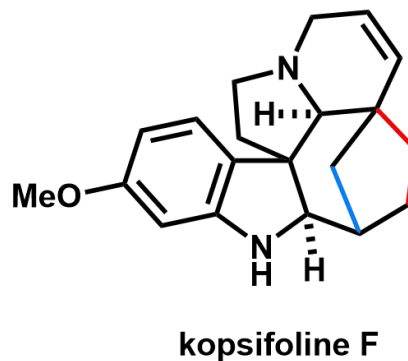
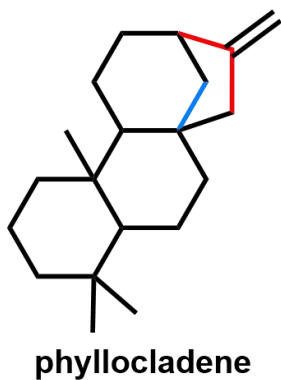
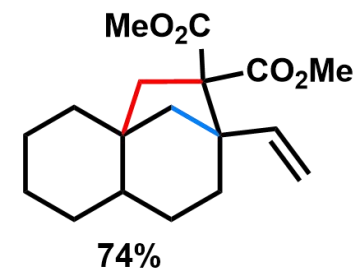
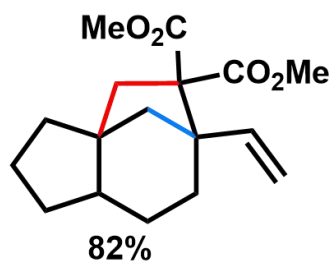
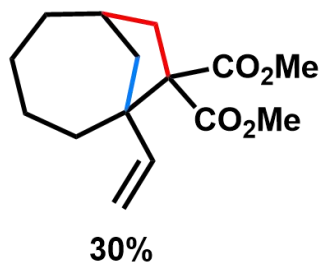
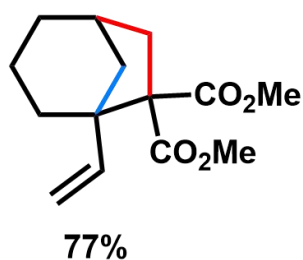
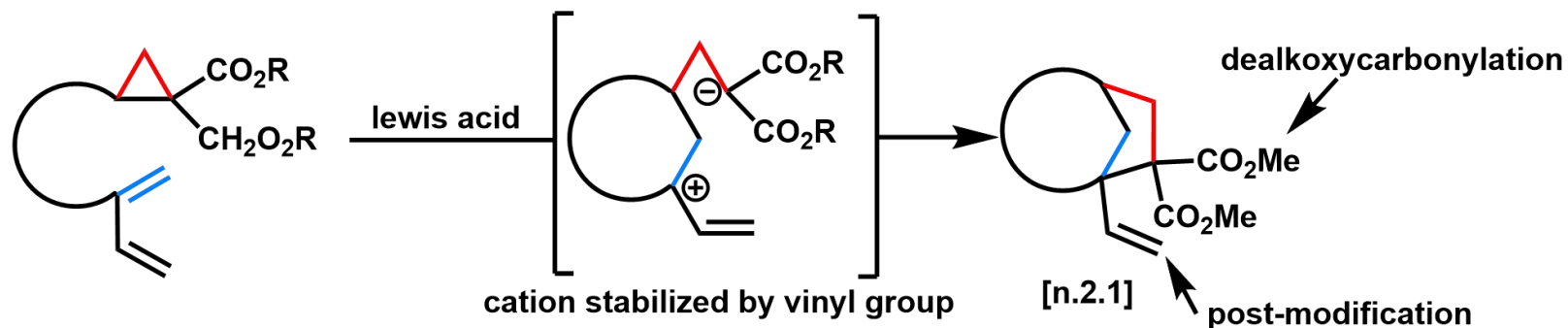


cleavage of highly strained cyclopropane may be a driving force too.

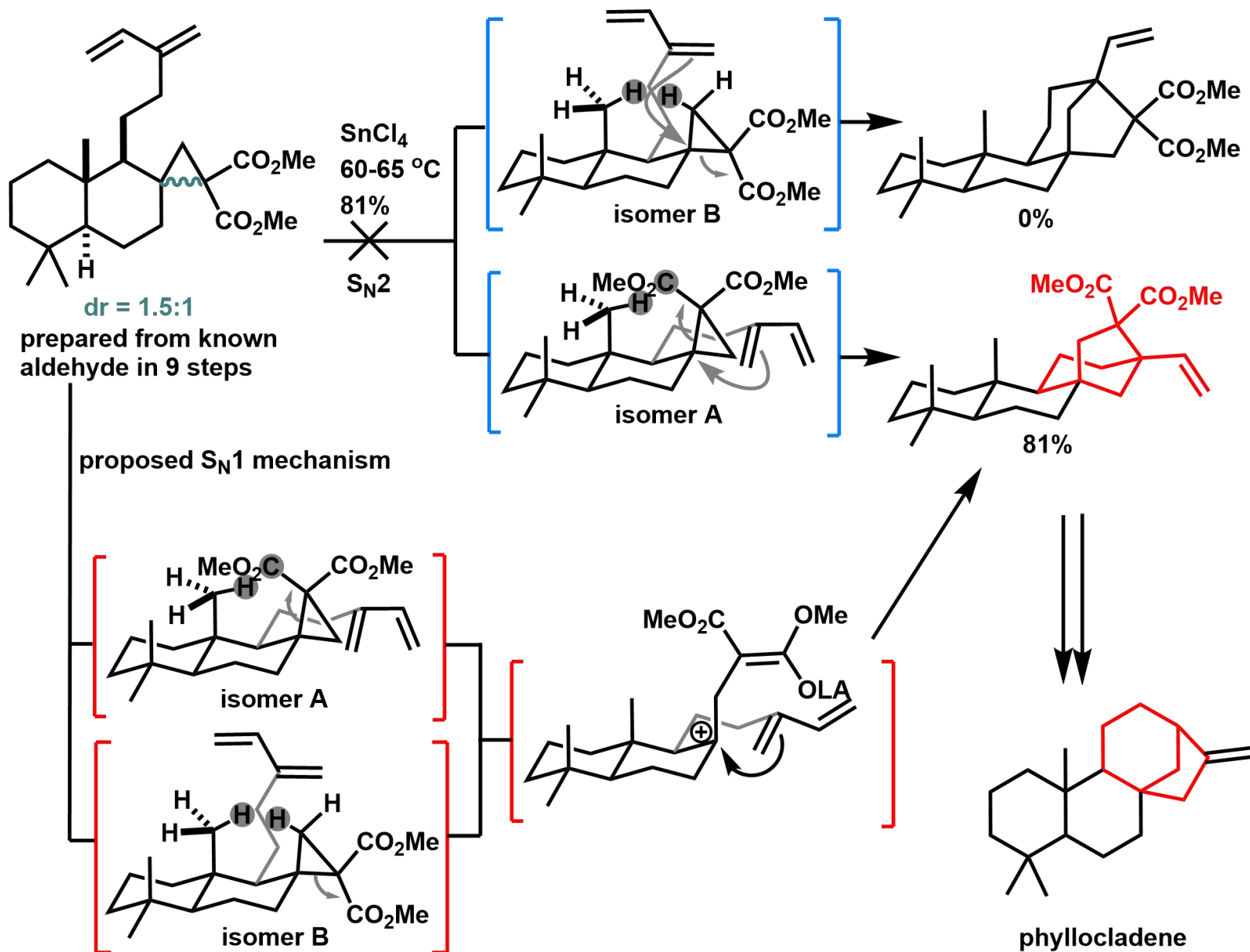
Stabilization of carbenium is necessary for high reaction yield.

