

Problem Session (1) [Answer]

2017.7.15 Keshu Zhang

Topic: Formal total synthesis of Lycojaponicum C

0. Introduction

0-1. Isolation

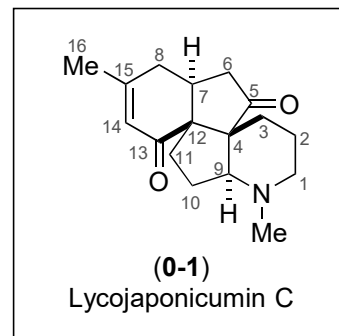
from the traditional Chinese medicine, *Lycopodium japonicum*.
(Li, Y. *et al. Org. Lett.* **2012**, *14*, 2614.)

0-2. Biological activity

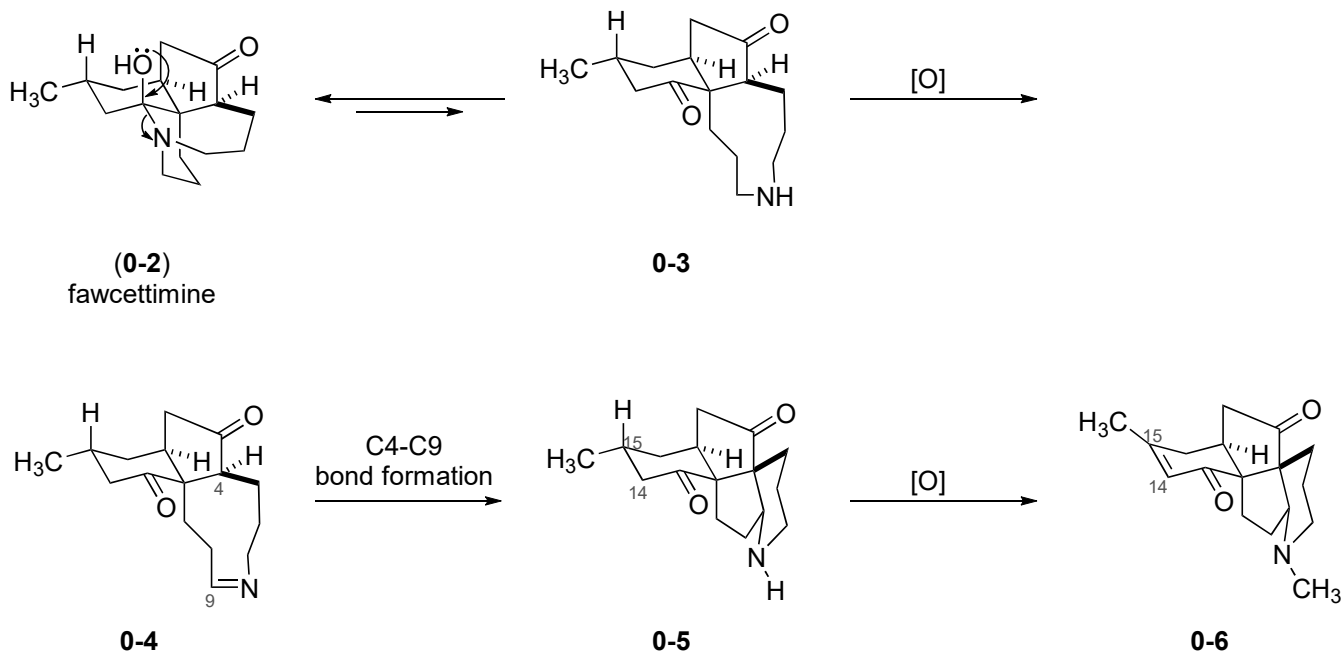
moderate inhibition toward lipopolysaccharide (LPS)-induced pro-inflammatory factors.
inactive against acetylcholinesterase.
(Li, Y. *et al. Org. Lett.* **2012**, *14*, 2614.)

0-3. Structural features:

-Fused 6/5/5/6 tetracyclic skeleton.
-A bicyclic [3.3.0] core with four continuous stereocenters,
including two vicinal quaternary carbon centers.



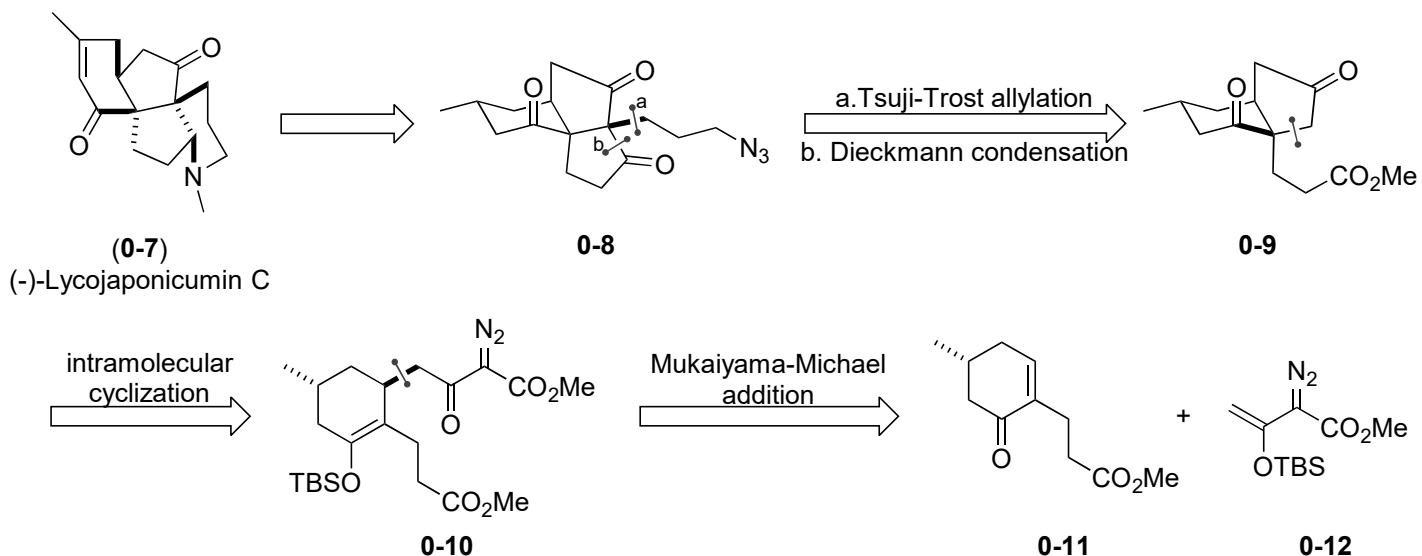
0-4. Proposed biosynthetic pathway for the Lycojaponicum skeleton.



Li, Y. *et al. Org. Lett.* **2012**, *14*, 2614.

