

Problem Session (2) -Answer-

2020.10.31. Yuto Hikone

Topic: Macrocycles

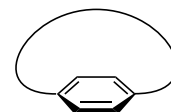
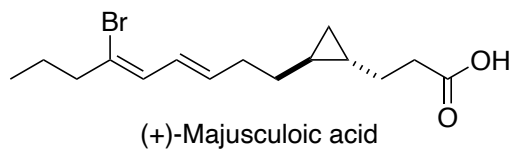
Introduction

1. Stereocontrol using macrocycle

(+)-Majusculoic acid : enantiomer of naturally occurring antifungal cyclopropane fatty acid.

total synthesis : Zhang (2018).

-> construction of stereocenters of small linear molecule by novel macrocyclic dimerization strategy.



2. Flexible macrocycle from rigid steroid

cyclophane : macrocyclic hydrocarbon consisting of an aromatic unit and an aliphatic chain.

Some paracyclophane can be derived from steroid.

-> Stereochemistry of flexible paracyclophane can be controlled by the stereocenters originally installed on rigid steroid.

1.

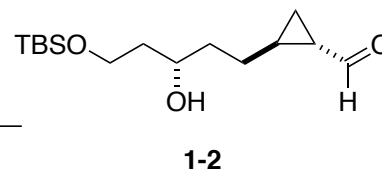
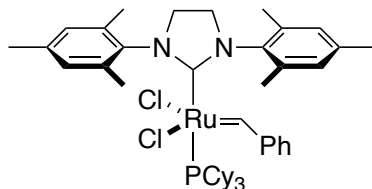
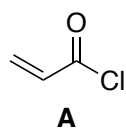
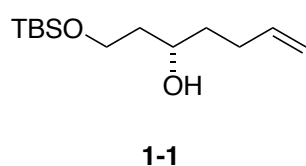
1. **A** (2 eq.), Et₃N (8 eq.), CH₂Cl₂, 0 °C, 72%

2. Grubbs 2nd (5 mol%)
CH₂Cl₂ (0.02 M), reflux, 53%

3. CH₂N₂ (5 eq.), Pd(OAc)₂ (10 mol%), Et₂O
-78 °C to rt, 80%

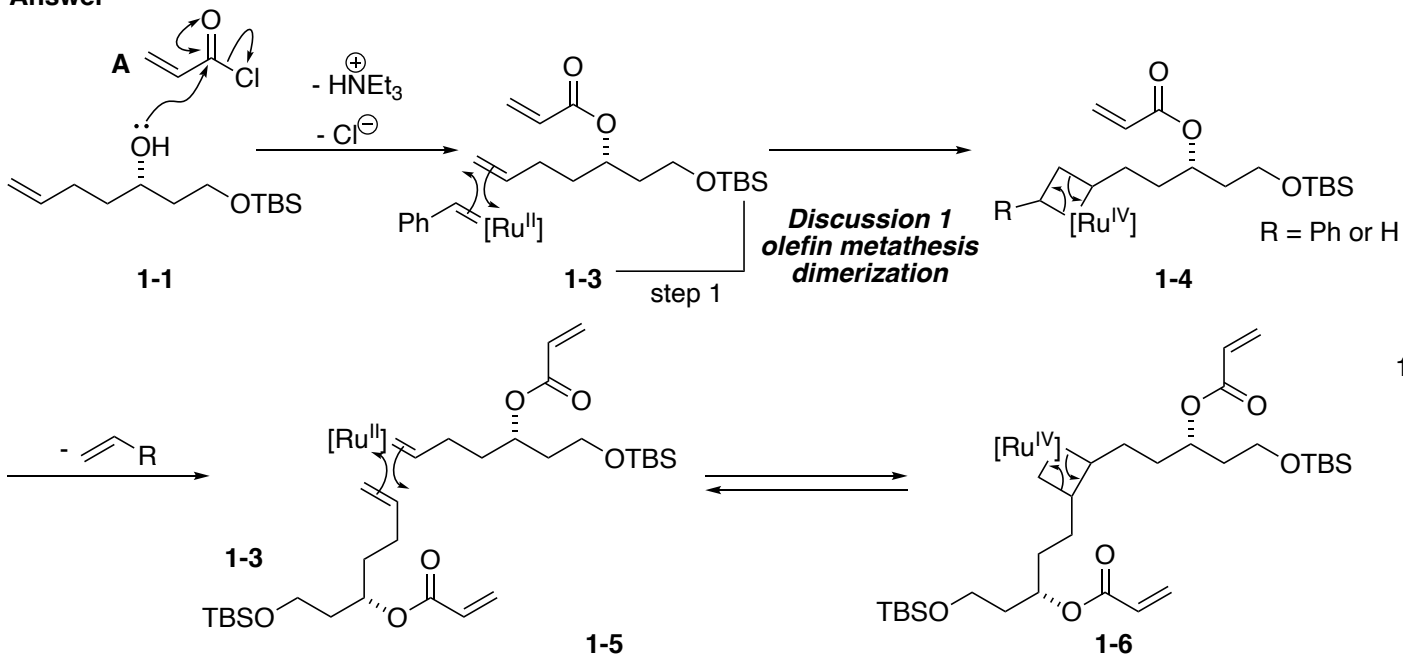
4. HN(OMe)Me·HCl (6 eq.), *i*-PrMgCl (12 eq.)
THF, 0 °C;

