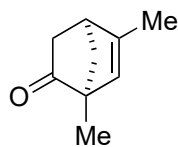


Problem Session (6)

2021/5/15 Takumi Fukuda

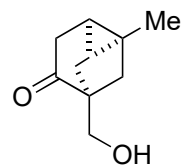
Please provide the reasonable reaction mechanisms and explain the stereoselectivities.

(1)



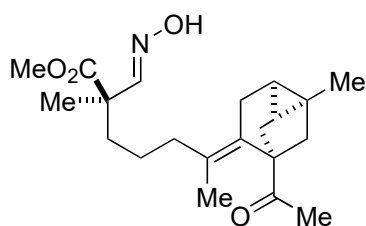
1-1

- $i\text{-Pr}_2\text{NH}$ (1.15 eq), $n\text{-BuLi}$ (1.1 eq)
THF, $-78\text{ }^\circ\text{C}$;
TMSCl (1.4 eq), $-78\text{ }^\circ\text{C}$ to rt
 - ZnEt_2 (2.0 eq), CH_2I_2 (2.0 eq)
 Et_2O , $0\text{ }^\circ\text{C}$ to rt
 - $n\text{-Bu}_4\text{NOH}$ (1.3 eq), MeOH, rt;
 $\text{H}_2\text{NOMe}\cdot\text{HCl}$ (1.5 eq), rt, 65% (3 steps)
-
- $\text{PhI}(\text{OAc})_2$ (2.0 eq), $\text{Pd}(\text{OAc})_2$ (15 mol%)
AcOH/Ac₂O (1/1), $90\text{ }^\circ\text{C}$, 71%
 - 1.0 M aq. HCl (7.0 eq)
acetone, $60\text{ }^\circ\text{C}$, 92%



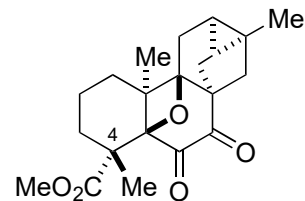
1-2

(2)



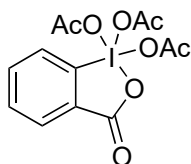
2-1

- $\text{PhI}(\text{OAc})_2$ (1.5 eq), MeOH, $0\text{ }^\circ\text{C}$;
extract with toluene^a; $120\text{ }^\circ\text{C}$, 52%
 - $\text{Me}_3\text{O}\cdot\text{BF}_4$ (2.5 eq), CH_2Cl_2 , rt;
TMSOTf (5.0 eq), Et₃N (8.0 eq), rt, 69%
 - H_2 (1 atm), Pd/C (140+280 wt%)
EtOAc/AcOH (5/1), $80\text{ }^\circ\text{C}$, 72%
 - KOt-Bu (20 eq), O₂ (bubbling)
THF, $-78\text{ }^\circ\text{C}$; PPh₃ (5.0 eq)
 $-78\text{ }^\circ\text{C}$ to rt, 72% (dr = 14:1^b)
-
- Dess-Martin periodinane (3.0 eq)
 $t\text{-BuOH}$ (3.0 eq), CH_2Cl_2 , rt;
SiO₂ (725 wt%)
hexane/EtOAc (3/1), rt, 74%
 - $\text{PhI}(\text{OH})\text{OTs}$ (10 eq)
NaHCO₃ (10 eq), CH_2Cl_2 , rt, 72%



(-)-Mitrephorone A (2-2)

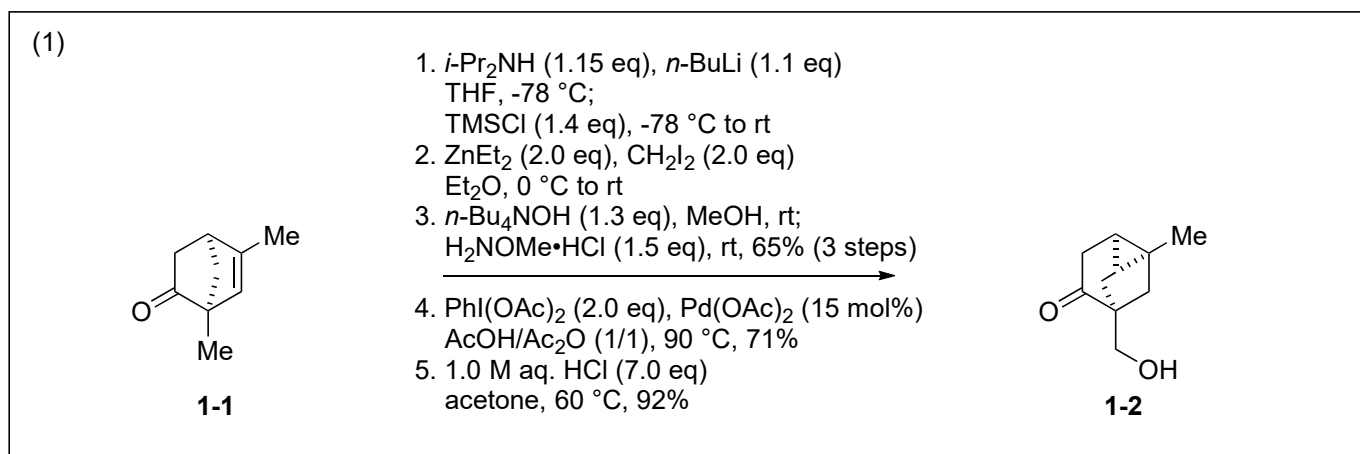
- a) After addition of saturated aq. Na₂S₂O₃/saturated aq. NaHCO₃/H₂O (1/1/2), the resultant mixture was extracted with toluene (x2). Combined organic layers were washed with brine/H₂O (1/1) and dried over Na₂SO₄. The filtrate was heated to $120\text{ }^\circ\text{C}$.
- b) The configuration of the newly formed stereogenic center was not assigned.



