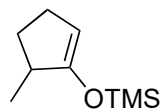


Problem Session (6)

2020/05/16 Akira Hirose

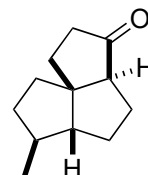
Please provide the reaction mechanisms.

1



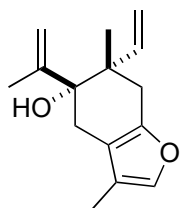
1-1 (racemic)

1. **1-2** (1.6 eq), TiCl₄ (2.0 eq)
Ti(O*i*-Pr)₄ (2.0 eq), CH₂Cl₂, -78 °C
 2. 10 % HCl aq/THF (1/1), rt
77% (2 steps, a mixture of 2 isomers)
dr. ca. 2 : 1
 3. KOH (3.4 eq), EtOH, rt
57%
(+undesired epimer: 29%)
-
4. hv, ethylene, CH₂Cl₂, -78 °C
73%
 5. AlCl₃ (2.0 eq), CH₂Cl₂, rt
93%



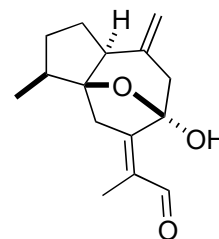
1-3 (racemic)

2



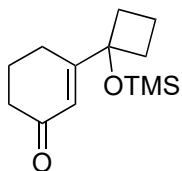
2-1

1. toluene, 140 °C
40%
(+undesired *endo*-olefin isomer: 8%)
2. CDCl₃, 5 °C
(reacted with O₂)
99%



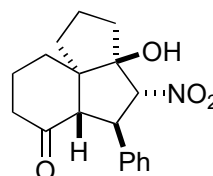
2-2

3

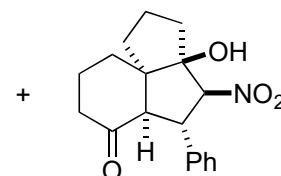


3-1 (racemic)

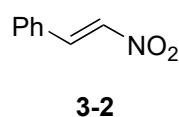
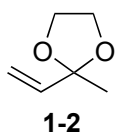
- TMSOTf (1.5 eq), CH₂Cl₂, 0 °C;
TiCl₄ (2.2 eq), **3-2** (1.2 eq), -78 °C
-
- 72% (**3-3** : **3-3'** = 11.4 : 1)



3-3 (racemic)



3-3' (racemic)

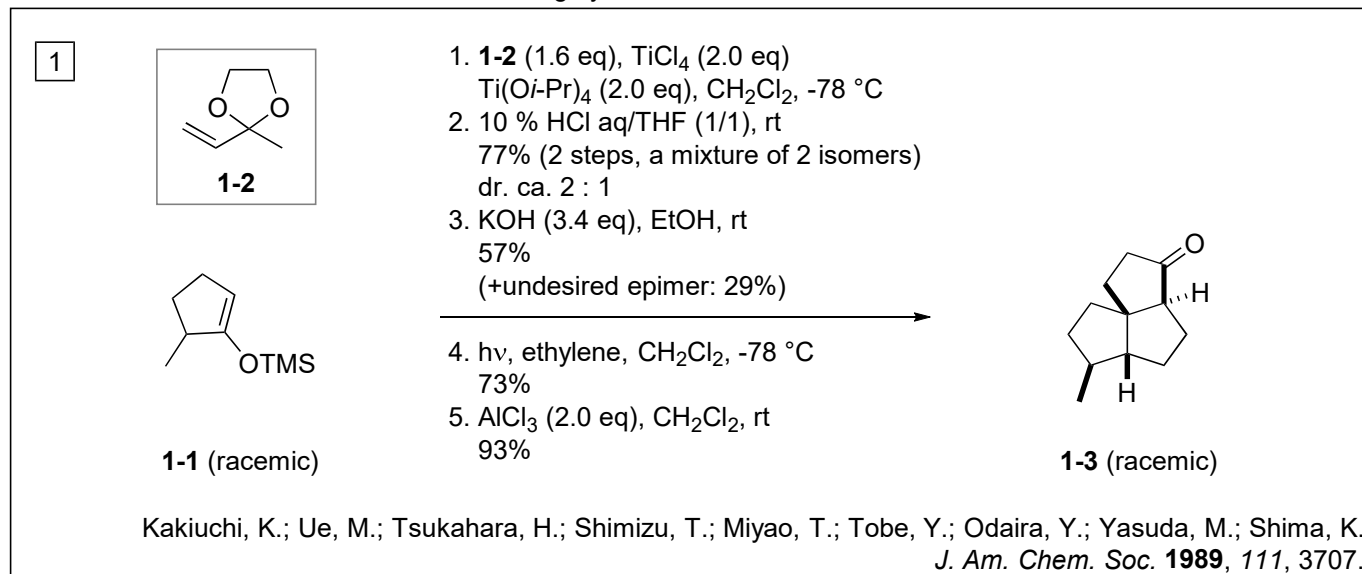


Problem Session (6)