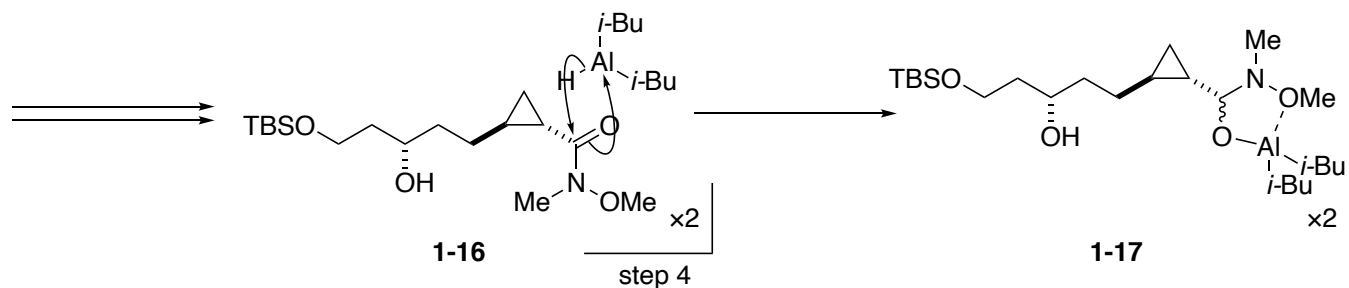
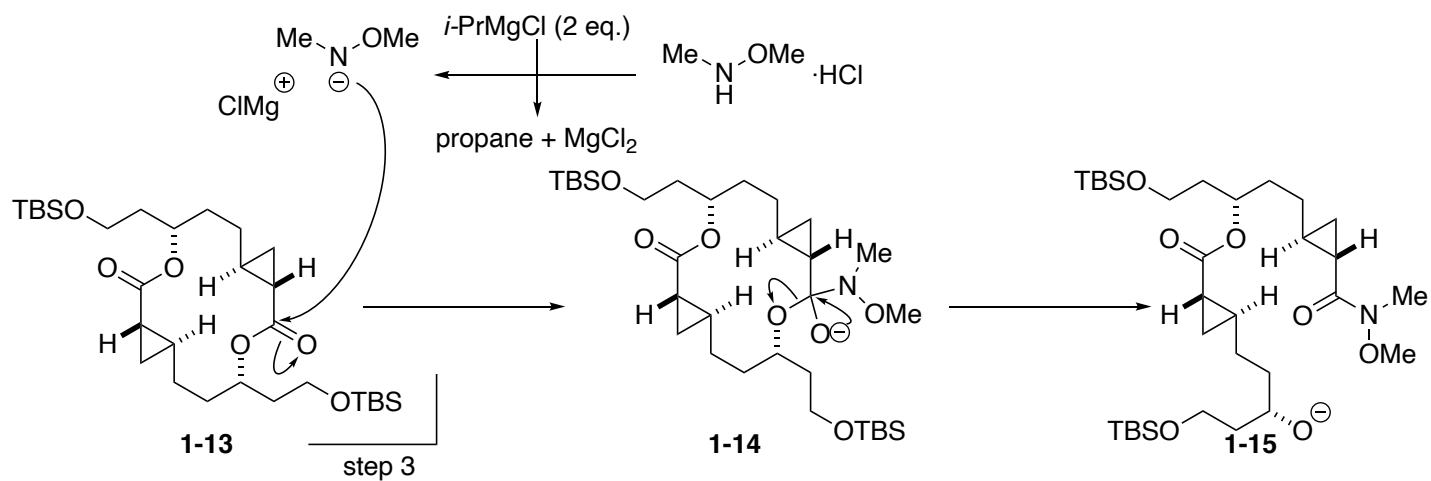
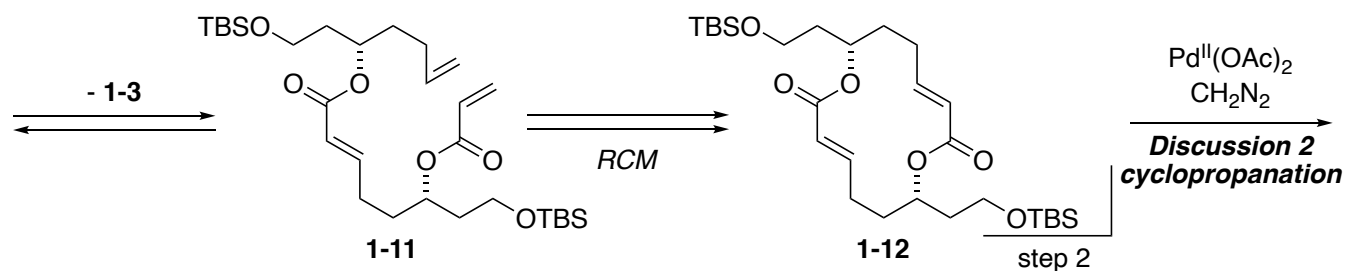
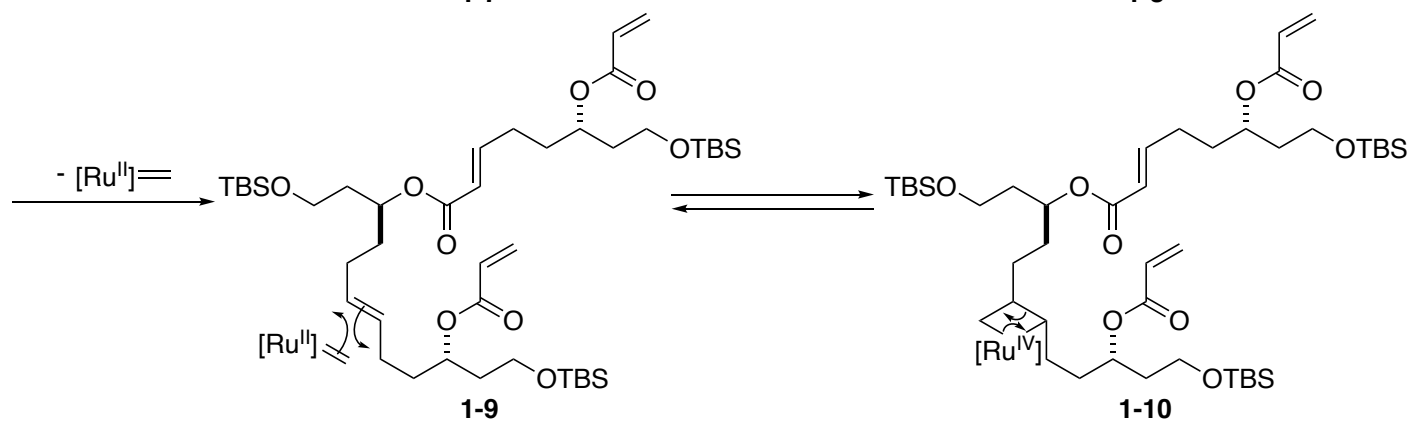
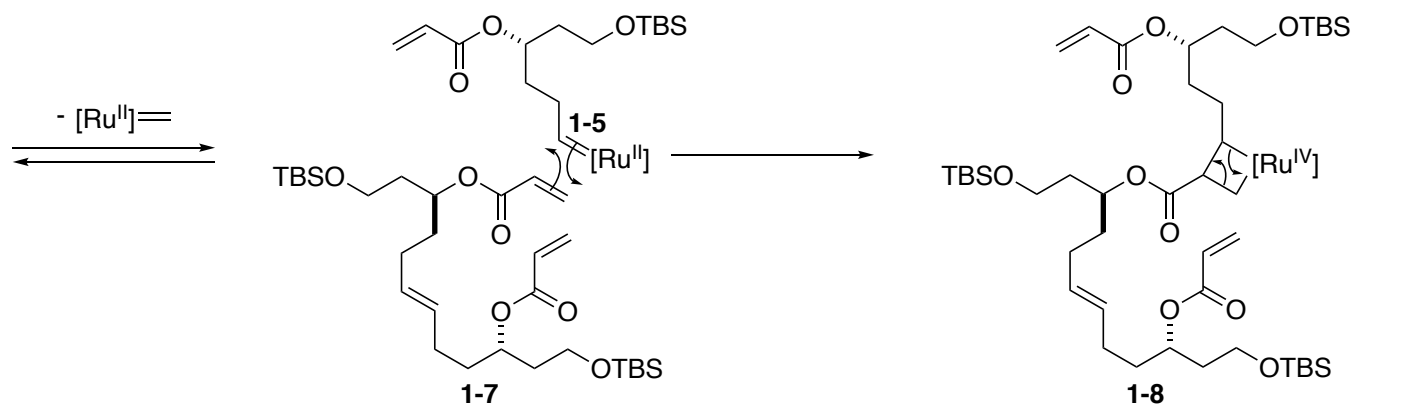
5. *i*-Bu₂AlH (2.5 eq.), Et₂O, 81% (2 steps)

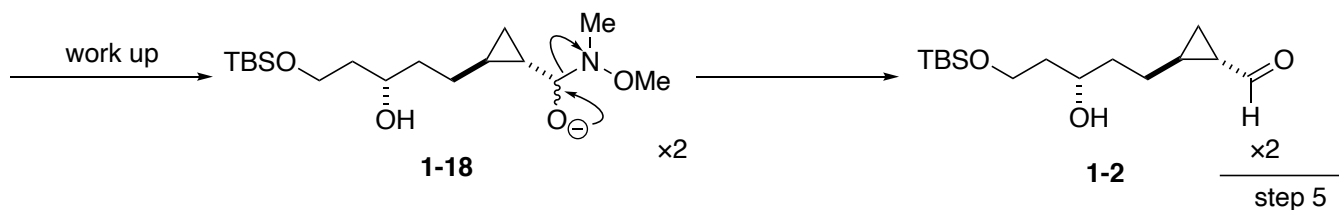


Chen, R.; Li, L.; Lin, N.; Zhou, R.; Hua, Y.; Deng, H.; Zhang, Y. *Org. Lett.* **2018**, *20*, 1477.

