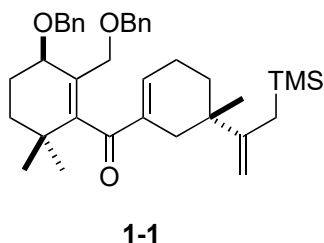


Problem Session (5)

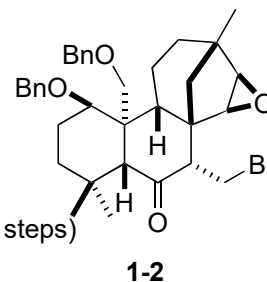
2020/06/13 Shimizu Shinsuke

Please provide following reaction mechanisms and stereoselectivities

Problem 1

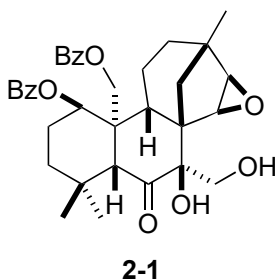


1. EtAlCl_2 (1.5 eq), $\text{CH}_2\text{Cl}_2/\text{THF}$ (20/1), 40 °C
2. O_2 , TPP (0.0056 eq), Fluorescent lamps, CDCl_3 , rt
; Ac_2O (0.9 eq), pyridine (0.3 eq)
DMAP (0.024 eq), rt, 43% (2 steps)
3. $\text{RhCl}(\text{PPh}_3)_3$ (1.0 eq), toluene, reflux, 67%
4. vinyl lithium, Et_2O , -78 °C to 0 °C
5. *m*-CPBA (2 eq), NaHCO_3 (10 eq), CH_2Cl_2 , 0 °C, 70% (2 steps)
6. NBS (1.2 eq), silica gel (200 wt%), CH_2Cl_2 , rt, 89%

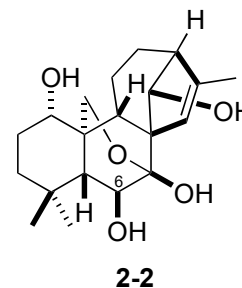


Vinyl lithium was prepared by mixing vinyl bromide (4 eq) and *t*-butyl lithium (7.5 eq).

Problem 2

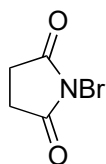


1. EtAlCl_2 (3 eq), toluene, 0 °C to rt, 57%
2. LiAlH_4 (5 eq), Et_2O , 0 °C to rt
3. NaIO_4 (1.38 eq), $\text{THF}/\text{pH 7 buffer}$ (2/1)
10 °C to rt, 69% (2 steps)
4. $(\text{MeO})_2\text{CMe}_2$ (20 eq), PTSA (0.1 eq)
acetone, rt; NaHCO_3 (20 eq); evaporation
; DMP (3.5 eq), CH_2Cl_2 , 0 °C; DMP (2 eq), rt, 66%
5. *i* Bu_2AlH (2 eq), $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$ (3/1), -100 °C to rt
; Red-Al (3 eq), THF, rt; aq.HCl, MeOH, rt, 56%
(diastereomer at C6: 21%)

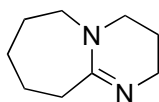


pH 7 buffer: 0.2 M aqueous solution of a 1:1 mixture of $\text{Na}_2\text{HPO}_4:\text{NaH}_2\text{PO}_4$

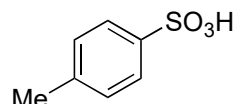
In these reactions, enantio excess (ee) were not mentioned.



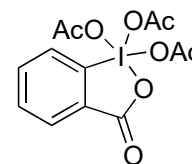
NBS



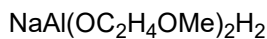
DBU



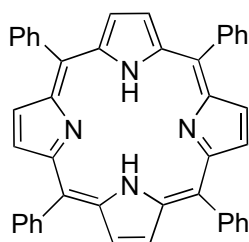
PTSA



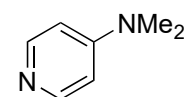
DMP



Red-Al



TPP



DMAP

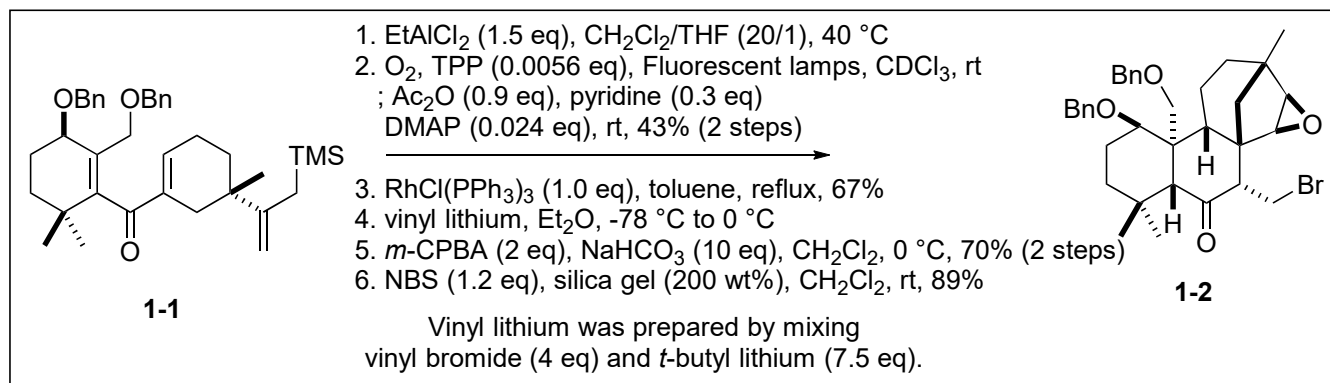
Problem Session (5) Answer

2020/06/13 Shimizu Shinsuke

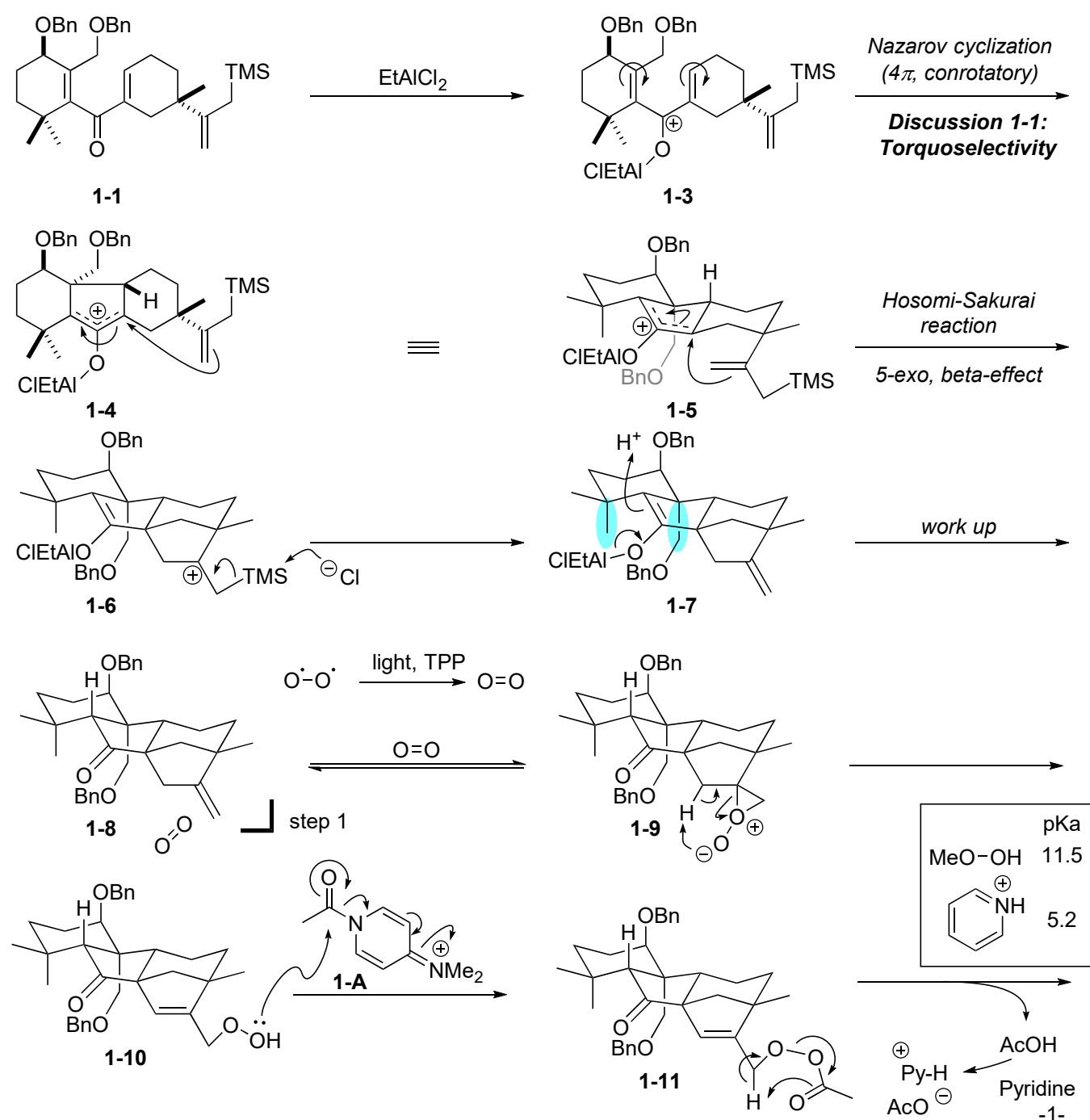
topic: Total synthesis of oridonin by Luo group

