

# **Synthesis and Application of B-N Containing Tryptophan**

**2021.6.12. Literature Session  
M1 Yosuke Nakata**

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**1. Introduction**

**2. Background**

**3. Synthesis and Application of B-N Containing Tryptophan  
(*Chem. Sci.* 2019, 10, 4994)**

# Contents

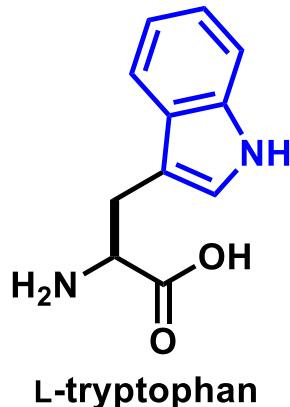
**1. Introduction**

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(*Chem. Sci.* 2019, 10, 4994)**

# Tryptophan : Characteristic Residue of Protein

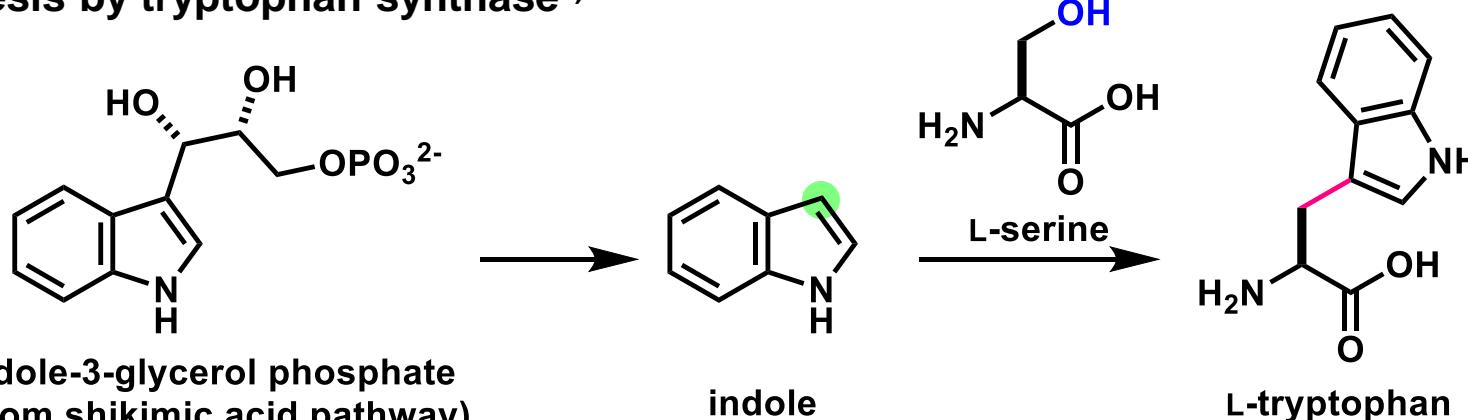
Natural amino acid possessing **indole ring**



- intrinsic fluorescence<sup>1)</sup>  
 $\lambda_{\text{em}}=348 \text{ nm}$  (Phe :  $\lambda_{\text{em}}=282 \text{ nm}$ , Tyr :  $\lambda_{\text{em}}=303 \text{ nm}$ )
- electron rich aromatic ring<sup>2)</sup>  
 $\pi$ -interaction, high redox potential

Relatively rare amino acid  
...1% of amino acids in found protein<sup>3)</sup>

Biosynthesis by tryptophan synthase<sup>4)</sup>



1) Teale, J. W. F.; Weber, G. *Biochme. J.* **1957**, 65, 476.

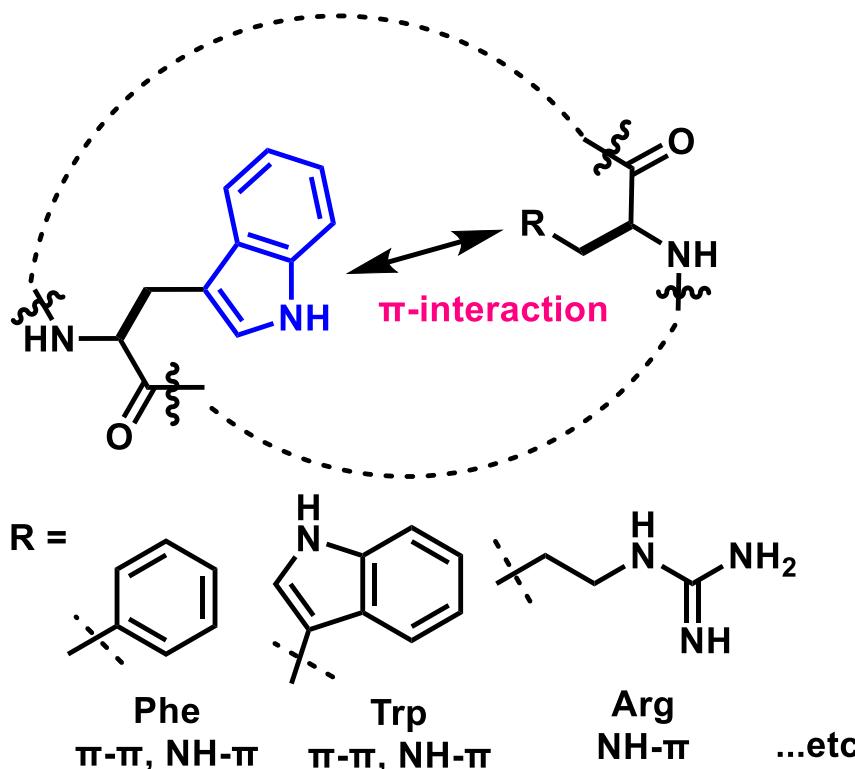
2) (a) Ruan, C.; Rodgers, T. M. *J. Am. Chem. Soc.* **2004**, 126, 14600. (b) Dougherty, A. D. *J. Nutr.* **2007**, 137, 1504S. (c) Dougherty, A. D. *J. Org. Chem.* **2008**, 73, 3667. (d) Burley, K. S.; Petsko, A. G. *Science*. **1985**, 229, 23.

3) McCaul, P. C.; Ludescher, D. R. *Photochem. Photobiol.* **1999**, 70, 166.

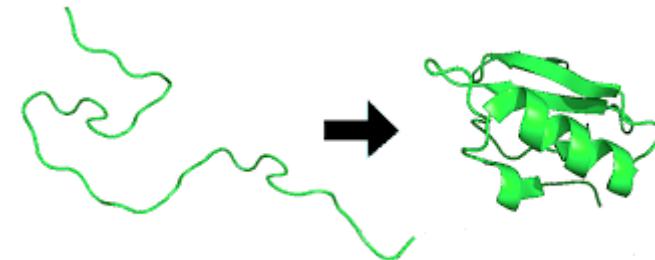
4) Dunn, M. F.; Schlichting, I.; Barends, R. T. *Curr. Opin. Chem. Biol.* **2008**, 12, 593.

# Biological Function and Application of Trp

stabilize peptides by  $\pi$ -interaction<sup>1)</sup>

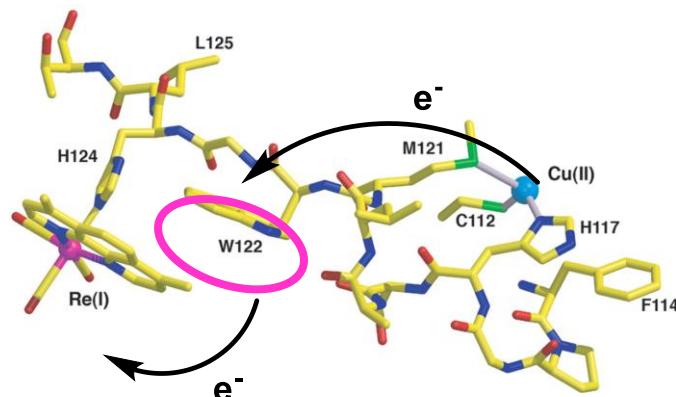


Probe protein folding<sup>2)</sup>



Wavelength or total emission of fluorescence changes by protein folding.

Accelerate electron transfer<sup>3)</sup>



Trp plays a very important role in peptides and proteins despite its rarity.



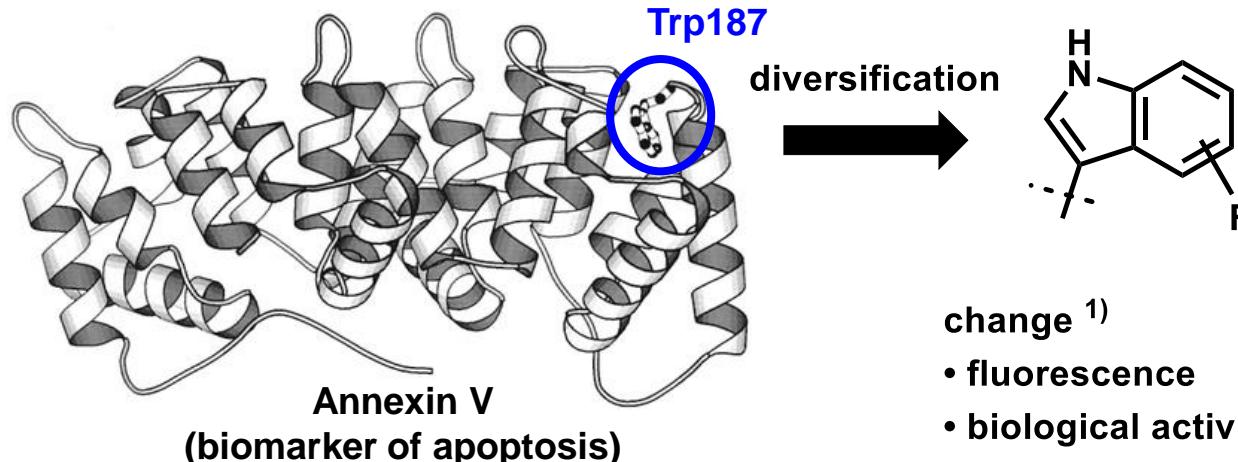
**Useful target for protein study using unnatural amino acids (UAAAs)**

1) Santiveri, M. C.; Jiménez, A. M. *Biopolymers*. 2010, 94, 779.

2) Royer, A. C. *Chem, Rev.* 2006, 106, 1769.

