

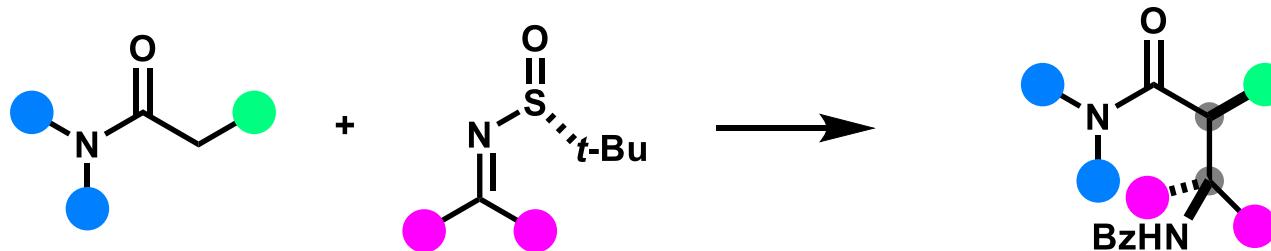
**Surrogate Asymmetric Mannich Reaction  
with Sulfinimines and Stereodivergent  
Olefination with Sulfur Ylides  
by Nuno Maulide's group**

2022/8/27 Yuya Shiga

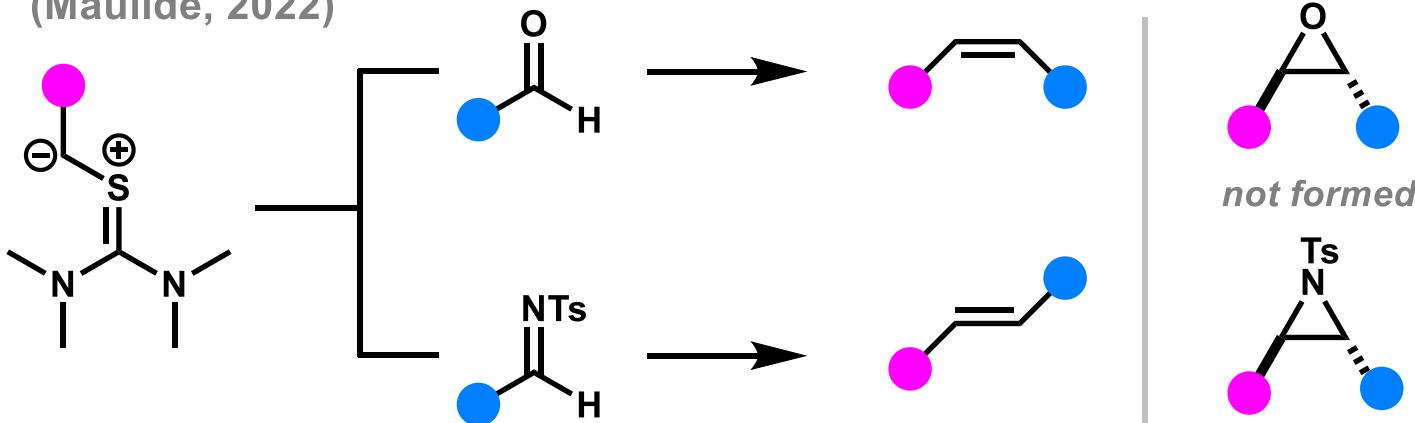
# Contents

## 1. Introduction

## 2. Surrogate Asymmetric Mannich Reaction with Sulfinimines. (Maulide, 2022)



## 3. Stereodivergent Olefination with Sulfur Ylides. (Maulide, 2022)



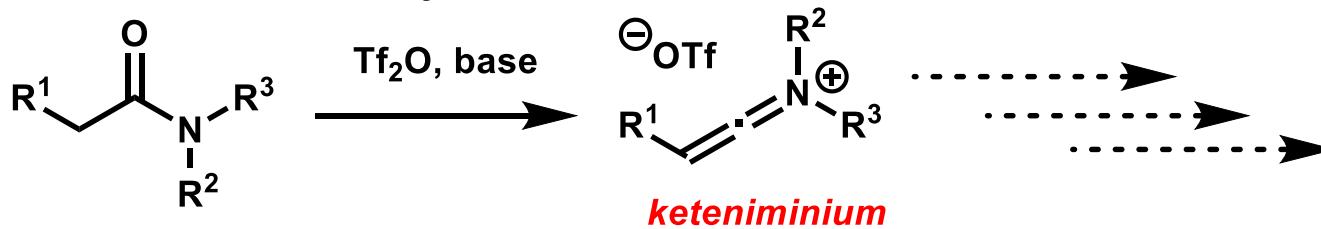
# Nuno Maulide



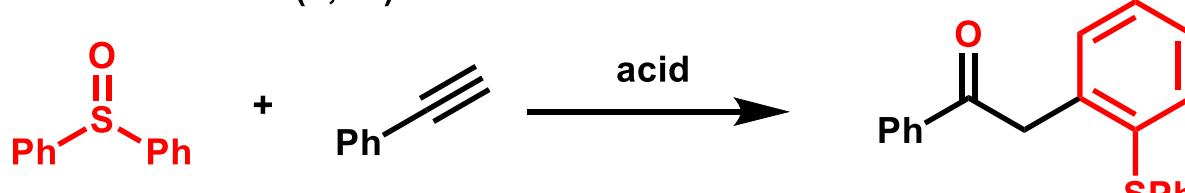
- 2003-2004 : Master's Degree, the Ecole Polytechnique  
2004-2007 : Ph. D, the Université catholique de Louvain (Prof. István E. Markó)  
2007-2008 : Postdoc, Stanford University (Prof. Barry M. Trost)  
2009-2013 : Group Leader, Max-Planck Institute for Coal Research  
2013- : Full Professor, the University of Vienna