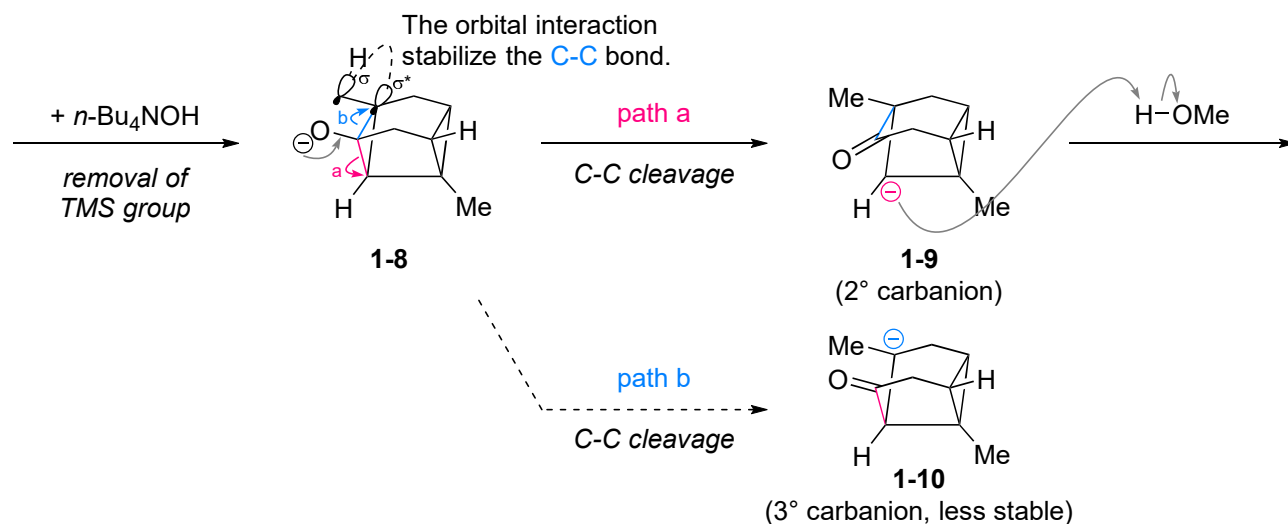
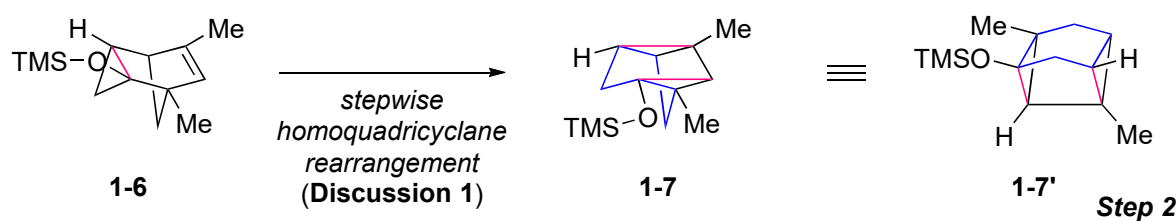
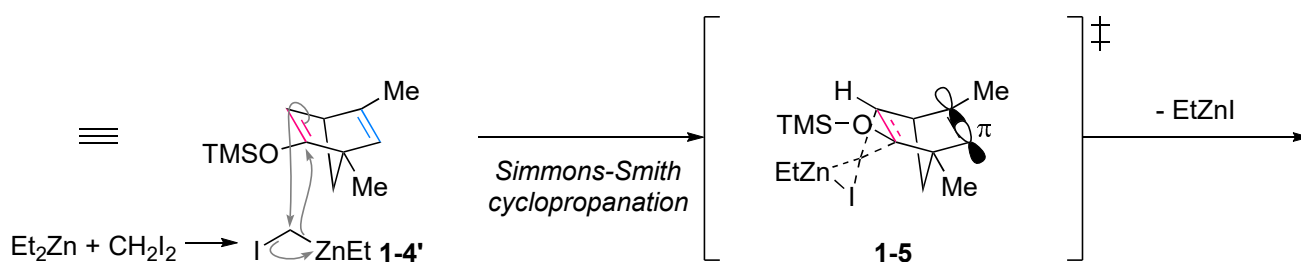
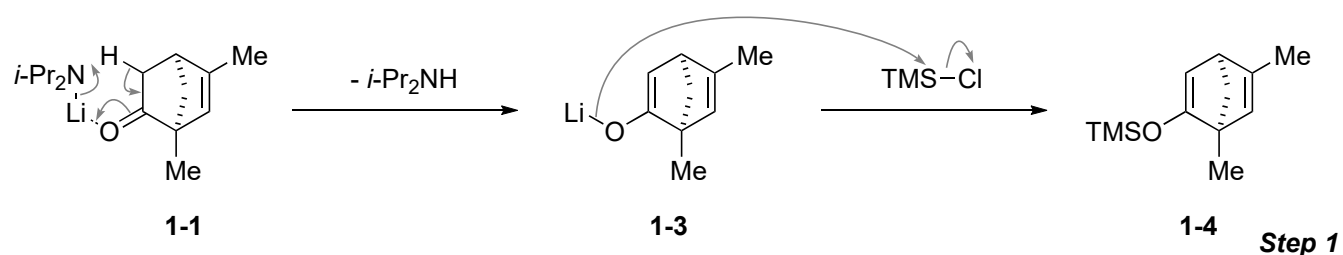
Dess-Martin periodinane