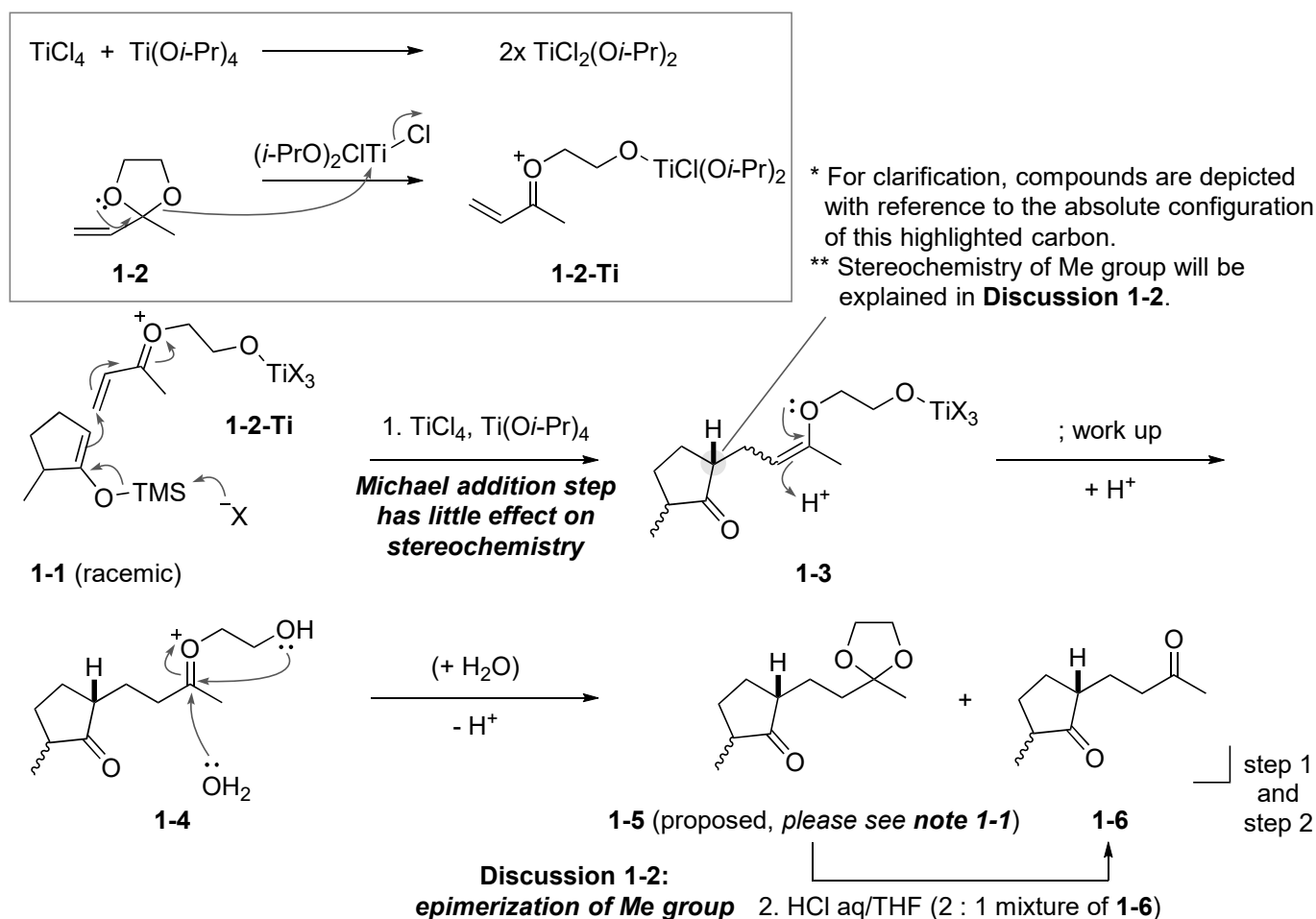
2020/05/16 Akira Hirose

Topic: Construction fused ring systems by using ring expansion/transannular reaction

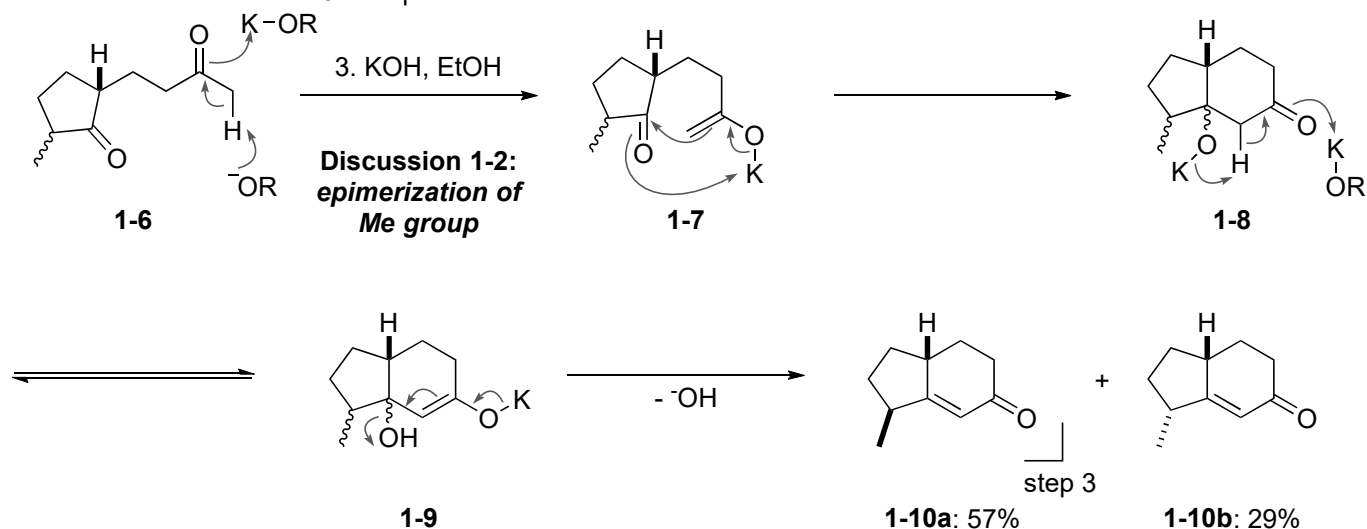
Problem 1. Construction of 5/5/5-*cis*-fused ring system