## Research Topic:

1. Activation of amides or ynamides.

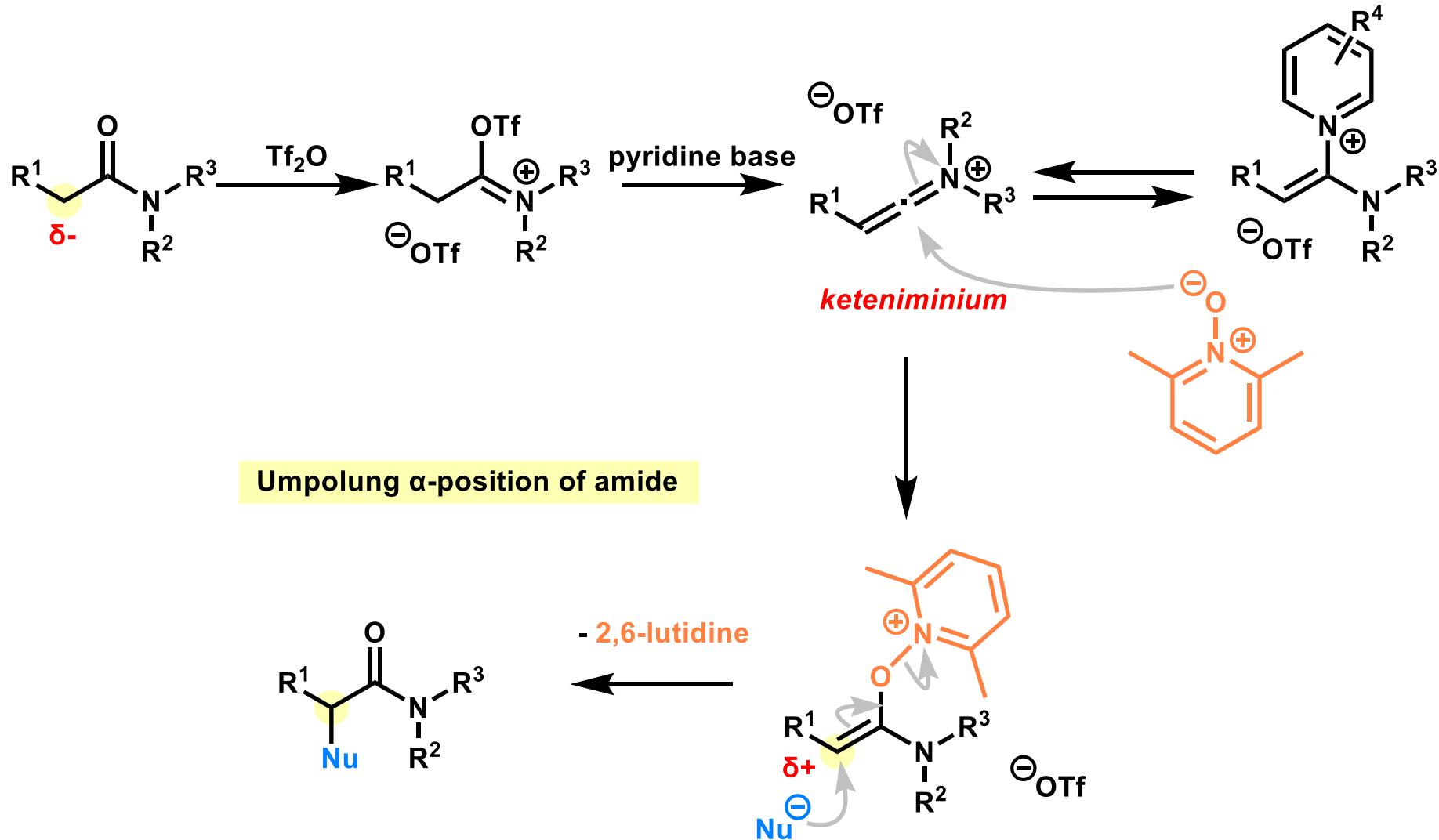


2. Reaction of Sulfur (II, IV).



3. Total Synthesis

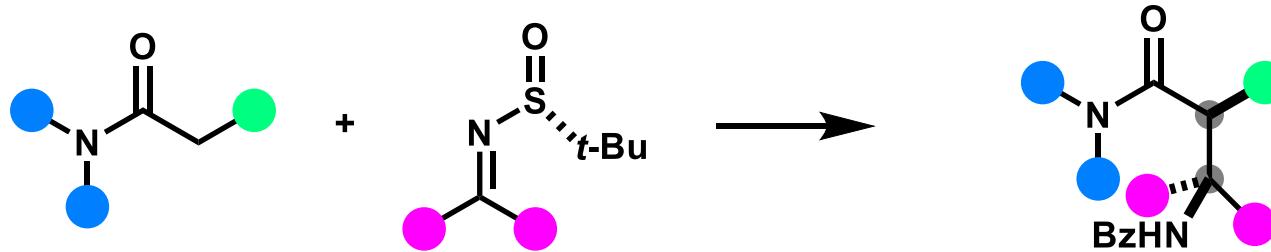
# Activation of Amide via Keteniminium



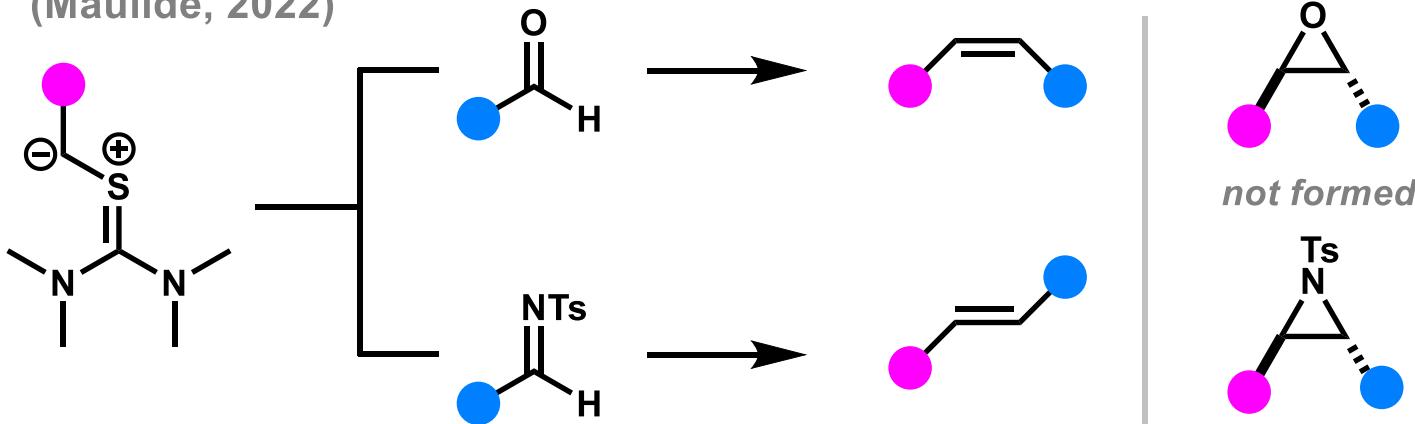
# Contents

## 1. Introduction

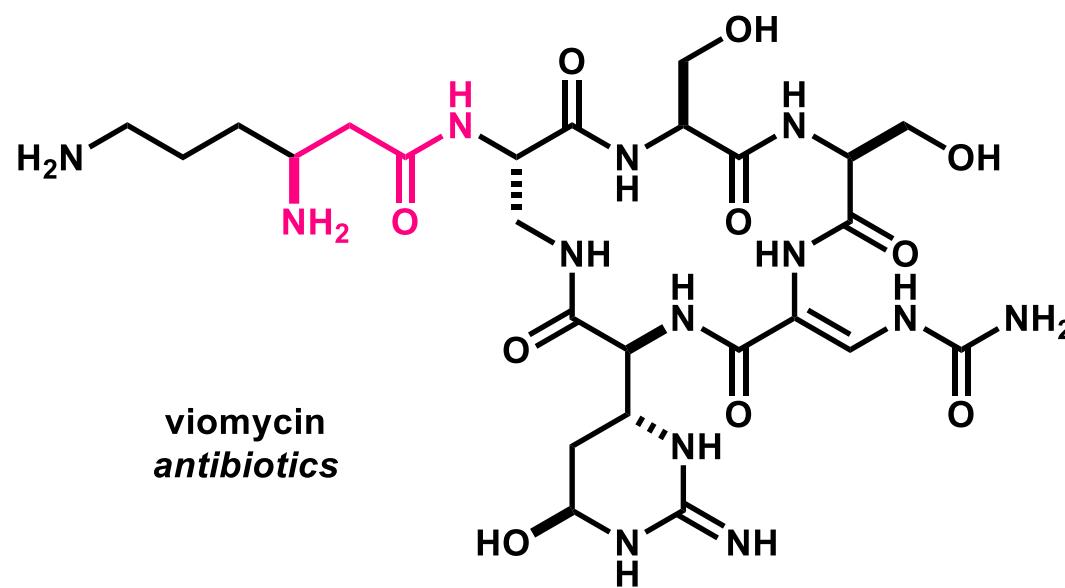
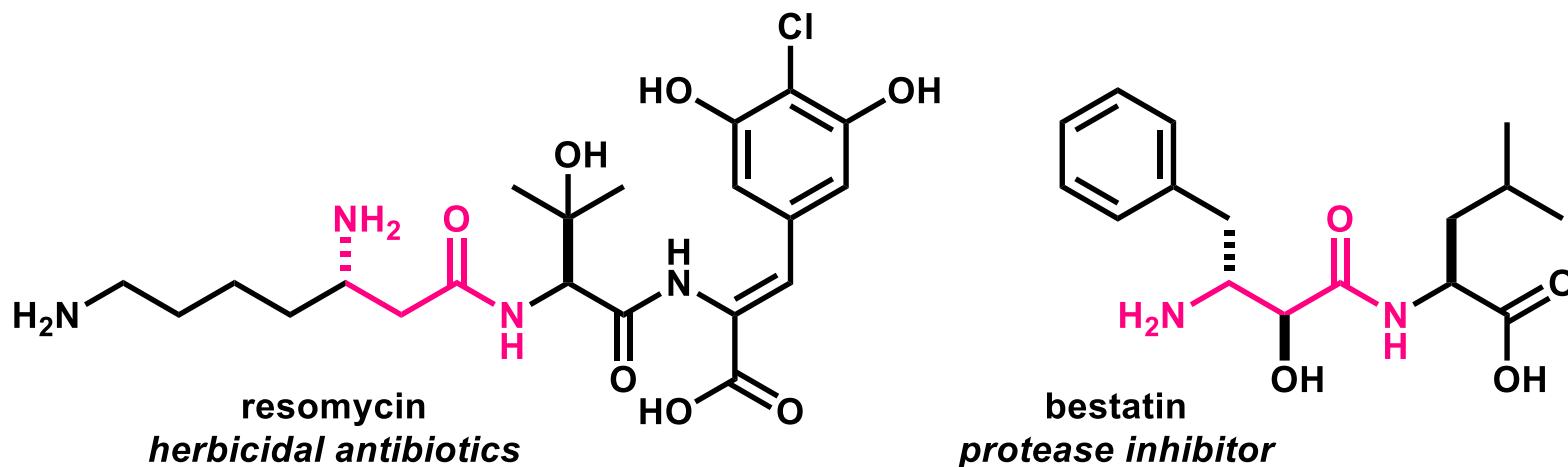
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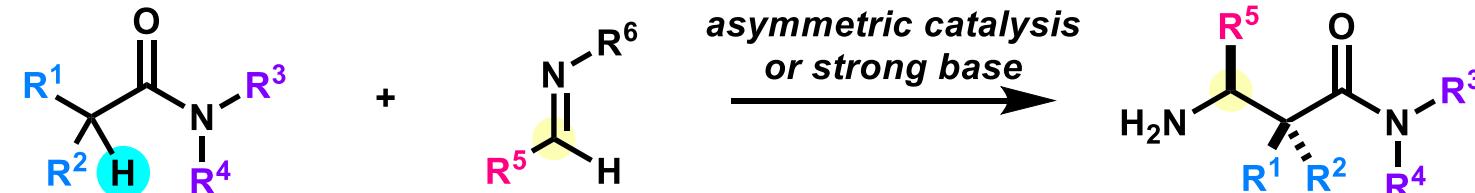
## 3. Stereodivergent Olefination with Sulfur Ylides. (Maulide, 2022)



# $\beta$ -Amino Acids as Natural Products and Drug Scaffolds

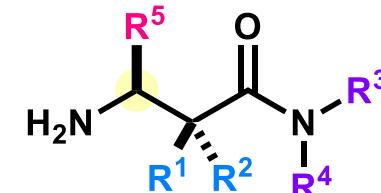


# Classical Mannich Reaction for the Construction of $\beta$ -Amino Amides



$pK_a = \sim 35$

aldimines



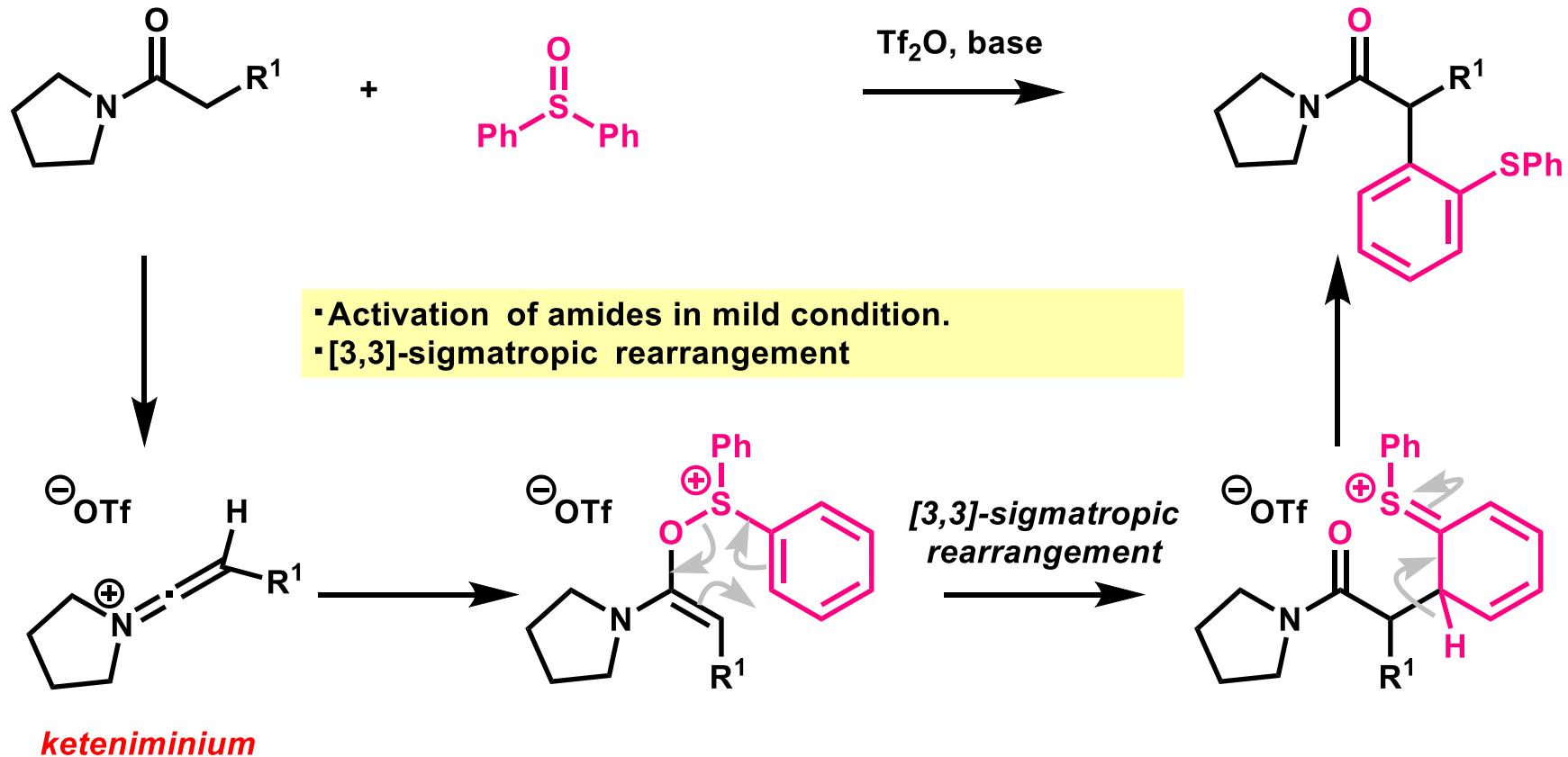
$pK_a = \sim 35$

ketimines

- Strong base is required
- The construction of quaternary carbon is unsolved.