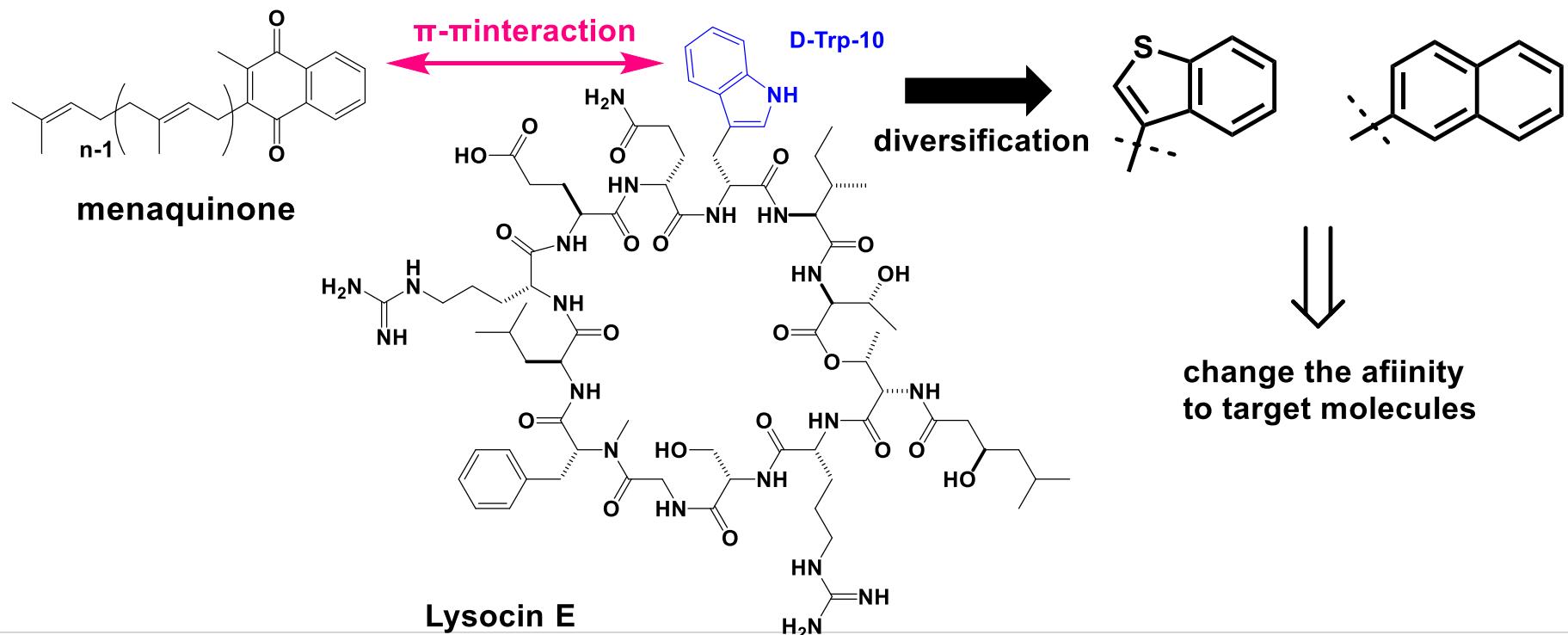
3) Shih, C.; Museth, K. A.; Abrahamsson, M.; Blanco-Rodriguez, M. A.; DiBilio, J. A.; Sudhamsu, J.; Crane, R. B.; Ronayne, L. K.; Towrie, M.; Vlcek, A. Jr.; Richards, H. J.; Winkler, R. J.; Gray, B. H. *Science*. 2008, 320, 1760.

# Application of UAAs to Trp residues



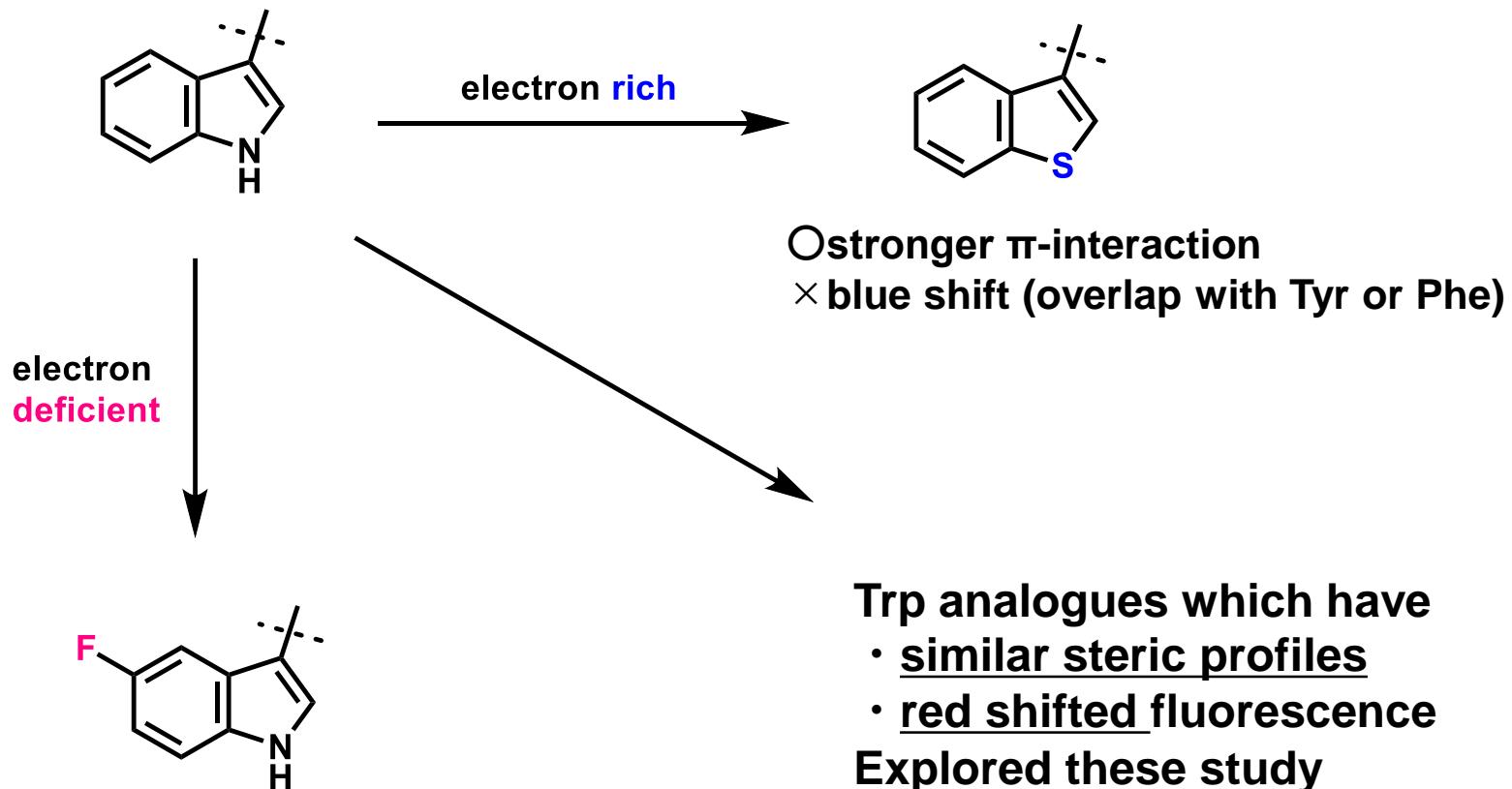
change 1)

- fluorescence
- biological activity



# New Chemical Space for UAAs

## Trp alteration



○ red shift  
× low incorporation efficiency<sup>1)</sup>

Trp analogues which have

- similar steric profiles
- red shifted fluorescence

**Explored these study**

1) Pratt, A. E.; Ho, C. *Biochemistry*, 1975, 14, 3035.

# Contents

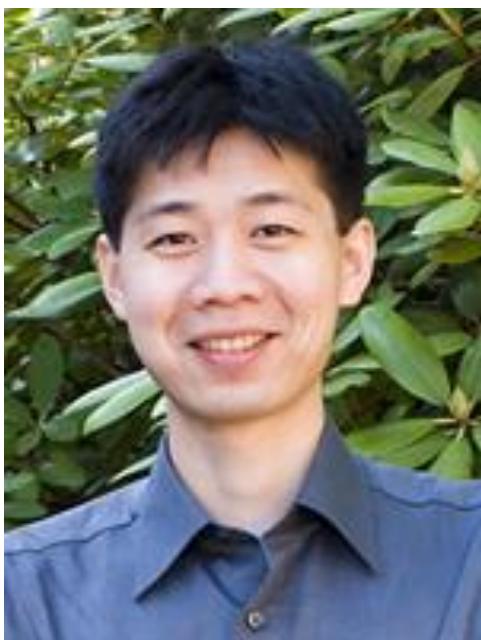
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(*Chem. Sci.* 2019, 10, 4994)

# Author's Profile<sup>1)</sup>

## Prof. Shin-Yuan Liu



### Research Career

**1995-1997 1st diploma in chemistry @Vienna University of Technology**

**1998-2003 Ph.D. in organic chemistry @Massachusetts Institute of Technology  
(Prof. Gregory C. Fu)**

**2003-2006 Postdoctoral fellow @Massachusetts Institute of Technology (Prof. Daniel G. Nocera)**

**2006-2012 Assistant Professor @University of Oregon**

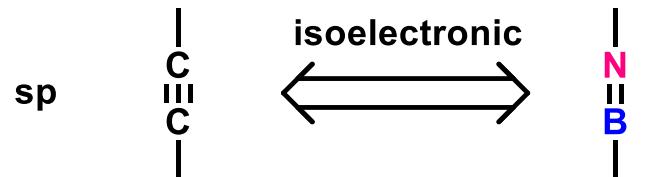
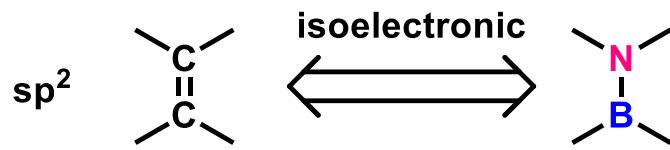
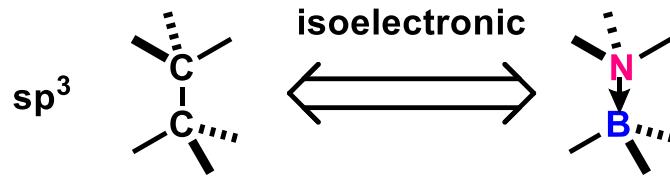
**2012-2013 Associate Professor @University of Oregon**

**2013- Full Professor @Boston College**

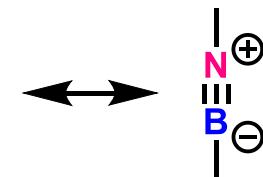
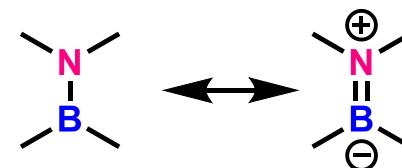
### Research Interests:

**Development of boron(B)-nitrogen(N)-containing heterocycles, specifically azaborines.**

# BN/CC Isosterism<sup>1)</sup>



element valence electron	<b>B   C   N</b>
3   4   5	



BN bond is **highly polarized** though total valence electron count was the same as CC bond.