0-5. Total syntheses of Lycojaponicum C

0-5-1. Tu's group (asymmetric)

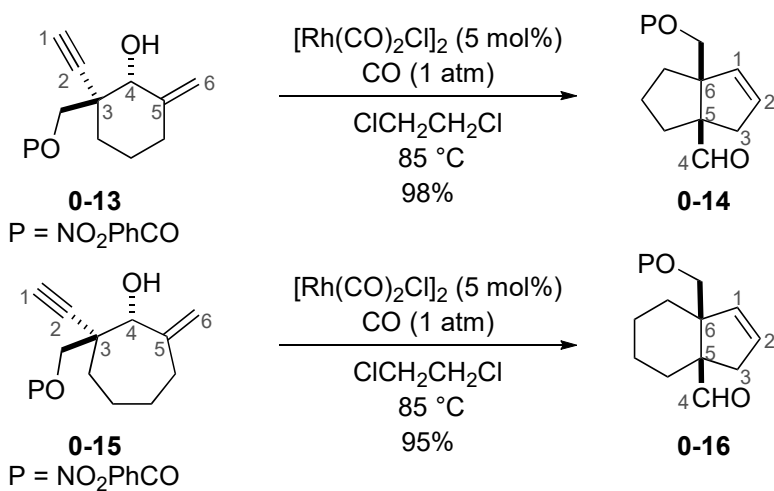


Hou, S.-H.; Tu, Y.-Q.; Liu, L.; Zhang, F.-M.; Wang, S.-H.; Zhang, X.-M. *Angew. Chem., Int. Ed.* **2013**, *52*, 11373.

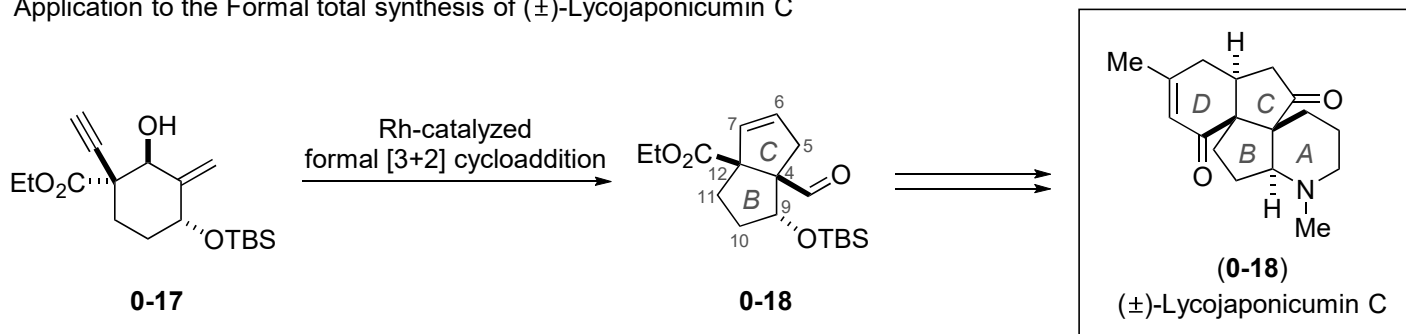
0-5-2. Yang's group (formal synthesis, racemic, **Problem session**)

Authors developed the Rh-catalysed [3+2] cycloaddition and applied it to the total synthesis.
(160514_PS_Komei_Sakata_lingzhiol;)

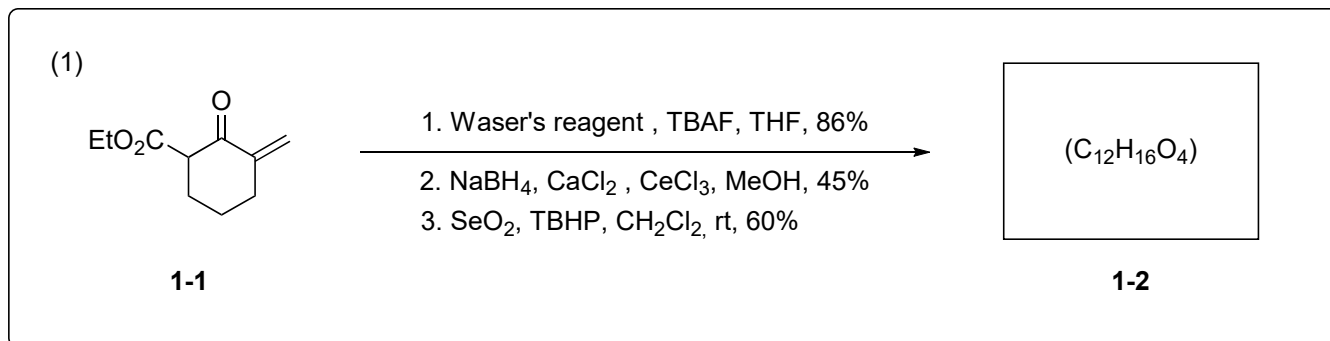
Rh-catalysed [3+2] cycloaddition (*Nat. Commun.* **2014**, *5*, 5707.)



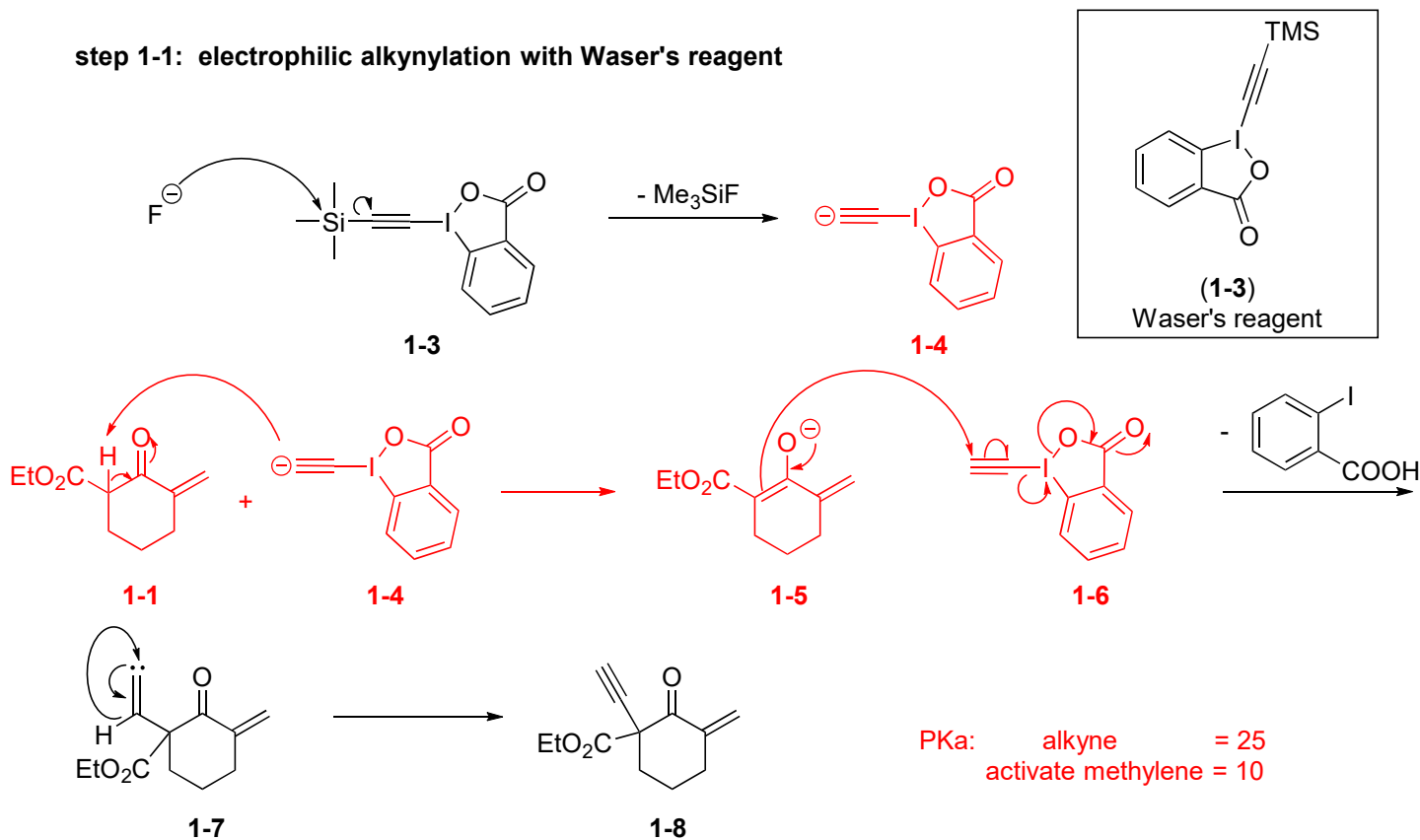
Application to the Formal total synthesis of (±)-Lycojaponicum C



