

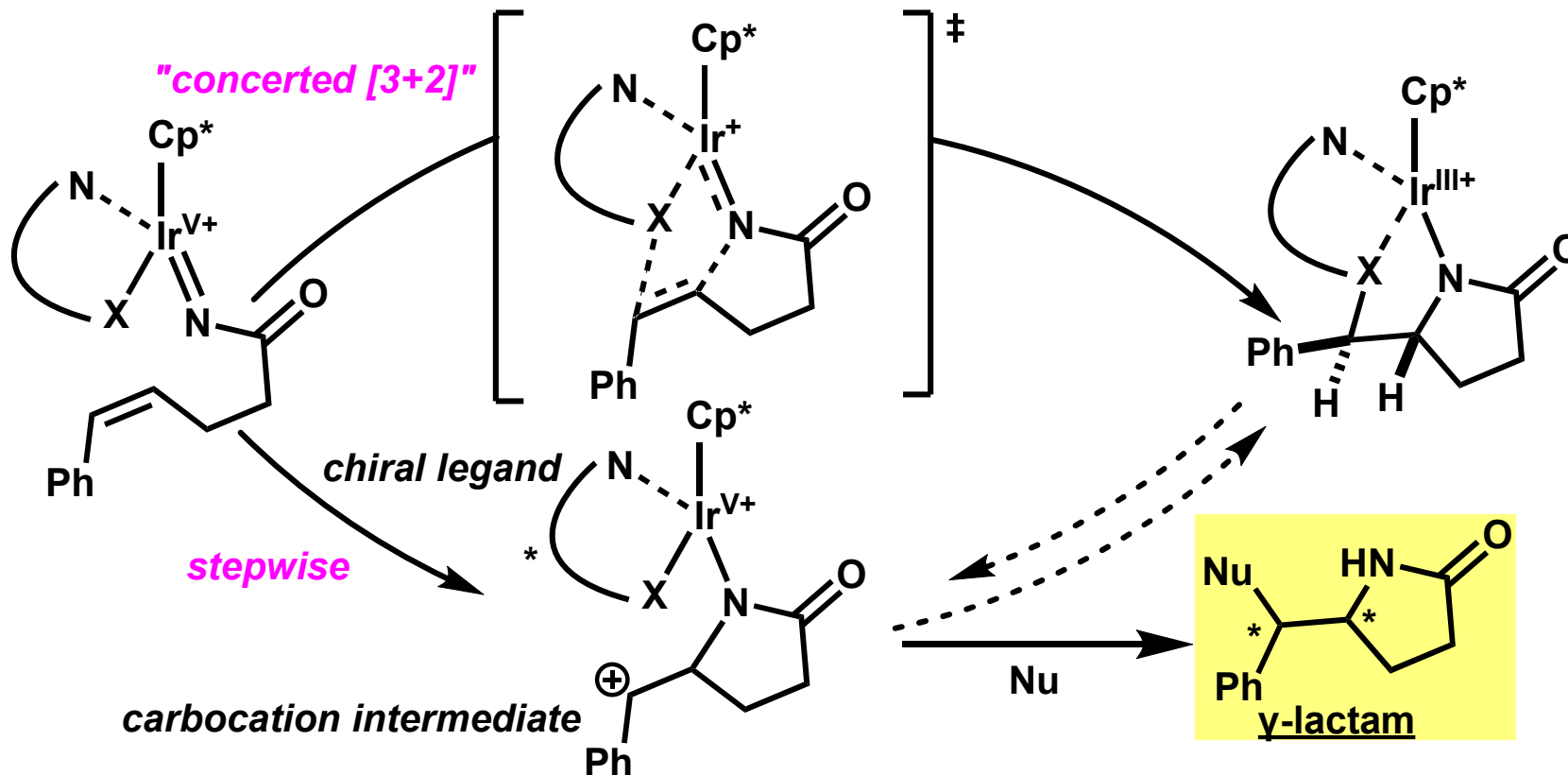
Access to β - and γ -Lactams via Olefin Difunctionalization

2022.5.7. LS Junichi Taguchi

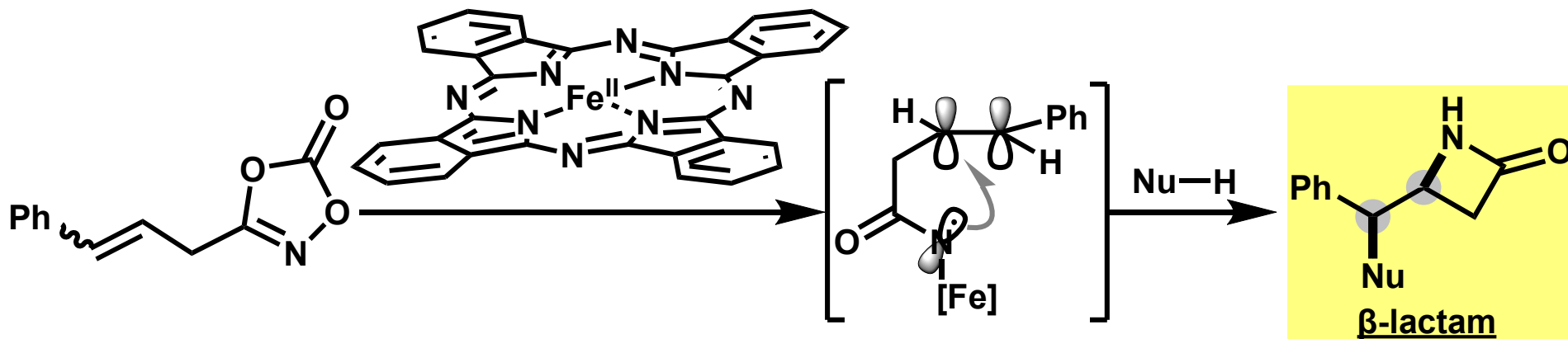
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1. Introduction (An Alternative to Acyl Azides)

2. Stereodefined Access to γ -Lactams via Olefin Difunctionalization



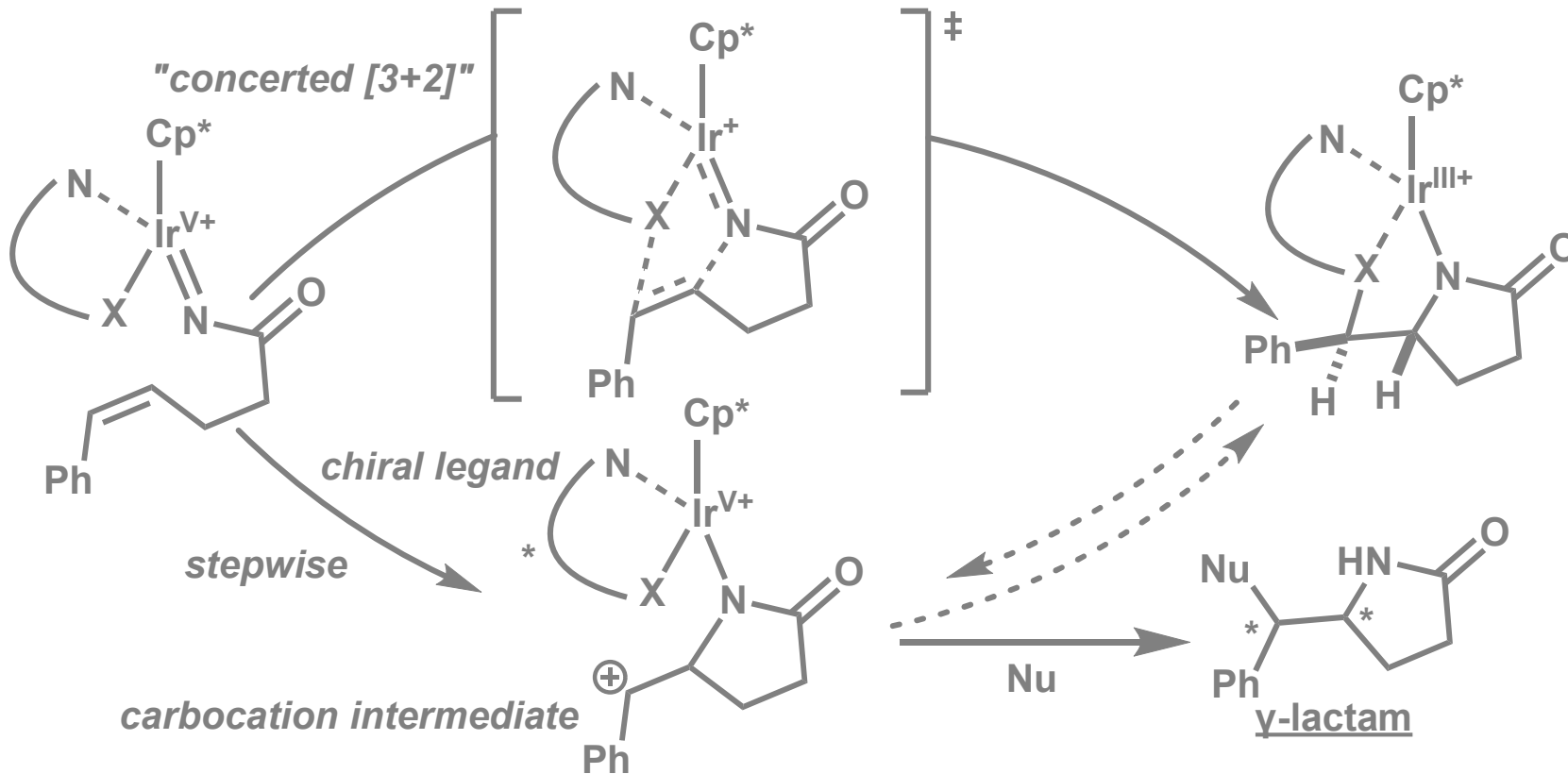
3. Access to β -Lactams via Iron-Catalyzed Olefin Oxyamidation Enabled by the π -Accepting Phthalocyanine Ligand (Main Paper) (*J. Am. Chem. Soc.* 2022, 144, 1872)



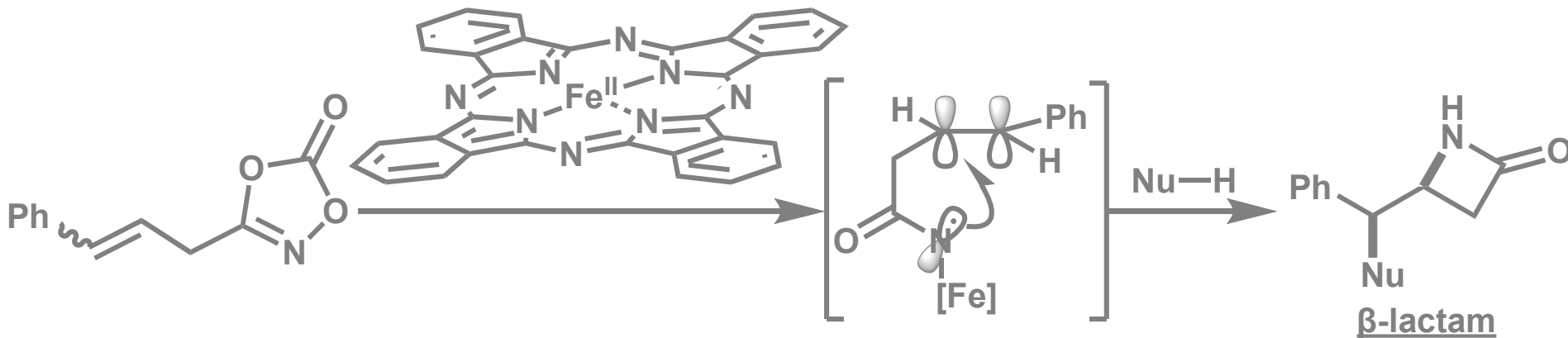
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Prof. Sukbok Chang

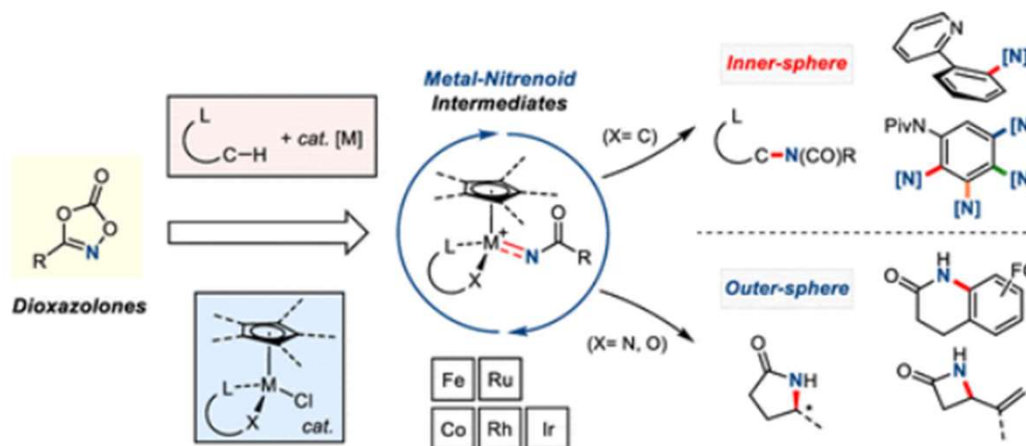
Education and Professional Careers:

1985 B.S. @ Korea university
1987 M.S. @ KAIST (Advisor: Prof. Sunggak Kim)
1996 Ph.D. @ Harvard University
(Advisor: Prof. Eric. N. Jacobsen)
1996-1998 Post Doctoral Fellow @ Caltech
(Advisor: Prof. Robert H. Grubbs)
1998-2002 Assistant Professor @ Ewha Womans University
2002- Professor @ KAIST
2018- Distinguished Professor @ KAIST



Research Interests:

1. Site-selective C-H activation strategies especially for C-C, C-N and C-O bond formations
(Today's topic)

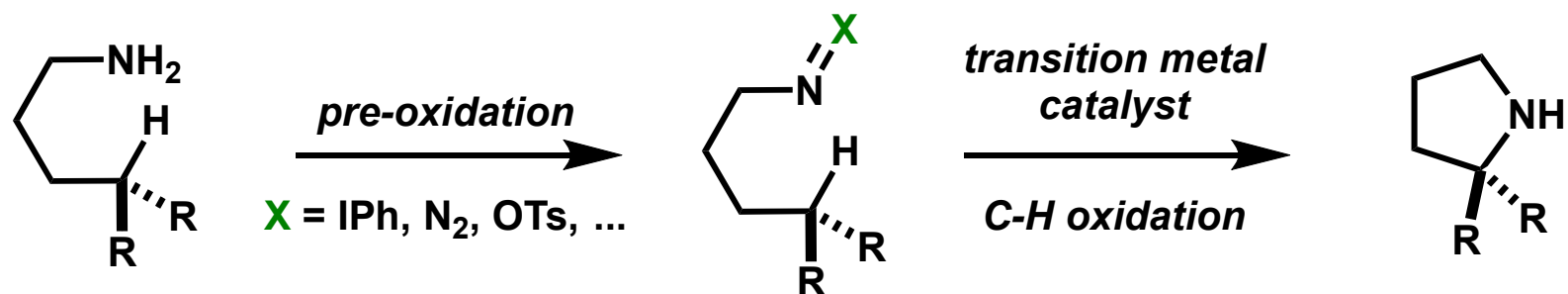


2. Methane functionalization

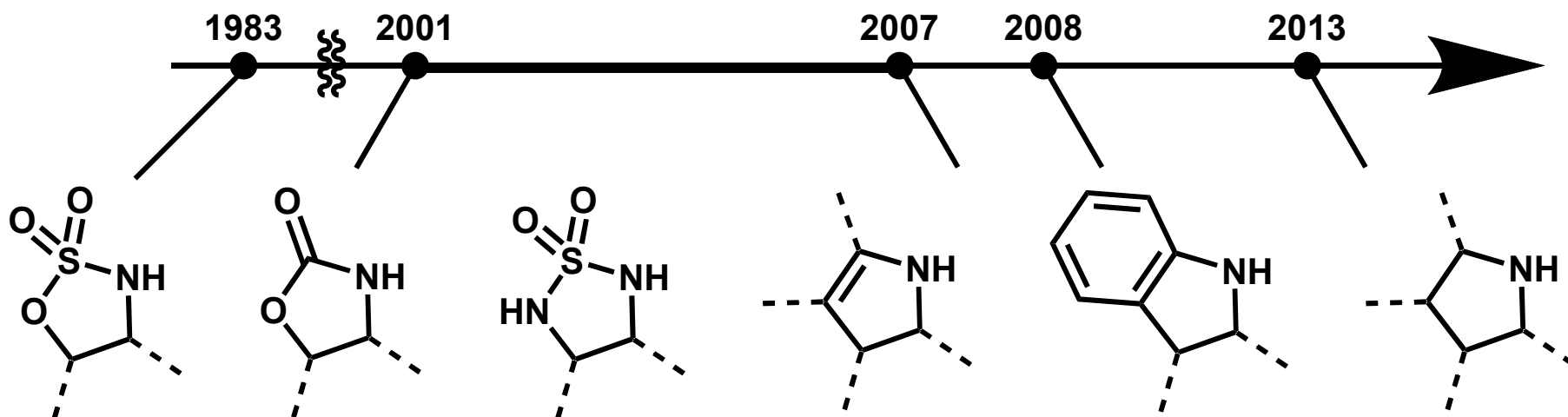
3. Catalytic selective defunctionalization

C-H Amination

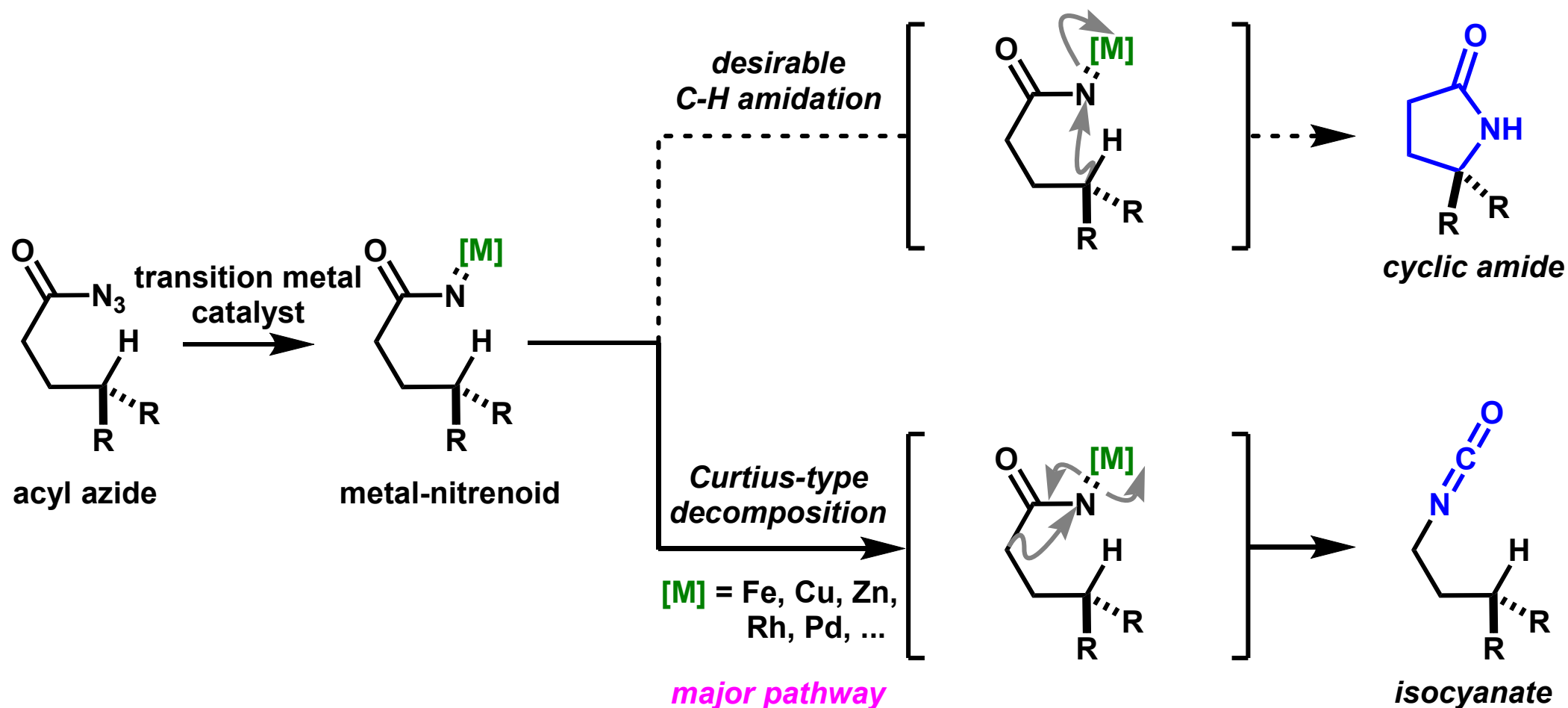
Intramolecular insertion of nitrenes into carbon-hydrogen bonds



history of C-H amination



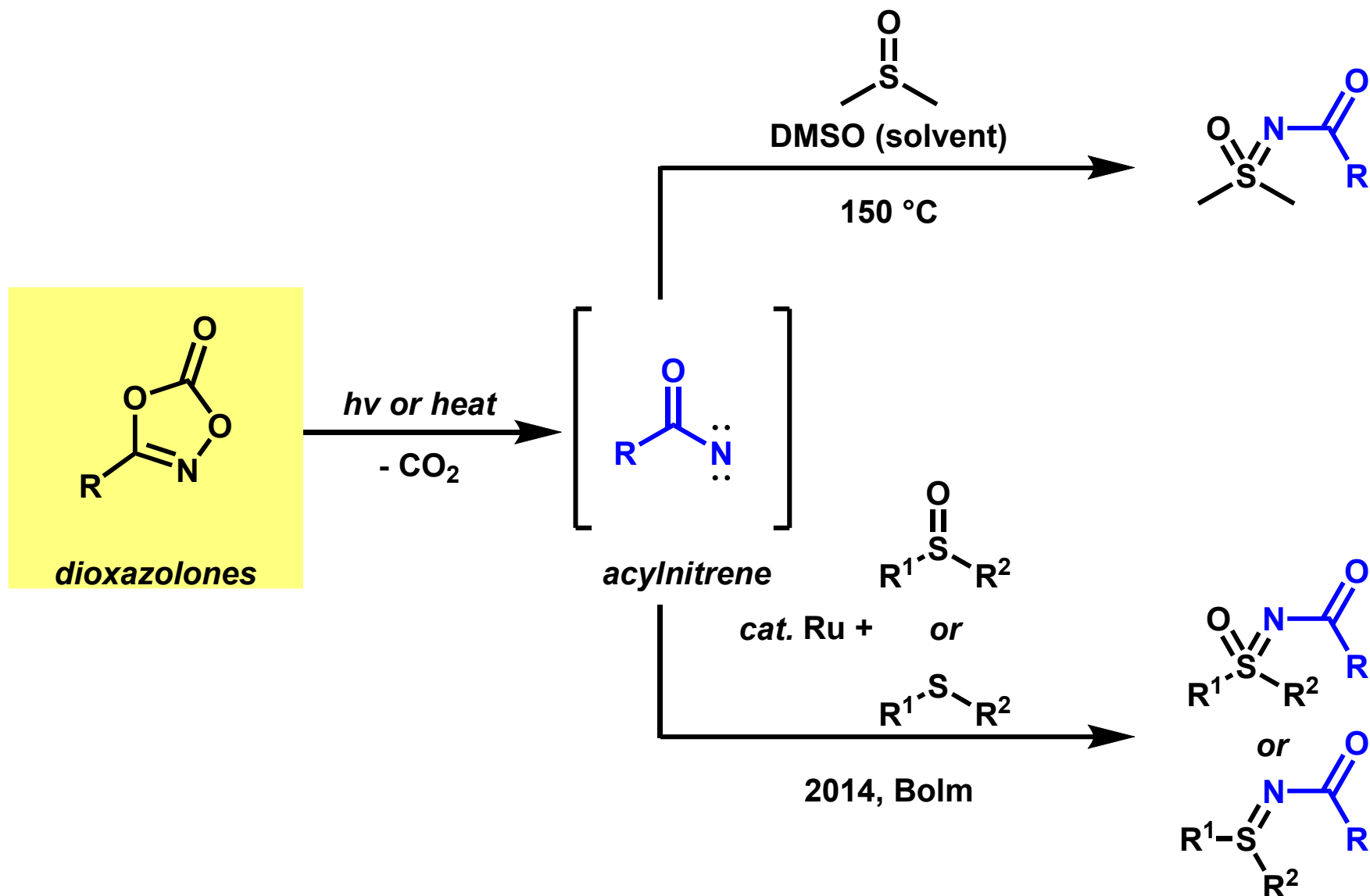
Why Is C-H Amidation Challenging?