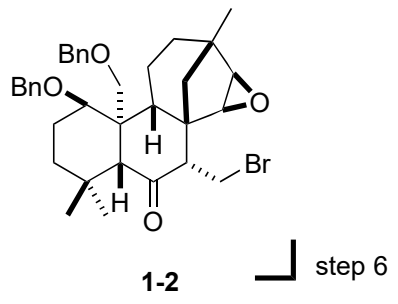
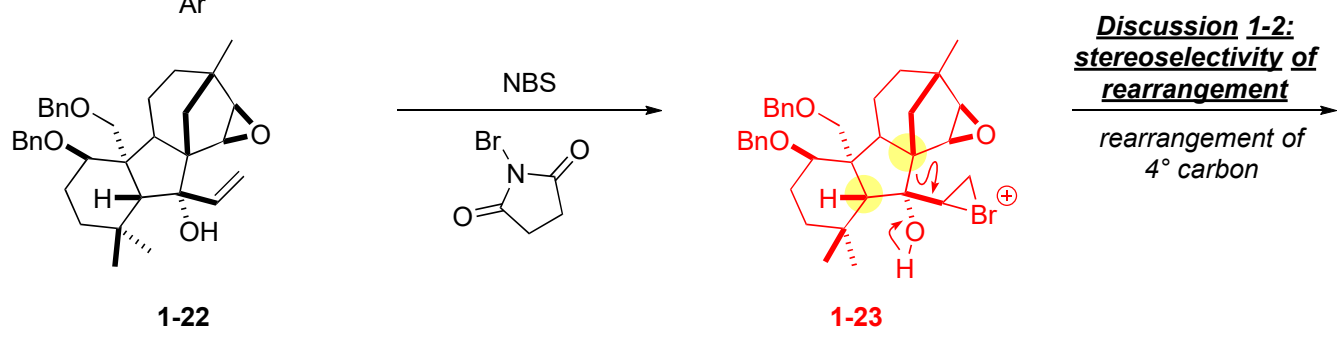
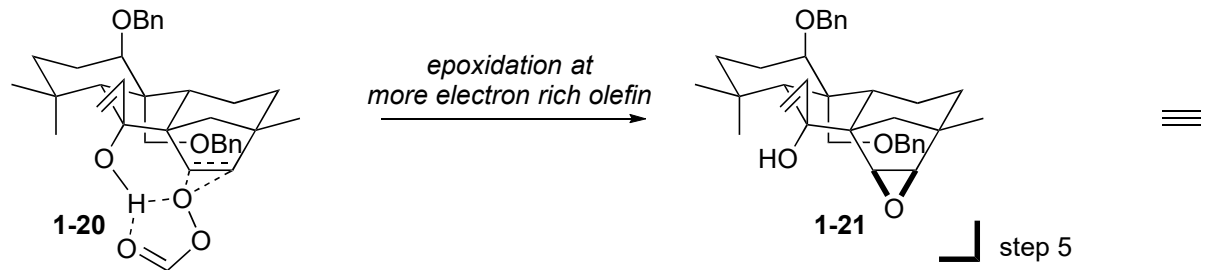
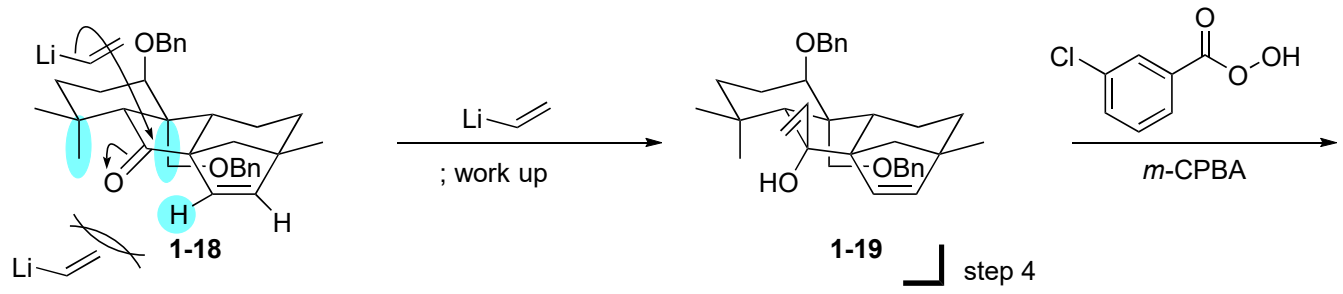
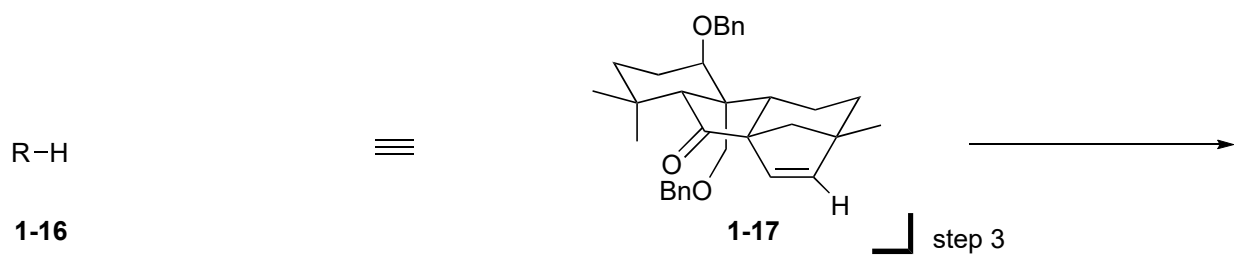
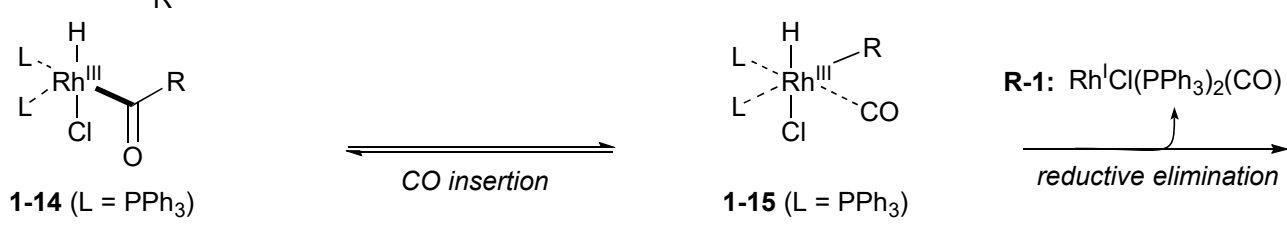
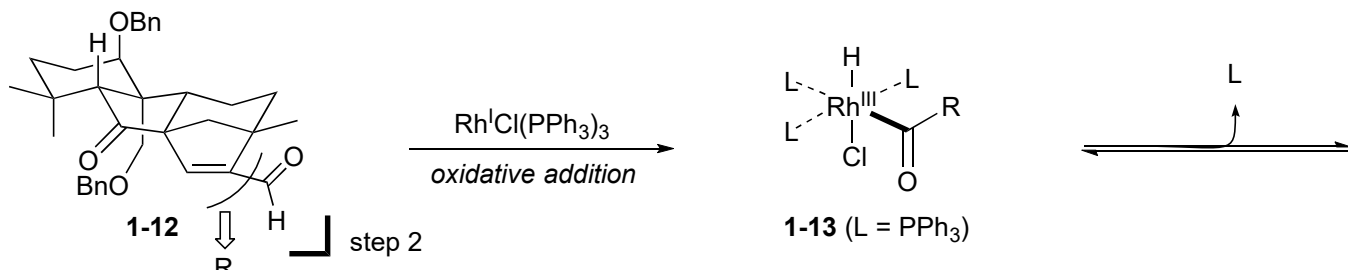
(Kong, L.; Su, F.; Yu, H.; Jiang, Z.; Lu, Y.; Luo, T. *J. Am. Chem. Soc.* **2019**, *141*, 20048.)