Answer

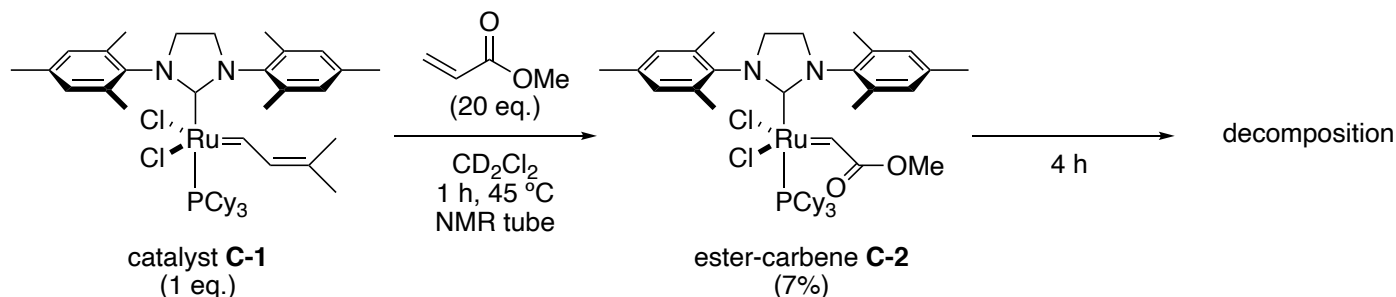






Discussion 1: olefin metathesis dimerization

1-1. Cross metathesis with α,β -unsaturated ester

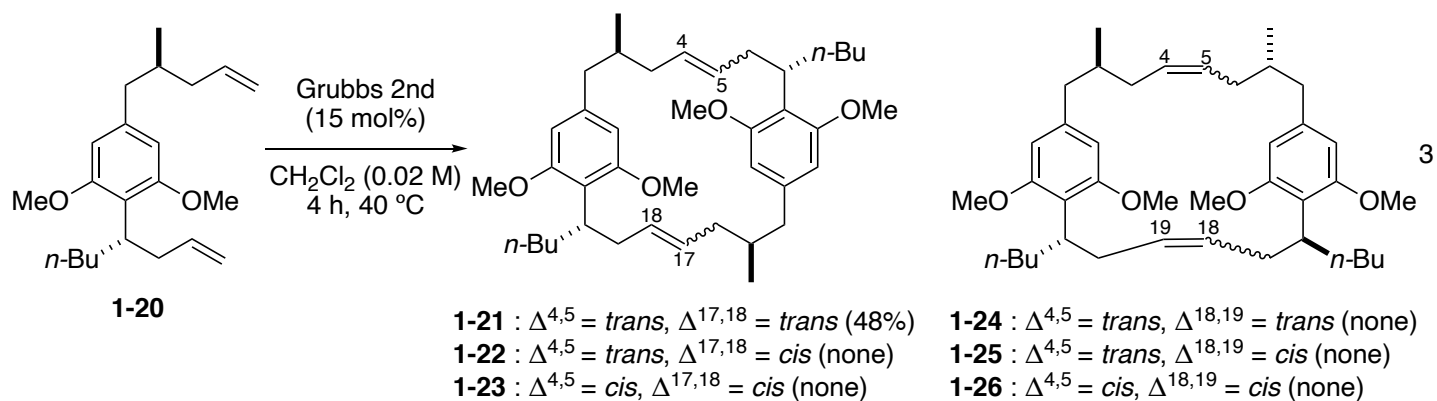
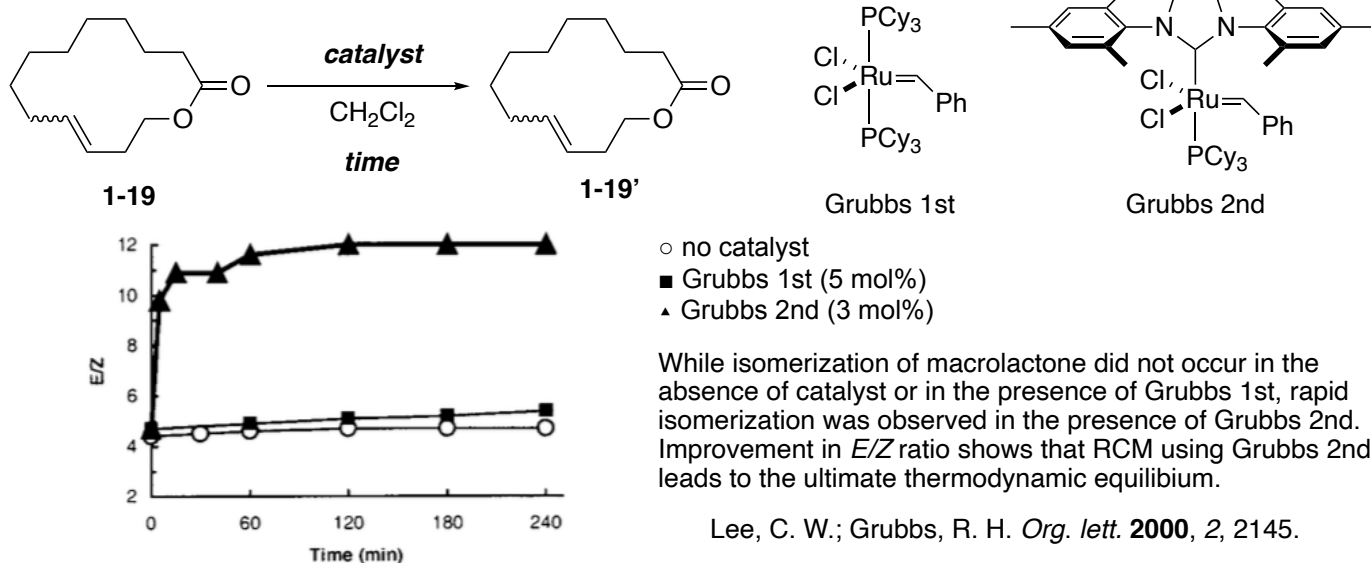


Chatterjee, A. K.; Morgan, J. P.; Scholl, M.; Grubbs, R. H. *J. Am. Chem. Soc.* **2000**, *122*, 3783.

This experiment suggests that the ester-carbene complex is unlikely to be an intermediate in cross metathesis between olefin and α,β -unsaturated ester.

Thus, in the cross metathesis between **1-3s**, Ru carbene complex **1-5** is likely to be the intermediate.

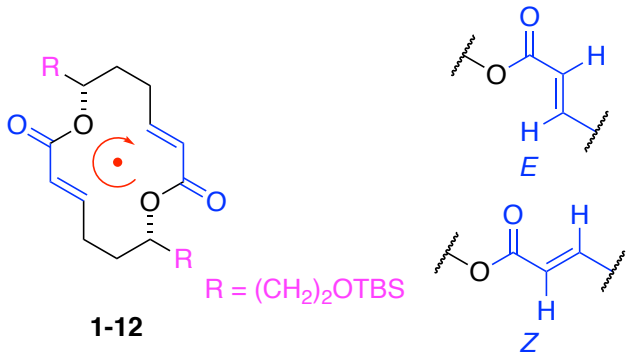
1-2. Dimerization



Smith, A. B.; Adams, C. M.; Kozmin, S. A. *J. Am. Chem. Soc.* **2001**, *123*, 990.

Discussion 2: cyclopropanation

2-1. Conformation of 14-membered macrolactone **1-12**



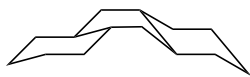
To consider appropriate conformations of 14-membered lactone **1-12**, make sure to grasp following points;

- **C₂ symmetry** (**confirmed from NMR spectra of 1-12**)
- **two geometrical isomers of enone (E or Z)**
- **branched alkyl groups (R)**

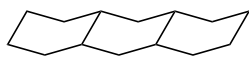


How to fulfill these points?

- **C₂ symmetry**
-> **chair-boat-chair (CBC)** or **boat-boat-boat (BBB)**
- **two geometrical isomers of enone (E or Z)**
-> **locate enones at appropriate positions**
- **branched alkyl groups (R)**
-> **equatorial position**



