

Photoactivation of Boron-Caged Prodrug via Phenyl Radical under Hypoxia

**2024.05.02 Literature Seminar
B6 Mizuki Sawada**

Contents

1. Introduction

Boron-caged prodrugs with anticancer activity

2. Main Paper



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Article

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Mechanism: Iridium(III) Anticancer Complex as an Example**

Moyi Liu, Yunli Luo, Junyu Yan, Xiaolin Xiong, Xiwen Xing,* Jong Seung Kim,* and Taotao Zou*

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Moyi Liu, Yunli Luo, Junyu Yan, Xiaolin Xiong, Xiwen Xing,* Jong Seung Kim,* and Taotao Zou*

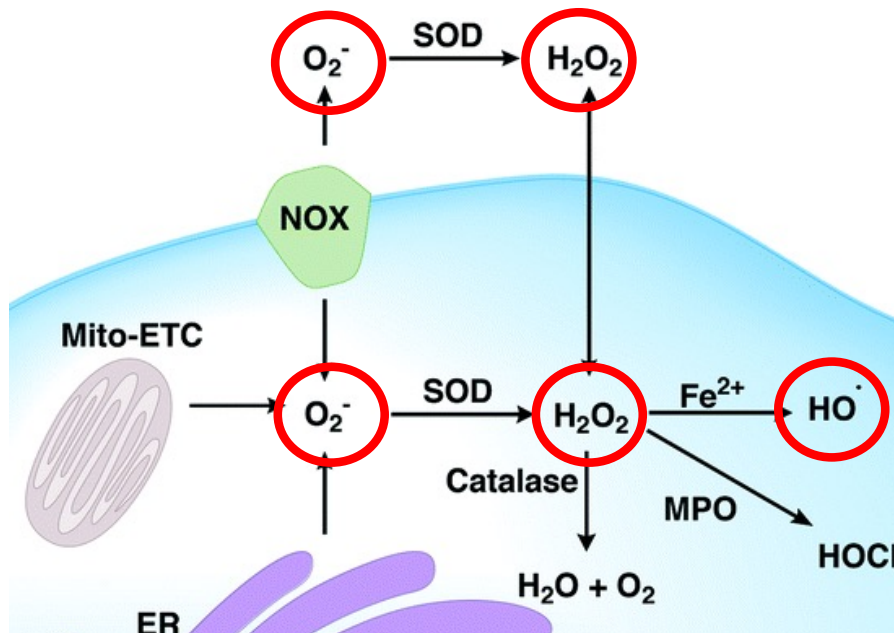
Cancer

Cancer = serious disease

About 10 million people were died of cancer in 2020 all over the world. ¹⁾

The features of cancer

(1) Increased level of ROS ²⁾

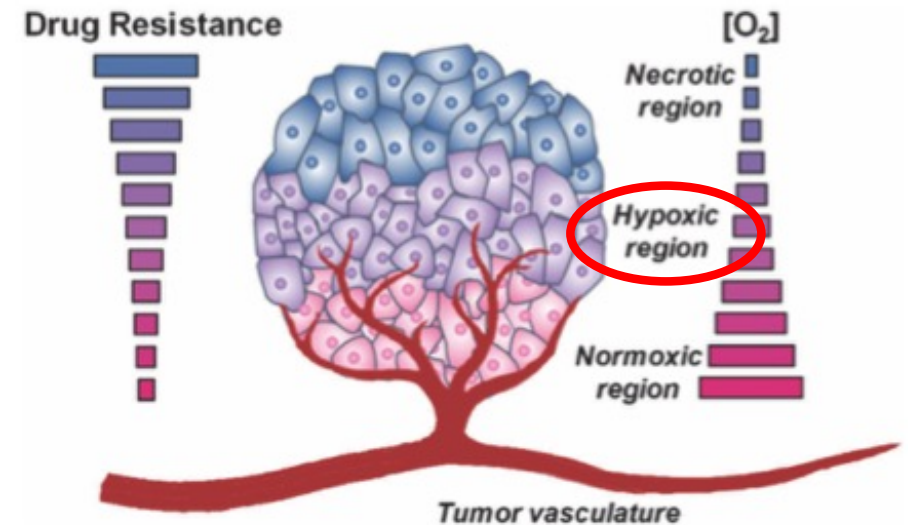


reactive oxygen species (ROS):

O_2^- , H_2O_2 , OH^\bullet , 1O_2

H_2O_2 : $\sim 0.5 \text{ nmol}/10^4 \text{ cells/h}$ in cancer cells ³⁾

(2) Hypoxic (low O_2 level) ⁴⁾



in cancer cell: 0.02–2% O_2

in normal cell: 2–9% O_2

1) https://japan-who.or.jp/factsheets/factsheets_type/cancer/ 2) Cadahía, J. P.; Previtail, V.; Troelsen, N. S.; Clausen, M. H. *MedChemComm* **2019**, *10*, 1531. 3) Szatrowski, T. P.; Nathan, C. F.; *Cancer Res.* **1991**, *51*, 794. 4) Sharma, A.; Arambula, J. F.; Koo, S.; Kumar, R.; Singh, H.; Sessler, J. L.; Kim, J. S. *Chem. Soc. Rev.* **2019**, *48*, 771.

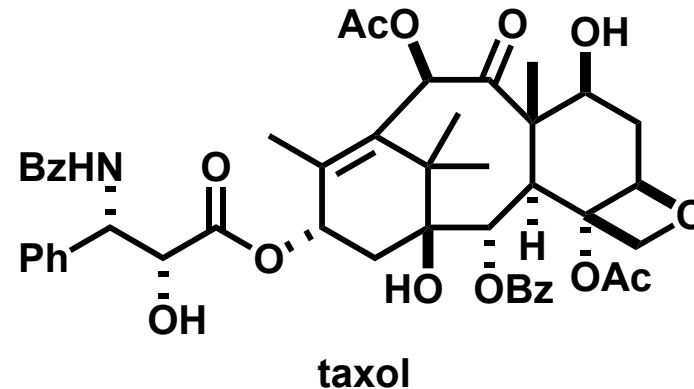
Anticancer Agents and Prodrugs

Chemotherapy with anticancer agents is one of the major treatments of cancer. However, anticancer drugs are sometimes toxic to normal cells and cause side effects



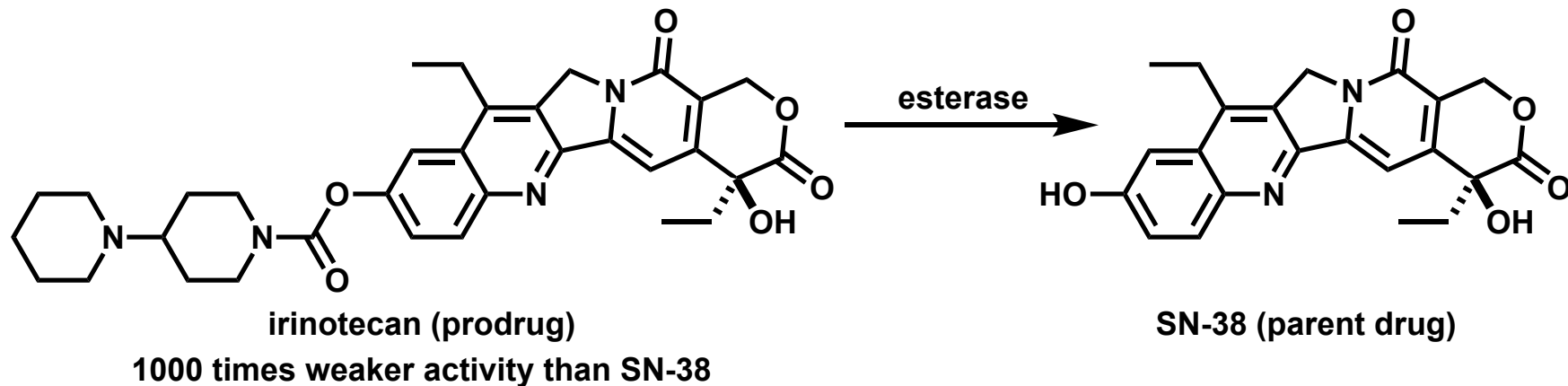
to decrease side effects...

An example of anticancer agent



Prodrug: a pharmacologically inactive derivative converted into an active parent drug in vivo.

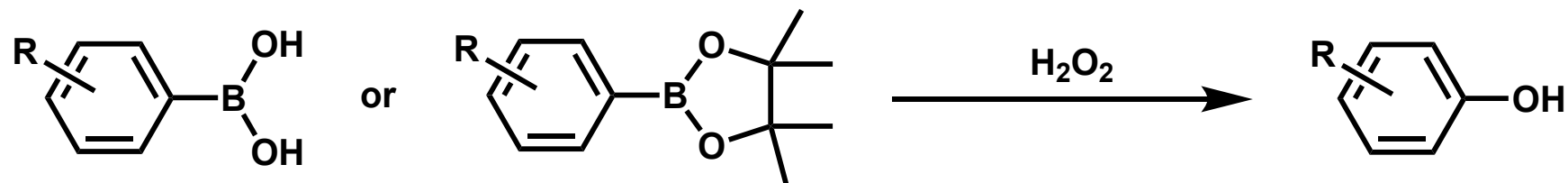
An example of clinically approved anticancer prodrug



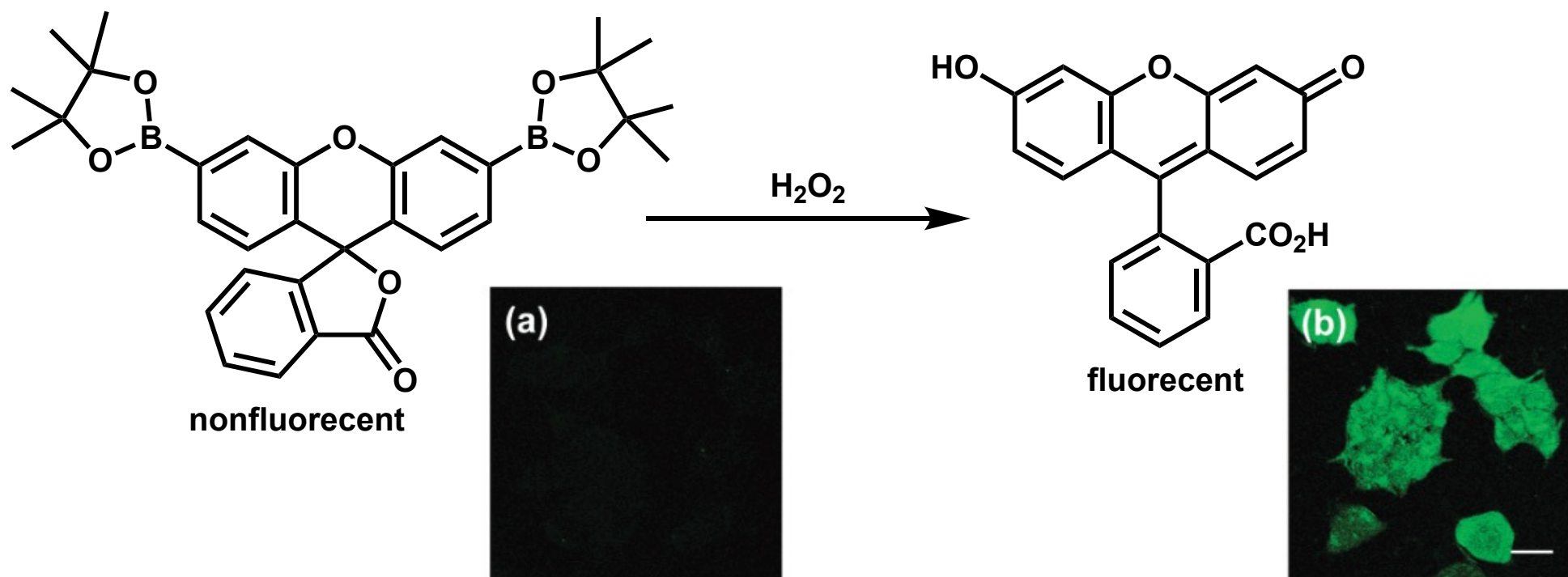
Prodrugs **selectively activated in tumor tissue** are effective to reduce side effects.

Boronic Acids or Esters and H₂O₂

Aryl boronic acids and esters are oxidized to phenol by H₂O₂.