The instability of acyl azides makes it difficult to construct cyclic amides via C-H functionalizations

Alternatives of Acyl Azides ~Dioxazolones~

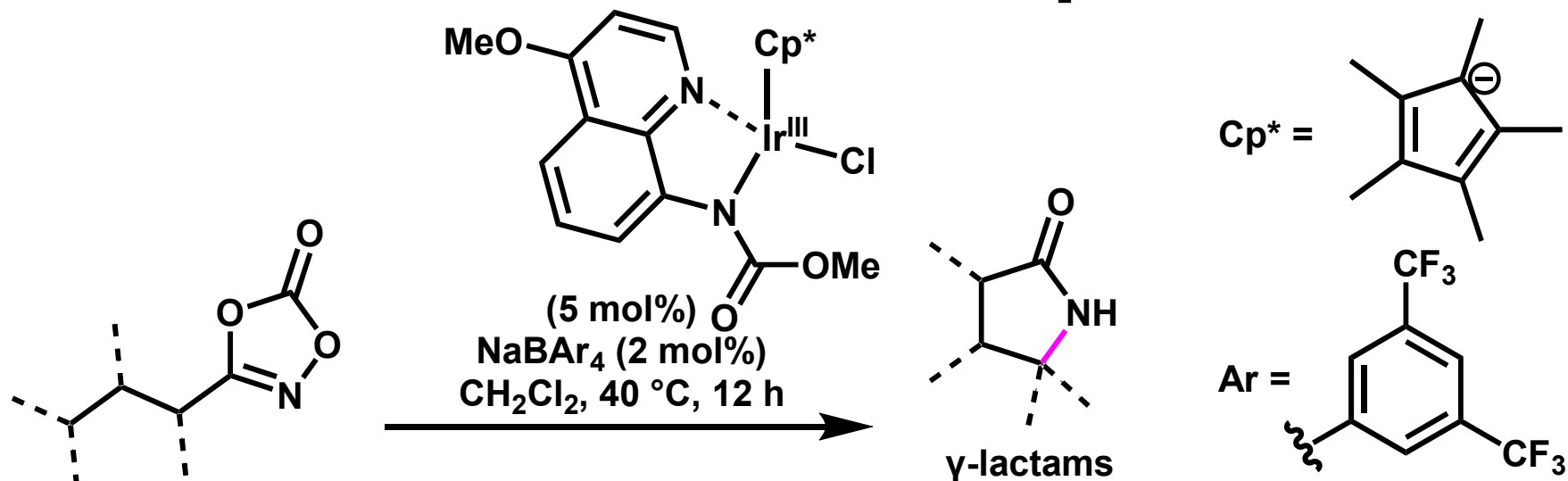
1968, Sauer and Mayer



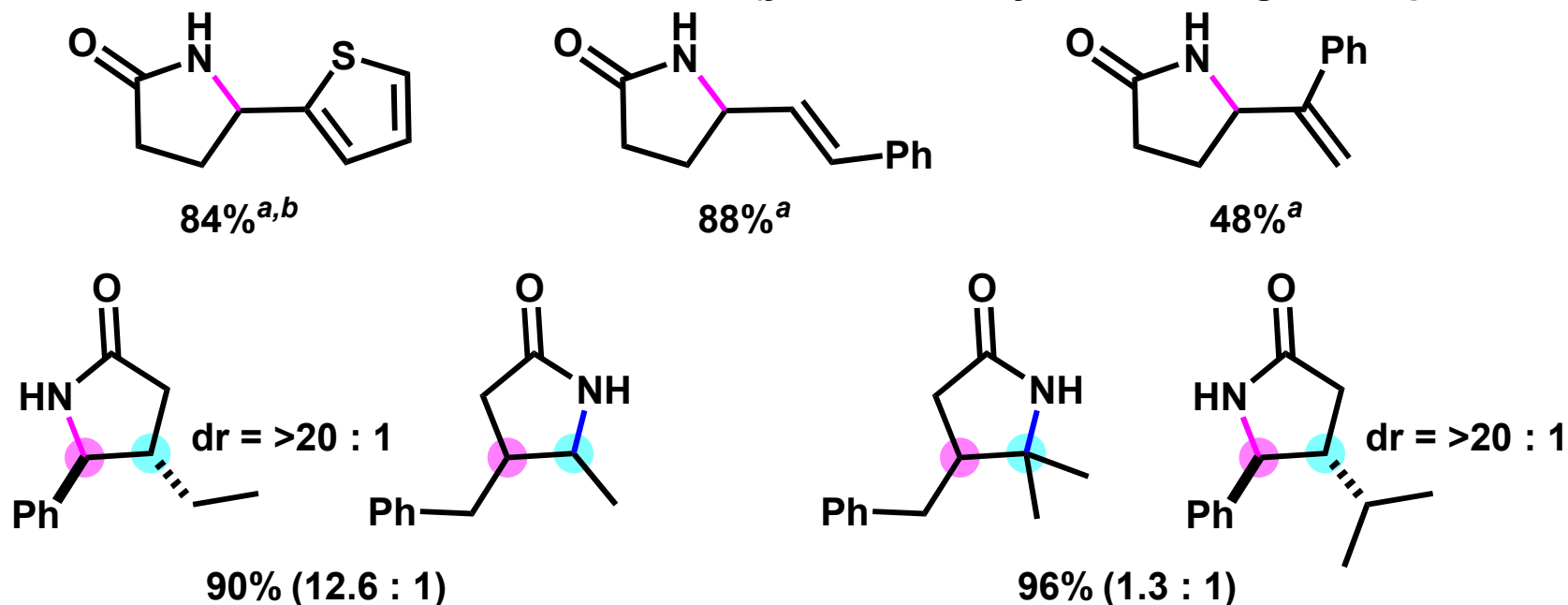
Inspired by these works, Prof. Sukbok Chang used dioxazolones as versatile amidating reagents.

1. Park, Y.; Park, K. T.; Kim, J. G.; Chang, S. *J. Am. Chem. Soc.* **2015**, *137*, 4534.
2. Sauer, J.; Mayer, K. K. *Tetrahedron Lett.* **1968**, *9*, 319.
3. Bizet, V.; Burglioni, L.; Bolm, C. *Angew. Chem. Int. Ed.* **2014**, *53*, 5639.

Substrate Scope



(yields with acyl azides are given in parentheses)



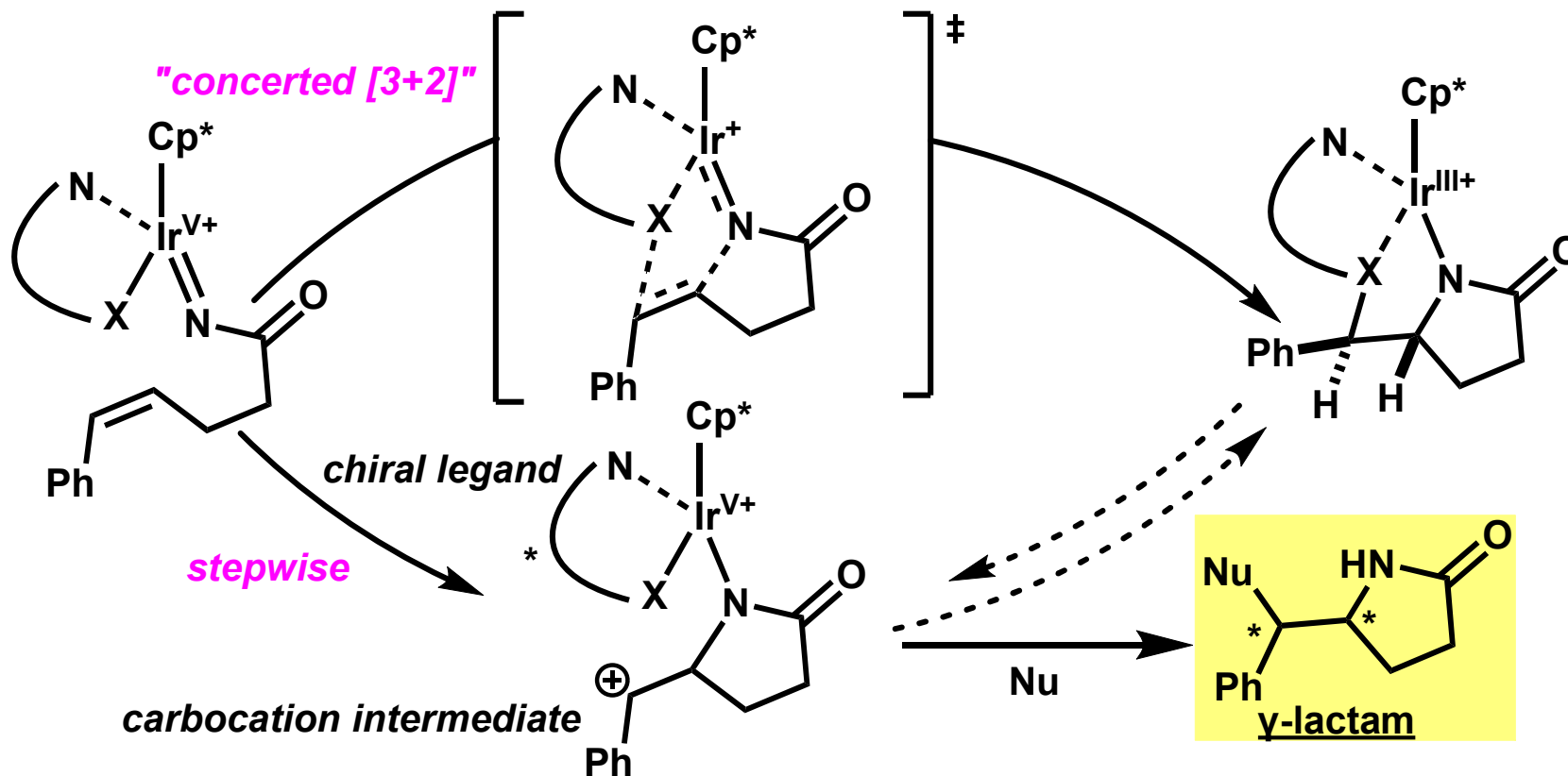
See also LS_191026_Tsukasa_Shimakawa

a) 10 mol% of the catalyst was used. b) 80 °C for 48 hours

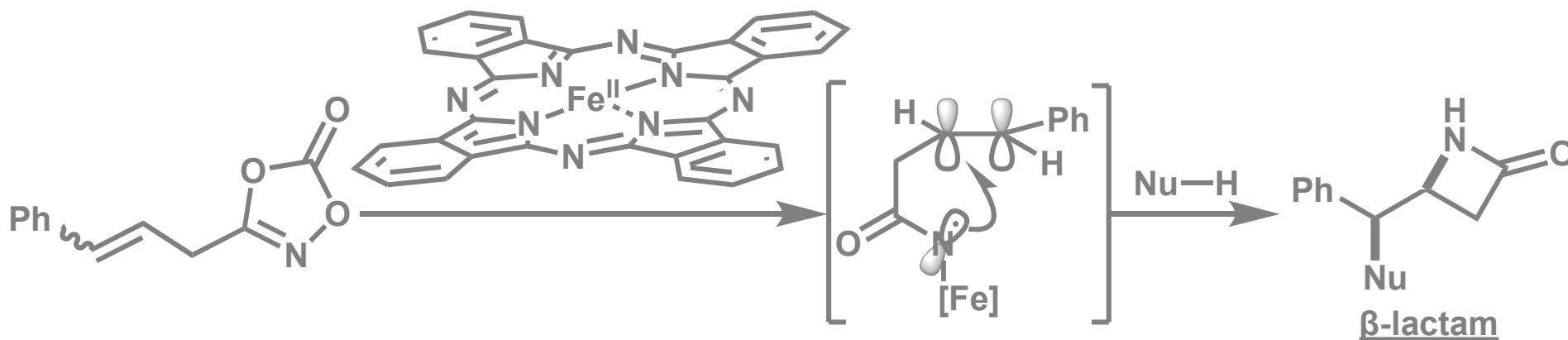
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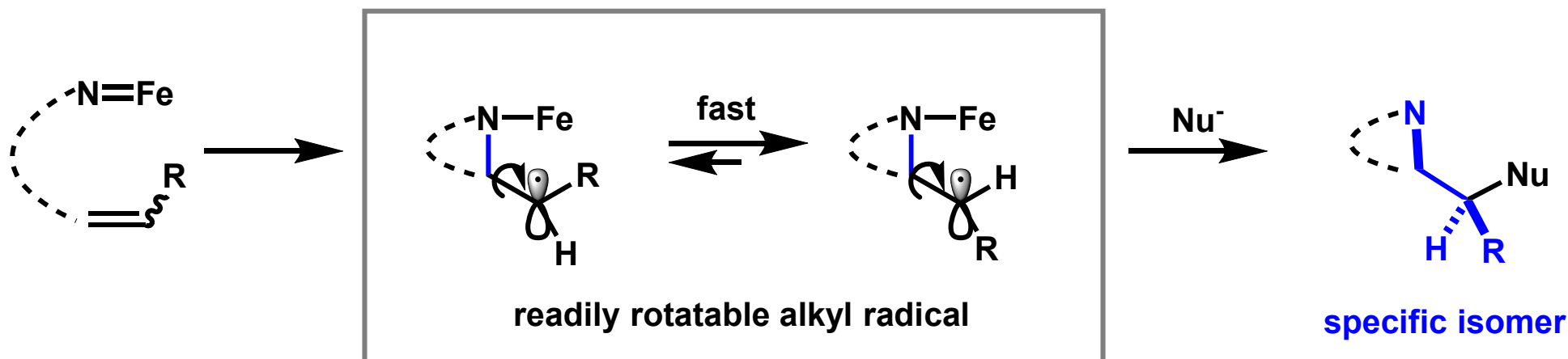
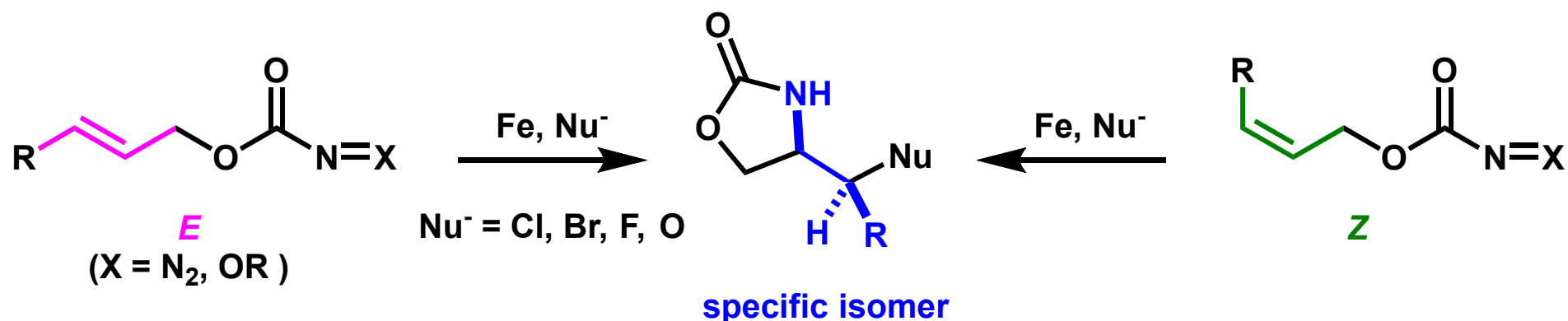
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Olefin Difunctionalization via a Nitrenoid Transfer

- a powerful tool for forging versatile aminated molecules with the installation of an additional functional group at the C=C bonds in a single operation.

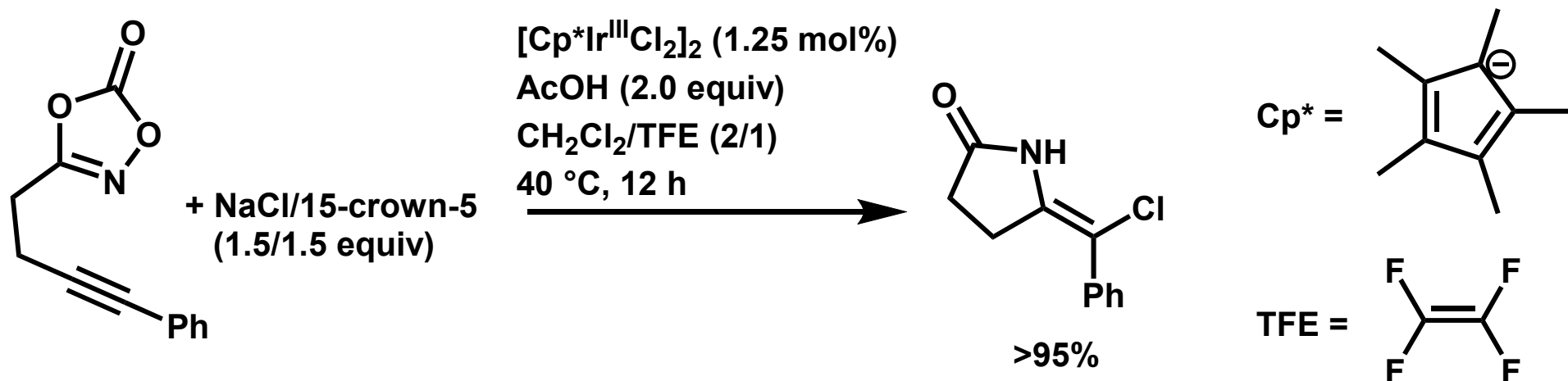
Previous studies: *stereoconvergent* nitrene transfer to olefins (Bach, Xu)



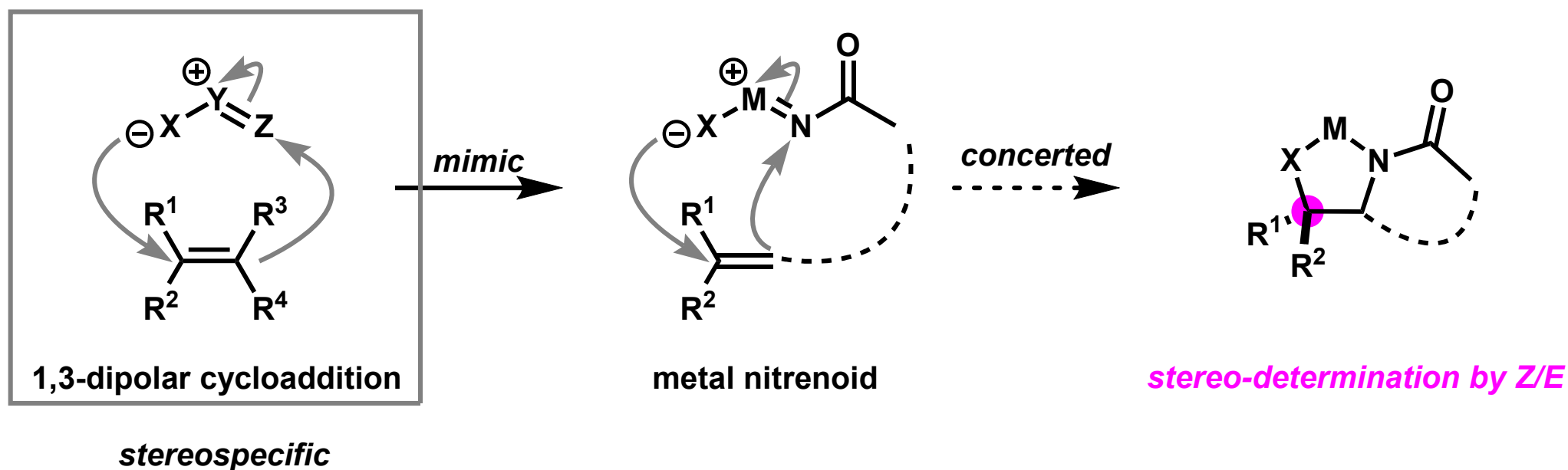
1. Hong, S. Y.; Chang, S. *J. Am. Chem. Soc.* **2019**, *141*, 10399. 2. (a) Kluegge, J.; Herdtweck, E.; Bach, T. *Synlett* **2004**, 2004, 1199. (b) Liu, G. -S.; Zhang, Y. -Q.; Yuan, Y. -A.; Xu, H *J. Am. Chem. Soc.* **2013**, *135*, 3343.

Working Hypothesis

previous work: Ir(III)-Catalyzed Haloamidation of Alkynes Enabled by Ligand Participation

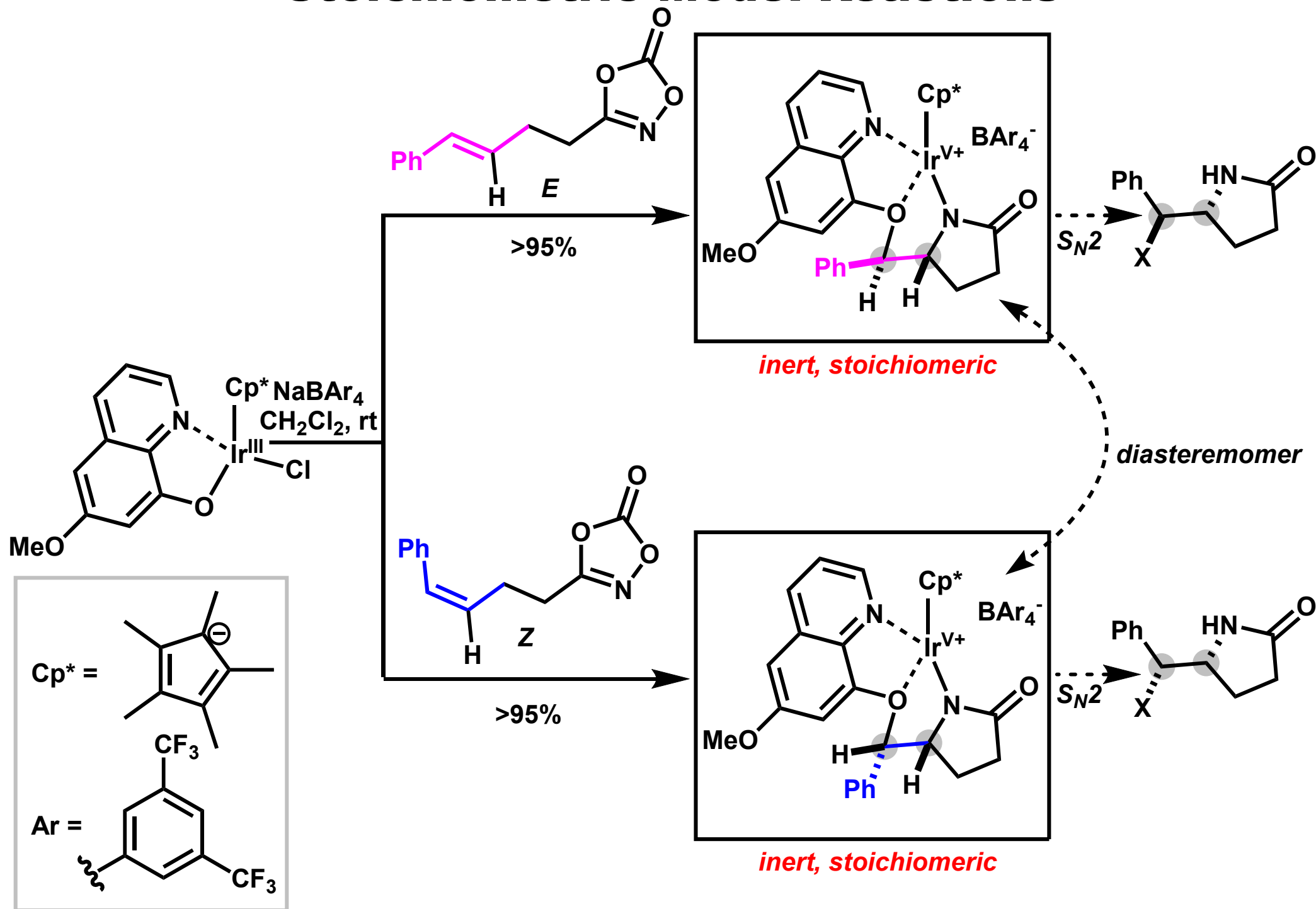


metal nitrenoids as a motif of 1,3-dipoles



1. Hong, S. Y.; Son, J.; Kim, D.; Chang, S. *J. Am. Chem. Soc.* **2018**, 140, 12359. 2. Hong, S. Y.; Chang, S. *J. Am. Chem. Soc.* **2019**, 141, 10399.