BN/CC isosteres have

- similar steric profile
- different electronic character (dipole moment, reactivity, hydrogen bonding)

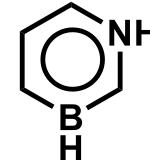
# BN-arenes<sup>1)</sup>



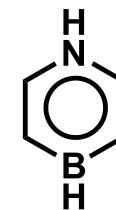
1,2-azaboreine

different properties from natural arenes

- relatively low aromaticity
- high energy HOMO
- red shifted absorption and emission



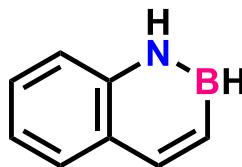
1,3-azaboreine



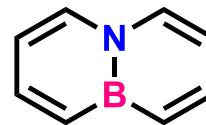
1,4-azaboreine



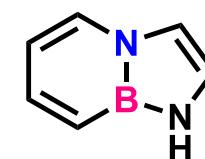
Liu (2013)



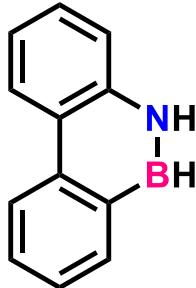
Dewar (1959)



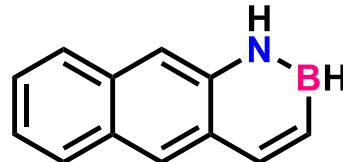
Dewar (1964)



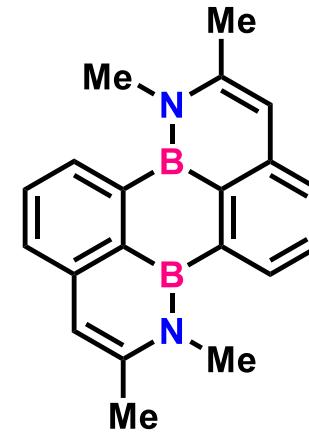
Liu (2011)



Dewar (1958)



Liu (2014)

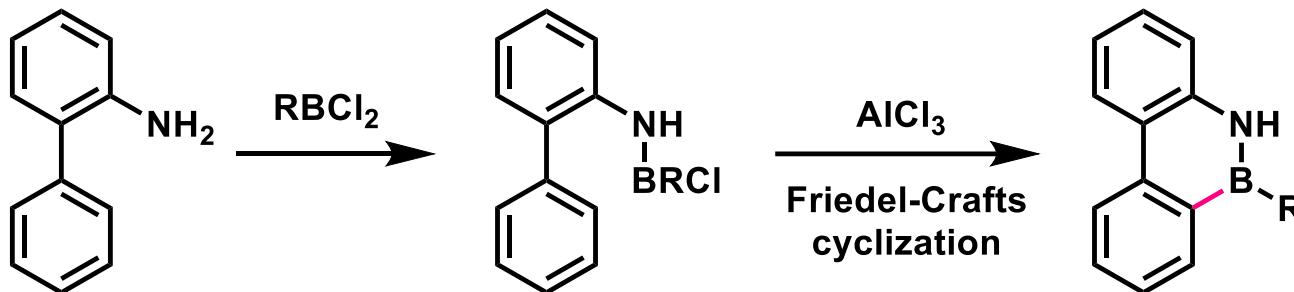


Wagner (2019)

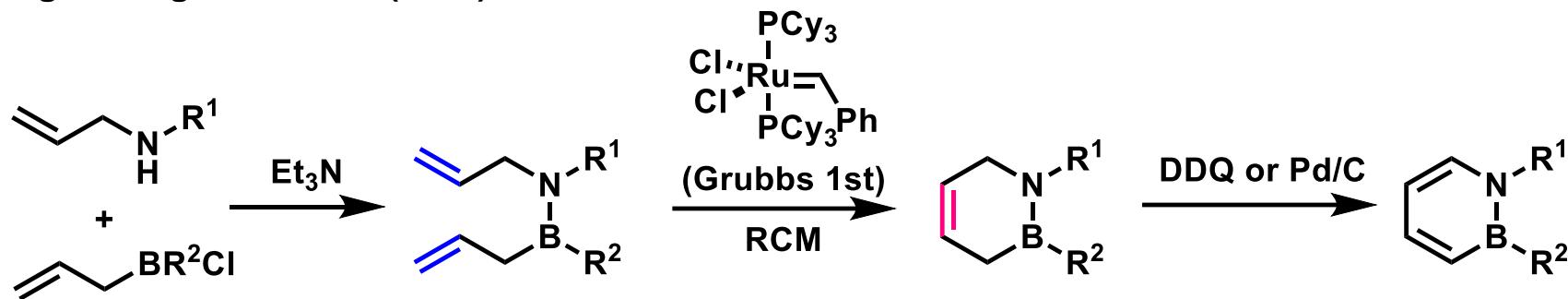
1) Abengózar, A.; G-García, P.; F-Rodríguez, A. M.; Sucunza, D.; Vaquero, J. J. *Advances in Heterocyclic Chemistry*. 2021.

# Synthesis of BN-arenes

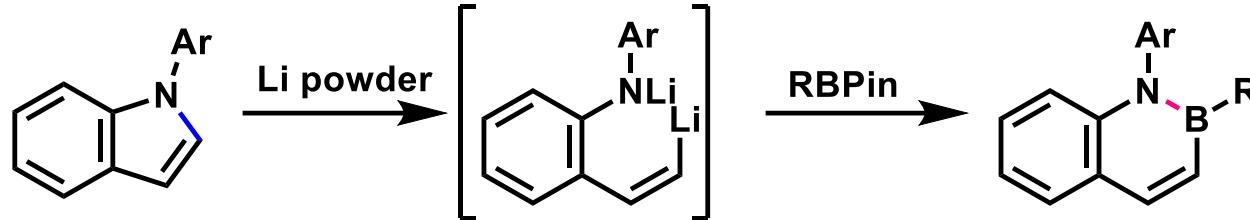
## 1. Borylative cyclization<sup>1)</sup>



## 2. Ring-closing metathesis (RCM)<sup>2)</sup>



## 3. Ring expansion<sup>3)</sup>



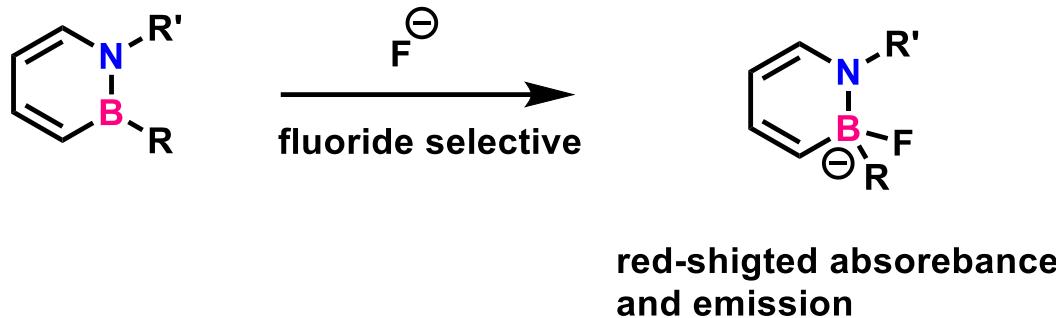
1) Dewar, S. J. M.; Kubba, P. V.; Pettit, R. *J. Chem. Soc.* **1958**, 3073.

2) Ashe, S. A. III.; Fang, X. *Org. Lett.* **2000**, 2, 2089.

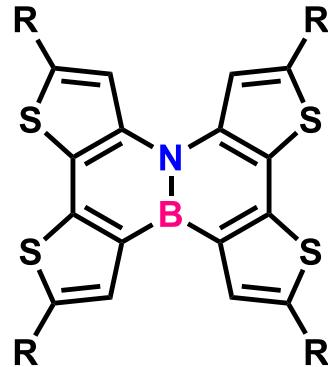
3) Tsuchiya, S.; Saito, H.; Nogi, K.; Yorimitsu, H. *Org. Lett.* **2019**, 21, 3855.

# Application of BN-arenes

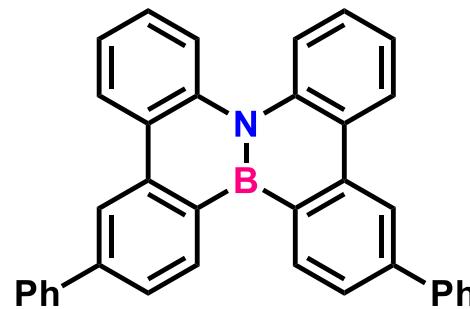
Fluoride sensor<sup>1)</sup>



Optoelectronic materials<sup>2), 3)</sup>



organic field-effect transistor (OFET)



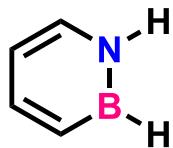
organic light-emmiting diode (OLED)

1) Agou, T.; Sekine, M.; Kobayashi, J.; Kawashima, T. *Chem. A Eur. J.* **2009**, *15*, 5056.

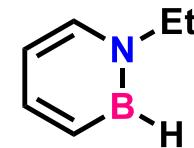
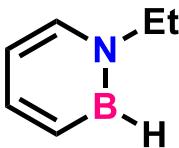
2) Wang, X.-Y.; Lin, H.-R.; Lei, T.; Yang, D.-C.; Zhuang, F.-D.; Wang, J.-Y.; Yuan, S.-C.; Pei, J. *Angew. Chem. Int. Ed.* **2013**, *52*, 3117.

3) Hashimoto, S.; Ikuta, T.; Shiren, K.; Nakatsuka, S.; Ni, J.; Nakamura, M.; Hatakeyama, T. *Chem. Mater.* **2014**, *26*, 6265.

# Biological Application of BN-arenes

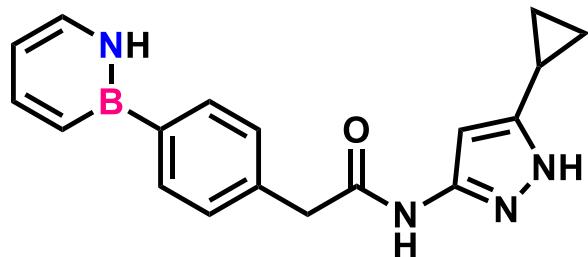


better ligand to T4 lysozome L99A<sup>1)</sup>



EbDH inhibitor<sup>2)</sup>

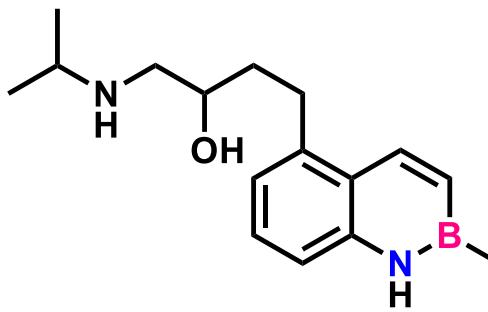
$IC_{50} = 2.8 \mu M$



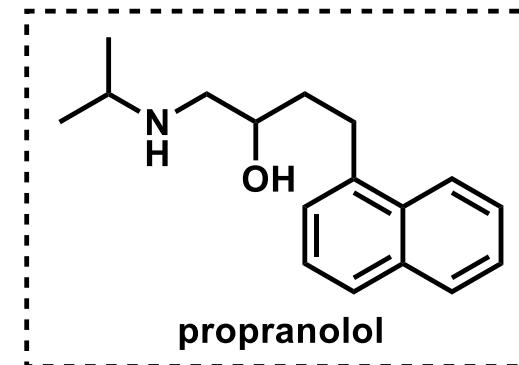
CDK2 inhibitor<sup>3)</sup>

$IC_{50} = 87 nM$

(all carbon analogus = 320 nm)



$\beta$ -blocker<sup>4)</sup>  
different ADMET profile  
from propranolol



propranolol

Biological application of BN arenes showed effectiveness especially in medicinal chemistry.