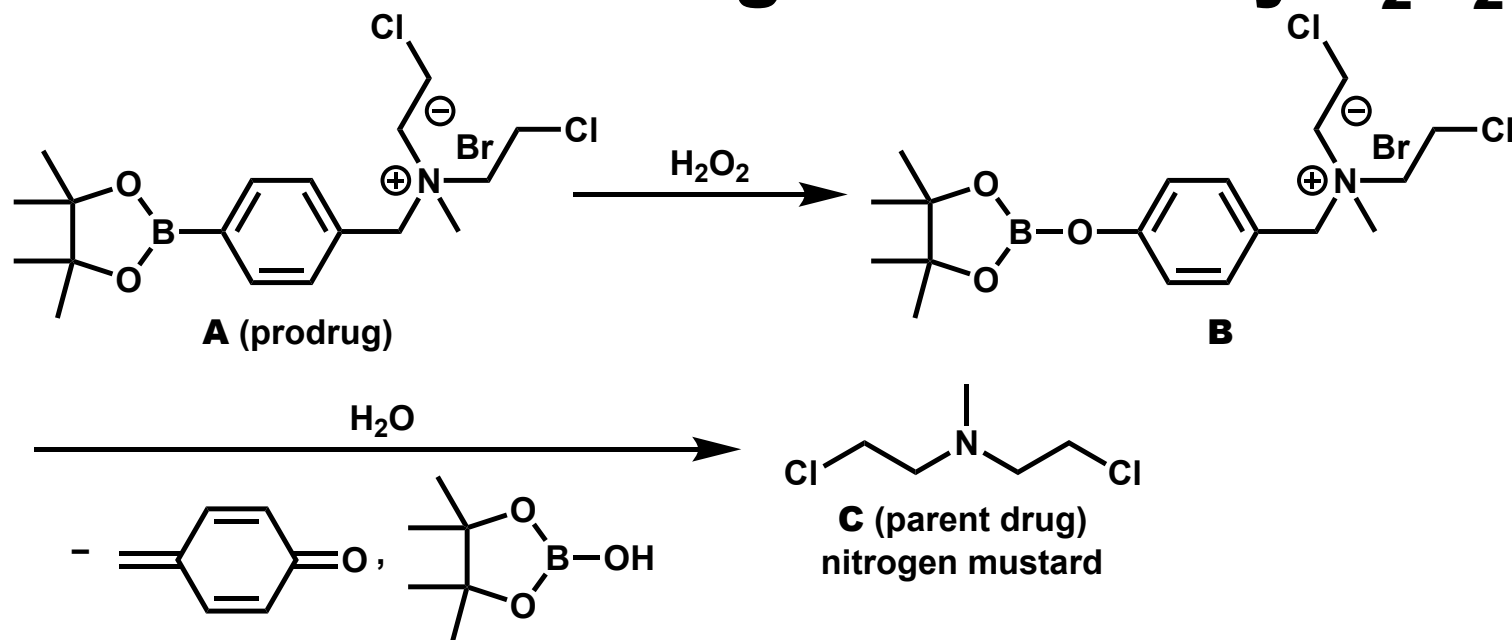


An example of application of this reactivity: H₂O₂-activated fluorescent probe ¹⁾

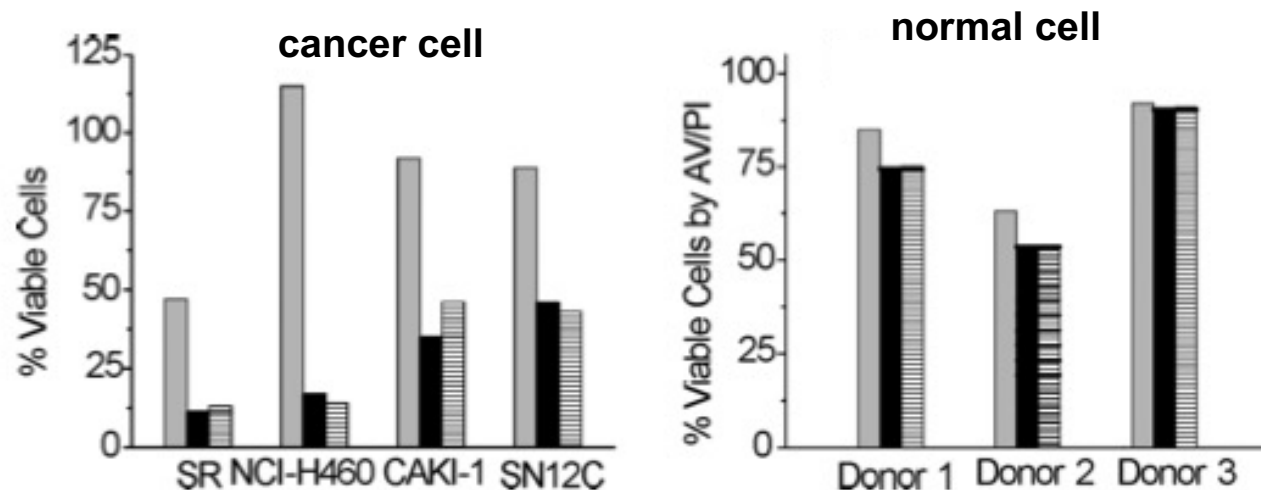


1) Miller, E. W.; Albers, A. E.; Pralle, A.; Isacoff, E. Y.; Chang, C. J. *J. Am. Chem. Soc.* **2005**, *127*, 16652.

First Example of Boronic Ester Anticancer Prodrug Activated by H₂O₂



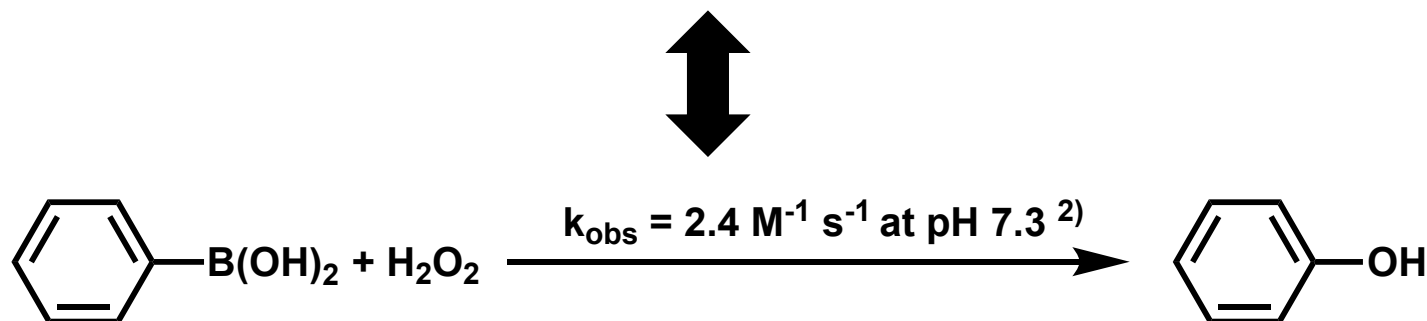
Cytotoxicity (black bar = **A**)



Compound **A** selectively inhibited the growth of cancer cells.

Limitation of Activation of Prodrugs by H₂O₂

[prodrug][H₂O₂] is expected to be less than 10⁻¹⁰ M².¹⁾
 -> **Very fast reaction** is suitable for efficient activation.



The reaction between phenyl boronic acid and H₂O₂ is **relatively slow**.

Spontaneously generated H₂O₂ is not enough to uncage boronic acid/ester prodrugs.

Strategy for overcoming this limitation¹⁾

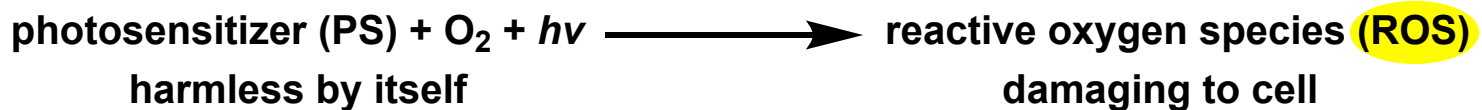
conjugation of ROS amplifiers, directing to lysosomes (higher H₂O₂ level than cytoplasm)...

Development of universal approach to efficiently activating boron-caged prodrug is needed.

1) Daum, S.; Reshetnikov, M. S. V.; Sisa, M.; Dumych, T.; Lootsik, M. D.; Bilyy, R.; Bila, E.; Janko, C.; Alexiou, C.; Herrmann, M.; Sellner, L.; Mokhir, A. *Angew. Chem., Int. Ed.* **2017**, *56*, 15545. 2) Graham, B. J.; Windsor, I. W.; Gold, B.; Raines, R. *Proc. Natl. Acad. Sci. USA.* **2021**, *118*, e2013691118.

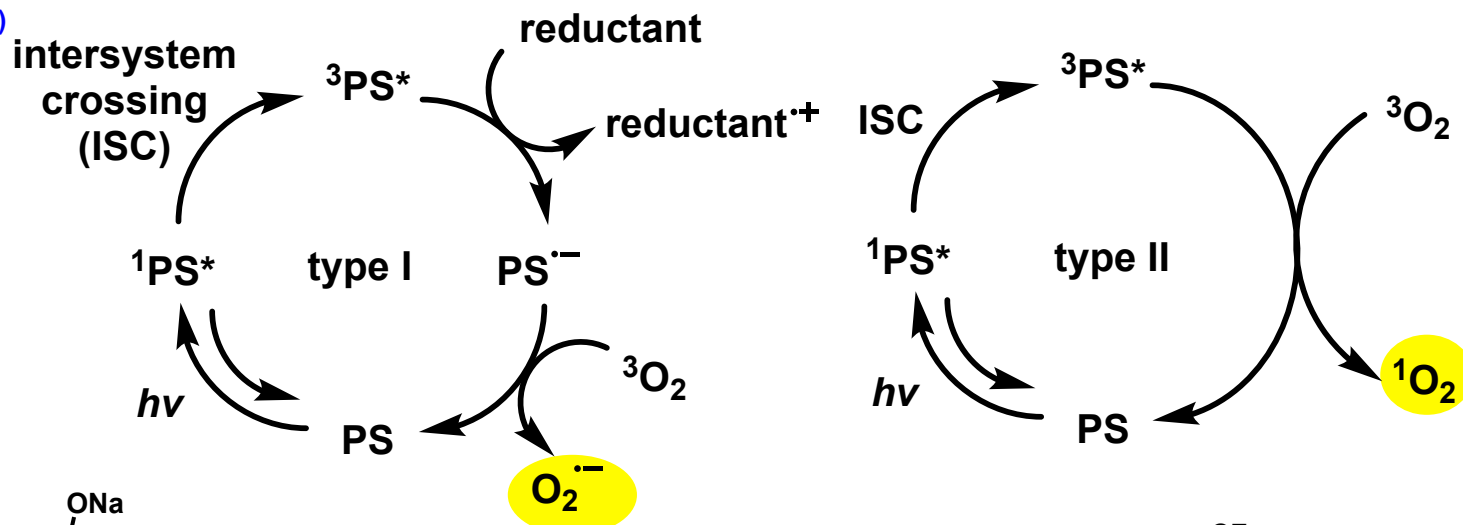
Generation of ROS by Photoactivation

Photodynamic therapy (PDT)

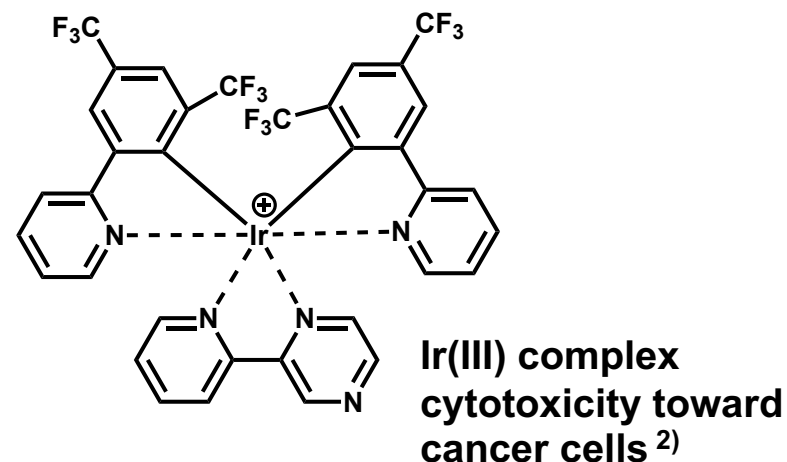
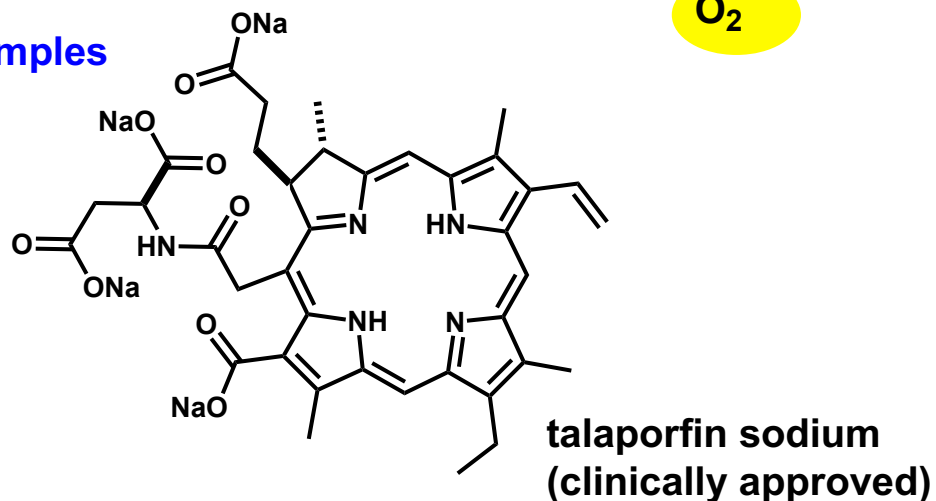


The cytotoxicity can be **spatiotemporally controlled** by photoirradiation only to diseased region.