1-1. The mechanisms of the 1st and 2nd steps: Mukaiyama-Michael reaction

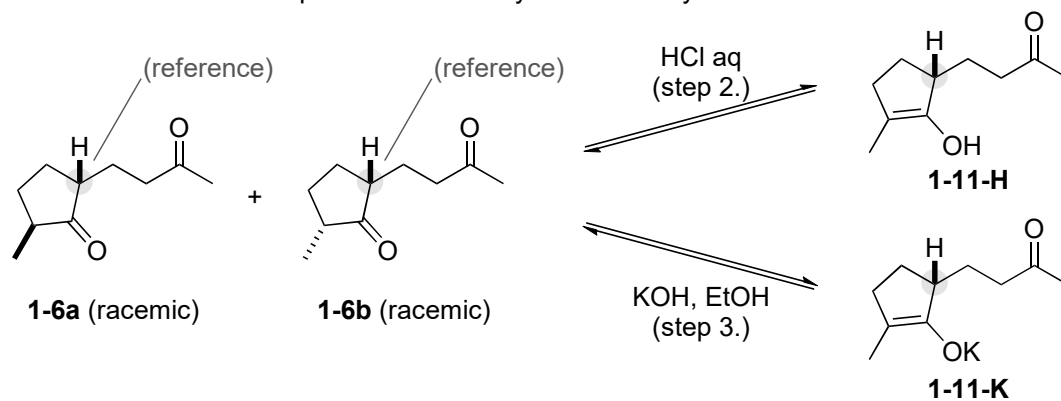


1-2. The mechanisms of the 3rd step: Robinson annulation

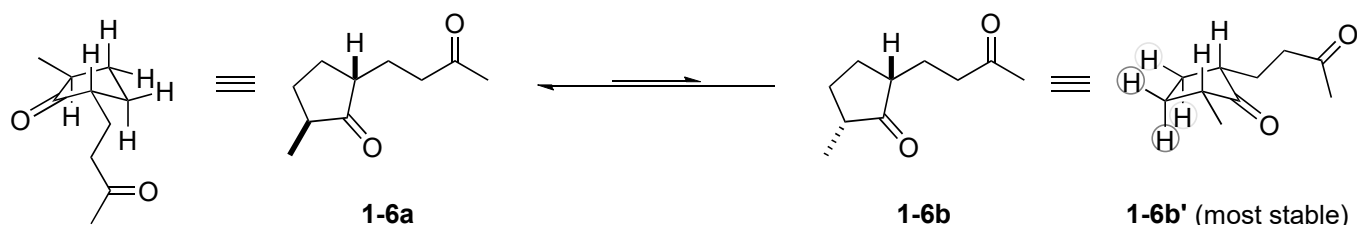
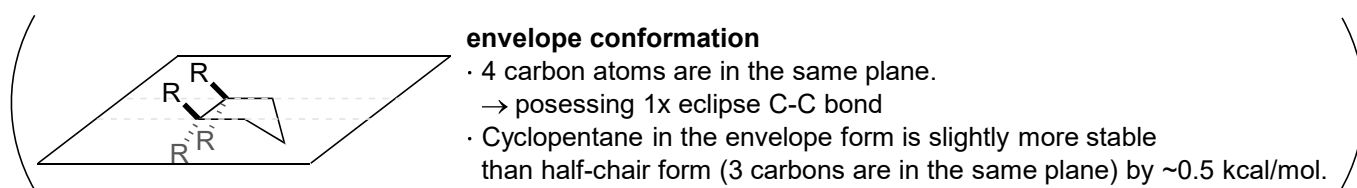


Discussion 1-2. Stereoselective 1-10a production

In step 2 (acidic conditions) and step 3 (basic conditions), **1-6a** and **1-6b** can epimerize via enol(ate) **1-11**.
 → Ratio of **1-6a** : **1-6b** depends on thermodynamic stability of both of them.



→ Considering the envelope conformation of cyclopentanone **1-6s**, **1-6a** would be slightly more stable than **1-6b** due to less H-H torsional strain.



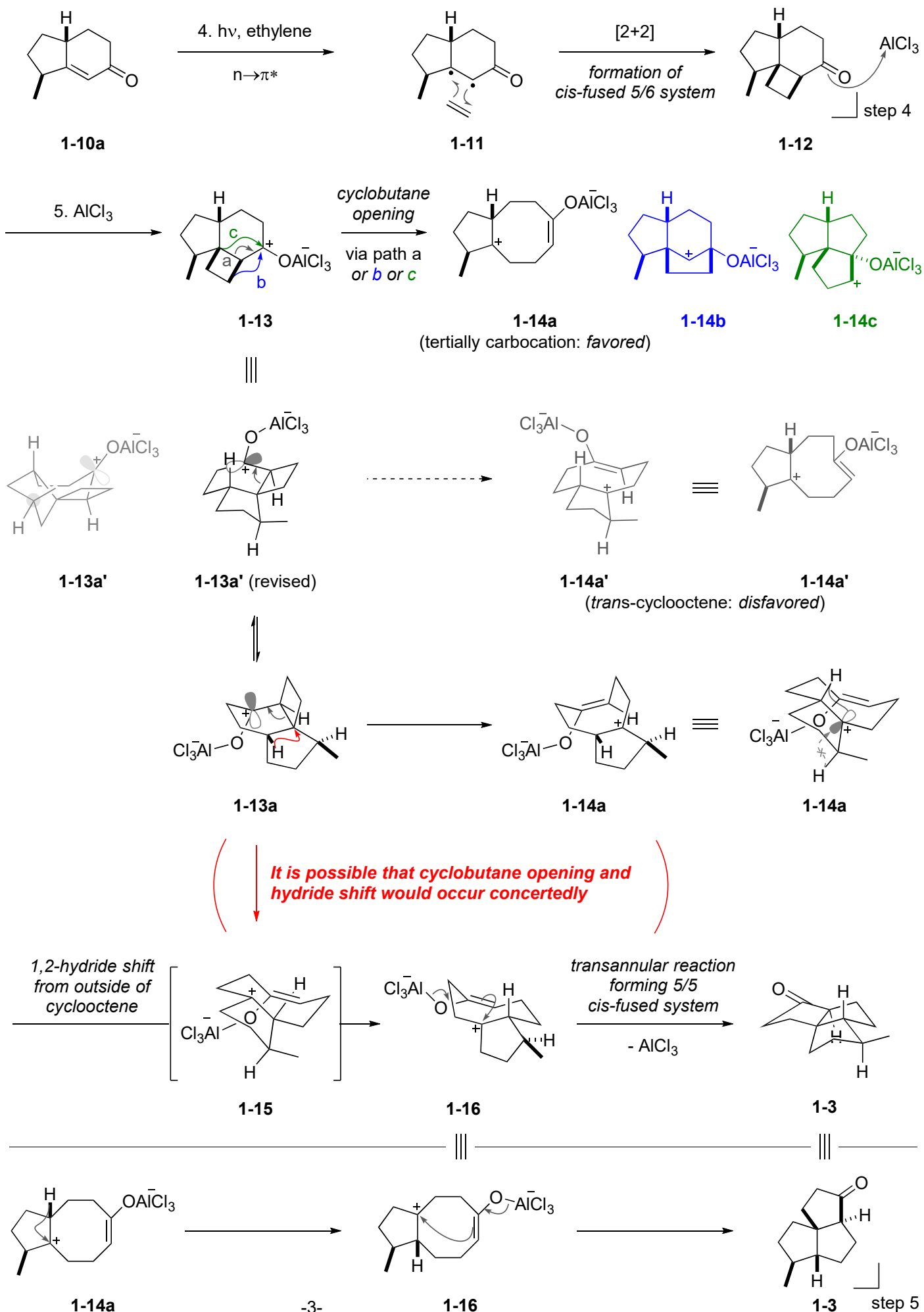
1-6a' (most stable)