→Installing BN bond into biomolecules will lead to broader application.

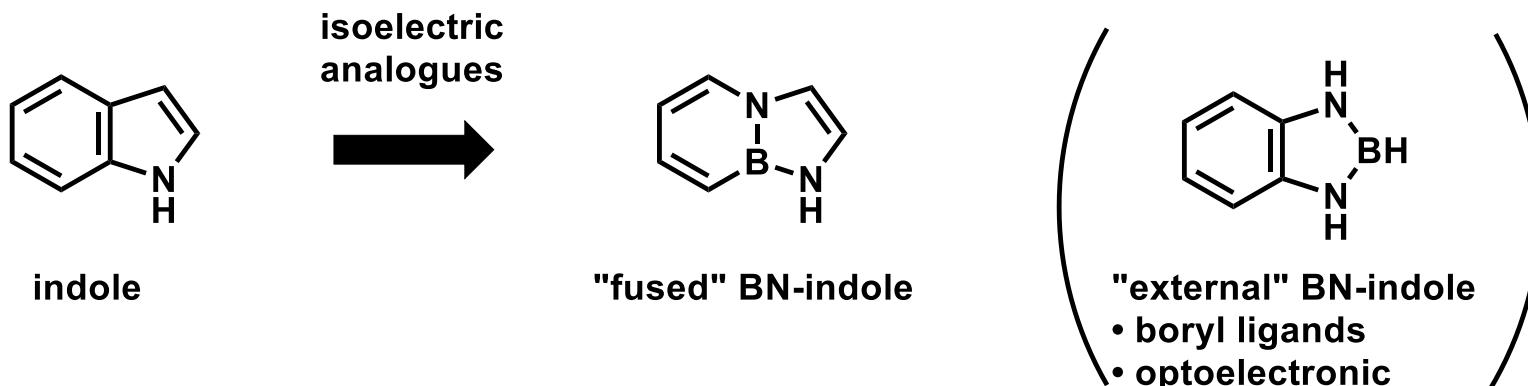
1) Liu, L.; Marwitz, V. J. A.; Matthews, W. B.; Liu, S.-Y. *Angew. Chem. Int. Ed.* **2009**, *48*, 6817.

2) Knack, H. D.; Marshall, L. J.; Harlow, P. G.; Dudzik, A.; Szaleniec, M.; Liu, S.-Y.; Heider, J. *Angew. Chem. Int. Ed.* **2013**, *52*, 2599.

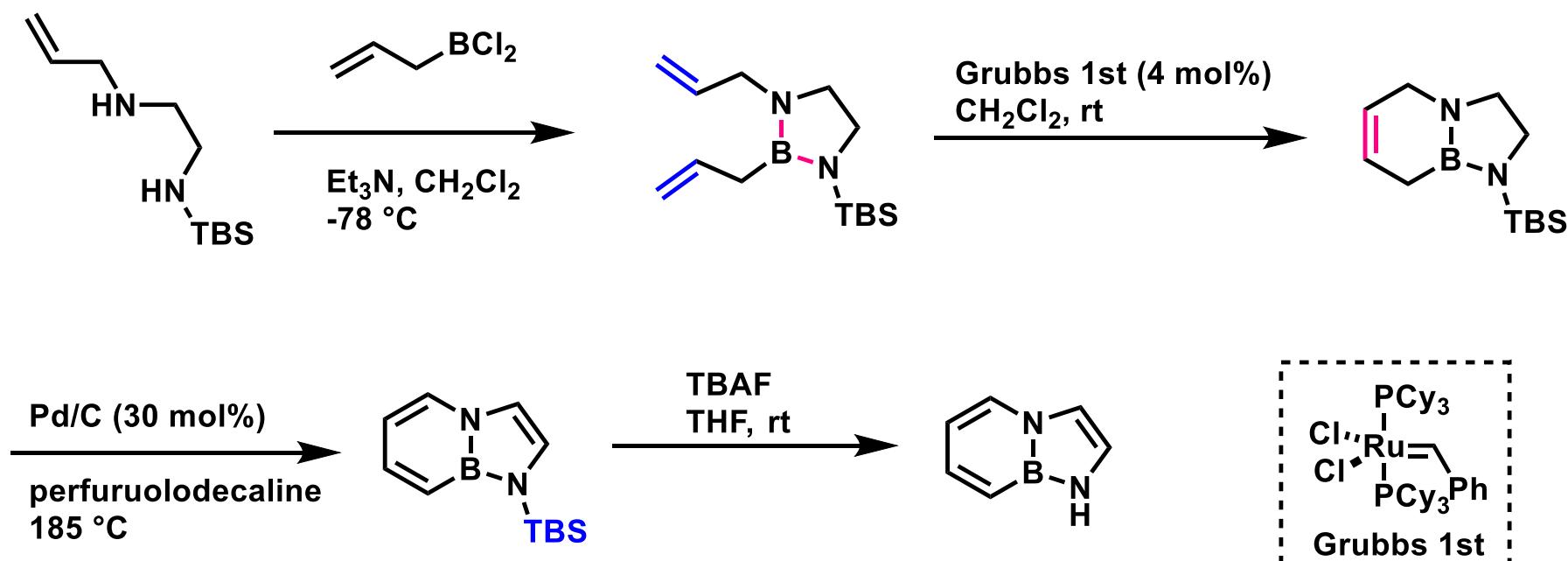
3) Zhao, P.; Nettleton, O. D.; Karki, G. R.; Zecri, J. F.; Liu, S.-Y. *ChemMedChem*, **2017**, *5*, 358.

4) Rombouts, R. J. F.; Tovar, F.; Austin, N.; Tresadern, G.; Trabanco, A. A. *J. Med. Chem.*, **2015**, *58*, 9287.

# BN-indoles<sup>1)</sup>



## synthesis of "fused" BN-indole<sup>2)</sup>

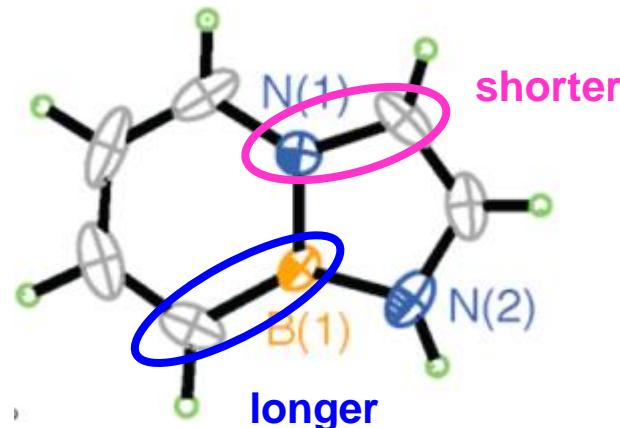


1) Abbey, R. E.; Liu, S.-Y. *Org. Biomol. Chem.* 2013, 11, 2060.

2) Abbey, R. E.; Zakharov, N. L.; Liu, S.-Y. *J. Am. Chem. Soc.* 2011, 133, 11508.

# Properties of BN-indole (1)

X-ray diffraction structure (crystal of  $\pi$ -complex with electron deficient arene)

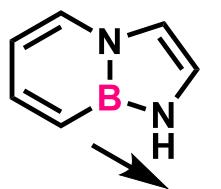


Highly planner and aromatic

Compared with natural indole

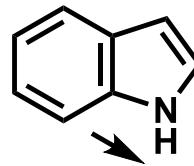
- N-C<sub>2</sub> bond (pink circled) was shorter
- B-C<sub>7</sub> bond (blue circled) was longer  
→ larger/shorter Atom radius of B/N