Mechanism ¹⁾



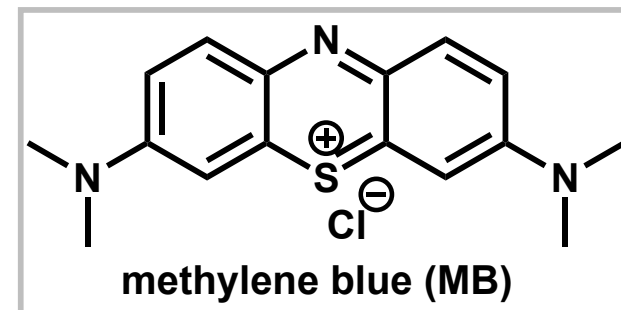
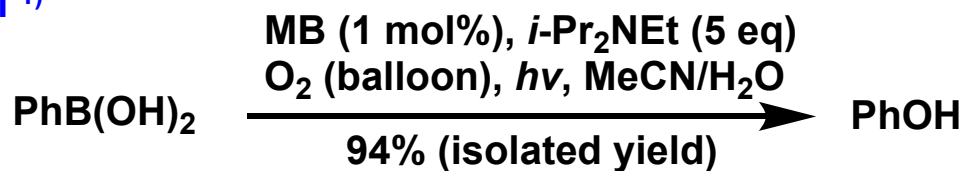
Examples



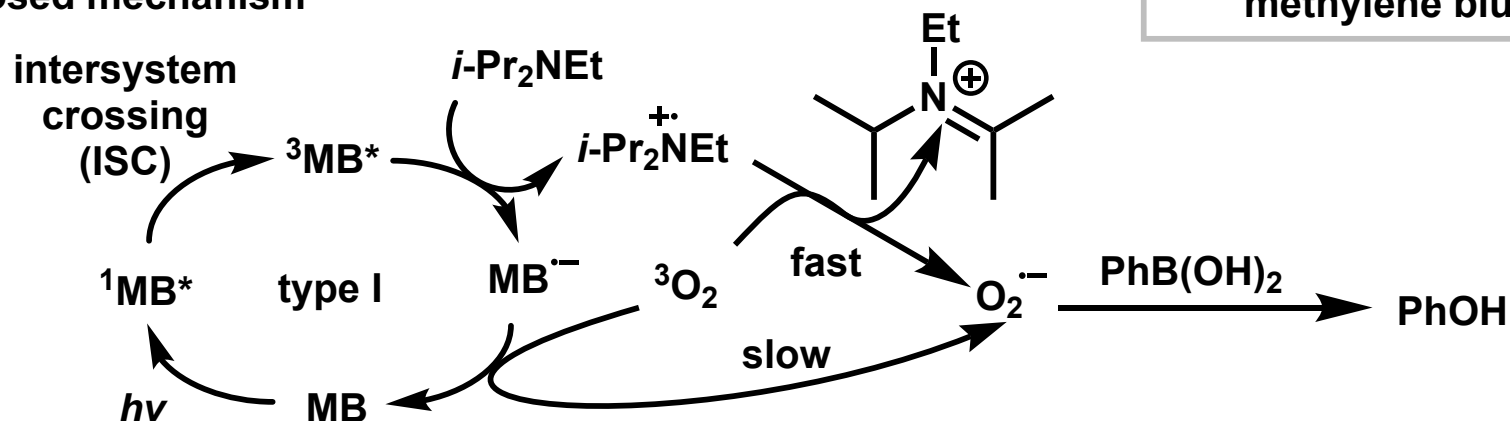
1) Sharman, W. M.; Allen, C. M.; Lier, J. E. *Drug Discov. Today*. **1999**, *4*, 507. 2) Bevernaegie, R.; Doix, B.; Bastien, E.; Diman, A.; Decottignies, A.; Feron, O.; Elias, B. *J. Am. Chem. Soc.* **2019**, *141*, 18486.

Photo-uncaging of Boronic Acids and Esters

example 1 ¹⁾



proposed mechanism



example 2 ²⁾

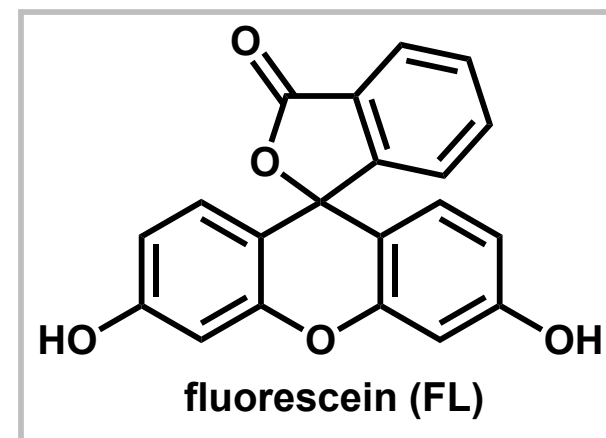
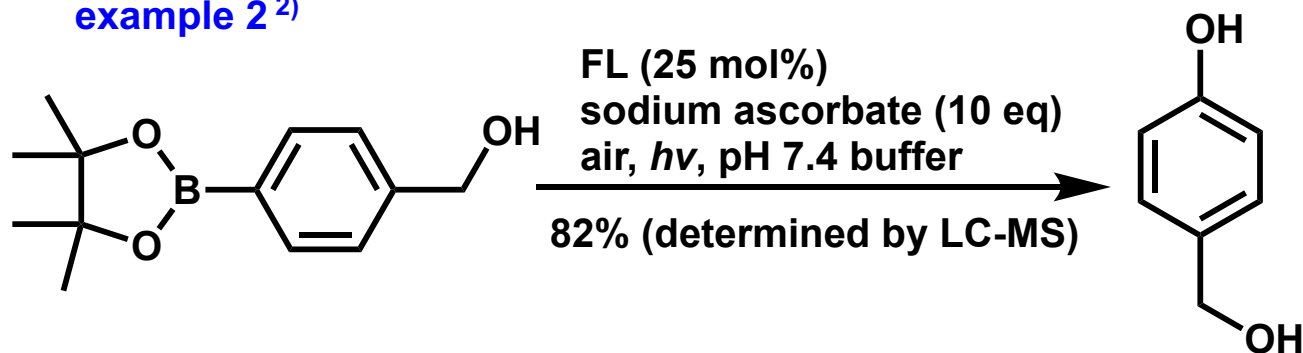


Photo-uncaging of boronic acids and esters **in hypoxic condition** has not been reported.

1) Pitre, S. P.; McTiernan, C. D.; Ismaili, H.; Scaiano, J. C. *J. Am. Chem. Soc.* **2013**, *135*, 13286. 2) Wang, H.; Li, W.-G.; Zeng, K.; Wu, Y.-J.; Zhang, Y.; Xu, T.-L.; Chen, Y. *Angew. Chem. Int. Ed.* **2019**, *58*, 561.

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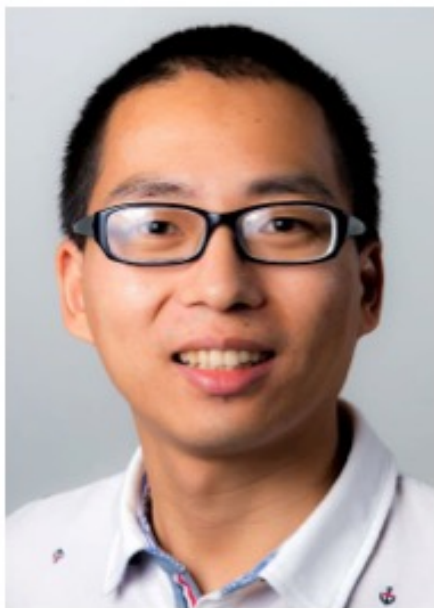
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Photoactivation of Boronic Acid Prodrugs via a Phenyl Radical Mechanism: Iridium(III) Anticancer Complex as an Example

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Prof. Taotao Zou



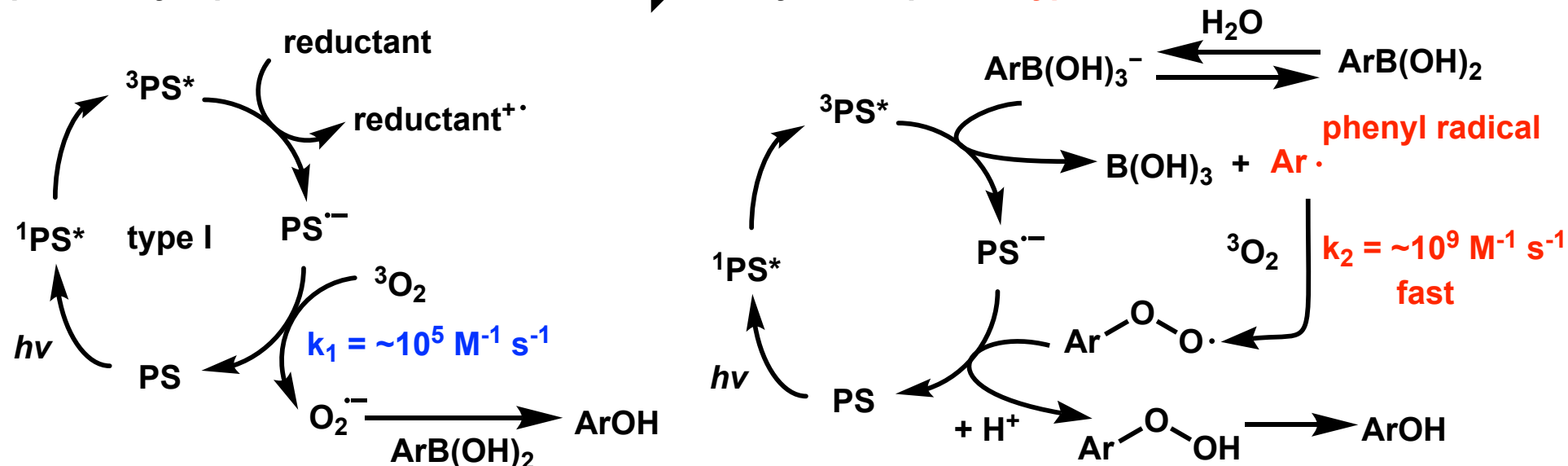
Career

- 2010** Bachelor @ Wuhan University (Prof. Chuluo Yang)
2015 Ph.D. @ The University of Hong Kong (Prof. Chi-Ming Che)
2015.04-2015.10 Postdoctoral fellow
@ University of Warwick (Prof. Peter J. Sadler)
2015.03-2017.04 Postdoctoral fellow
@ The University of Hong Kong (Prof. Chi-Ming Che)
2017.04-2018.02 Research Associate @ The Scripps Research Institute
(Prof. Xiang-Lei Yang and Prof. Paul Schimmel)
2018.03-2018.07 Assistant Professor @ The Chinese University of Hong Kong
2018.07- Professor @ Sun Yat-Sen University

Research topic: inorganic chemical biology, medicinal inorganic chemistry

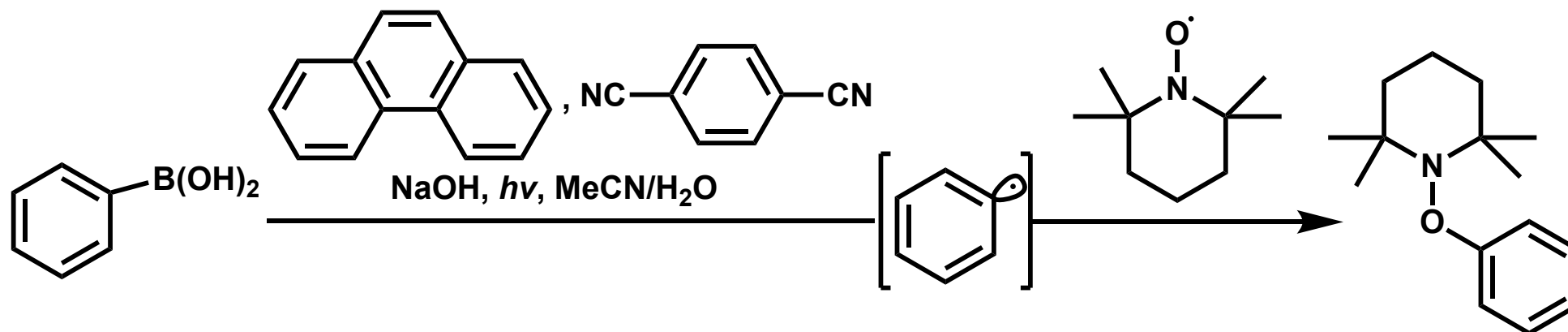
Concept

previously reported in **normoxia** \longrightarrow newly developed in **hypoxia**



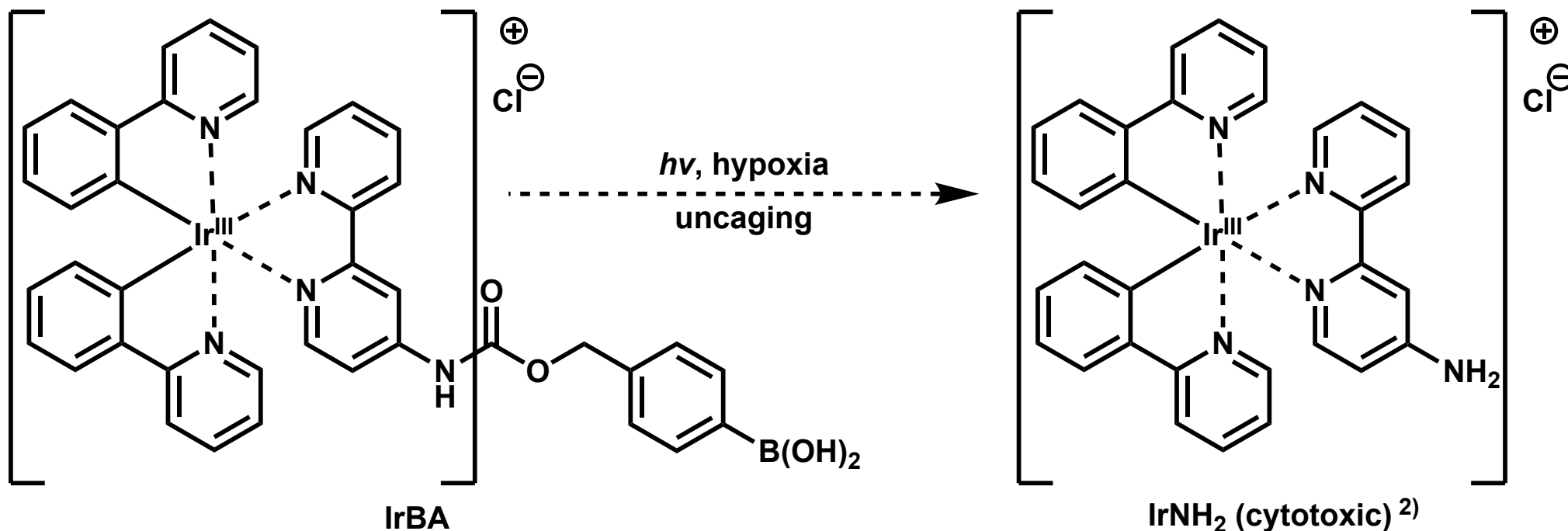
The use of highly reactive **phenyl radical** was planned to efficient uncage under **hypoxia**.