Stoichiometric Model Reactions

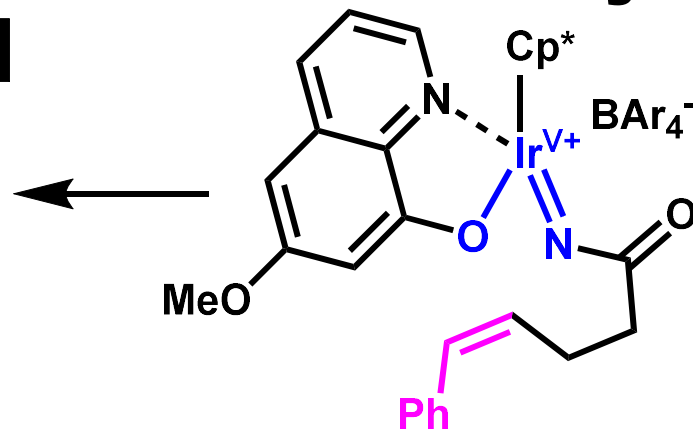


FMO Analysis

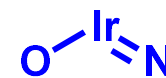
"dipolarophile"



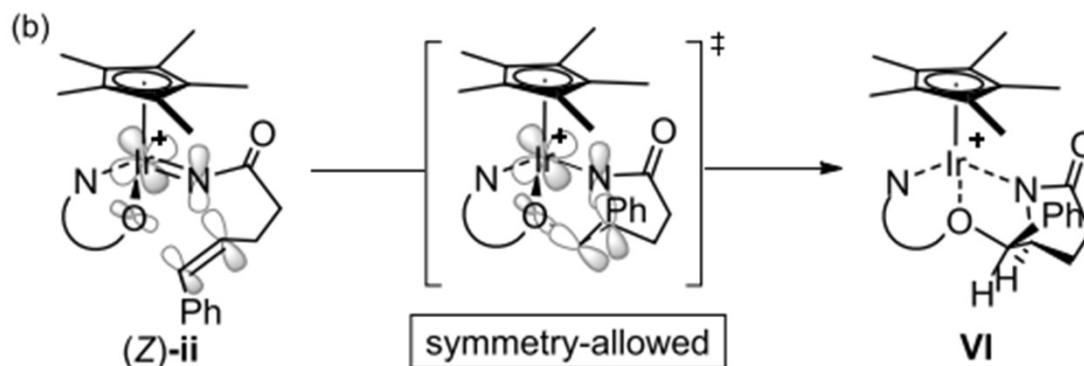
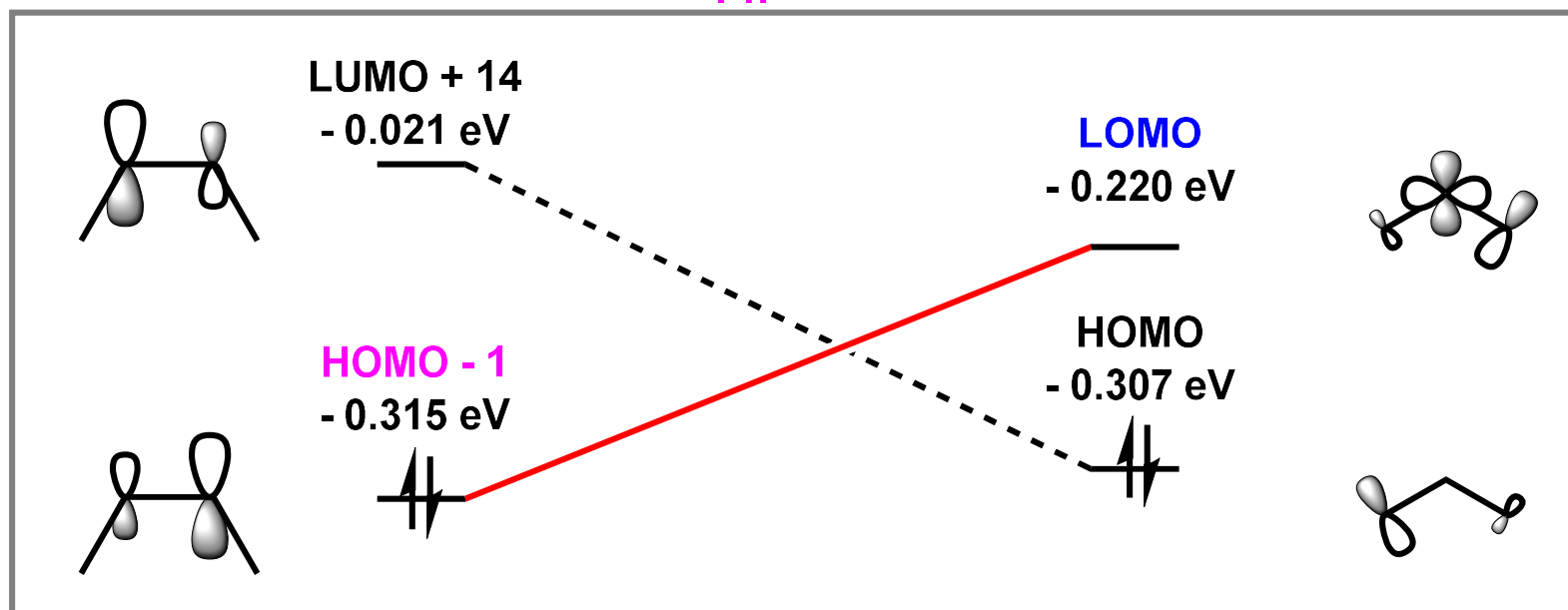
(alkene)



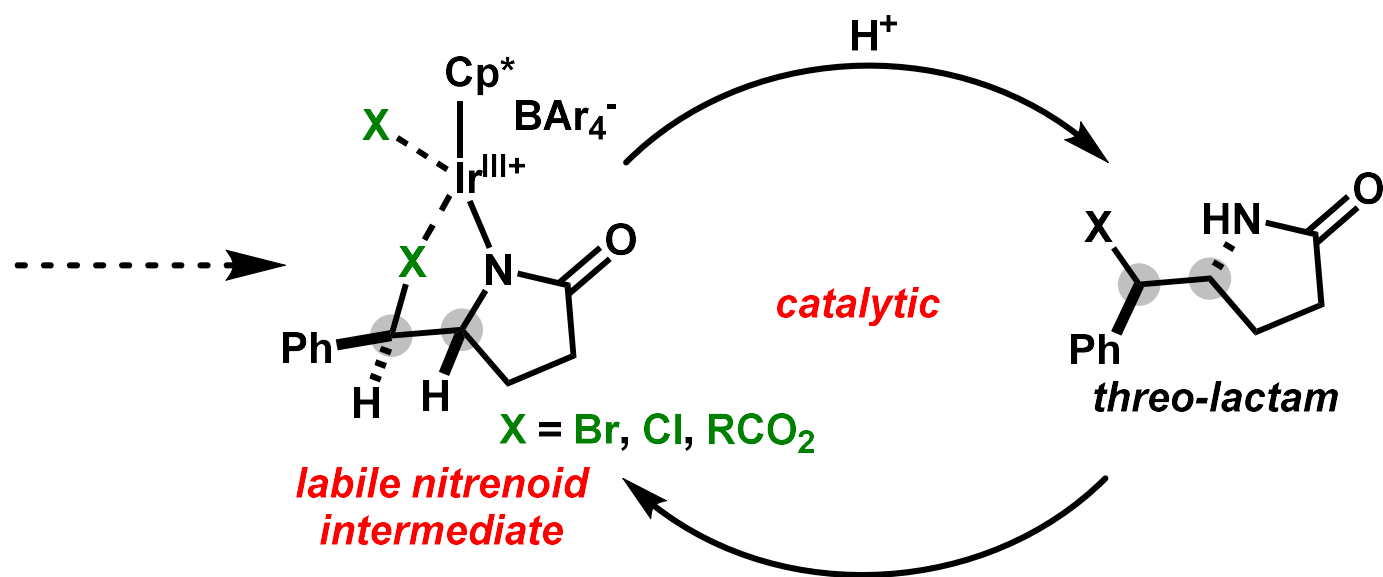
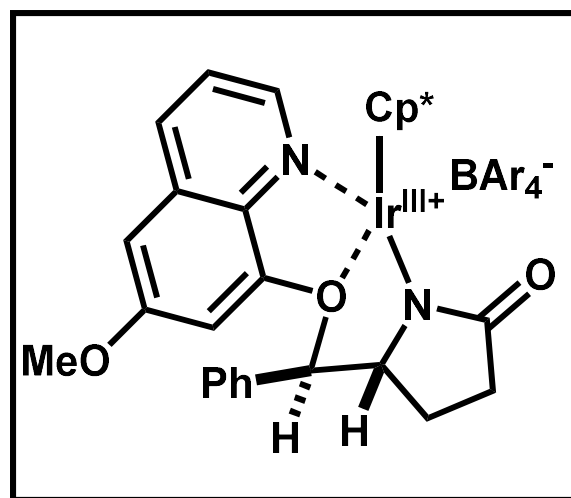
"1,3-dipole"



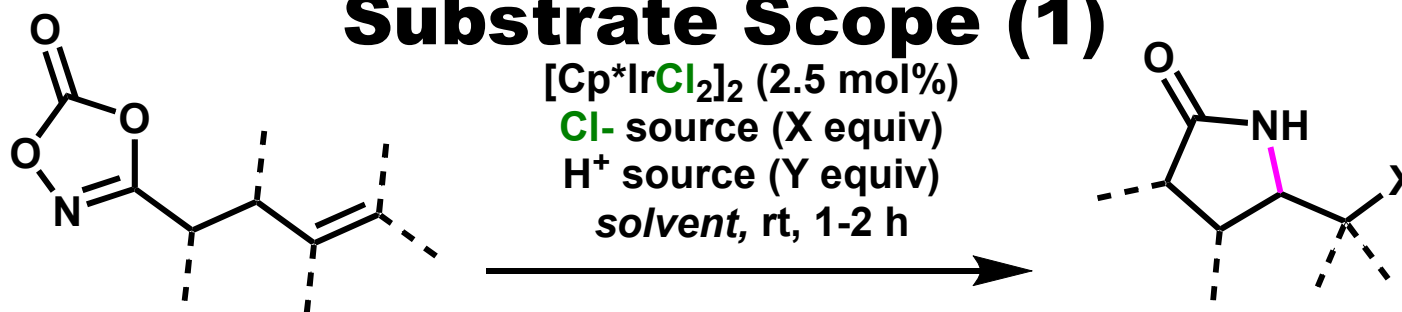
(Ir-nitrenoid)



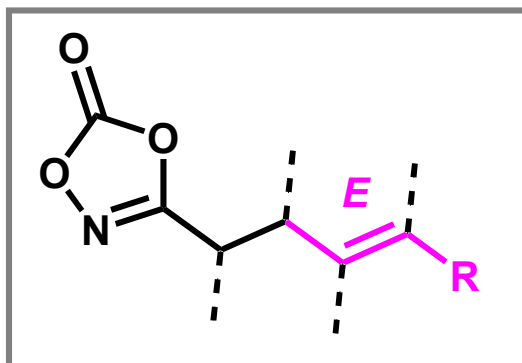
Catalyst Design (1)



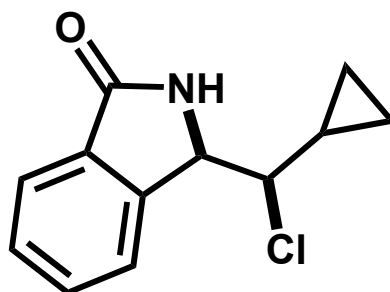
Substrate Scope (1)



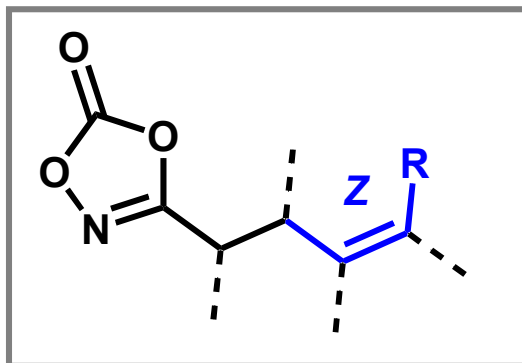
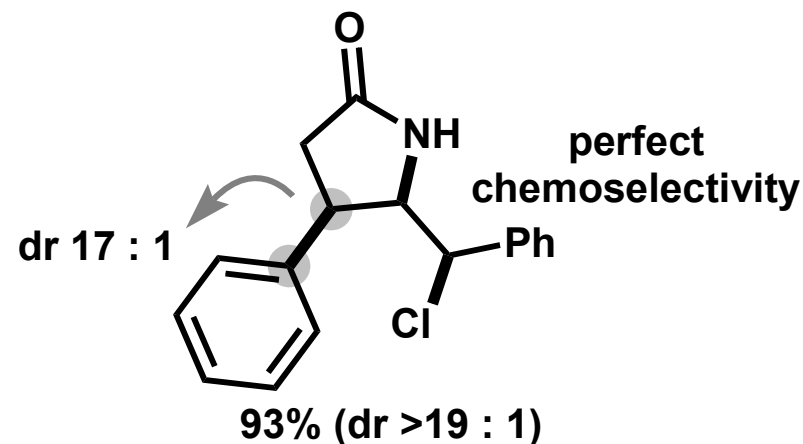
A slight erosion in deastereoselectivity was accompanied in these cases mainly due to the olefin isomerization induced by the acid.



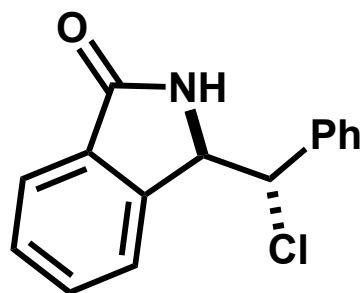
HCl (X = Y = 1.0)
 $\text{CH}_2\text{Cl}_2/\text{TFE}$ (1/1)



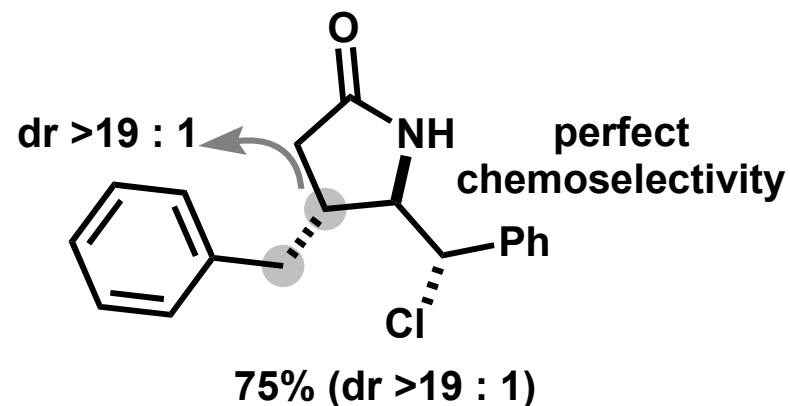
72%^b (dr 9.1 : 1)



$\text{NaCl}/15\text{-crown-5}$ (X = 1.5)
 HOAc (Y = 2.0), HFIP



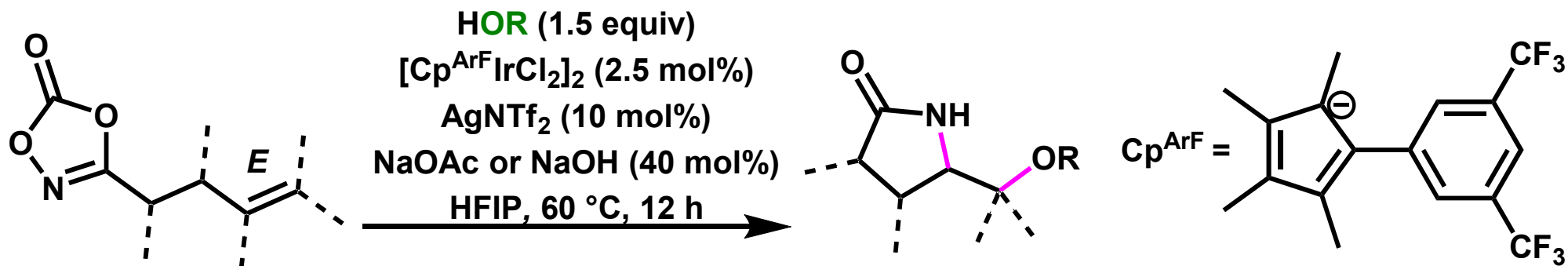
80% (dr 15 : 1)



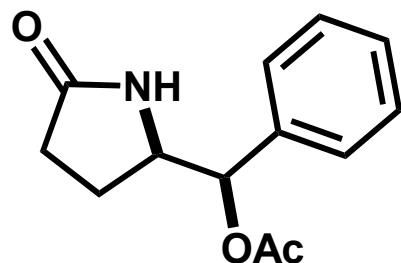
a) Ratio of diastereomers, determined by ^1H NMR analysis of the crude reaction mixture.

b) 5 mol% of $[\text{Cp}^*\text{Ir}^{\text{III}}\text{Cl}_2]_2$ was used.

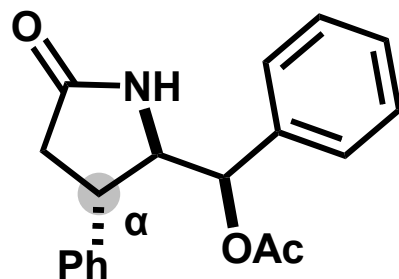
Substrate Scope (2)



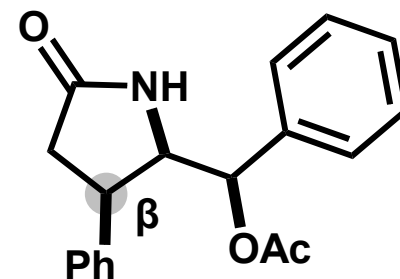
variation of dioxazolones



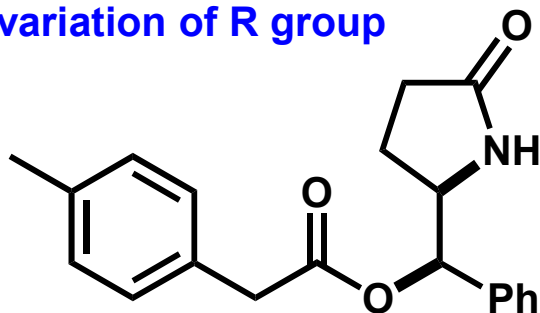
90% (dr >19 : 1)



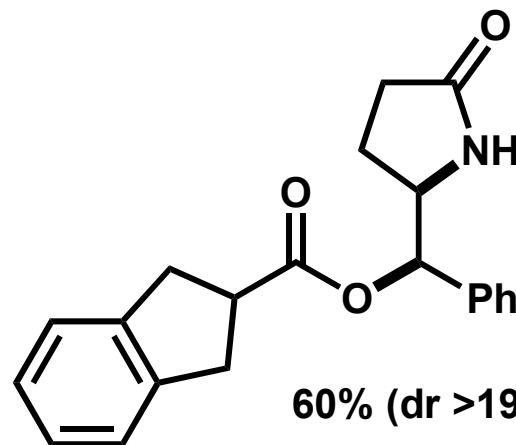
80% ($\alpha : \beta = 3 : 1$, dr = >19 : 1)



variation of R group



55% (dr >19 : 1)



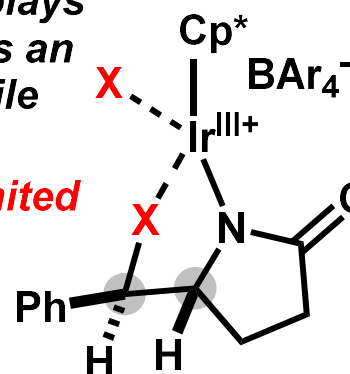
60% (dr >19 : 1)

Limitations and Solutions

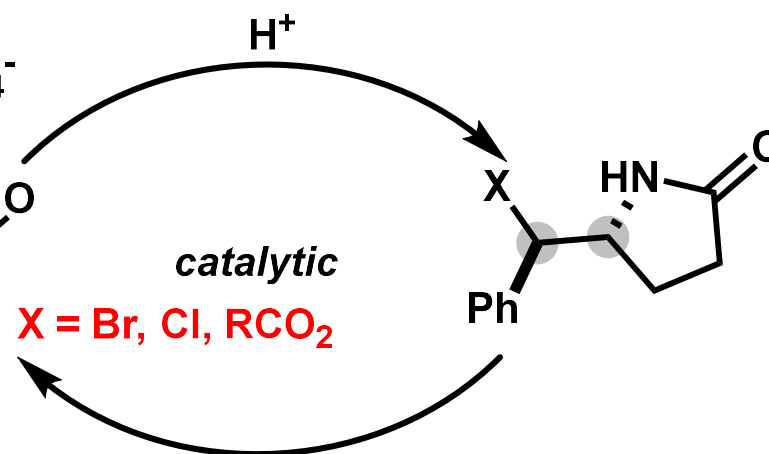
limitations:

An auxiliary ligand plays an additional role as an internal nucleophile

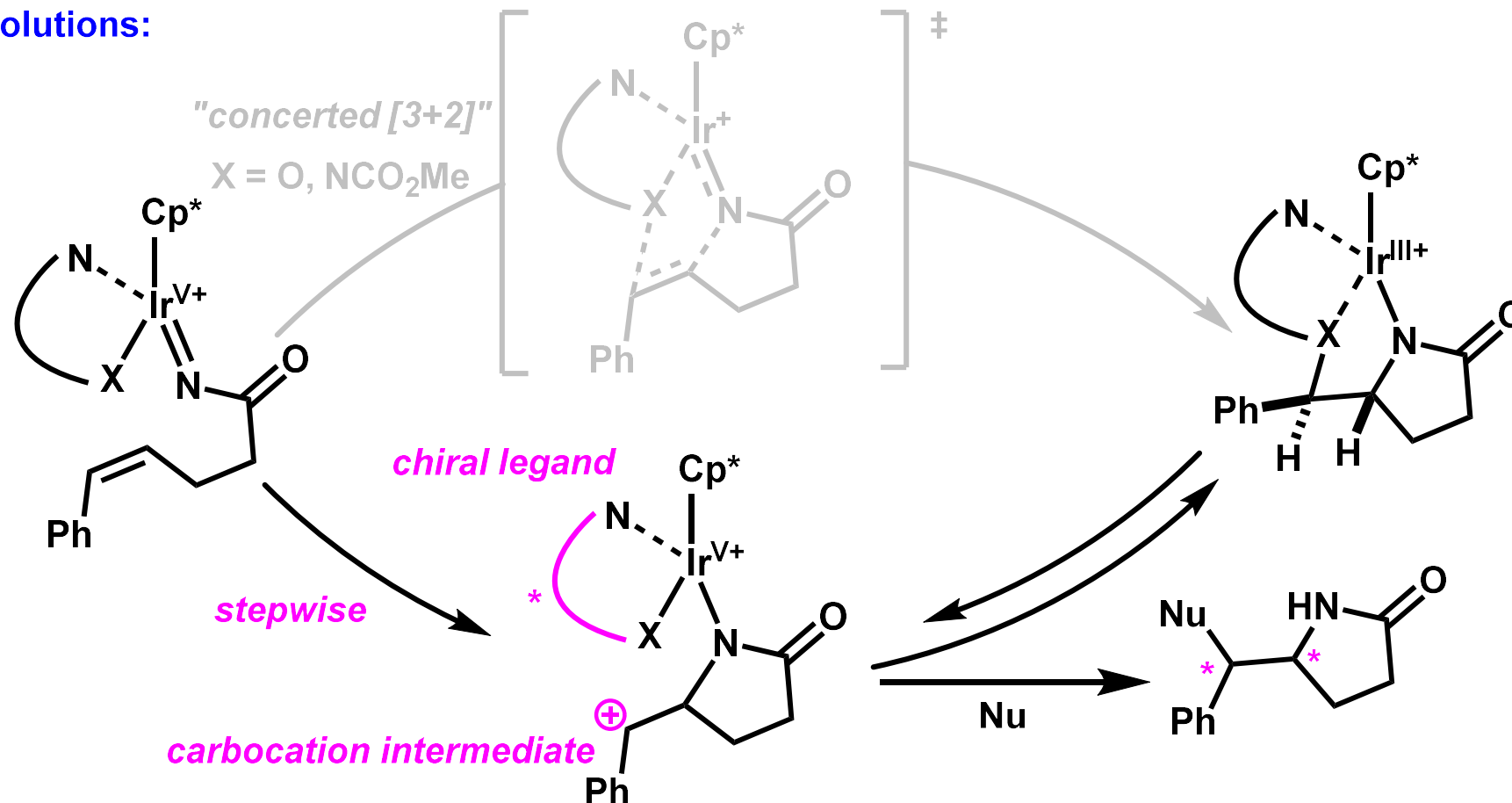
= Variations of nucleophiles are limited



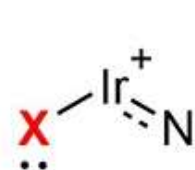
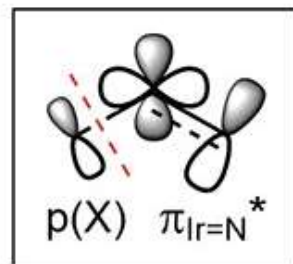
no enantioselectivities



solutions:



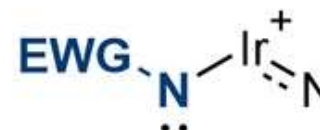
Catalyst Design (2): Electron-withdrawing Group



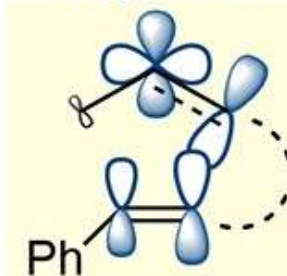
X-type variations



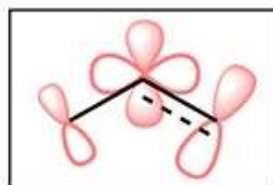
"Low lying $p(X)$ "



stepwise

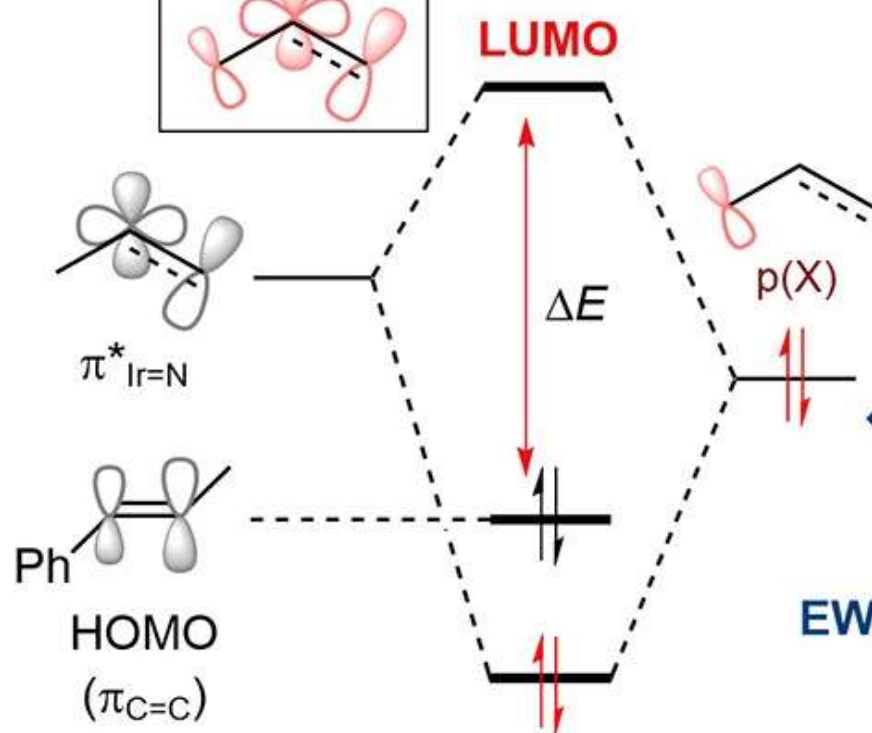
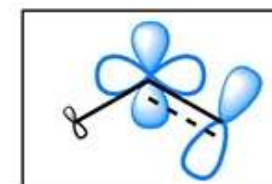


three-centered

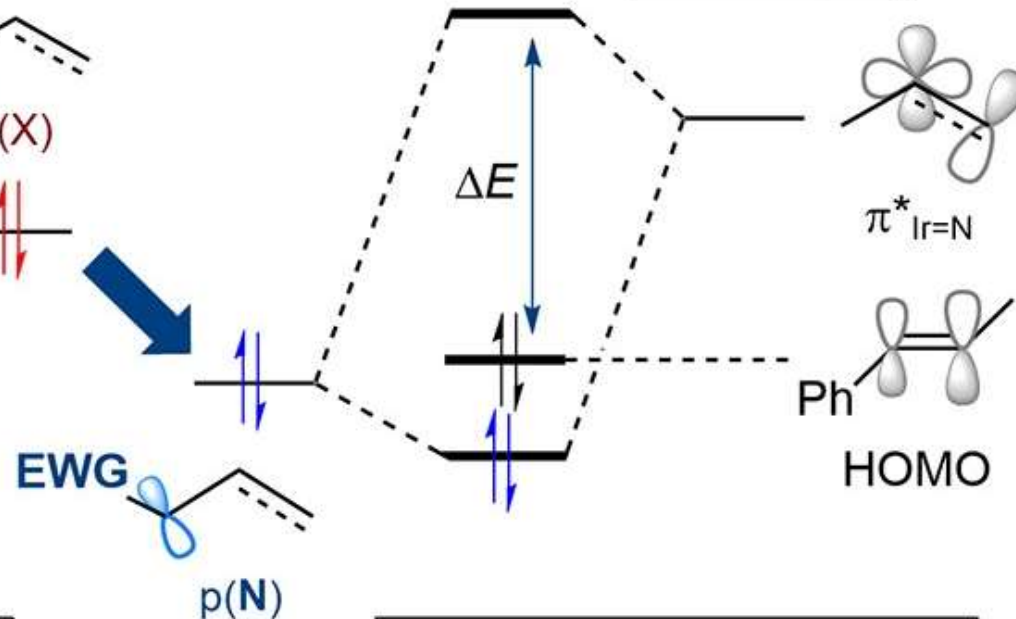


vs.

