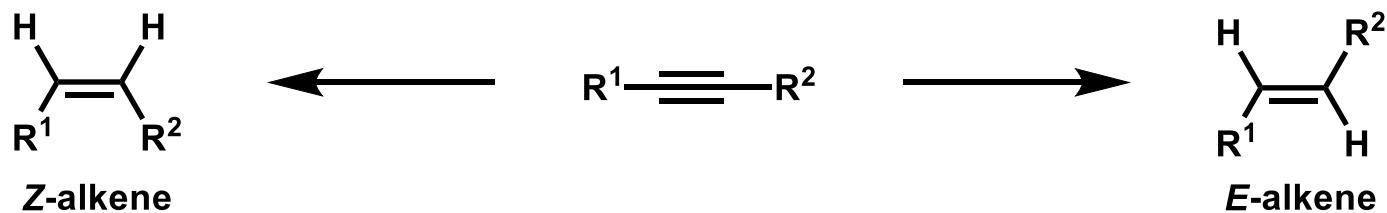


Stereo-selective Transfer Hydrogenation of Alkynes into Alkene



**LS 170715
M2 Shinsuke Shimizu**

Contents

1. Introduction of transfer hydrogenation (TH)

2. Ligand-Controlled TH of alkynes

3. TH of ynamides (main paper)

Contents

1. Introduction of transfer hydrogenation (TH)

2. Ligand-Controlled TH of alkynes

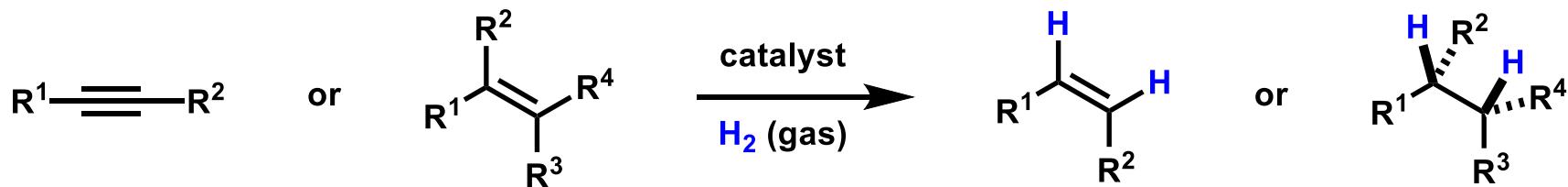
3. TH of ynamides (main paper)

Introduction of hydrogenation (1)

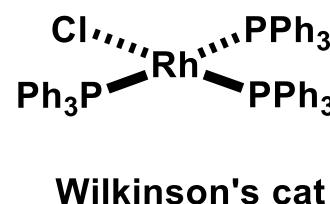
direct hydrogenation

transfer hydrogenation

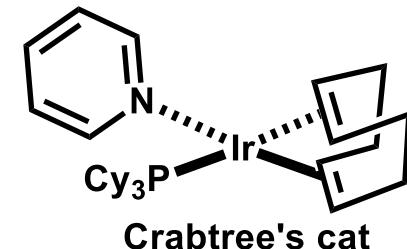
direct hydrogenation (with H₂ gas as hydrogen source)



catalyst... Pd/C
Pd/CaCO₃/Pb(OAc)₂ (Lindlar cat)
Pd/BaSO₄ (Rosenmundt reduction)
Pd/(OH)₂ (Pearlman cat.)
Wilkinson cat
Crabtree cat, etc...



Wilkinson's cat



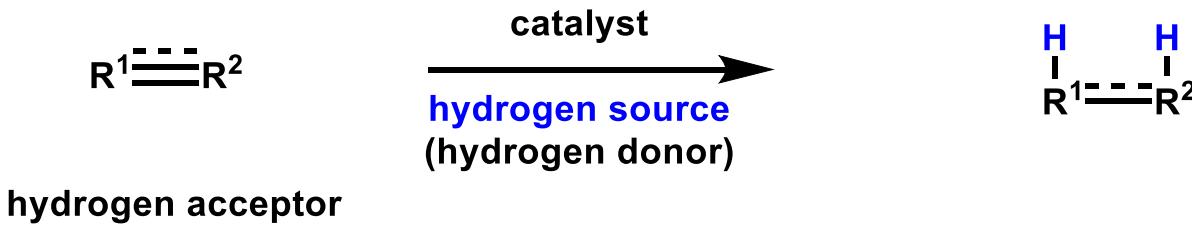
Crabtree's cat

Introduction of hydrogenation (2)

direct hydrogenation

transfer hydrogenation

transfer hydrogenation (non-H₂ hydrogen source)



catalyst...

Pd, Ni, Co, Ir, Au, etc...

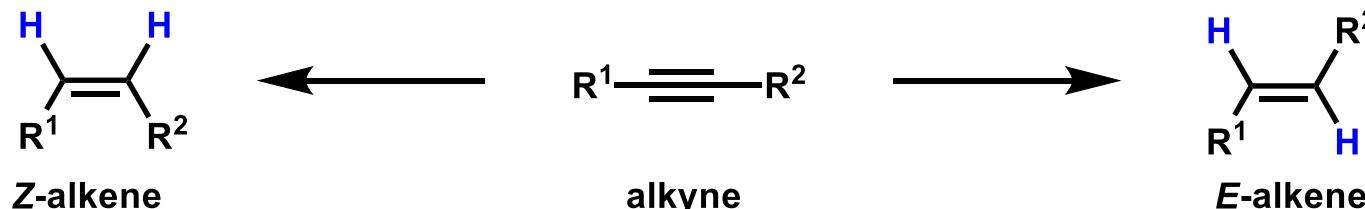
hydrogen source... HCO_2H , MeOH, EtOH, *i*-PrOH
 NH_3BH_3

convenient and powerful method

- Not require pressurized H_2 gas, nor elaborate experimental setups.
- **hydrogen source** are readily available, inexpensive, and easy to handle.

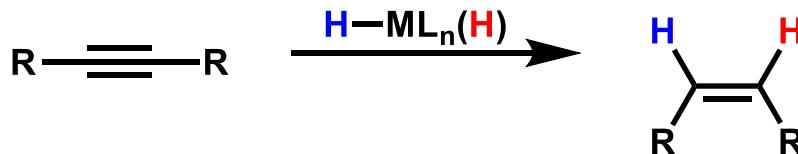
Stereoselective semi-reduction of alkyne

- semi-reduction



- Z-alkene

- Lindlar cat -> see also 121208_LS_Yuki_Nakagawa_Semihydrogeantion_with_Pd.BN
- *cis*-hydrometalation/reductive-elimination



- E-alkene

- Birch reduction -> traditional and powerful method,
but low functional-group tolerance because of the harsh condition.

-> remaining a major challenge

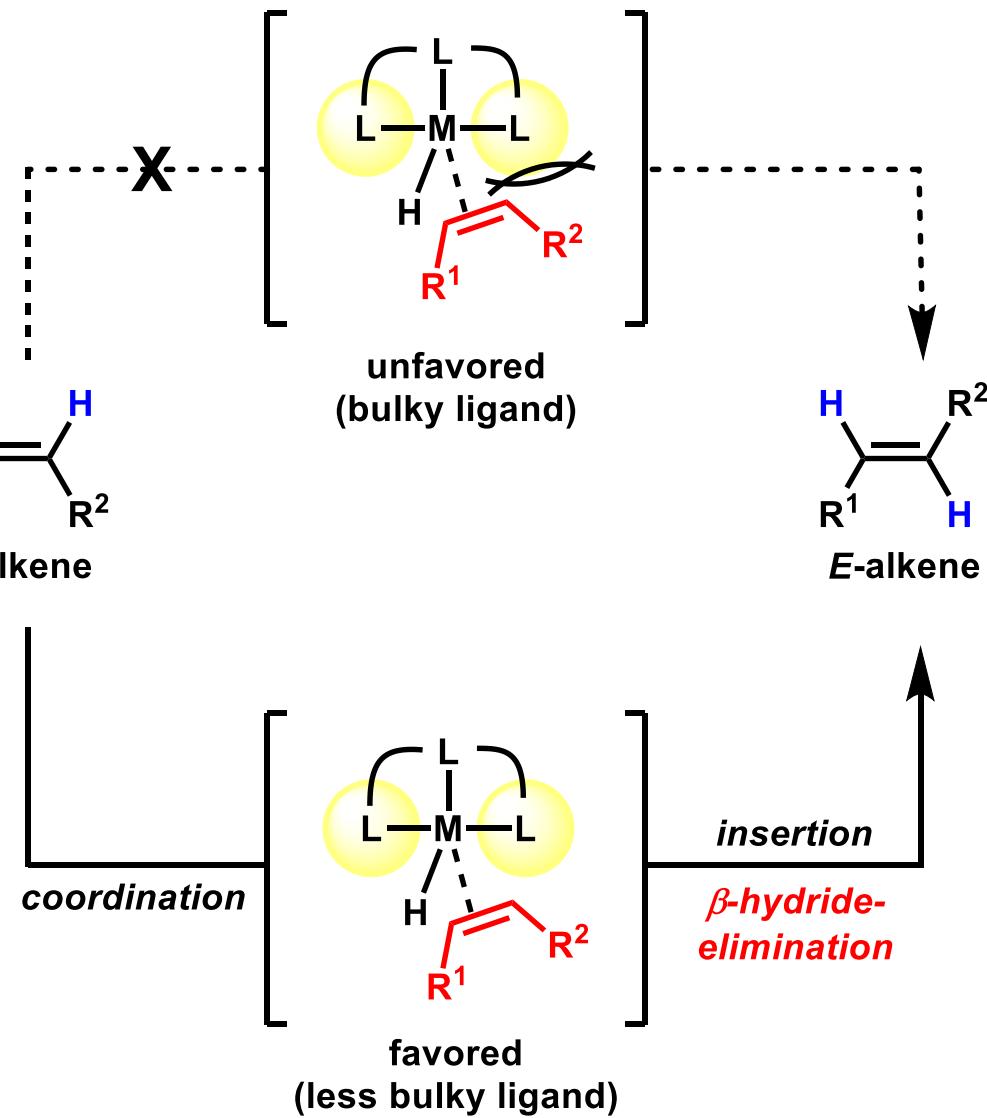
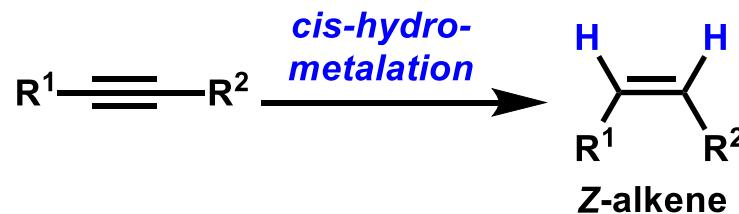
Contents

- 1. Introduction of transfer hydrogenation (TH)**
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Concept of this research

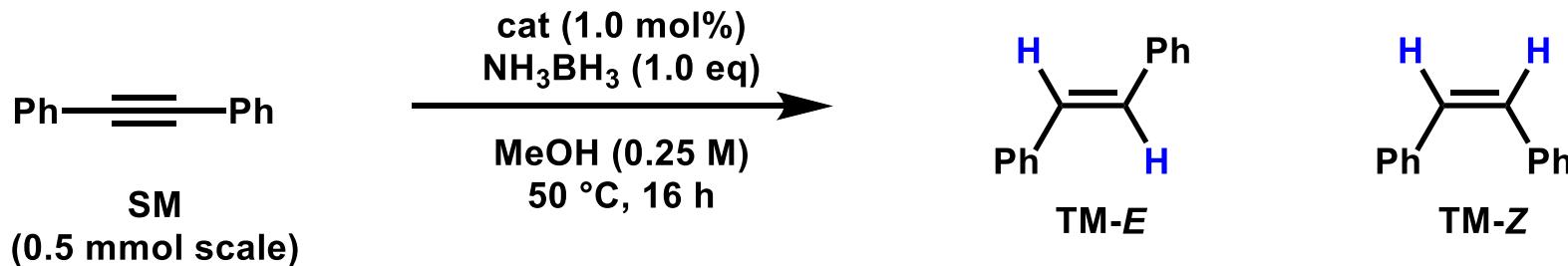
- E-alkene is possible to be generated via...