Boronate anion was photocatalytically converted into phenyl radical. ¹⁾



1) Iwata, Y.; Tanaka, Y. Kubosaki, S. Morita, T.; Yoshimi, Y. *Chem. Commun.* **2018**, 54, 1257.

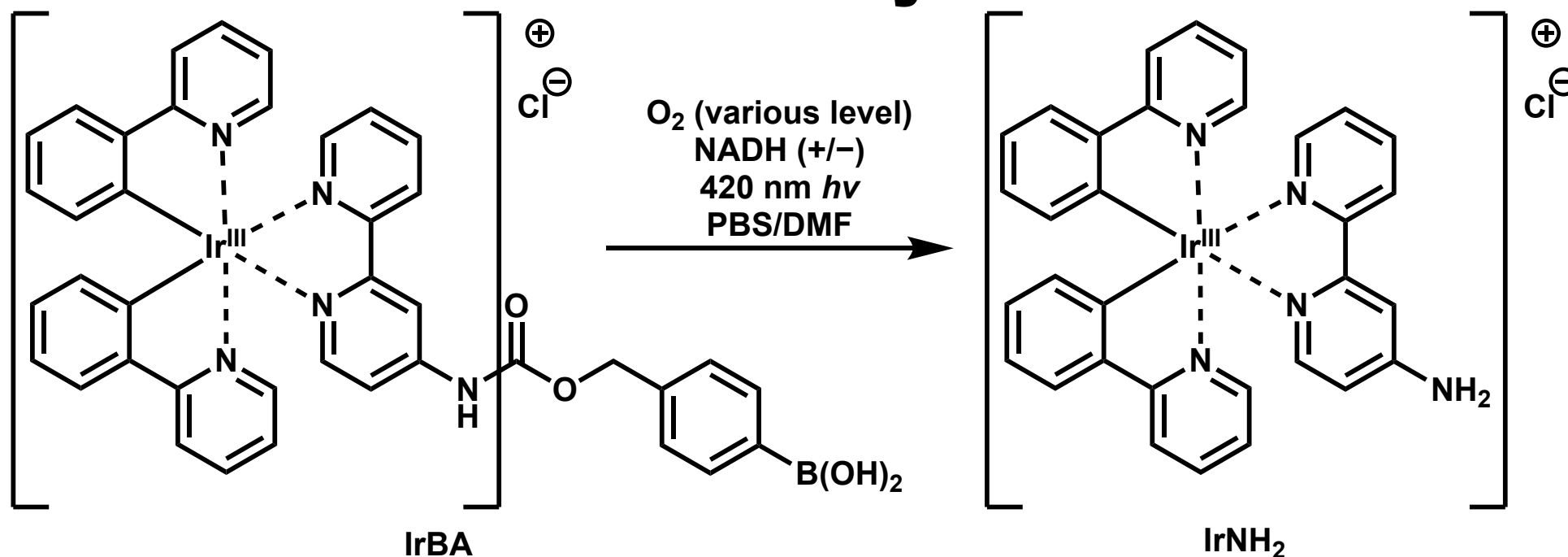
IrBA as Proof of Concept



Ir(III) complex has
 photocatalytic activity¹⁾ = work as a photosensitizer
 cytotoxicity²⁾ = work as an anticancer drug

1) Bevernaegie, R.; Doix, B.; Bastein, E.; Diman, A.; Decottignies, A.; Feron, O.; Elias, B. *J. Am. Chem. Soc.* **2019**, *141*, 18486. 2) Kuang, S.; Liao, X.; Zhang, X.; Rees, T. W.; Guan, R.; Xiong, K.; Chen, Y.; Ji, L.; Chao, H. *Angew. Chem. Int. Ed.* **2019**, *59*, 3315.

Photoreactivity of IrBA



entry	O ₂ level	NADH	time required for full conversion of IrBA to IrNH ₂
1	21%	+	< 10 min
2	21%	-	15 min
3	1%	-	15 min
4	0.1%	-	30 min
5	0.02%	-	30 min
6	0% (degassed)	-	no reaction (reaction time is not reported)

IrBA was successfully converted to IrNH₂ without external reductant and under hypoxic condition.

Analysis of Reaction Intermediate (1): Boronate Anion

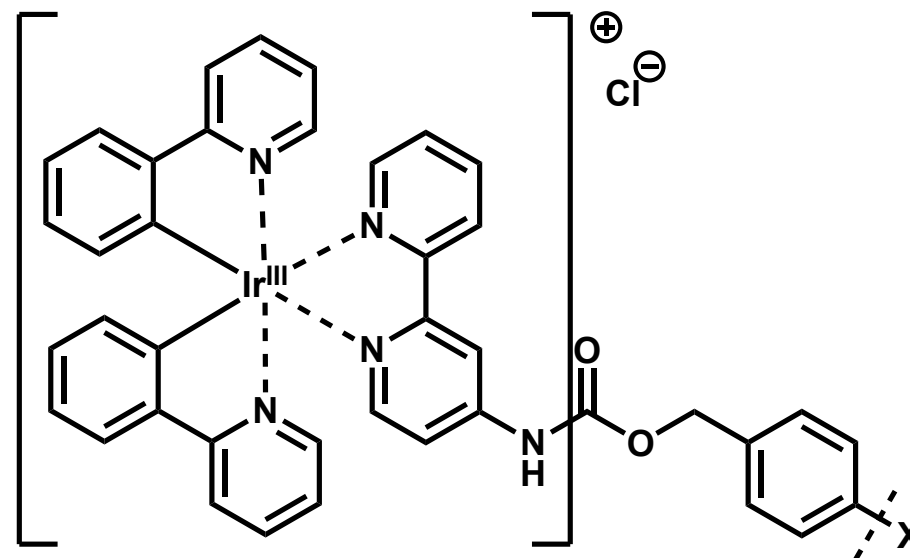
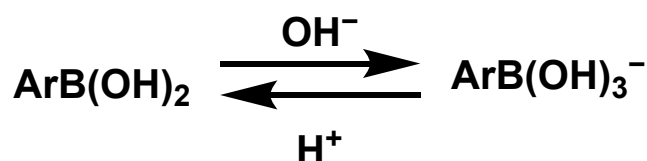
Redox potential (vs Ag/AgCl)

$$[p\text{-MeOCH}_2\text{PhB(OH)}_2]^{*+}/[p\text{-MeOCH}_2\text{PhB(OH)}_2] = + 2.23 \text{ V}$$

$$[p\text{-MeOCH}_2\text{PhB(OH)}_3]^{*+}/[p\text{-MeOCH}_2\text{PhB(OH)}_3]^- = + 0.85 \text{ V}$$

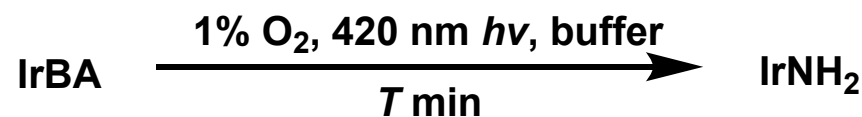
$$[\text{Ir}^{\text{III}}806]^{*+}/[\text{Ir}^{\text{II}}806] = + 1.02 \text{ V}$$

Only boronate anion can be oxidized by $[\text{Ir}^{\text{III}}806]^{*+}$.



X = H: Ir806, X = B(OH)₂: IrBA

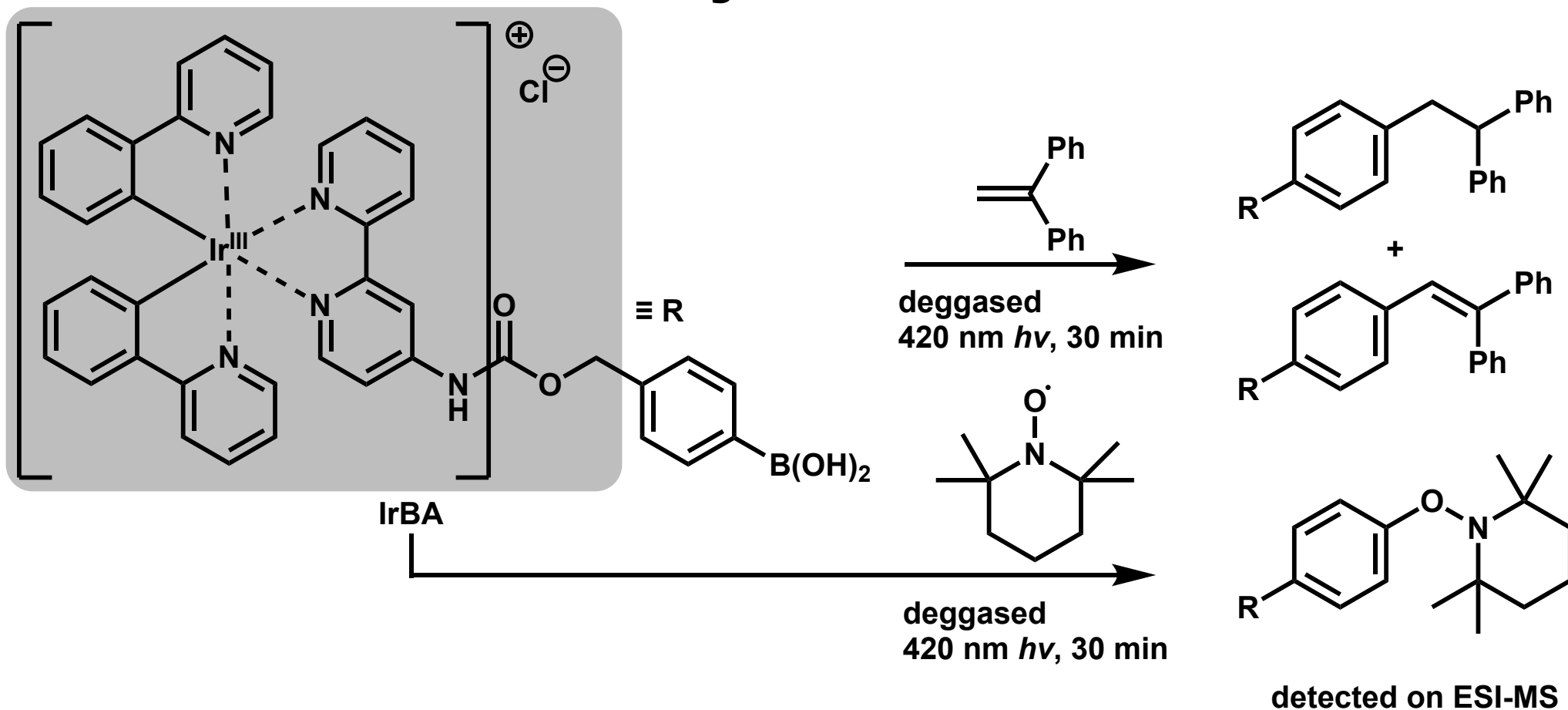
Reaction rate under various pH conditions



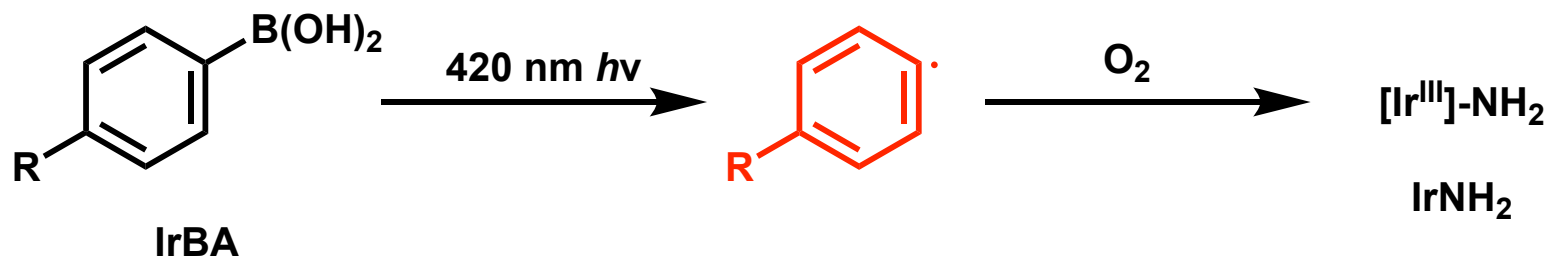
entry	pH	T	yield
1	4.5	30	31%
2	6.0	30	60%
3	8.0	< 10	> 99%

Boronate anion seems to be involved in photoreaction.