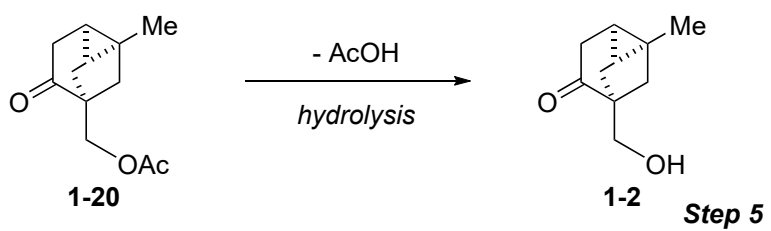
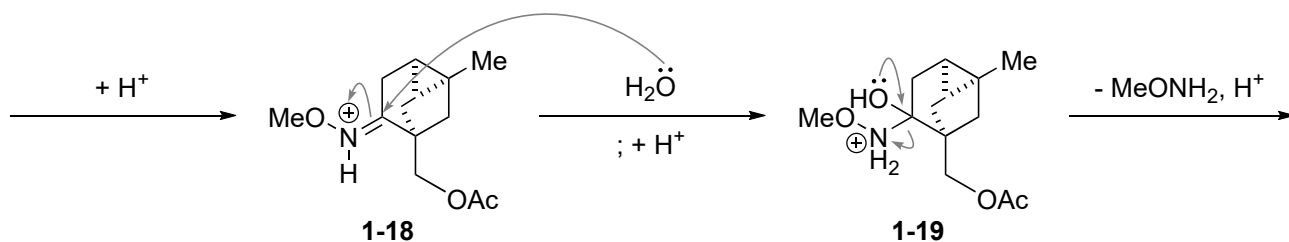
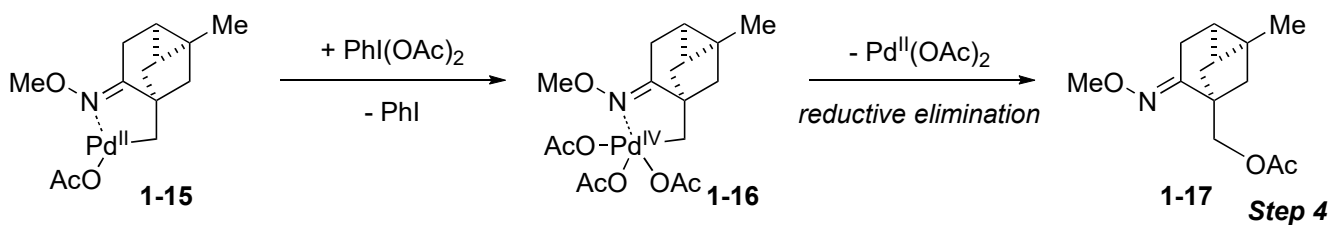
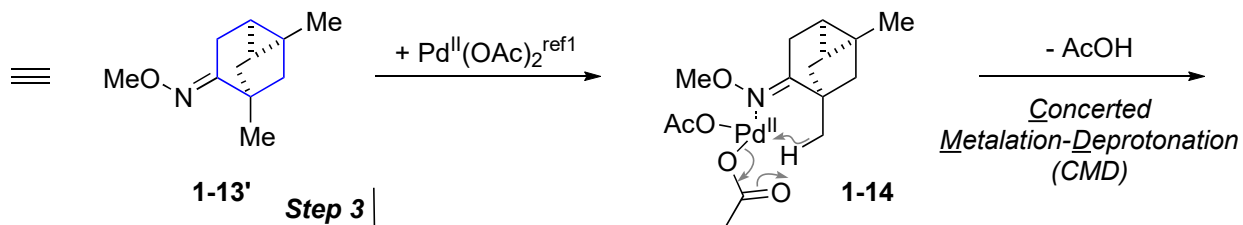
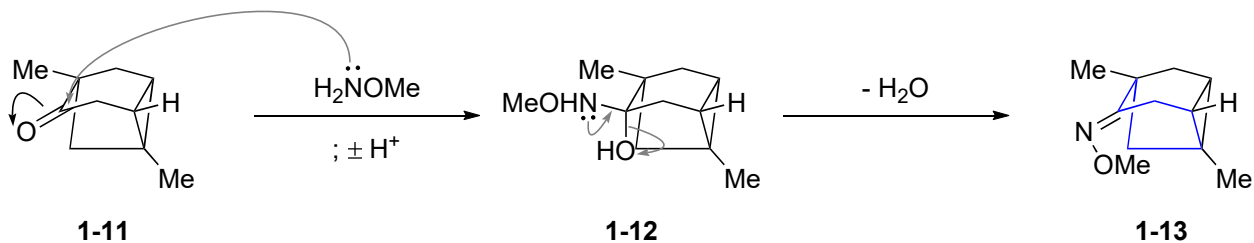
Topic: Total synthesis of (-)-Mitrephorone A by Carreira's group

1-1. Reaction mechanism



Schneider, M.; Richter, M. J. R.; Krautwald, S.; Carreira, E. M. *Org. Lett.* **2019**, *21*, 8705.





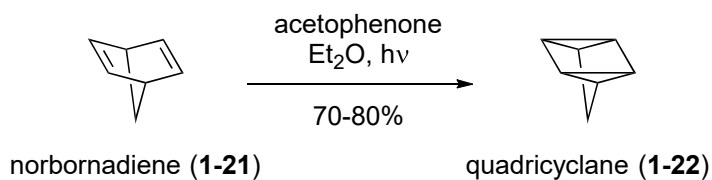
1-2. Discussion

1-2-1. Homoquadricyclane rearrangement

1-2-1-1. Synthesis and reactivity of quadricyclane

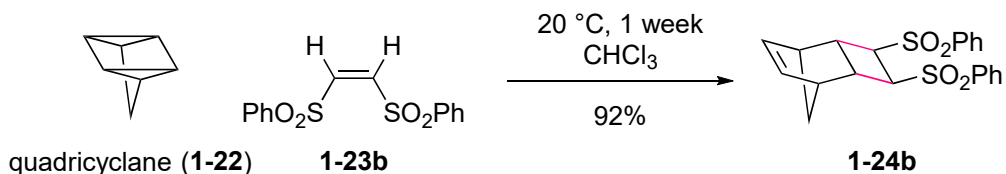
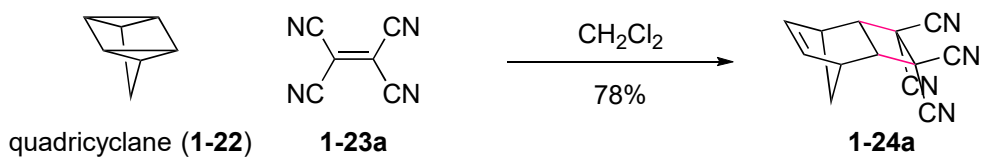
◆ Synthesis of quadricyclane

Smith, C. D. *Org. Synth.* **1971**, *51*, 133.