Substrate scope (2)

non-benzofused substrates activated with vinyl group



Synthesis of phyllocladene



Contents

0. Introduction

1. Lewis acid catalyzed intramolecular [3+2] cross-cycloaddition of cyclopropane 1,1-diesters with alkenes.

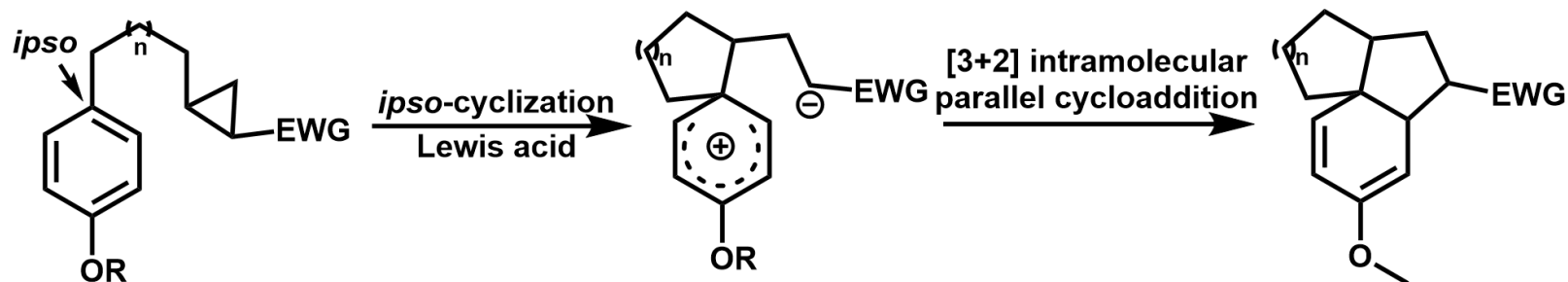
(*Angew. Chem. Int. Ed.* **2013**, *52*, 2032.)

2. Lewis acid catalyzed dearomative [3+2] parallel-cycloaddition of cyclopropane with benzene.

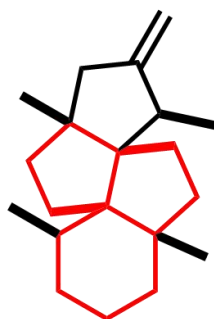
(*Nat. Commun.* **2024**, *15*, 9206.)

dearomative (3+2) intramolecular parallel cycloaddition of donor-acceptor cyclopropane

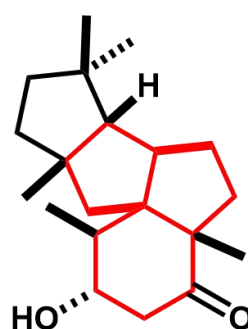
Design concept



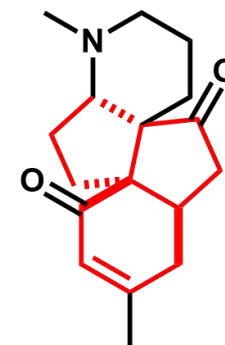
Overcome the inherent rearomatization of aryl compounds;
The angular tricycle skeletons are widely embedded in complex natural products