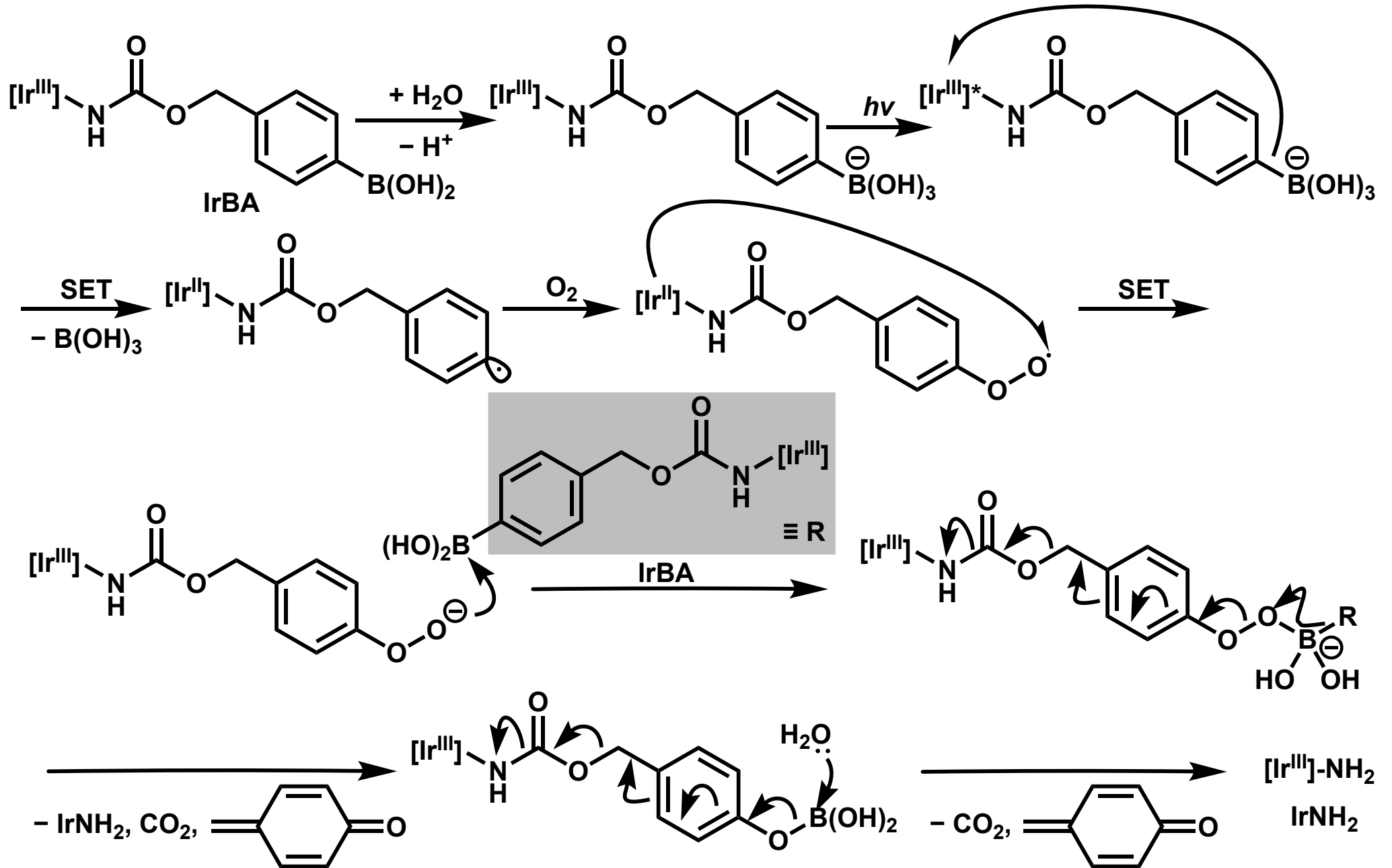
Analysis of Reaction Intermediate (2): Phenyl Radical



These results indicate the generation of **phenyl radical intermediate** by photoactivation of IrBA.

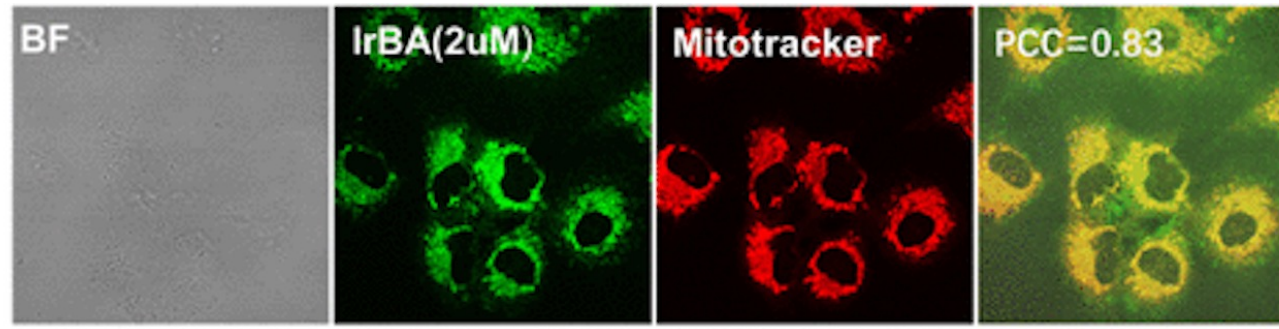


Proposed Reaction Mechanism



Formation of Phenyl Radical in Cell

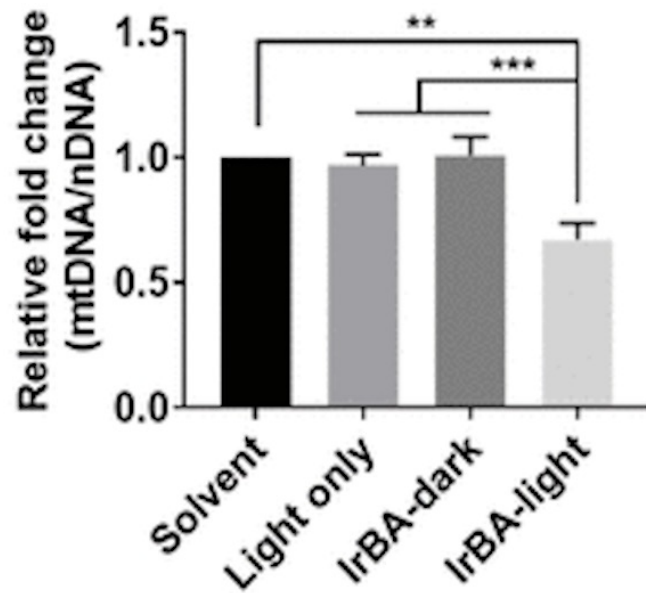
IrBA was accumulated in mitochondria in A549 cell (human non-small-cell lung cancer cell).



Phenyl radical is known to damage DNA.

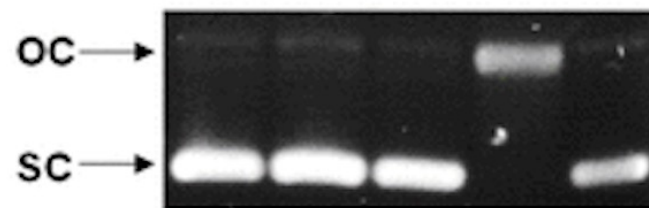
mtDNA damage by IrBA + $h\nu$ in live cell

O_2 : < 0.1%, $h\nu$: 420 nm

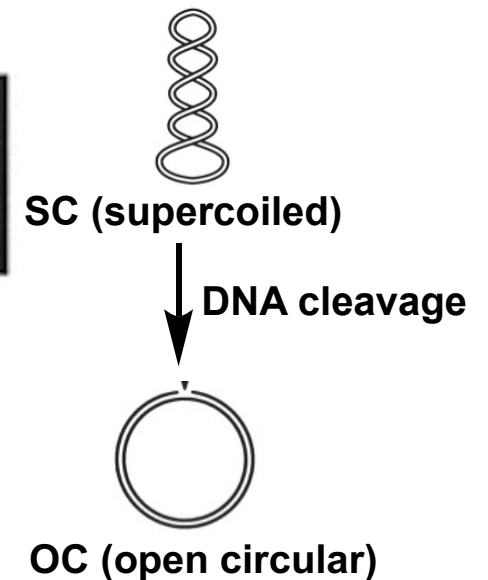


Inhibition of DNA damaging activity by TEMPO

O_2 : < 0.1%, $h\nu$: 420 nm

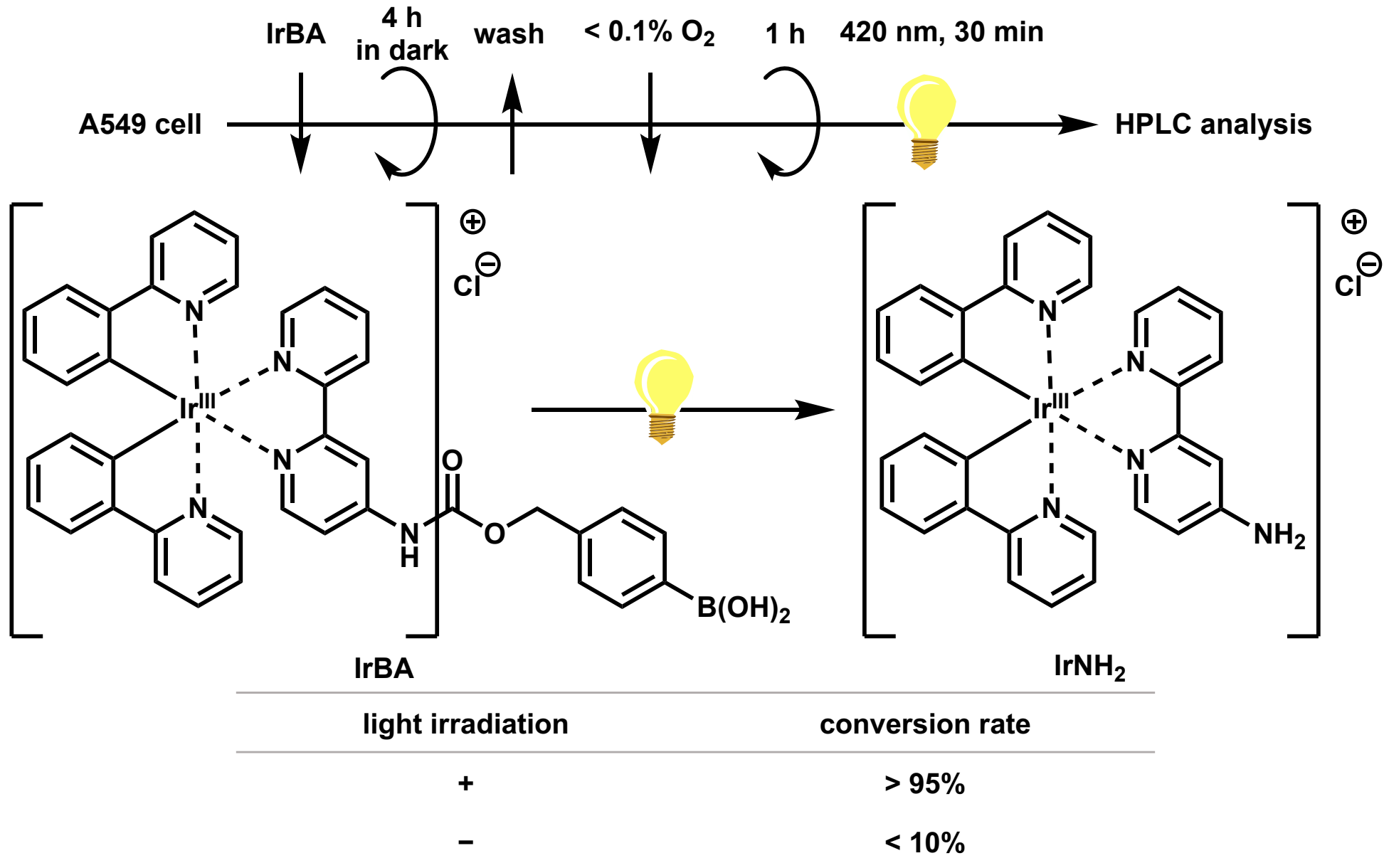


	1	2	3	4	5
DNA	+	+	+	+	+
light	-	+	-	+	+
IrBA	-	-	+	+	+
TEMPO	-	-	-	-	+



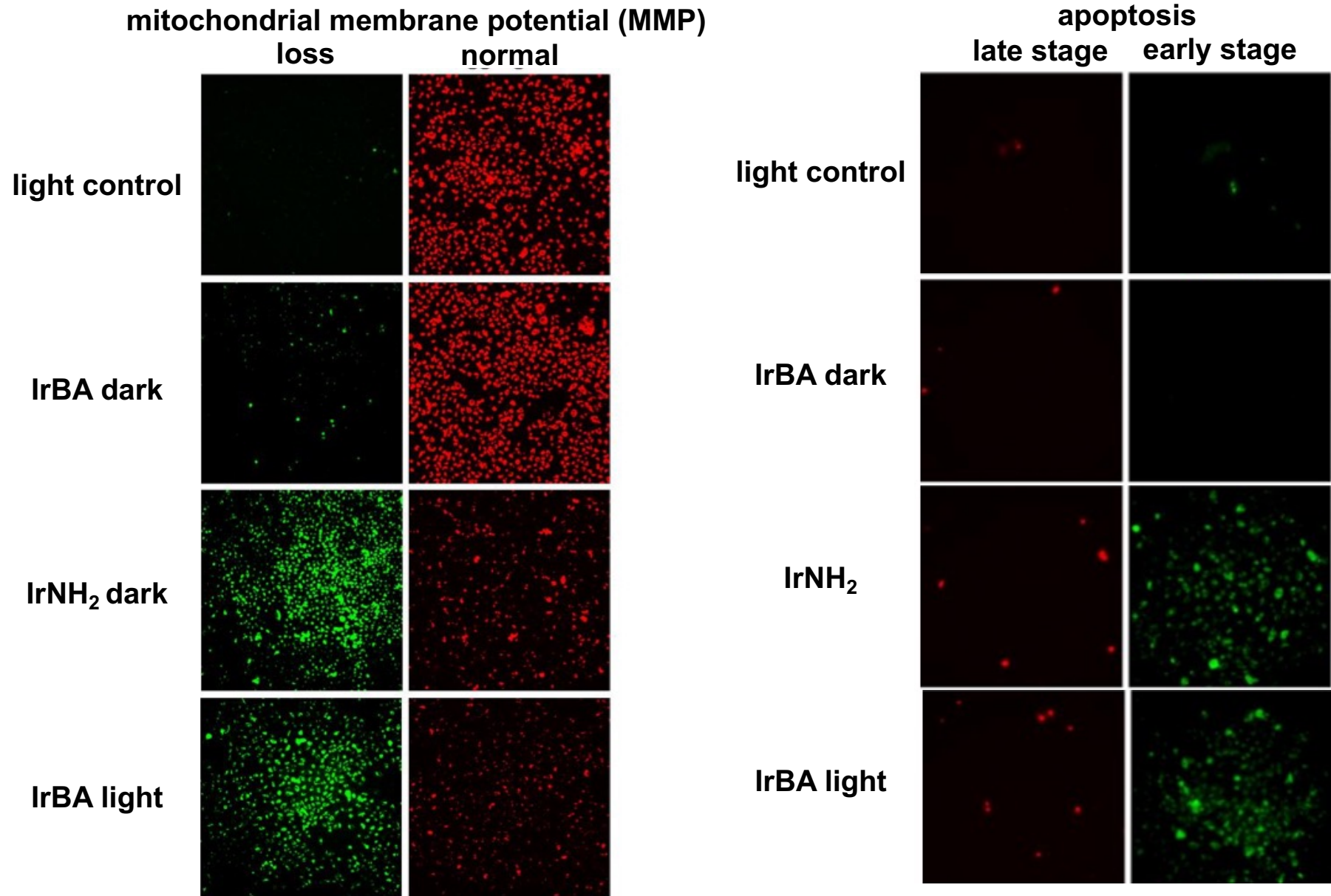
These results suggest that phenyl radical is generated under cellular environment.