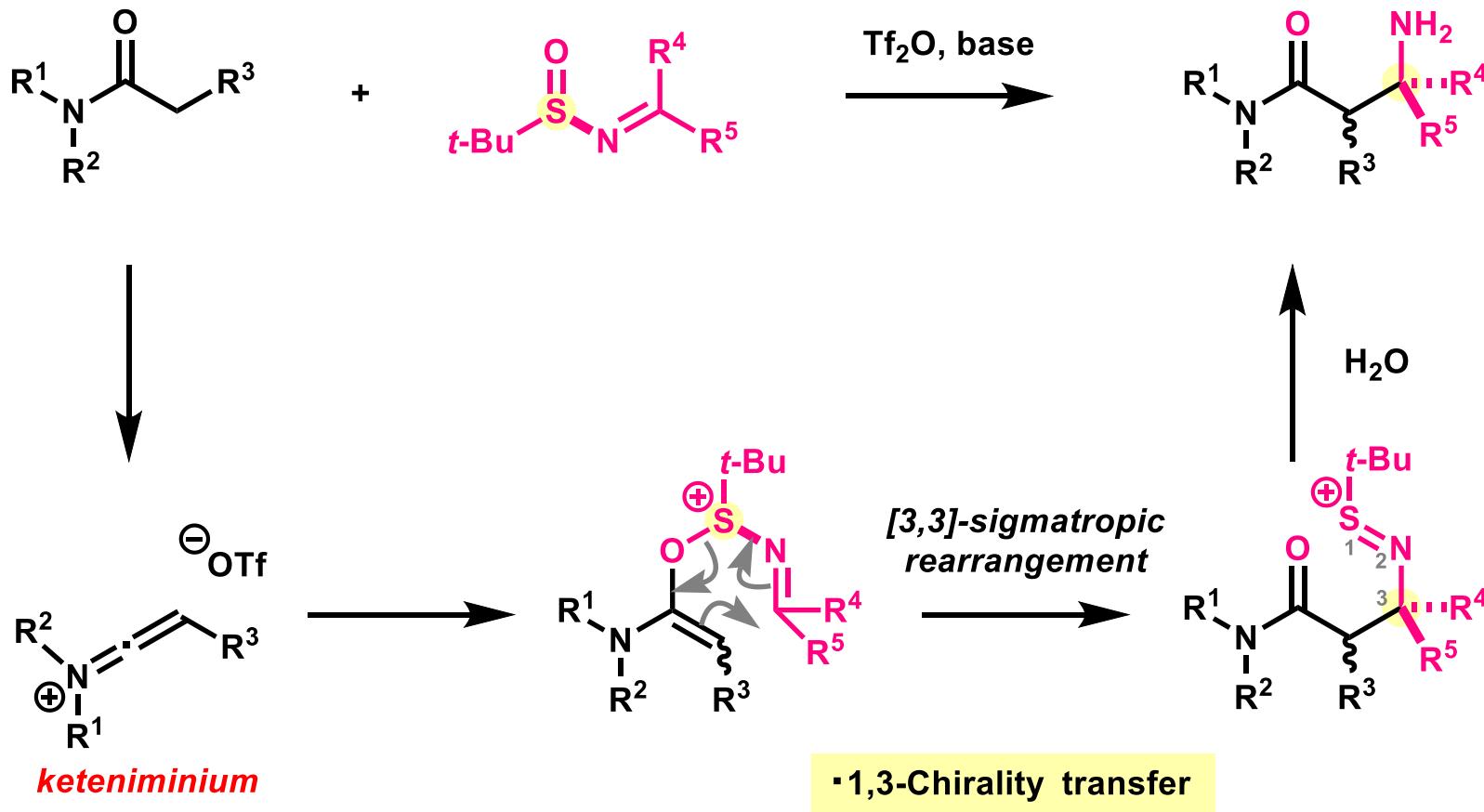
1) Yin, L.; Brewitz, L.; Kumagai, N.; Shibasaki, M. *J. Am. Chem. Soc.* **2014**, 136, 17958. 2) Sun, B.; Balaji, P. V.; Kumagai, N.; Shibasaki, M. *J. Am. Chem. Soc.* **2017**, 139, 8295.

# $\alpha$ -Arylation of Amides by Sulfonium Rearrangement

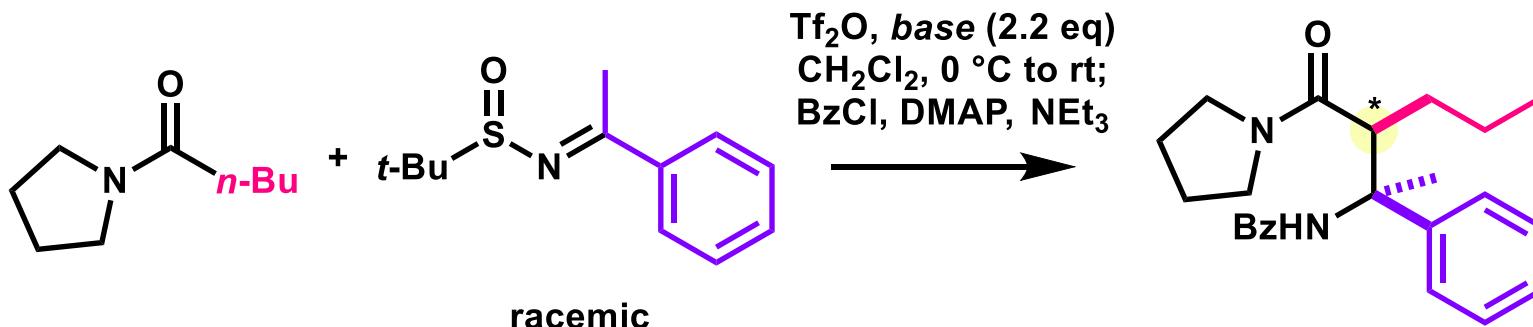


1) Huang, X.; Klimczyk, S.; Maulide, N. *Synthesis* **2012**, 2012, 175.

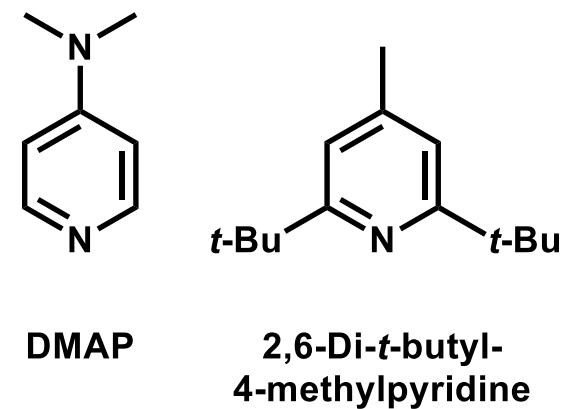
# A Sulfonium Rearrangement Approach for the Construction of $\beta$ -Amino Amides



# Optimization of the Reaction Conditions

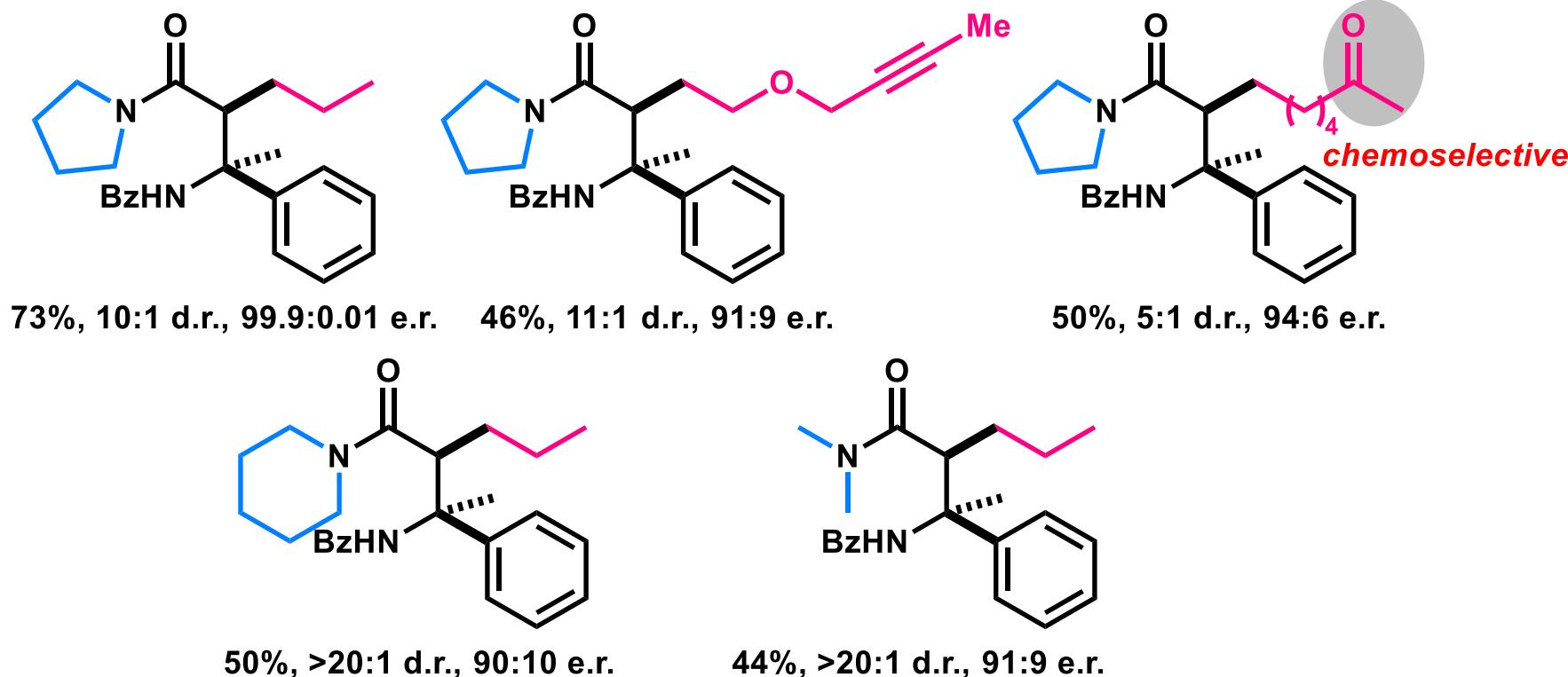
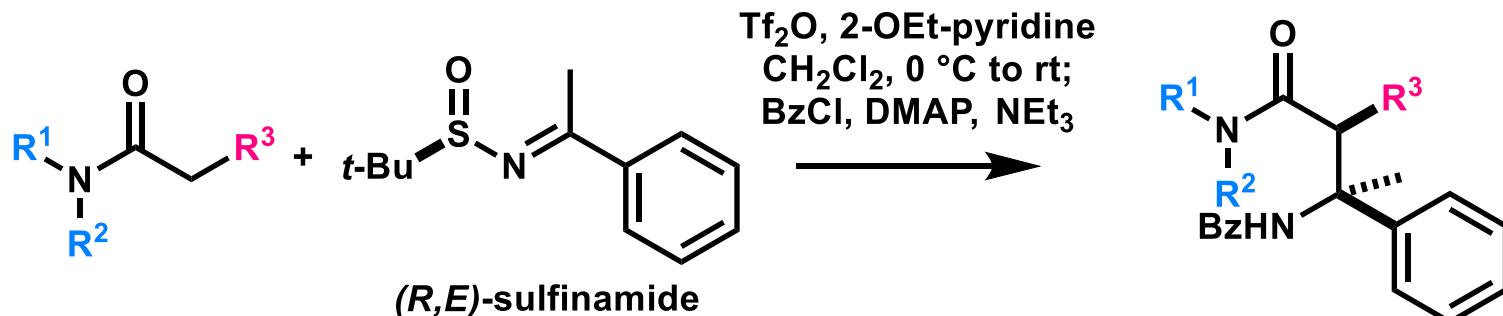


entry	base	yields	d.r.*
1	2,6-Di- <i>t</i> -butyl-4-methylpyridine	trace	-
2	2-Ph-pyridine	36%	7:1
3	2-Me-pyridine	50%	10:1
4	2-OEt-pyridine	73%	10:1
5*	2-OEt-pyridine	55%	10:1