symmetry-controlled

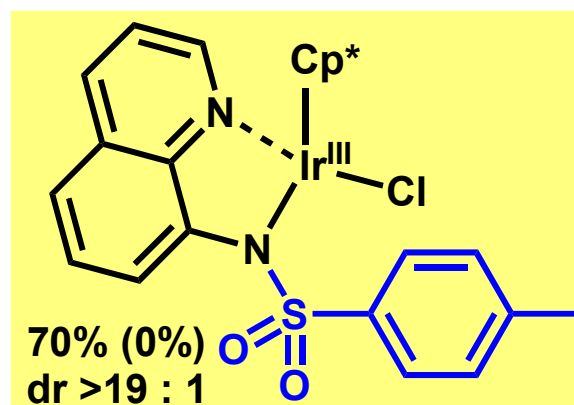
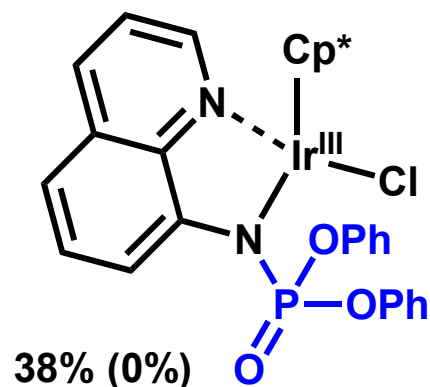
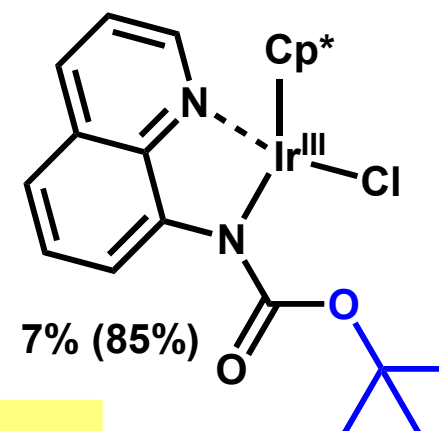
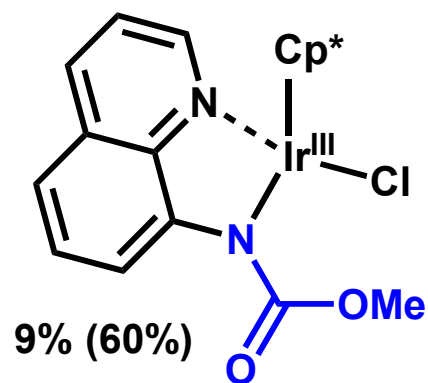
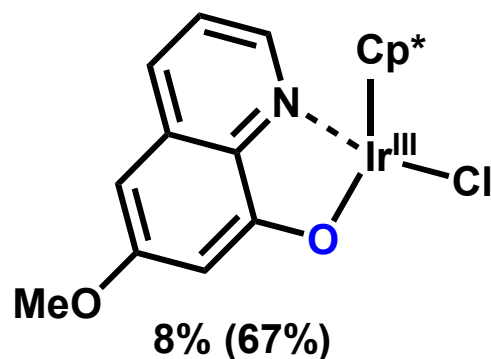
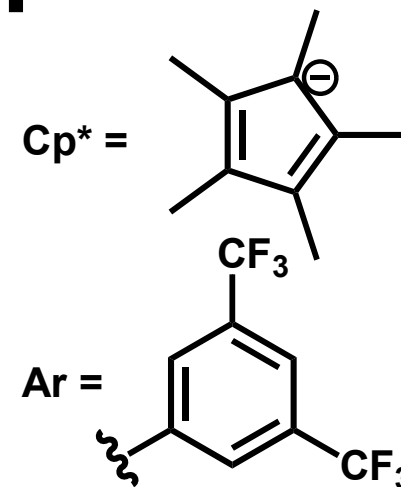
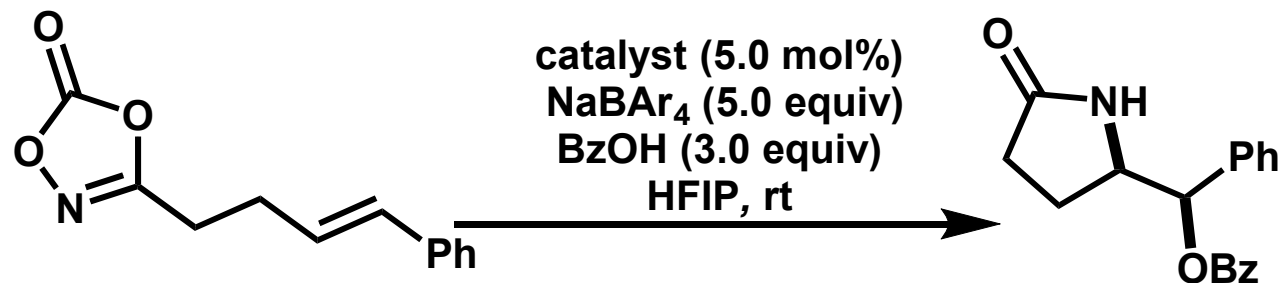


larger HOMO/**LUMO** gap



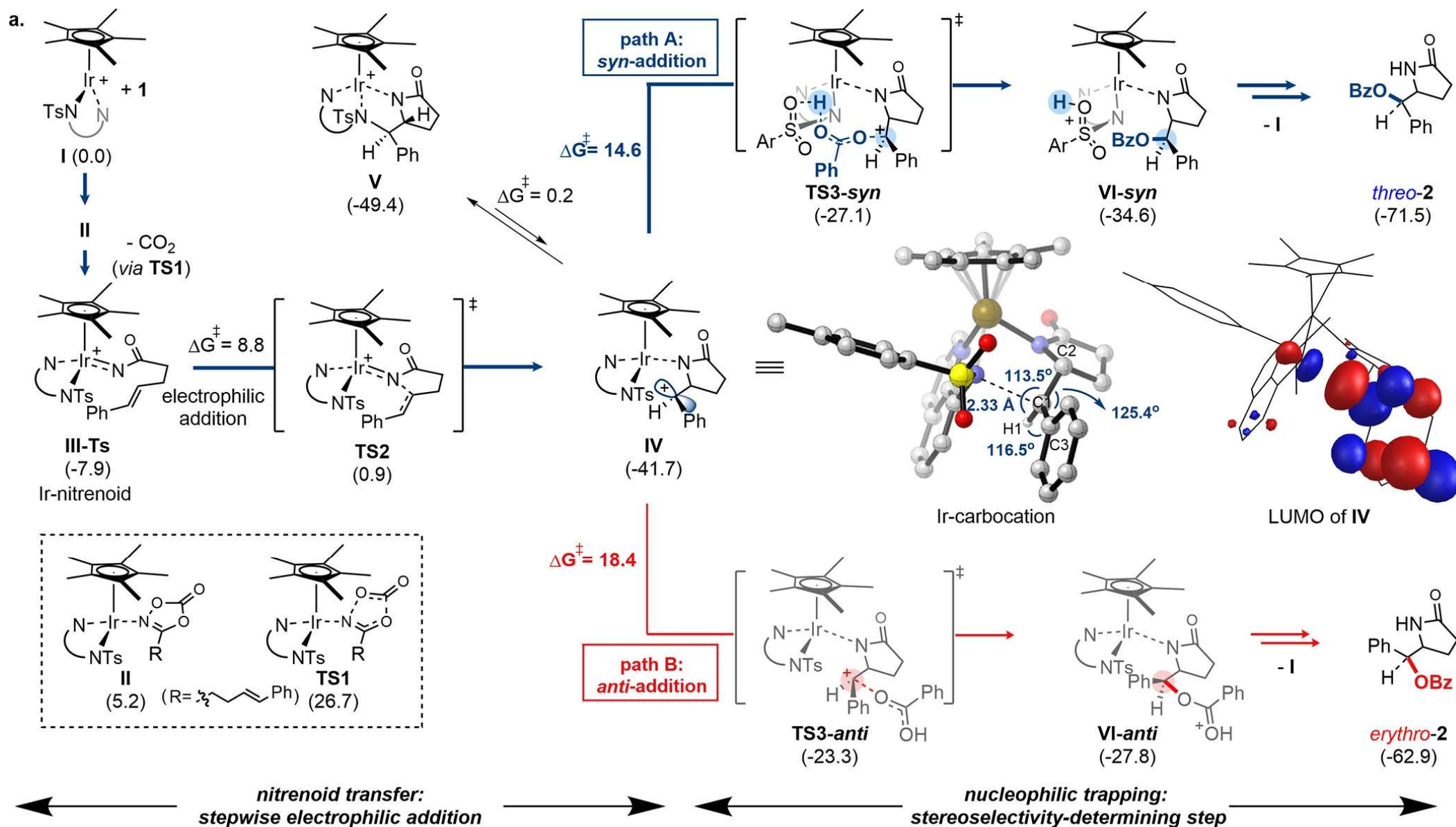
smaller HOMO/**LUMO** gap

Catalyst Optimization



Yields were based on a ^1H NMR analysis of the crude reaction mixture using 1,1,2,2-tetrachloroethane as an internal standard.

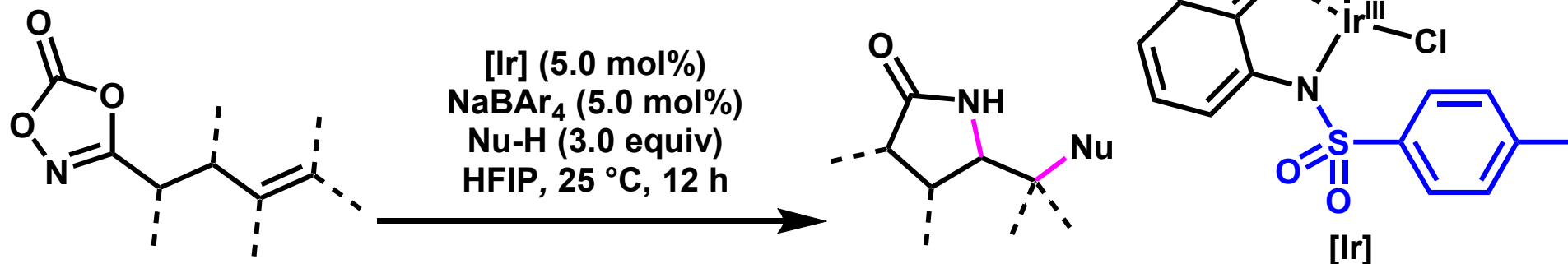
Computational Modeling



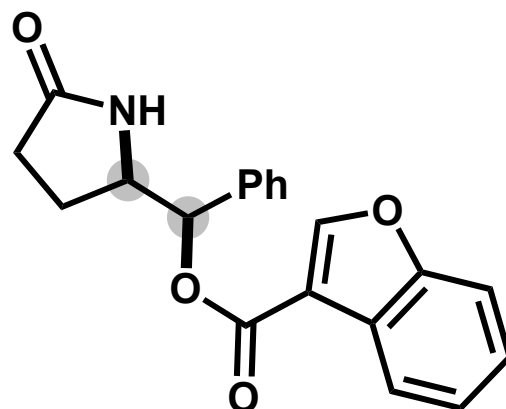
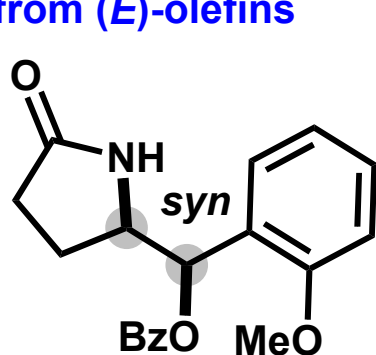
PCM(HFIP)-M06/SDD+6+6-311++G**//M06/LanI2dz+6-31G**

A hydrogen bond between a sulfonyl oxygen atom of the catalyst and an O-H hydrogen of benzoic acid plays an important role for determining a diastereoselectivity.

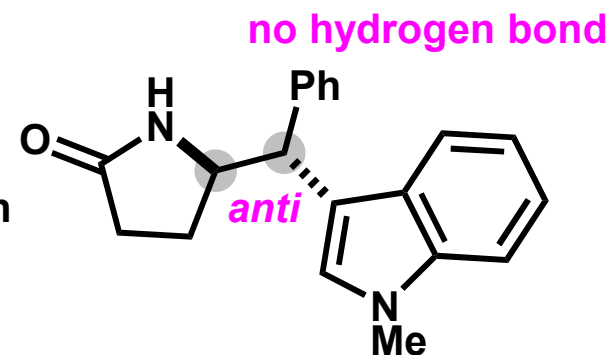
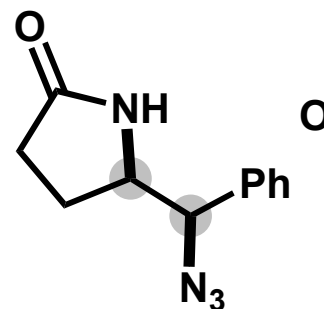
Substrate Scope (3)



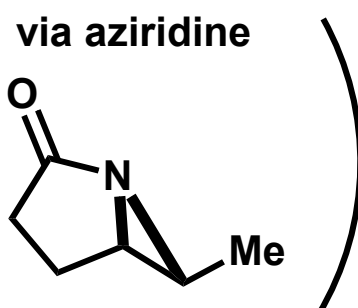
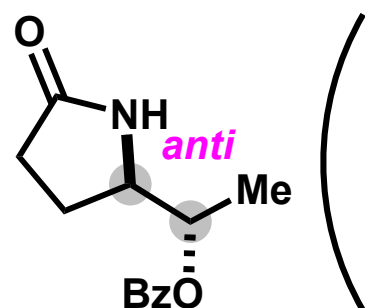
from (*E*)-olefins



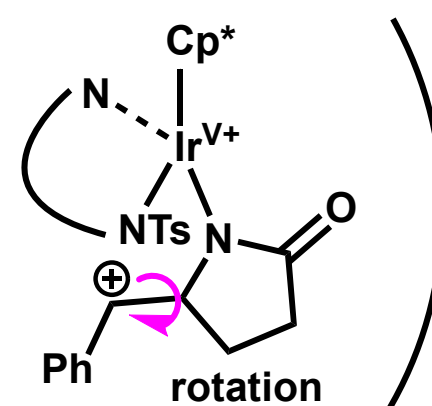
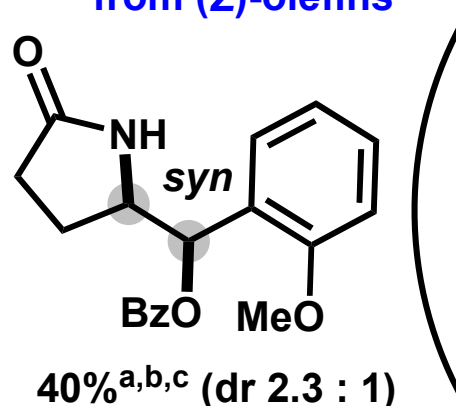
other nucleophiles



carbocation without resonance stabilization

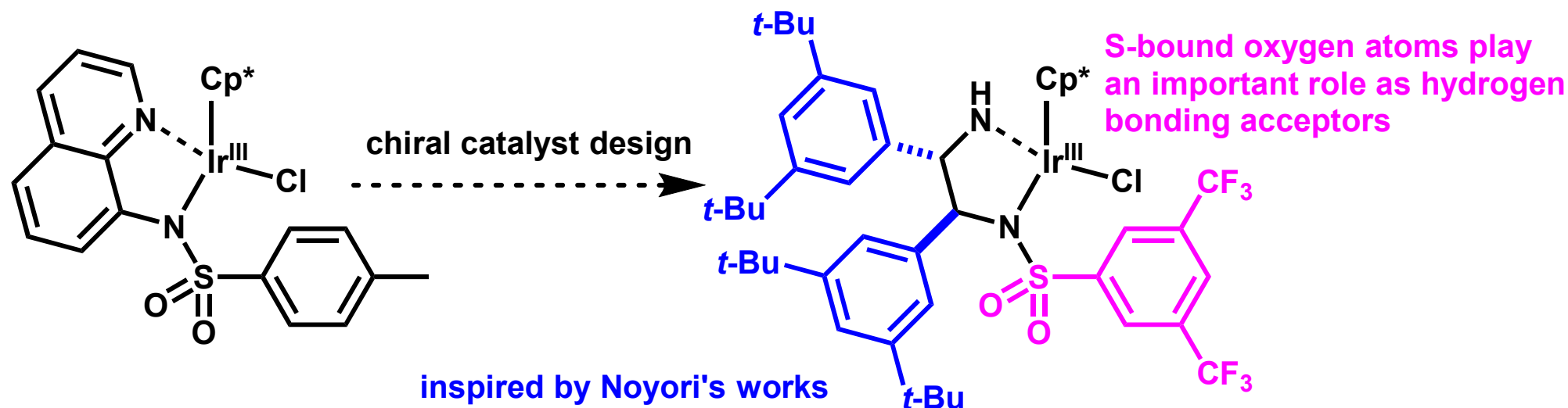


from (*Z*)-olefins

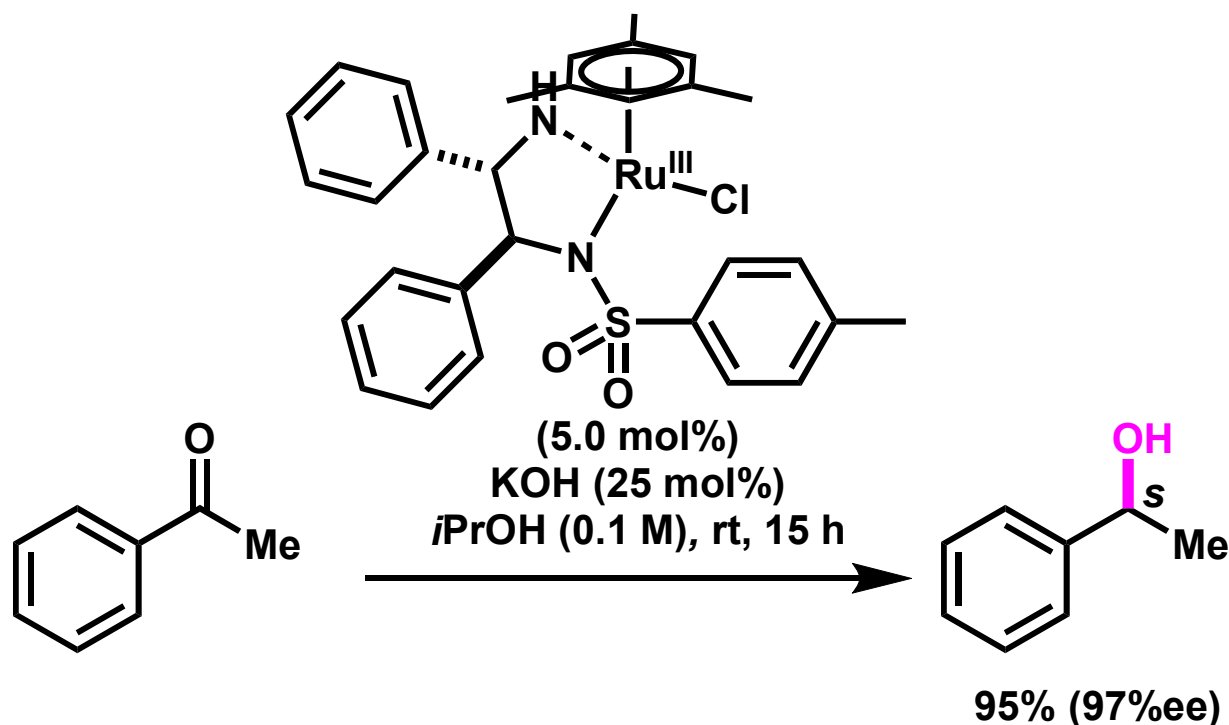


a) [Ir] (10 mol%) and NaBAR₄ (10 mol%) were used. b) NFTB solvent was used. c) Run at 60 °C.

Development of Asymmetric Oxyamidation

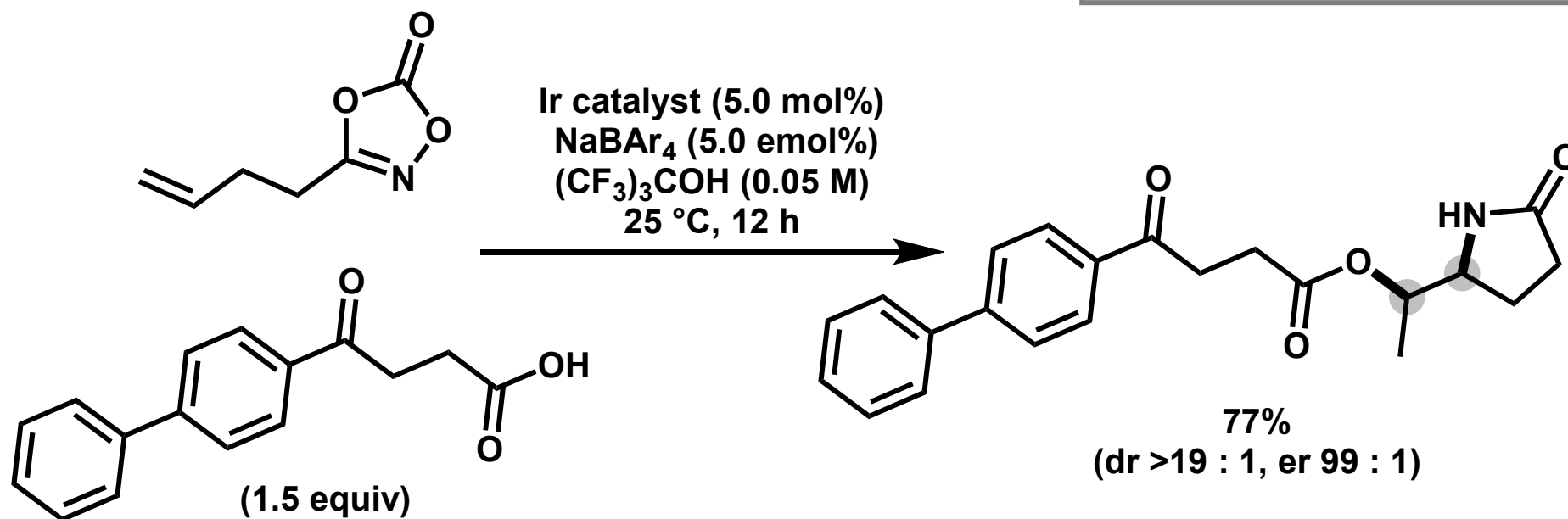
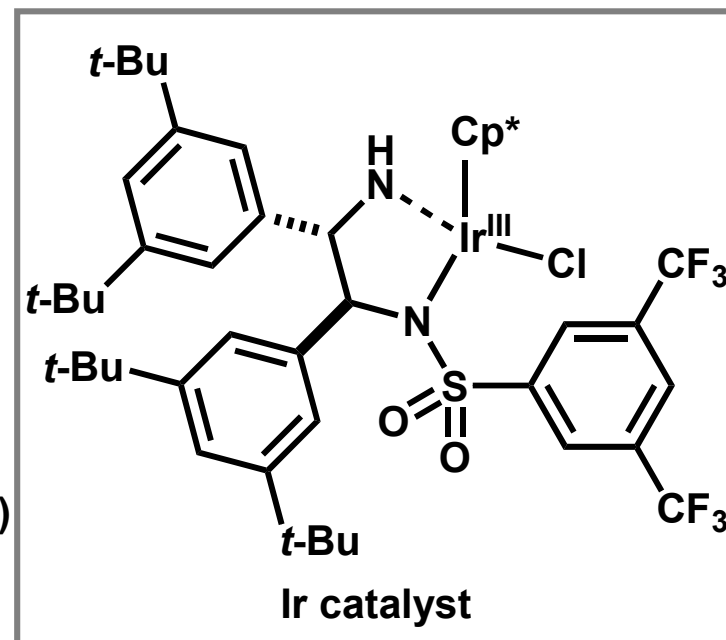
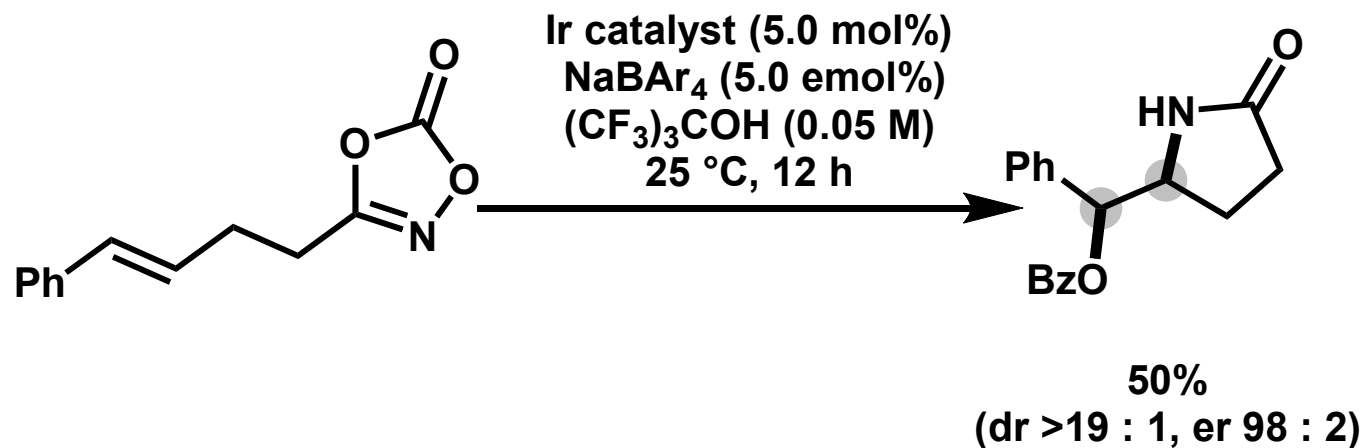


Asymmetric hydrogenation reaction by the Noyori group²



1. Kim, S.; Kim, D.; Hong, S. Y.; Chang, S. *J. Am. Chem. Soc.* **2021**, 143, 3993.
2. Hashiguchi, S.; Fujii, A.; Takehara, J.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1995**, 117, 7562.