Answer

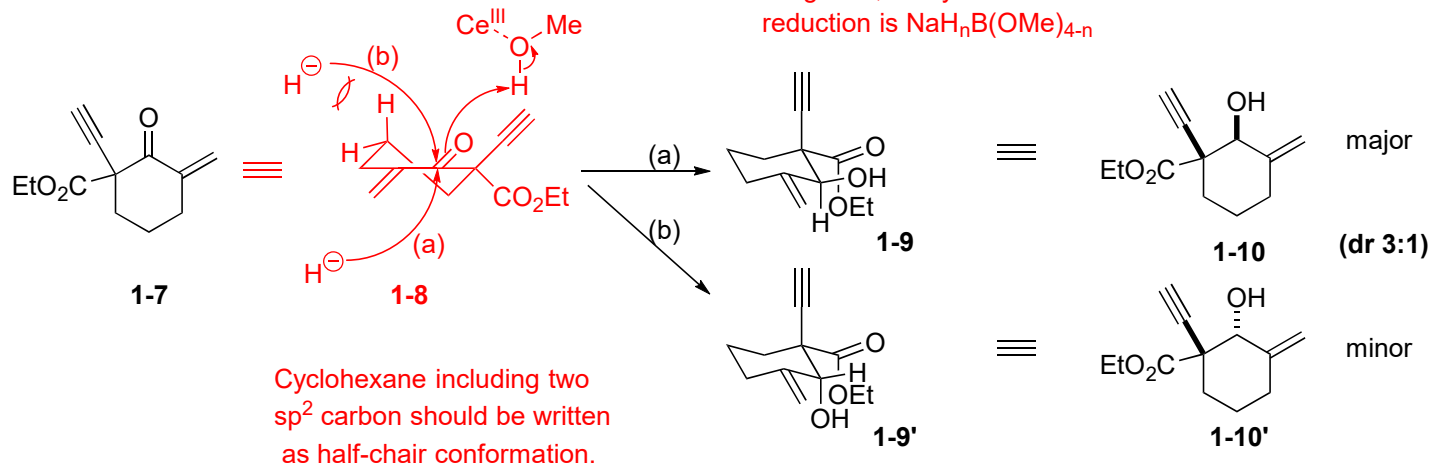


step 1-1: electrophilic alkylation with Waser's reagent

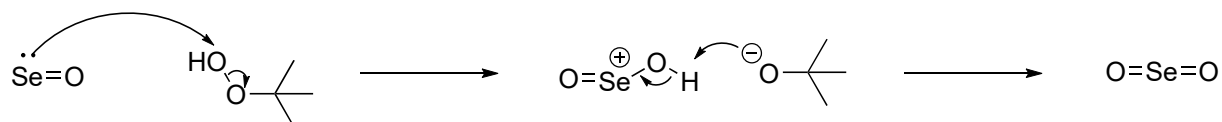
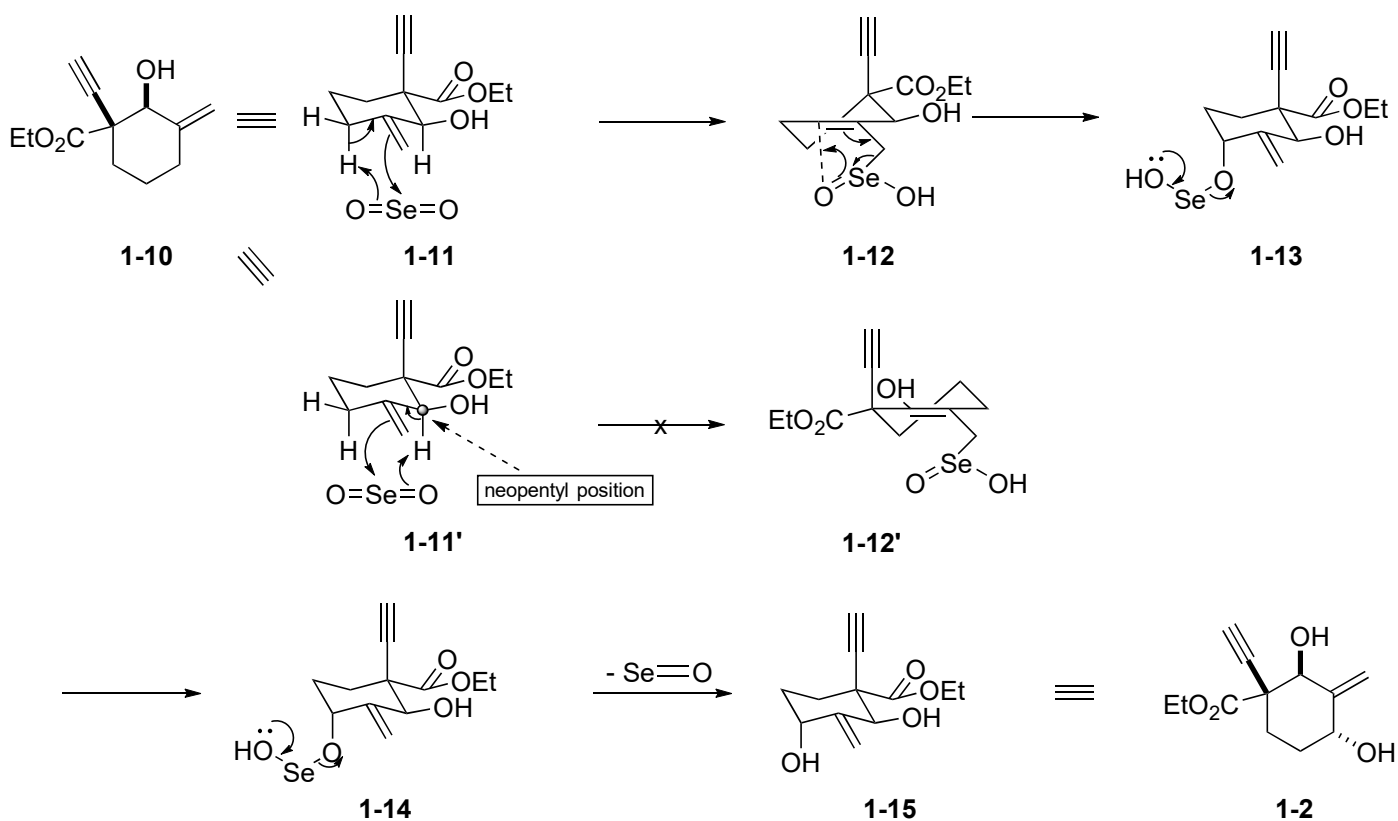