Problem 1



Answer 1



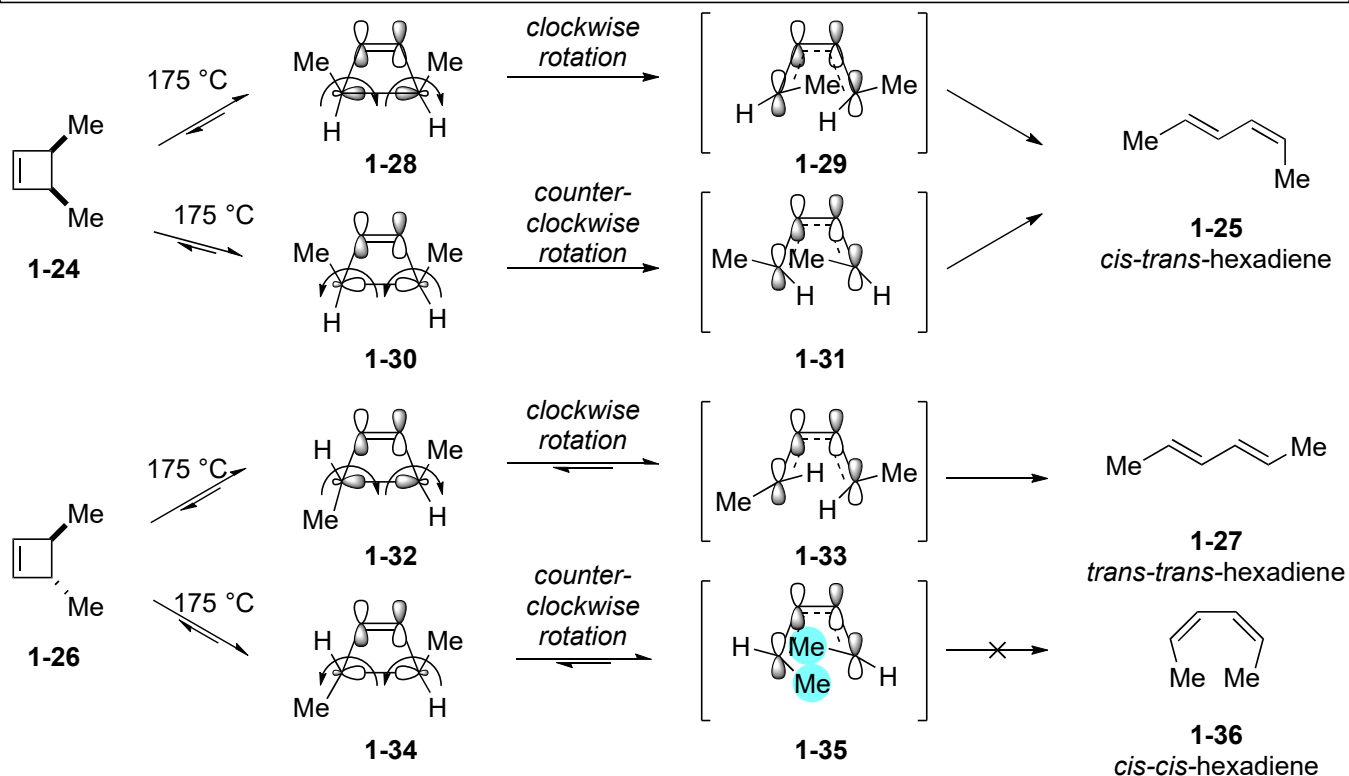
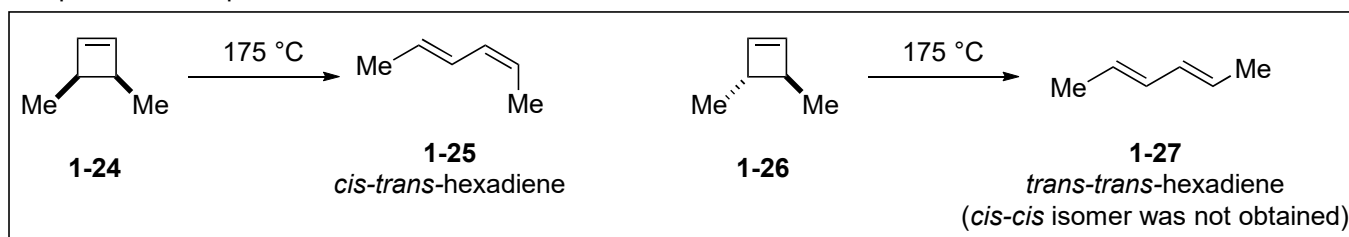


Discussion 1-2:
stereoselectivity of rearrangement
 rearrangement of 4° carbon

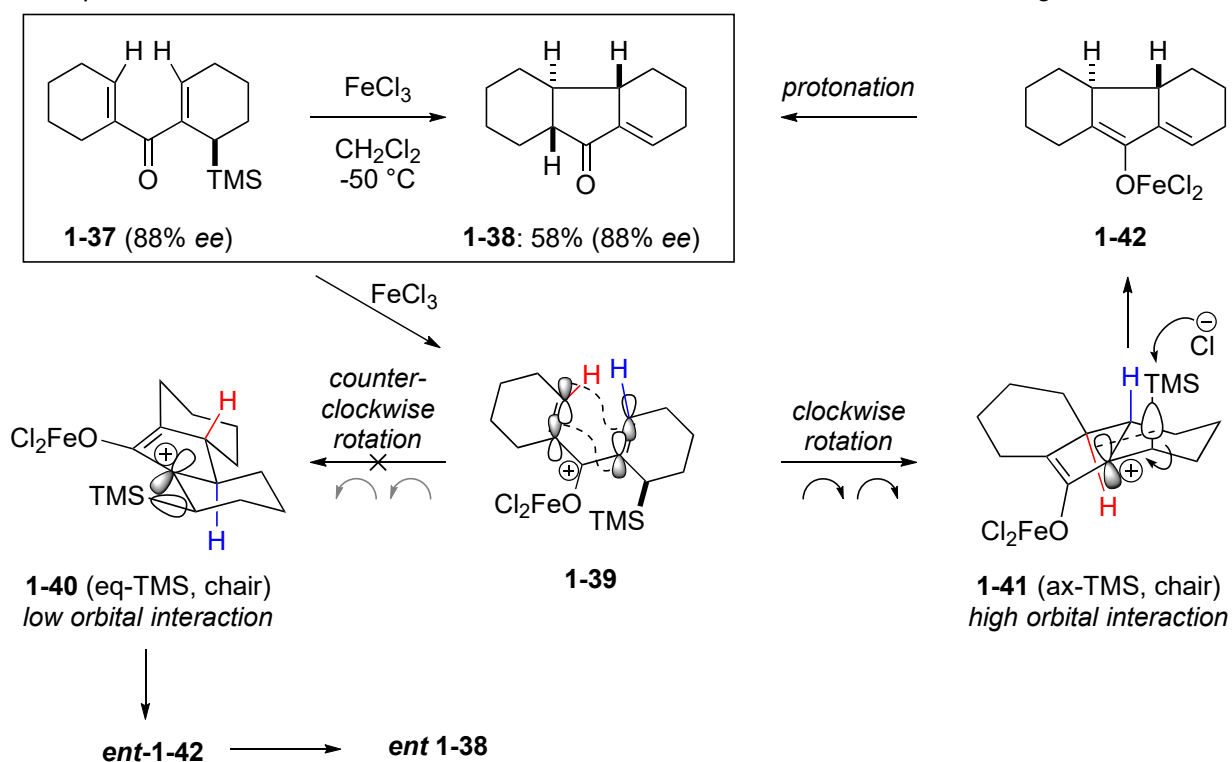
Discussion 1-1: Torquoselectivity of Nazarov cyclization (clockwise or counterclockwise?)

Torquoselectivity: steric repulsion and stereoelectronic effect are the driving force for the selectivity.