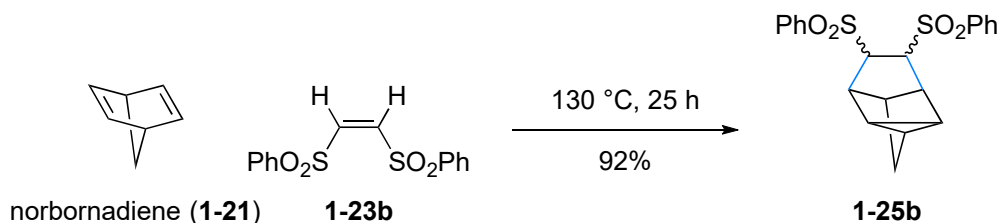
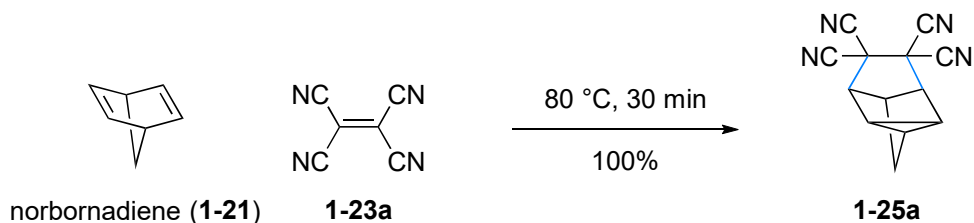


◆ Reactivity of quadricyclane

Petrov, V. A.; Vasil'ev, N. V. *Current. Org. Syn.* **2006**, *3*, 215.



Quadricyclane (**1-22**) reacts with variety of electron deficient olefins stereoselectively with the formation of *exo*-tricyclonenes.

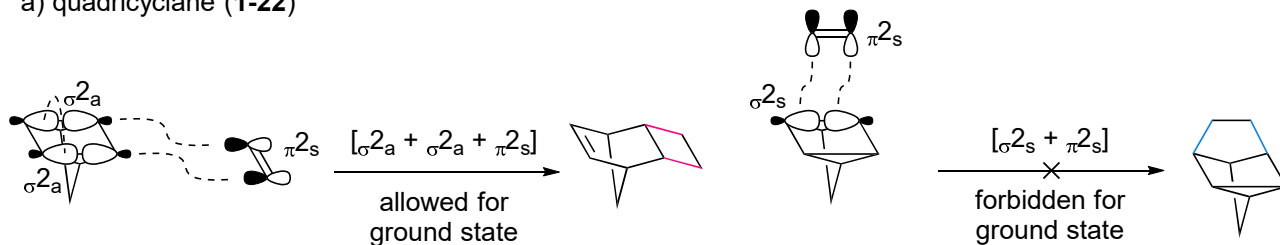


On the otherhand, the cycloaddition reactions of norbornadiene (**1-21**) and electron deficient olefins usually proceed at higher temperature and result in selective formation of tetracyclonanes (Also see 150509_PS_Shunichiroh_KATOH_Woodward_Hoffmann_rules).

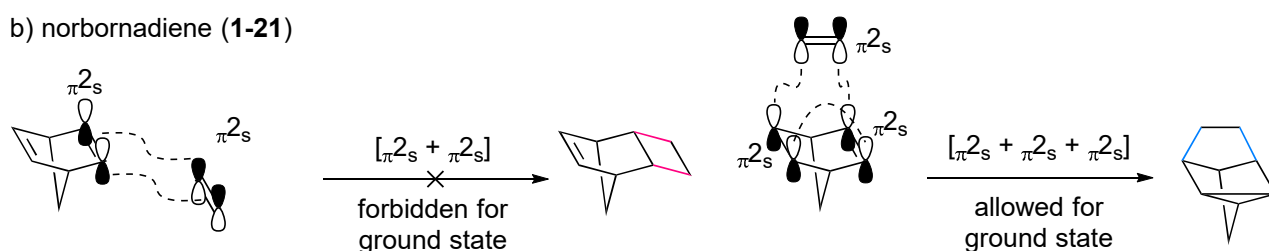
◆ Rationale for reactivity

These reactivity can be explained by Woodward-Hoffmann rules.

a) quadricyclane (**1-22**)

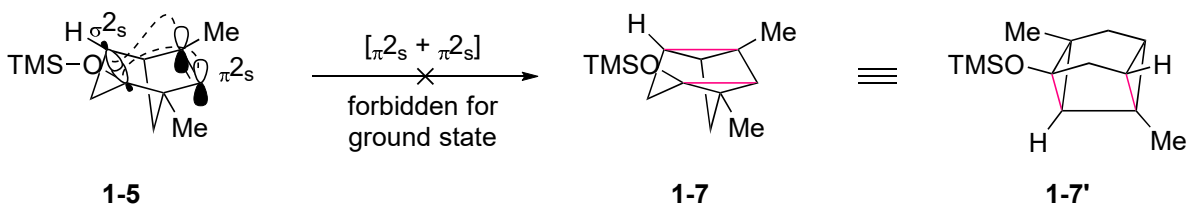


b) norbornadiene (**1-21**)

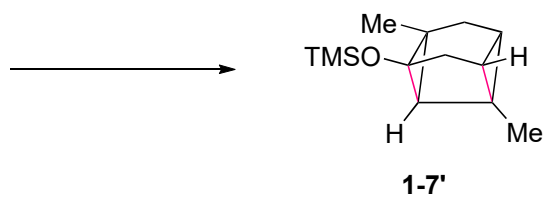
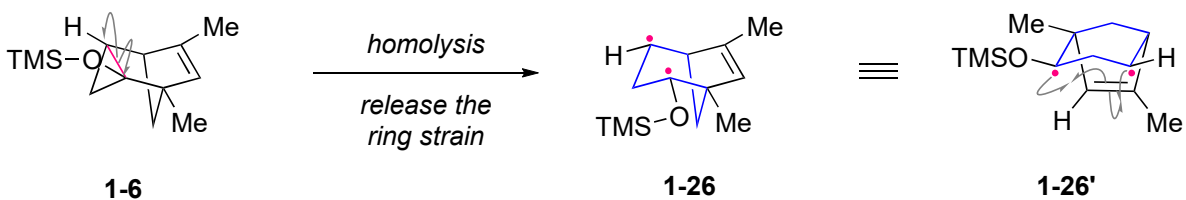


1-2-1. Homoquadricyclane rearrangement

1-2-2. Stepwise vs concerted

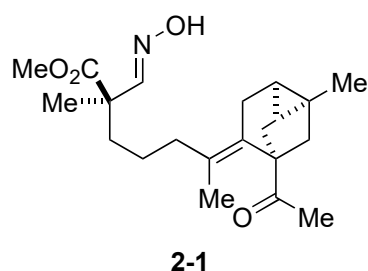


The reaction proceeds through a stepwise mechanism.