- (1) *cis-hydrometalation*
- (2) *β-hydride-elimination*



- the coordination and insertion of Z-alkene intermediate require a less steric hindered metal center.

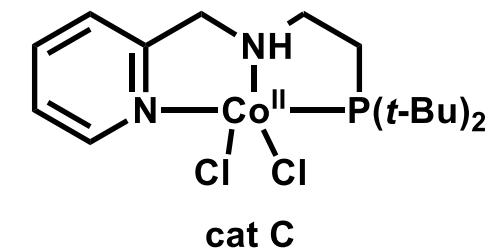
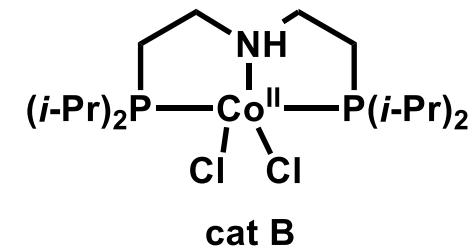
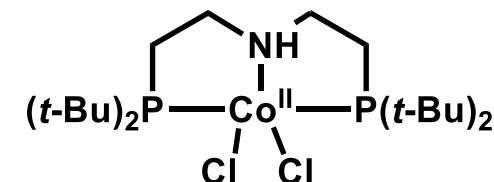
Co-Catalyzed TH of Alkyne



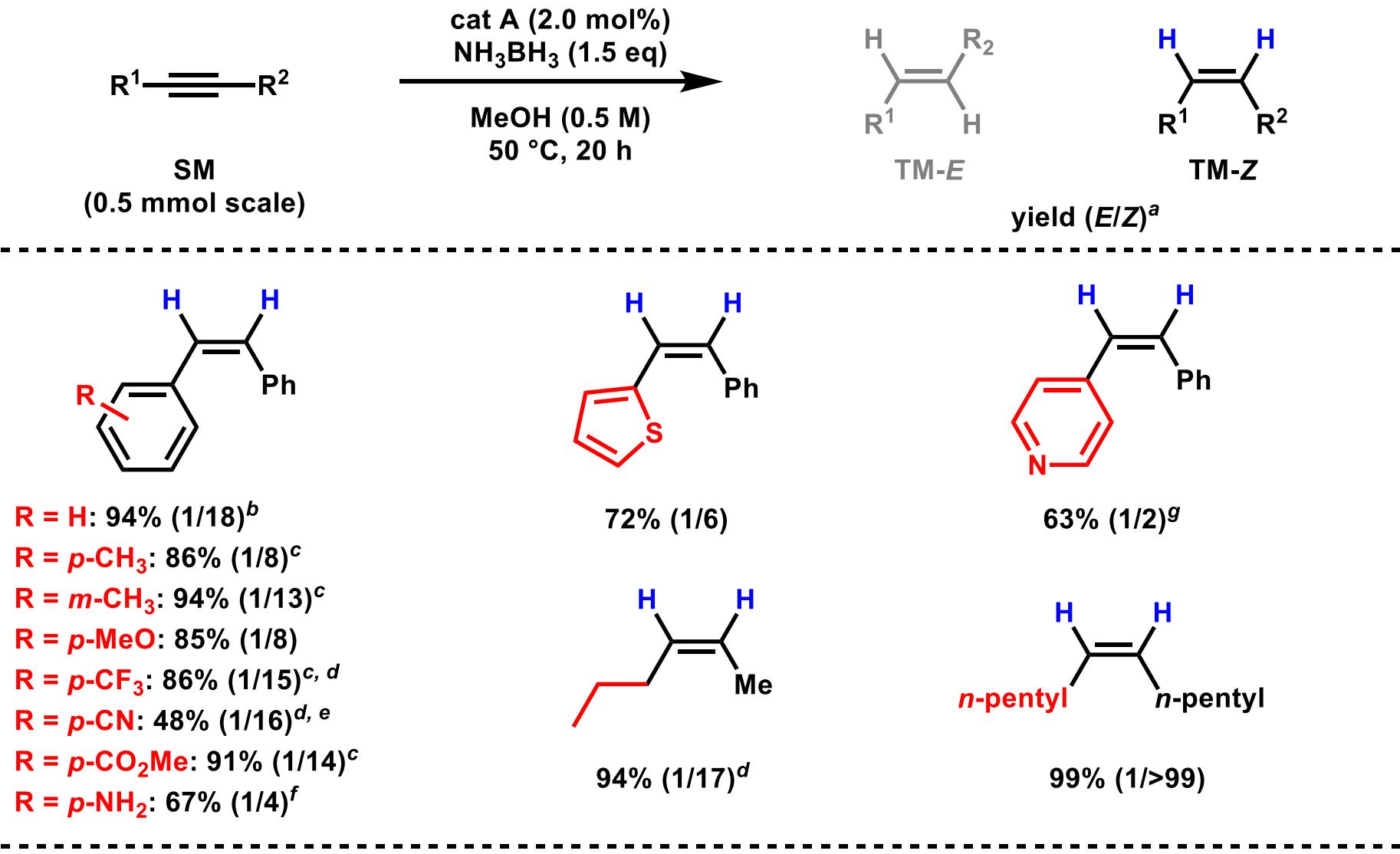
entry	catalyst	yield / % ^a		conv. SM / %
		TM-E	TM-Z	
1	none	ND ^b	ND	100
2	CoCl ₂	2	24	30
3	cat A	5	94	99
4	cat B	92	8	100
5	cat C	100	ND	100

a: determined by GC analysis using biphenyl as the internal standard.

b: ND means "not detected".

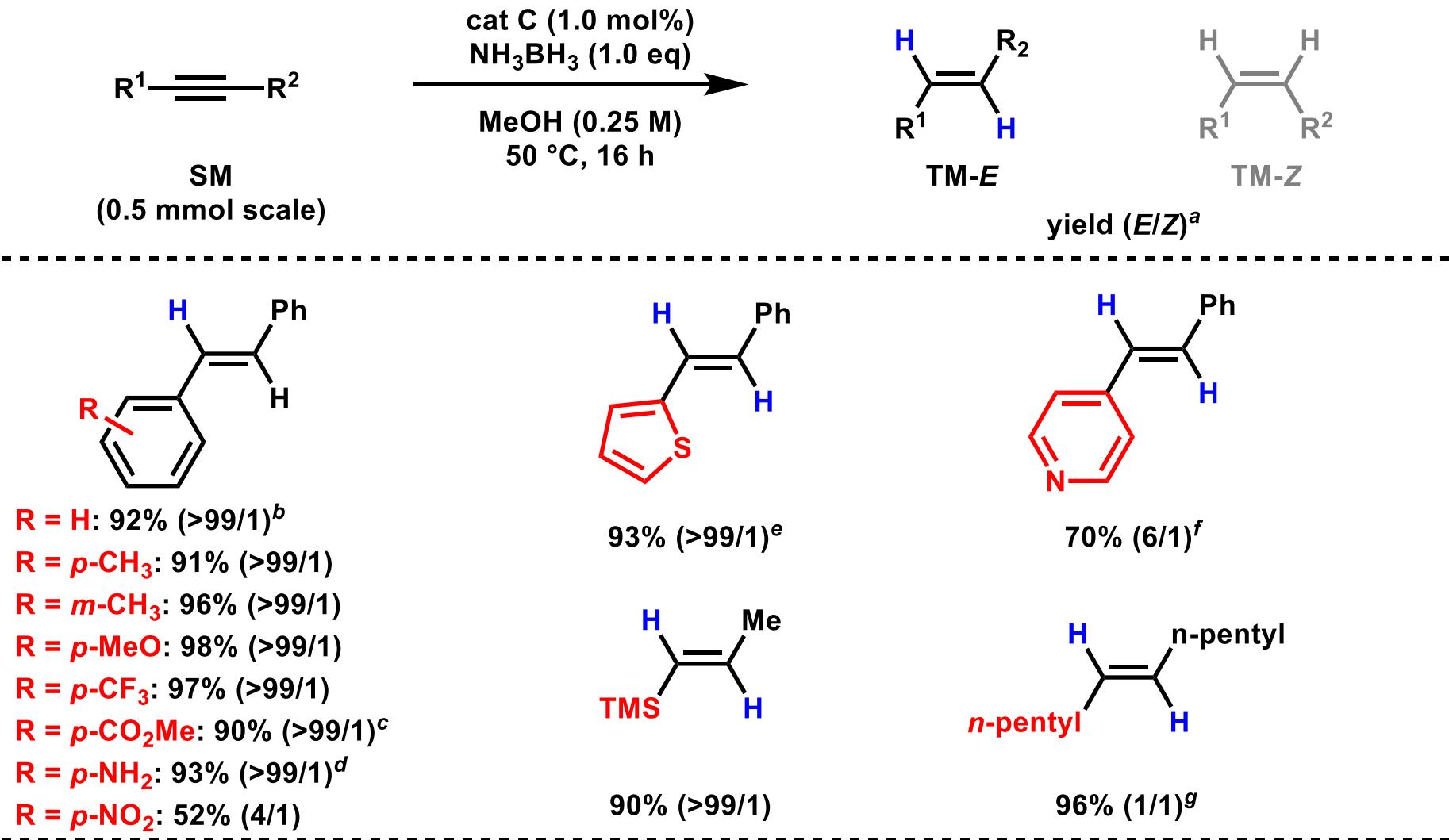


Scope of the Z-selective hydrogenation



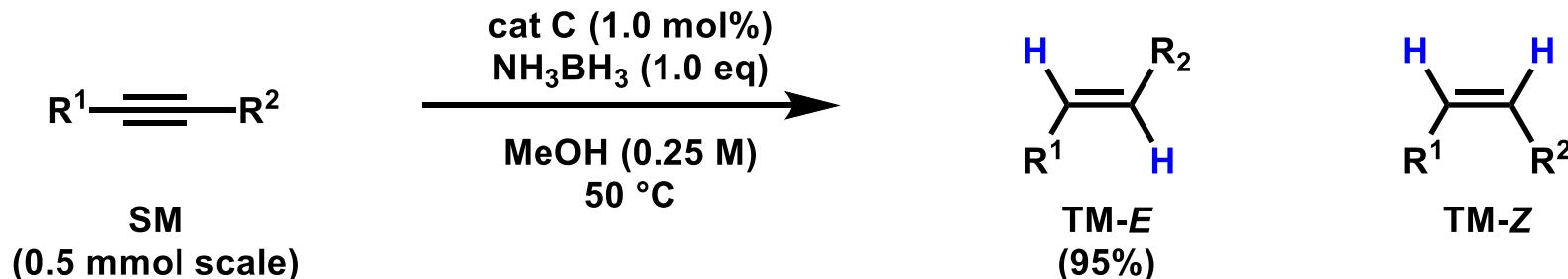
a: isolated yield. **b:** NH_3BH_3 (1.2 eq), cat A (1.0 mol%). **c:** 60 °C. **d:** GC yield. **e:** NH_3BH_3 (2.6 eq). **f:** NH_3BH_3 (2.4 eq), cat A (3.0 mol%). **g:** NH_3BH_3 (1.0 eq), cat A (1.5 mol%).