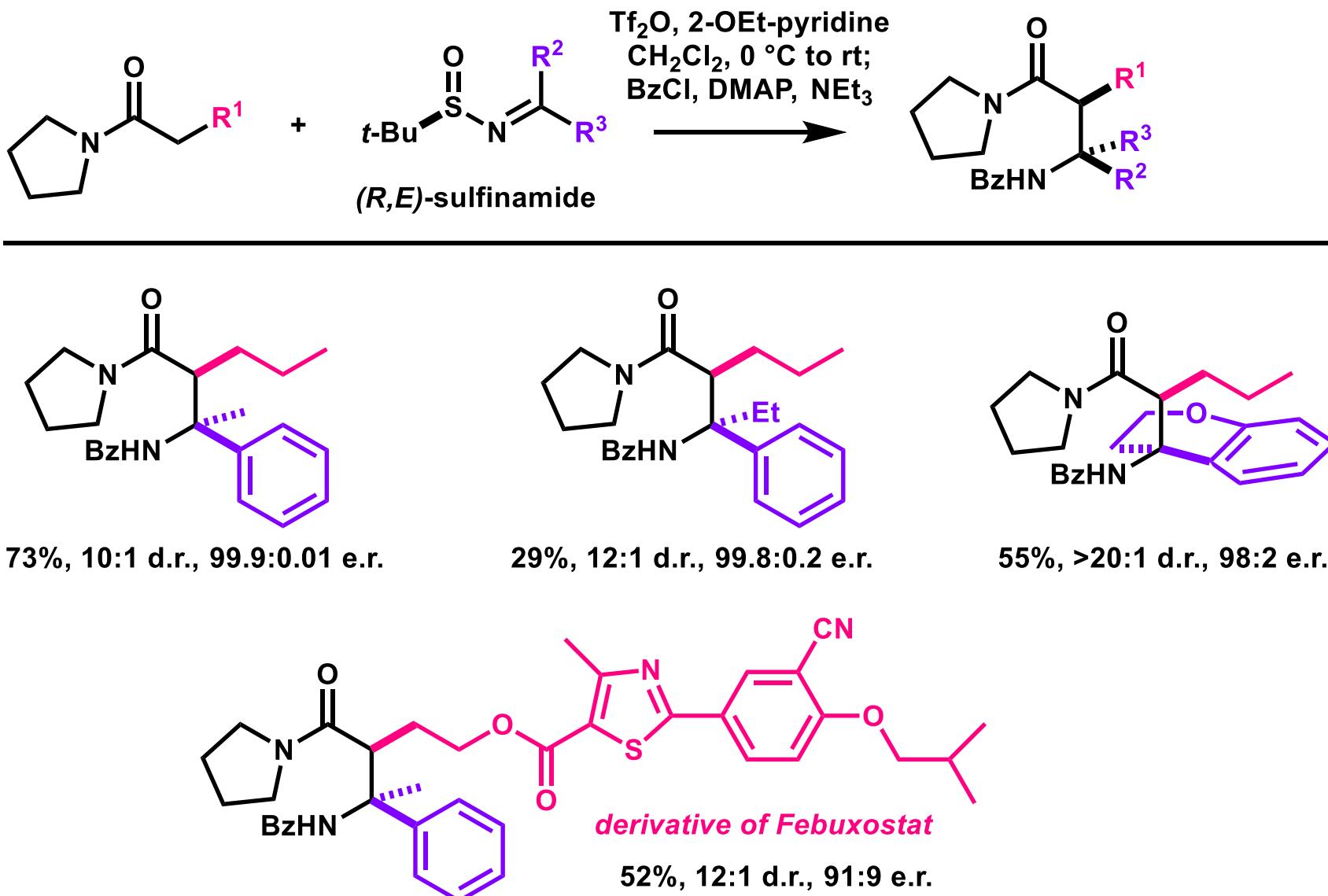
\* 3.0 equivalents of sulfonimine were used.

# Substrate Scope (I)



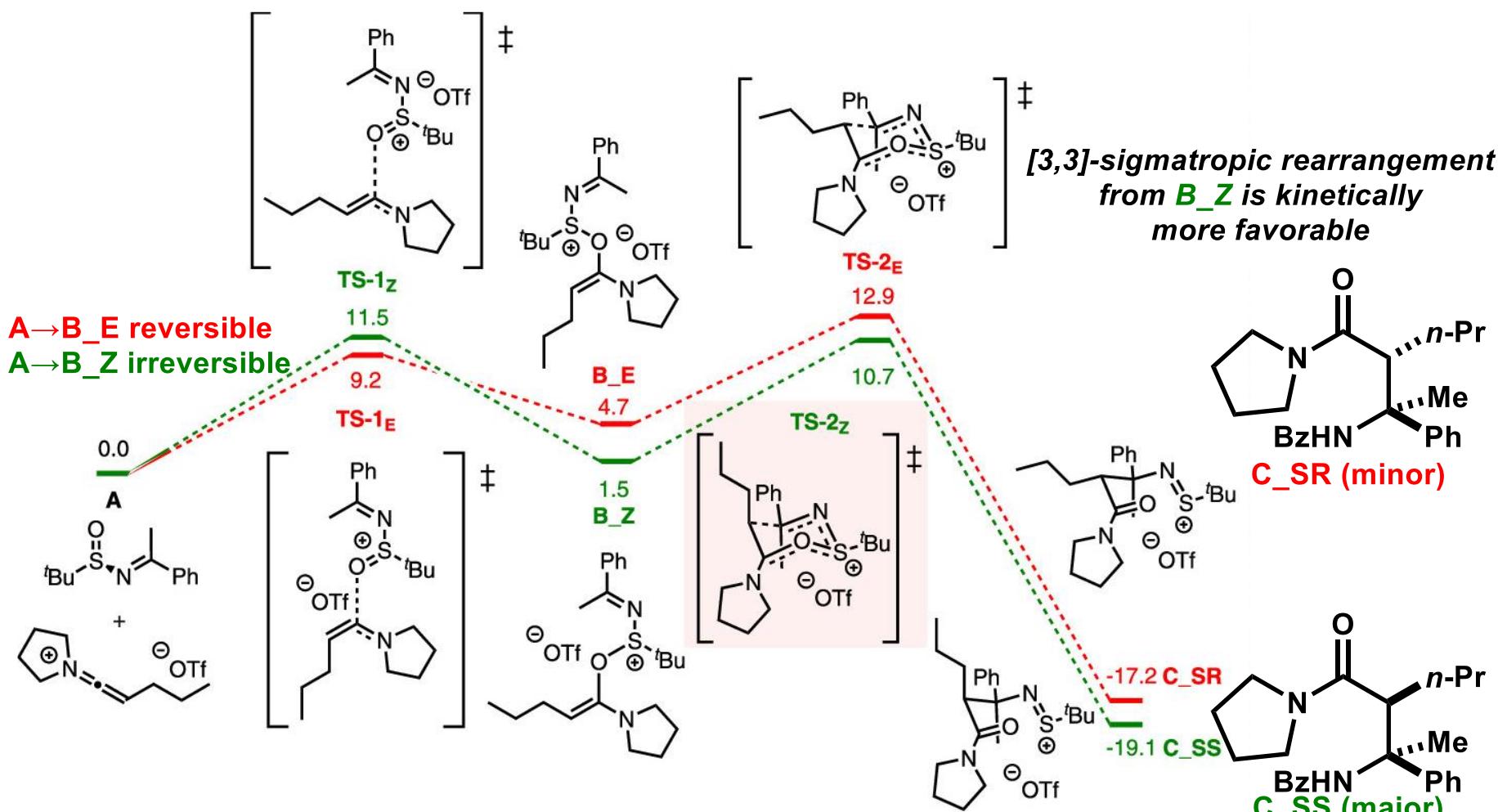
1) Feng, M.; Mosiagin, I.; Kaiser, D.; Maryasin, B.; Maulide, N. *J. Am. Chem. Soc.* **2022**, *144*, 13044.

## Substrate Scope (II)



1) Feng, M.; Mosiagin, I.; Kaiser, D.; Maryasin, B.; Maulide, N. *J. Am. Chem. Soc.* **2022**, *144*, 13044.

# Mechanistic Insight into the Diastereoselectivity

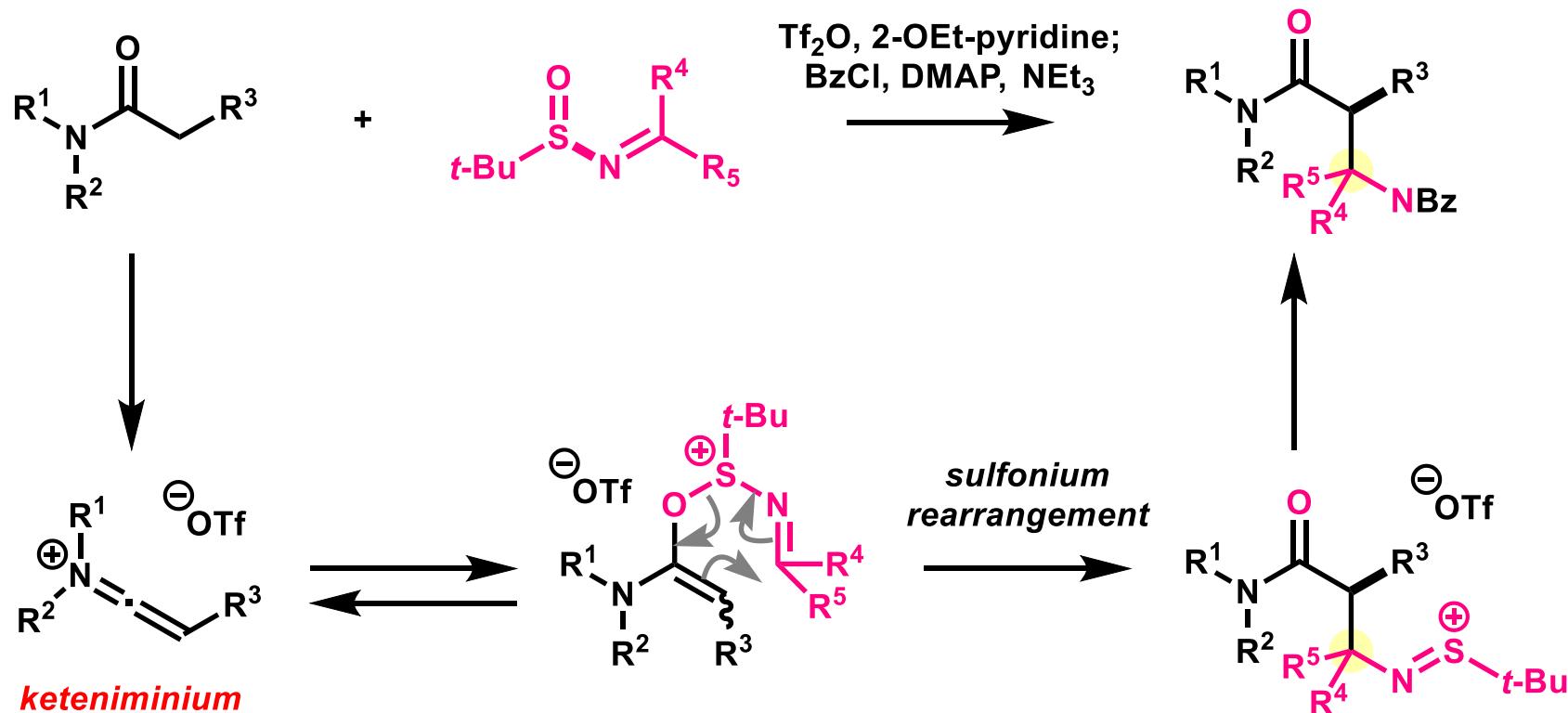


The diastereoselectivity is determined by the Z/E isomerism of the transient intermediate B.

Calculations were conducted at DLPNO-CCSD(T)/def2-TZVP//B3LYP-D3(BJ)/def2-SVP level of theory

1) Feng, M.; Mosiagin, I.; Kaiser, D.; Maryasin, B.; Maulide, N. *J. Am. Chem. Soc.* 2022, 144, 13044.