waihoensene
200919_PS_Yuma_Komori

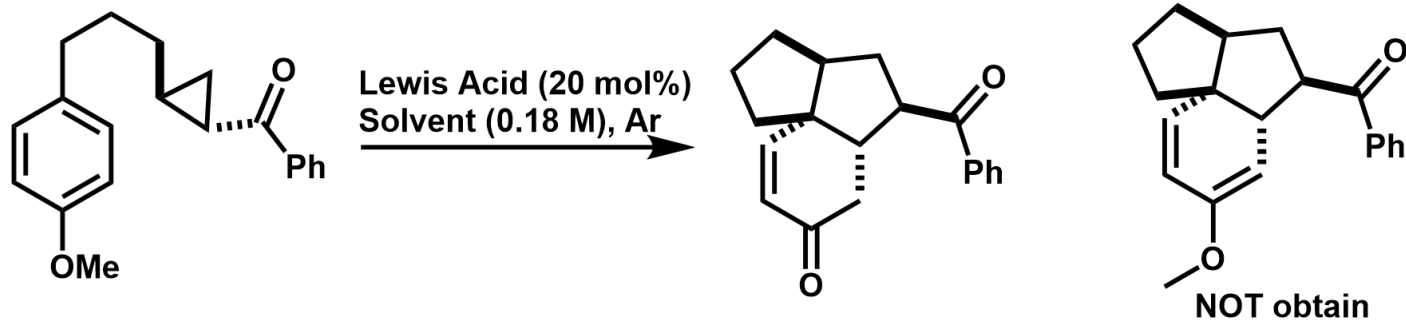


conidiogenone
221029_LS_Manaka_Matsumoto



lycojaponicum C
170715_PS_Keshu_Zhang

Optimization

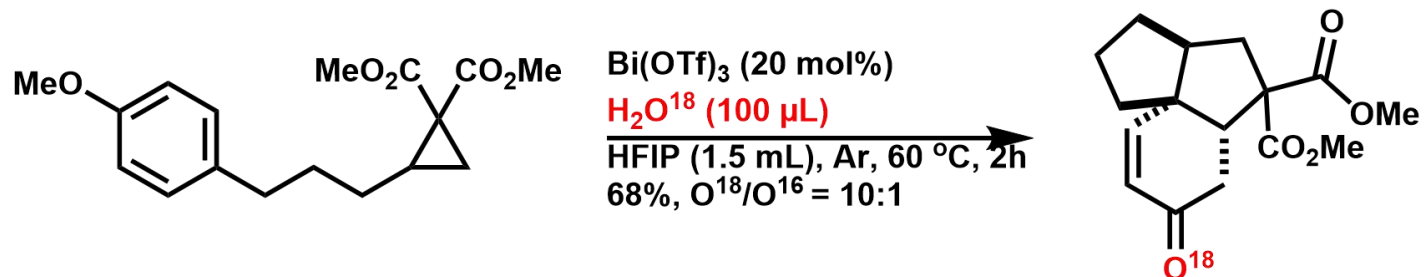


Entry	Lewis Acid	Solvent	Temperature	time (h)	Yield [%]
1	Sc(OTf) ₃	DCE	50	18	-
2	Yb(OTf) ₃	DCE	50	18	-
3	Cu(OTf) ₃	DCE	50	18	6
4	Al(OTf) ₃	DCE	50	18	28
5	AgSbF ₆	DCE	50	18	29
6	FeCl ₃	DCE	50	18	28
7	Bi(OTf) ₃	DCE	50	18	27
8	Al(OTf) ₃	HFIP	50	18	67
9	AgSbF ₆	HFIP	60	18	47
10	FeCl ₃	HFIP	60	18	48
11	Bi(OTf) ₃	HFIP	60	18	74
12*	Bi(OTf) ₃	HFIP	60	2	79

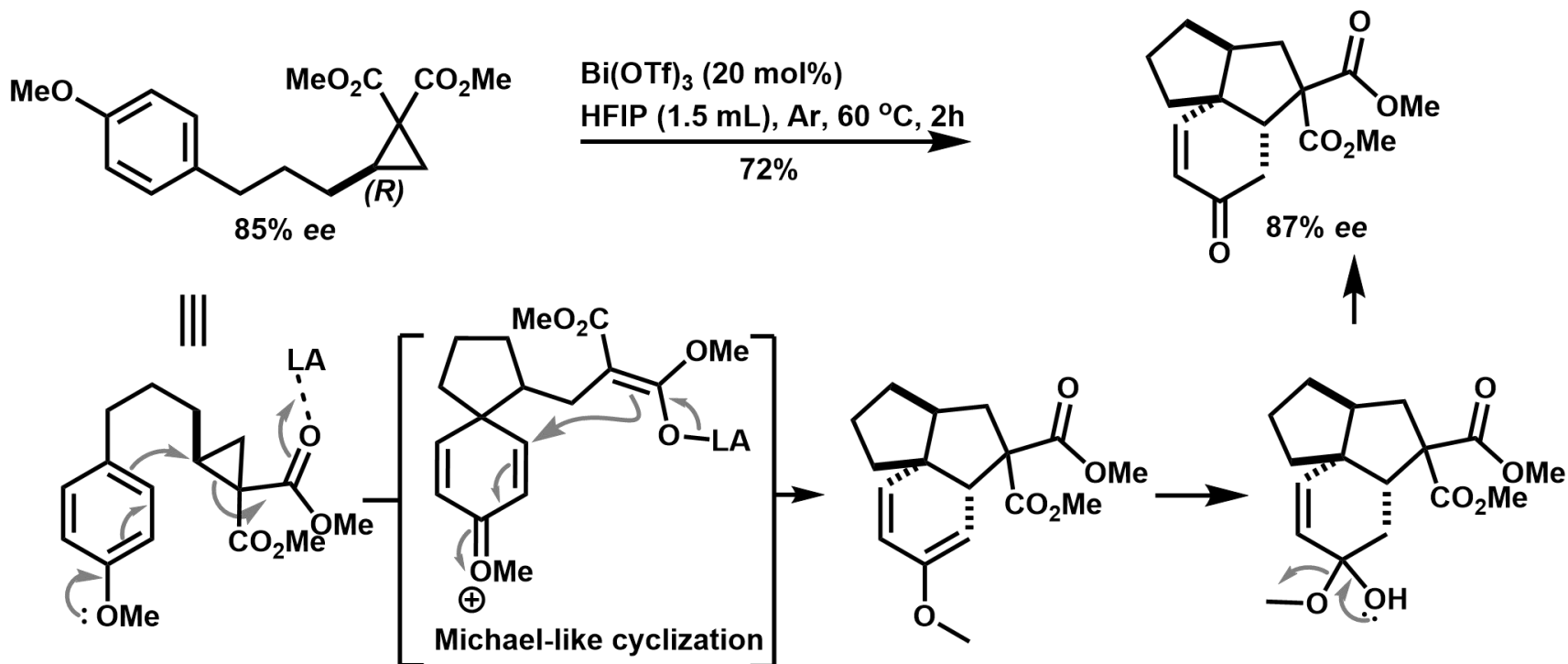
* Gram scale preparation: with addition of H₂O (1.5 eq),
2 grams of cyclopropane afforded 1.33 grams product.

Mechanism

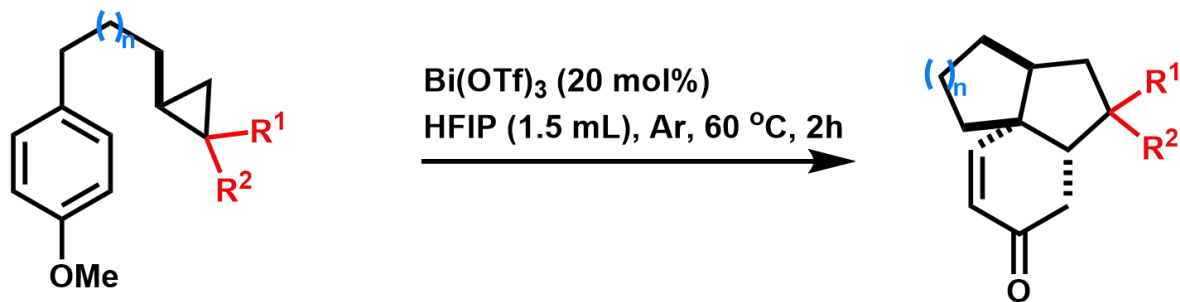
1. O¹⁸-labeled experiment



2. Asymmetric experiment

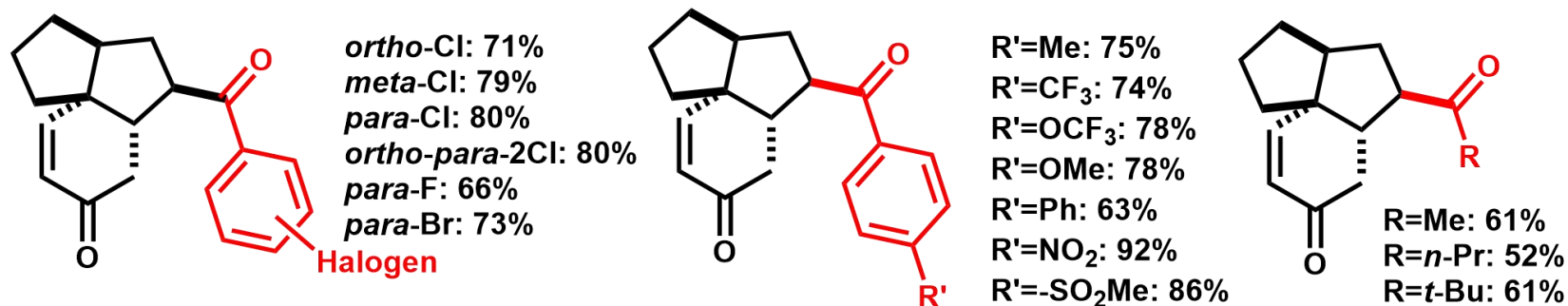


Substrate scope (1)