CBC-1



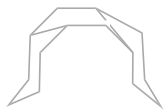
BBB-1



BBB-3

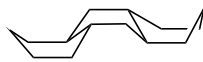


BBB-5



CBC-2

chair-boat-chair



BBB-2

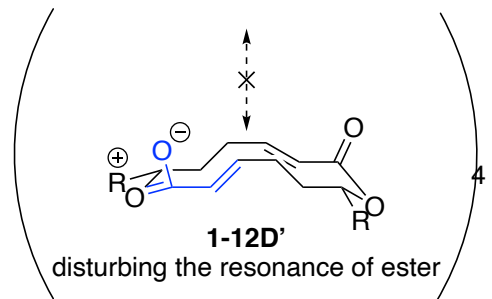
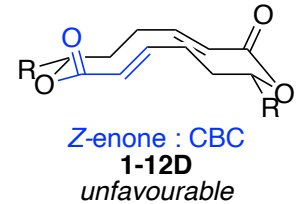
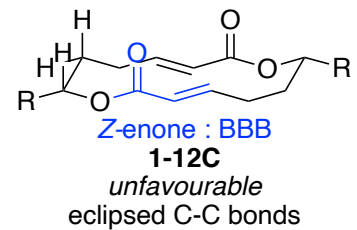
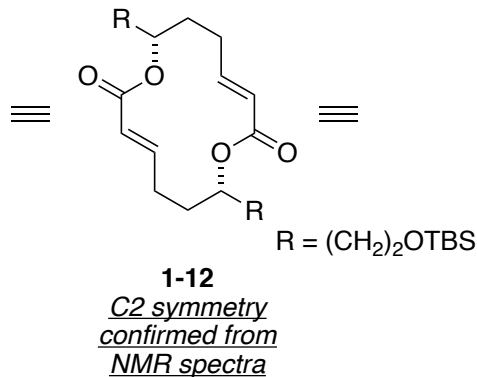
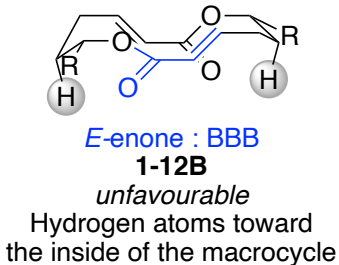
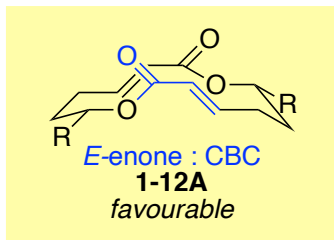
boat-boat-boat



BBB-4

Considering the steric repulsions, CBC1, BBB1 or BBB2 are the most plausible conformations.

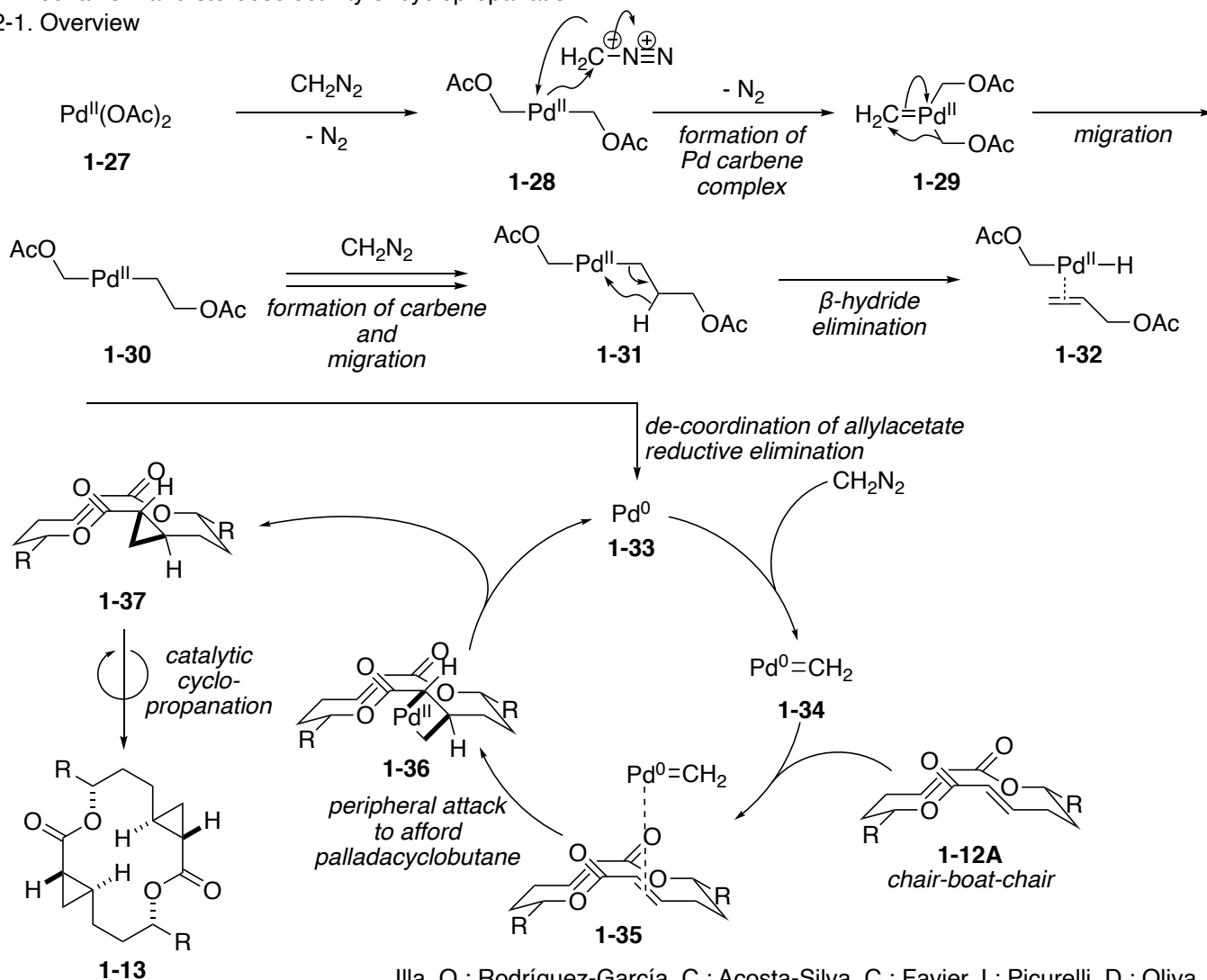
-> Taken these points into consideration, the conformations of 14-membered macrolactone **1-12** are depicted as follows.



Among these four conformations, **1-12A** is thermodynamically most stable.

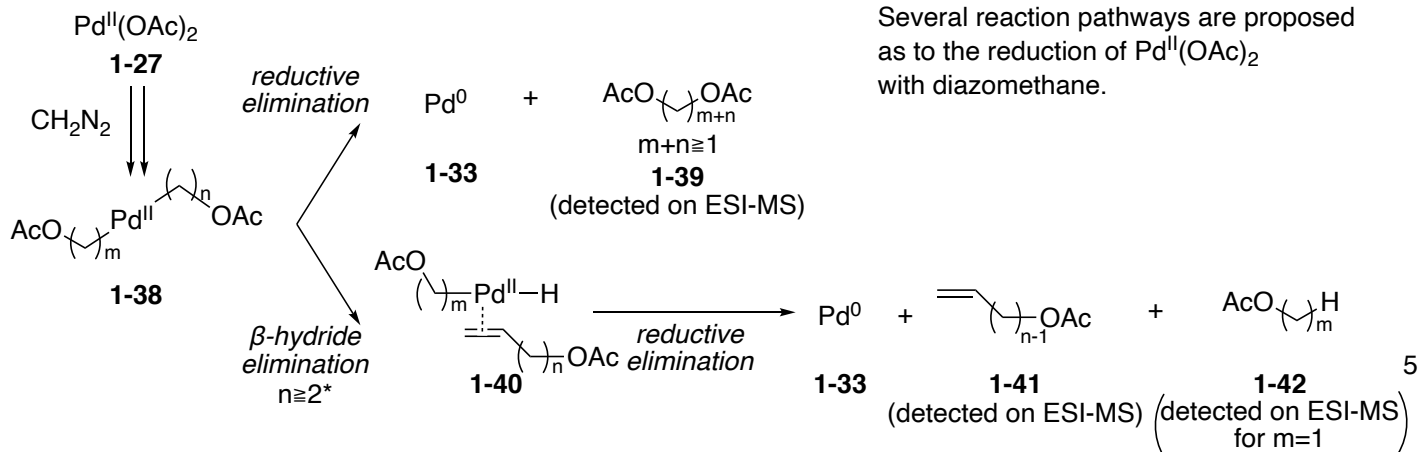
2-2. Mechanism and stereoselectivity of cyclopropanation

2-2-1. Overview

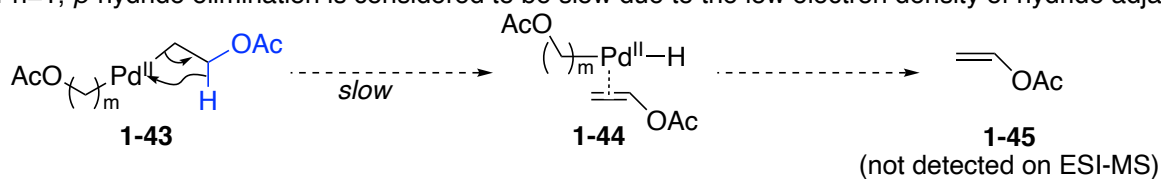


Illa, O.; Rodríguez-García, C.; Acosta-Silva, C.; Favier, I.; Picurelli, D.; Oliva, A.; Gómez, M.; Branchadell, V.; Ortuño, R. M. *Organometallics* **2007**, *26*, 3306.