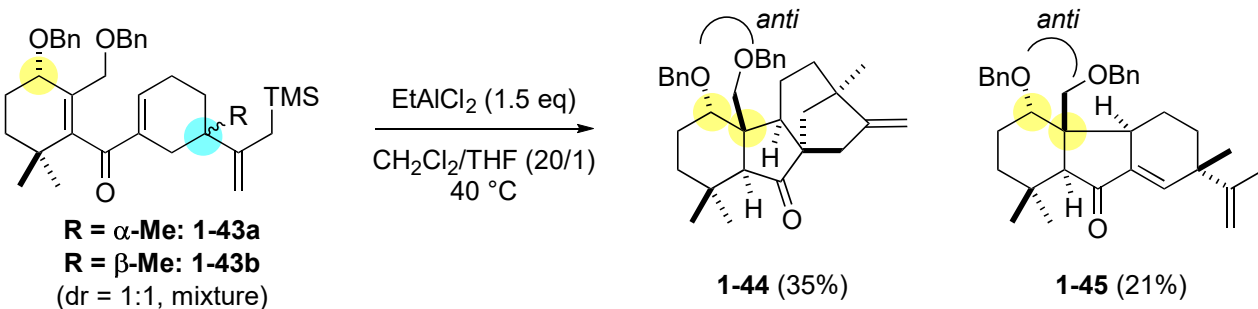
example of steric repulsion: Winter, R. E. K. *Tetrahedron Lett.* **1965**, 17, 1207.



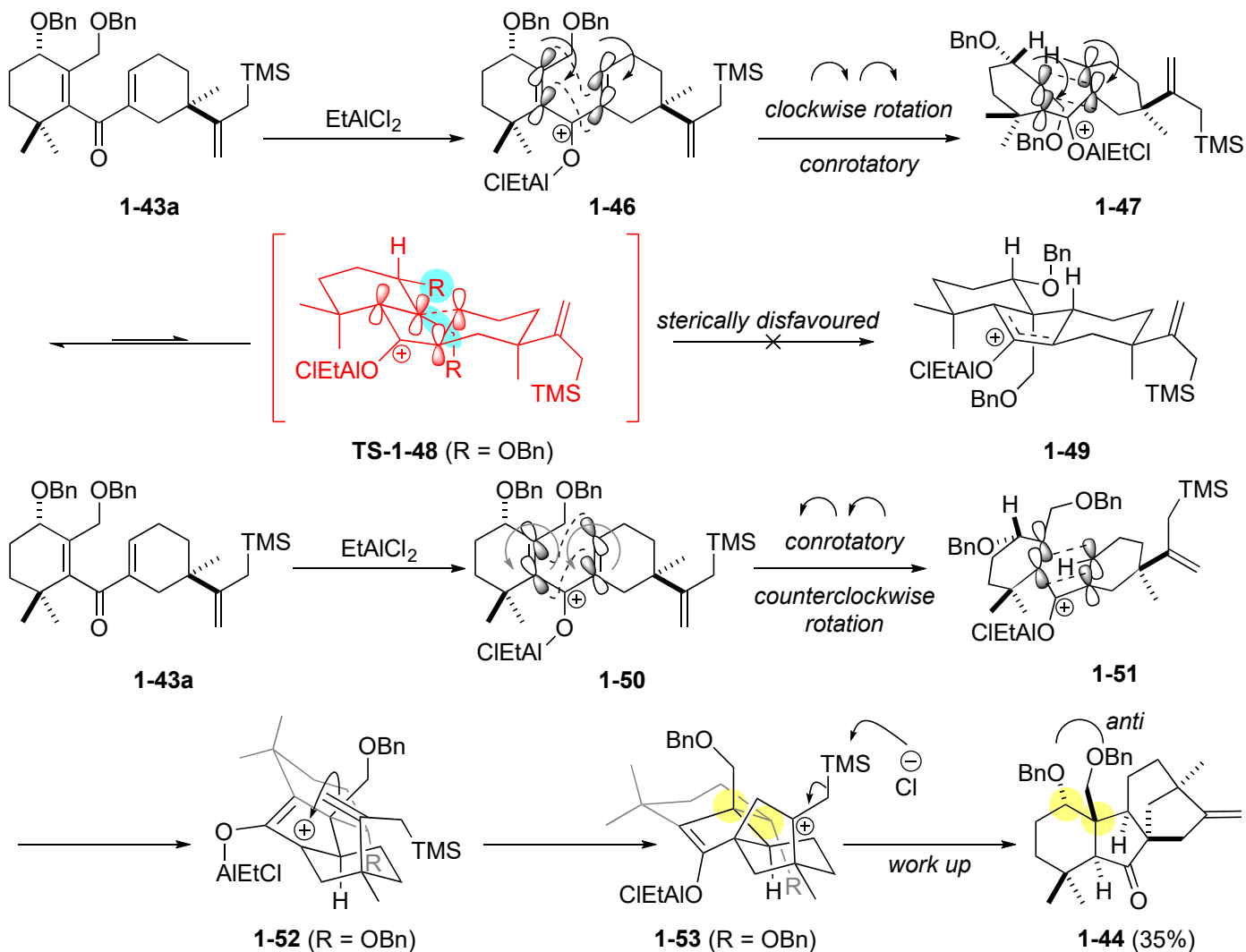
example of stereoelectronic effect: Denmark, S. E.; Wallace, M. A.; Walker, C. B. *J. Org. Chem.* **1990**, 55, 5543.



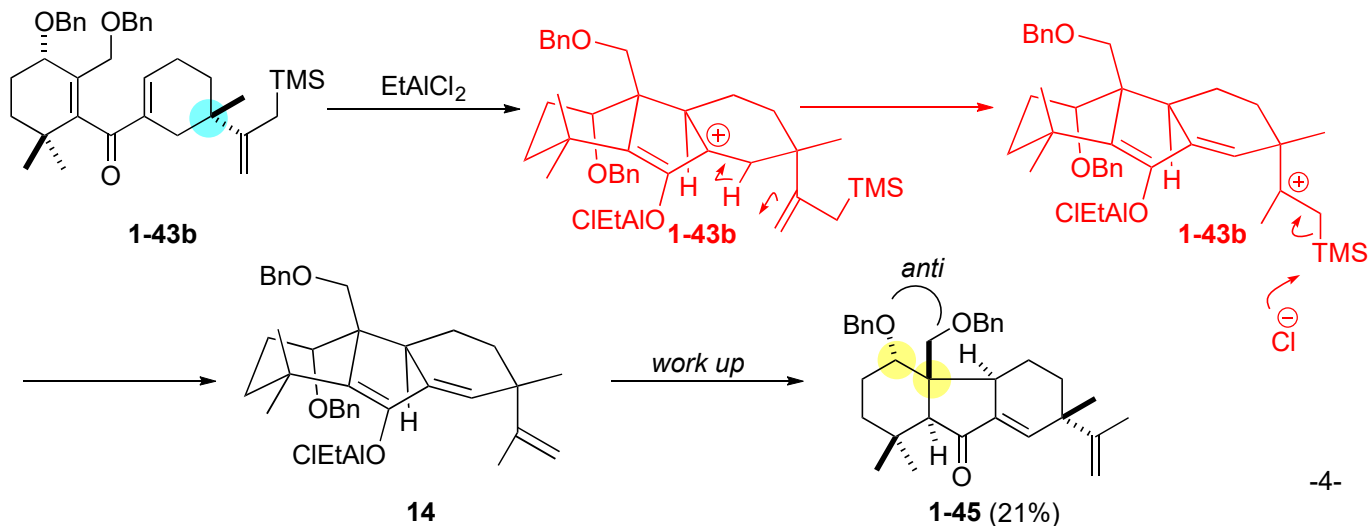
Model experiment for Nazarov cyclization and Hosomi-Sakurai reaction