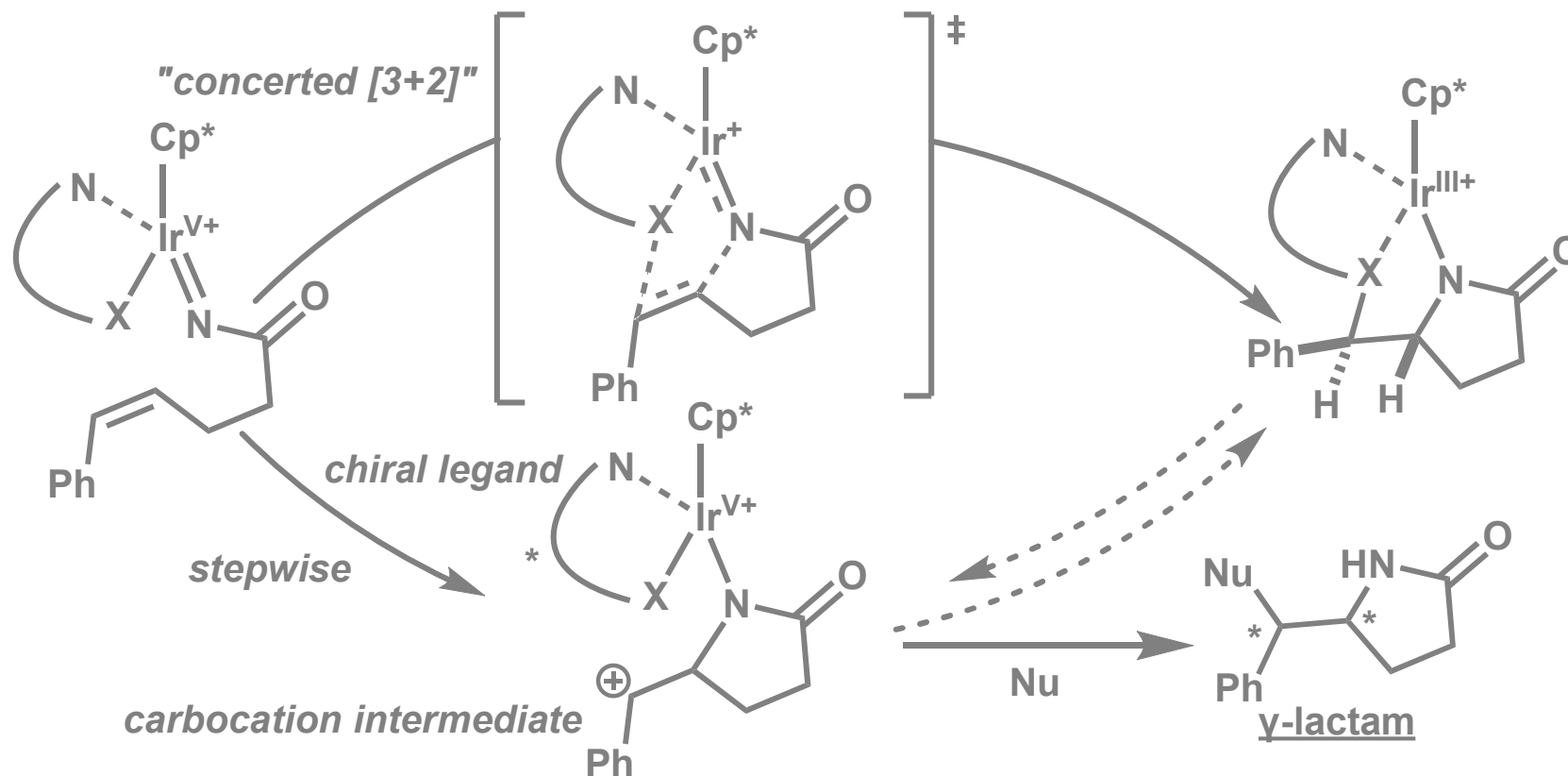
Substrate Scope (4)



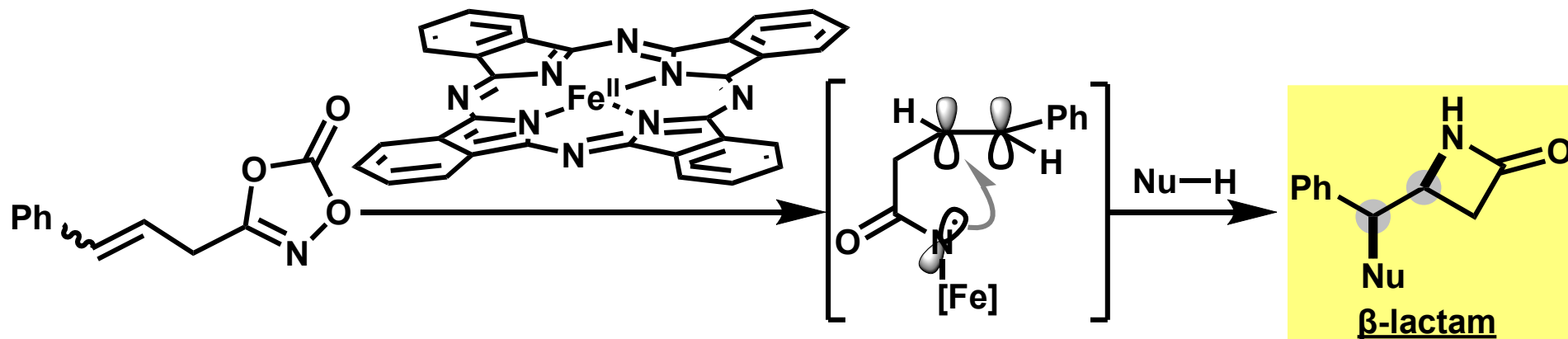
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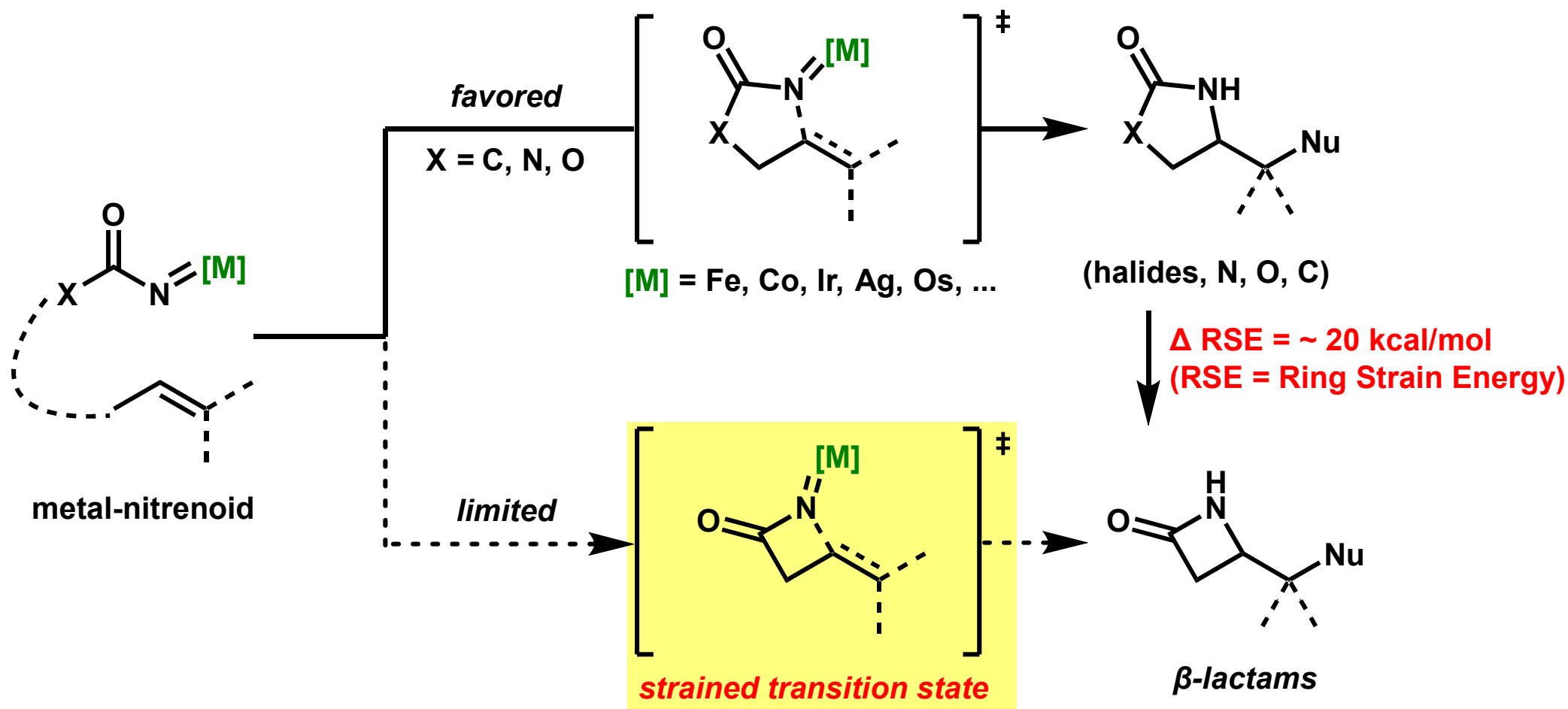
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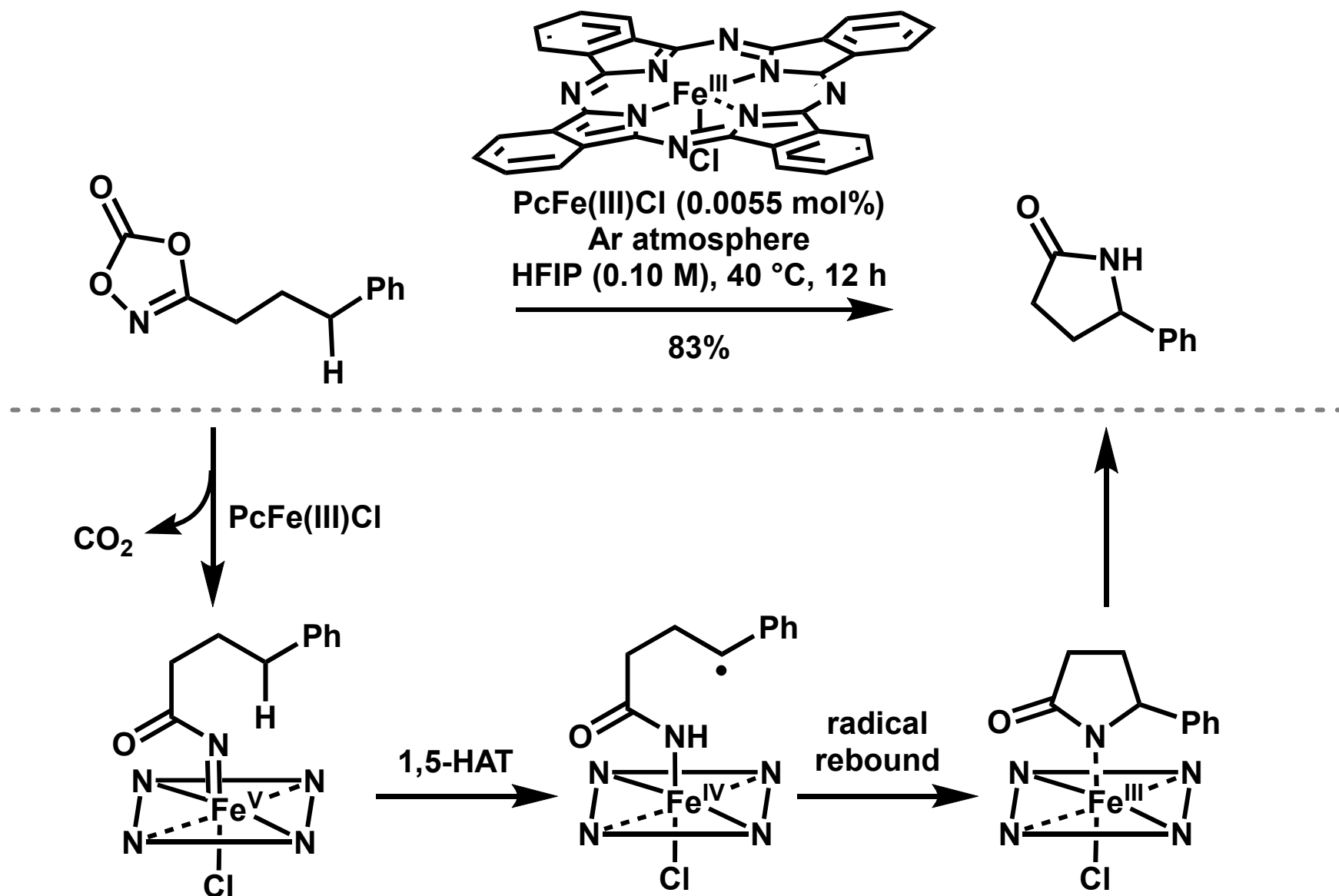


Access to β -Lactams via Metal-Catalyzed Olefin Difunctionalization Approaches



→ How to realize this challenging conversion?

Iron Catalyst System for Intramolecular C-H Amidation

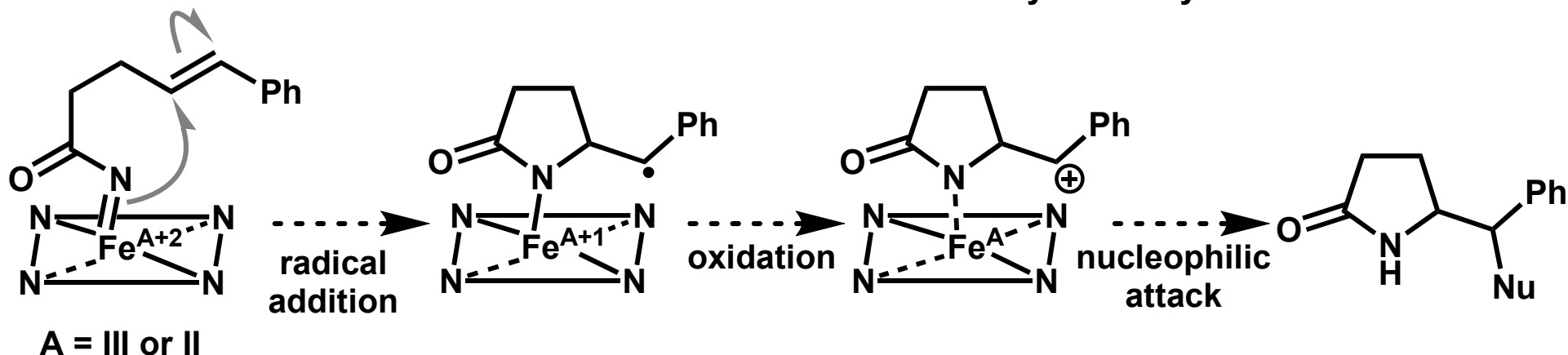


→Radical reaction is a powerful tool to construct C-C bonds.

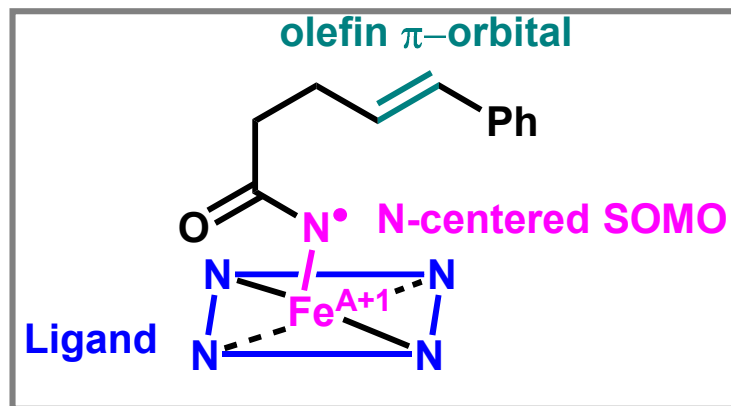
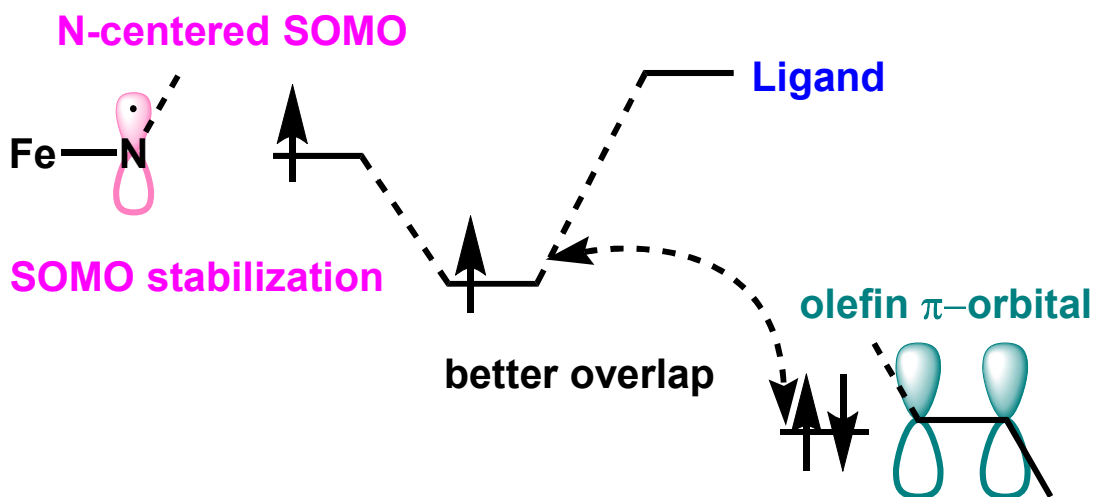
The Strategy for Olefin Oxiamidation

1. Powerful radical reaction → **Fe** catalyst

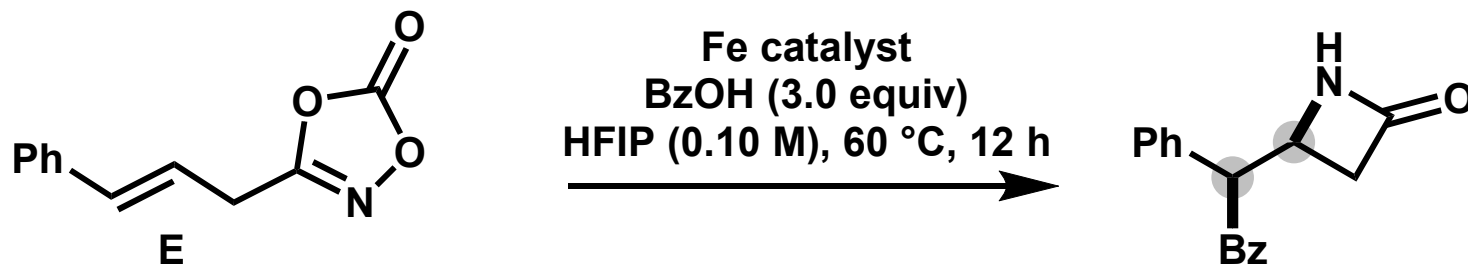
2. The formation of a carbocation intermediate → The **oxidation** by Fe catalyst



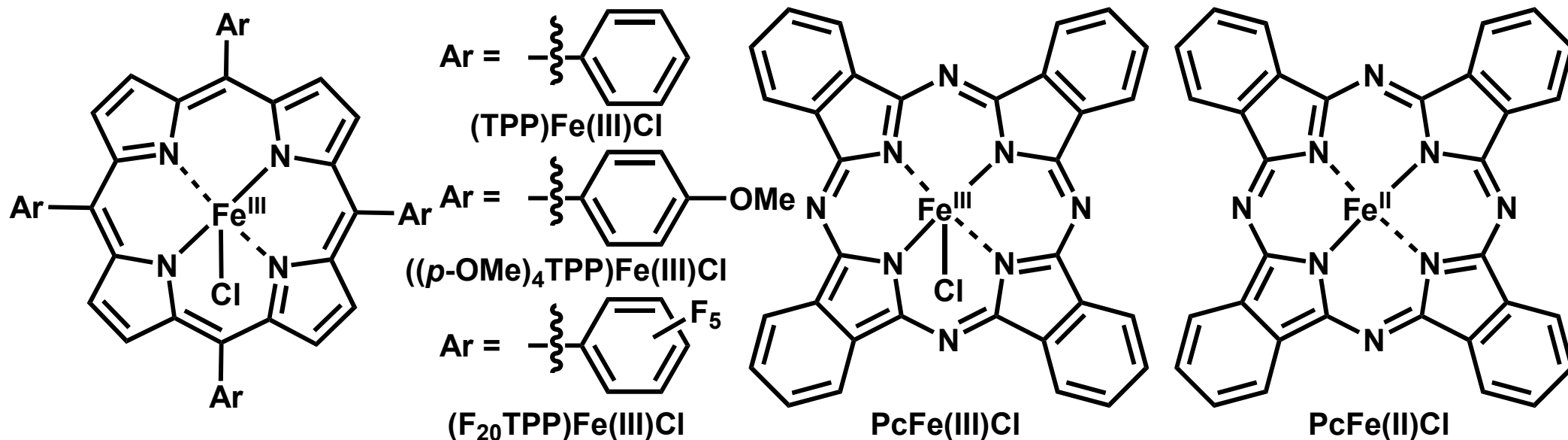
3. Potent π -accepting abilities of ligands enhance orbital interactions between the N-centered SOMO and the olefin π -orbital.