• 2x 1,3-diaxial interaction

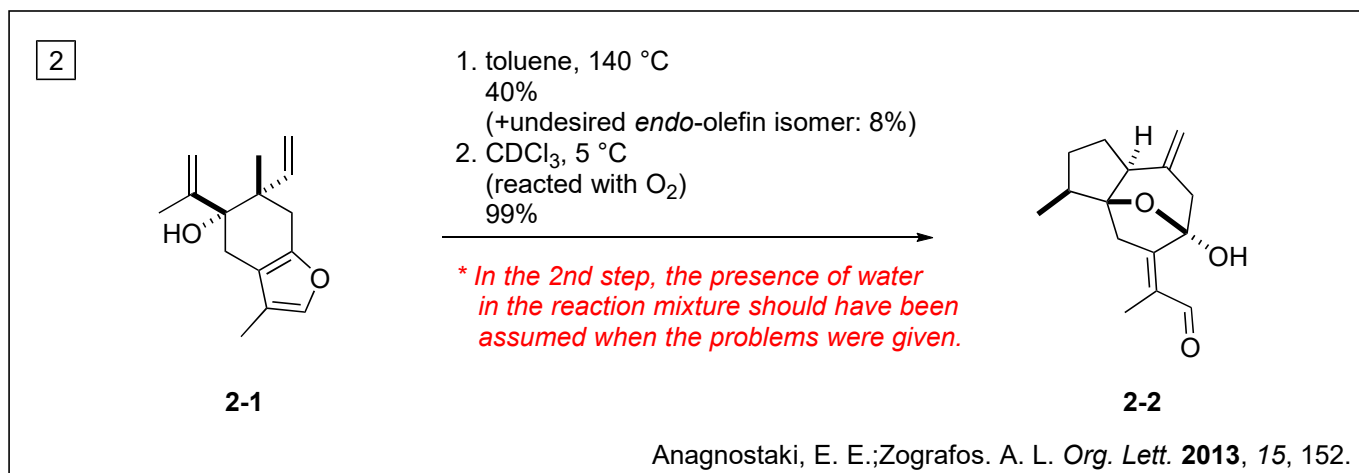
• 1x 1,3-diaxial interaction
 • **2x H-H torsional strain**

→ This difference of the stability would determine the ratio of **1-6** mixture (ca. 2 : 1) and lead to stereoselectivity of Robinson annulation (**1-10a**: 57%, **1-10b**: 29%).