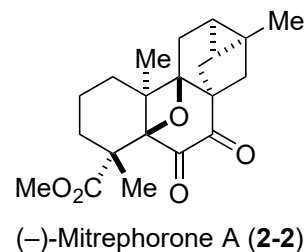


2-1. Reaction mechanism

(2)

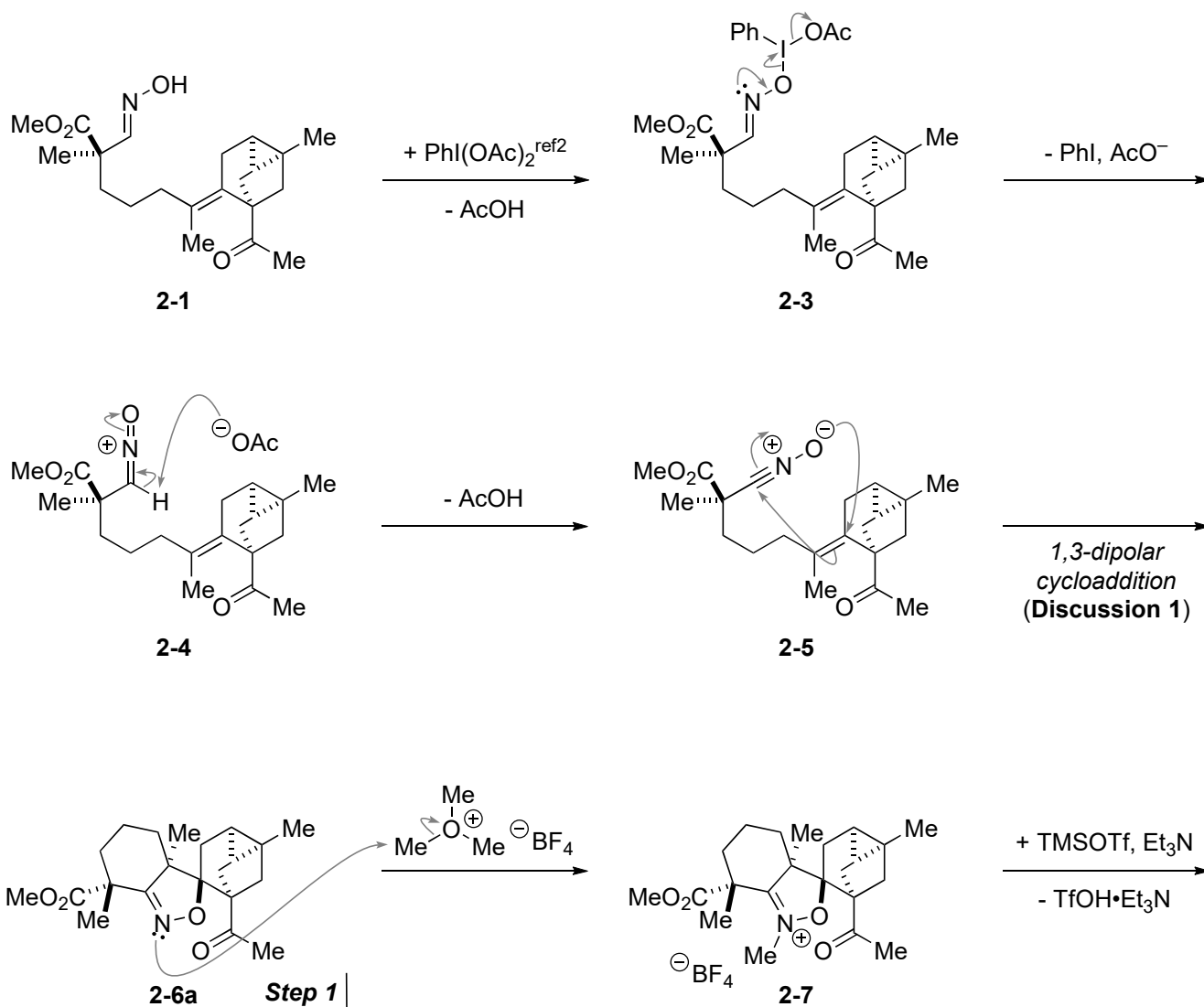


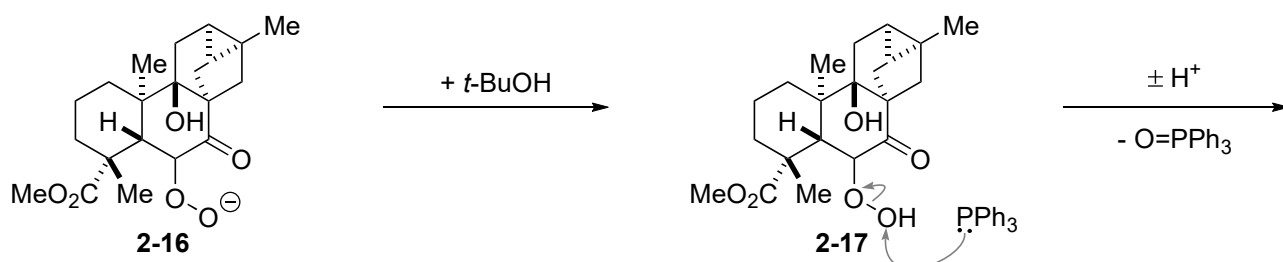
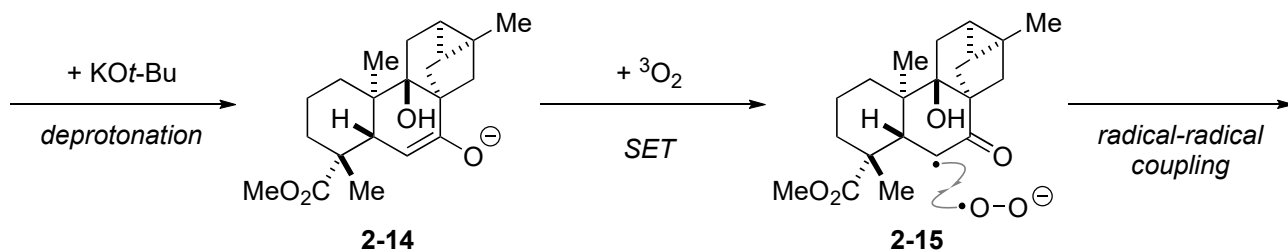
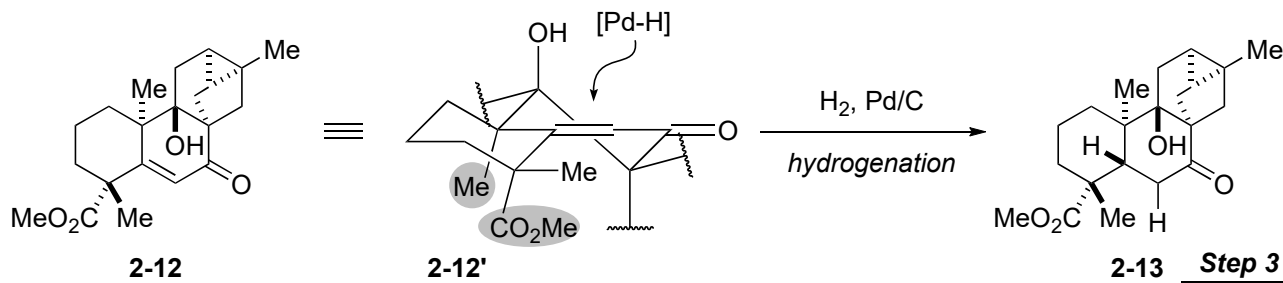
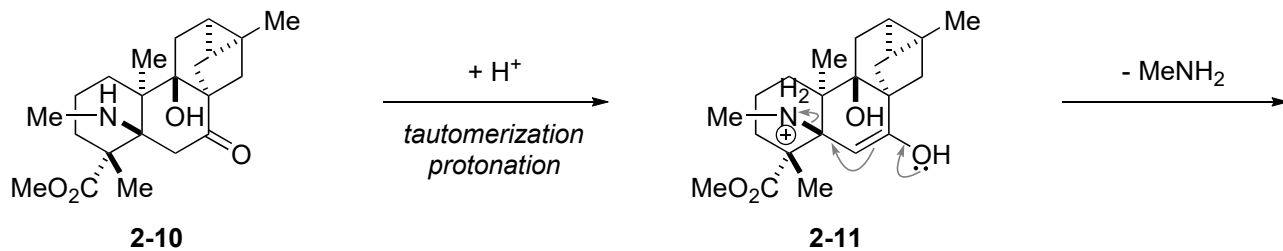
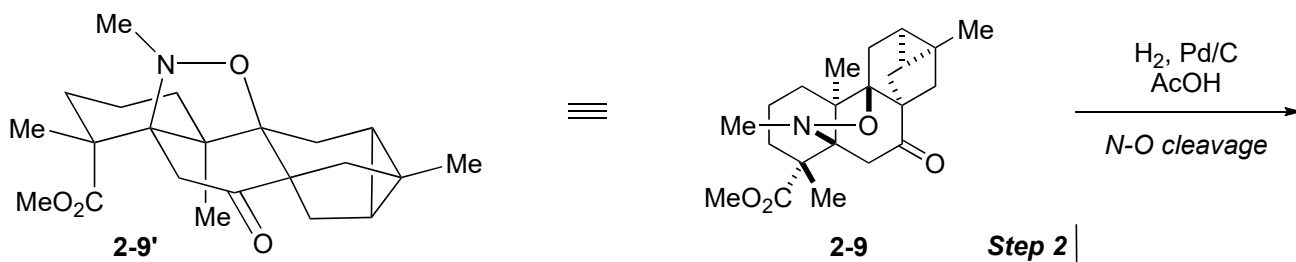
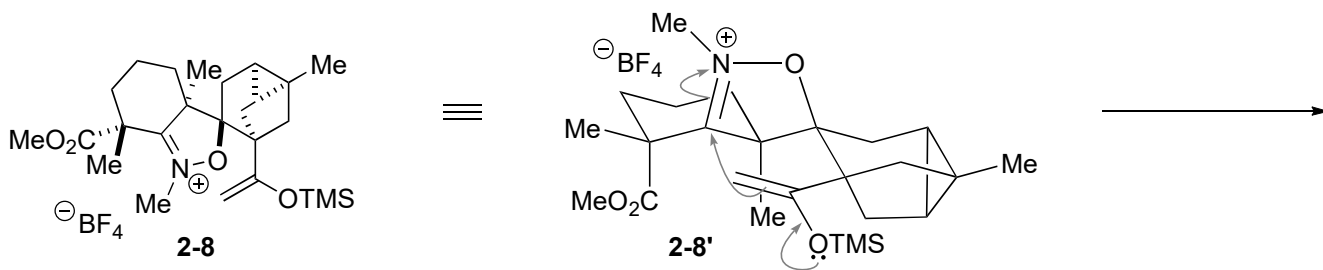
1. $\text{PhI}(\text{OAc})_2$ (1.5 eq), MeOH, 0 °C; extract with toluene^a; 120 °C, 52%
2. $\text{Me}_3\text{O}\cdot\text{BF}_4$ (2.5 eq), CH_2Cl_2 , rt; TMSOTf (5.0 eq), Et_3N (8.0 eq), rt, 69%
3. H_2 (1 atm), Pd/C (140+280 wt%), EtOAc/AcOH (5/1), 80 °C, 72%
4. KO^t-Bu (20 eq), O_2 (bubbling) THF, -78 °C; PPh_3 (5.0 eq) -78 °C to rt, 72% (dr = 14:1^b)
5. Dess-Martin periodinane (3.0 eq) *t*-BuOH (3.0 eq), CH_2Cl_2 , rt; SiO_2 (725 wt%) hexane/EtOAc (3/1), rt, 74%
6. $\text{PhI}(\text{OH})\text{OTs}$ (10 eq) NaHCO_3 (10 eq), CH_2Cl_2 , rt, 72%