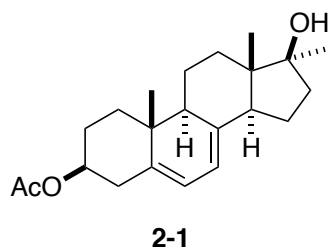
2-2-2. Details of the reaction mechanism



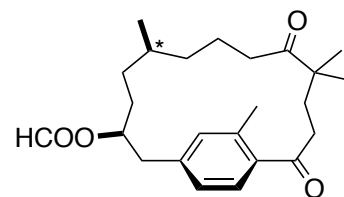
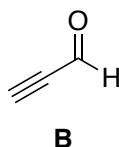
* For $n=1$, β -hydride elimination is considered to be slow due to the low electron density of hydride adjacent to OAc.



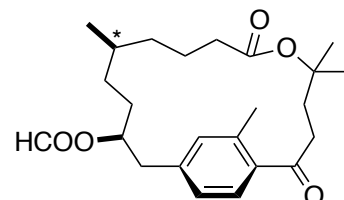
2.



1. **B** (2.5 eq.), $\text{BF}_3 \cdot \text{OEt}_2$ (1 eq.)
 CH_2Cl_2 , 0 °C to rt
2. xylene, reflux, 46% (2 steps)
3. NaBH_4 (2 eq.), $\text{CH}_2\text{Cl}_2/\text{EtOH}$, rt, 65%
4. H_2 , Pt/C, MeOH, rt, 85% (dr = calc. 8:1)
5. HCOOH , reflux, 80%
6. O_3 , CH_2Cl_2 , -78 °C; PPh_3 (1.1 eq.), rt



34% (dr = 8:1)

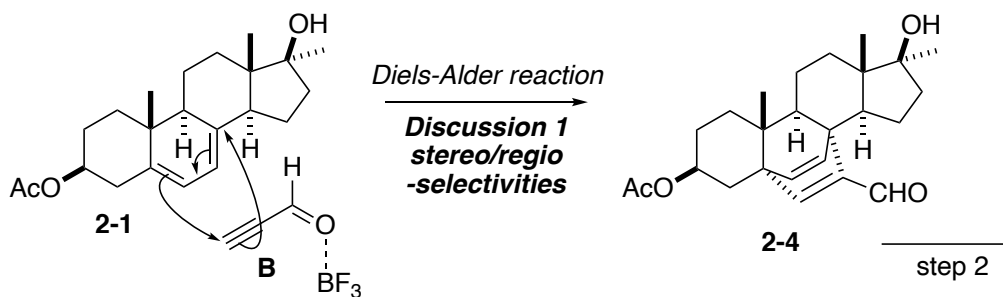


6%

Bäurle, S.; Blume, T.; Mengel, A.; Parchmann, C.; Skuballa, W.; Bäsler, S.; Schäfer, M.; Sülzle, D.; Wrona-Metzinger, H. *Angew. Chem. Int. Ed.* **2003**, *42*, 3961.

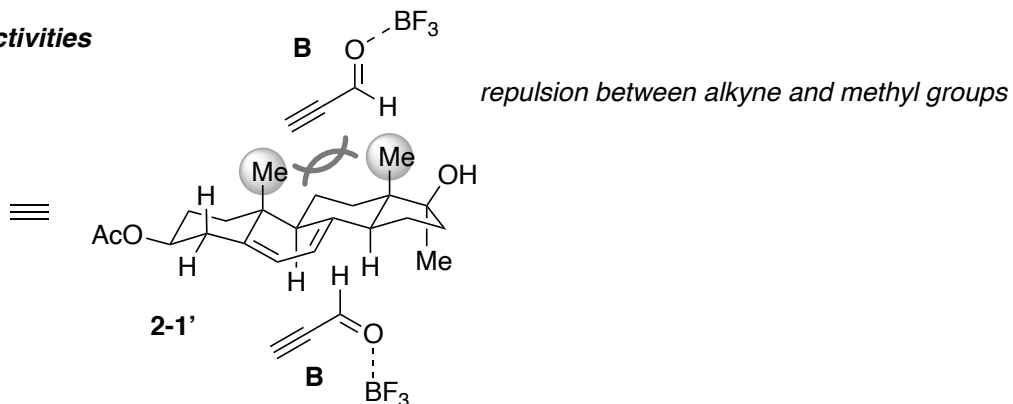
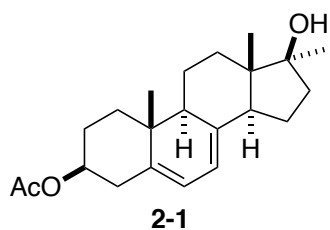
Answer