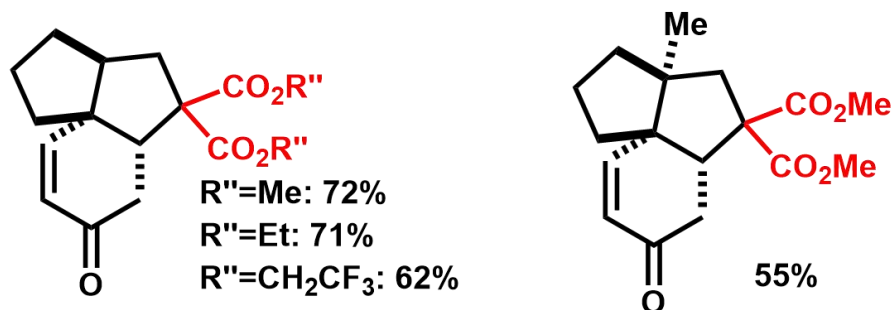


1. variation on acceptor ($n=1$)

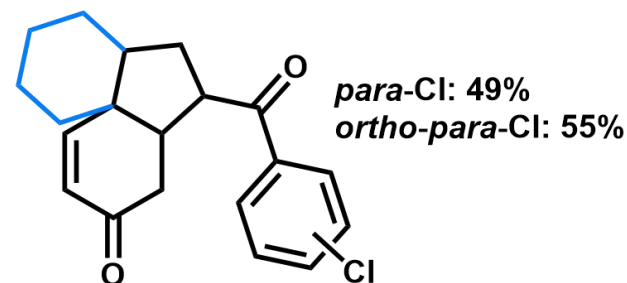
1.1 *mono*-acceptor on cyclopropane ($R^2=H$)



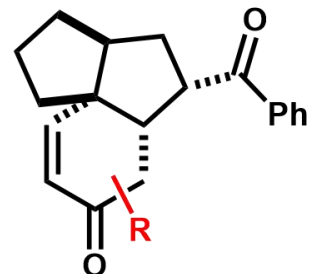
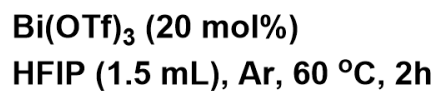
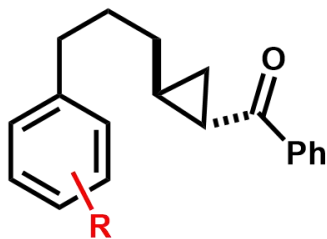
1.2 *double*-acceptor on cyclopropane ($R^1=R^2=CO_2R$)



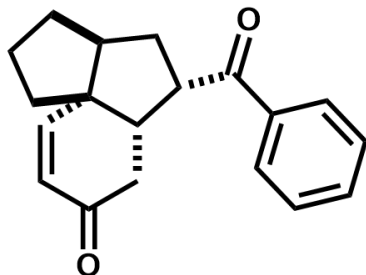
2. variation on tether ($n=2$)



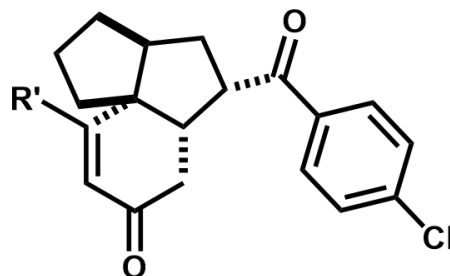
Substrate scope (2)



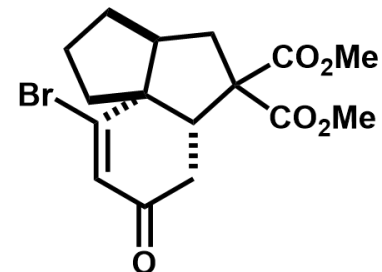
3. variation on benzene



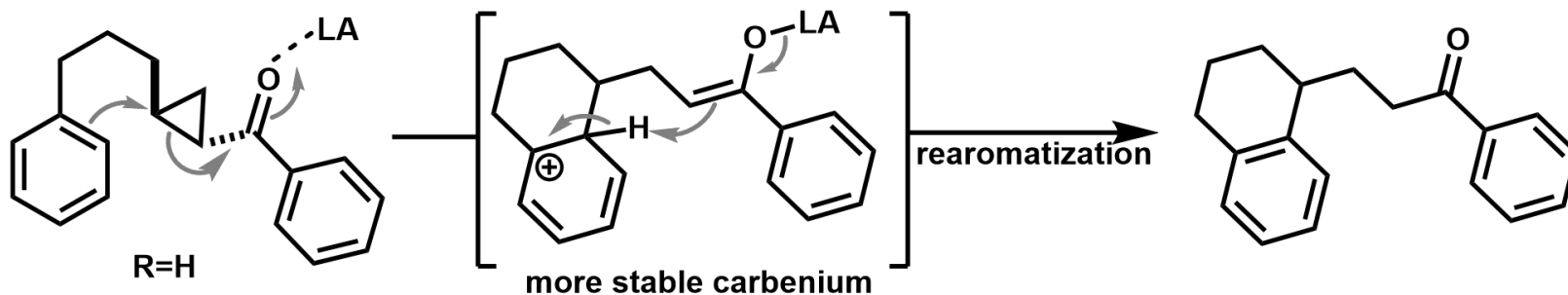
$\text{R} = \text{para-OCH}_2\text{CH}_2\text{Br}$: 74%
 $\text{R} = \text{para-OH}$: 61%