Reaction Optimization

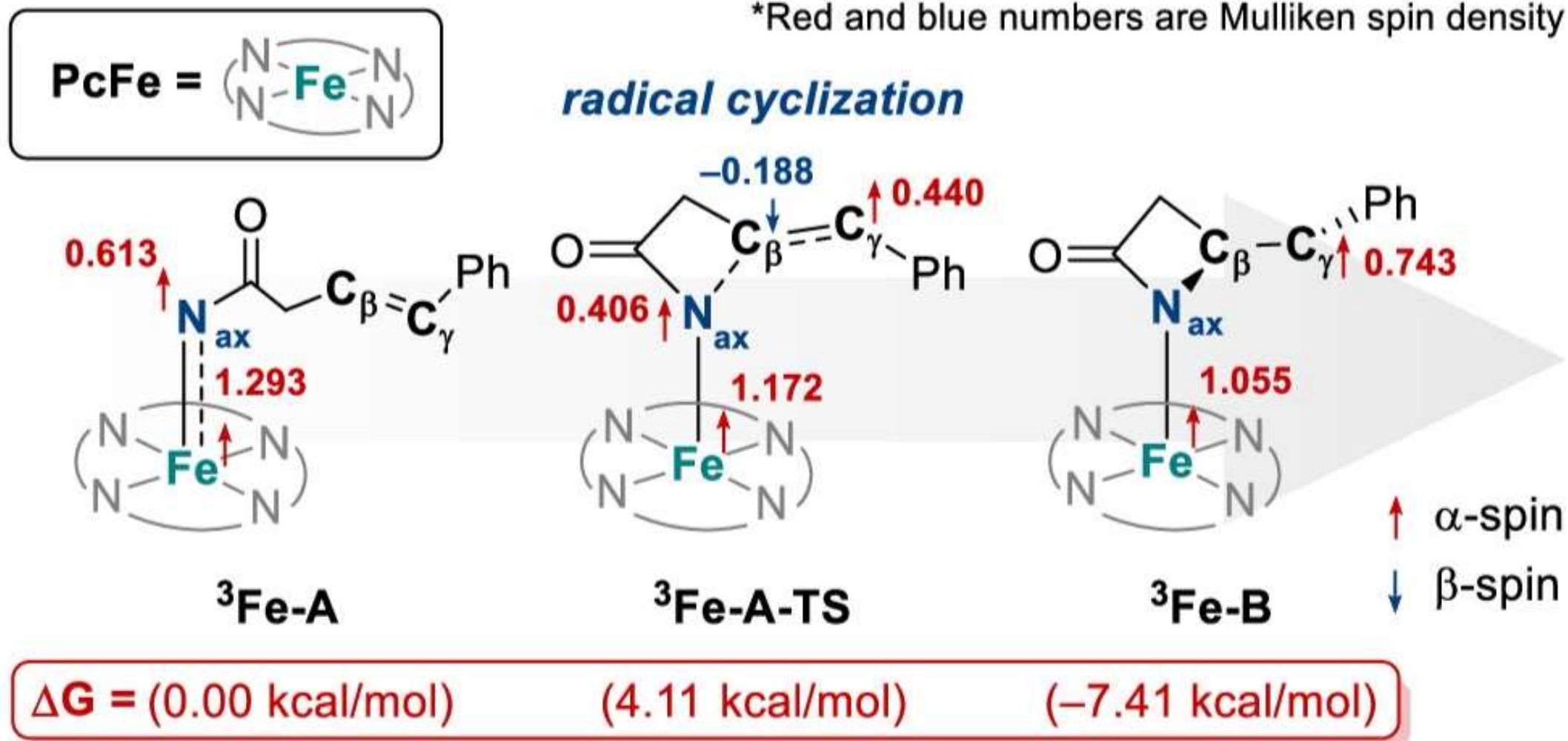


entry	Fe catalyst (mol%)	conv. (%)	yield (%)	d.r.
1	FeCl ₂ (10)	72	N.D.	
2	(TPP)Fe(III)Cl (5.0)	37	11	4:1
3	((p-OMe) ₄ TPP)Fe(III)Cl (5.0)	37	9	5:1
4	(F ₂₀ TPP)Fe(III)Cl (5.0)	40	13	5:1
5	PcFe(III)Cl (1.0)	>95	70	10:1
6	PcFe(II) (1.0)	>95	77	10:1



Mechanistic Investigations (1)

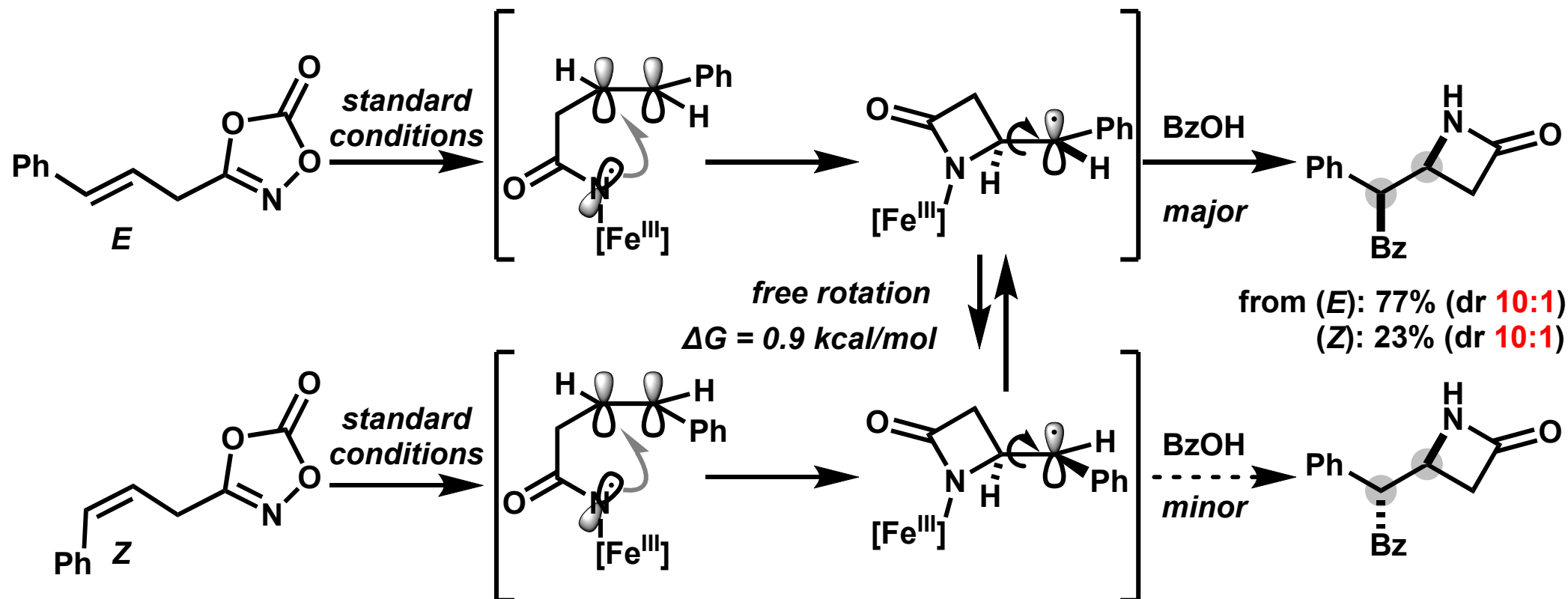
DFT calculation for PcFe-catalyzed 4-exo-trig-cyclization



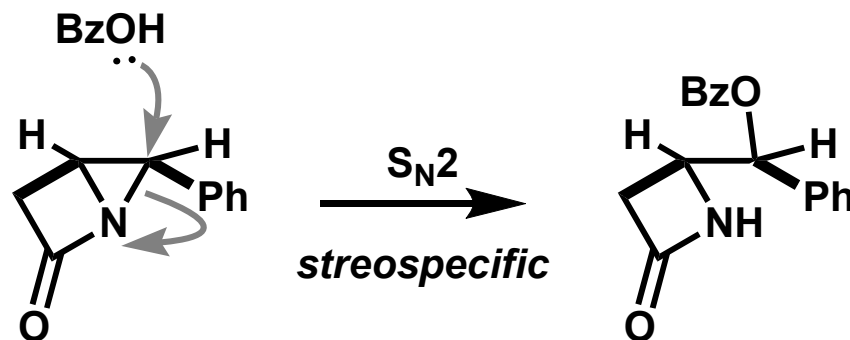
*Mulliken spin density provides a means of estimating partial atomic charges from calculations carried out by the methods of computational chemistry.

Mechanistic Investigations (2)

Stereoconvergent lactamization

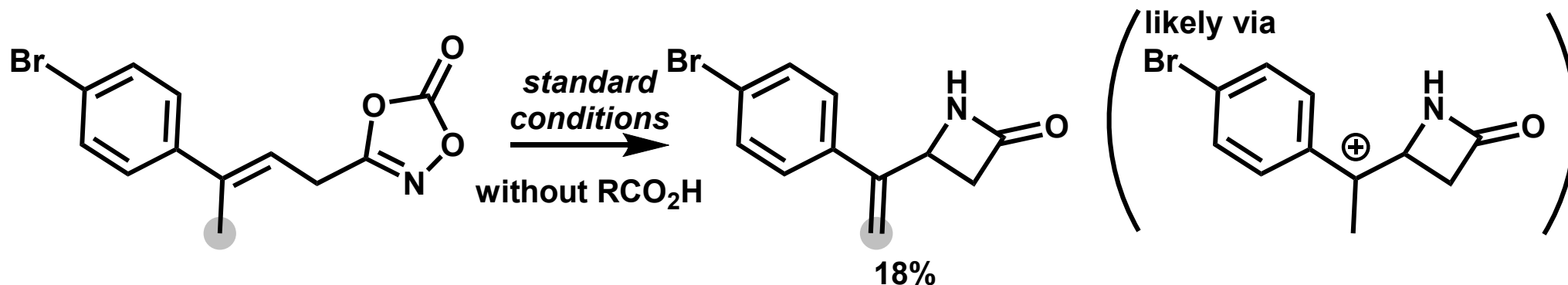


Two-step aziridination/ring-opening mechanism could be ruled out in the present oxyamidation, where the reaction would instead be stereospecific to the olefin geometry.

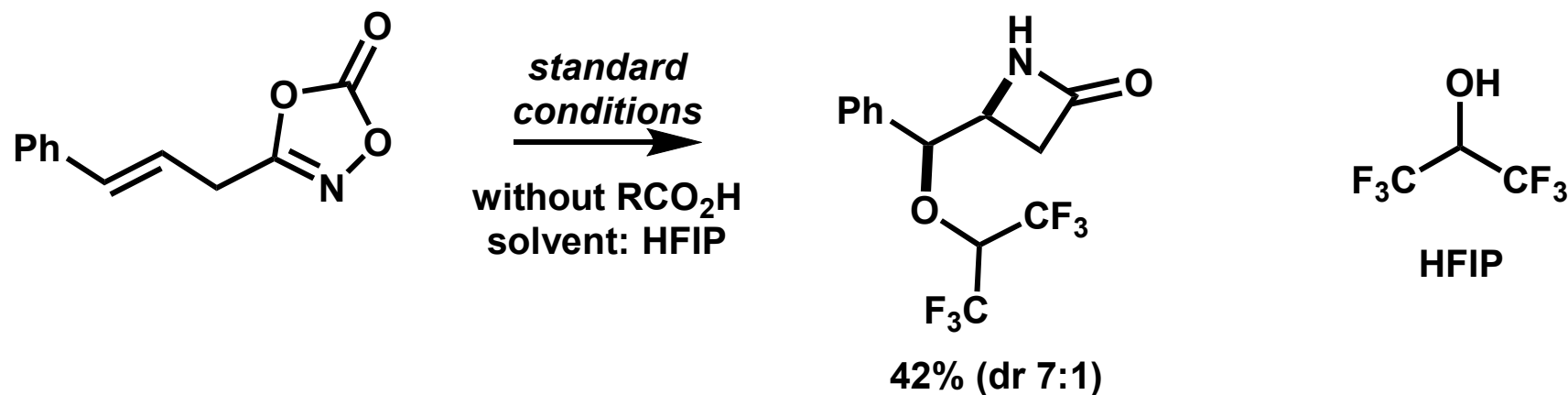


Mechanistic Investigations (3)

Allylic β -lactam formation

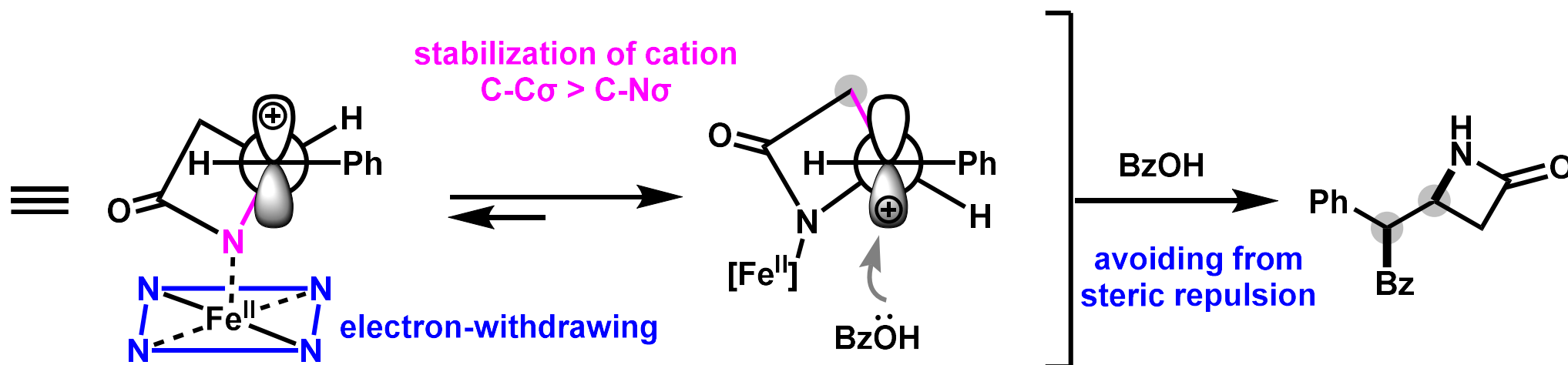
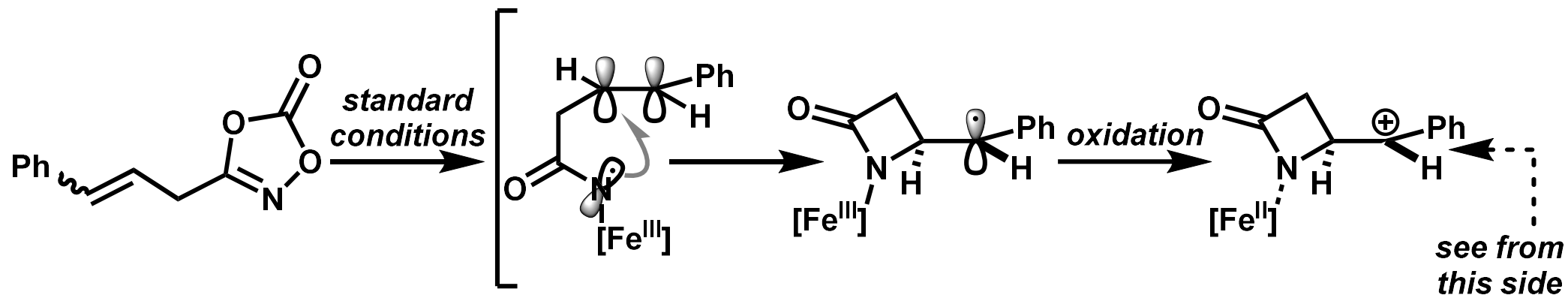


Solvolysis



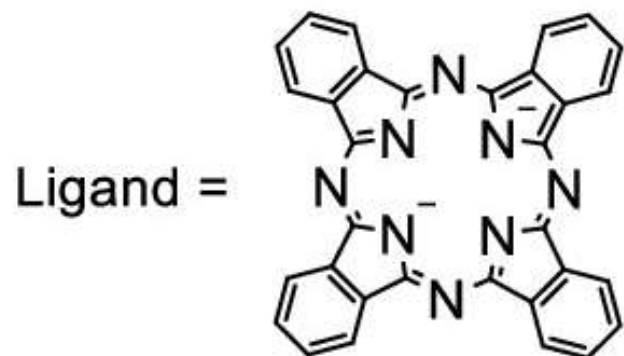
These results are suggestive of the carbocationic intermediacy.

Rationale for Stereoselectivity (My Opinion)

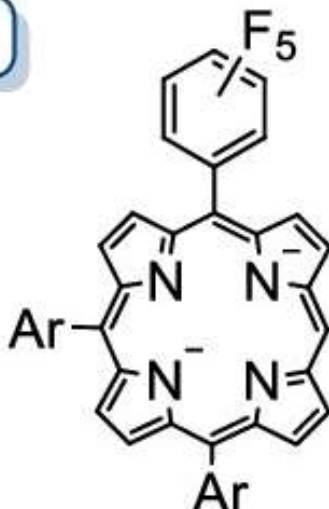


Ligand effects (1)

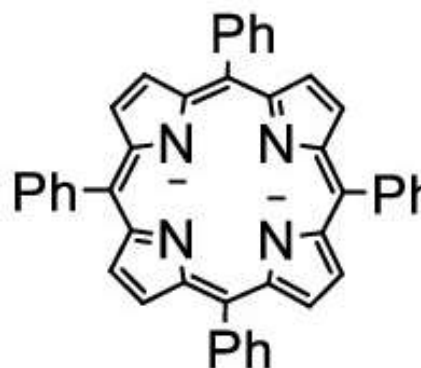
Catalyst: [Ligand]Fe(III)Cl



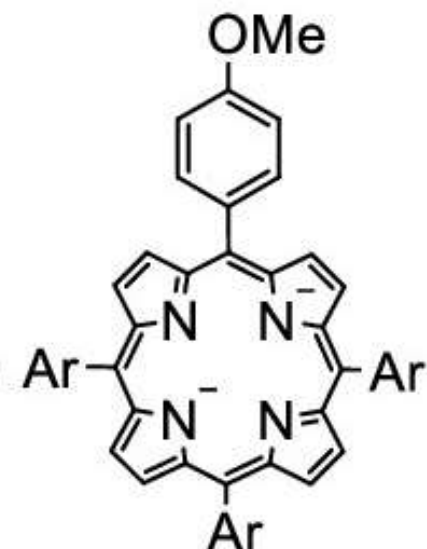
Pc²⁻



F₂₀TPP²⁻



TPP²⁻



(p-OMe)₄TPP²⁻

positive redox potential (vs Fc/Fc⁺)

E_{1/2, II/III} = -0.050 V

-0.382 V

-0.546 V

-1.169 V

Yield/Conv. = 0.71

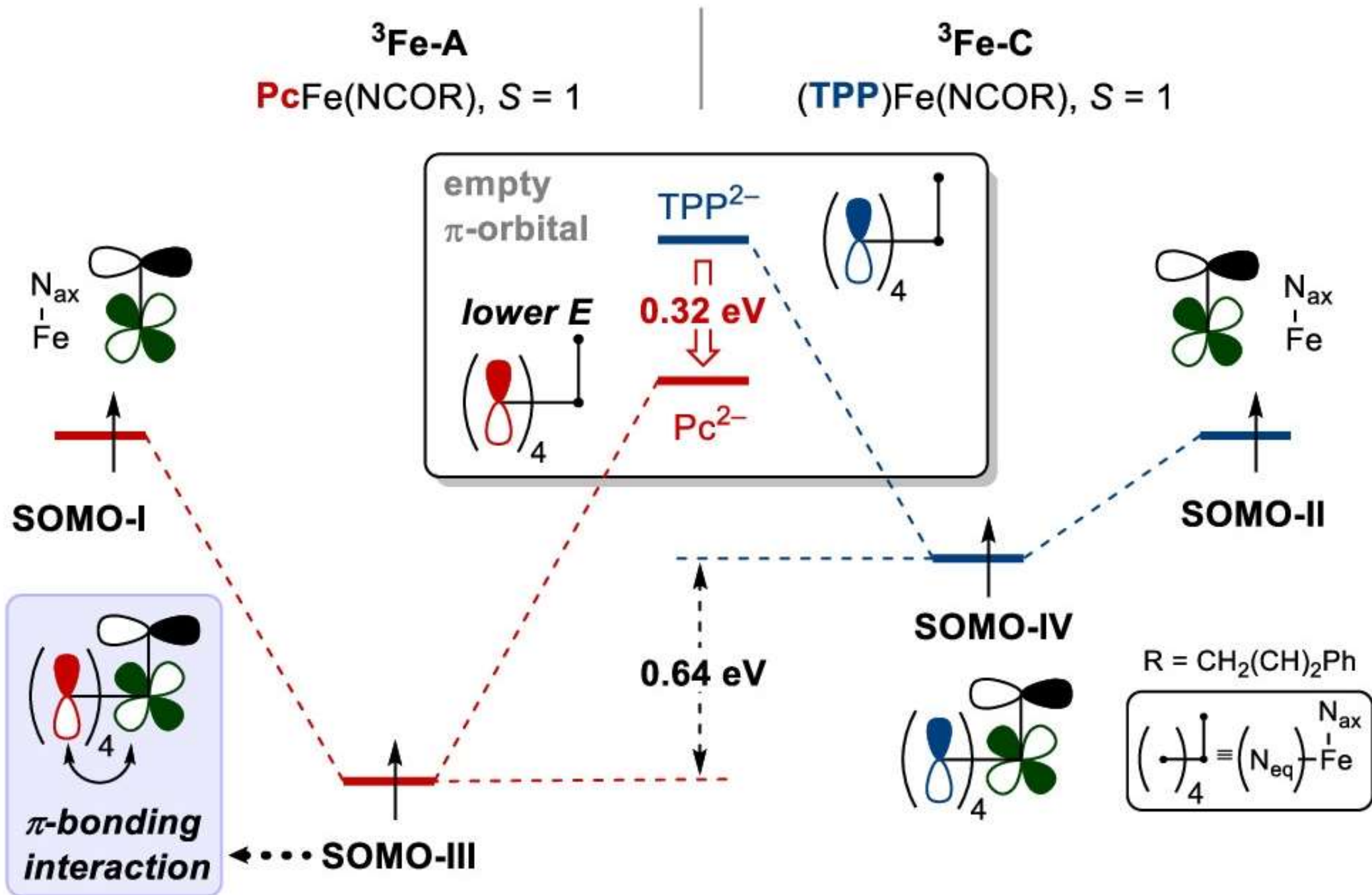
0.33

0.30

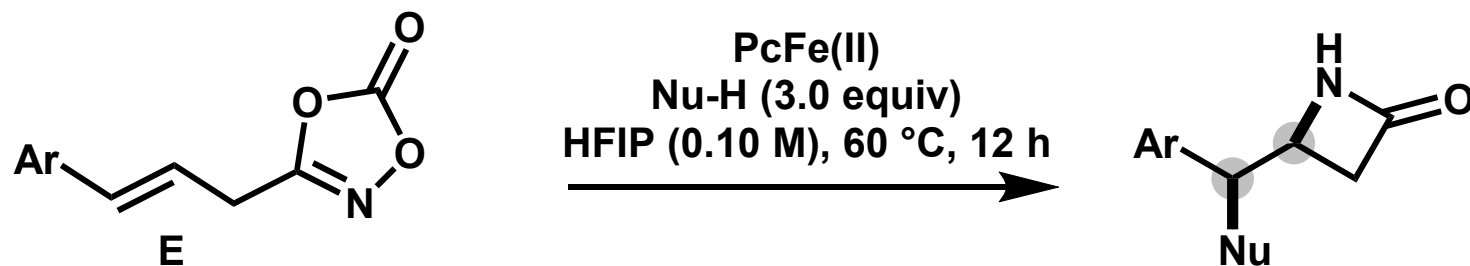
0.24

Among these ferric complexes employed, PcFe(III)Cl exhibited the most positive reduction potential, which can be assigned to Fe(III)/Fe(II) reduction.

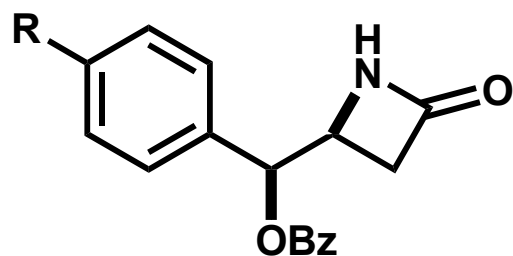
Ligand effects (2)



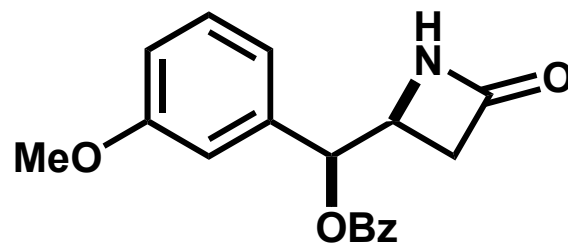
Substrate Scope



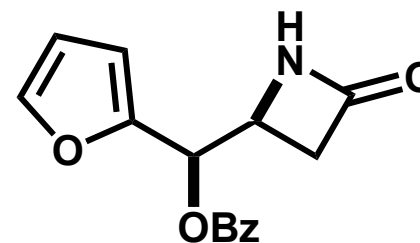
variation of dioxazolones



R = Br: 80% (dr 11:1)
R = Me: 70% (dr 2.4:1)

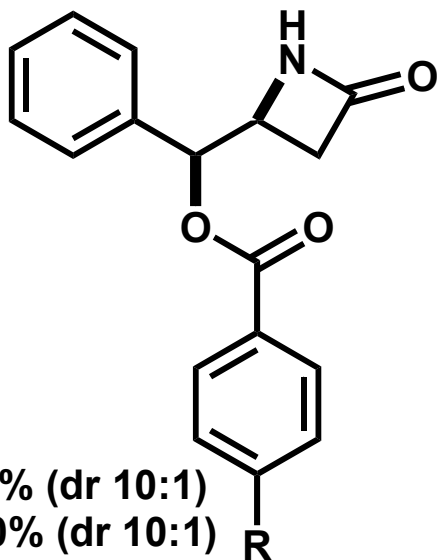


73% (dr 11:1)

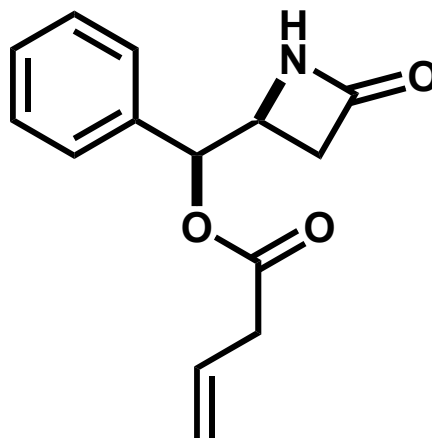


49% (dr 1.4:1)

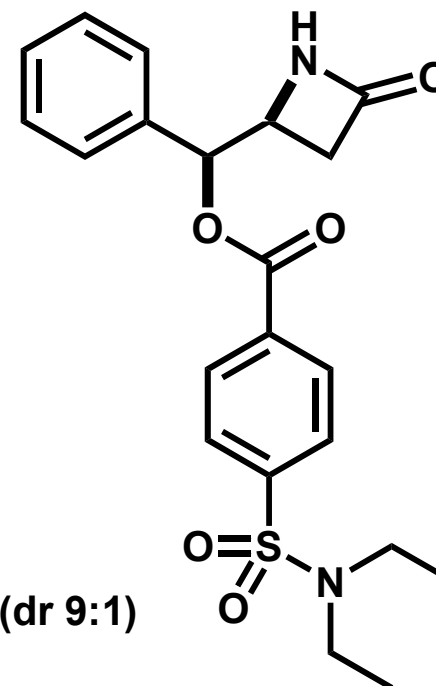
variation of carboxylic acid



R = NO₂: 75% (dr 10:1)
R = OMe: 70% (dr 10:1)

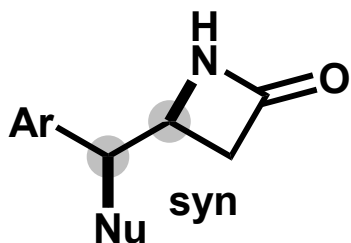


65% (dr >20:1)

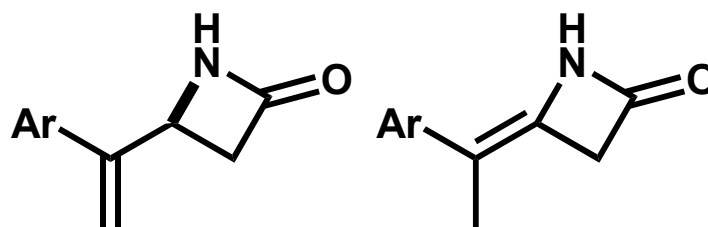
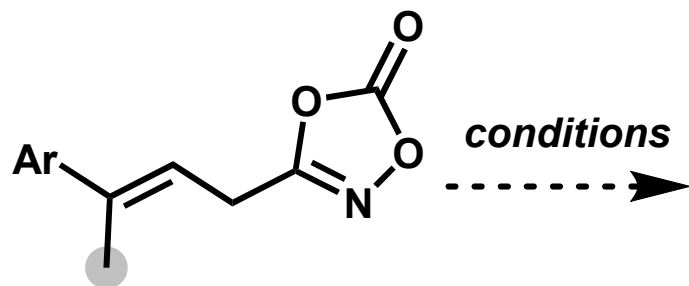
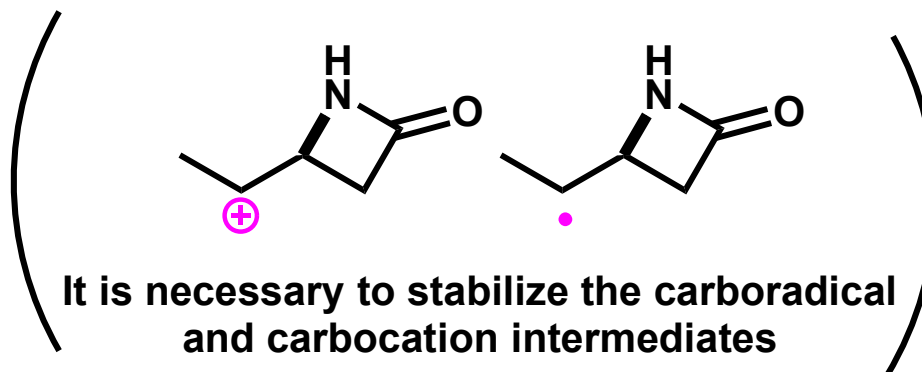
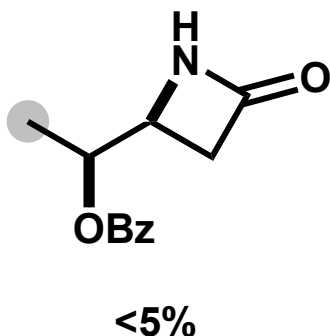


78% (dr 9:1)

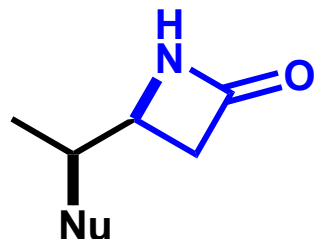
Limitation (My Opinion)



In most cases, only syn products are obtained.