Release of IrNH₂ in Cell under Hypoxia (1)



IrBA was successfully converted into IrNH₂ by light irradiation in cancer cell under hypoxia.

Release of IrNH₂ in Cell under Hypoxia (2)



IrBA was photochemically activated to induce MMP loss and apoptosis under hypoxia.

Cytotoxicity against Cancer Cells

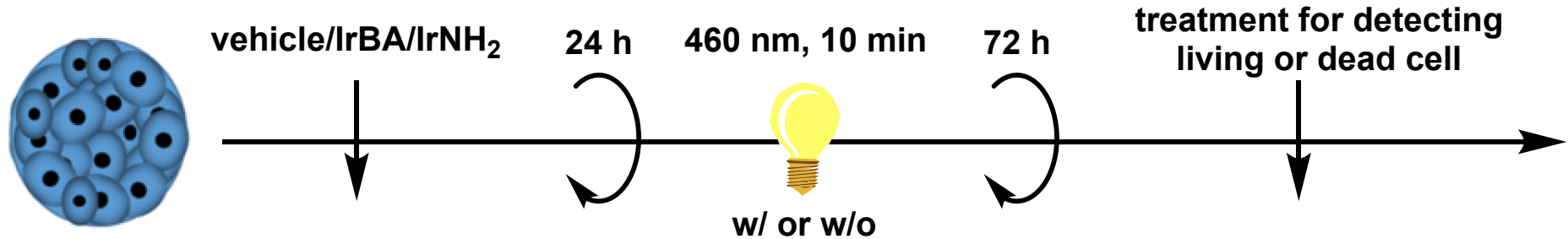
cell	compound	IC ₅₀ (μM) under hypoxia (O ₂ < 0.1%)		
		dark	light	PI
A549	IrBA	> 150	4.4 ± 0.3	> 34
	IrNH ₂	9.6 ± 0.6	2 ± 0.4	4.5
MCF-7	IrBA	> 200	6.0 ± 1.4	> 33
	IrNH ₂	10.2 ± 1.4	2.7 ± 0.2	4
A375	IrBA	75.6 ± 4.2	2.0 ± 0.1	38
	IrNH ₂	4.6 ± 0.4	0.7 ± 0.2	6.5

photo index (PI) = $IC_{50, \text{dark}}/IC_{50, \text{light}}$

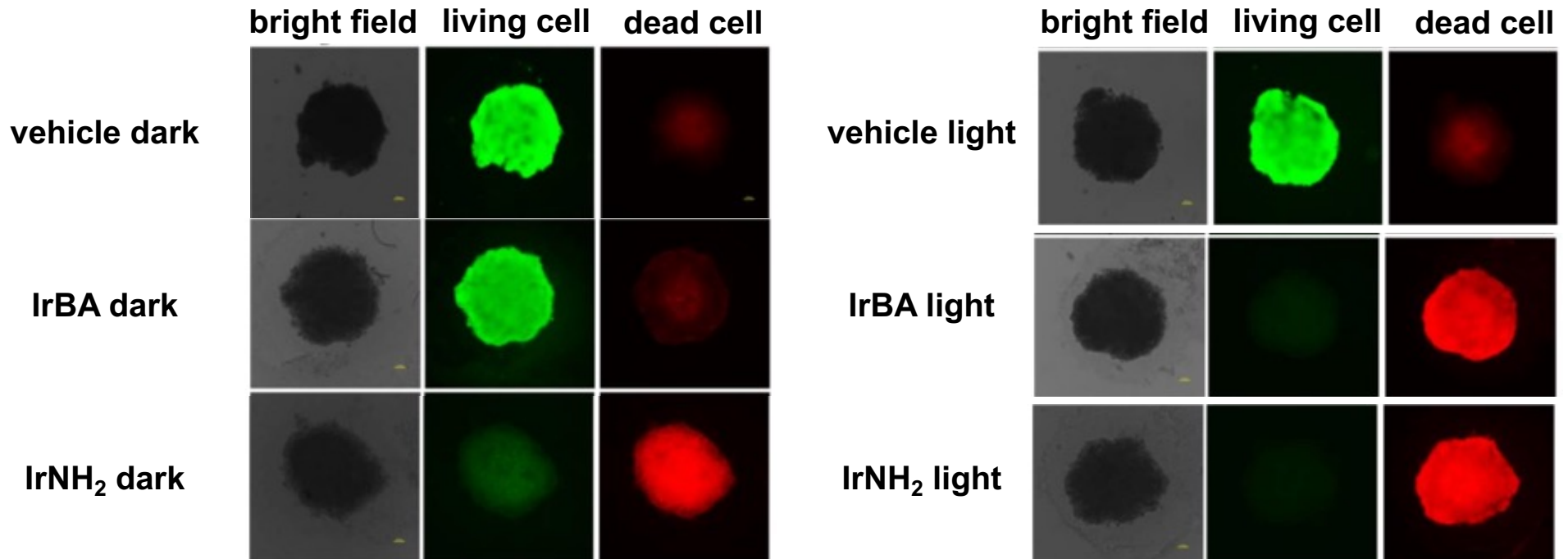
MCF-7: human breast adenocarcinoma cell, A375: human melanoma cell

Photoactivated IrBA showed cytotoxicity compared to IrNH₂ under dark.
 Since PI value of IrBA was more than 30, cytotoxicity of IrBA can be controlled by light irradiation.

Cytotoxicity against 3D Tumor Model

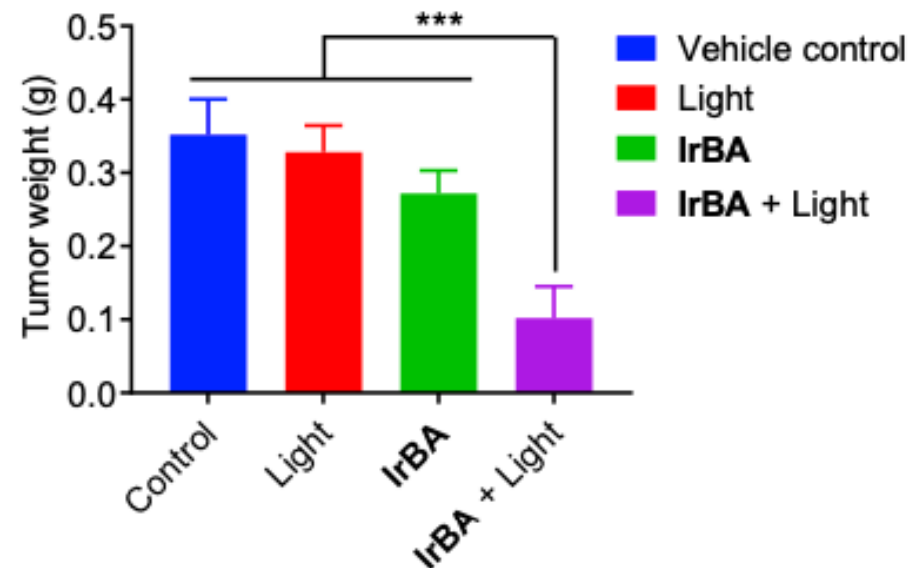
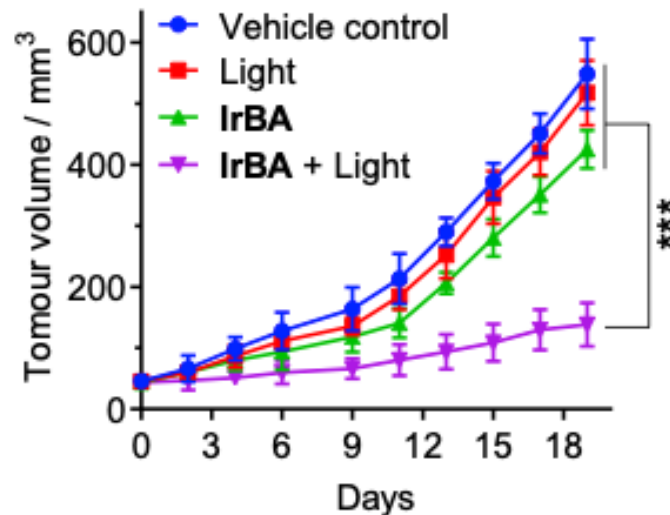
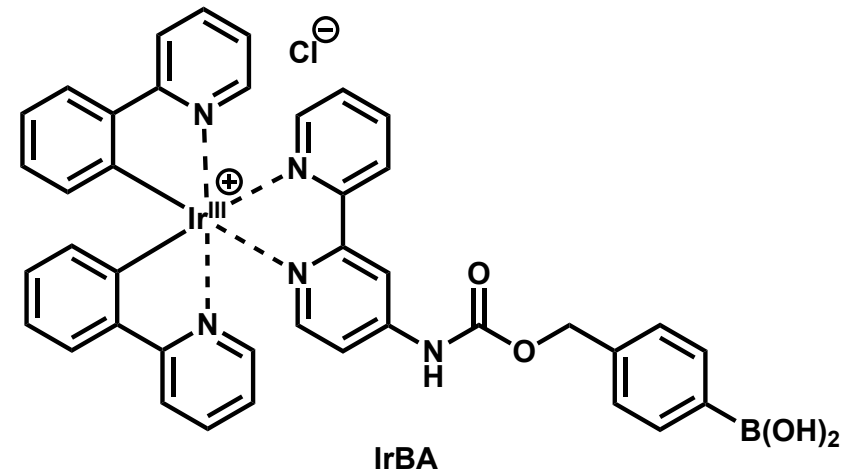
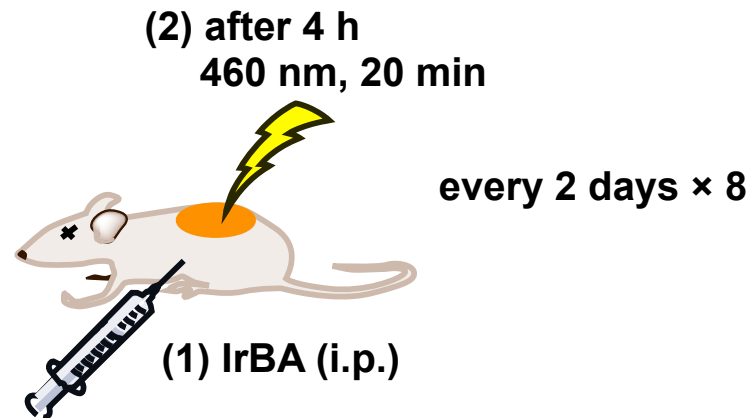