$\text{R}' = \text{Me}$: 69%
 $\text{R}' = \text{OMe}$: 26%



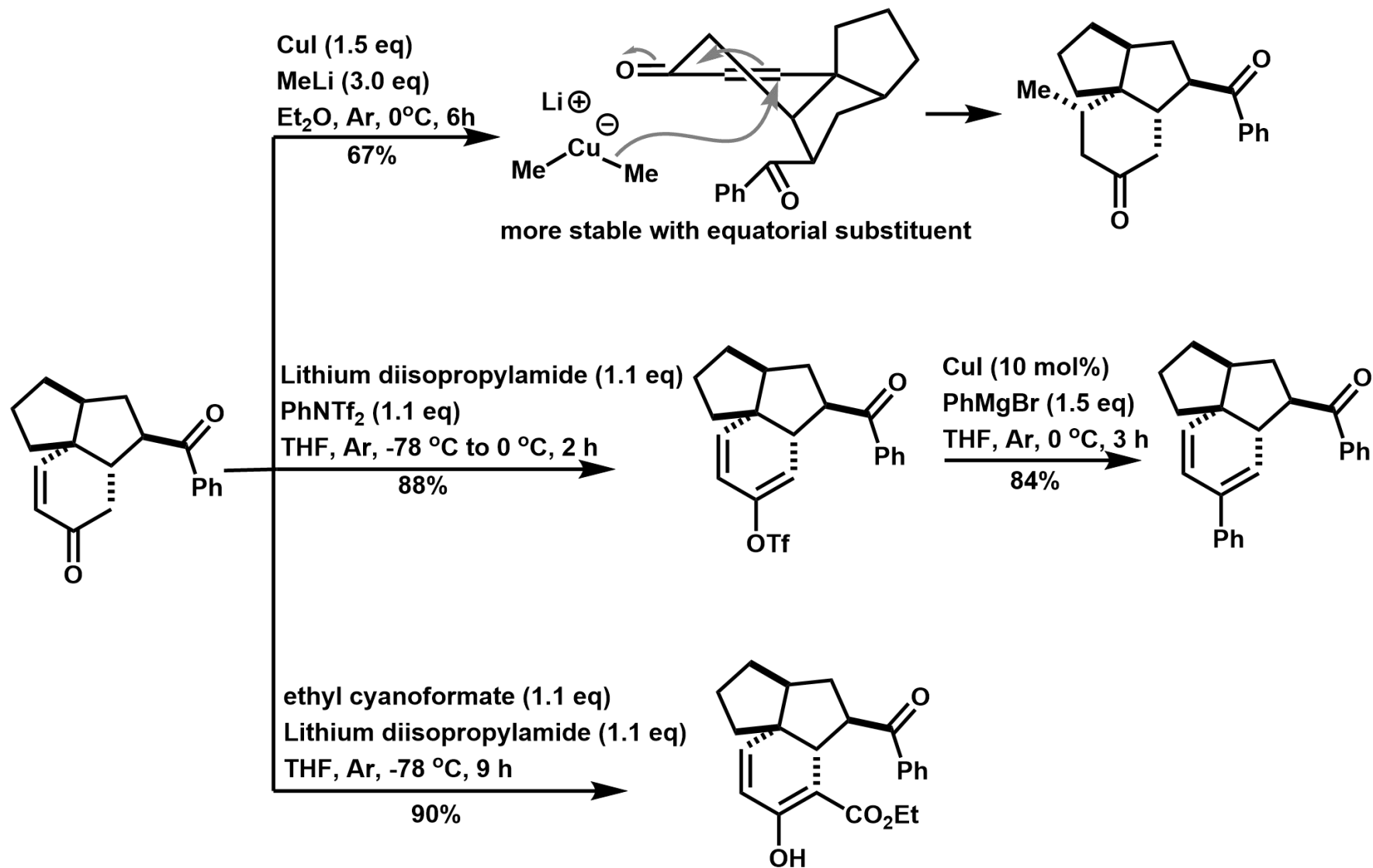
84%



In absence of directing group on benzene, only the *ortho*-Friedel-Crafts product was obtained.

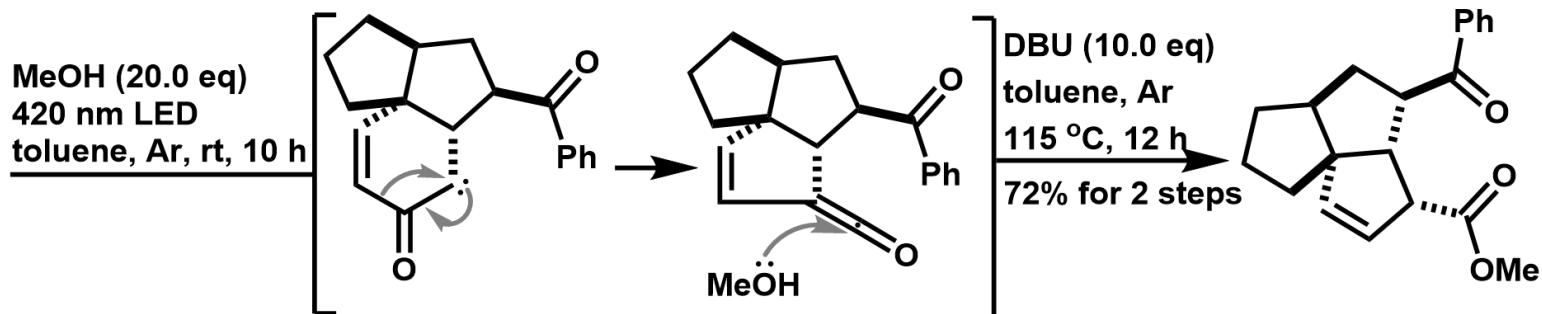
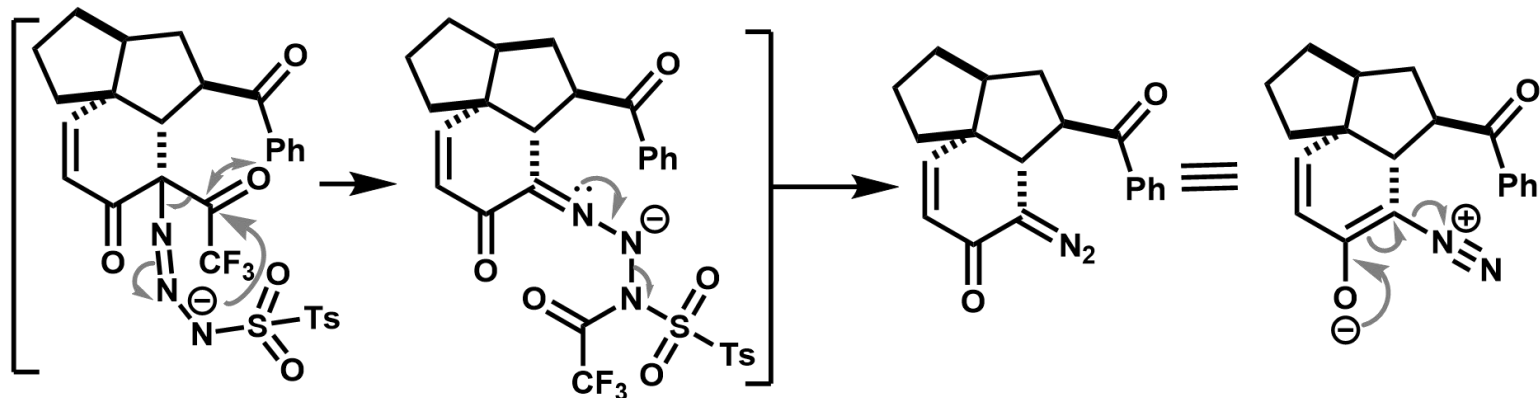
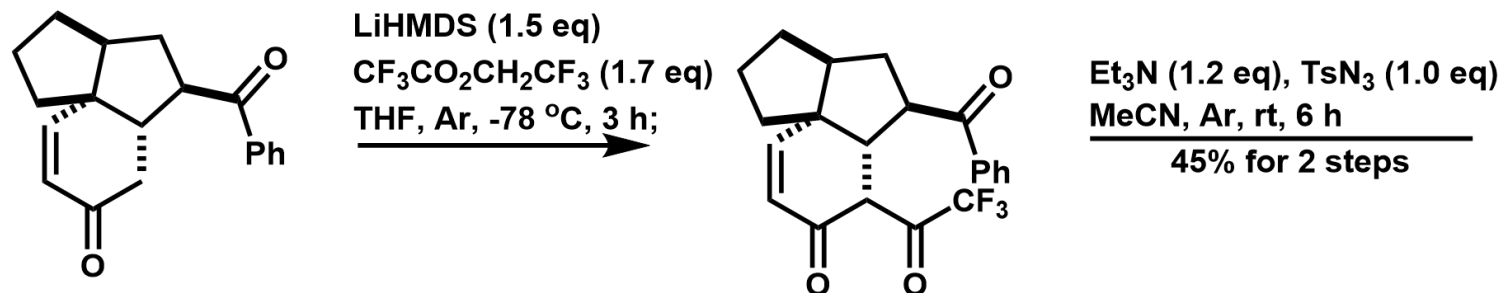
Post-functionalization (1)