step 1-2: Luche reduction

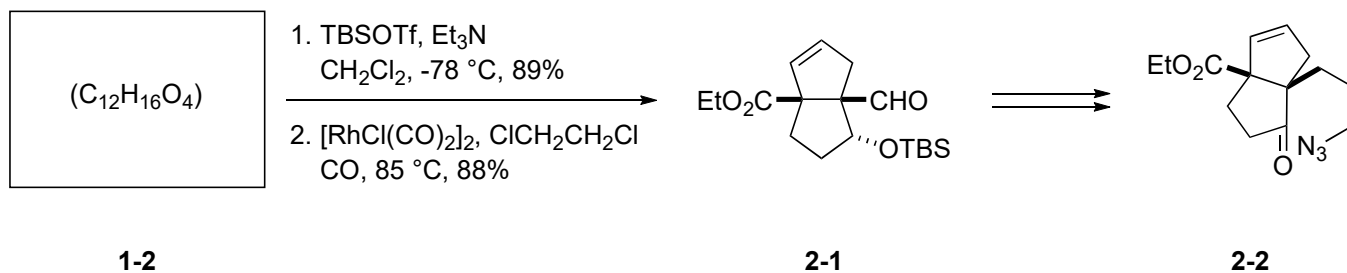
CeCl₃ is able to accelerate the exchanging of ligands, so hydride source of Luche reduction is NaH_nB(OMe)_{4-n}



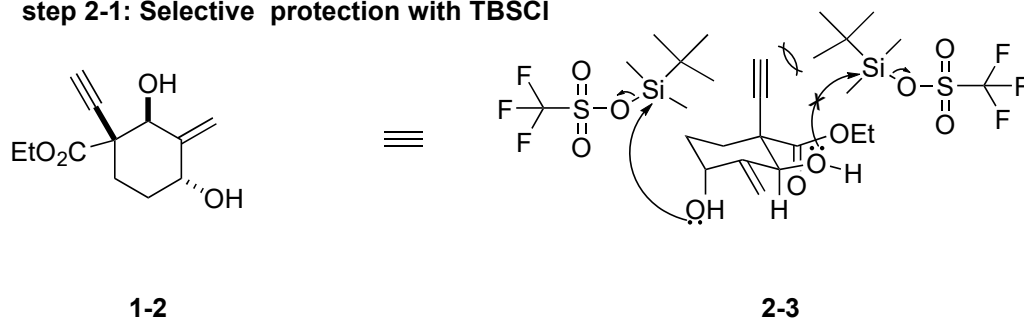
step 1-3: Oxidation with selenium dioxide



(2)

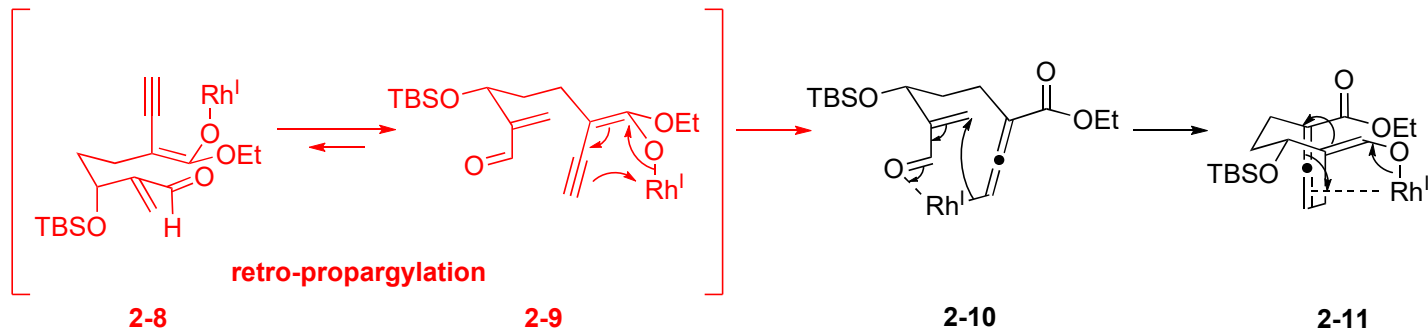
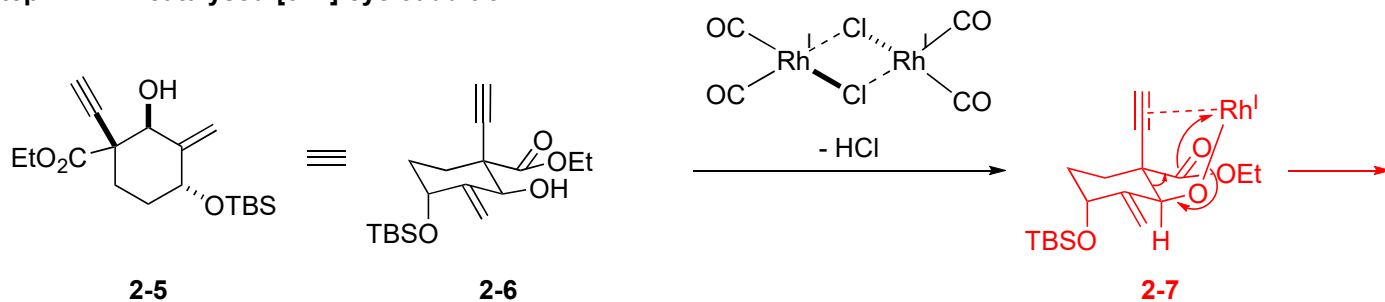