A549 tumor spheroid
3D tumor model with hypoxia region



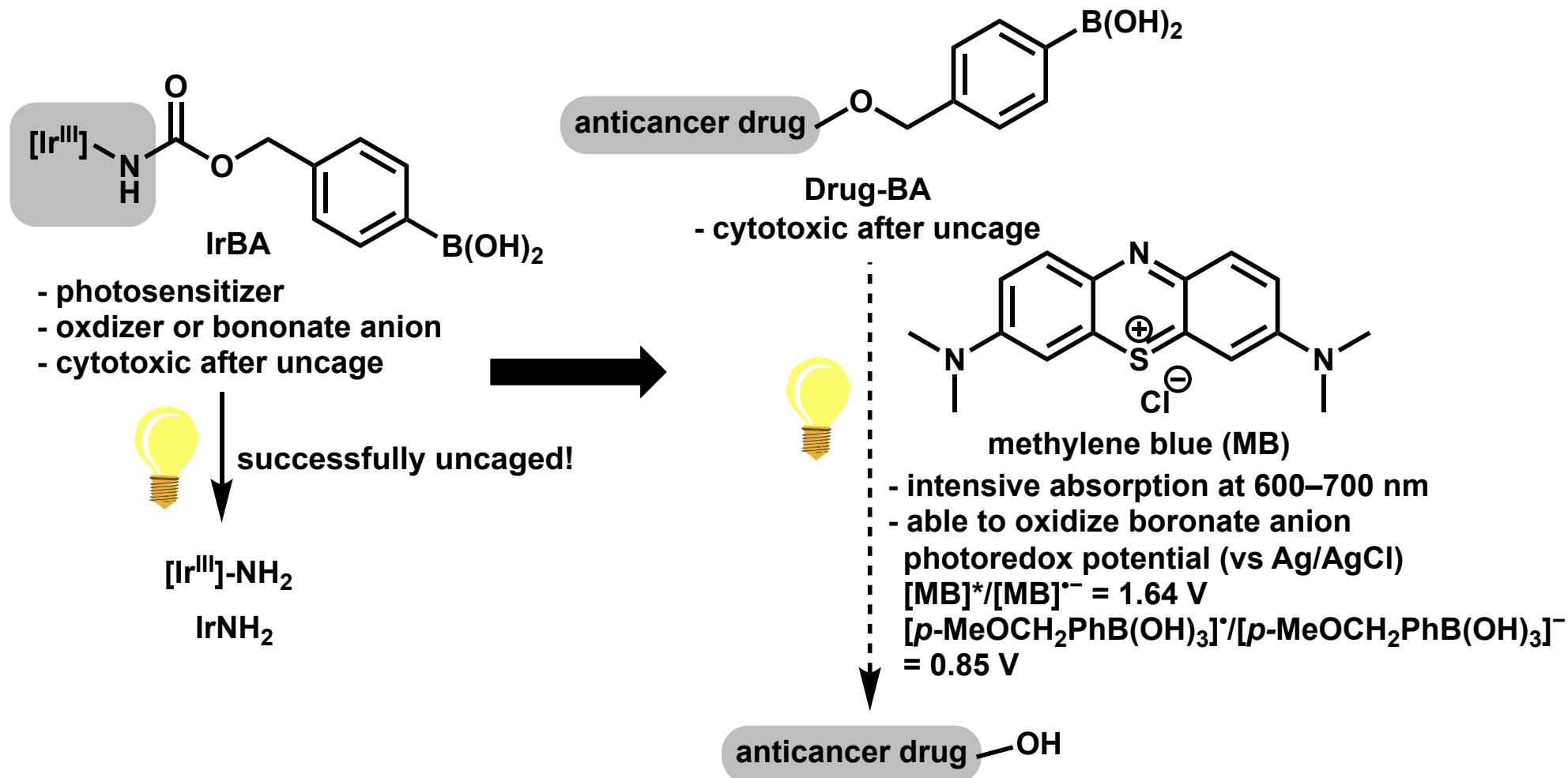
IrBA was activated by photoirradiation to show cytotoxicity against 3D tumor model including hypoxia region.

Antitumor Activity *in vivo*



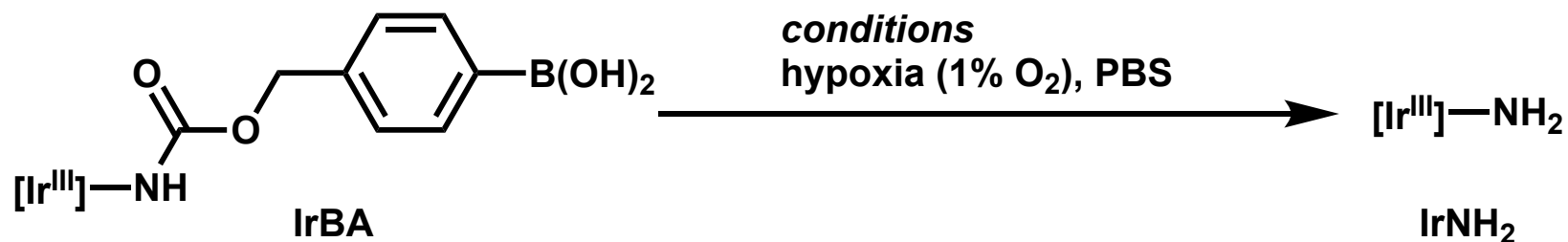
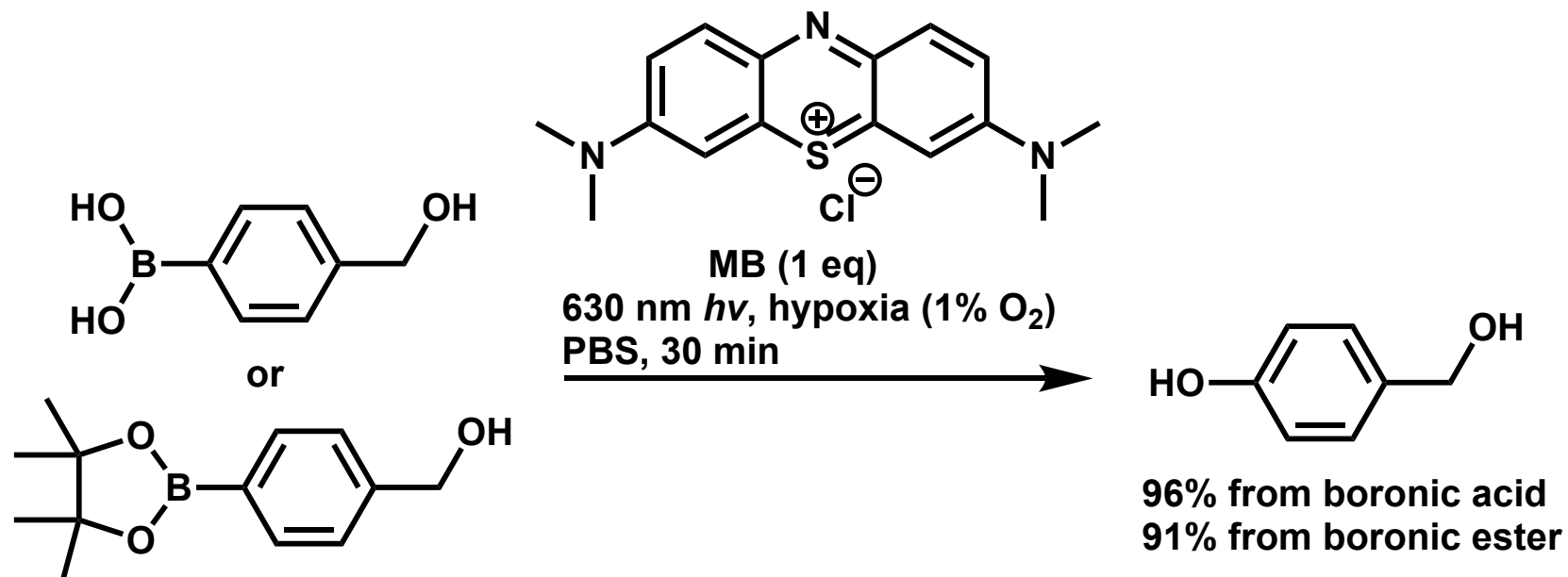
- Combination of IrBA and light irradiation could suppress tumor growth in mice.
- IrBA without light showed low tumor inhibiting activity.
- > Intratumoral ROS was insufficient to activate IrBA.
- No mouse death or mouse body weight less was observed.

Photoactivation by Extramolecular photosensitizer



Extramolecular photosensitizer, MB, was used to show the generality of this strategy; this uncaging strategy can be applied for activation of boron-caged prodrugs without photosensitivity.

Photoactivation by Methylene Blue (1)



conditions	result
MB (1 eq), 630 nm $h\nu$, 30 min	IrBA: full conversion within 30 min, IrNH ₂ : 92%
H ₂ O ₂ (10 eq)	IrBA: full conversion after 3 h

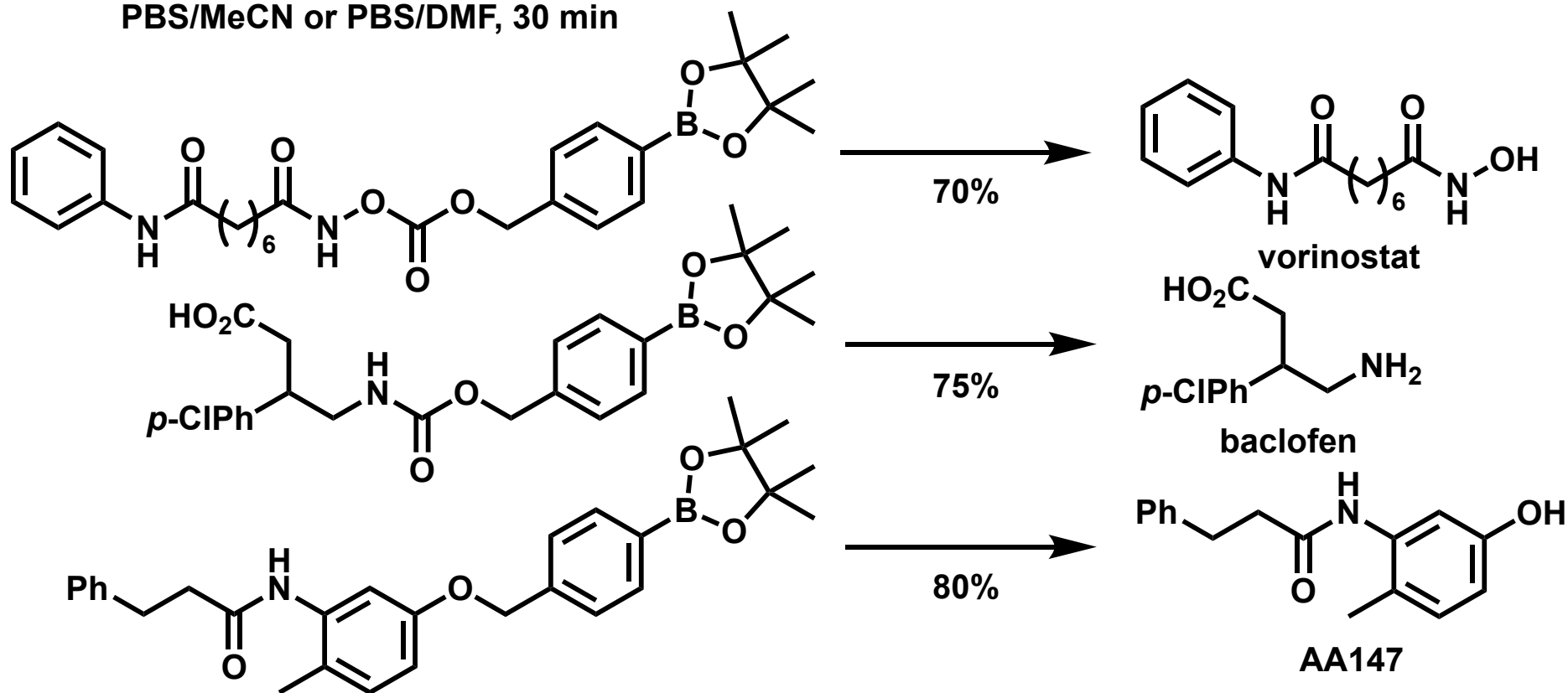
Yield was determined by HPLC analysis.

Boronic acids and esters were successfully uncaged by MB and 630 nm $h\nu$ under hypoxia.

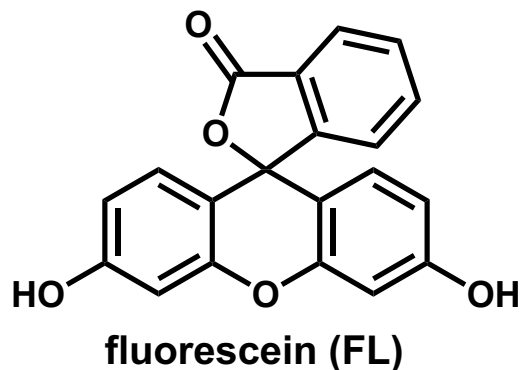
Photoactivation by Methylene Blue (2)

Condition: MB (1 eq), 630 nm $h\nu$, hypoxia (1% O₂)
PBS/MeCN or PBS/DMF, 30 min

Yield was determined by HPLC analysis.