acidity of NH



pK<sub>a</sub>

~30



20.95

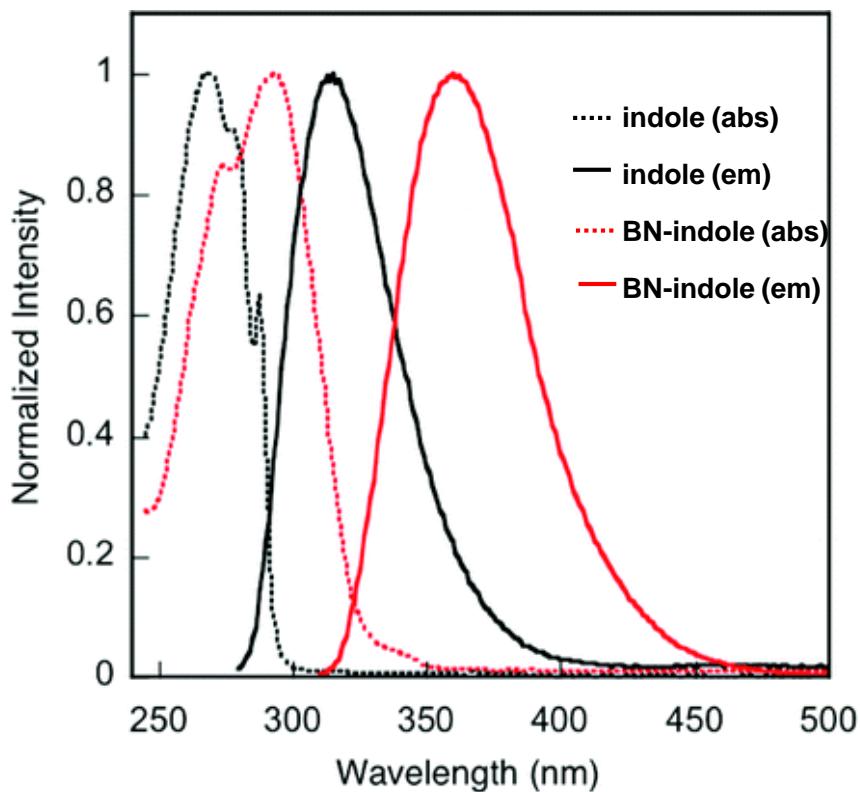
lower acidity

→ inductive effects exerted by the neighboring boron atom

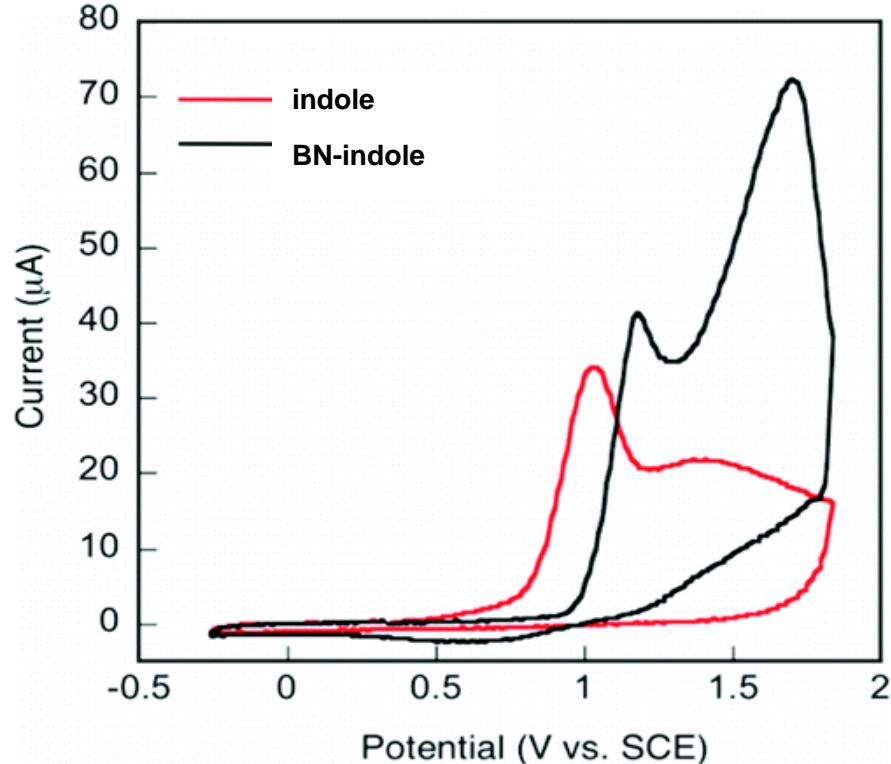
electronegativity  
B : 2.04 C : 2.55

# Properties of BN-indole (2)

## Fluorescence



## Cyclic voltammetry



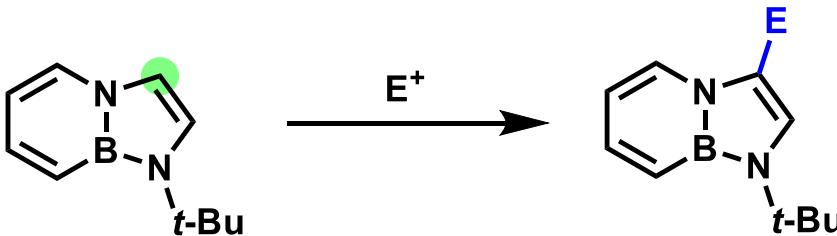
	$\lambda_{\text{abs}}$ (nm)	$\lambda_{\text{em}}$ (nm)	Stokes Shift ( $\text{cm}^{-1}$ )	$E$ (V)
indole	268	315	5570	1.18
BN-indole	293	360	6350	1.04

The wavelengths were red shifted and the oxidation potential was lower.

→ indicate higher energy HOMO and smaller HOMO-LUMO gap.

# Reactivity of BN-indole (1)

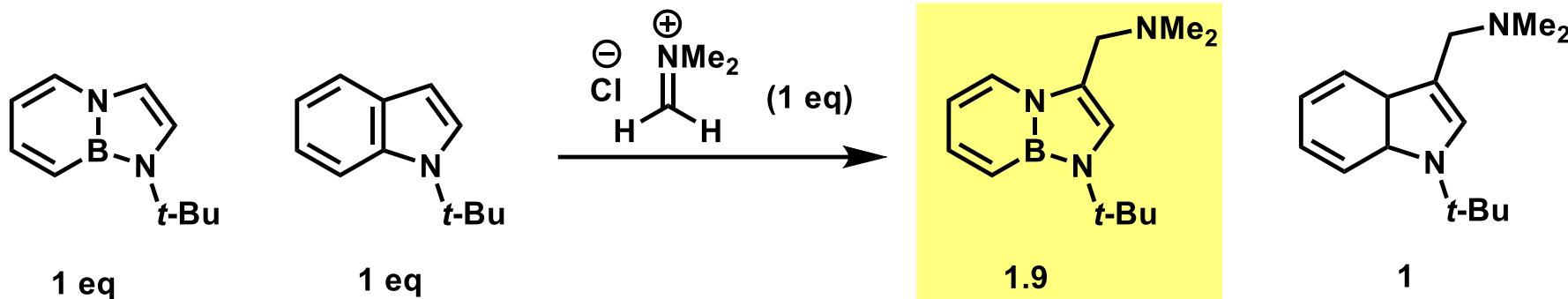
electrophilic aromatic substitution (EAS) at 3-position



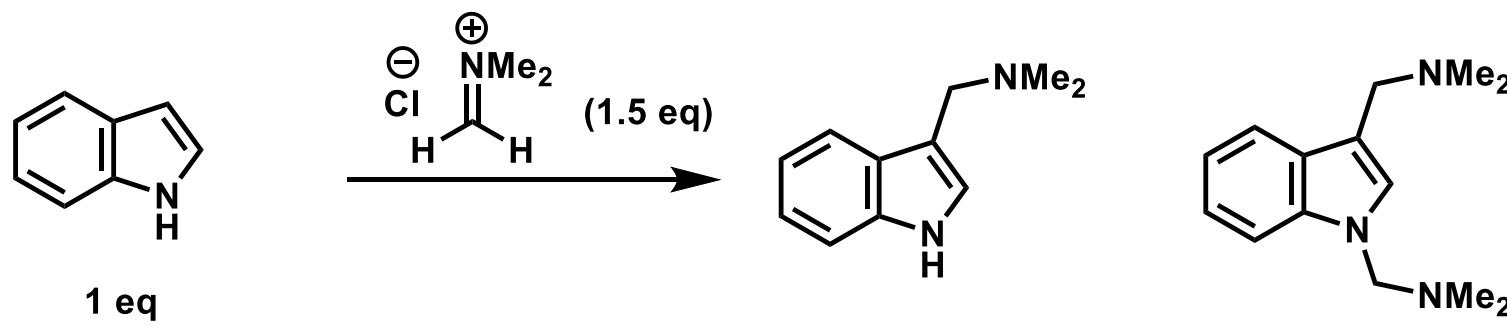
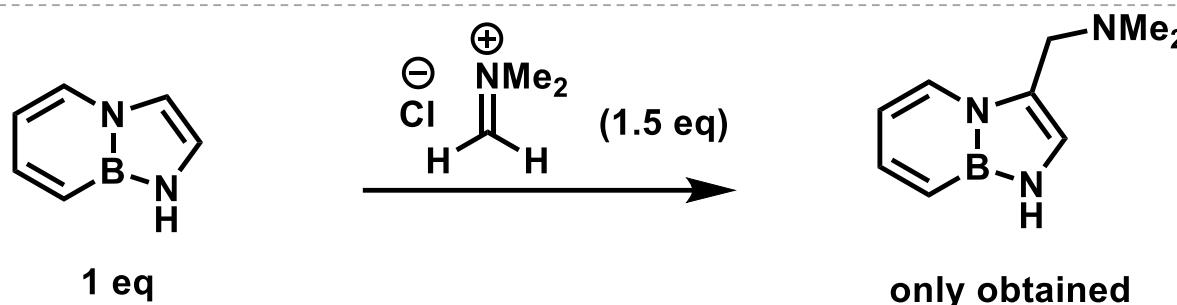
E <sup>+</sup>	E		E <sup>+</sup>	E	yield
Br <sub>2</sub>		39	CD <sub>3</sub> OD/D <sub>2</sub> O		39 <sup>b</sup>
		53			23 <sup>c</sup>
		57			

<sup>a</sup> ZrCl<sub>4</sub> was used as catalyst. <sup>b</sup> ~80% D enrichment <sup>c</sup> Et<sub>2</sub>AlCl was used as catalyst.

## **Reactivity of BN-indole (2)**



**3-position of BN indole is more nucleophilic.**



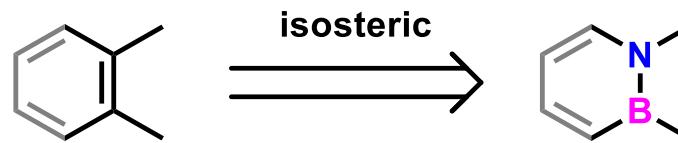
**1-position of BN-indole is less nucleophilic.**

## 2.1 :

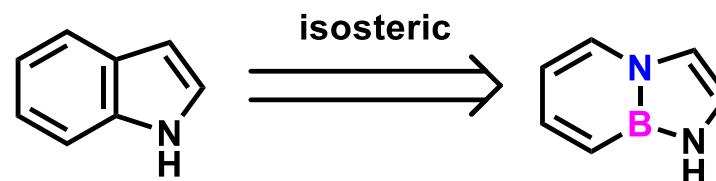
1

- 1) Chrostowska, A.; Xu, S.; Mazière, A.; Boknevitz, K.; Li, B.; Abbey, E. R.; Dargelos, A.; Graciaa, A.; Liu, S.-Y. *J. Am. Chem. Soc.* **2014**, 136, 11813.

# Short Summary



**BN isosterism is effective method for altering electronical character without altering steric profiles.**



**BN-indole showed different properties from natural indole**

- red shifted UV absorption and fluorescence
- higher redox potential
- higher and more selective reactivity with electrophile