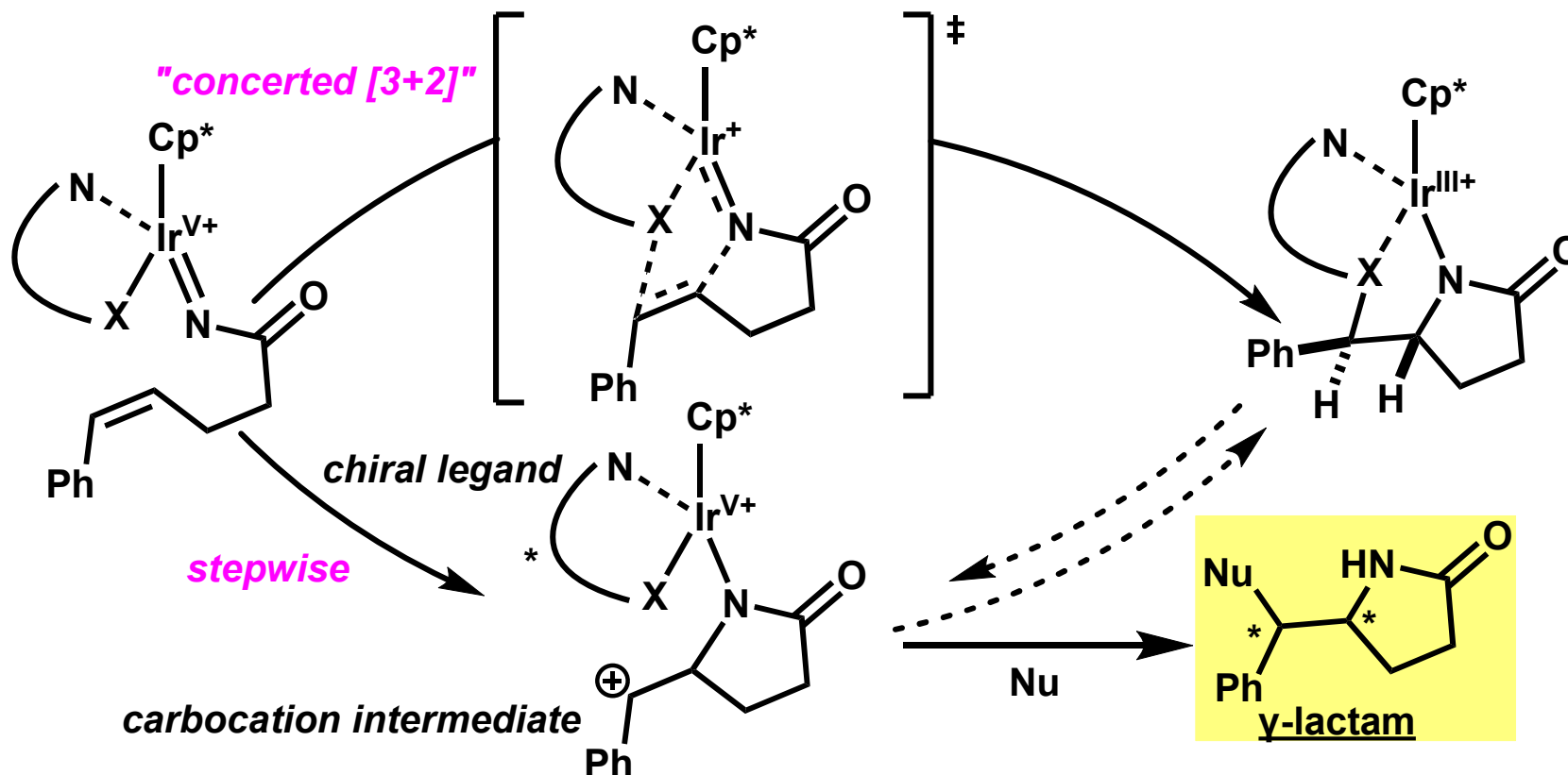
E-1-like elimination should be kept in mind.



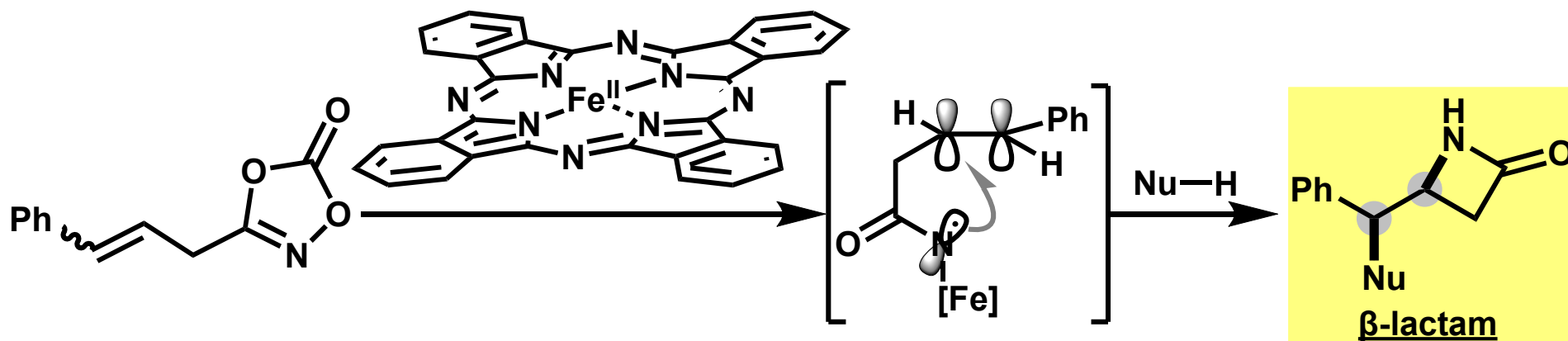
β -lactams are susceptible to nucleophilic attack.
→ Only weak nucleophiles (such as carboxylic acids) are able to be used in this reaction?

Summary

Stereodefined Access to γ -Lactams via Olefin Difunctionalization

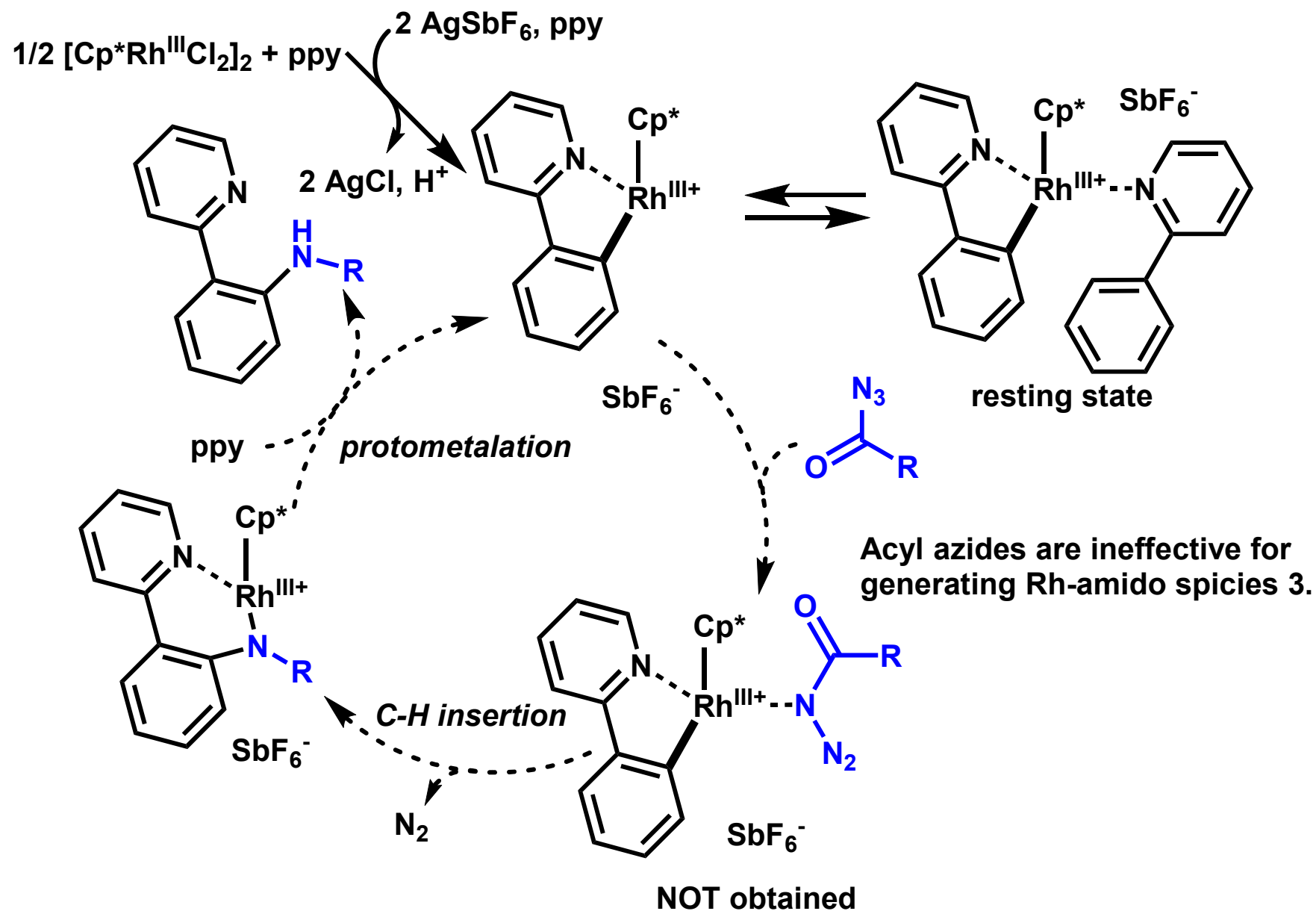


Access to β -Lactams via Iron-Catalyzed Olefin Oxyamidation Enabled by the π -Accepting Phthalocyanine Ligand



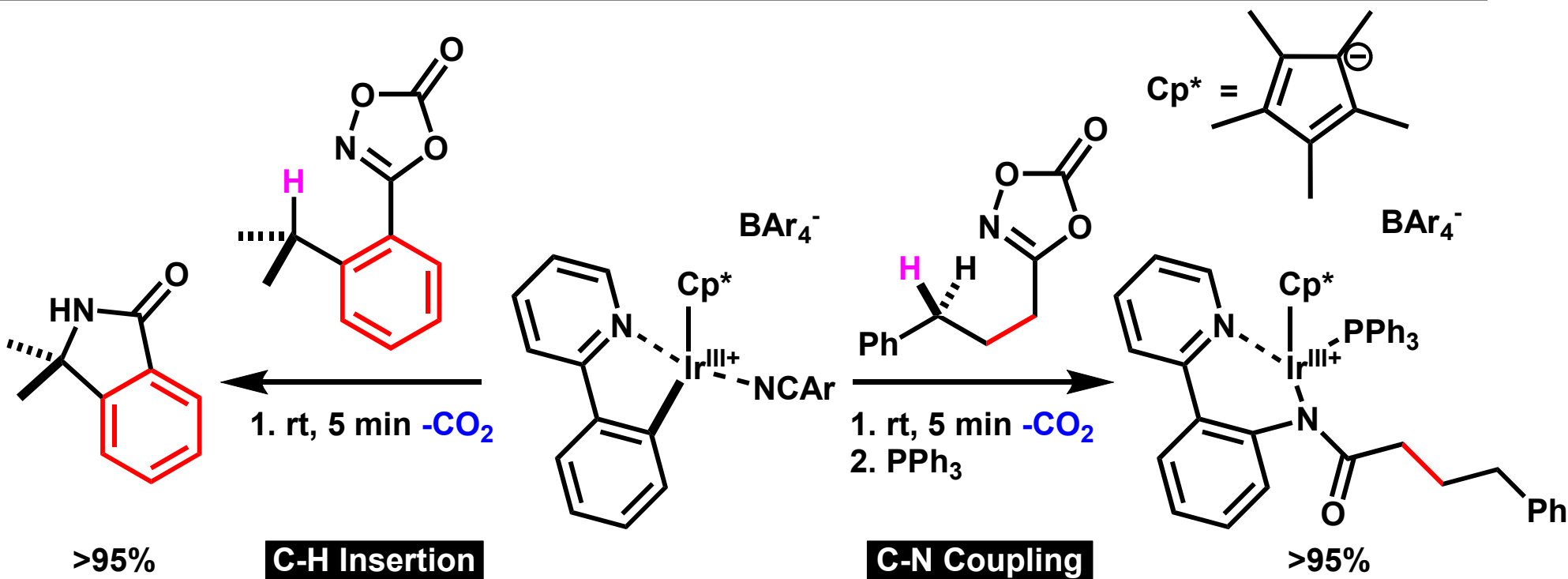
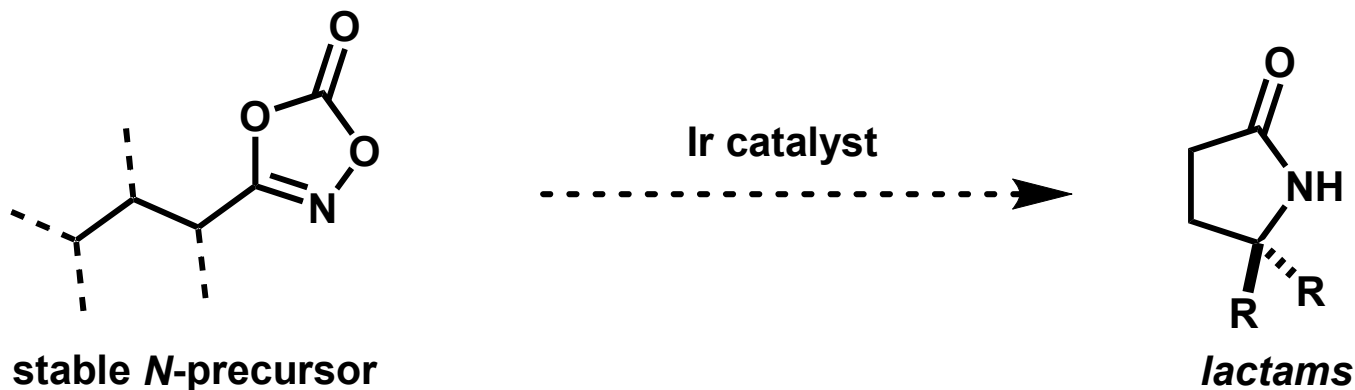
Appendix

Identification of a Rate-limiting Factor

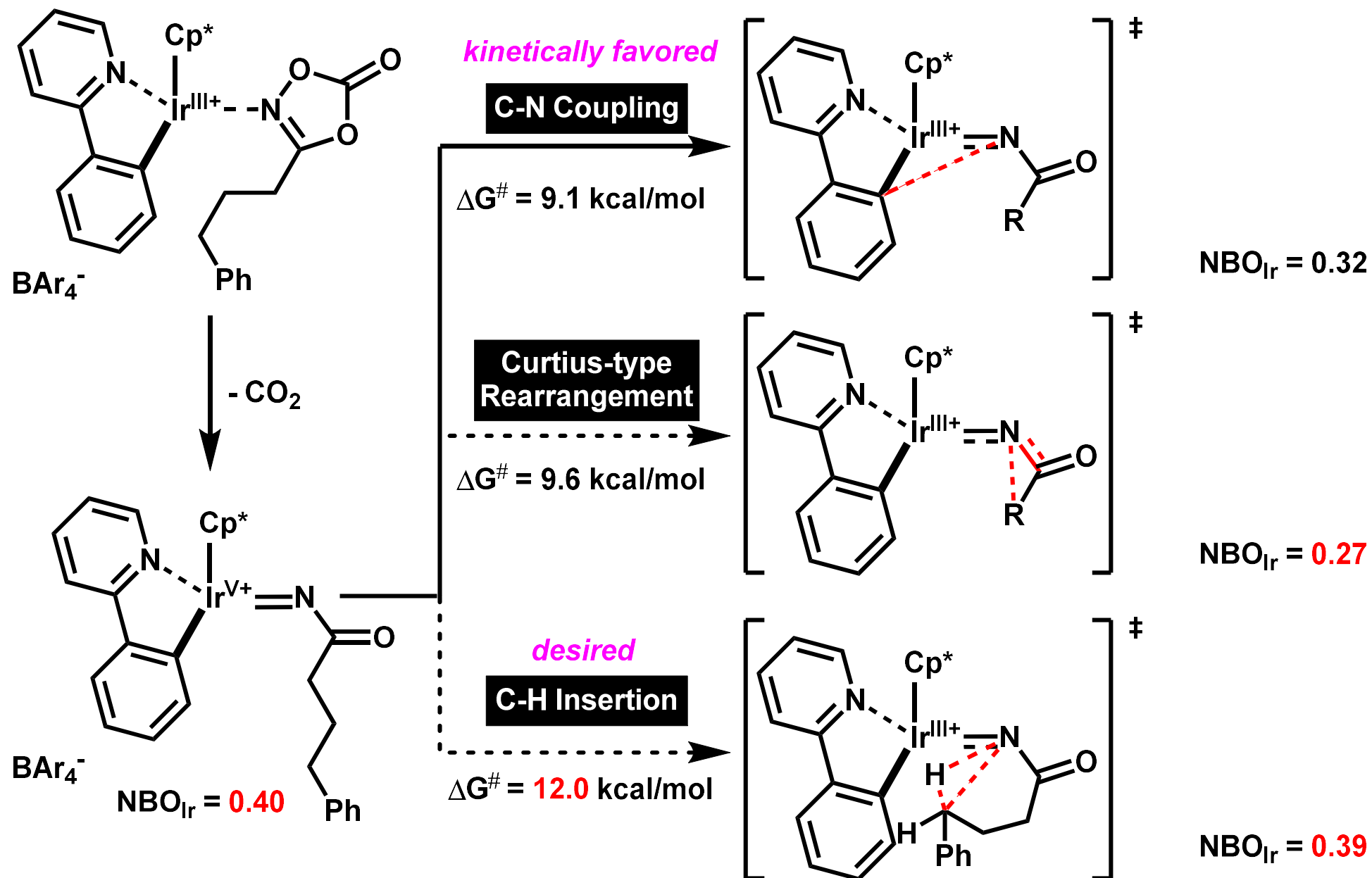


Viability of C-H Insertion

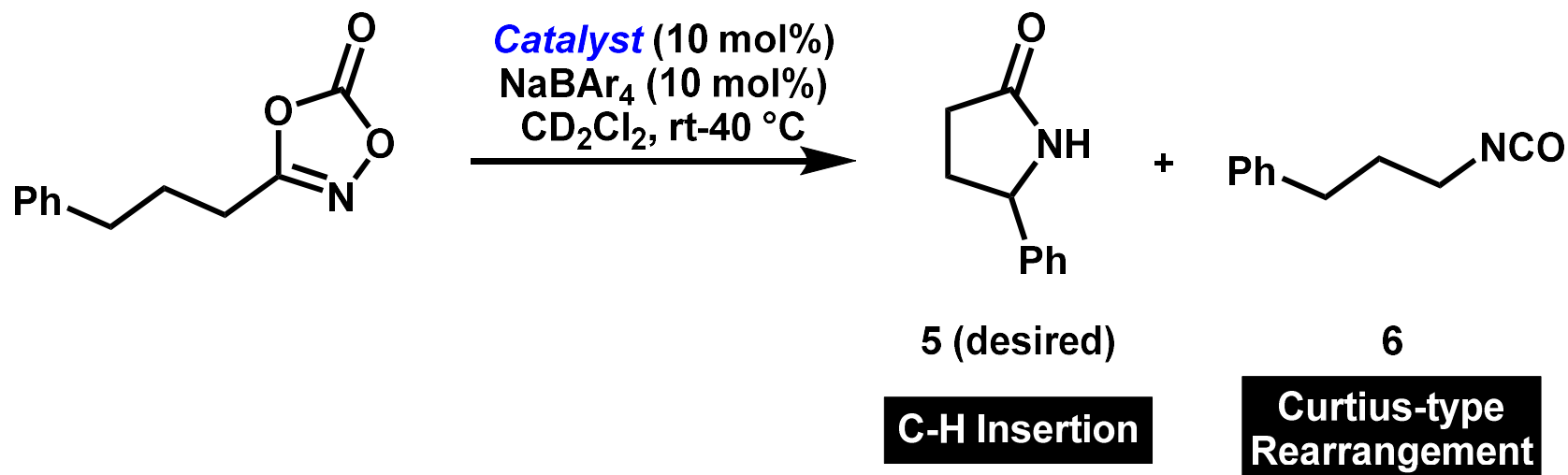
Aim:



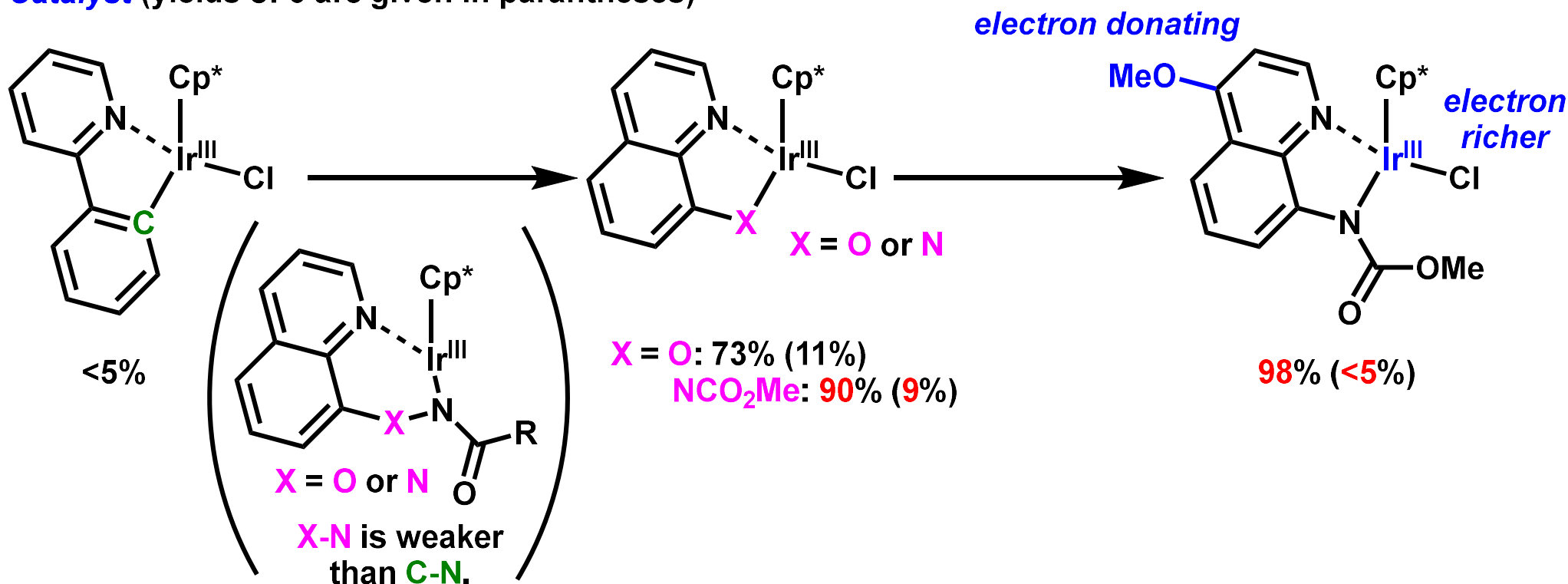
DFT-Calculated Competitive Working Models