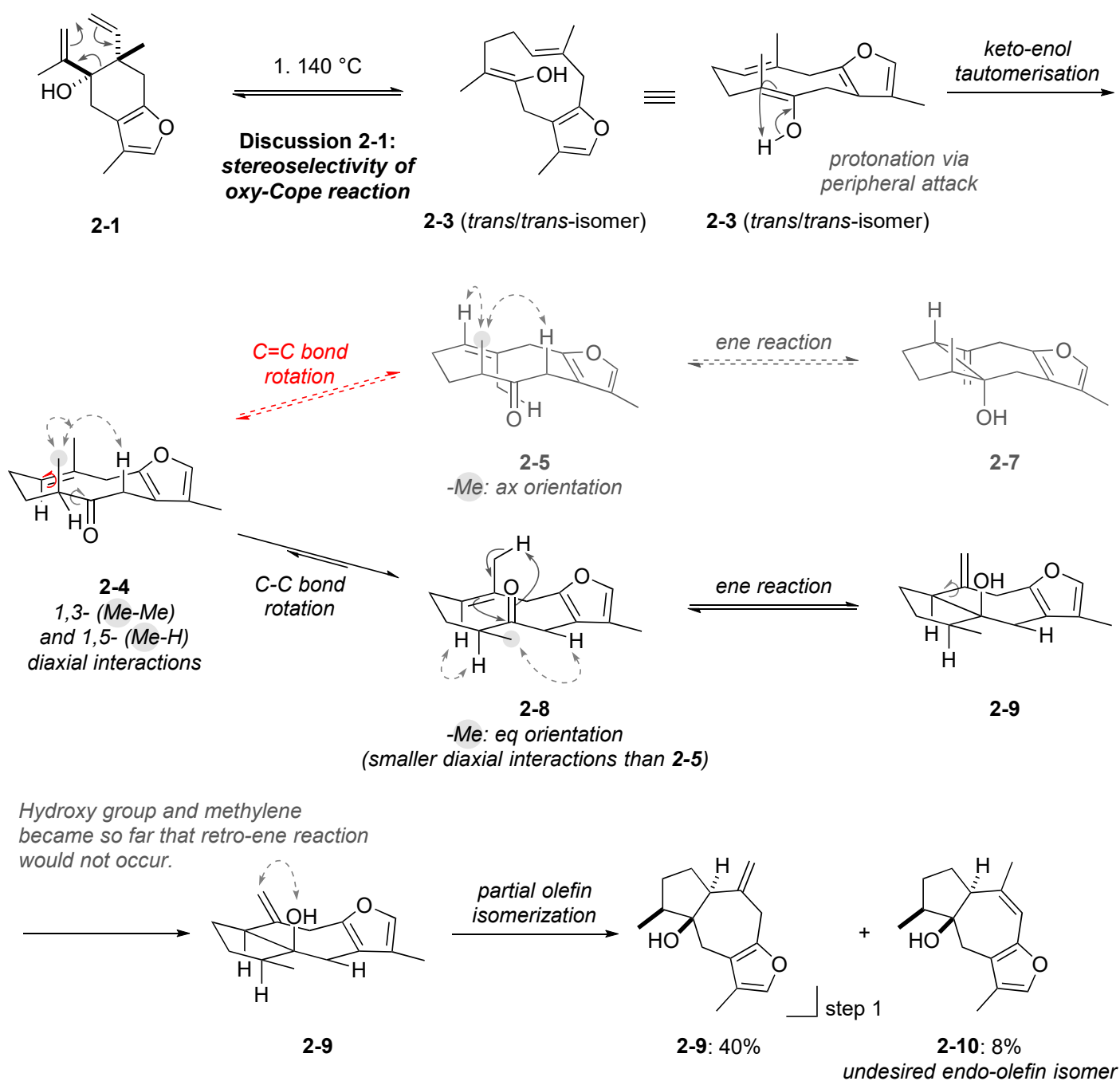
1-3. The mechanisms of the 4th and 5th steps



Problem 2. Synthesis of guaiane sesquiterpenes

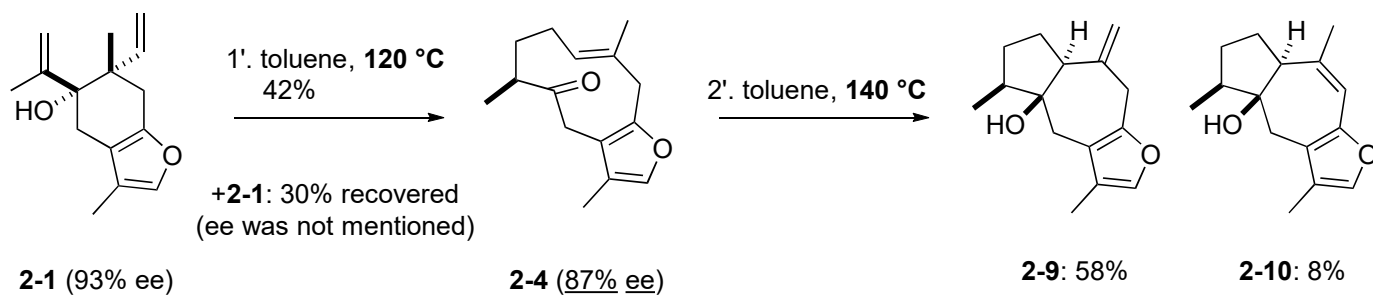


2-1. The mechanism of the 1st step: oxy-Cope/ene reaction



Discussion 2-1. Stereoselectivity of Cope reaction

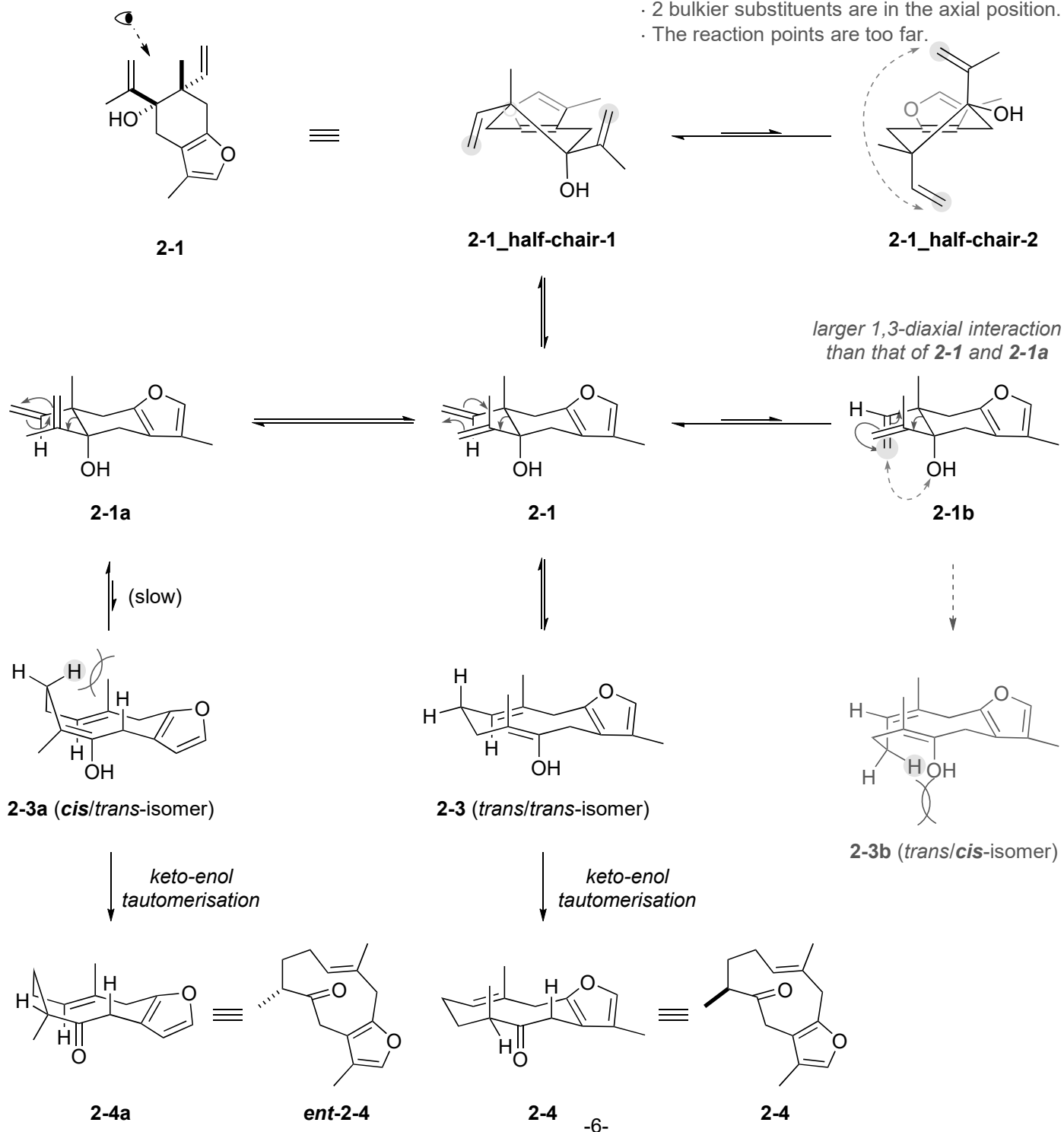
2-1-1. 2 step-conversion from 2-1 to 2-9