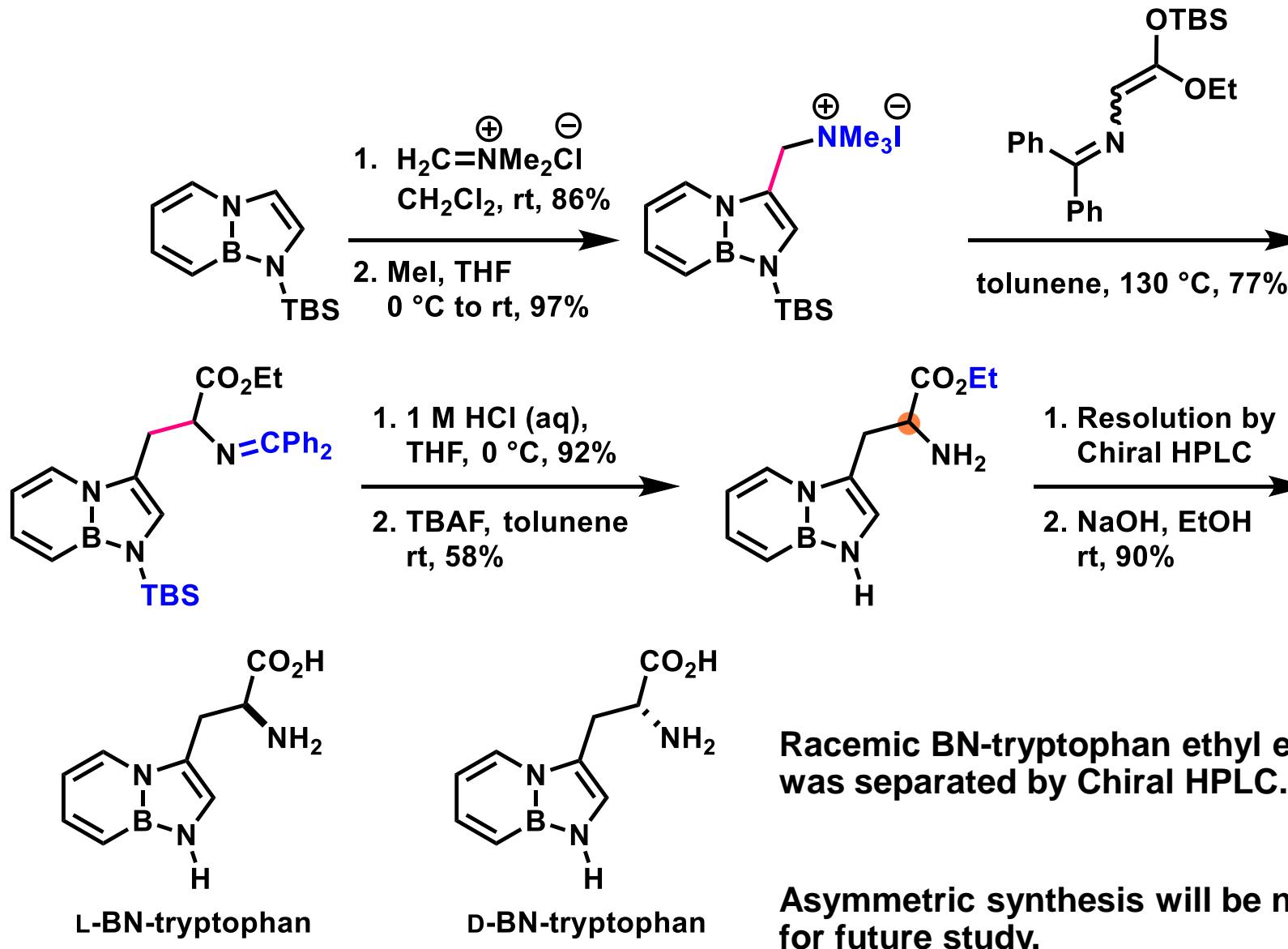
# Contents

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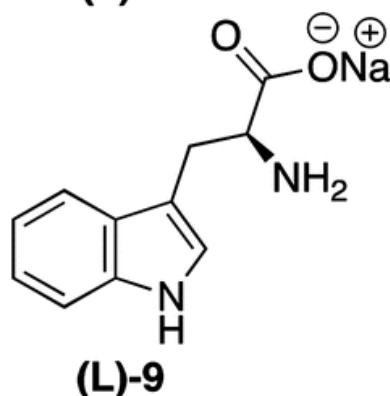
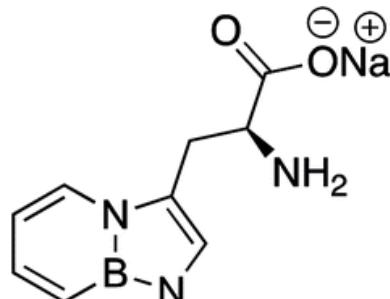
2. Background

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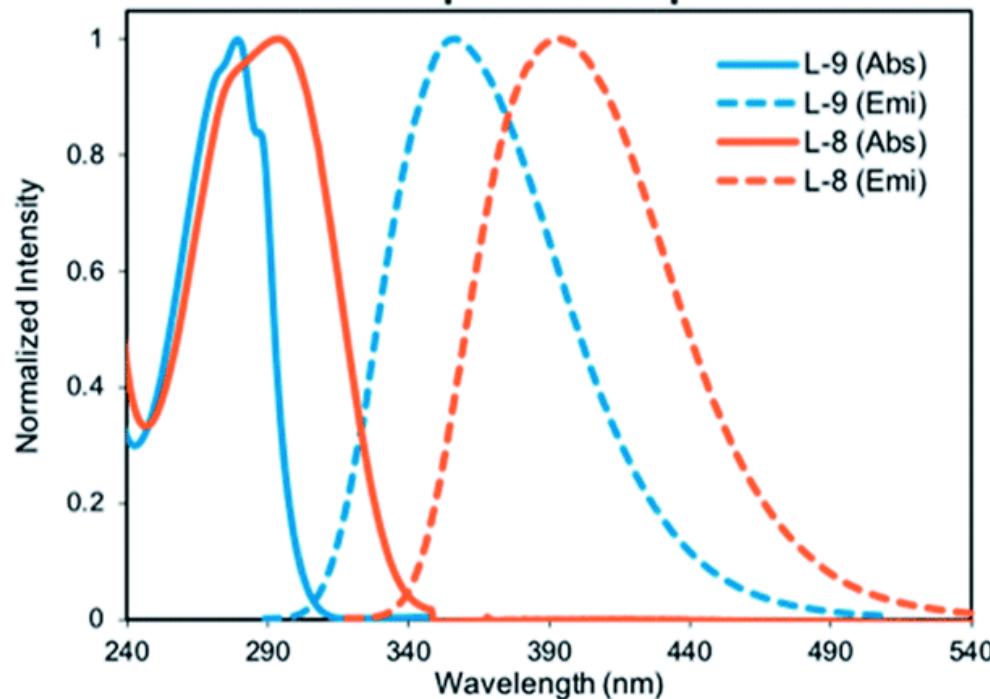
# Synthesis of BN-tryptophan



# Photophysical Properties of BN-Trp



Absorbance and Fluorescence Spectra  
of L-trp and BN-trp

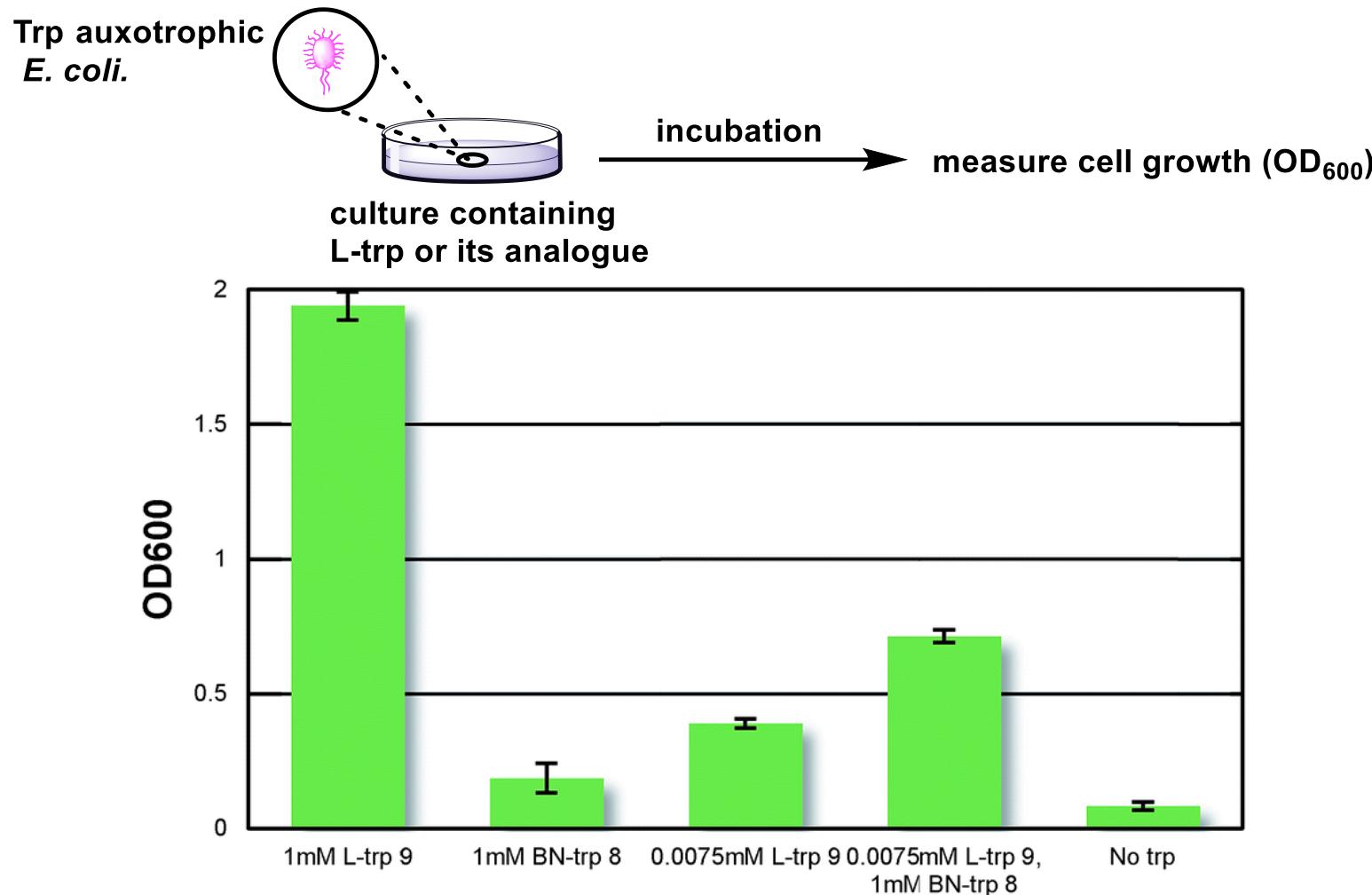


	$\lambda_{\text{abs}}$ (nm)	$\lambda_{\text{em}}$ (nm)	Stokes Shift ( $\text{cm}^{-1}$ )
Trp (L)-9	279	356	7752
BN-Trp (L)-8	295	394	8633

The wavelengths of absorption and emission were red shifted.