# Summary 1

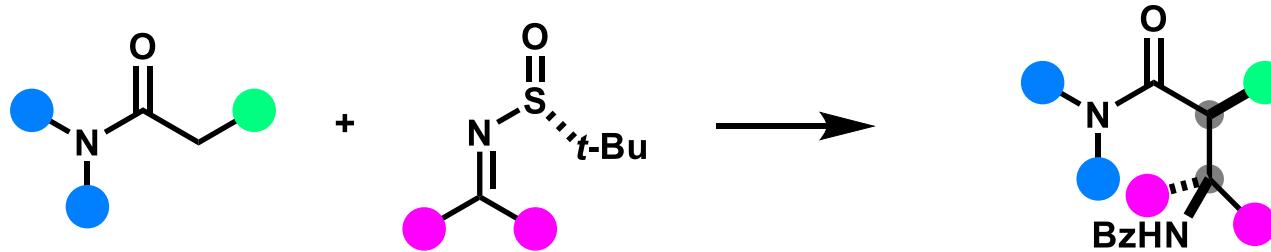


- surrogate asymmetric Mannich reaction
- quaternary stereocenters
- high diastereo- and enantioselective
- high functional-group tolerance

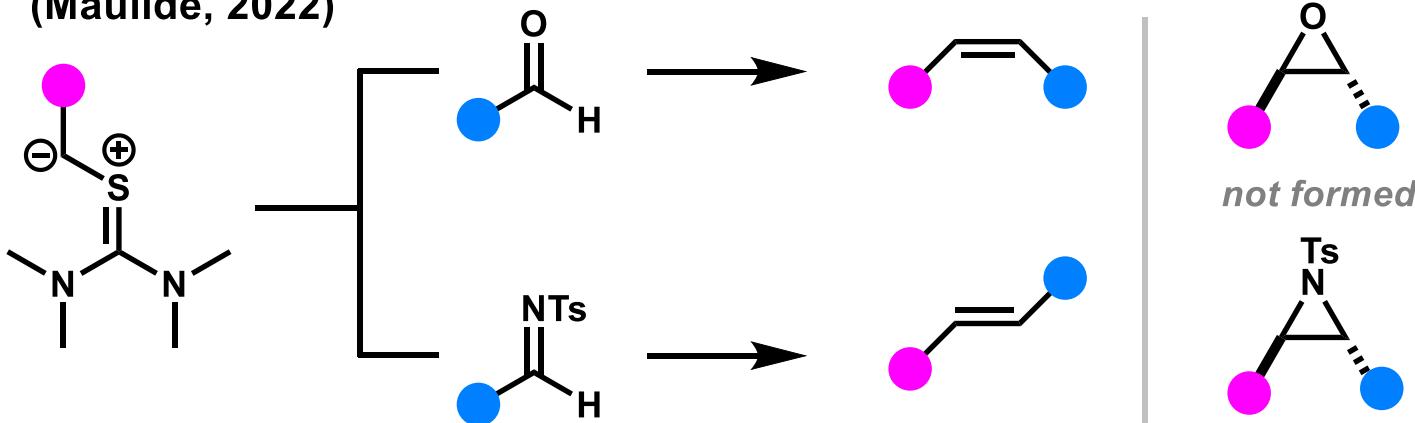
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## 1. Introduction

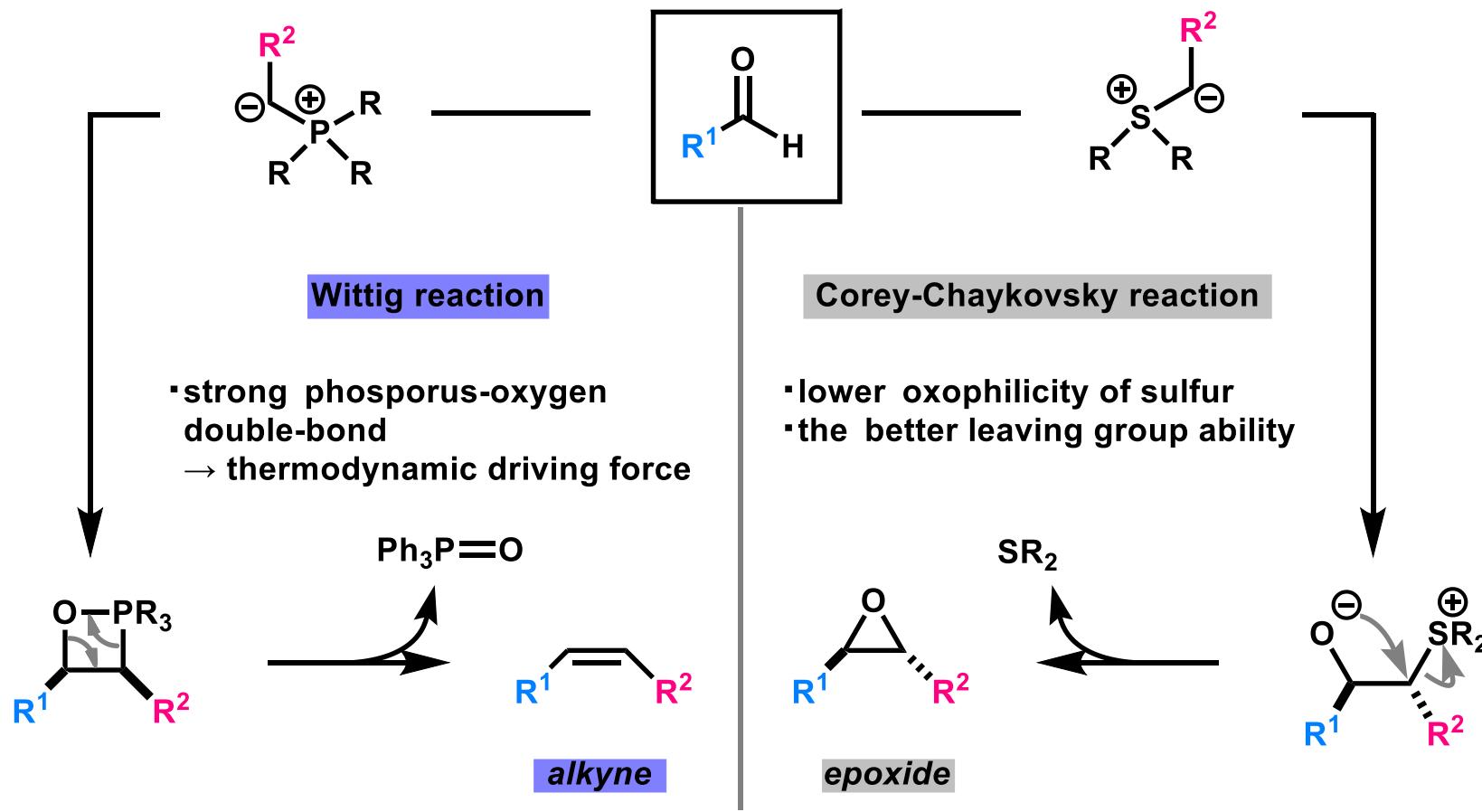
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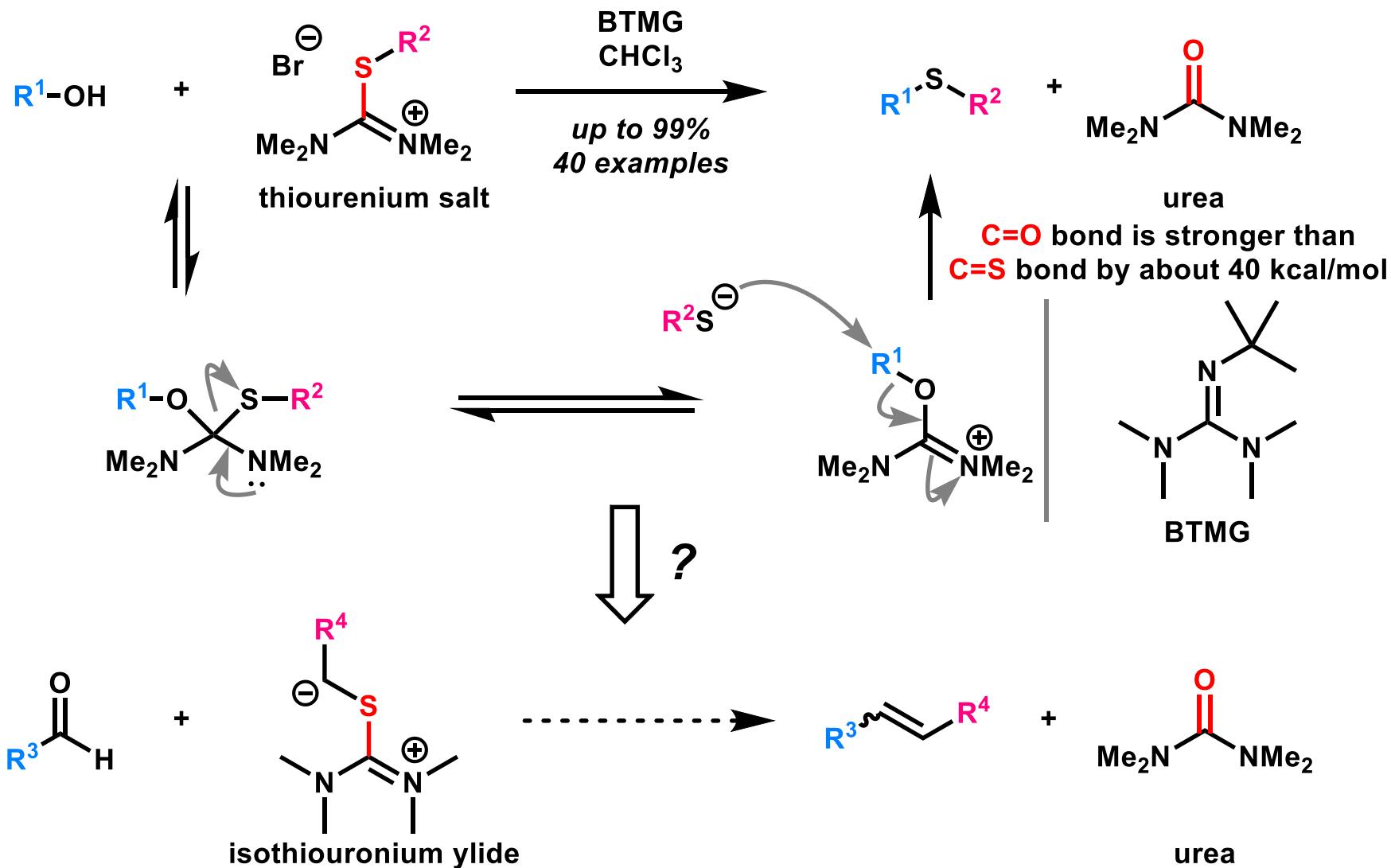


# P- and S-yliques: A Textbook Reactivity Dichotomy



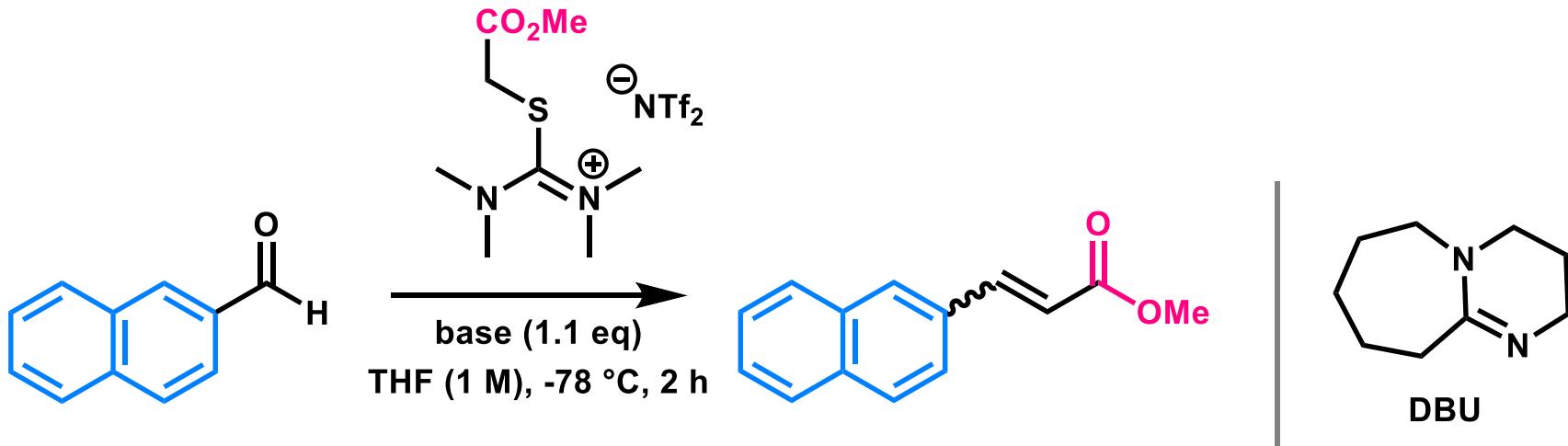
*Is it impossible to proceed olefination by S-ylide?*

# Previous Works about Thiouronium Salts



- 1) Merad, J.; Matyasovsky, J.; Stopka, T.; Brutiu, B. R.; Pinto, A.; Drescher, M.; Maulide, N. *Chem. Sci.* **2021**, 12, 7770. 2) Hadad, C. M.; Rablen, P. R.; Wiberg, K. B. *J. Org. Chem.* **1998**, 63, 8668.