Anagnostaki, E. E.; Zografos, A. L. *Org. Lett.* **2013**, *15*, 152.

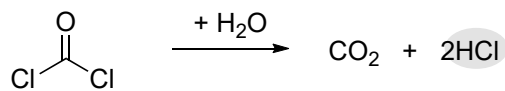
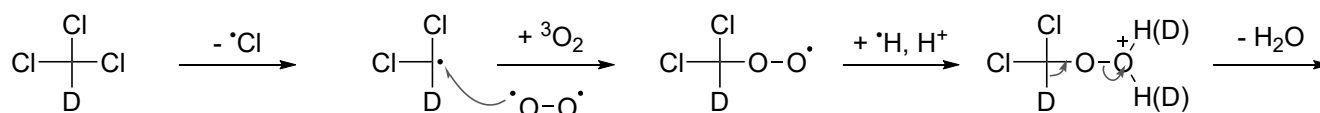
2-1-2. Plausible explanation for stereoselective producing of 2-3

- 2 bulkier substituents are in the axial position.
- The reaction points are too far.



2-2. The mechanism of the 2nd step: autoxidation of furan ring (the reaction with $^3\text{O}_2$)

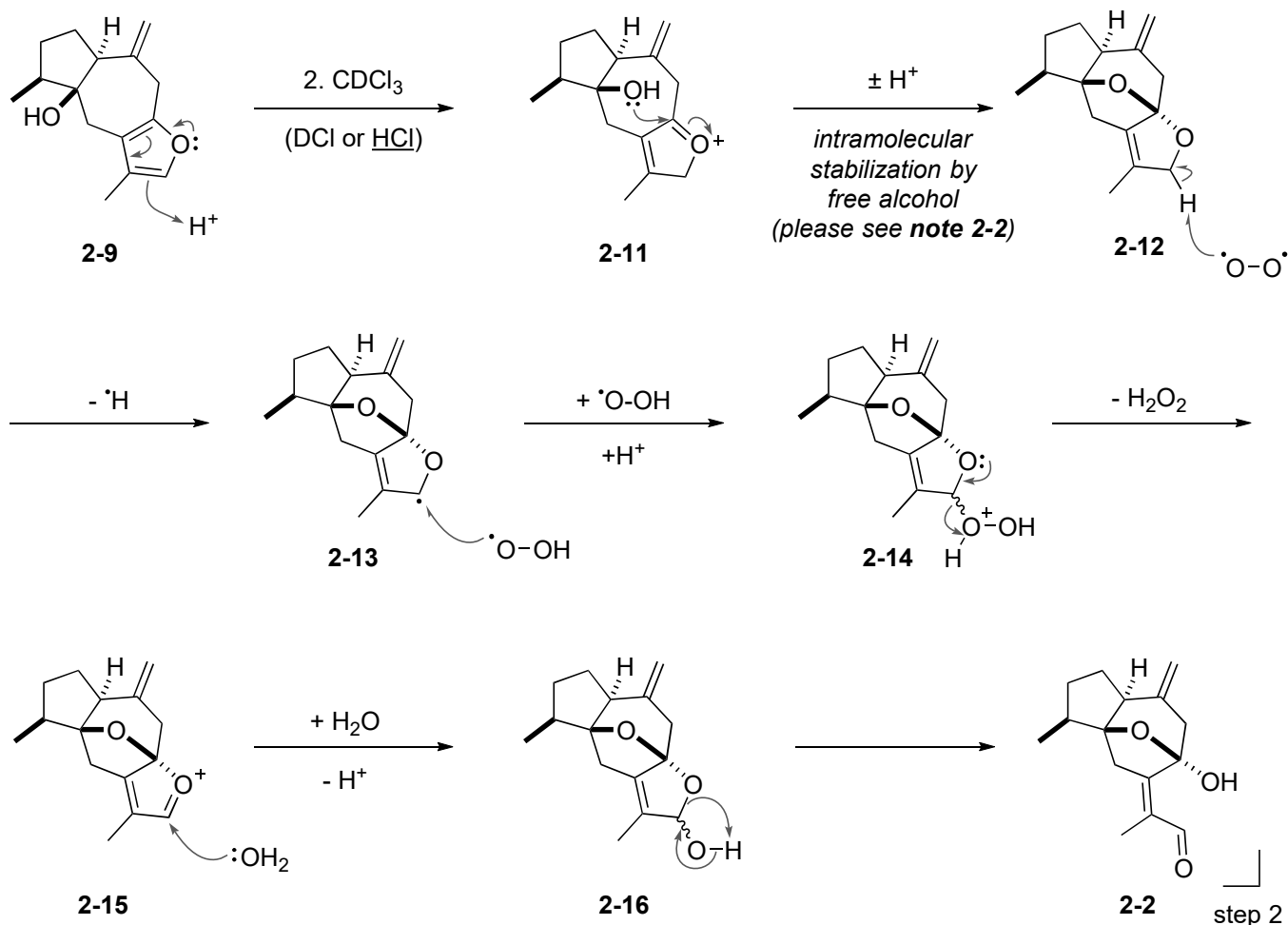
Oxidative decomposition of CDCl_3 to phosgene and DCI