step 2-1: Selective protection with TBSCl

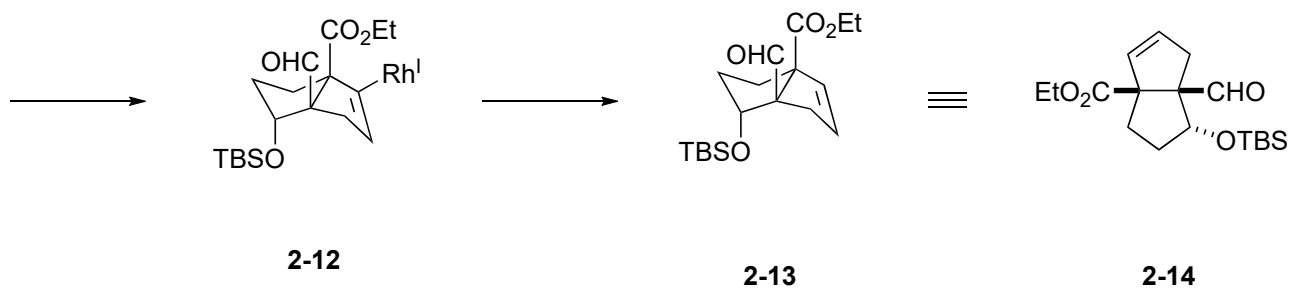




step 2-2: Rh-catalysed [3+2] cycloaddition

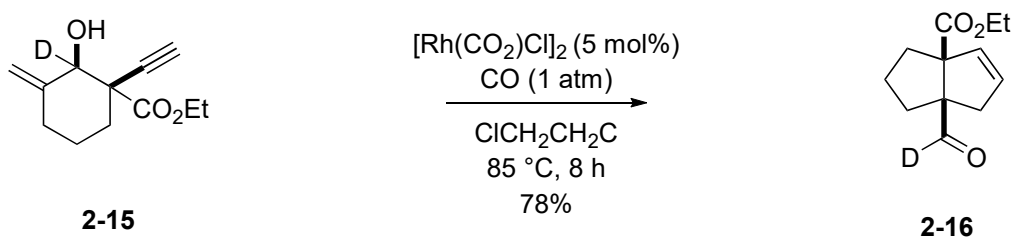


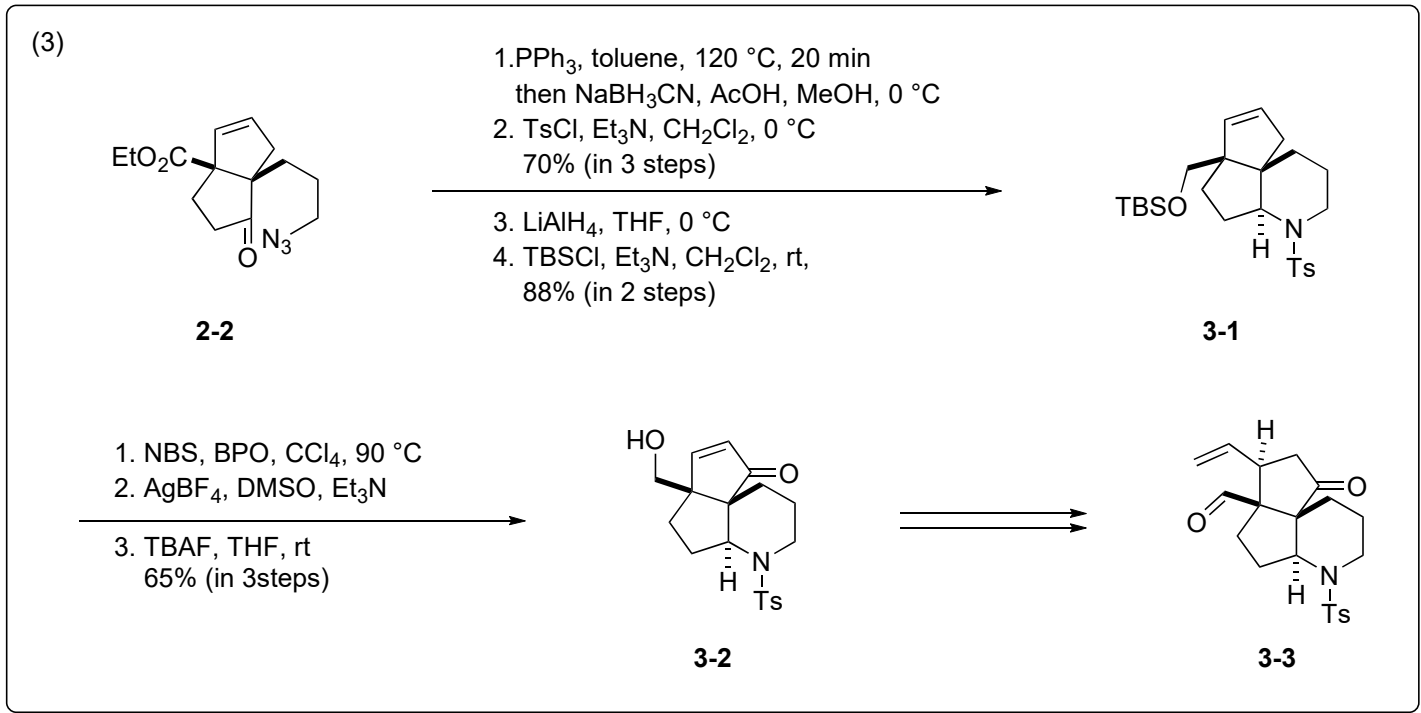
retro-propargylation should occur via rhodate (2-8) to give a reasonable stereochemical structure of Rh in 2-9



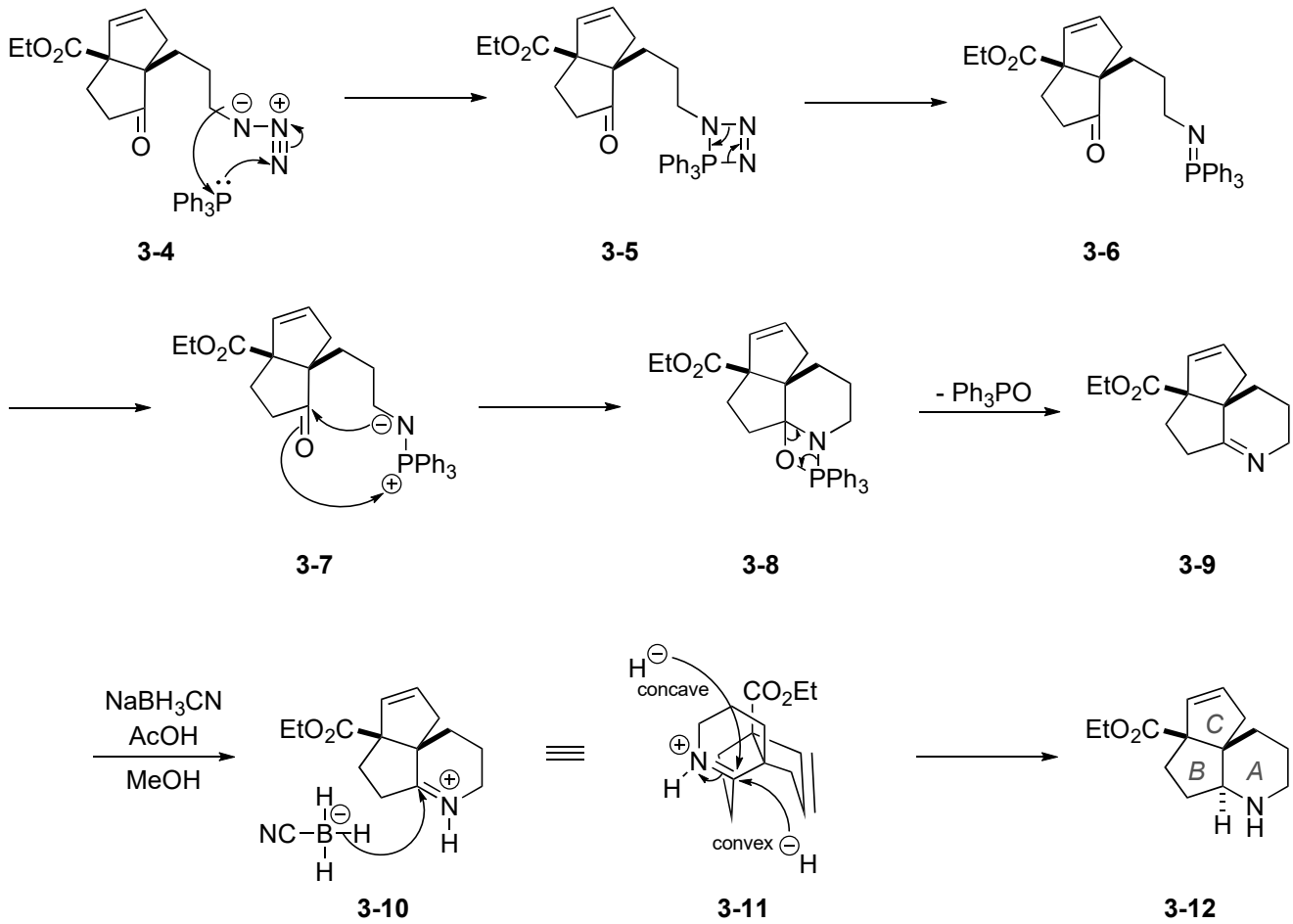
Evidence for retro-propargylation

Deuterium-labeling experiment (*Nat. Commun.* **2014**, *5*, 5707.)