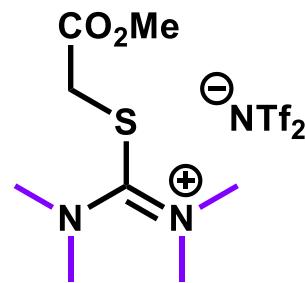
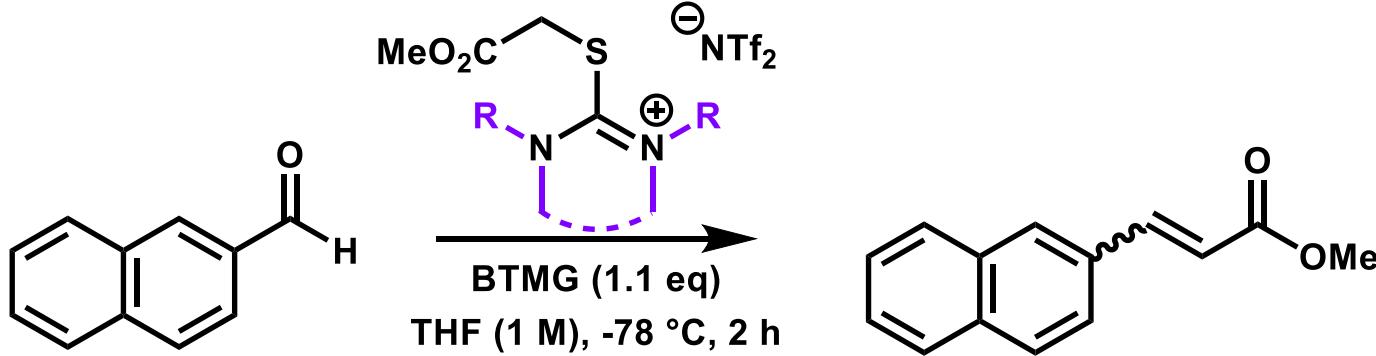
# Optimization for the Reaction Conditions



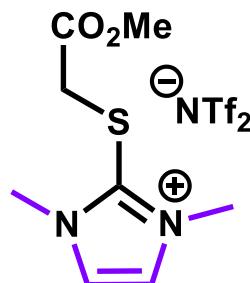
entry	base	solvent	olefin (E:Z)
1	KNTMS <sub>2</sub>	THF	93% (1.3/1)
2	LiN( <i>i</i> -Bu) <sub>2</sub>	THF	91% (1.2/1)
3	NEt <sub>3</sub>	THF	not observed
4	DBU	THF	55% (1/2.6)
5	BTMG	THF	93% (1/3.6)
6	BTMG	CH <sub>2</sub> Cl <sub>2</sub>	90% (2.2/1)

1) Merad, J.; Grant, P.; Stopka, T.; Sabbatani, J.; Meyrelles, R.; Preinfalk, A.; Matyasovsky, J.; Maryasin,  
B.; Gonzalez, L.; Maulide, N. *J. Am. Chem. Soc.* **2022**, 144, 12536.

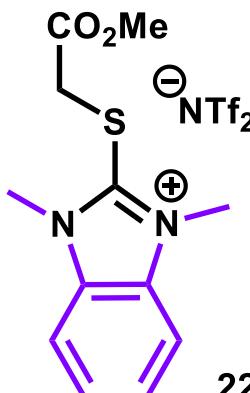
# Optimization of Isothiouronium Bistriflimides



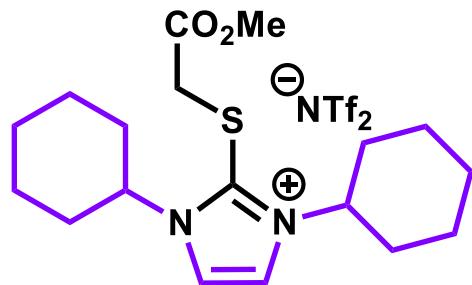
93% (E:Z=1/3.6)



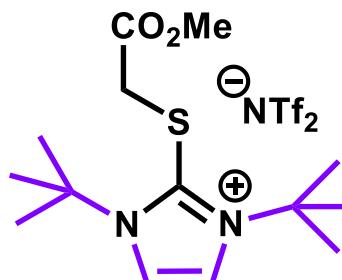
97% (E:Z=1/2.8)



22% (E:Z=1/1.5)



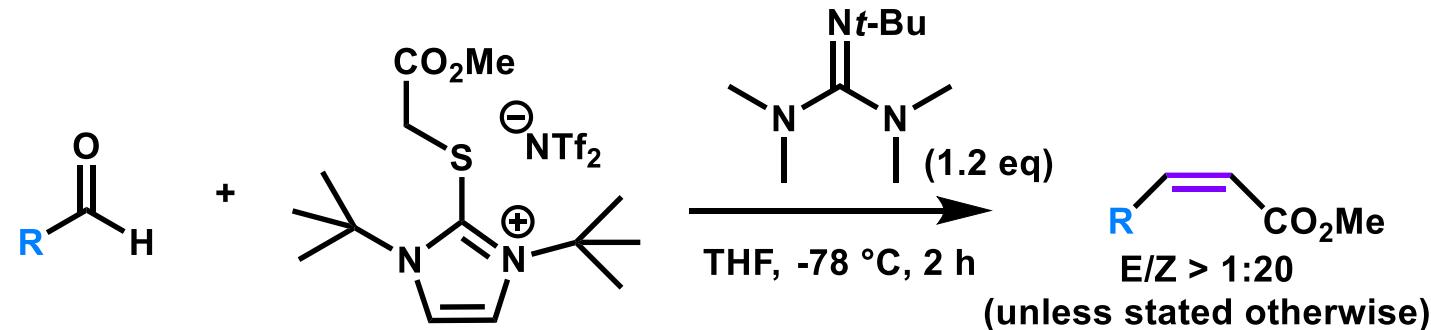
60% (E:Z=1/18)



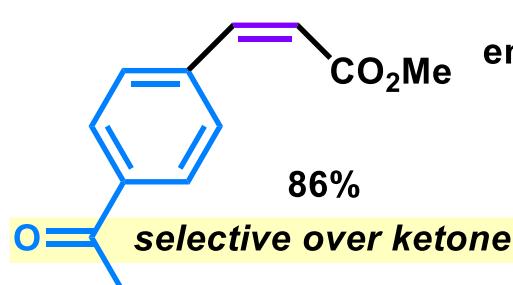
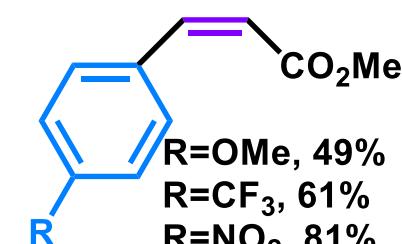
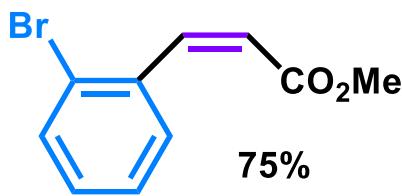
92% (only Z)

1) Merad, J.; Grant, P.; Stopka, T.; Sabbatani, J.; Meyrelles, R.; Preinfalk, A.; Matyasovsky, J.; Maryasin, B.; Gonzalez, L.; Maulide, N. *J. Am. Chem. Soc.* **2022**, 144, 12536.

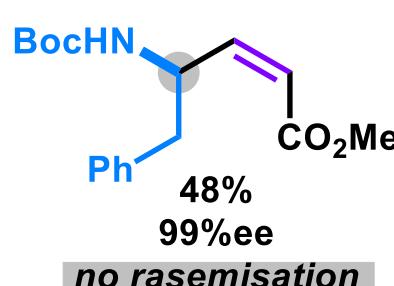
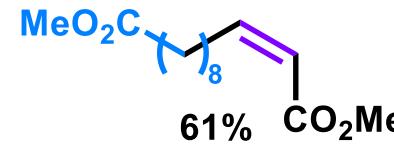
# Substrate Scope (I)



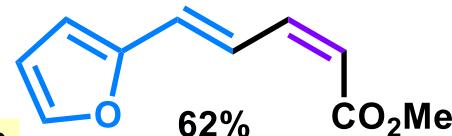
aryl aldehydes



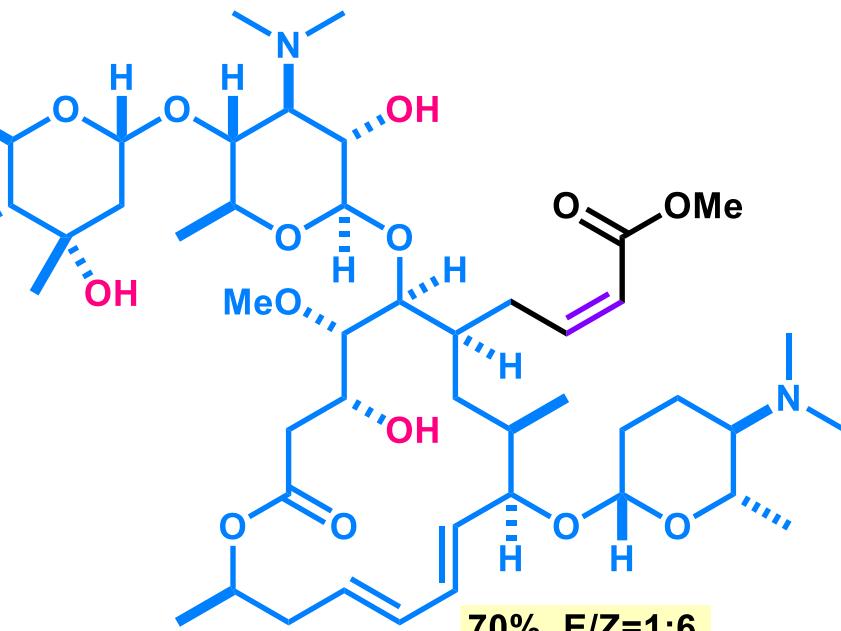
aliphatic aldehydes



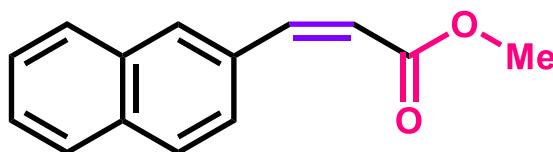
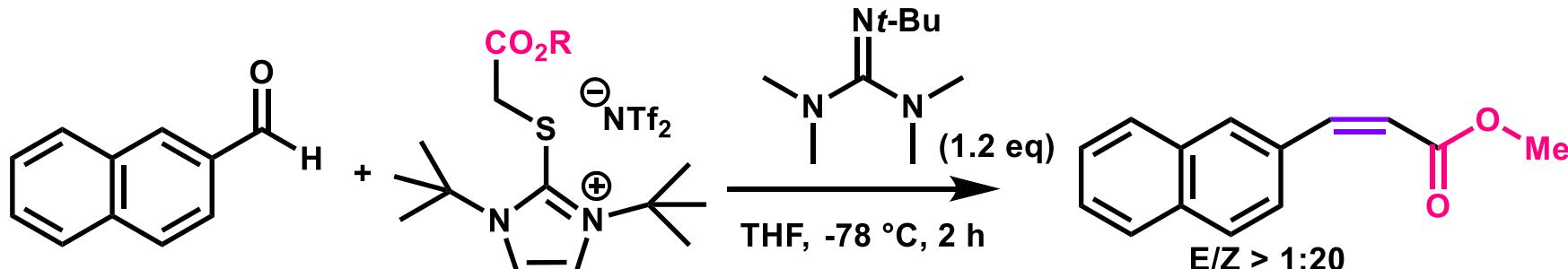
enal