Scope of the *E*-selective hydrogenation



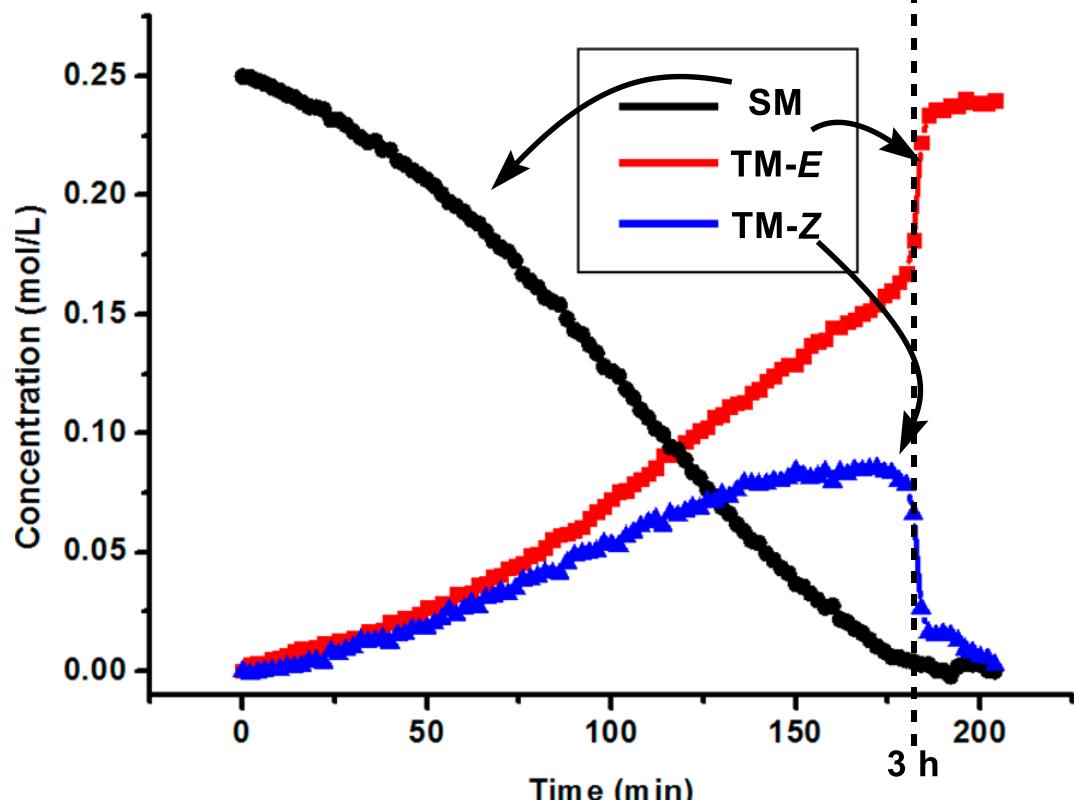
a: isolated yield. *b*: 5 mmol scale, cat C (0.2 mol%), MeOH (0.1 M). *c*: GC yield. *d*: NH₃BH₃ (0.8 eq), cat C (3.0 mol%), MeOH/THF (0.25 M, 3/1), 36 h. *e*: NH₃BH₃ (1.5 eq), cat C (2.0 mol%), 36 h. *f*: NH₃BH₃ (0.7 eq), 36 h. *g*: NH₃BH₃ (0.5 eq), cat B (4.0 mol%), MeOH (0.5 M), 40 h.

Time-course study of the *E*-selective semireduction



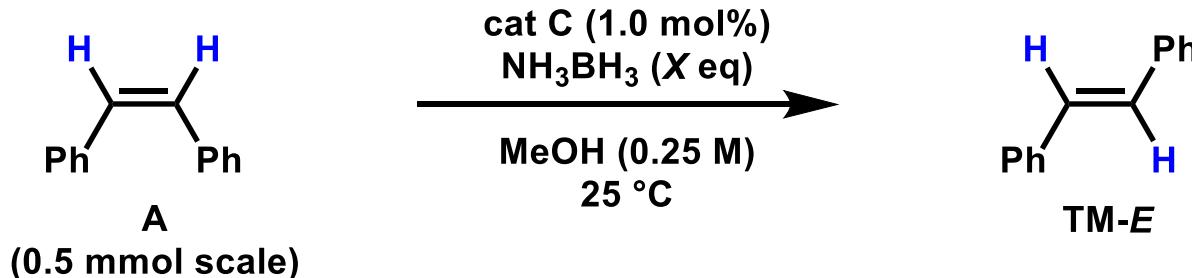
- Z-alkene intermediate (TM-Z) was generated in the first 3 h.
- TM-Z was further converted to the *E*-alkene (TM-E) via a isomerization.
- Decrease in the concentration of SM led to a dramatic increase in the rate of the isomerization step (Time: 3 h).

isomerization process was inhibited by the SM?



The course of the reaction could be recorded from the characteristic IR absorption of SM, TM-*E* and TM-Z.

Co-catalyzed isomerization of Z-alkene

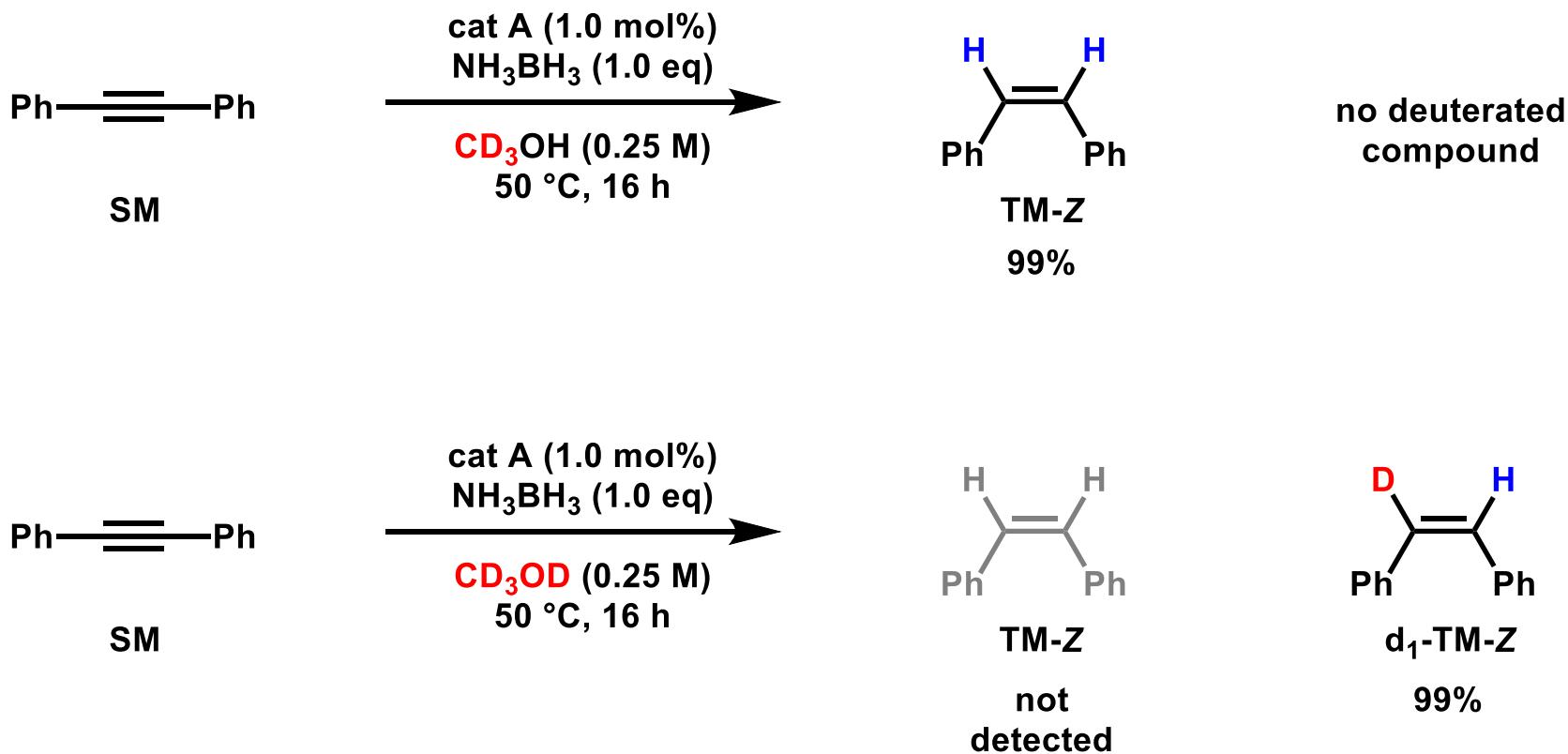


entry	time	NH ₃ BH ₃ (eq)	conv. A ^a (%)	yield TM- <i>E</i> ^a (%)	
1	1 h	1.0	100	100	
2	1 h	0.1	100	100	
3	16 h	0	0	ND ^c	
4 ^b	16 h	1.0	—	8	

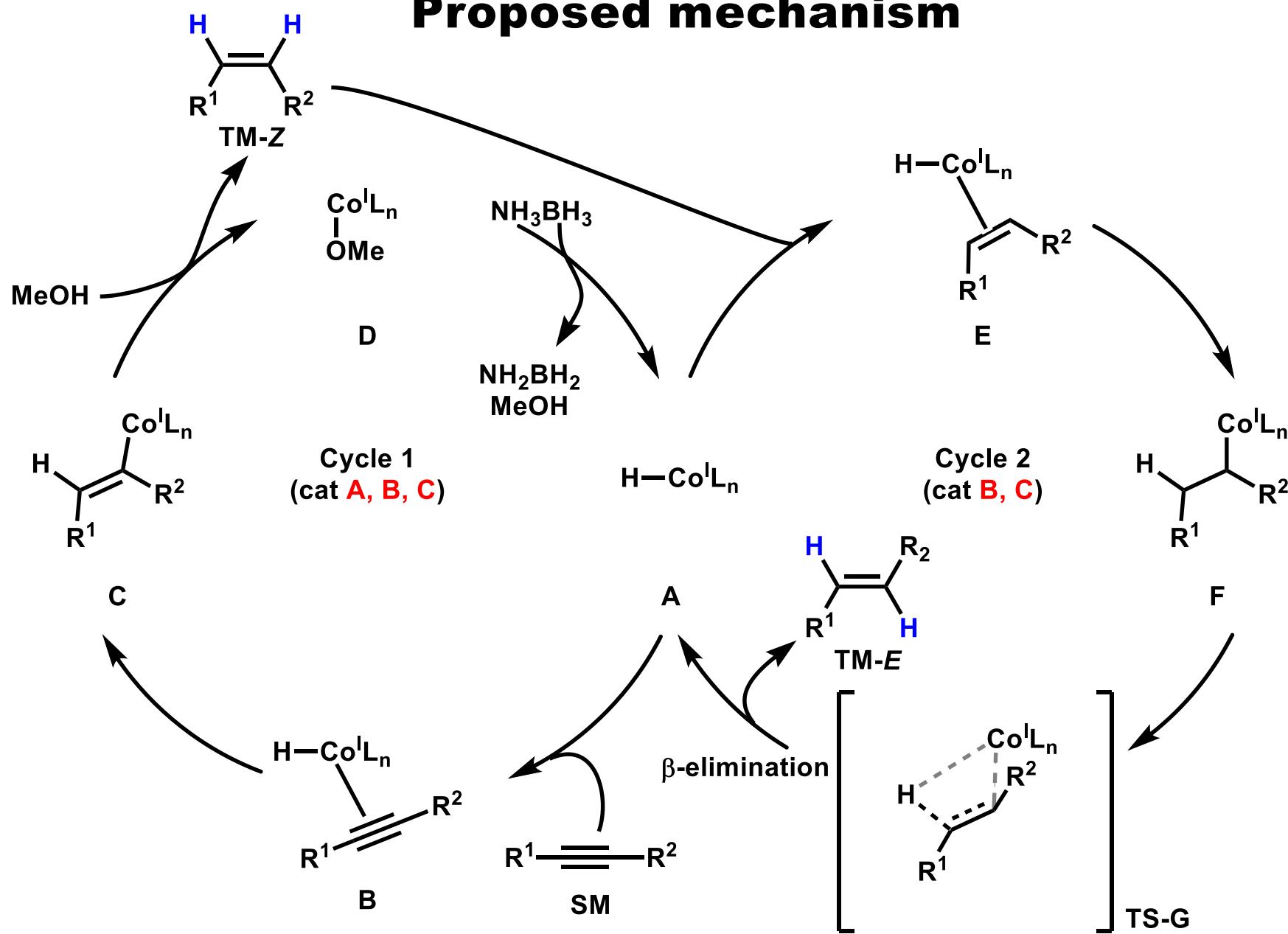
a: GC yields. *b*: **B** (1.0 eq) was added. *c*: ND means "not detected".