ring decoration



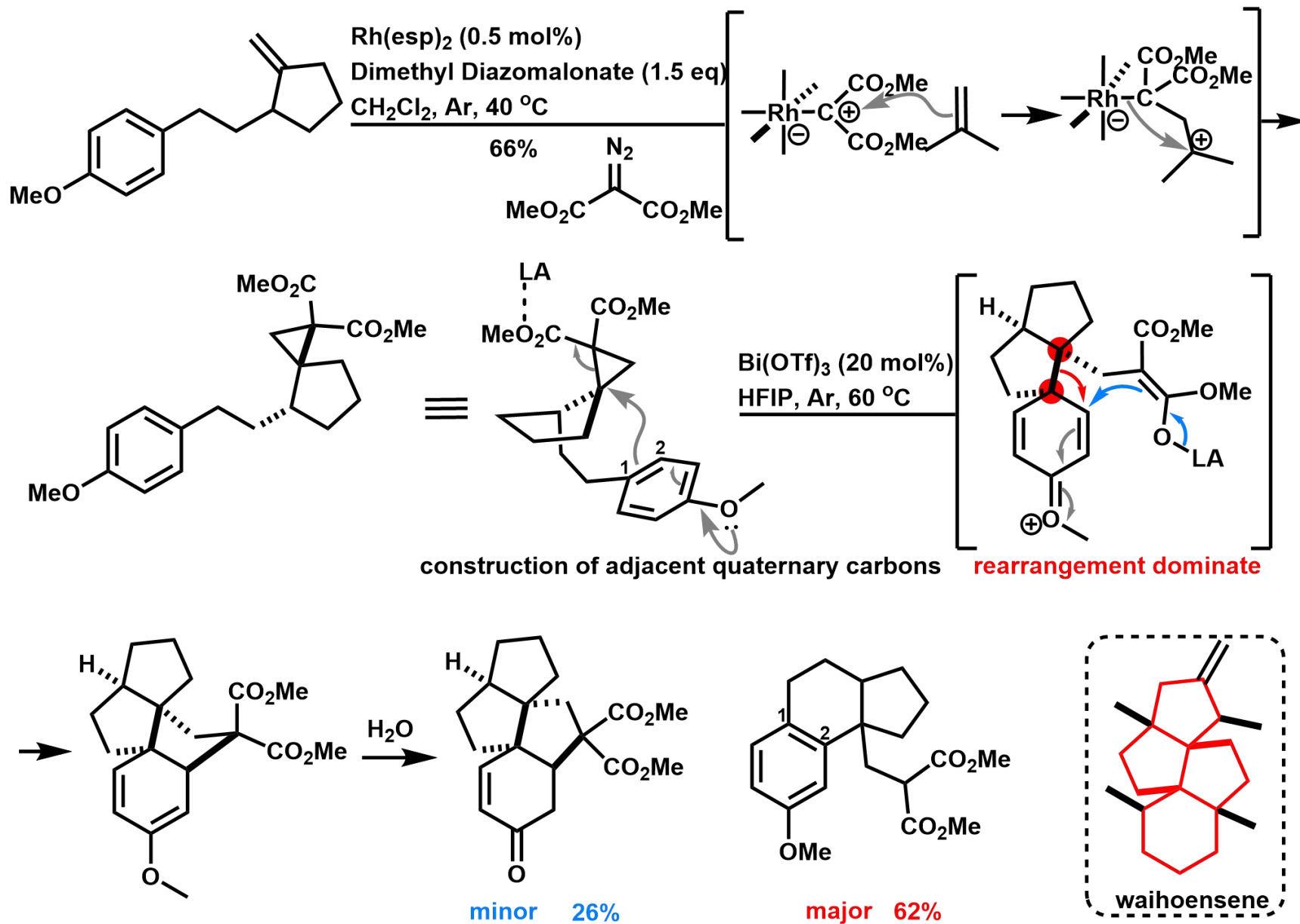
Post-functionalization (2)