(phosgene)

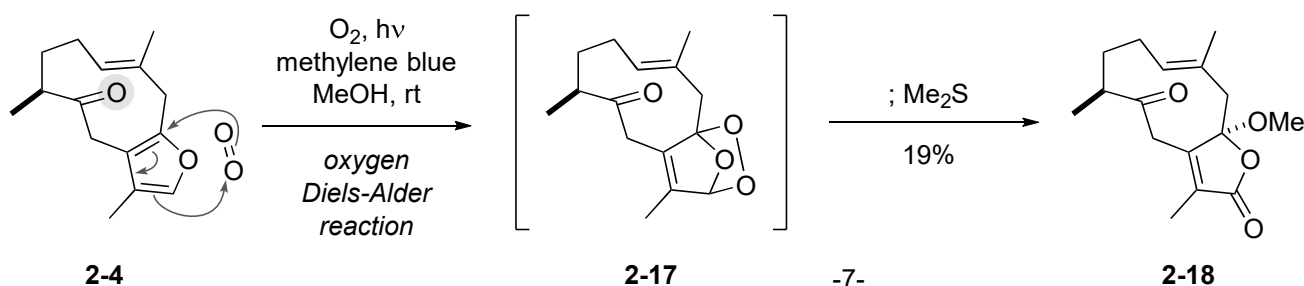
Chapman, A. T. *J. Am. Chem. Soc.* **1935**, *57*, 419.

(* Although the presence of water is not definitely mentioned in the paper, it is more natural to assume that water involved the reaction mechanisms.)