Deuterium-labeling experiments

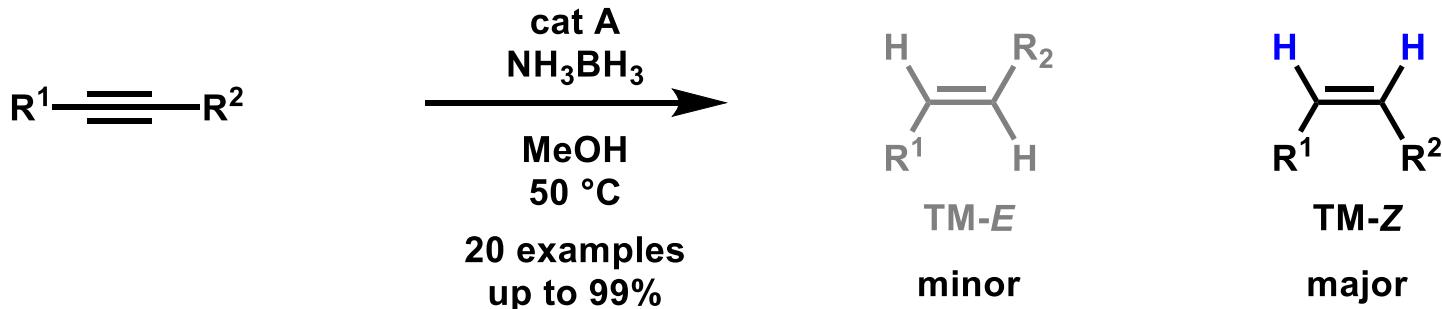


Proposed mechanism

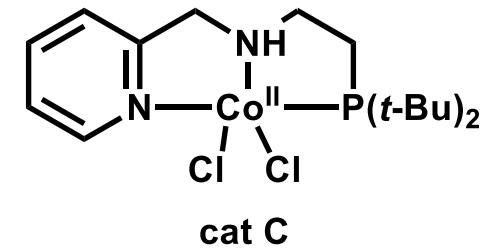
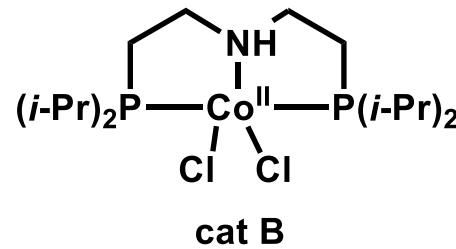
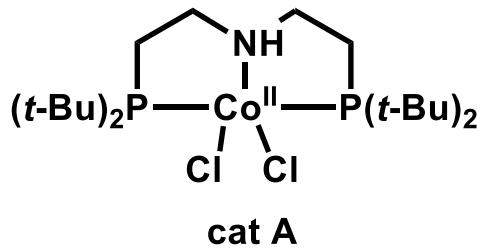
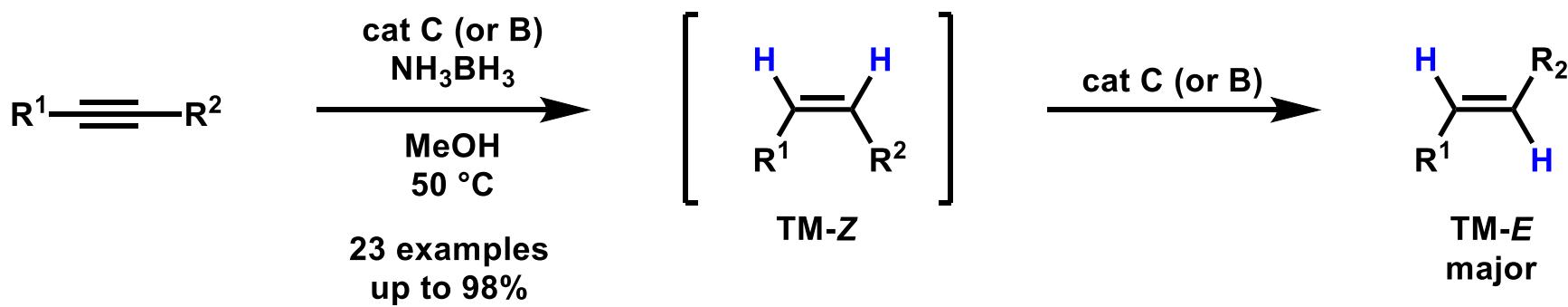


Short summary

- Z-selective semi-hydrogenation



- **E-selective semi-hydrogenation**



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K. C. Kumara Swamy



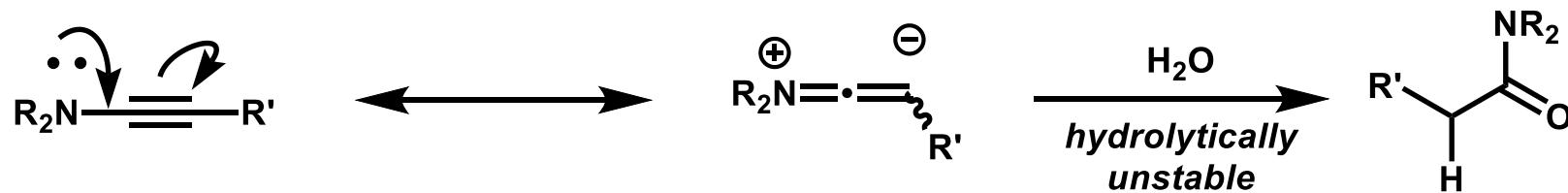
- 1957 Born in Kalasa (India)
- 19XX Ph.D from Indian Institute of Science (Bangalore) with Prof. S. S. Krishnamurthy
- 1986-89 P.D. at University of Massachusetts (Amherst, USA) with Prof. Robert R. Holmes.
- now Professor at the University of Hyderabad, School of Chemistry.

Current research interests

- (i) Chiral phosphorus compounds- Utility as chiral ligands for transition metals
- (ii) Phosphonates in Organic Synthesis
- (iii) Unusual Phosphorus Compounds
- (iv) Allene chemistry and palladium catalysis

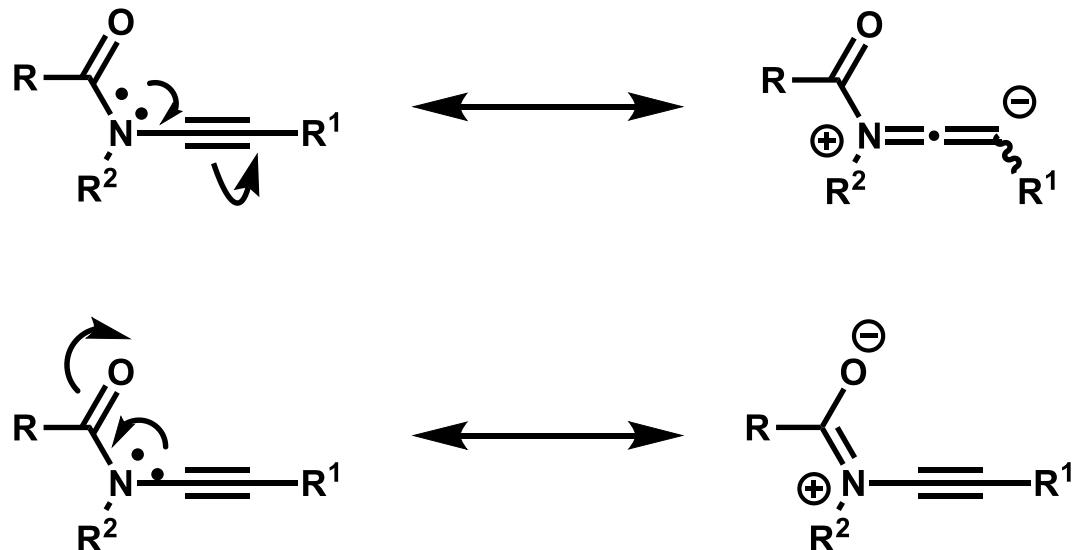
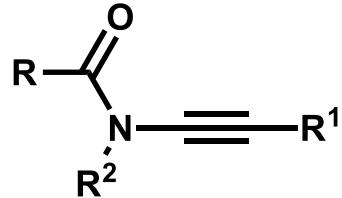
Background of ynamides (1)

- ynamine (1958~)



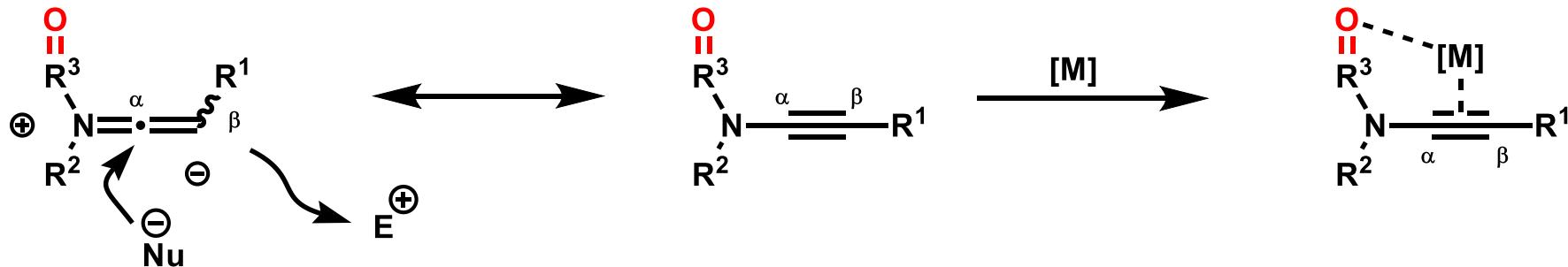
difficult in the experimental preparation and general handling of ynamine.