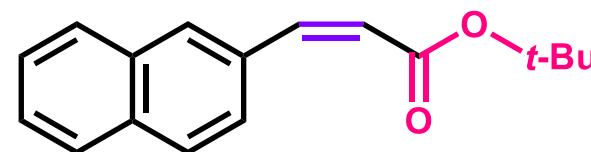
complex *unprotected* macrolide antibiotic



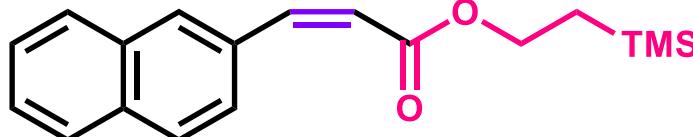
## Substrate Scope (II)



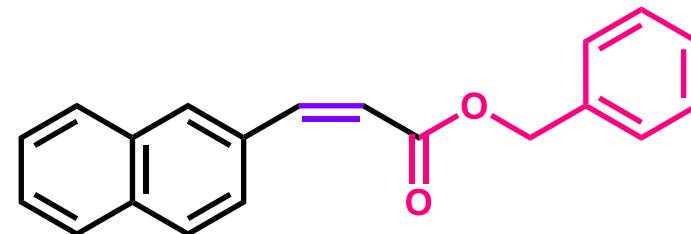
92%  
(88% gram scale)



83%



84%



85%

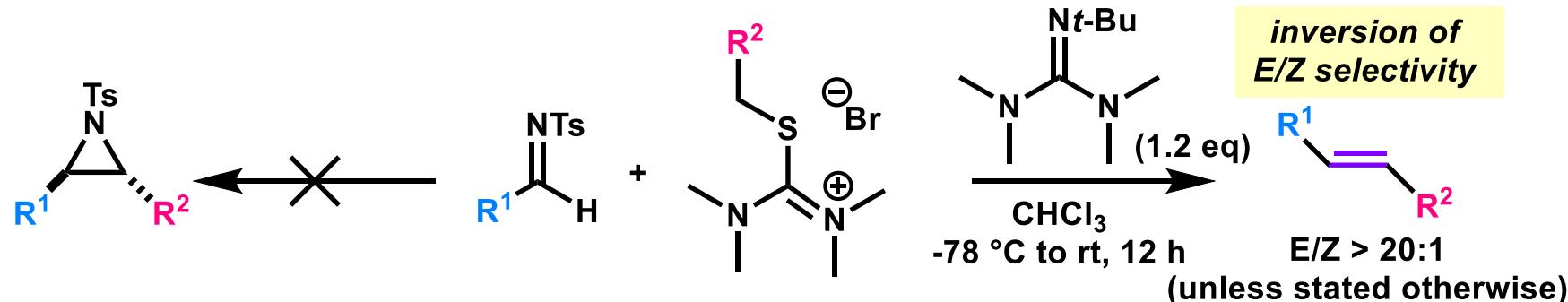
# Reaction of N-Tosylimines



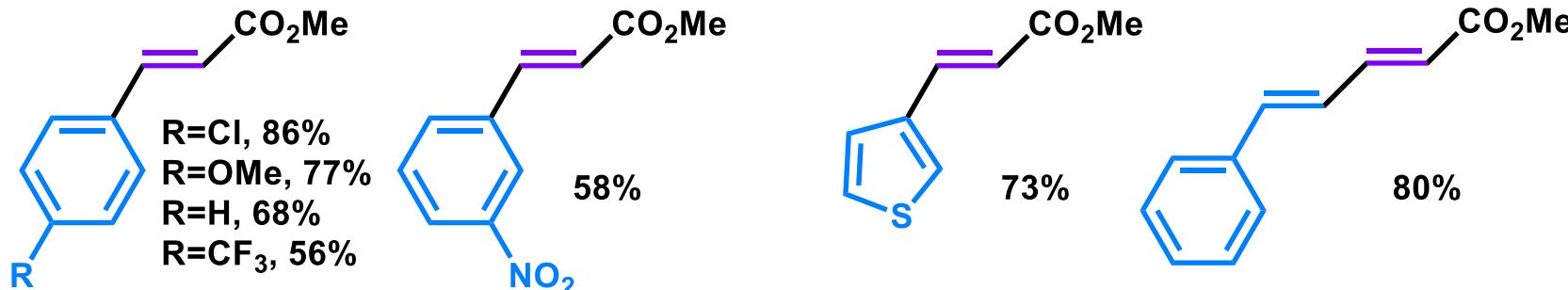
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1) Merad, J.; Grant, P.; Stopka, T.; Sabbatani, J.; Meyrelles, R.; Preinfalk, A.; Matyasovsky, J.; Maryasin,  
B.; Gonzalez, L.; Maulide, N. *J. Am. Chem. Soc.* **2022**, 144, 12536.

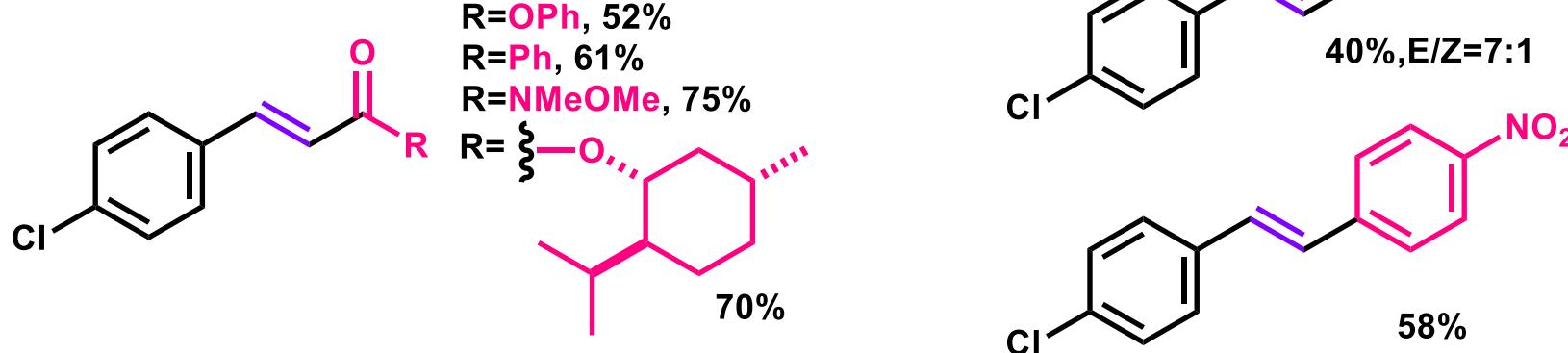
# Substrate Scope of E-Selective Olefination