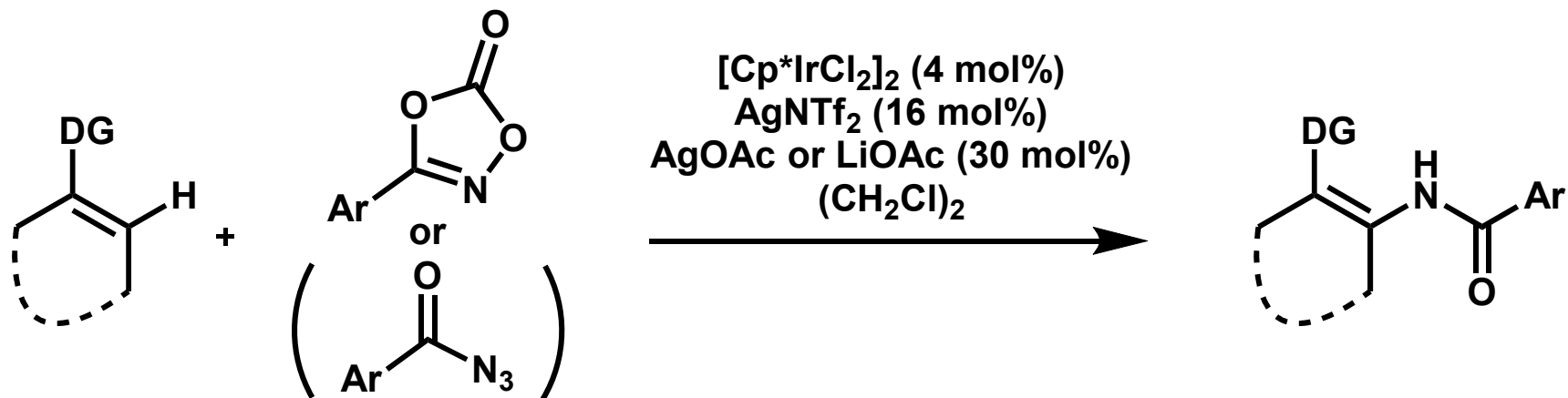
DFT-Calculated Competitive Working Models



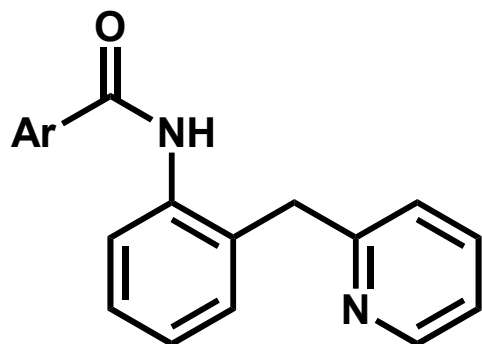
Catalyst (yields of 6 are given in parantheses)



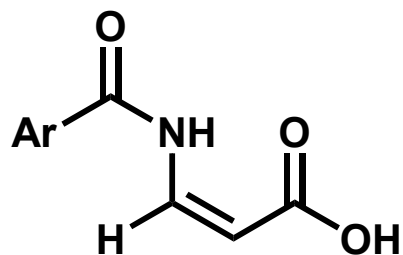
Ir(III)-catalyzed C-H Amidation



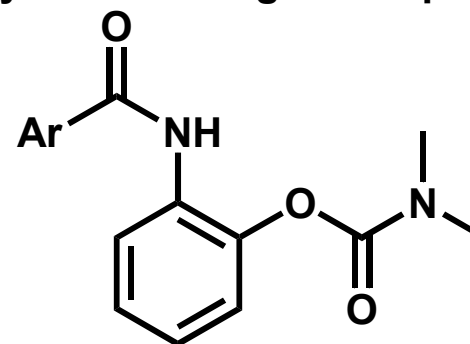
(yields with acyl azides are given in parentheses)



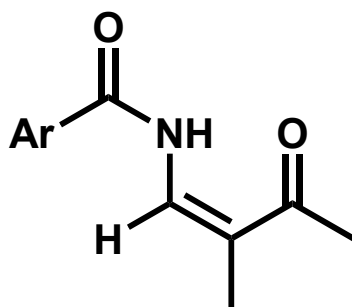
>95% (<5%)



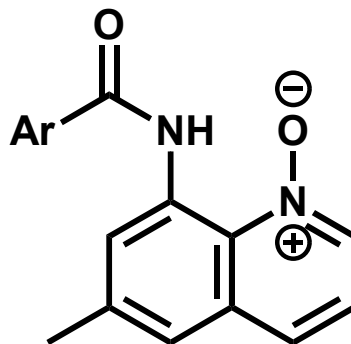
70% (<5%)



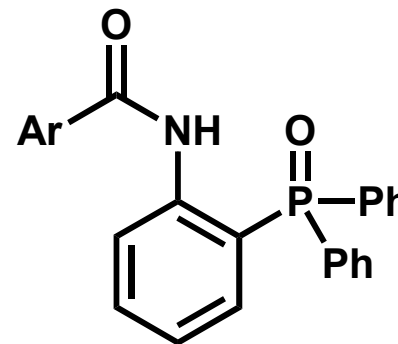
60% (<5%)



77% (25%)

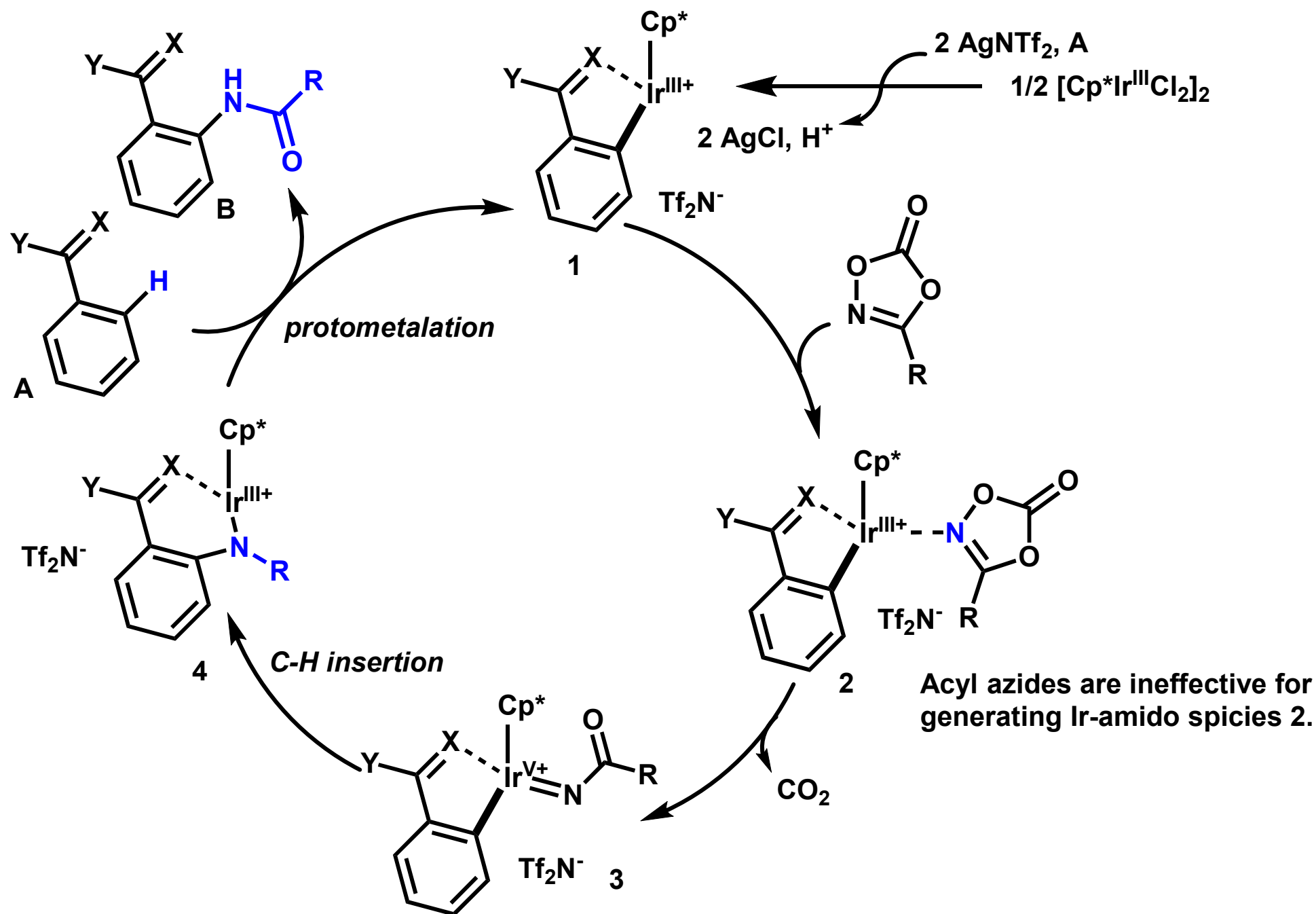


63% (<5%)

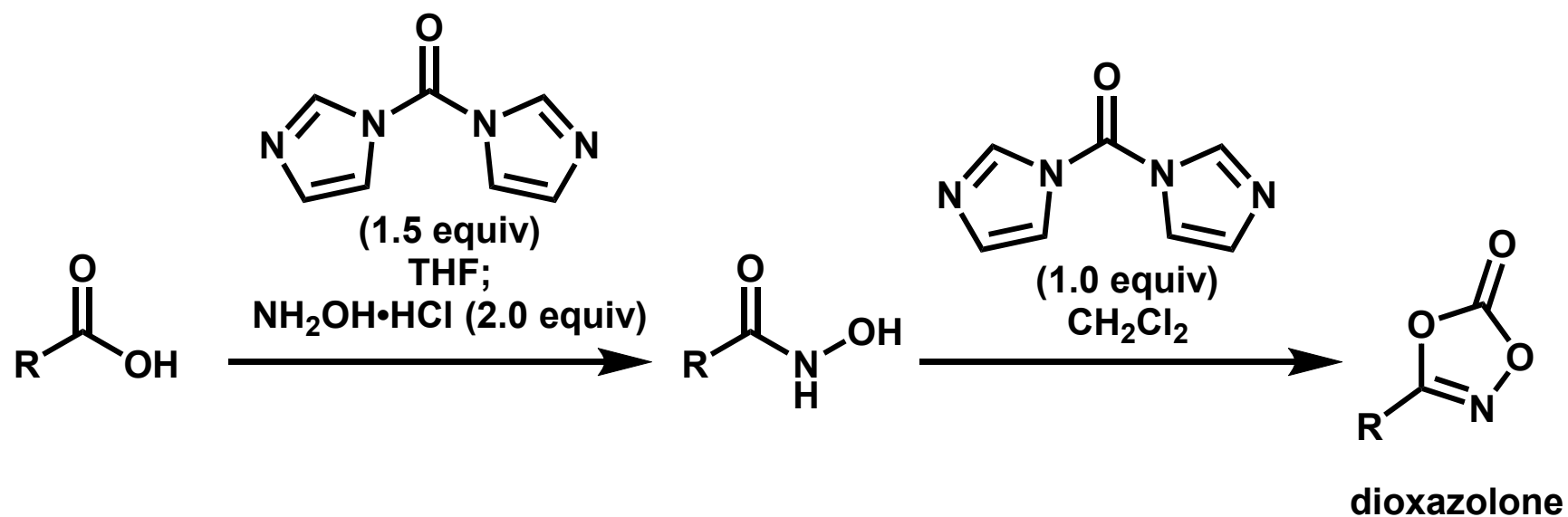


52% (9%)

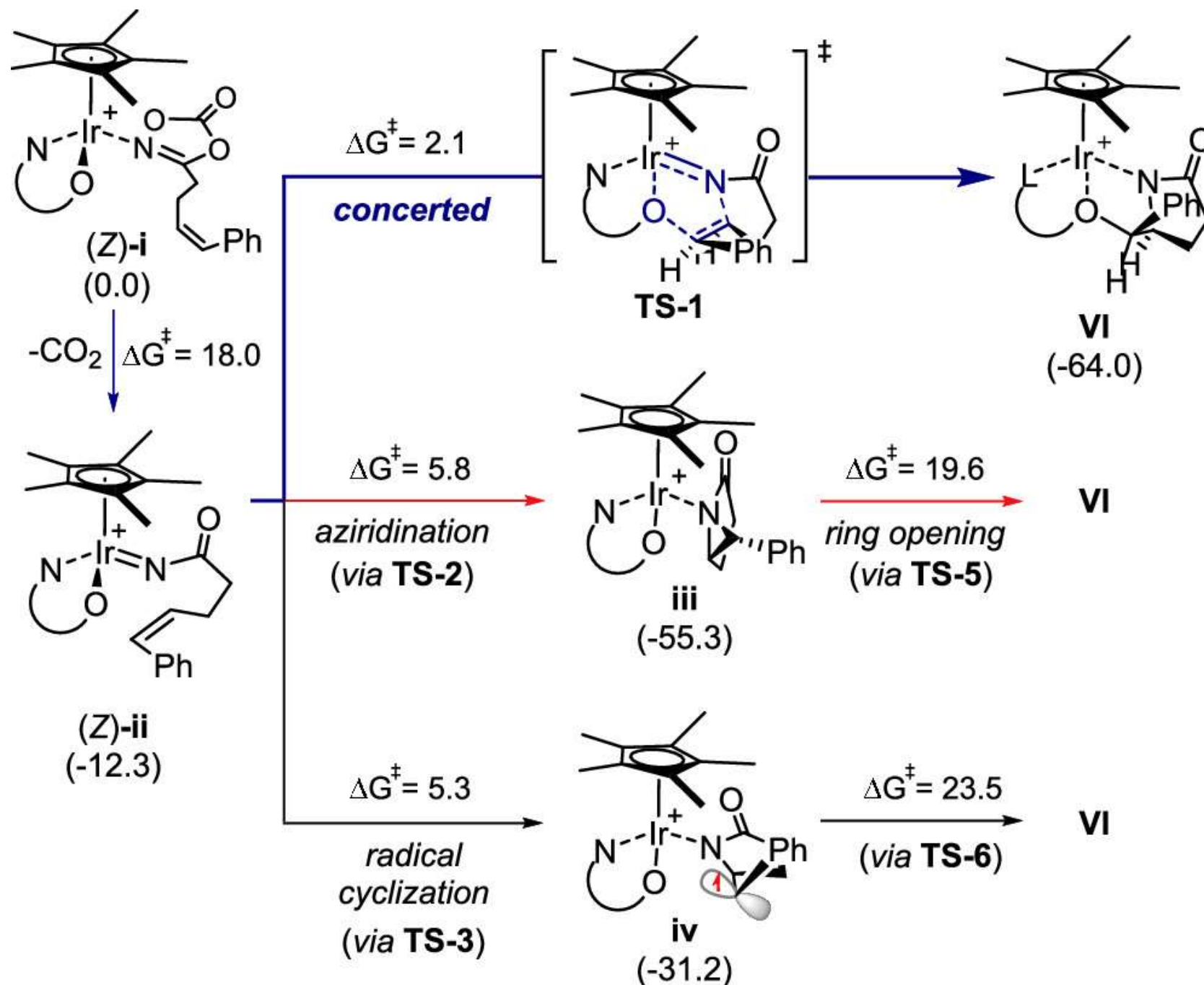
Proposed Reaction Mechanism



Synthesis of Dioxazolones



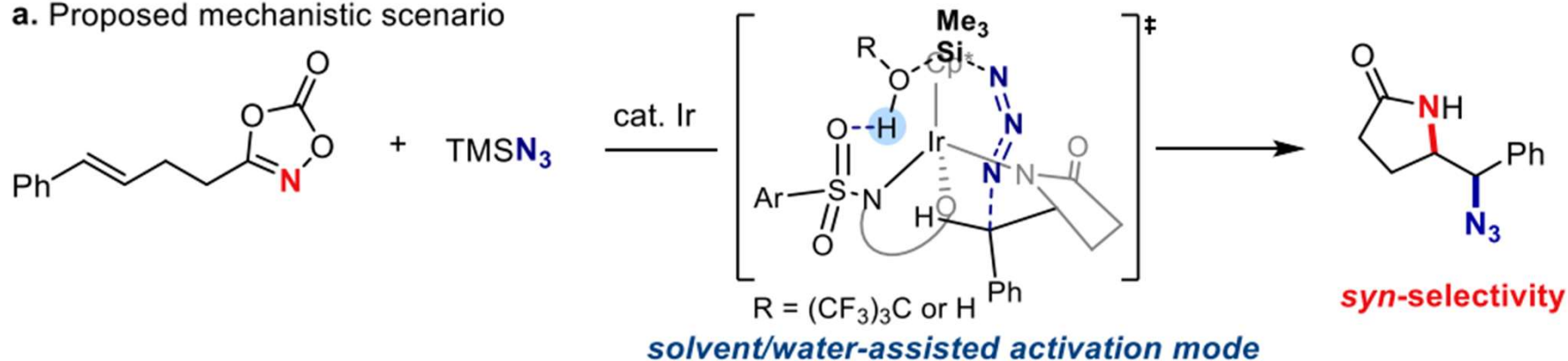
Plausible Working Modes of a Putative Ir-nitrenoid



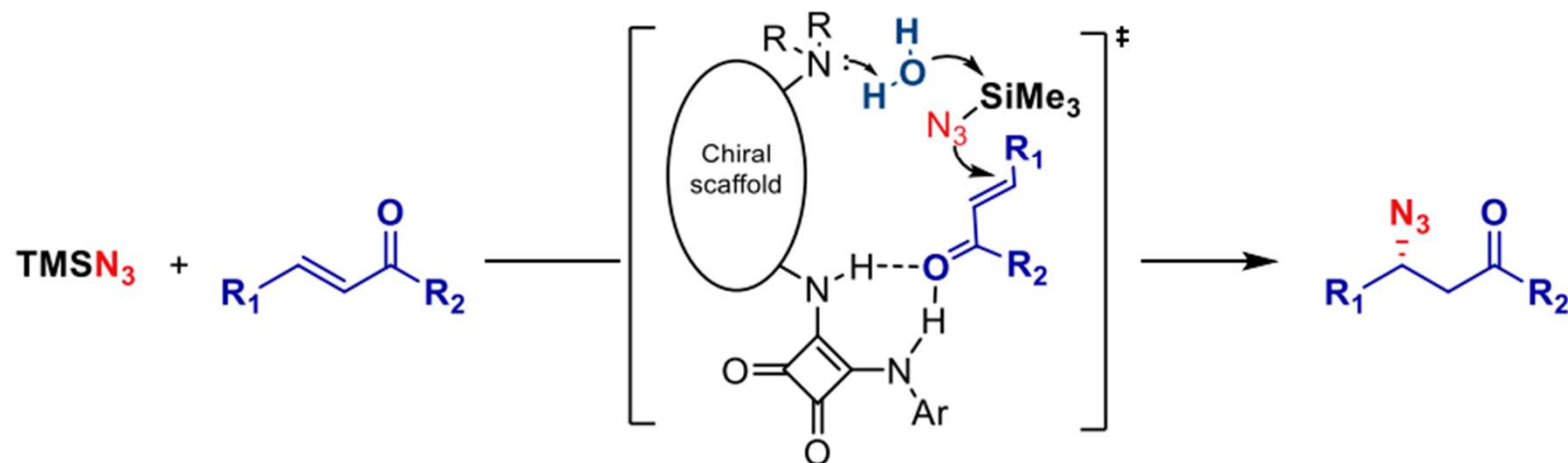
PCM(dichloromethane)-M06/SDD+6-311++G**//M06/LanI2dz +6-31G**

Proposed Mechanistic Mode for Syn-selective Pathway

a. Proposed mechanistic scenario

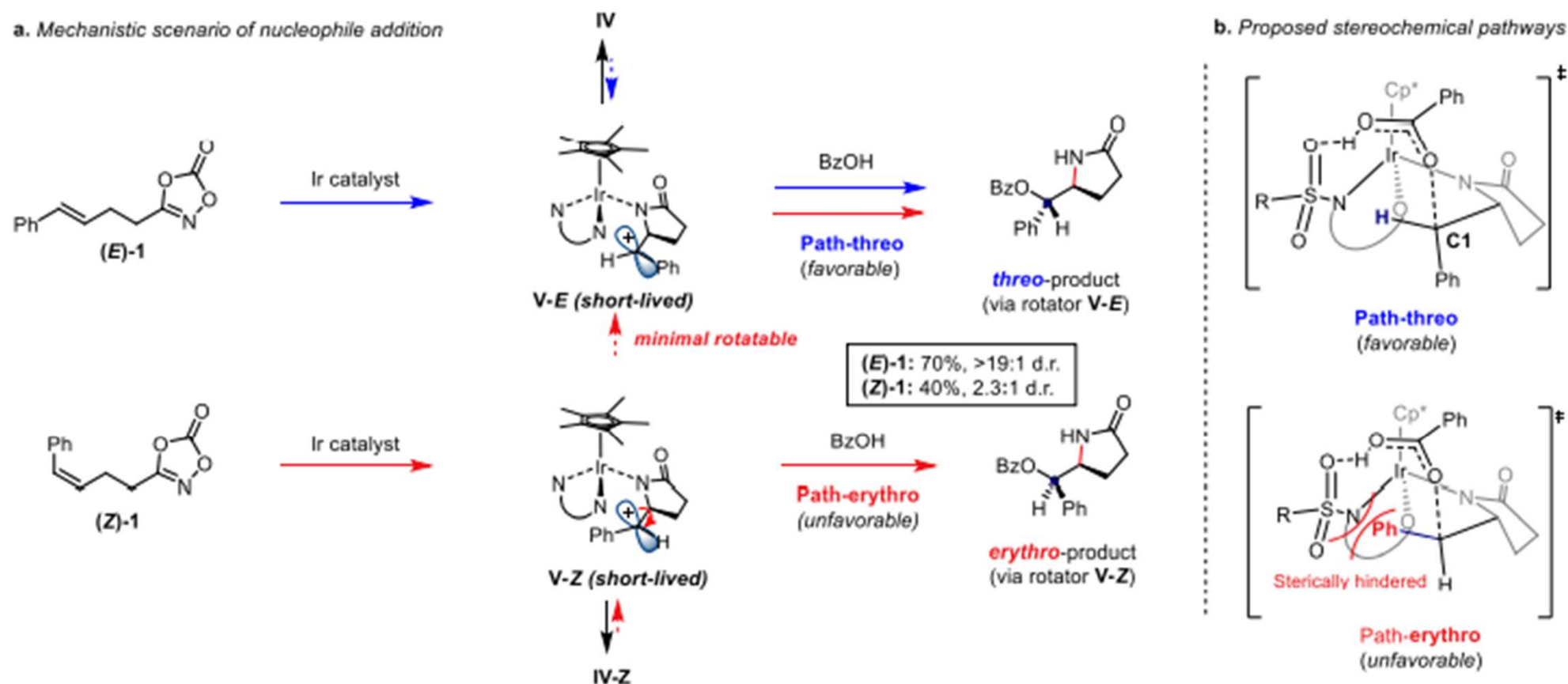


b. Water-assisted activation mode in catalysis (*Adv. Synth. Catal.* **2019**, 361, 4790)

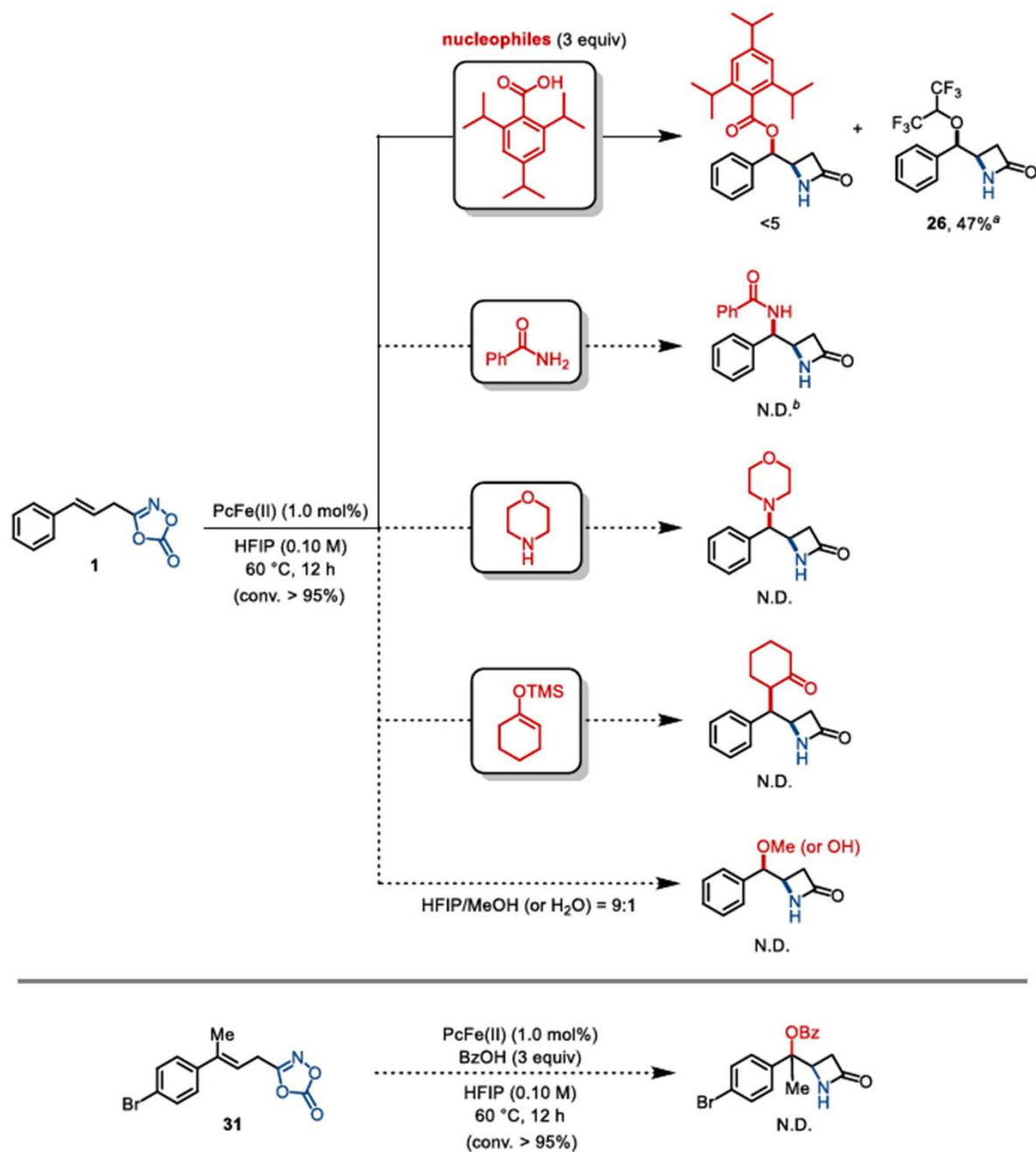


The Origin of the Partial Convergence

Scheme S2. The origin of the partial convergence.



Unsuccessful Results of PcFe(II)-catalyzed Olefin Difunctionalization



Detection of Triplet PcFe-imidyl Radical by Low Temperature X-band EPR