1st step

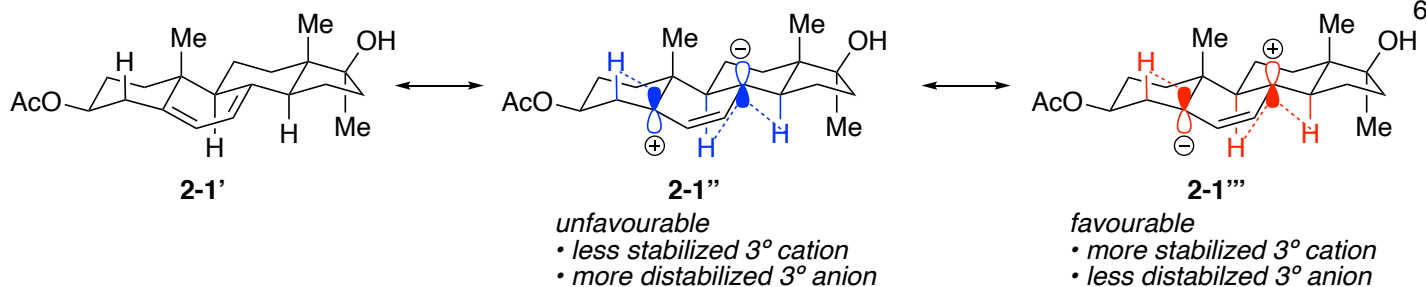


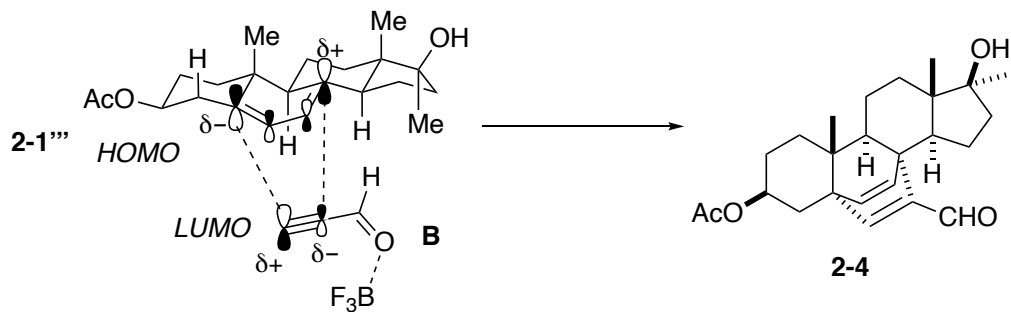
Discussion 1: stereo/regioselectivities

1-1. Stereoselectivity

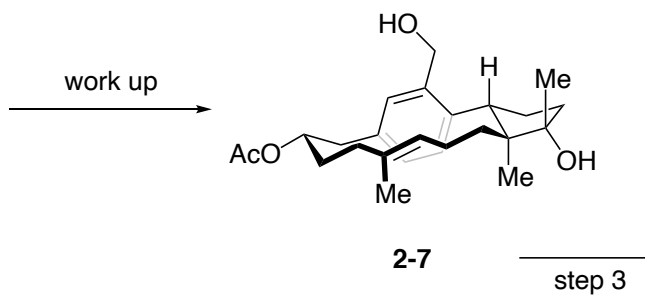
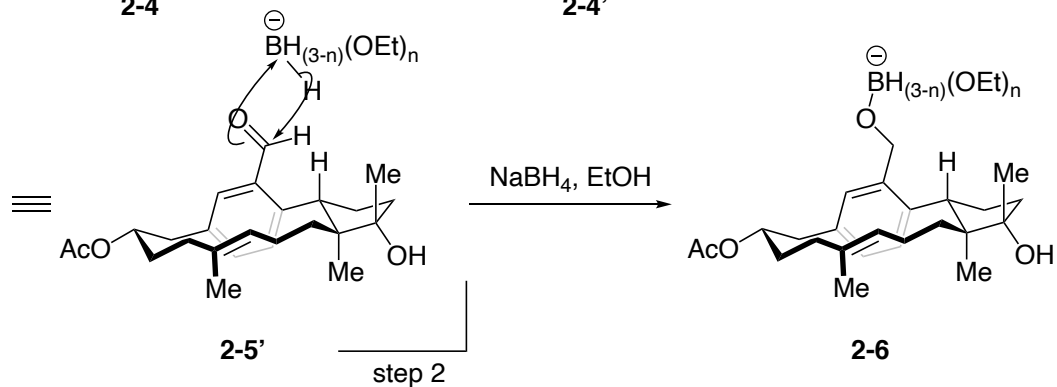
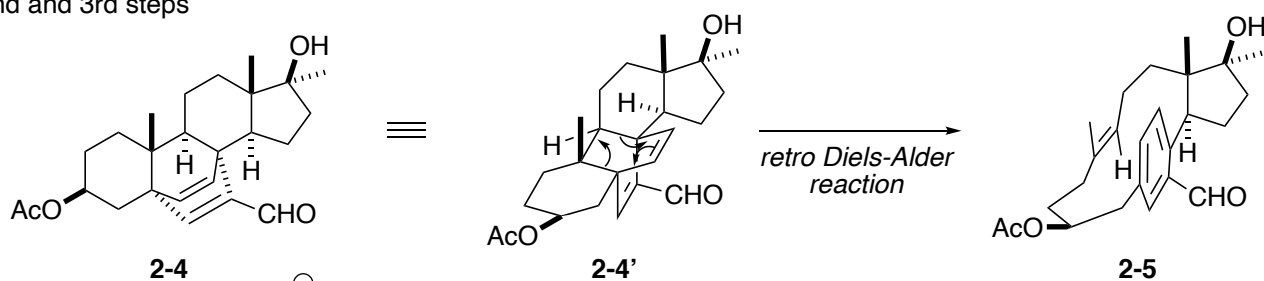