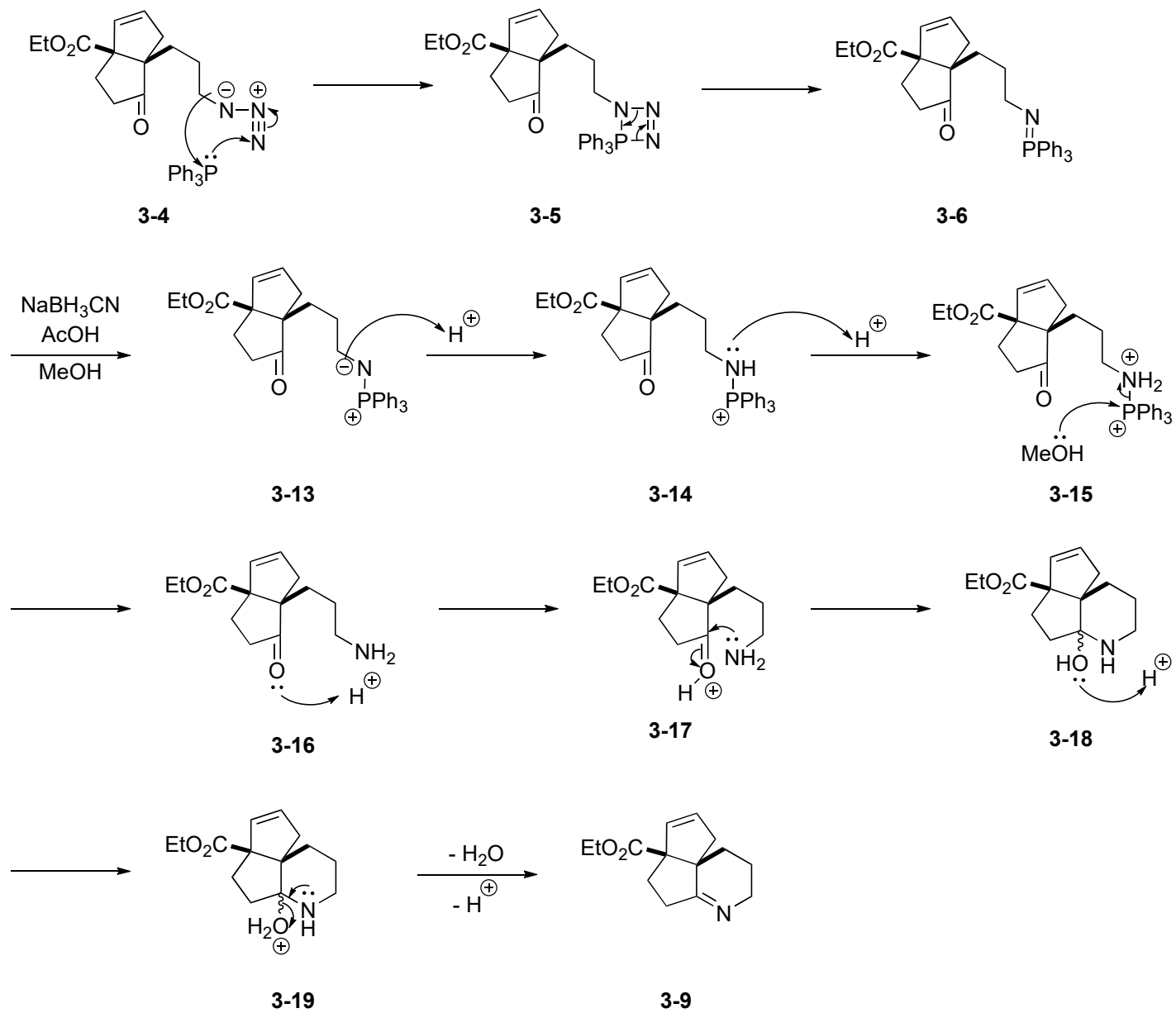




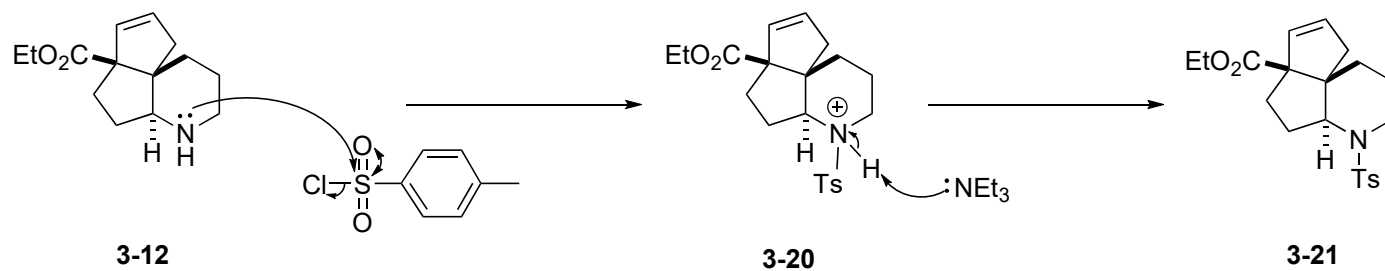
step 3-1-1: Aza-wittig Reaction



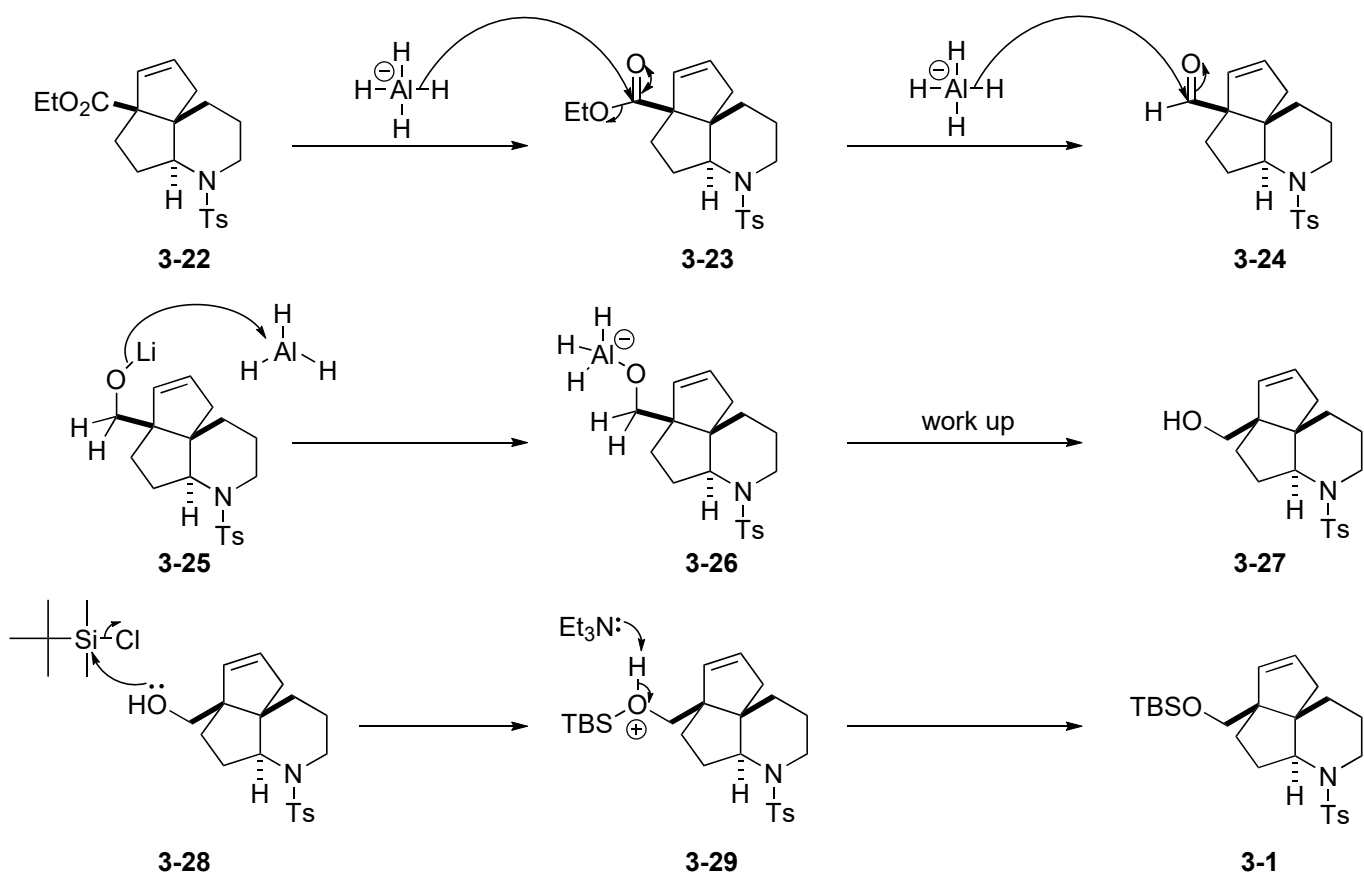
step 3-1-1': Staudinger Reaction and imine formation (from 3-4 to 3-9)