in water, pH 10

# Incorporation as a Surrogate of Trp

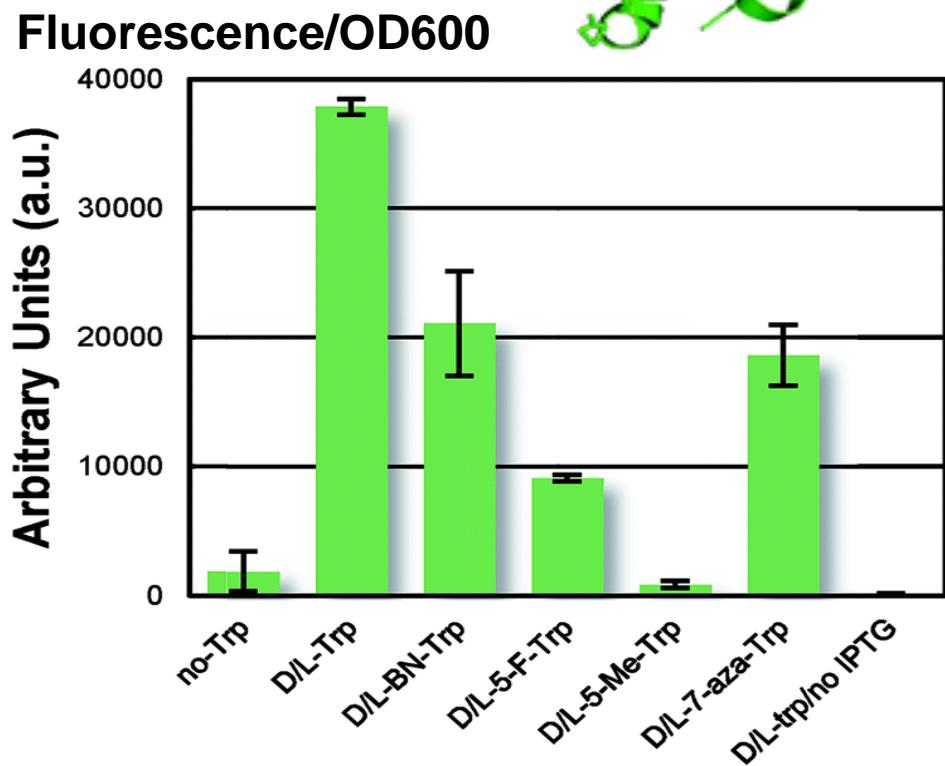
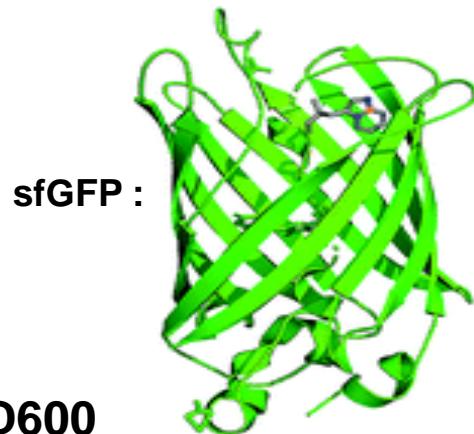
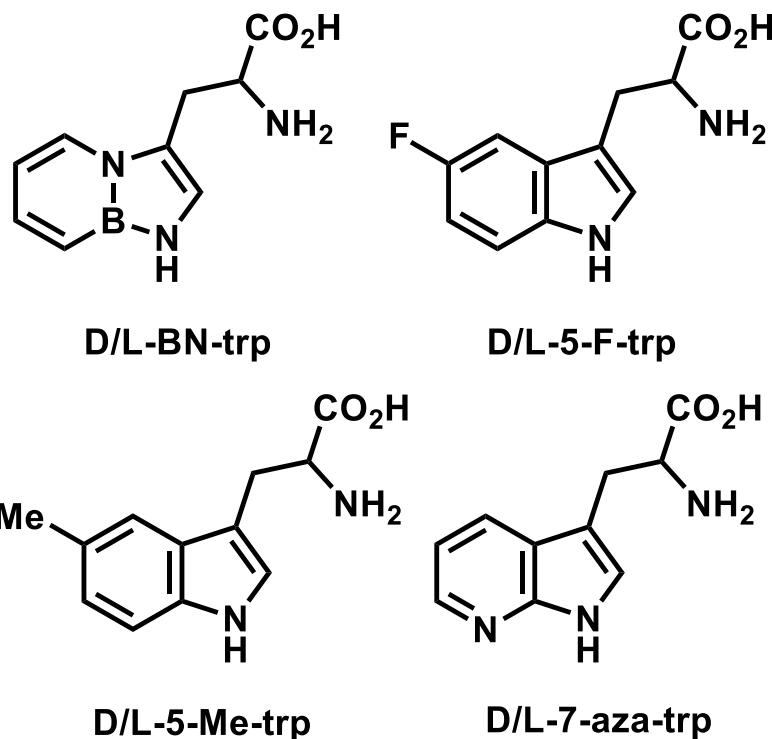
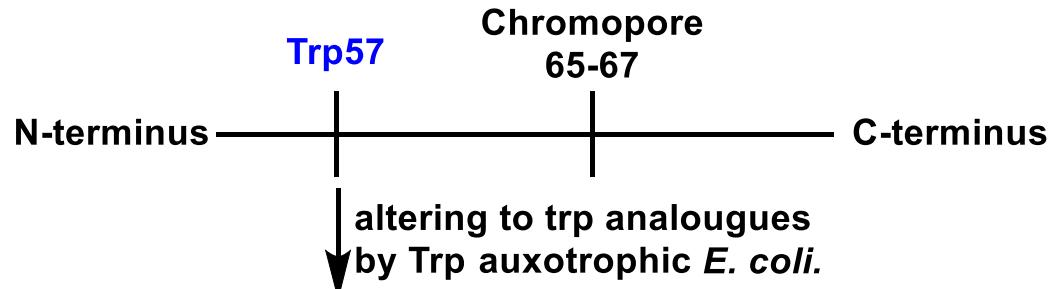


BN-trp has quite lower but sure cell growth ability.

- accepted as a substrate for the endogenous tryptophanyl-tRNA synthetase
- not tolerated for proteome-wide replacement

# Comparison with Other Trp Analogues

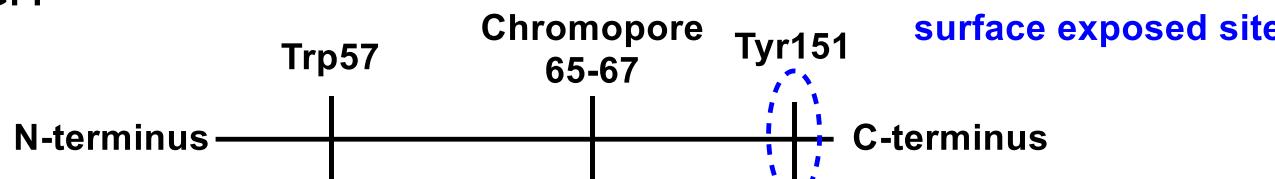
# **super folding green fluorescent protein (sfGFP)**



## BN-Trp shows similar or improved incorporation efficiency relative to other Trp analogues.

# Change of Reactivity

sfGFP



mutant sfGFP



protein A (Trp)

MS: 27620 [ $M_A$ ] $1 \text{ mM H}_2\text{O}_2, 5 \text{ min}$ 

protein B (BN-Trp)

MS: 27621 [ $M_B$ ]

protein A (Trp):

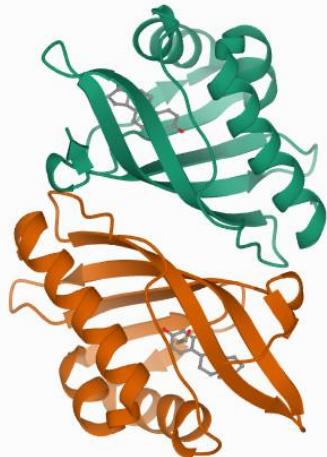
 $[M_A]$  was mainly detected.  $[M+O]$  was also detected.

protein B (BN-Trp):

 $[M_B]$  was almost lost.  $[M+O]$  was mainly detected.

Although protein A and B only differ by the presence of BN-trp, the reactivity against oxidant was quite different.

# Change of Fluorescence

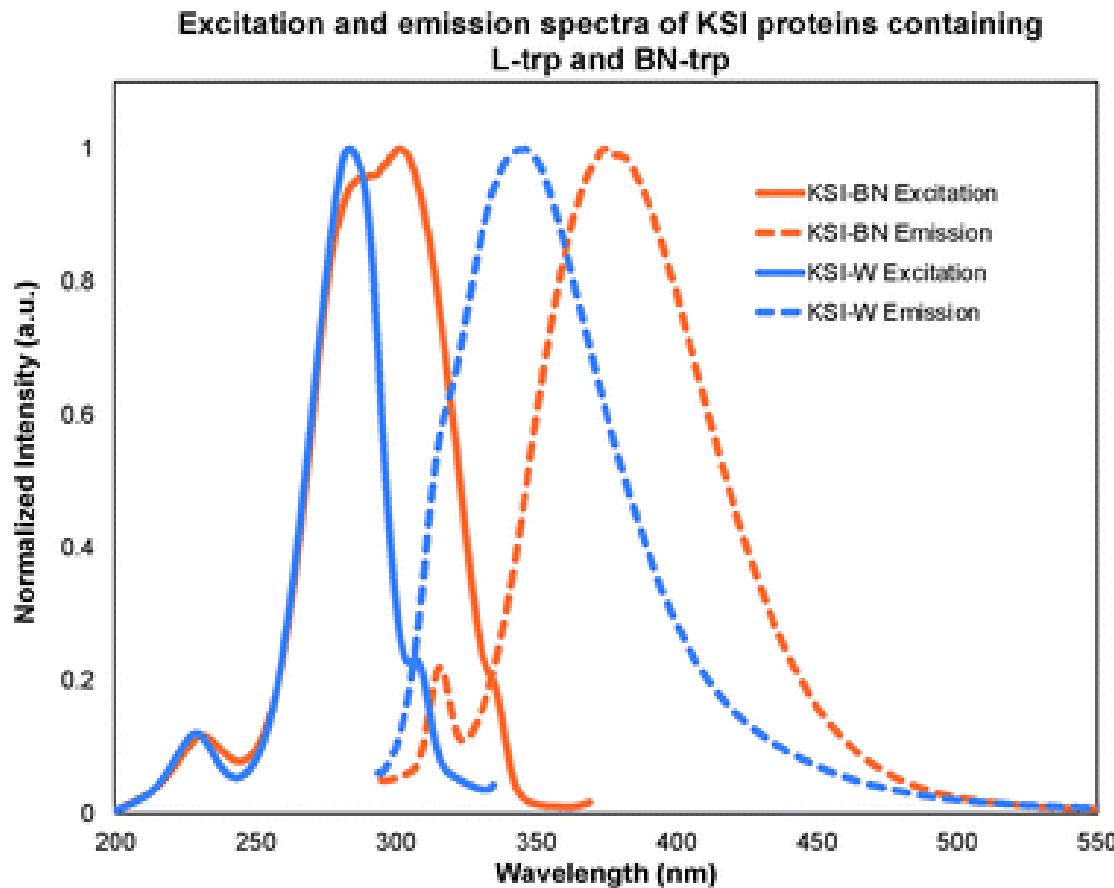


**Ketosteroid isomerase (KSI)**

- homo-dimeric protein
- two Trp residues in each monomer

**KSI-W : wild type**

**KSI-BN : BN-trp analogue**



The fluorescence of KSI-BN was red shifted.