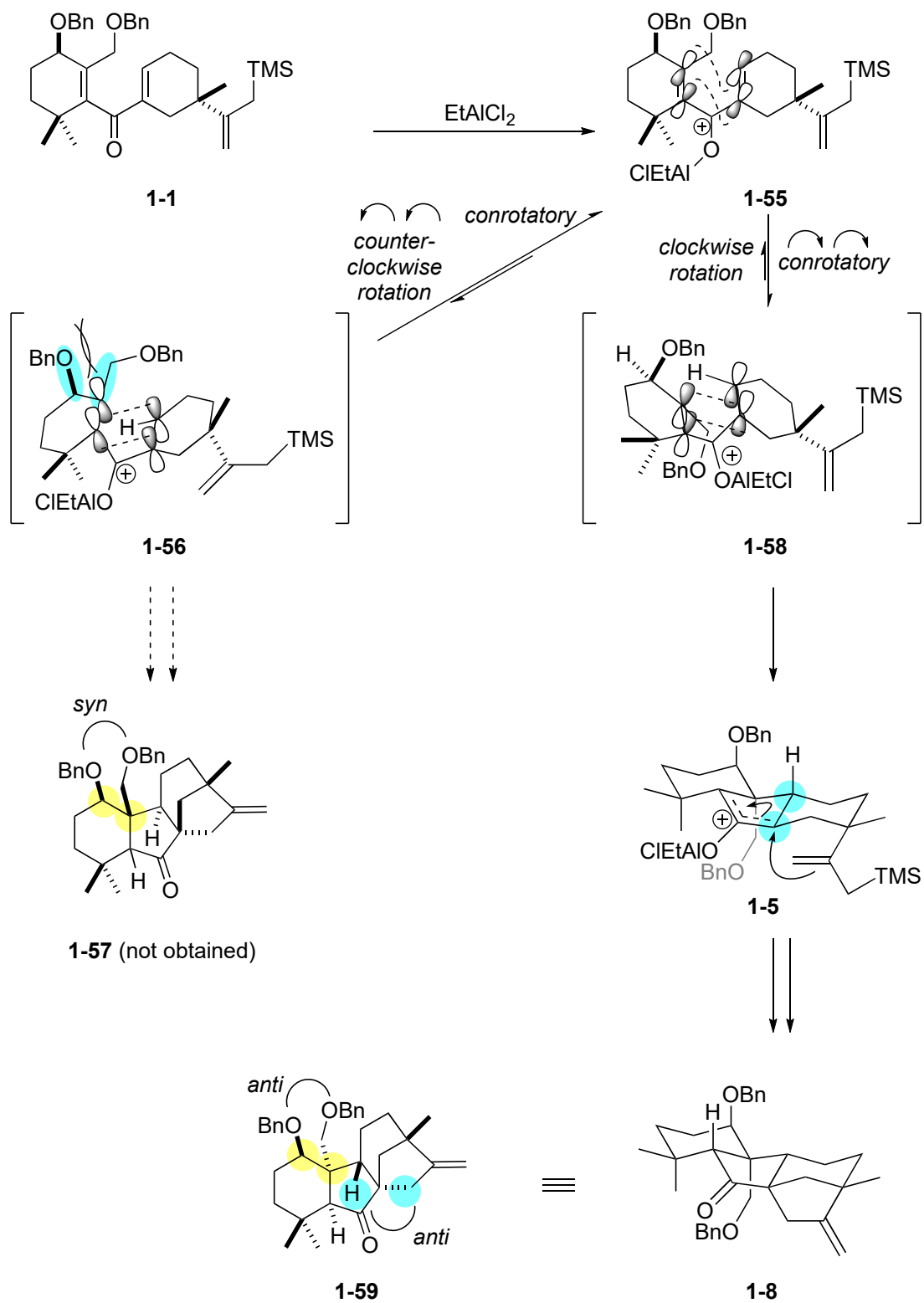


Torquoselectivity

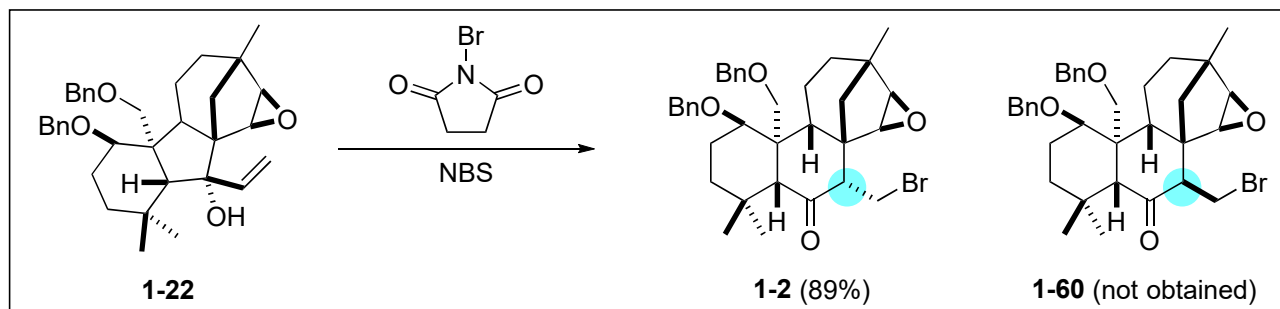


Hosomi-Sakurai reaction

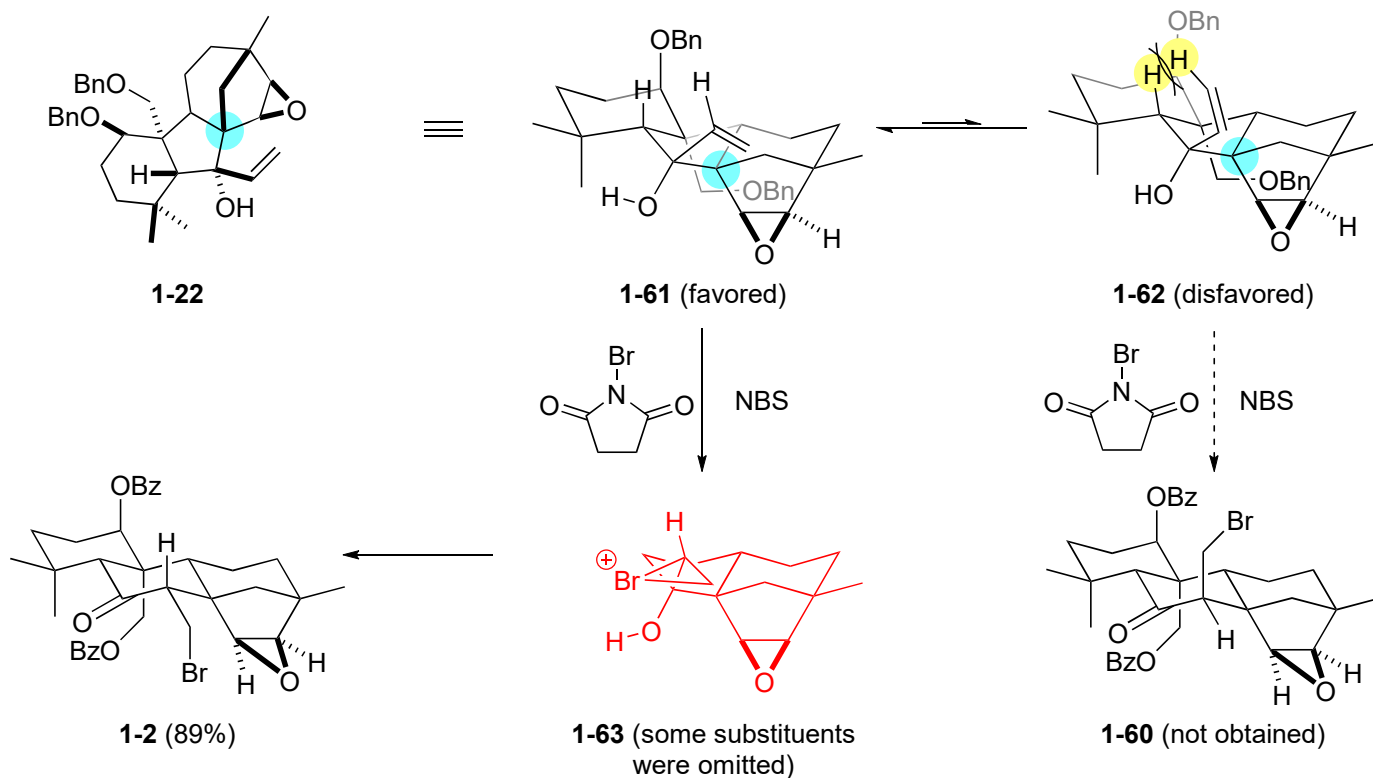




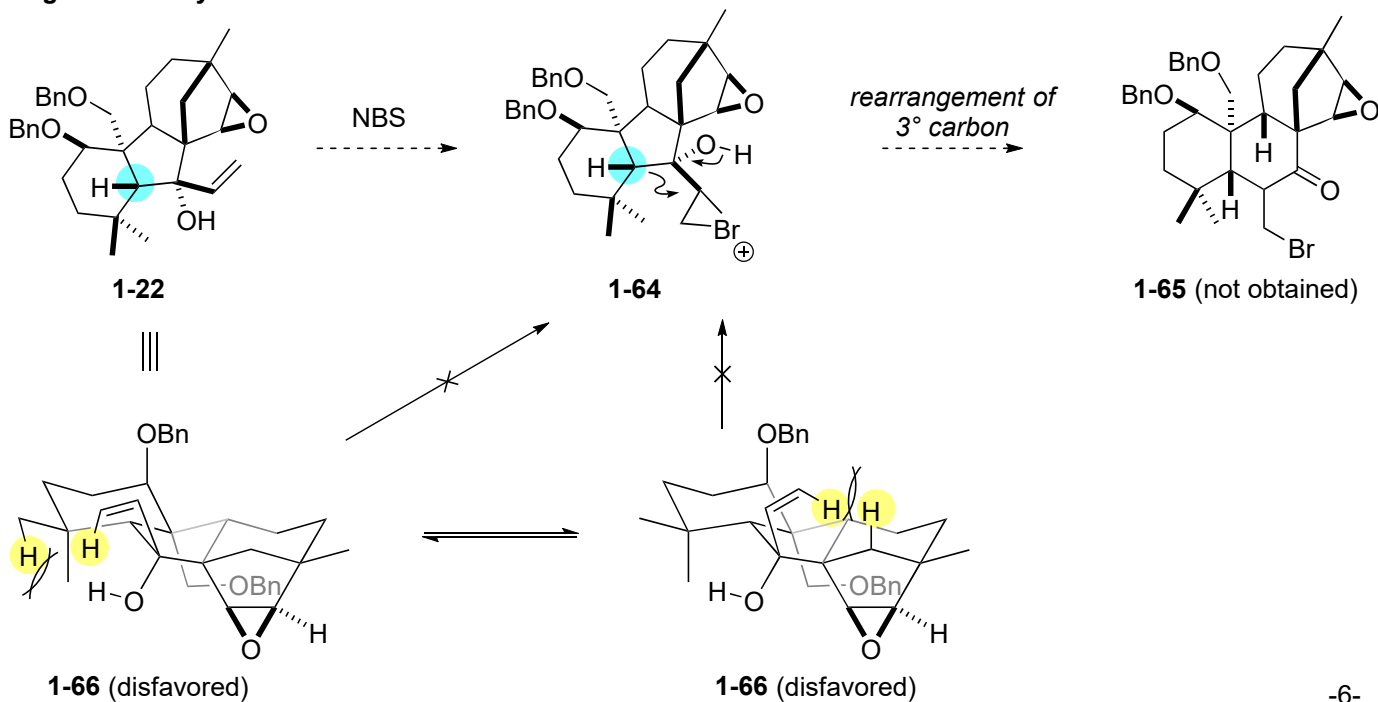
Discussion 1-2: stereoselectivity of rearrangement