Construction of the 5-5-5-tricyclic carbocycles via ring contraction



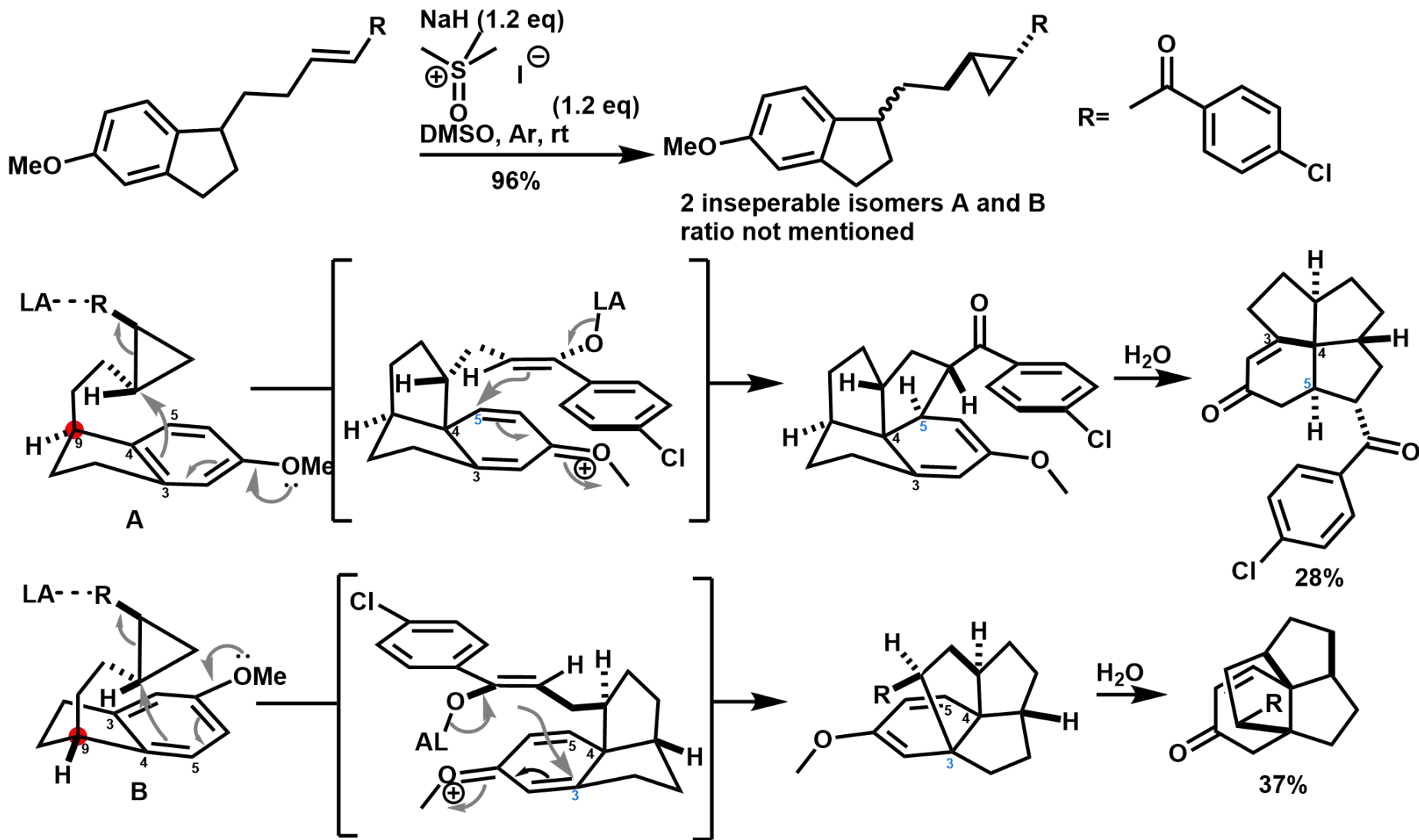
Post-functionalization (3)

Construction of the 5-5-5-6-tetracyclic carbocycles via tetra-substituted cyclopropane



Post-functionalization (4)

Construction of the 5-5-5-6-tetracyclic carbocycles

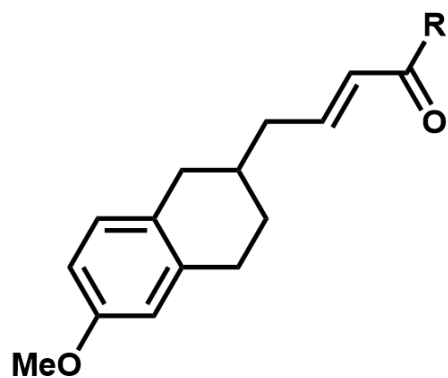


Condition: Bi(OTf)₃ (20 mol%), HFIP, Ar, 60 °C, 2 h

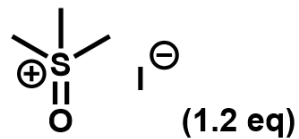
Skeleton types are substrate dependent, chiral center at C9 direct different regio-selectivity in the second cyclization.

Post-functionalization (5)

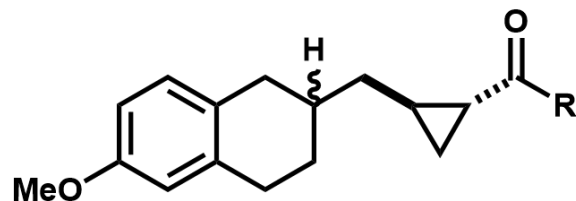
Construction of the 5-5-6-6-tetracyclic carbocycles.



NaH (1.2 eq)

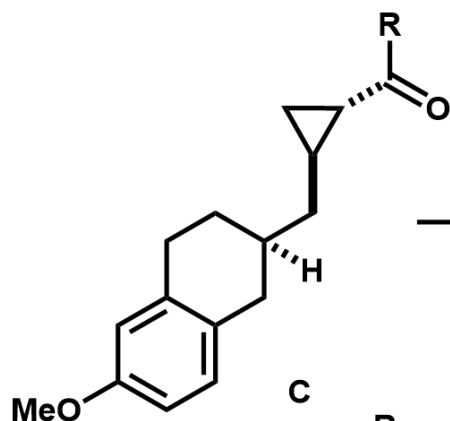


93%

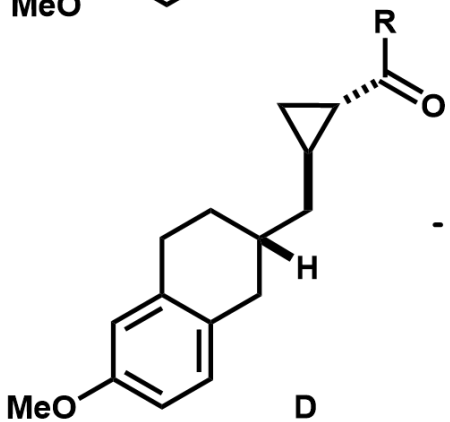


2 inseparable isomers C and D
ratio not mentioned

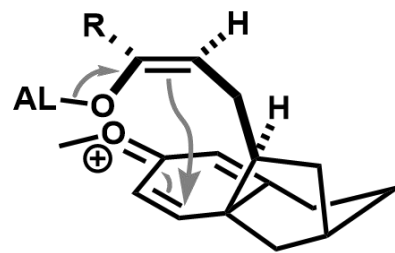
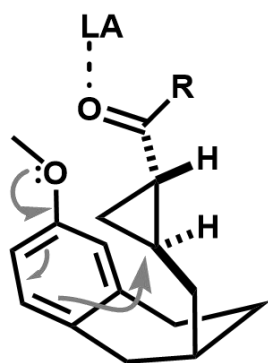
Bi(OTf)₃ (20 mol%)
HFIP, Ar, 60 °C