- ynamide (1972~)

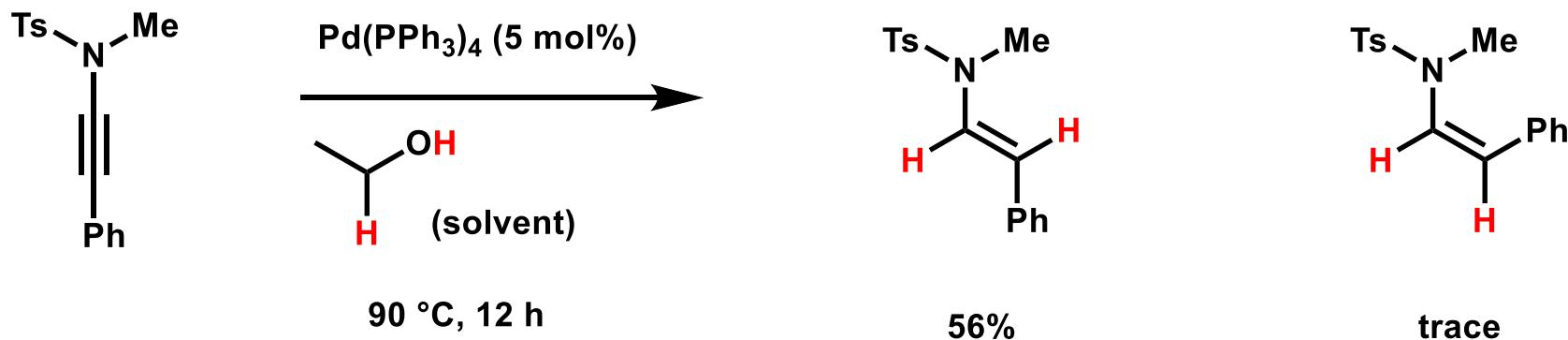


EWG
• stabilization
• directing group
• chiral auxiliary

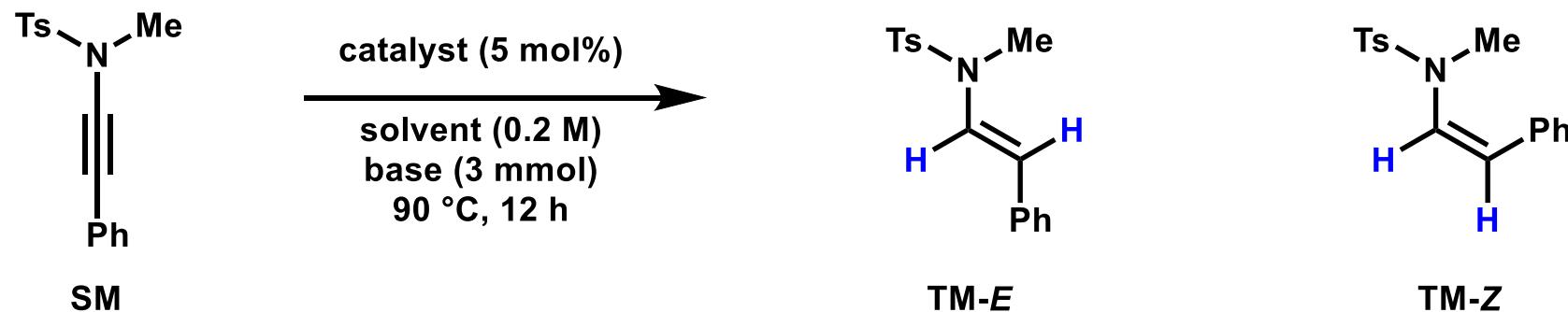
Background of ynamides (2)



-> see 120707_Answer_Tomomi_GOTO_Ynamide



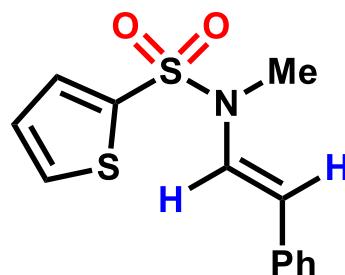
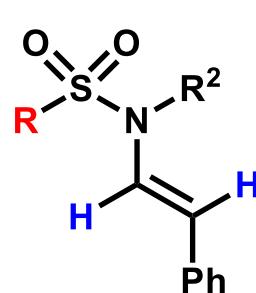
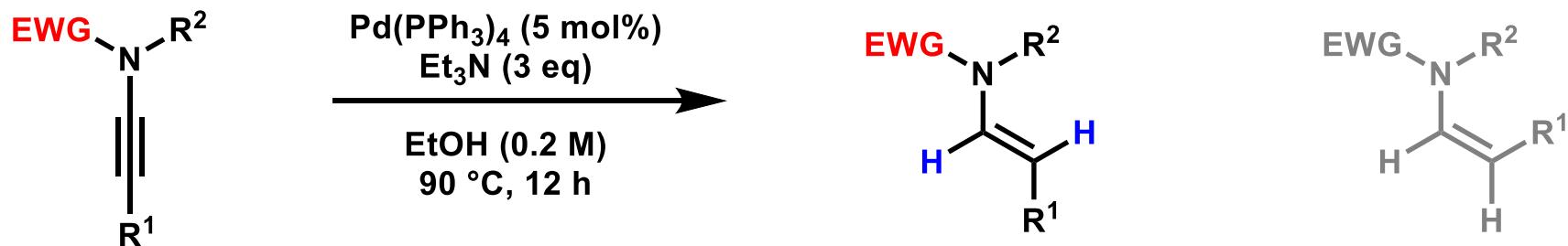
Pd-catalyzed hydrogenation of ynamide with ethanol



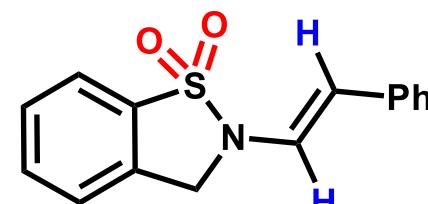
entry	catalyst	solvent	base	ratio			yield ^a
				TM-E	TM-Z	SM	
1	Pd(PPh ₃) ₄	EtOH	—	60	0	40	56%
2	Pd(PPh ₃) ₄	tol (or THF, DMSO)	—		no reaction		0
3	—	EtOH	—		no reaction		0
4	Pd(PPh ₃) ₄	<i>i</i> -PrOH	—	36	0	64	33%
5	Pd(PPh ₃) ₄	<i>n</i> -PrOH	—	54	0	46	52%
6	Pd(PPh ₃) ₄	<i>t</i> -BuOH	—		no reaction		0
7	Pd(PPh ₃) ₄	EtOH	NaOH	42	0	58	36%
8	Pd(PPh ₃) ₄	EtOH	Et ₃ N	100	0	0	92%

a: isolated yield.

Substrate Scope (1)



93%



96%

R = Tol: 92% ($R^2 = \text{Me}$)

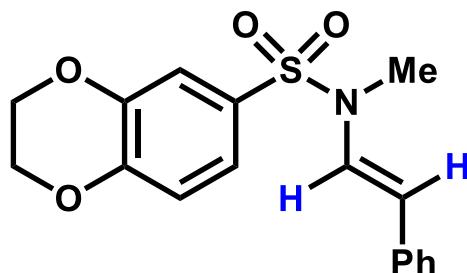
R = (p\text{-MeO})\text{Ph}: 94% ($R^2 = \text{Me}$)

R = 4-biphenyl: 86% ($R^2 = \text{Me}$)

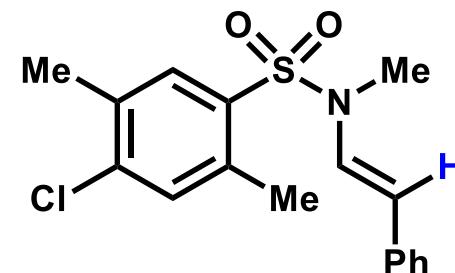
R = naphthyl: 90% ($R^2 = \text{Me}$)

R = Tol: 88% ($R^2 = \text{Bn}$)

R = Tol: 86% ($R^2 = \text{Ph}$)

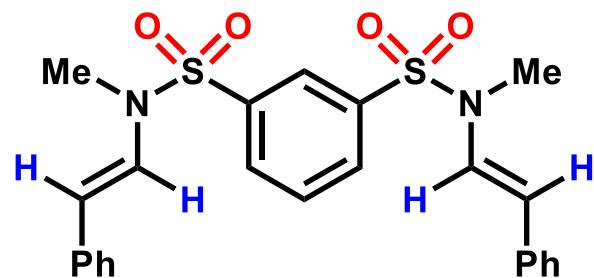
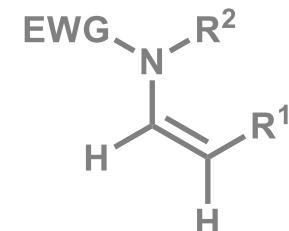
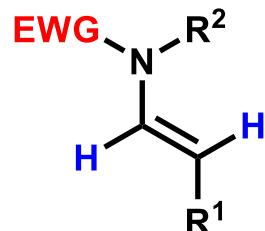
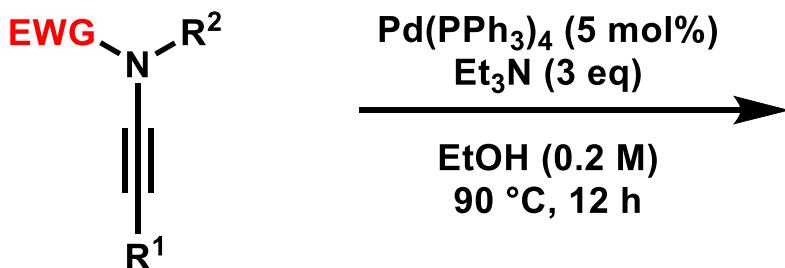


82%

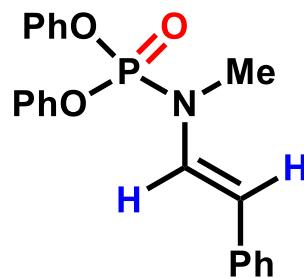


76%

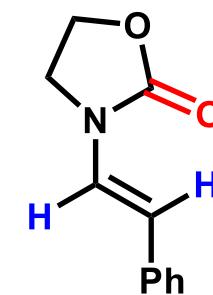
Substrate Scope (2)



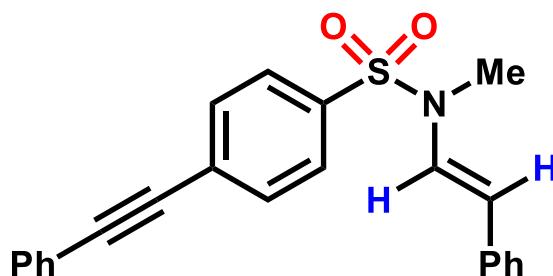
82%



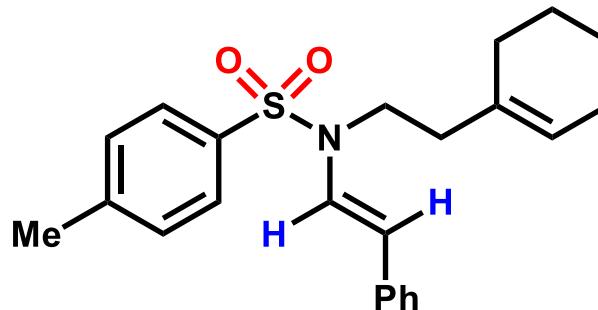
95%



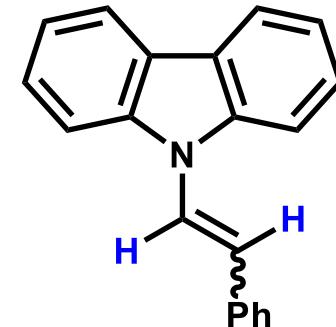
74% (*E/Z*: 6/1)