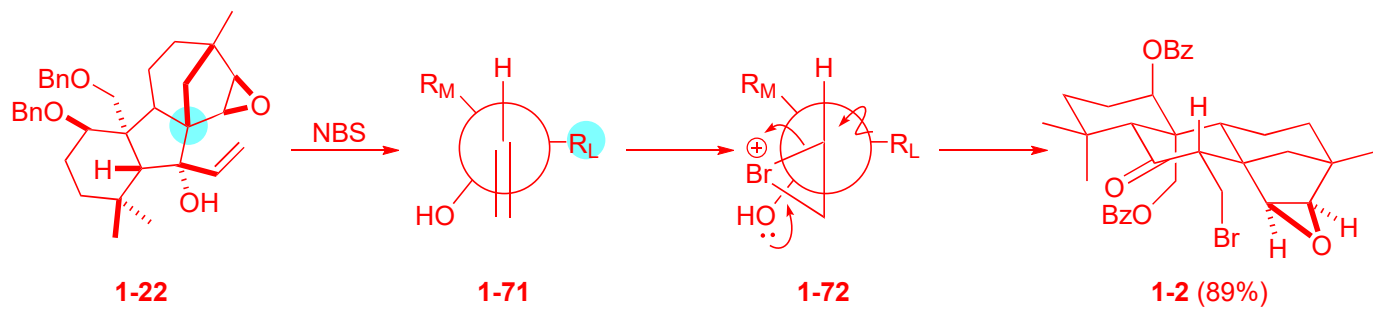
Stereoselectivity



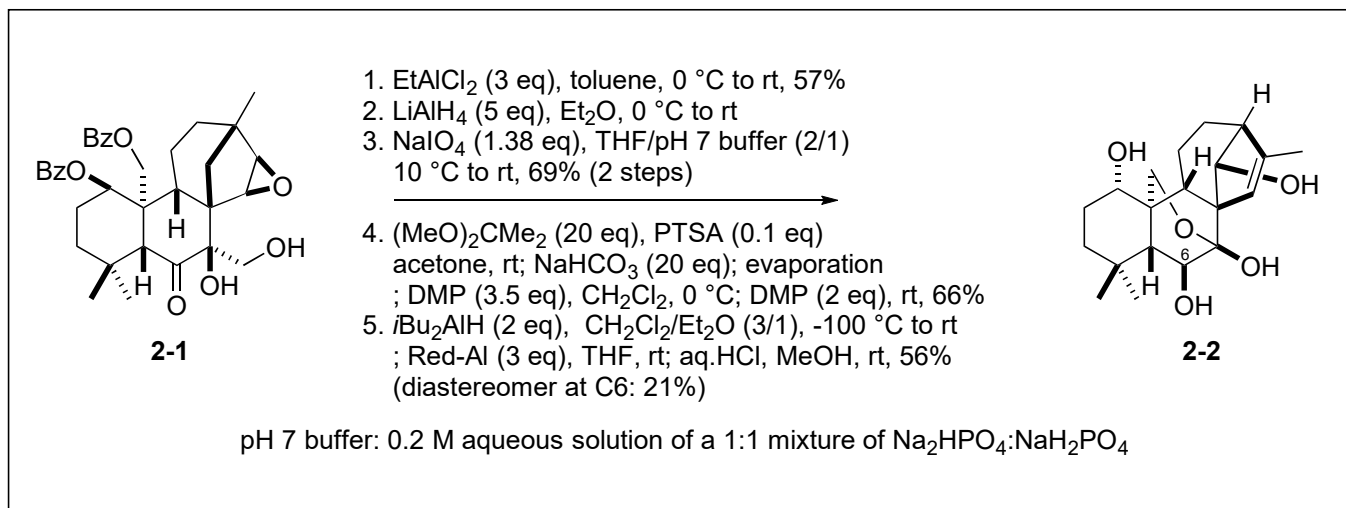
Regioselectivity



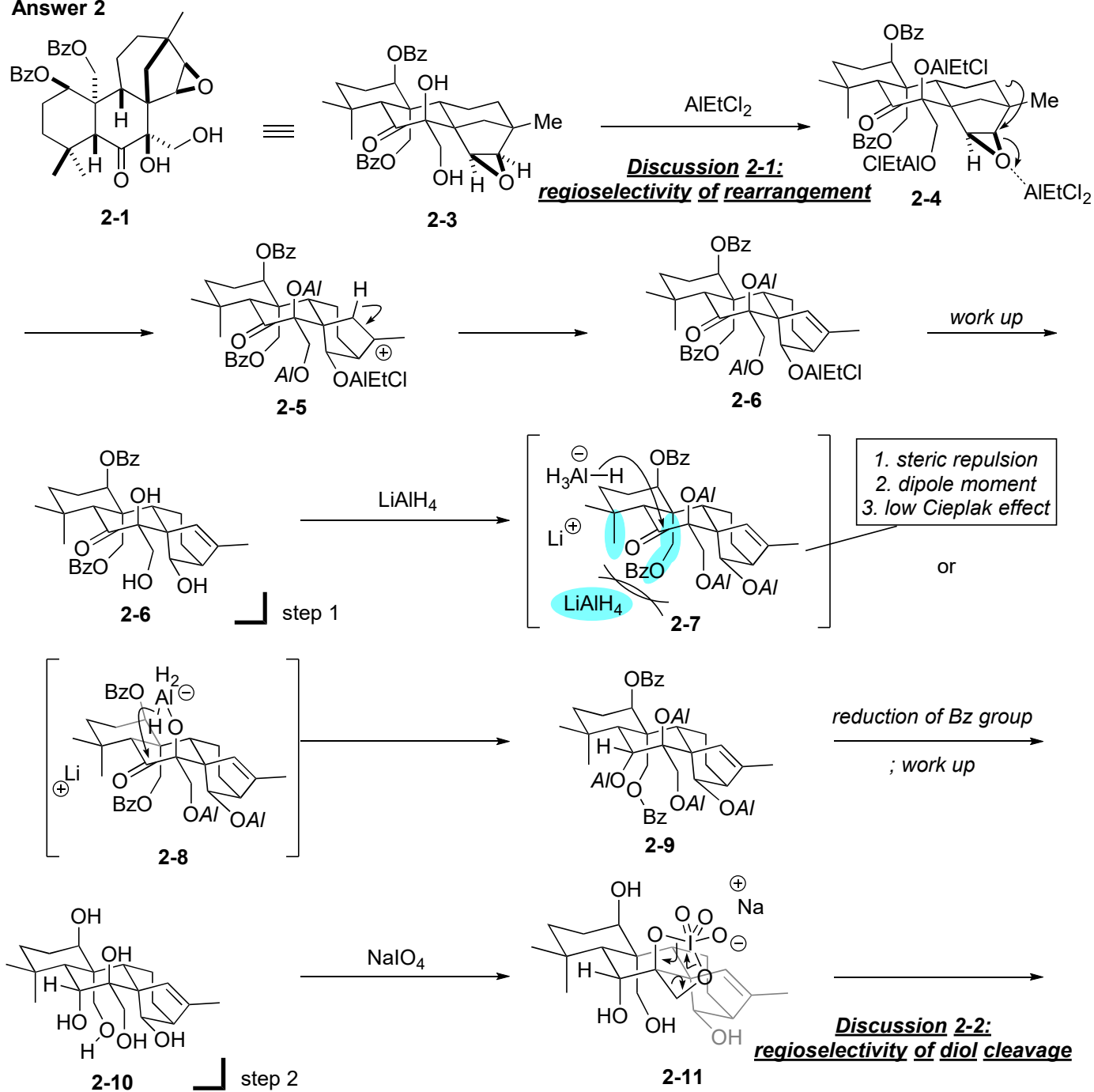
Other explanation

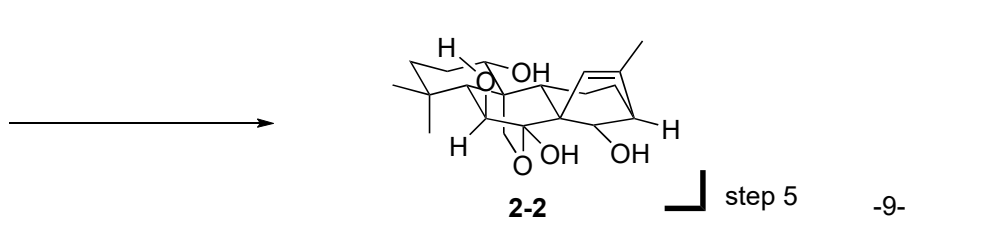
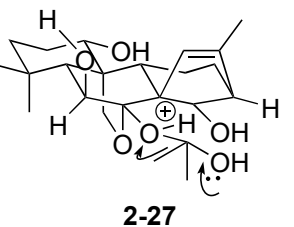
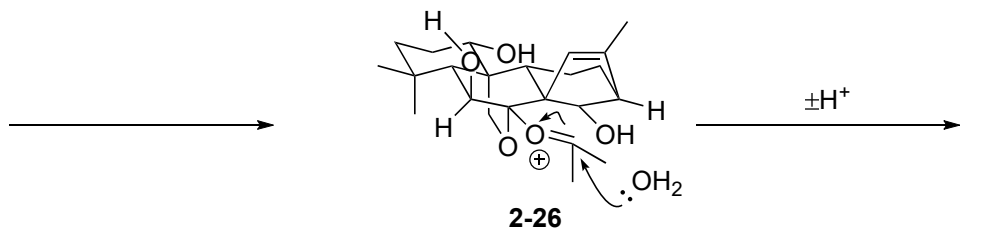
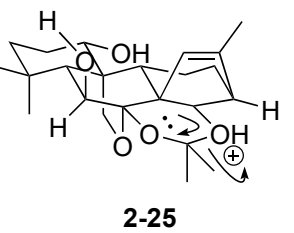
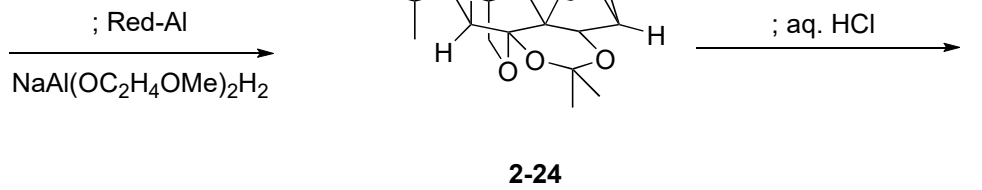