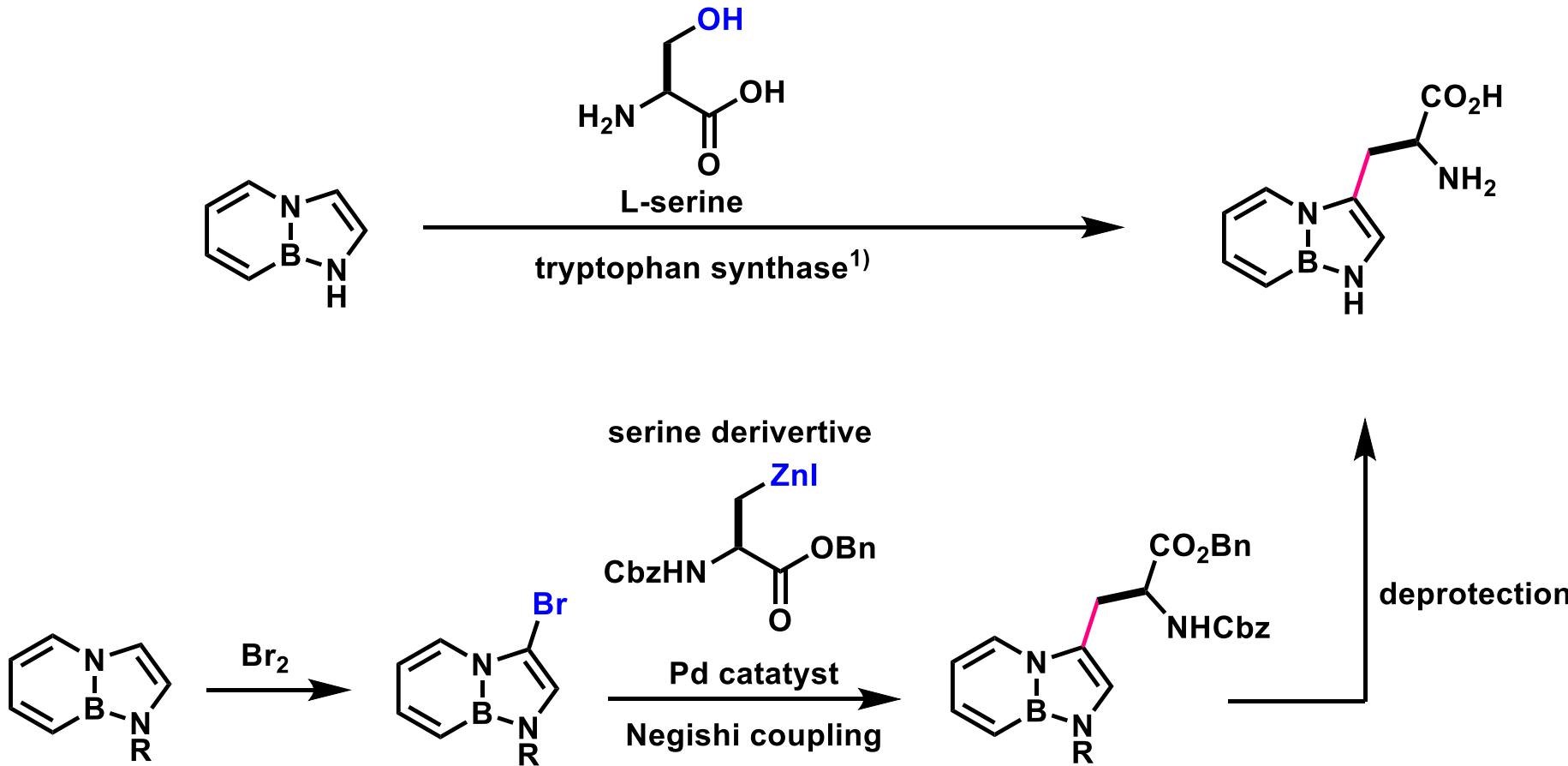
→ The Tyr fluorescence peak (316 nm) could be distinguished from the BN-Trp peak.

	$\lambda_{\text{ex}} \text{ (nm)}$	$\lambda_{\text{em}} \text{ (nm)}$
KSI-W	284	342
KSI-BN	285	372

# Future Study (1)

For further study of BN-tryptophan, methods for asymmetric synthesis of BN-trp will be needed.

My proposal methods are shown below.

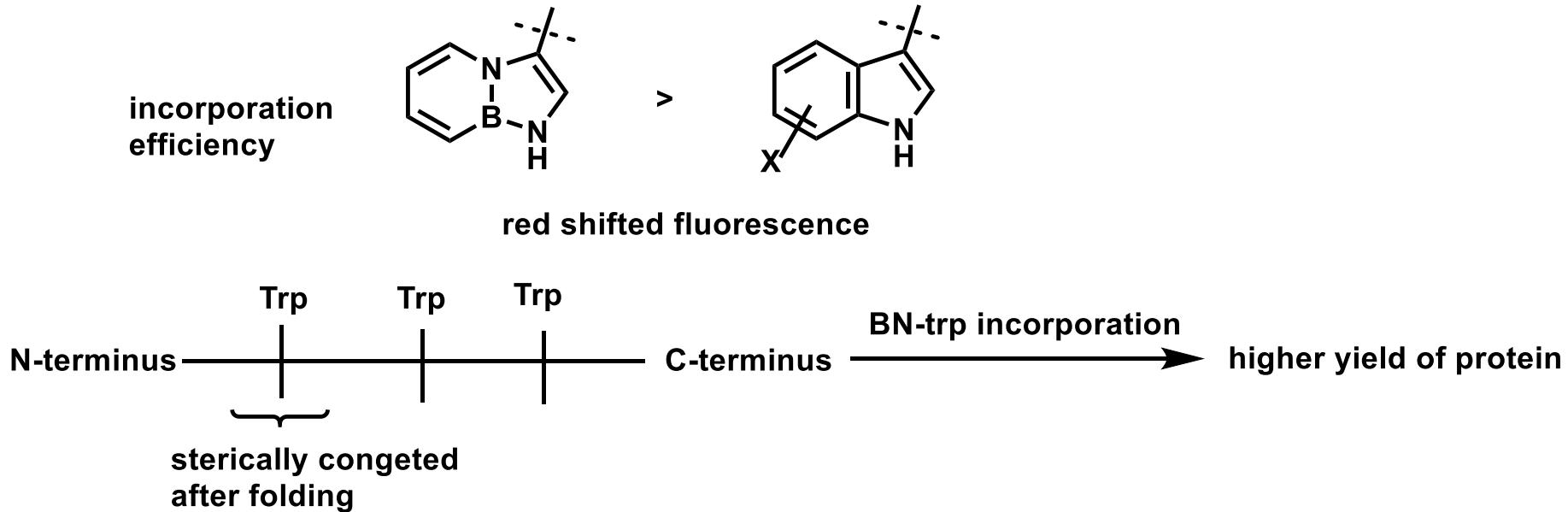


1) Sloan, J. M.; Phillips, S. R. *Bioorg. Med. Chem. Lett.* **1992**, 2, 1053.

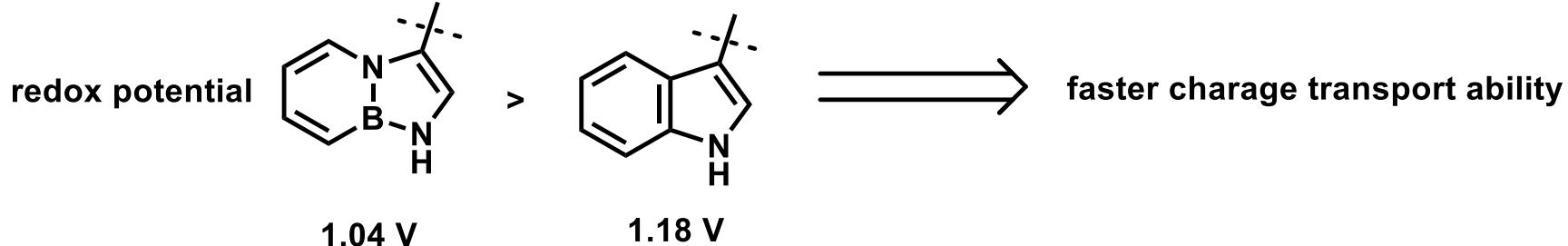
# Future Study (2)

Application of BN-tryptophan will

## 1. Probing proteins possessing multi Trps or Trp at sterically hindered position

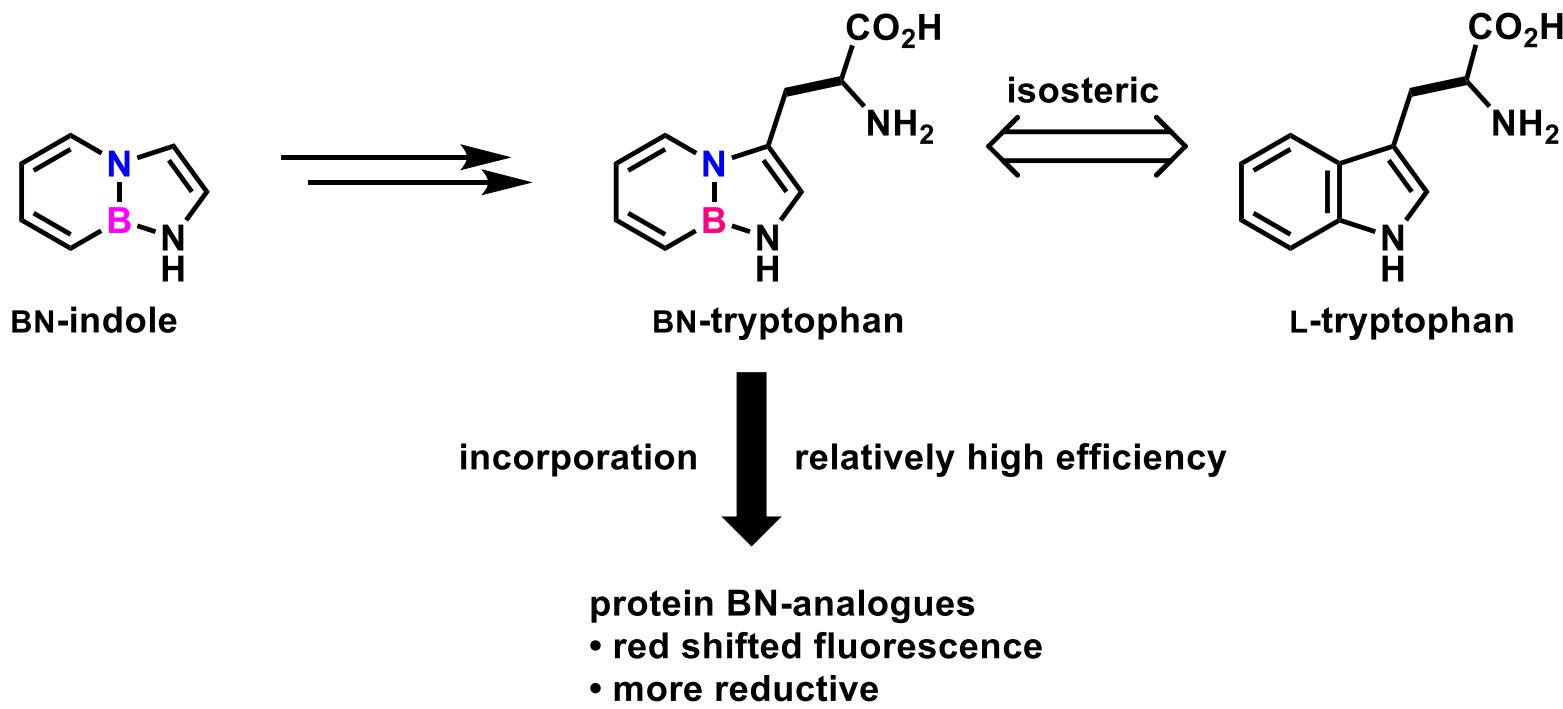


## 2. efficient charge transporter<sup>1)</sup>



1) Shih, C.; Museth, K. A.; Abrahamsson, M.; Blanco-Rodriguez, M. A.; DiBilio, J. A.; Sudhamsu, J.; Crane, R. B.; Ronayne, L. K.; Towrie, M.; Vlcek, A. Jr.; Richards, H. J.; Winkler, R. J.; Gray, B. H. *Science*. 2008, 320, 1760.

# Summary



**BN-Tryptophan was synthesized and incorporated into proteins.**

**Protein BN-analogues showed distinct photophysical property and reactivity.**

**BN-tryptophan will be expected to explore new chemical space for UAAs.**