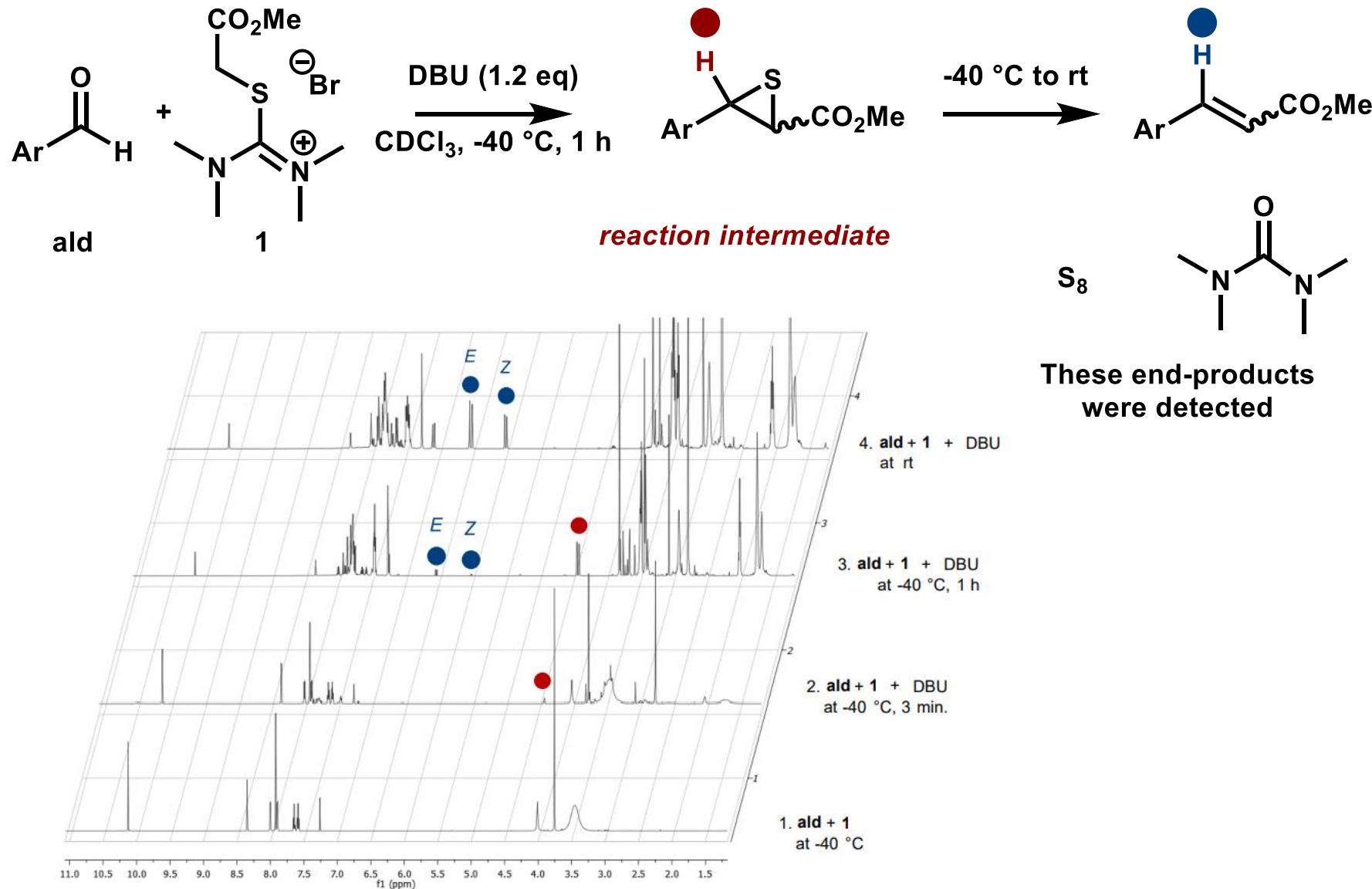
## Tosyl imine scope



## Isothiouronium scope

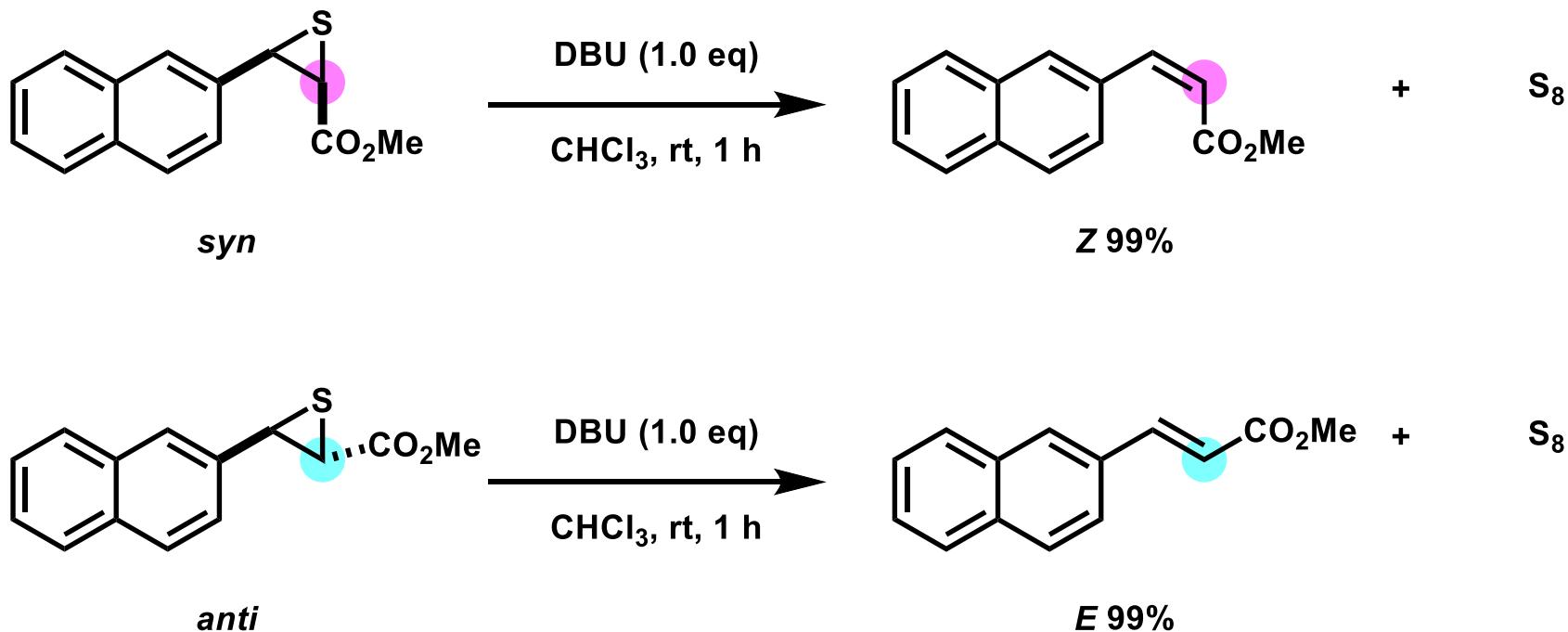


# Mechanistic Study



1) Merad, J.; Grant, P.; Stopka, T.; Sabbatani, J.; Meyrelles, R.; Preinfalk, A.; Matyasovsky, J.; Maryasin, B.; Gonzalez, L.; Maulide, N. *J. Am. Chem. Soc.* **2022**, 144, 12536.

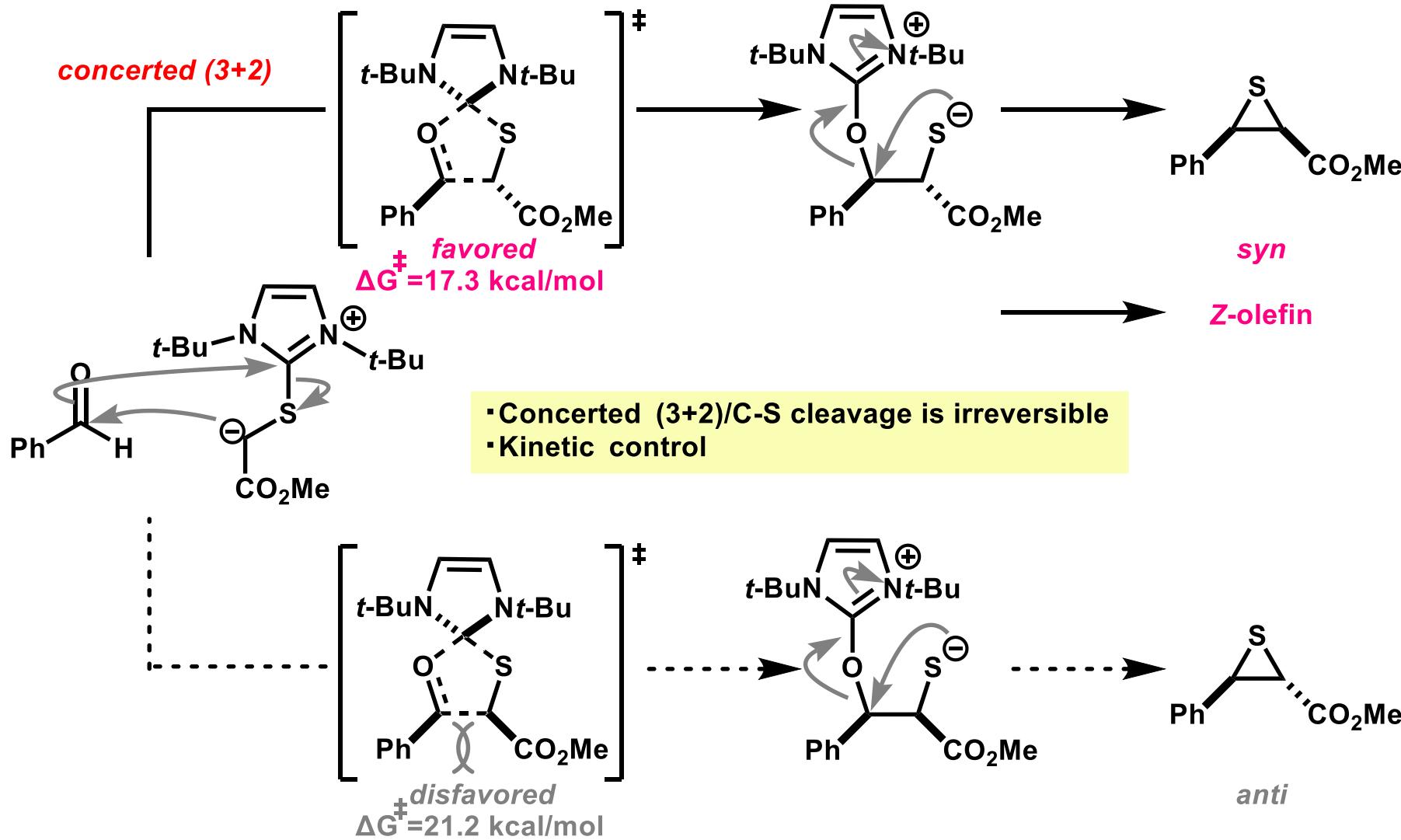
# Stereospecificity of Desulfurisation



Desulfurisation proceeds stereospecifically.

→ The selectivity of olefination is determined by the stereochemistry of episulfide, the reaction intermediate.

# Z-Selective Olefination via *syn*-Episulfide



Calculations were conducted at PBE0-D3BJ/def2-TZVP,SMD//PBE0-D3BJ/def2-SVP,SMD level of theory

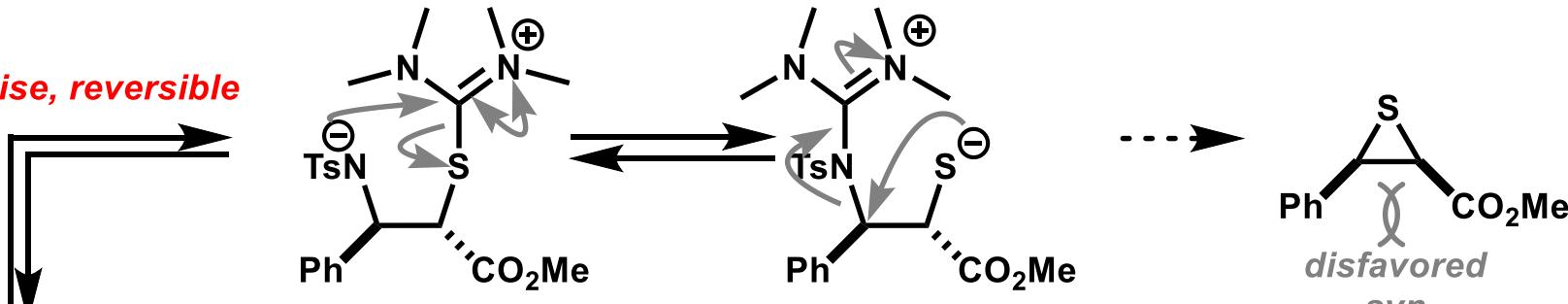


*E*-olefin

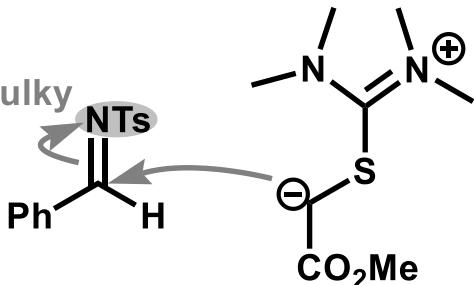
1) Merad, J.; Grant, P.; Stopka, T.; Sabbatani, J.; Meyrelles, R.; Preinfalk, A.; Matyasovsky, J.; Maryasin, B.; Gonzalez, L.; Maulide, N. *J. Am. Chem. Soc.* **2022**, 144, 12536.

# E-Selective Olefination via *anti*-Episulfide

*stepwise, reversible*

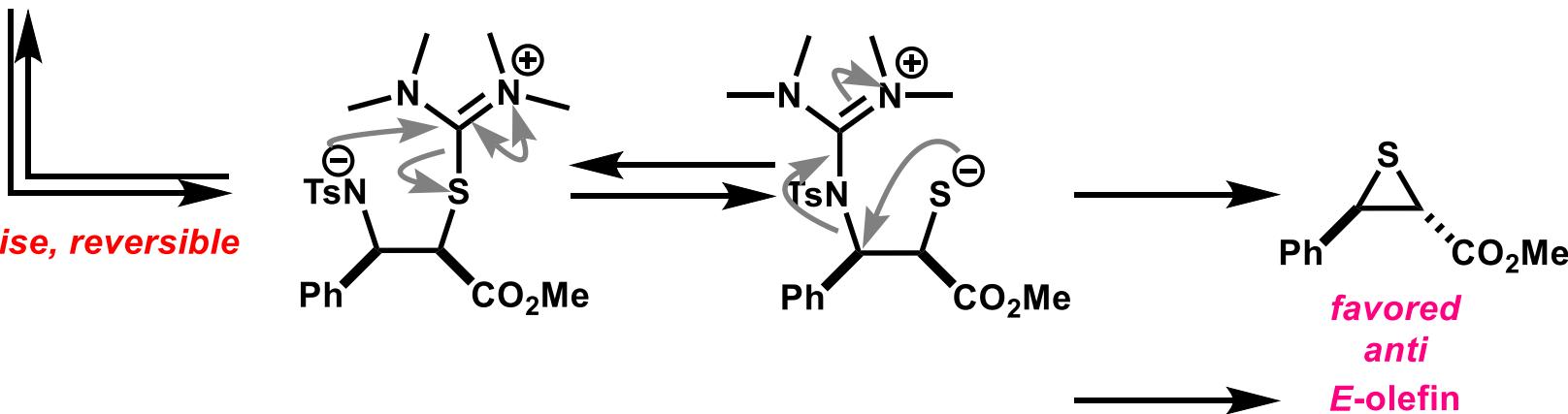


*bulky*



- Stepwise C-C formation /C-S cleavage is reversible
- Thermodynamic control

*stepwise, reversible*



1) Merad, J.; Grant, P.; Stopka, T.; Sabbatani, J.; Meyrelles, R.; Preinfalk, A.; Matyasovsky, J.; Maryasin, B.; Gonzalez, L.; Maulide, N. *J. Am. Chem. Soc.* **2022**, 144, 12536.

# Summary 2