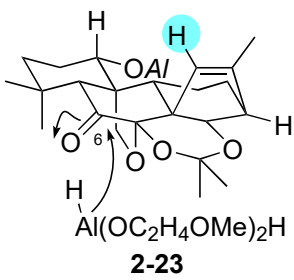
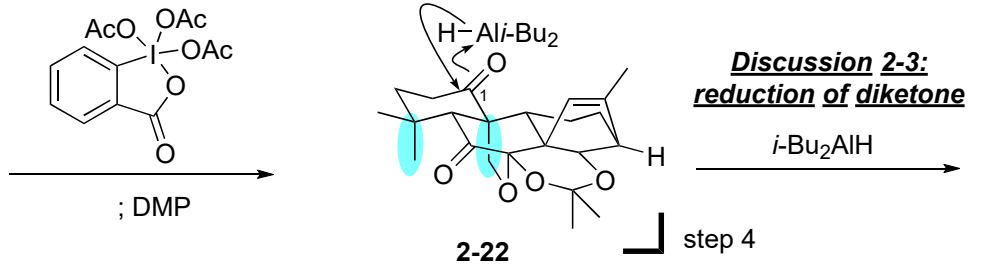
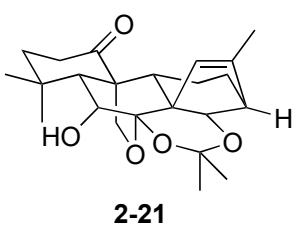
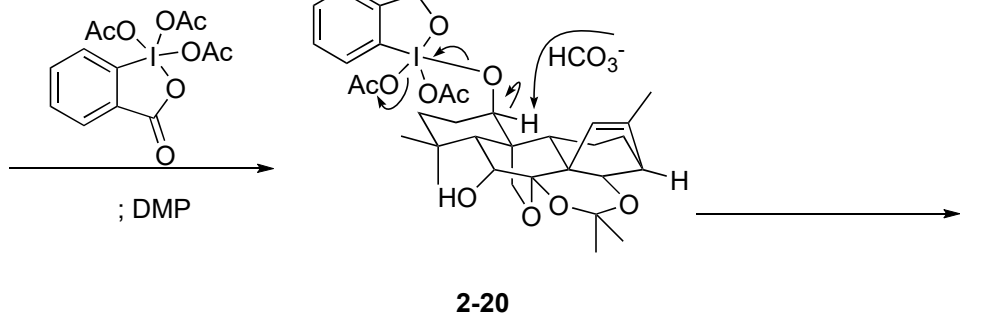
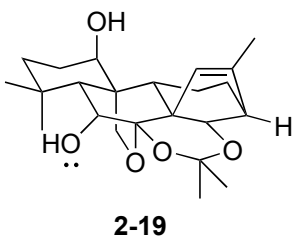
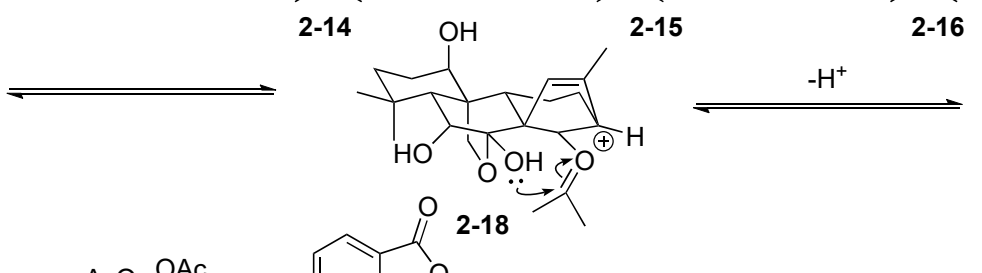
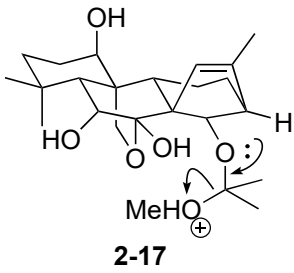
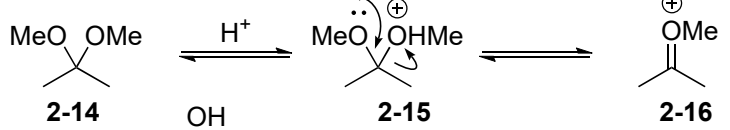
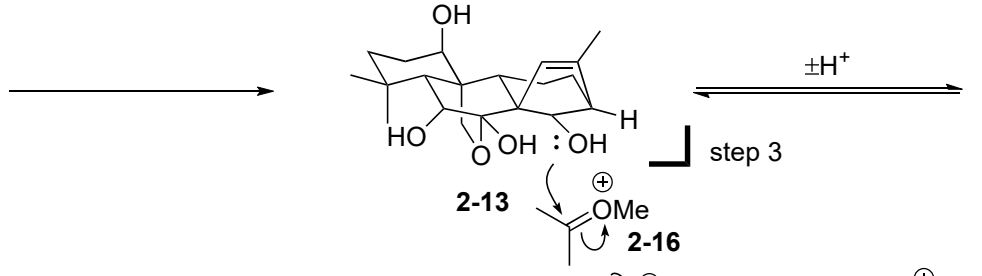
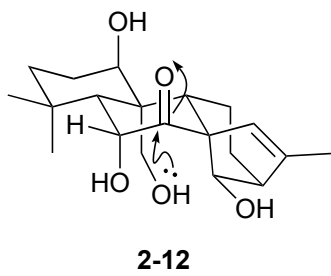