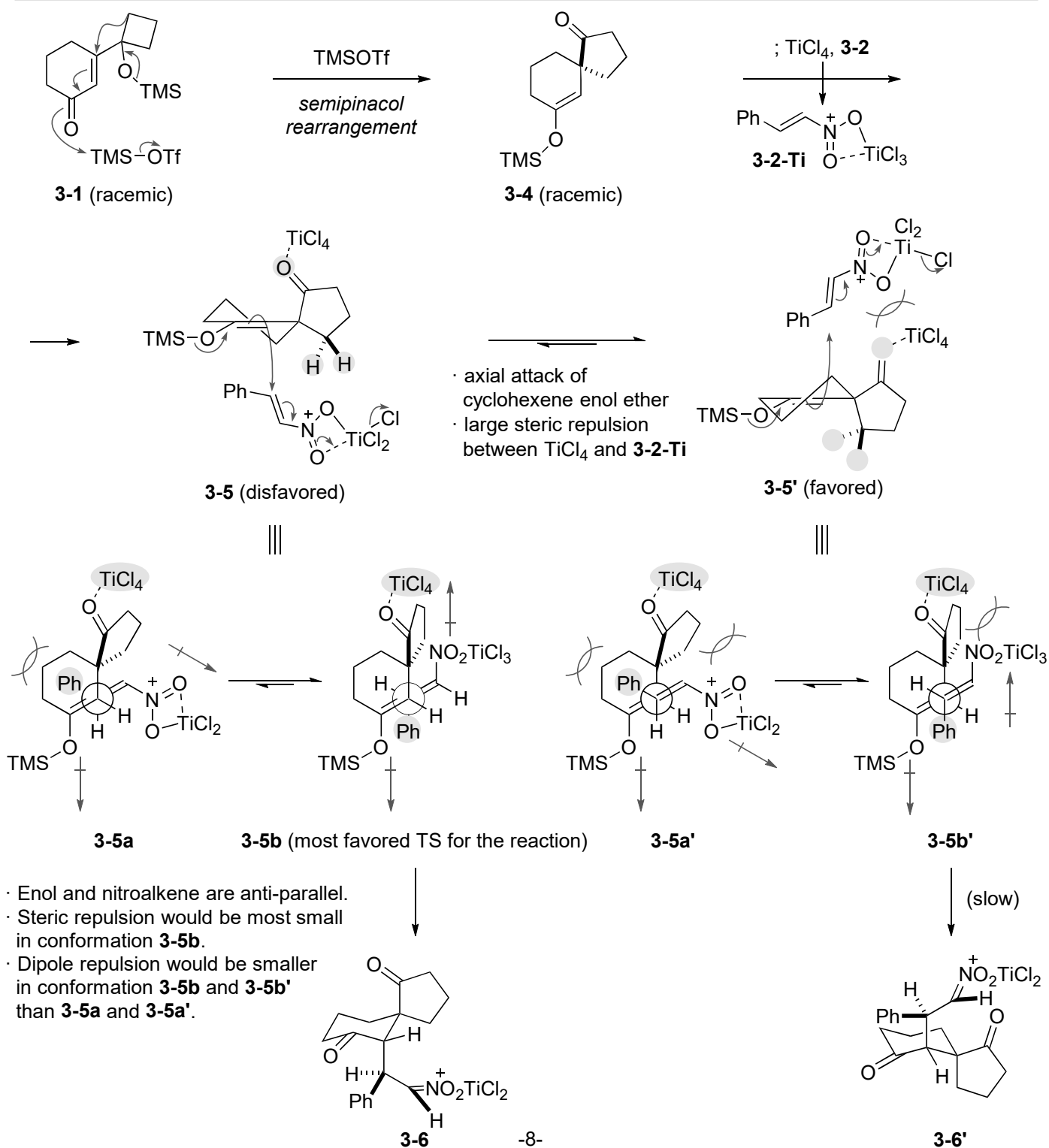
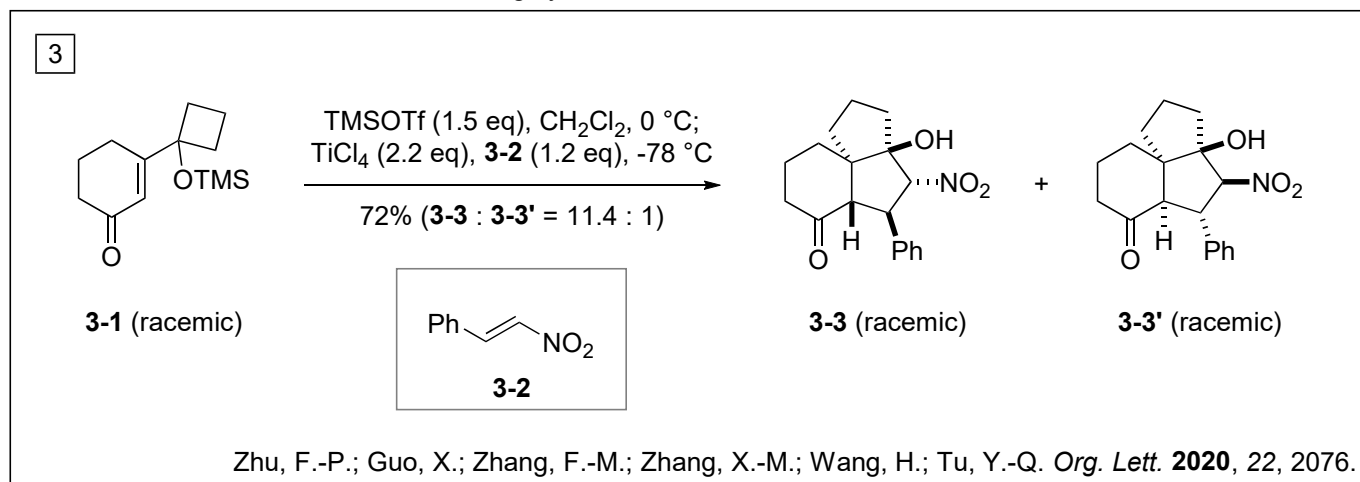


Note 2-2. Oxidation of the furan ring in 2-4 (the reaction with $^1\text{O}_2$)

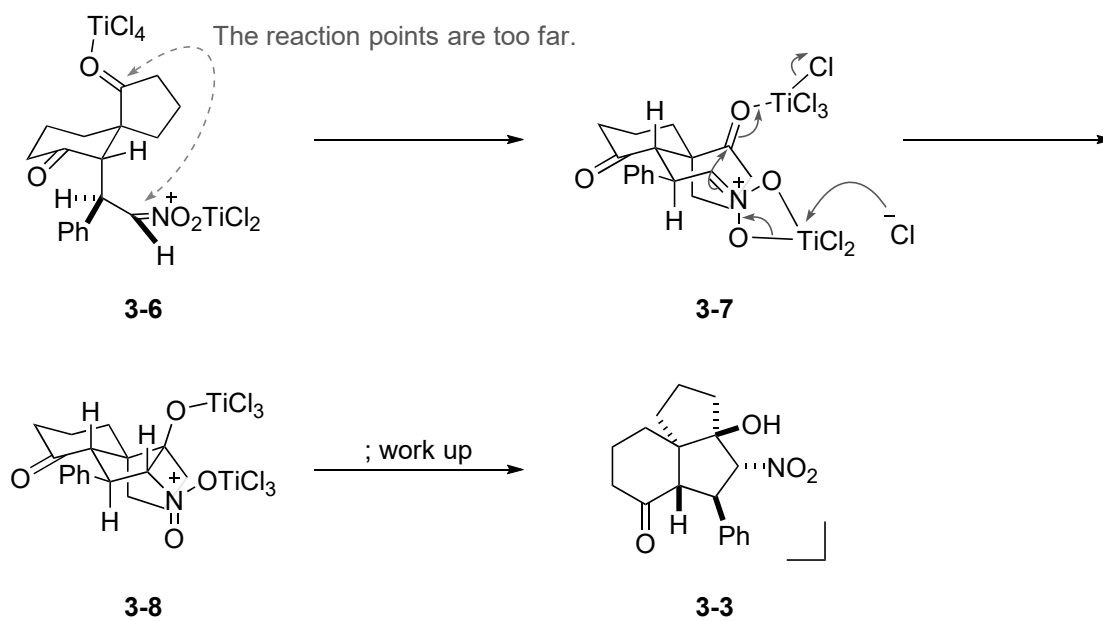
- Autoxidation of **2-4** (possessing ketone instead of free alcohol) did not occur.
- Oxidation of furan in **2-4** required production of singlet oxygen.



Problem 3. Construction of 6/5/5-fused ring system



3-1. The mechanisms from **3-6** to **3-3** (a major isomer)



3-2. The mechanisms from **3-6'** to **3-3'** (a minor isomer)