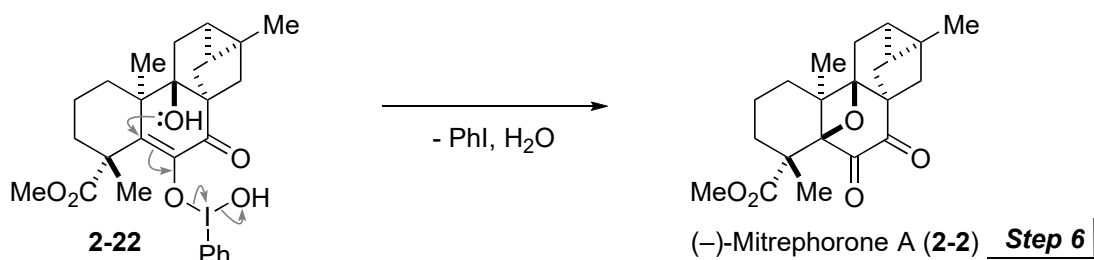
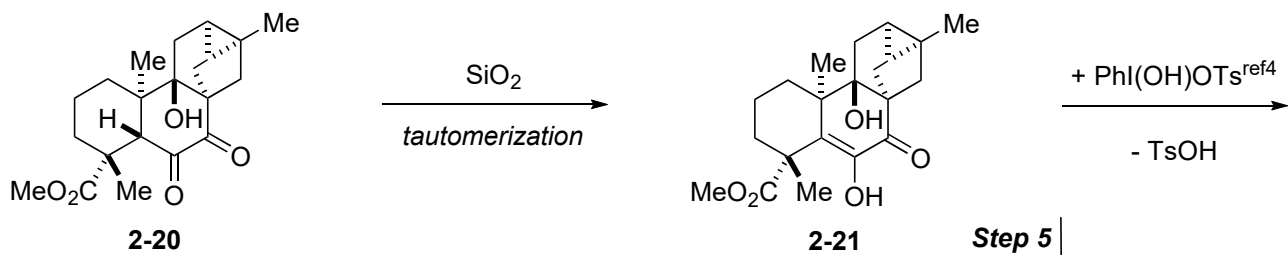
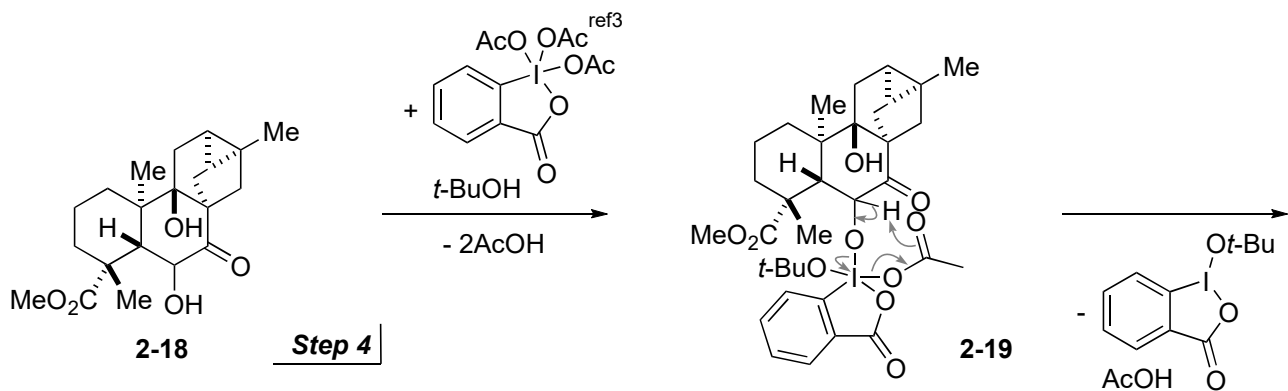


- a) After addition of saturated aq. $\text{Na}_2\text{S}_2\text{O}_3$ /saturated aq. $\text{NaHCO}_3/\text{H}_2\text{O}$ (1/1/2), the resultant mixture was extracted with toluene (x2). Combined organic layers were washed with brine/ H_2O (1/1) and dried over Na_2SO_4 . The filtrate was heated to 120 °C.
- b) The configuration of the newly formed stereogenic center was not assigned.

Schneider, M.; Richter, M. J. R.; Carreira, E. M. *J. Am. Chem. Soc.* **2020**, *142*, 17802.