Problem 2



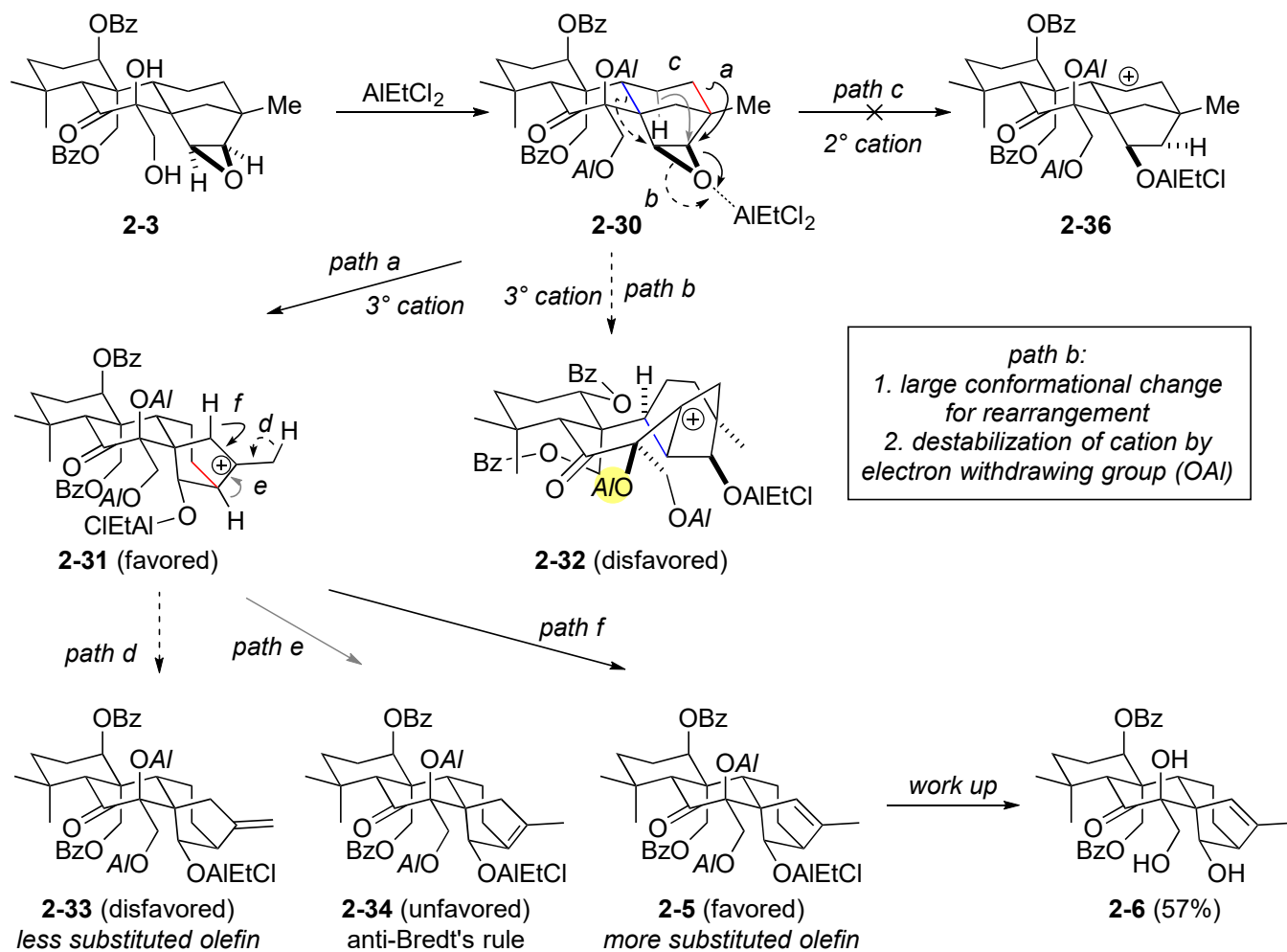
Answer 2



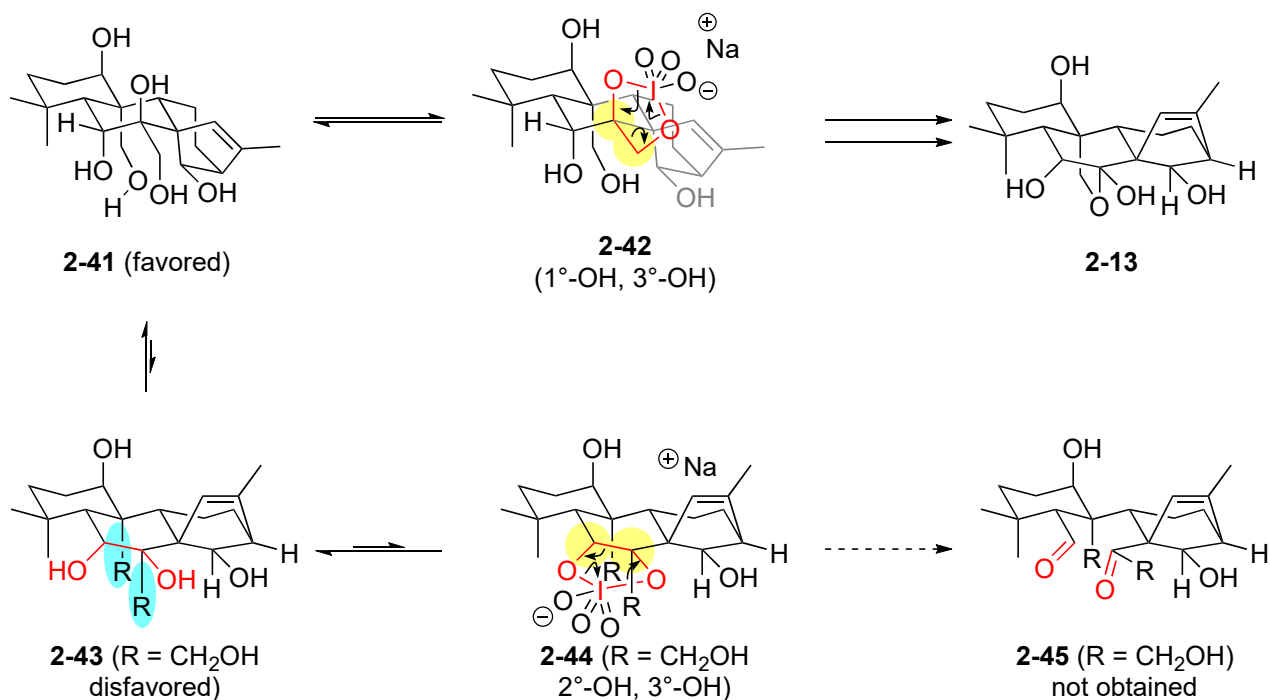


Discussion 2-3:
reduction of diketone

Discussion 2-1: regioselectivity of rearrangement



Discussion 2-2: regioselectivity of diol cleavage



Discussion 2-3: reduction of diketone

The order of reduction:

-> Reduction at C1-ketone proceeded first.

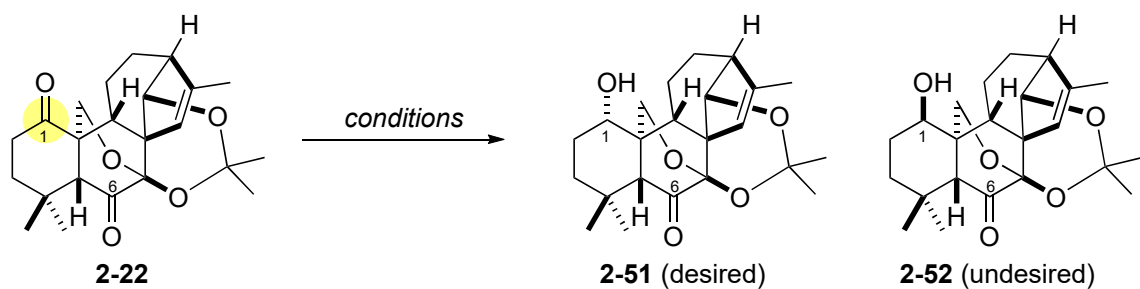


Table 2-1

entry	conditions	results
1	LiAlH ₄ (3 eq), THF, -78 °C	2-51:2-52=2.5:1
2	<i>i</i> -Bu ₂ AlH (3 eq), CH ₂ Cl ₂ /THF (3/1), -100 °C	2-51:2-52=8:1