1-2. Regioselectivity

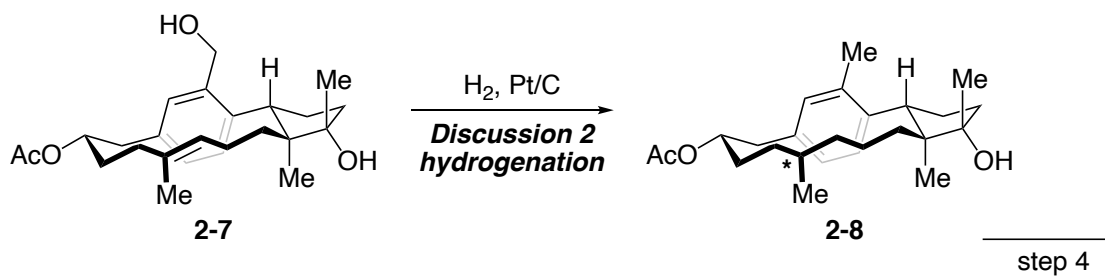




2nd and 3rd steps

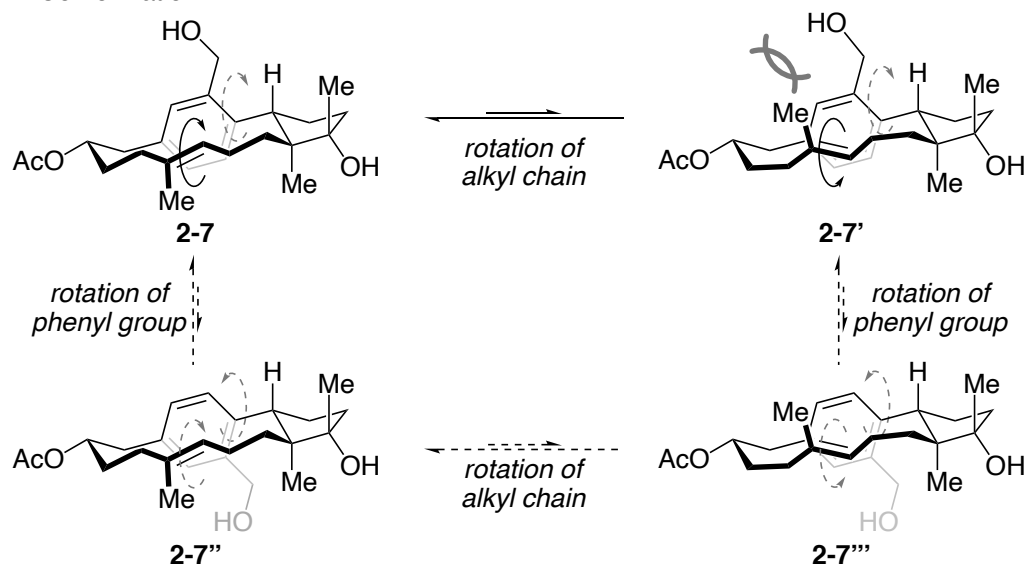


4th step



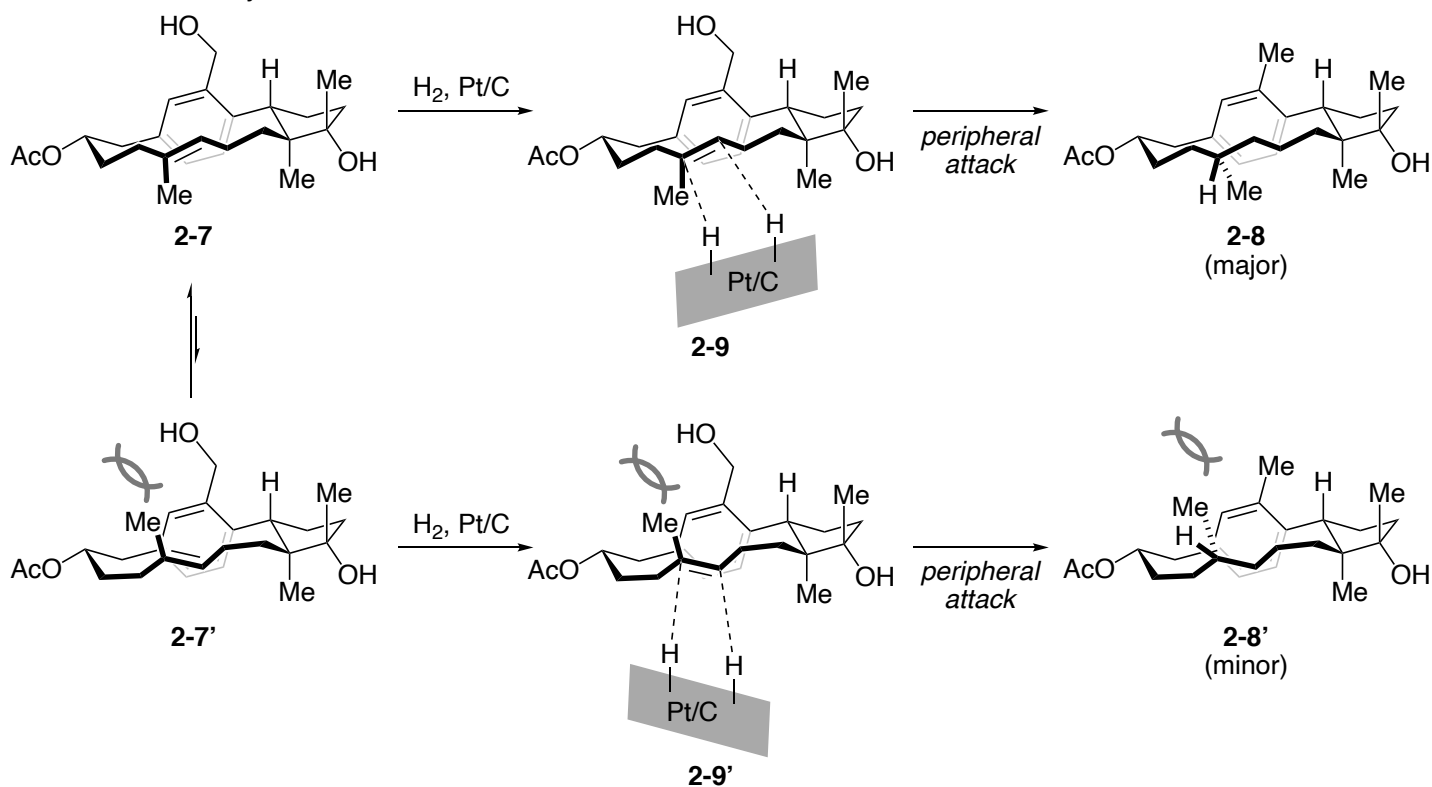
Discussion 2: hydrogenation

2-1. Conformation

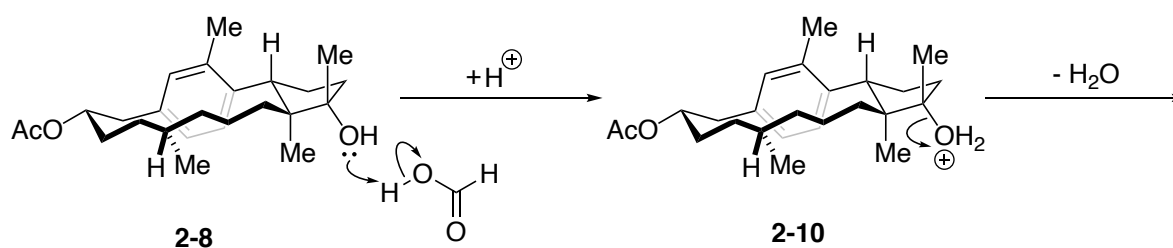


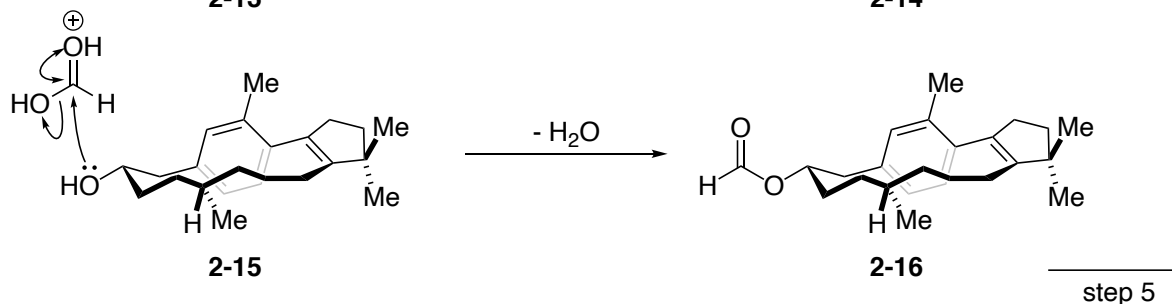
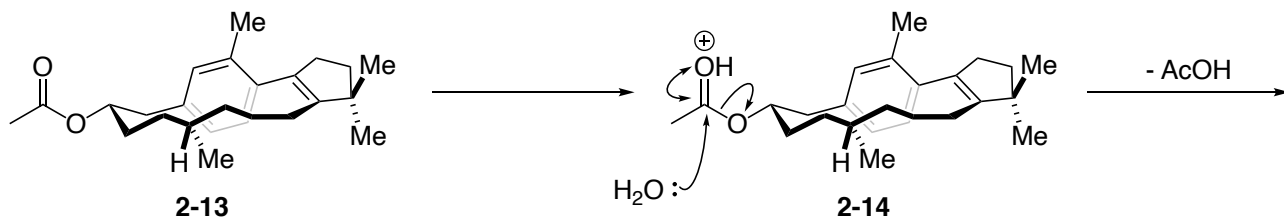
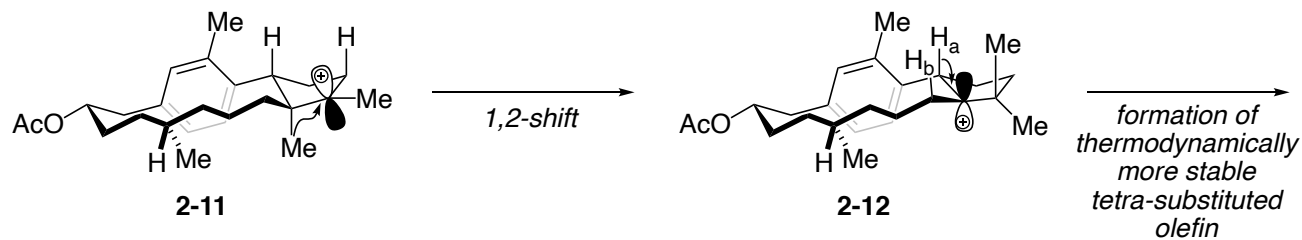
Due to the rigidity and bulkiness of phenyl group, bond rotation is permitted only as to alkyl chain.

2-2. Stereoselectivity

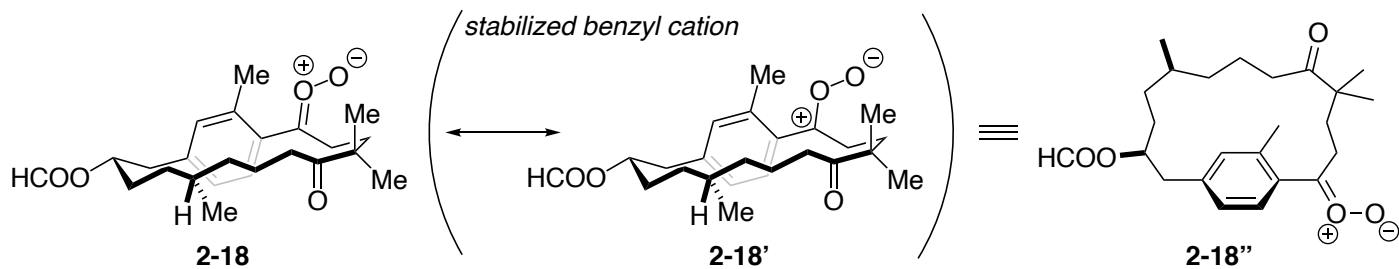
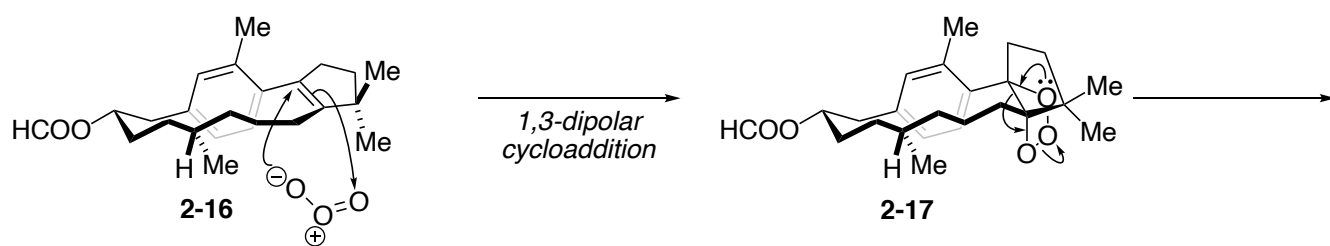