80%

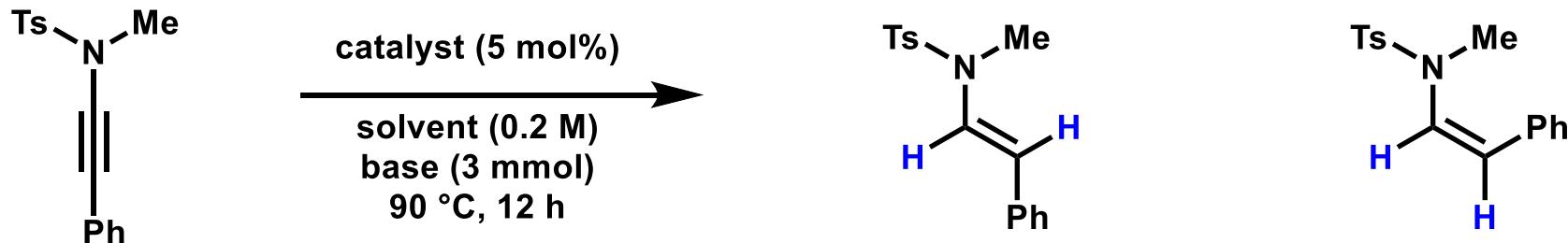


87%



96% (*E/Z*: 5/1)

Optimization of Pd catalyst



SM

TM-E

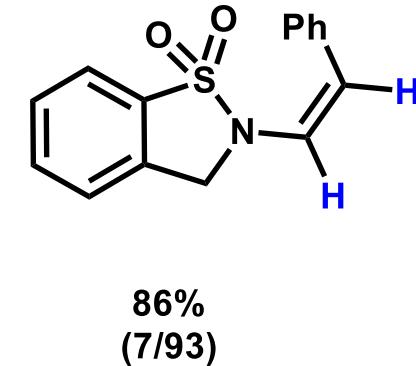
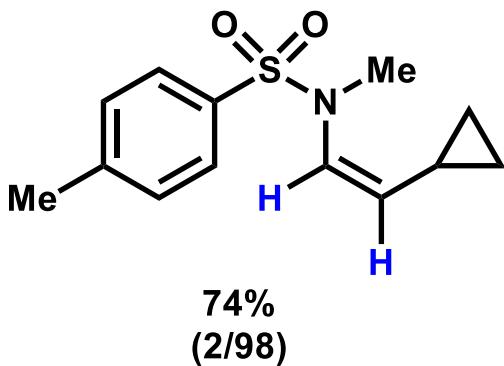
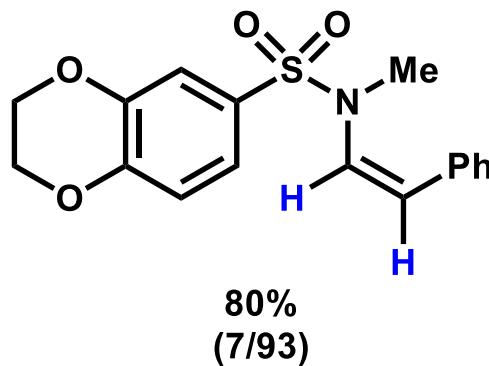
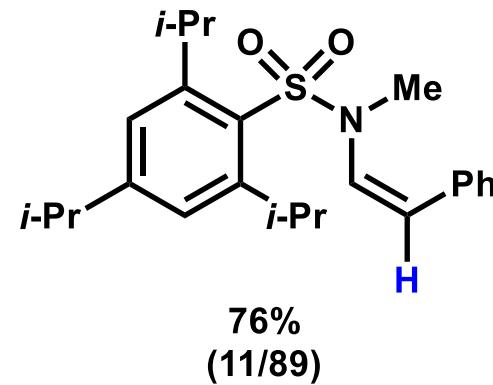
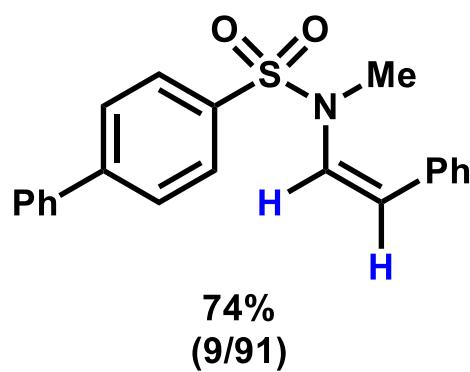
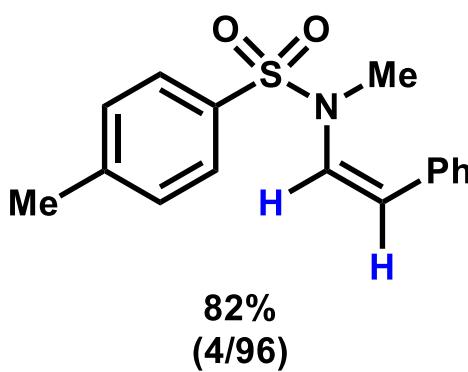
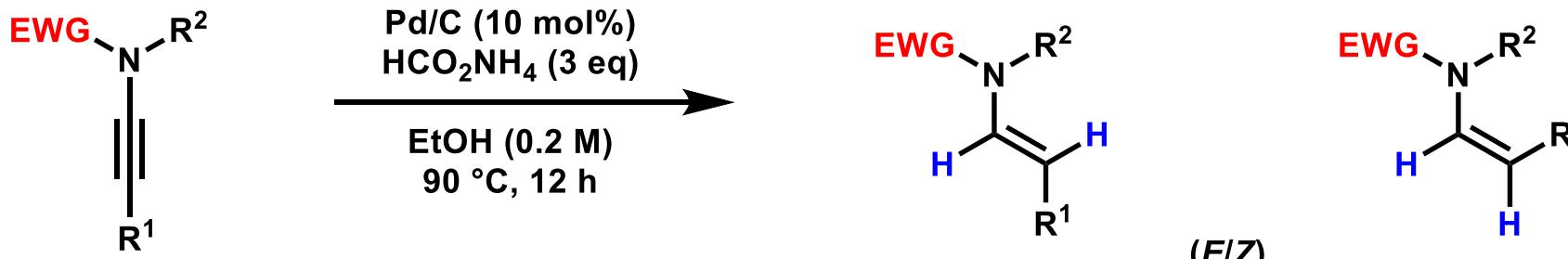
TM-Z

entry	catalyst	solvent	base	ratio (yield)			yield of TM-E ^a
				TM-E	TM-Z	SM	
1	PdCl ₂ (PPh ₃) ₂	EtOH	Et ₃ N	96	4	0	90% ^b
2	Pd(OAc) ₂ (PPh ₃) ₂	EtOH	Et ₃ N	90	10	0	84% ^b
3	Pd/C	EtOH	Et ₃ N	0	6	94	0
4	Pd ₂ (dba) ₃	EtOH	Et ₃ N	0	20	80	0
5 ^c	Pd(PPh ₃) ₄	EtOH	Et ₃ N	no reaction			0
6 ^d	Pd(PPh ₃) ₄	EtOH	Et ₃ N	76	0	24	72%
7	Pd(PPh ₃) ₄	EtOH	Et ₃ N	100	0	0	92

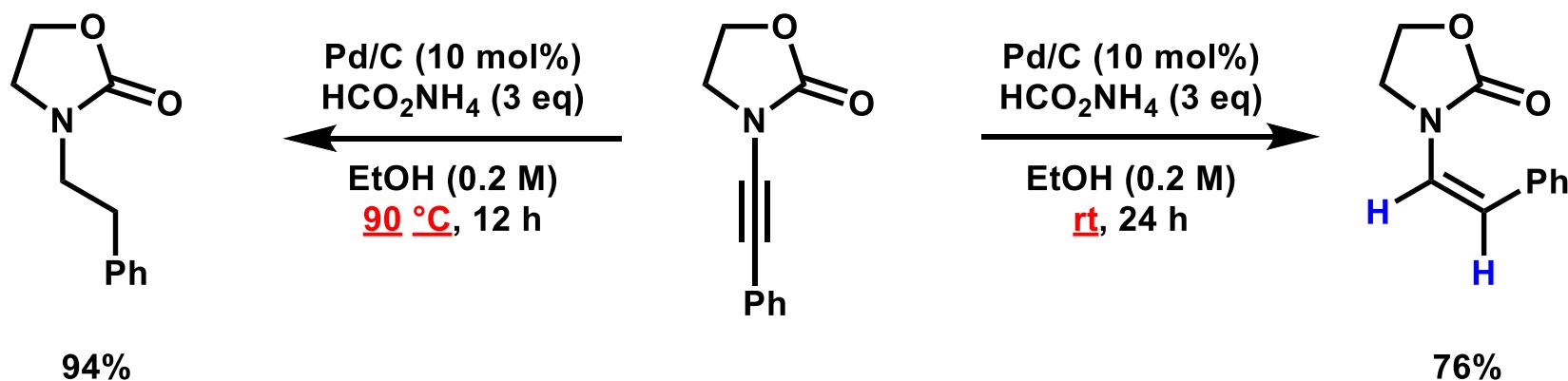
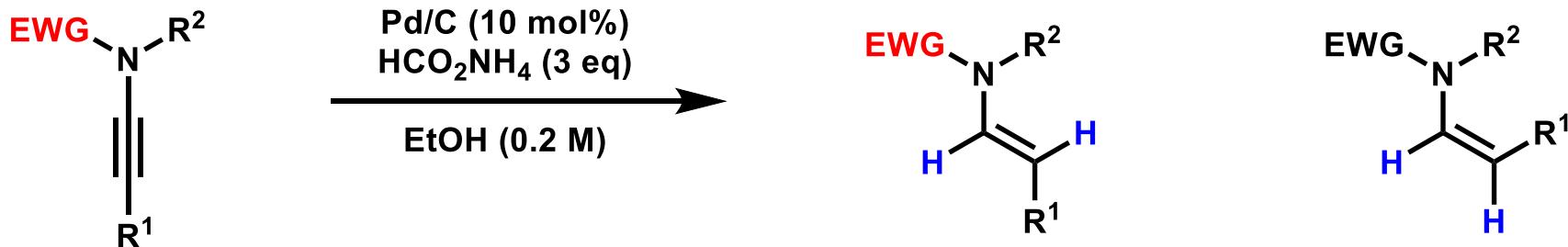
a: isolated yield. *b:* total (TM-E + TM-Z) yield. *c:* reaction performed at room temperature.

d: 3 mol% [Pd]-catalyst used.