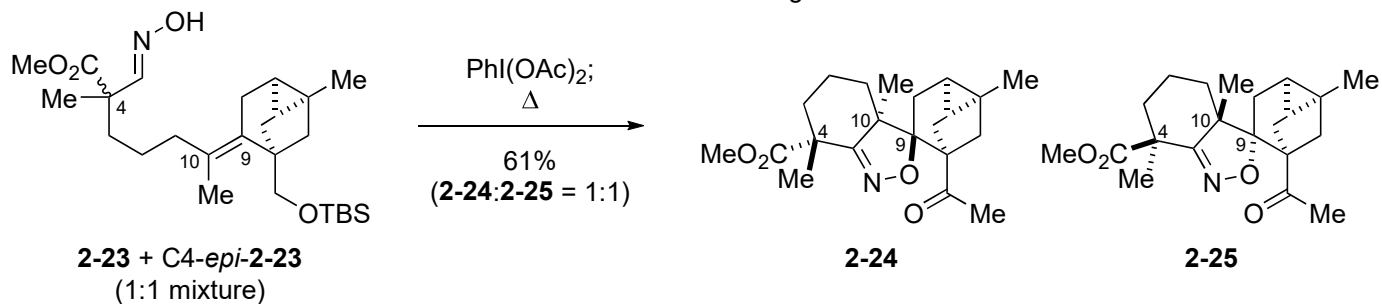
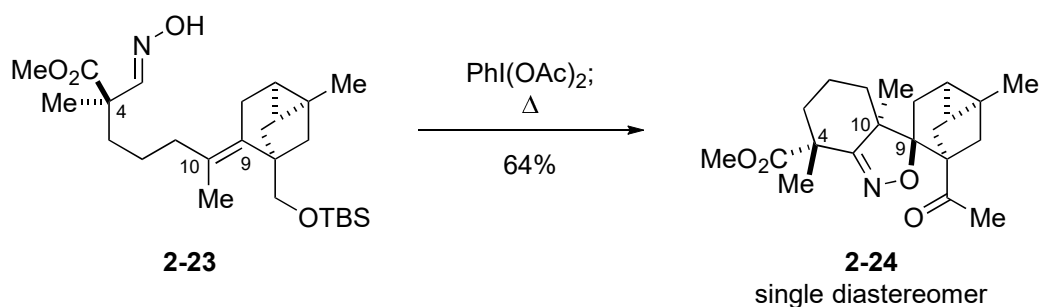




2-2. Discussion

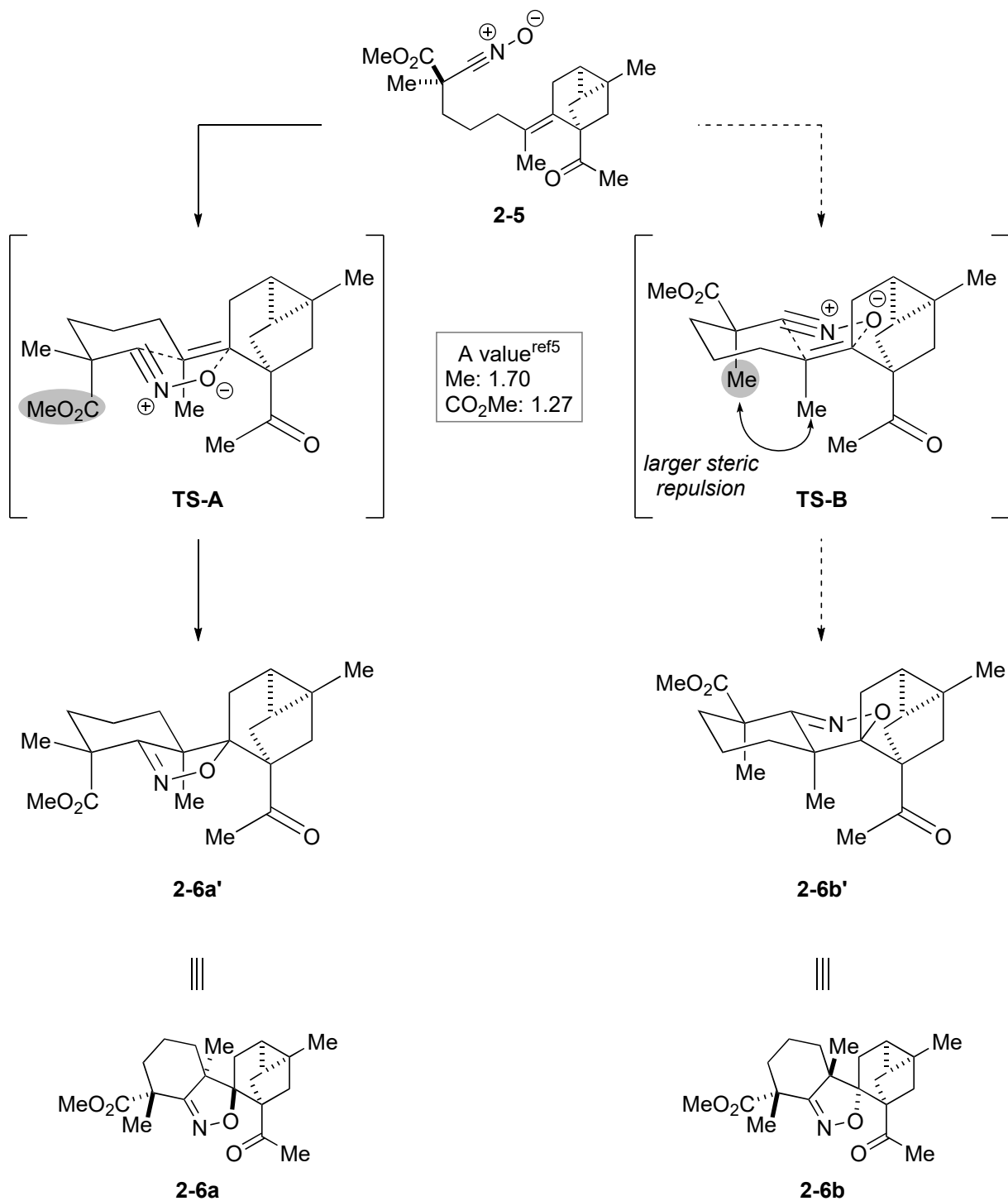
2-2-1. 1,3-Dipolar cycloaddition

2-2-1-1. Experimental results



Two diastereomers **2-24** and **2-25** have the same relative configuration at C4, C9, and C10 positions. This result shows that the facial selectivity in the 1,3-dipolar cycloaddition is fully controlled by the α stereocenter of the nitrile oxide (C4 position).

2-2-1-2. Rationale for the stereoselectivity of 1,3-dipolar cycloaddition



Reference

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- Dess, D. B.; Martin, J. C. *J. Org. Chem.* **1983**, *48*, 4155.
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- Hirsch, J. A. *Topics in Stereochemistry* **1967**, *3*, 199.