Three prodrugs were uncaged by MB and 630 nm $h\nu$ under hypoxia.



redox potential (vs Ag/AgCl)

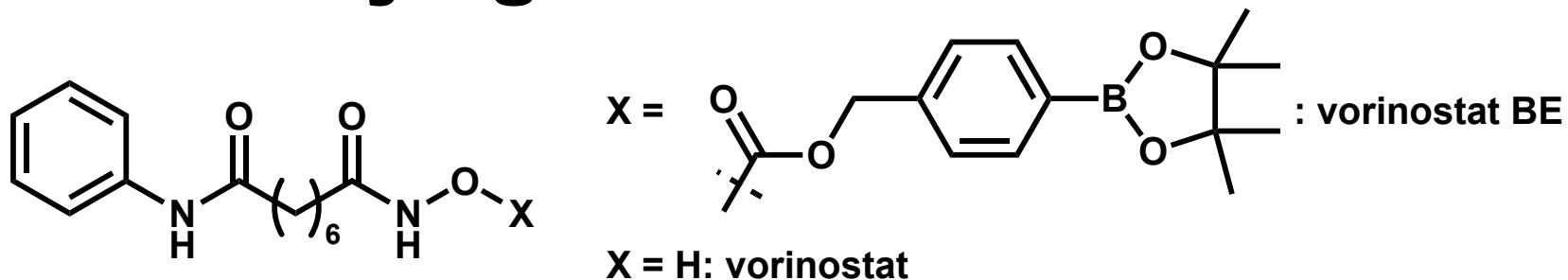
[FL]^{*}/[FL]⁻ = + 0.81 V, [FL]⁺/^{*}[FL] = - 1.03 V

[p-MeOCH₂PhB(OH)₃]^{*}/[p-MeOCH₂PhB(OH)₃]⁻ = + 0.85 V

O₂/[O₂]⁻ = - 0.83 V

“No obvious reaction” was occurred with FL and $h\nu$.
→ ruling out the involvement of O₂⁻

Cytotoxicity against Cancer Cells with MB

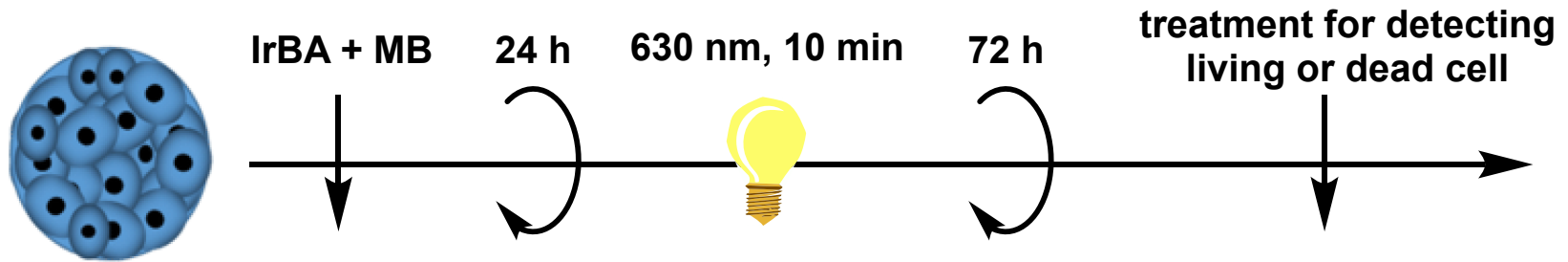


cell	compound	IC ₅₀ (μM) under hypoxia (O ₂ < 0.1%) with MB (2 μM)		
		dark	light	PI
A549	IrBA	> 150	9.3 ± 2.1	> 16
	vorinostat BE	> 200	23.6 ± 2.3	> 8.5
	vorinostat	15.7 ± 2.6	not reported	—
MCF-7	IrBA	> 200	13.6 ± 1.6	> 15
	vorinostat BE	> 200	40.6 ± 1.9	> 4.9
	vorinostat	35.5 ± 2.5	not reported	—
A375	IrBA	70.2 ± 3.4	3.6 ± 0.5	20

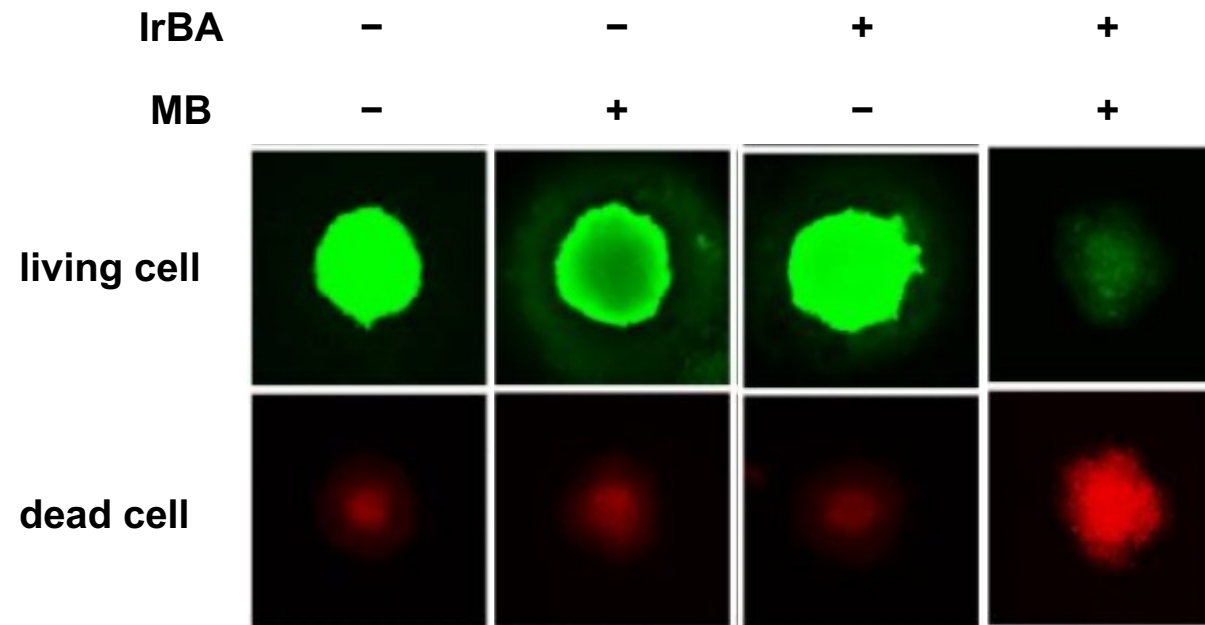
More than 90% cell survived when treated with MB (2 μM) and 630 nm light irradiation.

IrBA and vorinostat BE activated by the combination of MB and 630 nm *hν* showed cytotoxicity against cancer cells under hypoxia.

Cytotoxicity against 3D Tumor Model with MB

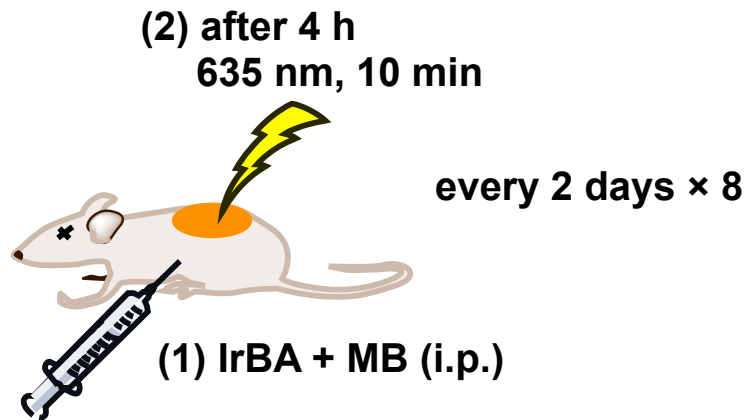


A549 tumor spheroid
3D tumor model with hypoxia resion

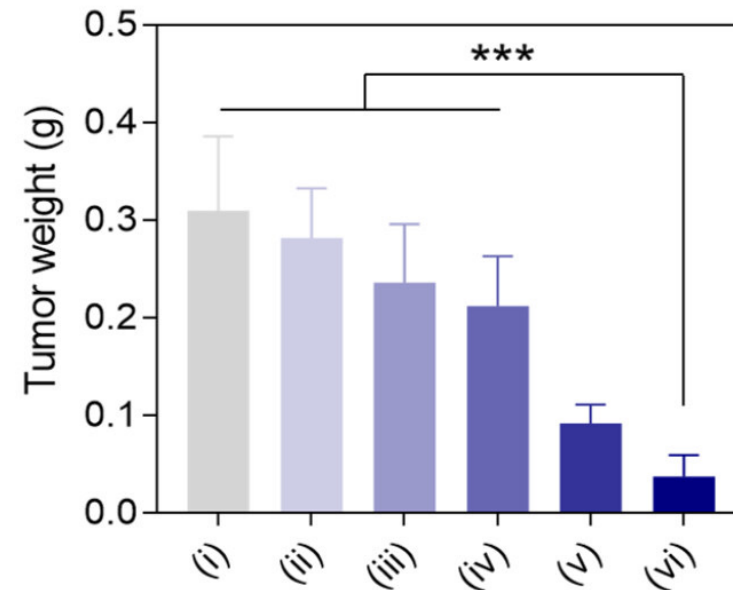
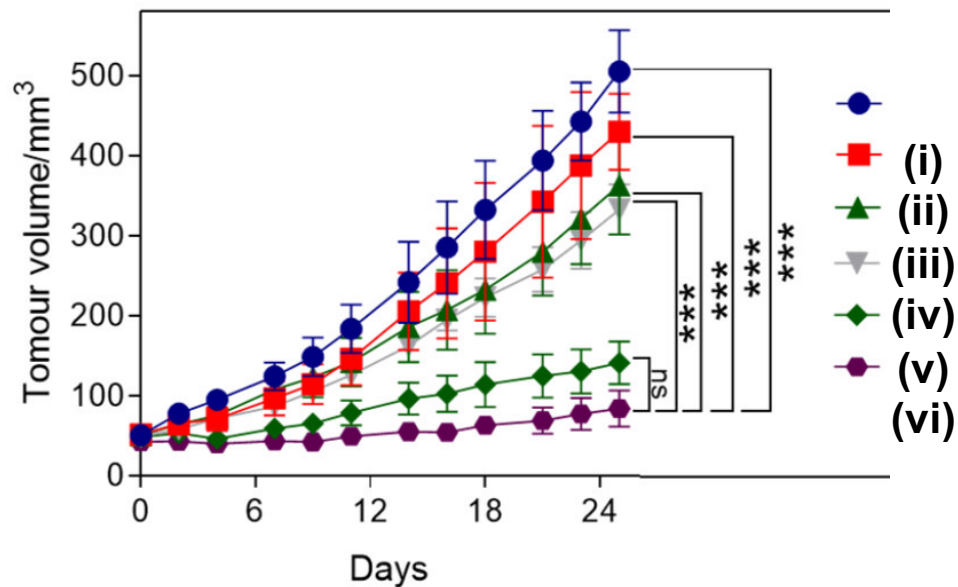


Only the combination of IrBA and MB with 630 nm *hν* showed cytotoxicity against tumor spheroid.

Antitumor Activity *in vivo* with MB

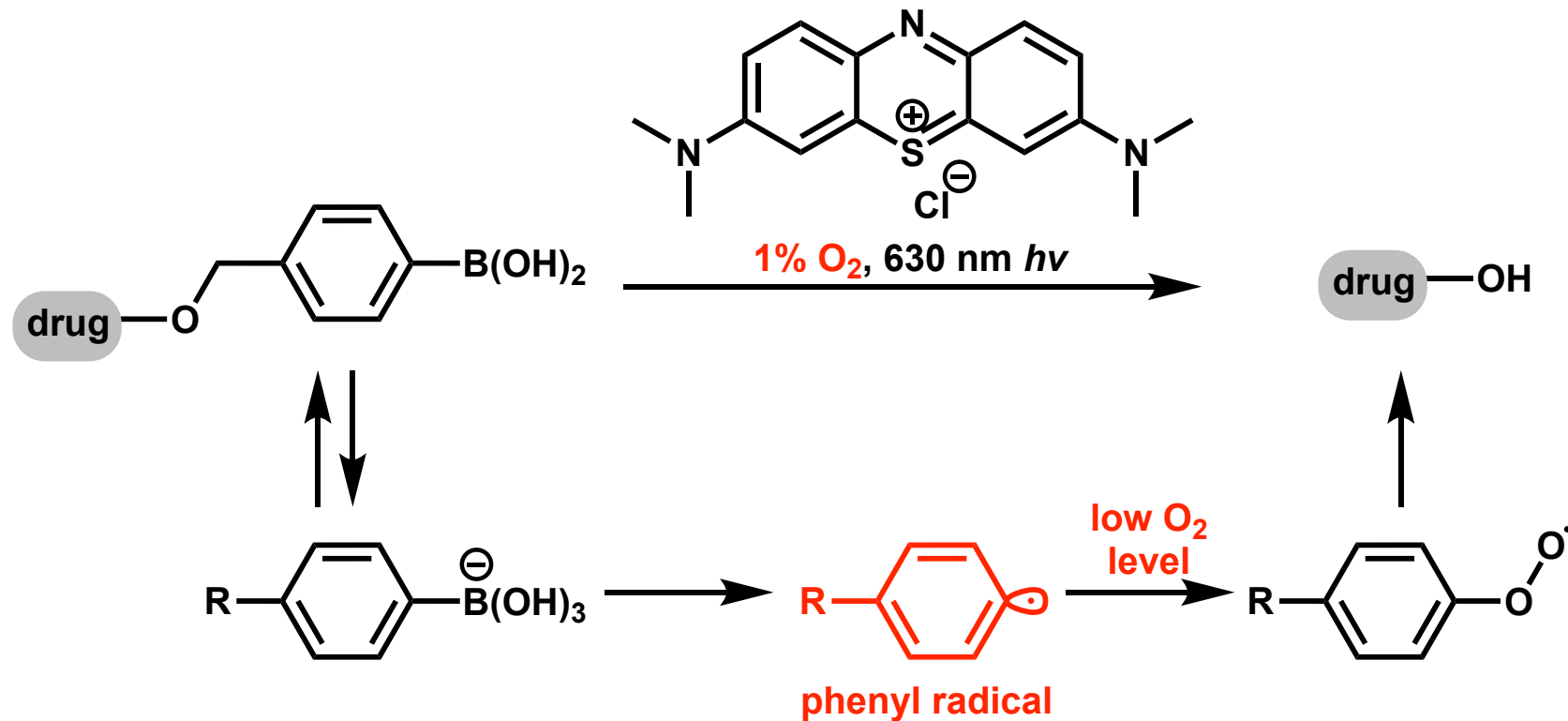


- (i) vehicle control
- (ii) IrBA + MB, dark
- (iii) IrBA + 635 nm *hv*
- (iv) MB + 635 nm *hv*
- (v) IrNH₂, dark
- (vi) IrBA + MB + 635 nm *hv*



- The combination of IrBA, MB and photoirradiation showed significant tumor growth inhibition activity compared to its parent drug IrNH₂.
- No mouse death and body weight loss occurred.

Summary



Boronic acids/esters prodrug can be photochemically uncaged under **hypoxia** via **phenyl radical**.

Boron-caged Ir(III) complex showed

- good PI value (= IC_{50, dark}/IC_{50, light}) against cancer cells **under hypoxia**
- good tumor growth inhibition activity and low side effects *in vivo* by combination with methylene blue and light irradiation.

This activation approach would be effective for cancer treatment.