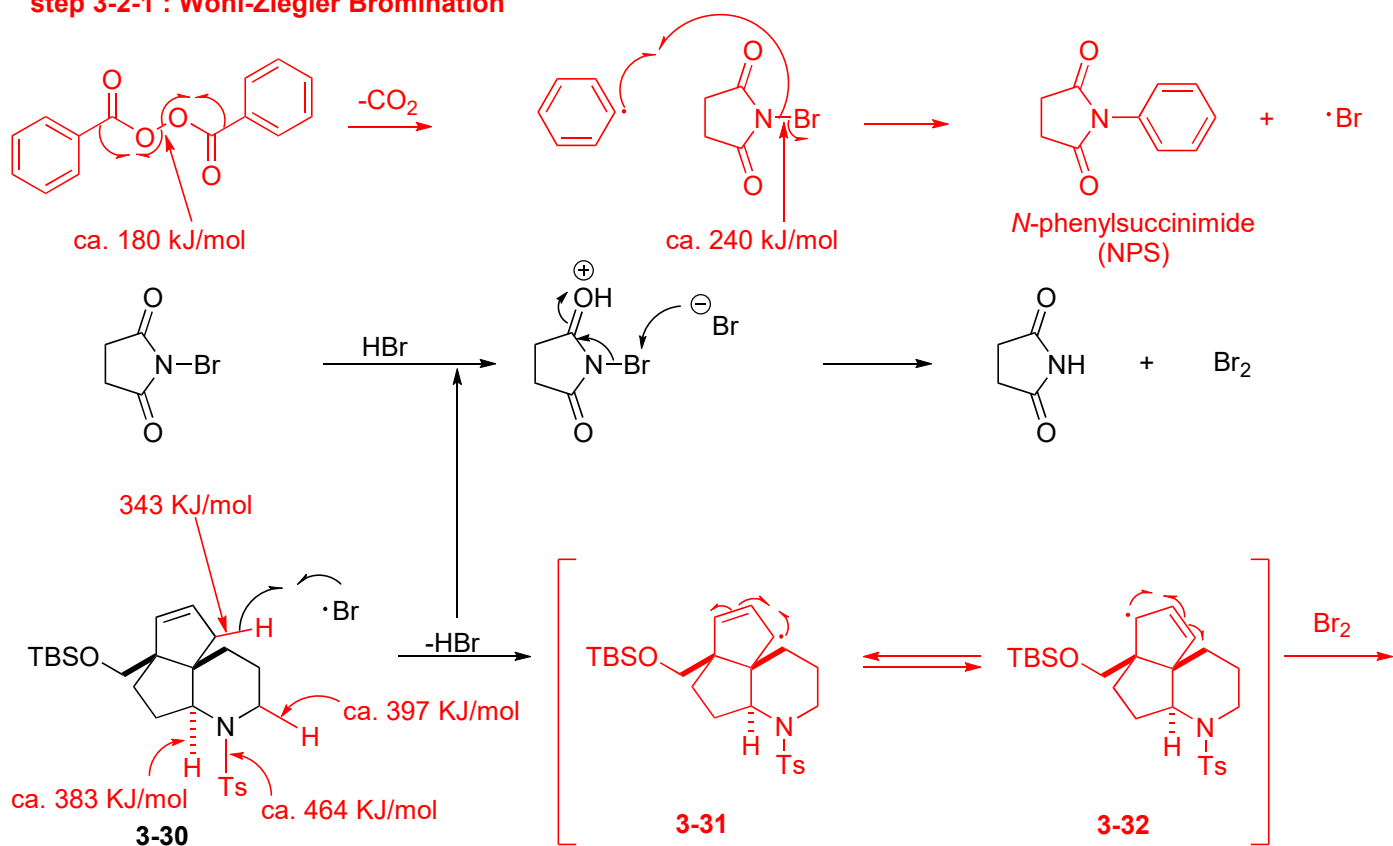
step 3-1-2 : Protection of amine with TsCl

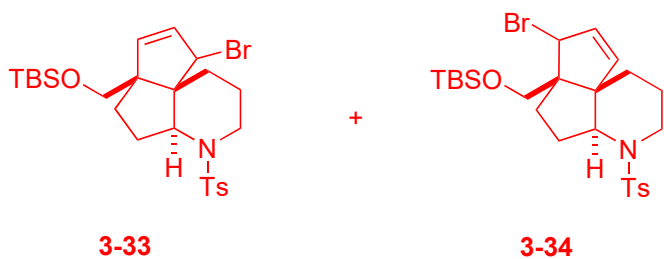


step 3-1-3, 4 : Reduction of ester with LiAlH₄ and Protection of corresponding alcohol with TBSCl