5th step

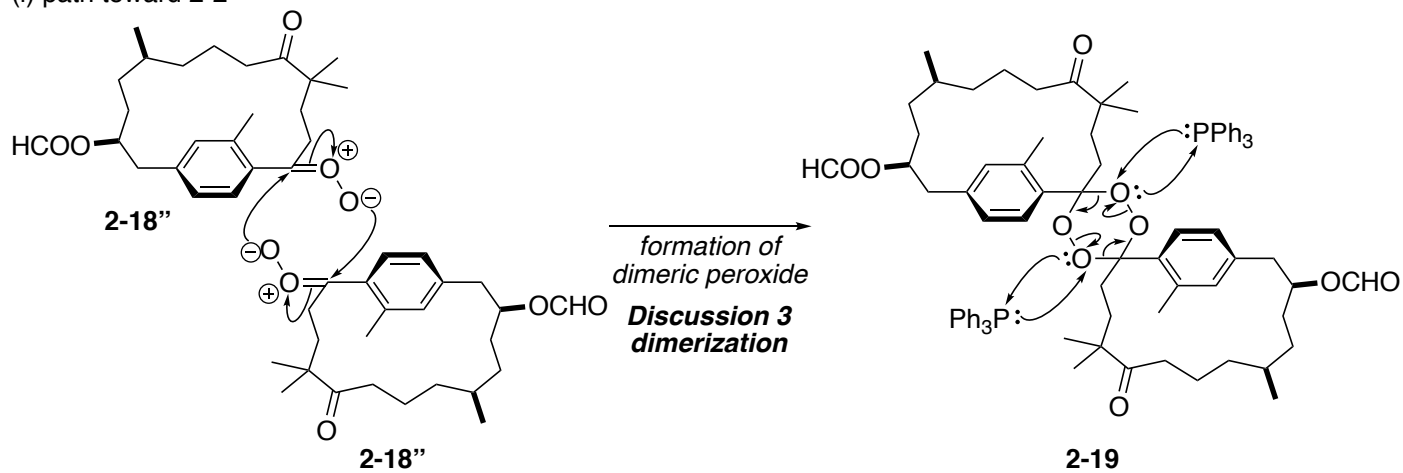


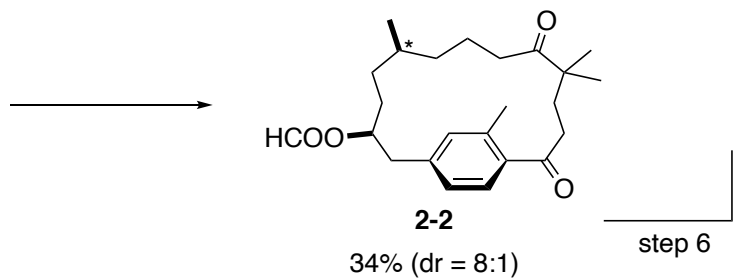


6th step

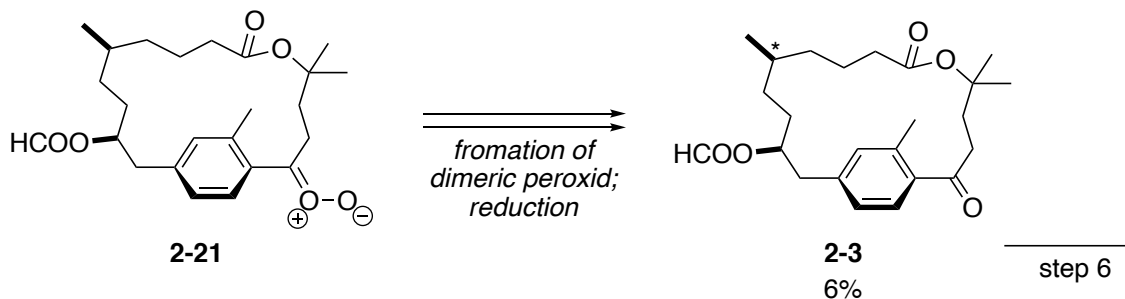
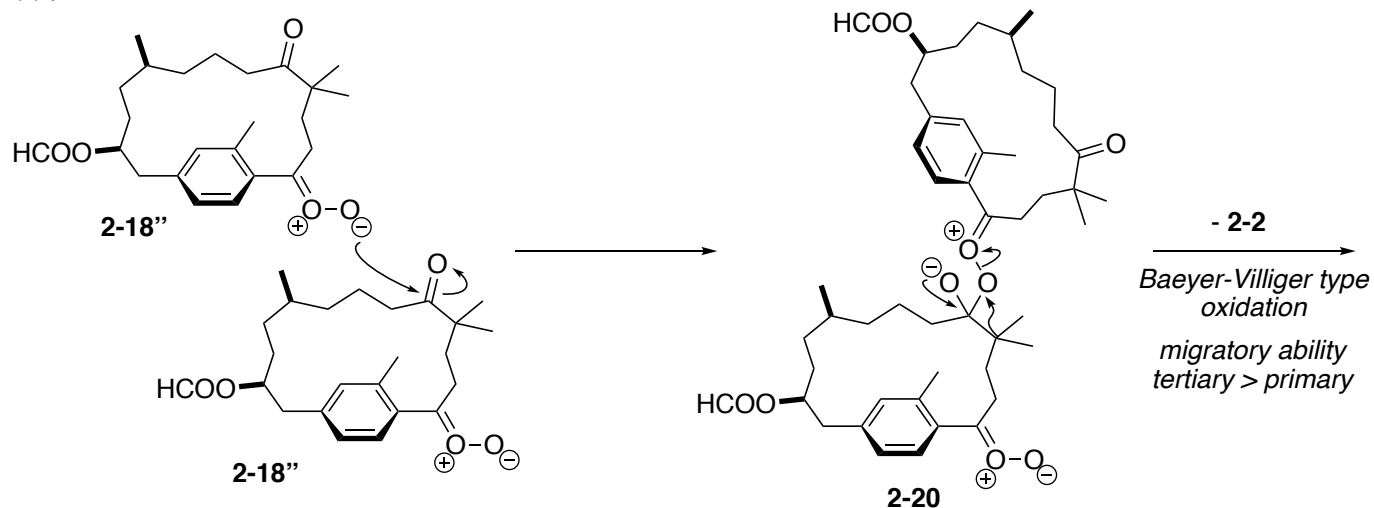


(i) path toward 2-2



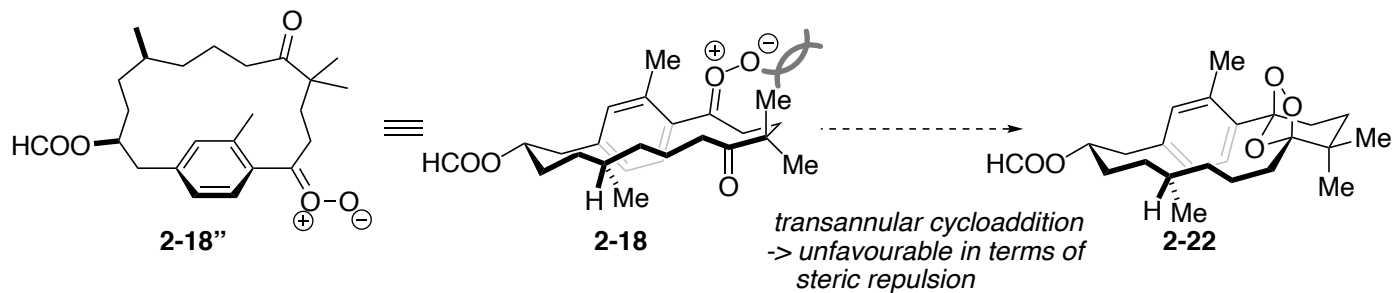


(ii) path toward **2-3**

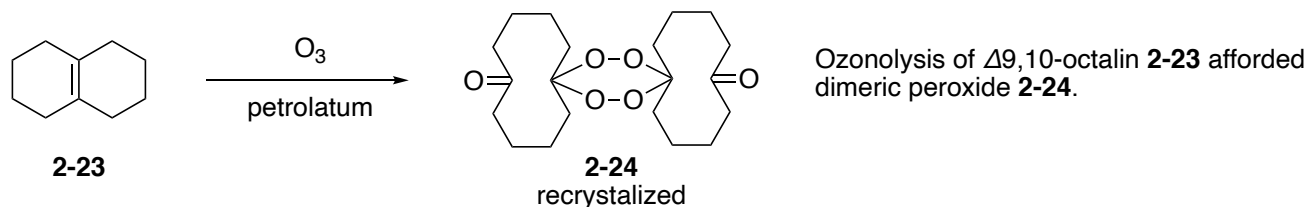


Discussion 3: dimerization

3-1. Formation of ozonide



3-2. Experimental evidence of dimeric peroxide



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