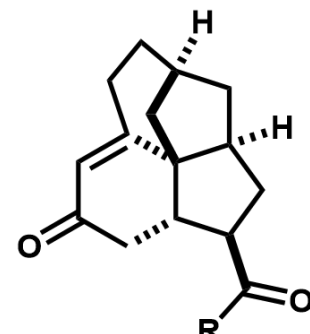
C



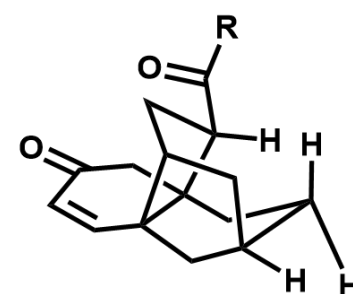
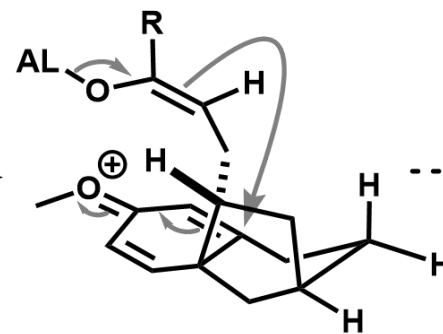
D



H₂O

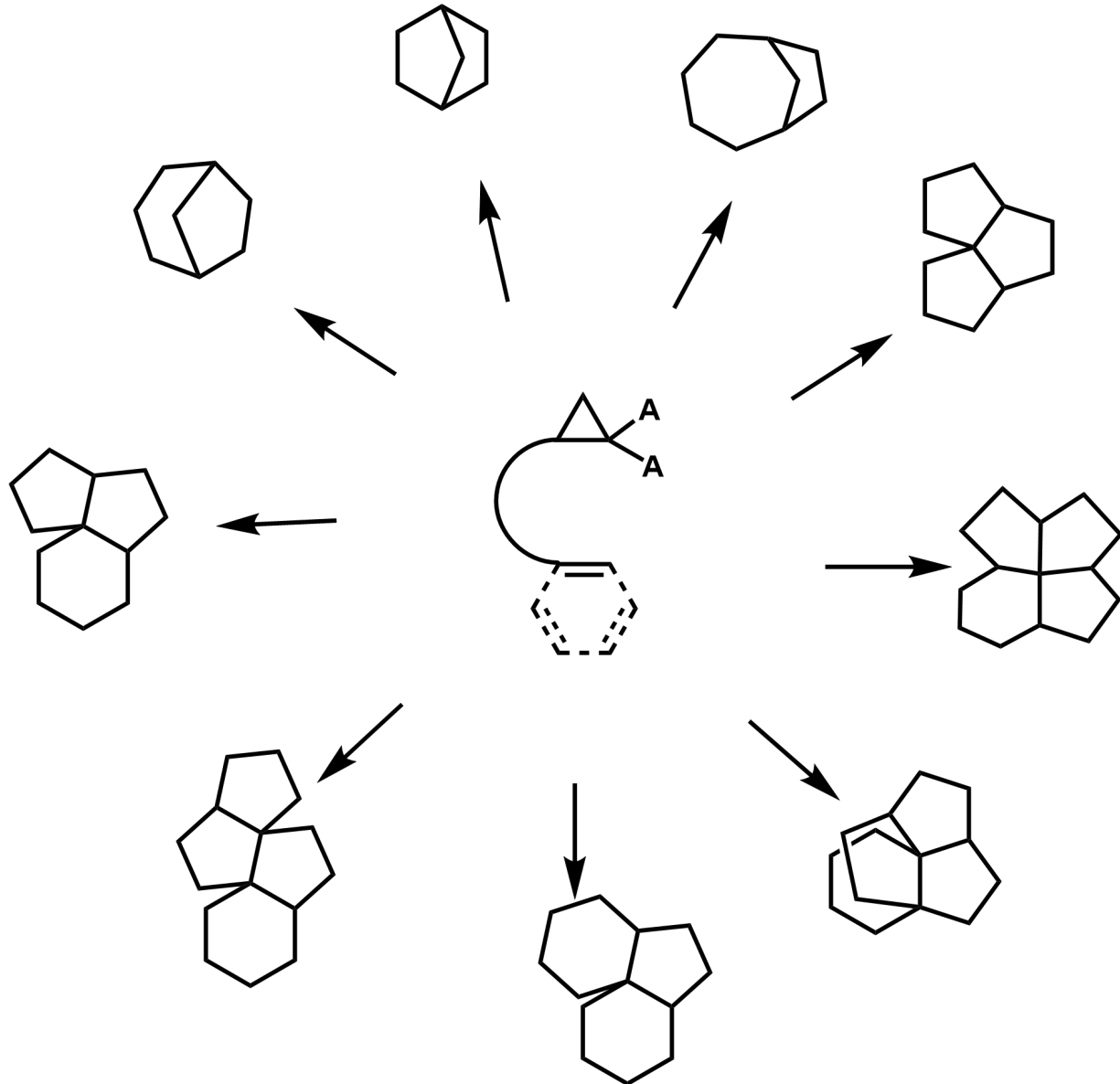


R=Phenyl: 47%
R=*para*-Cl-Phenyl: 53%

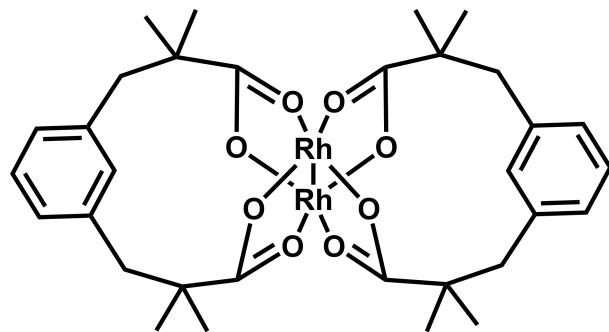


NOT obtain

Summary



Thank you!



$\text{Rh}(\text{esp})_2$