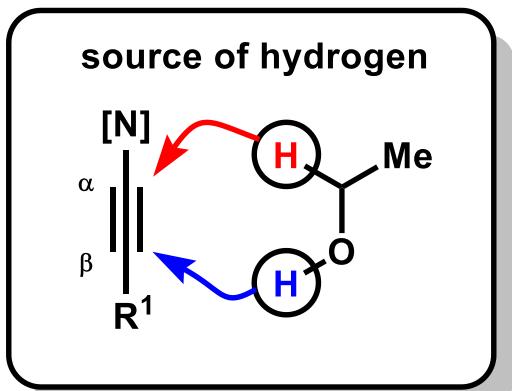
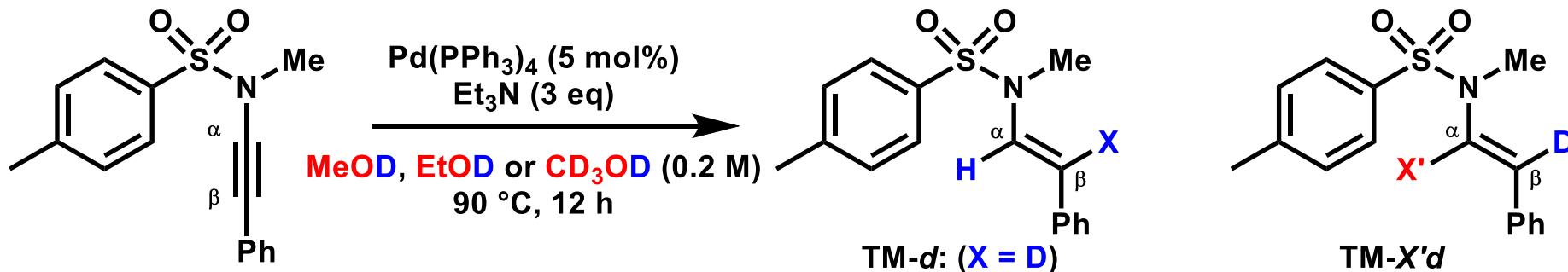
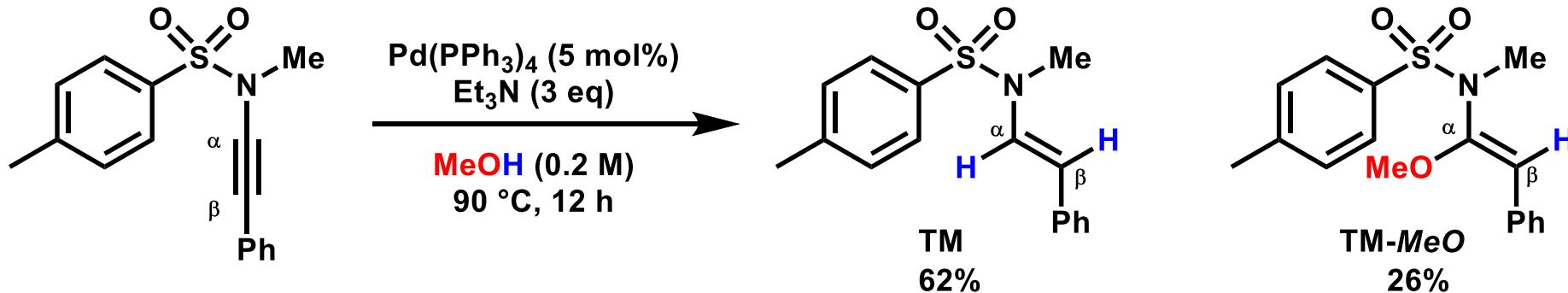
Scope of the Z-selective hydrogenation (1)



Scope of the Z-selective hydrogenation (2)



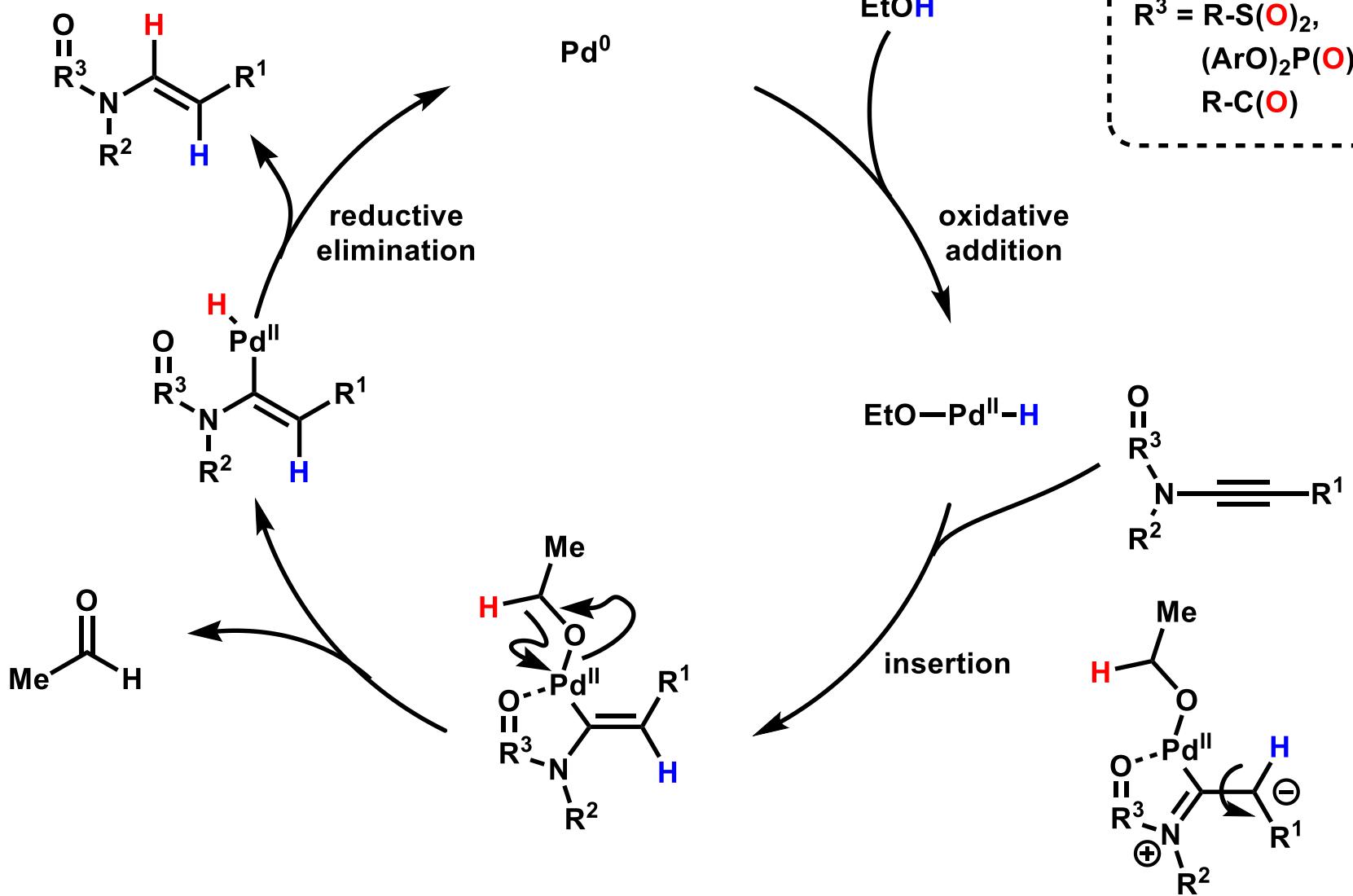
Control experiments



MeOD:	TM: 0% TM-d: 56%	detected (not isolated) $X = \text{OMe}$
EtOD ^a :	90% (TM/TM-d = 1/ ≥ 1)	NOT detected $X = \text{OEt}$
CD ₃ OD	no reaction	NOT detected $X = \text{OCD}_3$

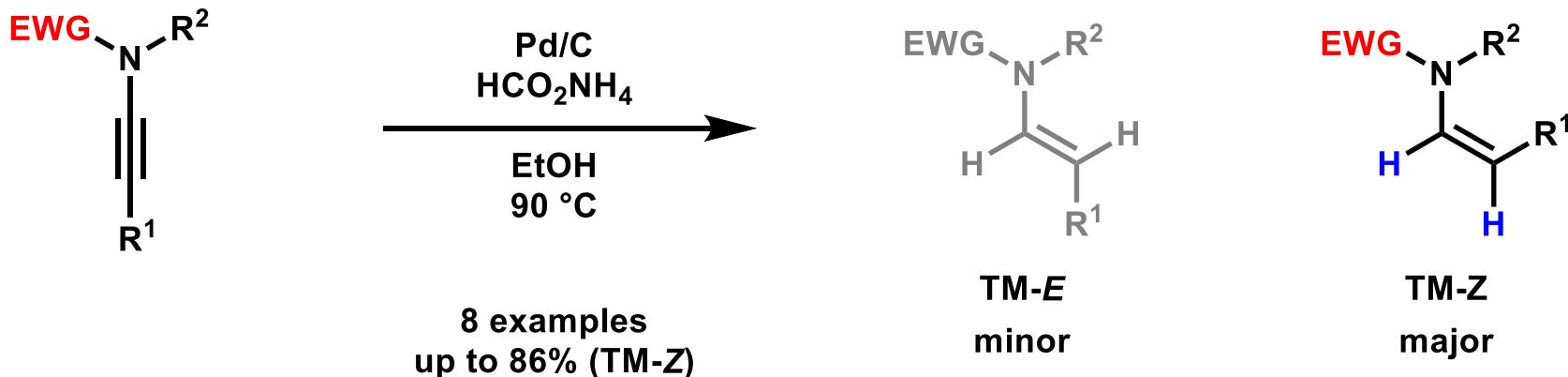
^a: ca 60% deuterated EtOD was used
(prepared by hydrolysis of $\text{Si}(\text{OEt})_4$ with D_2O).