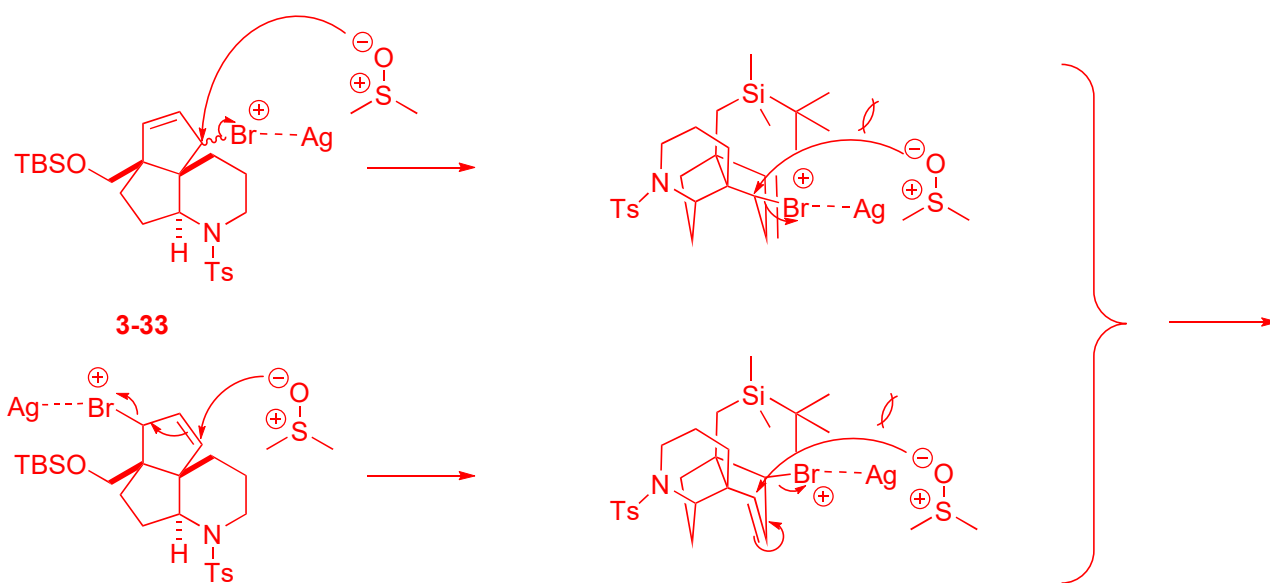
step 3-2-1 : Wohl-Ziegler Bromination



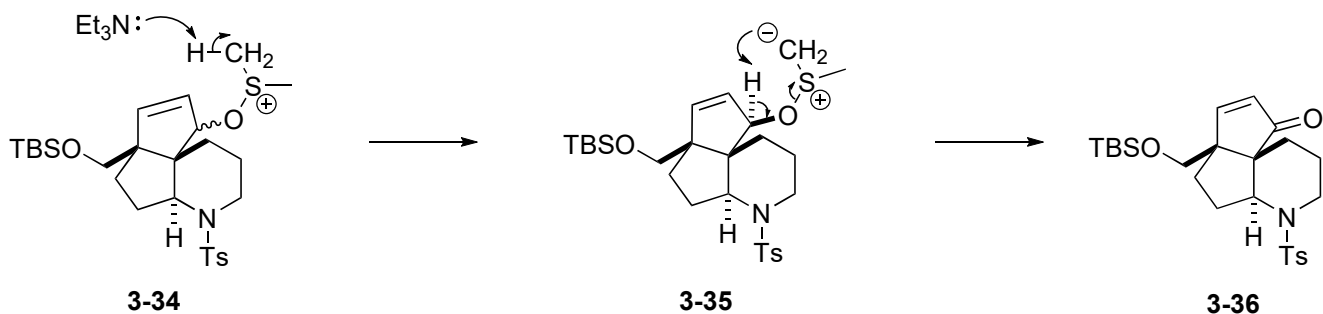


The product of Wohl-Ziegler Bromination was not been isolated, 3-33 and 3-34 are predicting product.

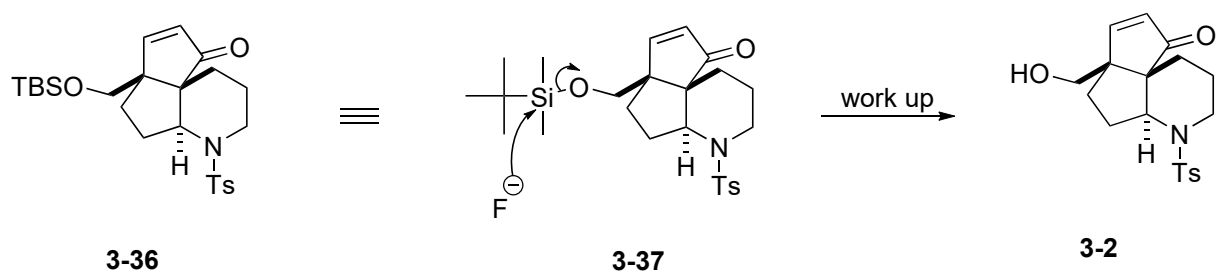
step 3-2-2 : Kornblum oxidation



3-34 Nucleophilic attack of DMSO to compound 3-33 and 3-34 can give identical compound 3-34 because of steric hindrance.



step 3-2-3 : deprotection of TBS with TBAF



4. Appendix

^{13}C -labeling experiment (González, D. F.; Brand, J. P. Waser, *J. Chem. Eur. J.* **2010**, *16*, 9457.)

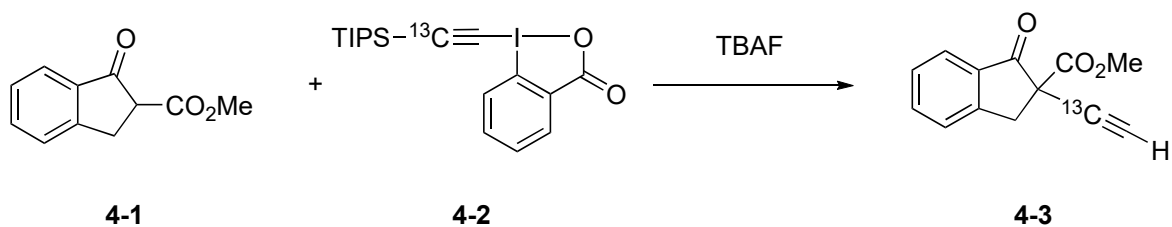
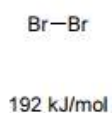
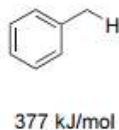
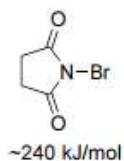
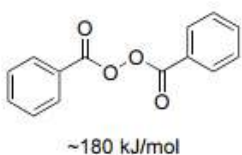


Table of Bond Dissociation Energies

1. T. L. Cottrell, *The Strengths of Chemical Bonds*, 2d ed., Butterworth, London, **1958**.
2. B. deB. Darwent, *National Standard Reference Data Series*, National Bureau of Standards, no. 31, Washington, **1970**.
3. S. W. Benson, *J. Chem. Educ.* **42**:502 (**1965**).
4. J. A. Kerr, *Chem. Rev.* **66**:465 (**1966**).

The mechanism of Wohl-Ziegler Bromination (160514_PS_Komei_Sakata_Ans_lingzhiol)

Bond dissociation energy



Reaction mechanisms