Proposed mechanism

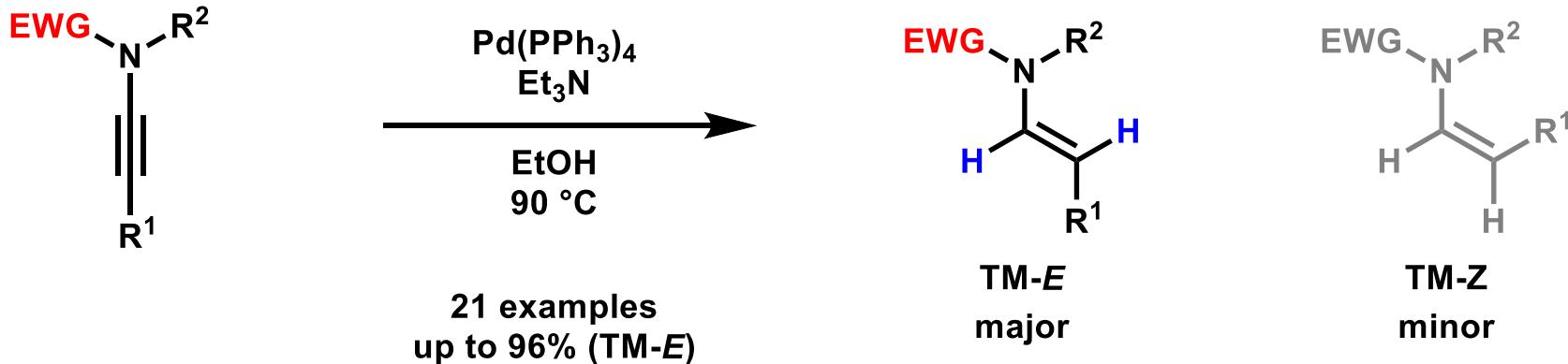


Summary

- Z-selective semi-hydrogenation

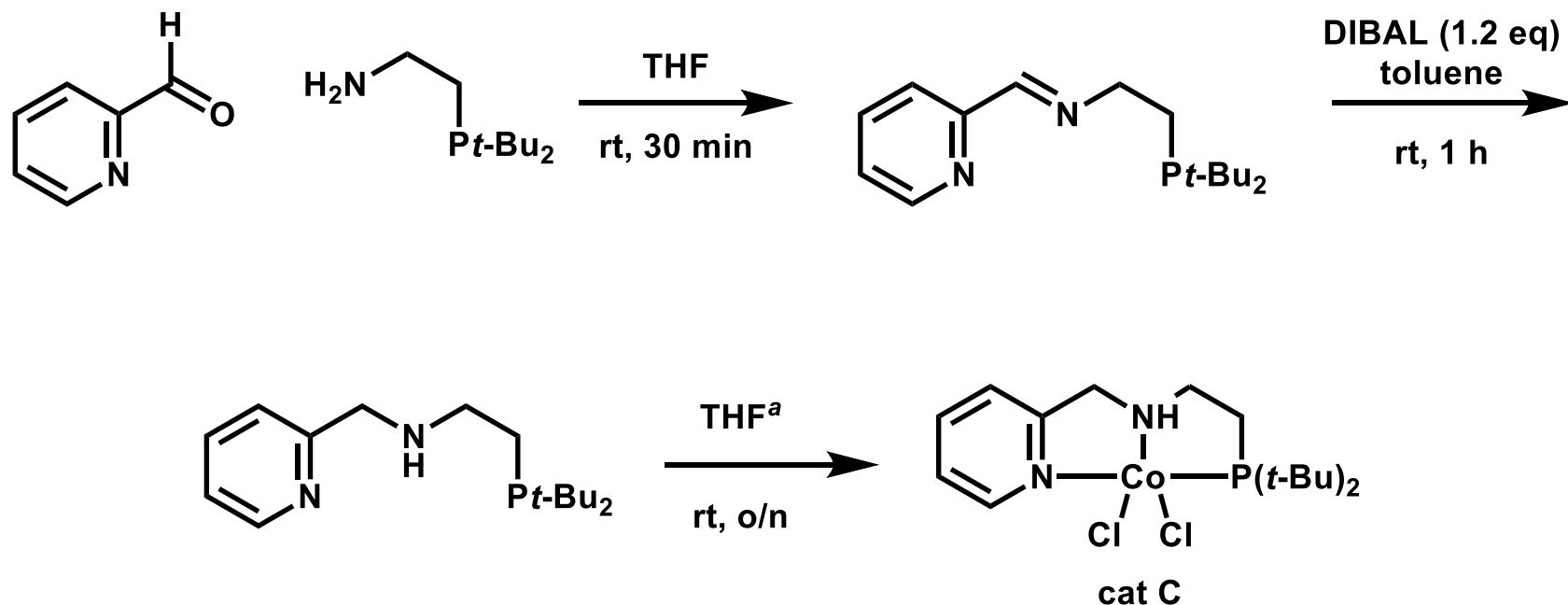


- E-selective semi-hydrogenation

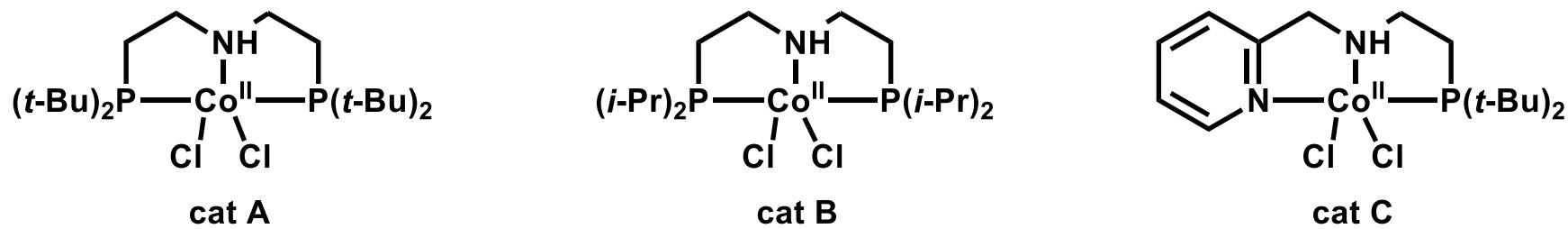


Appendix

Preparation of catalyst

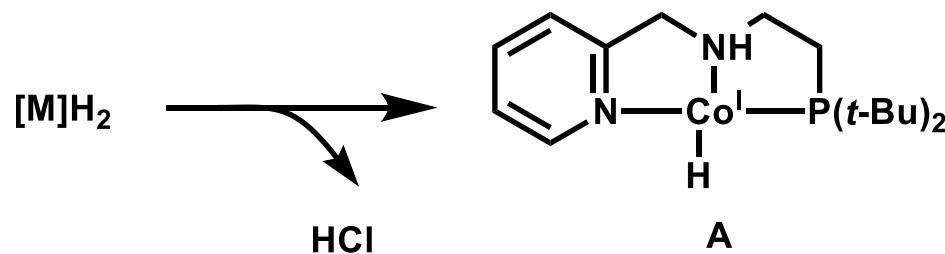
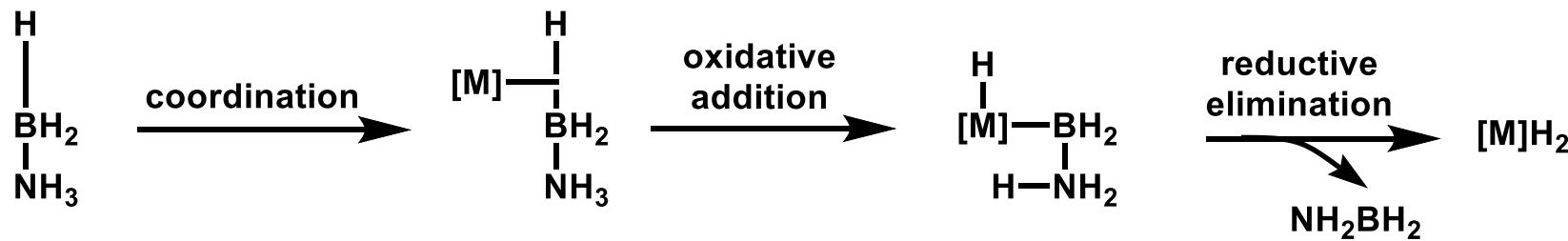
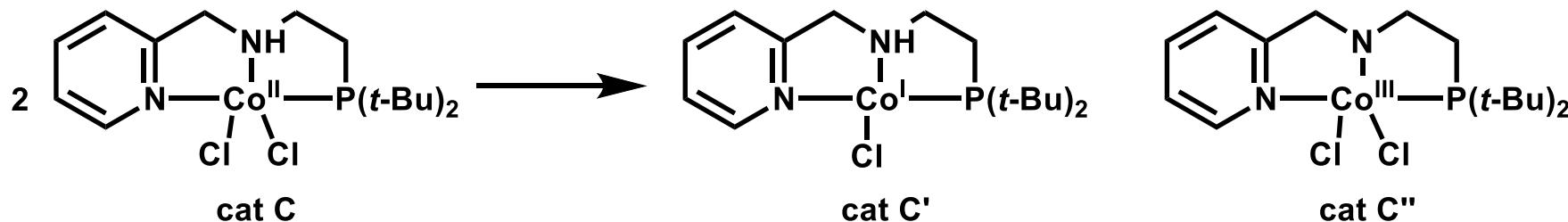


a: cat A and cat B were prepared under same condition

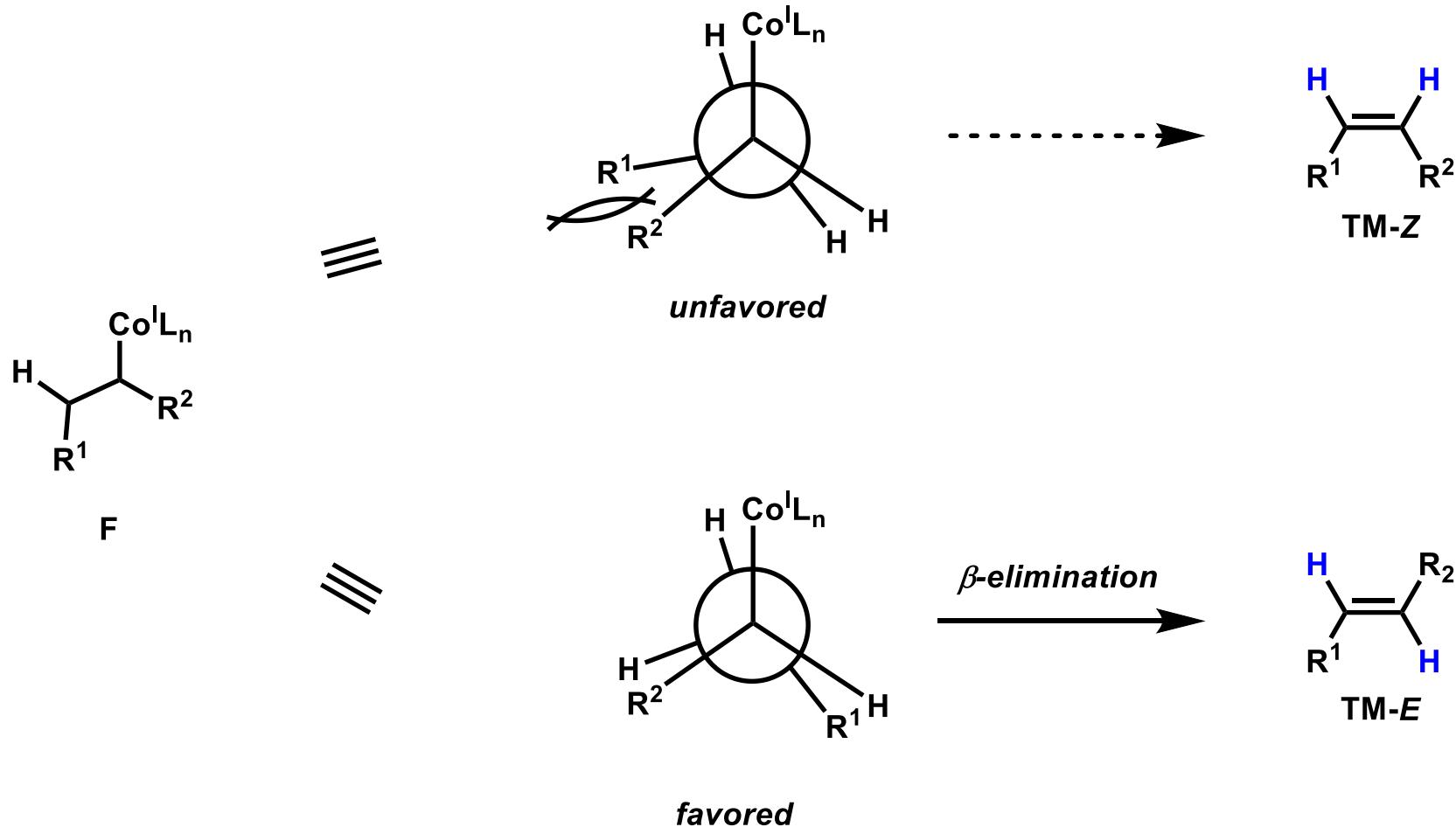


Activation of catalyst

disproportionation of Co(II)

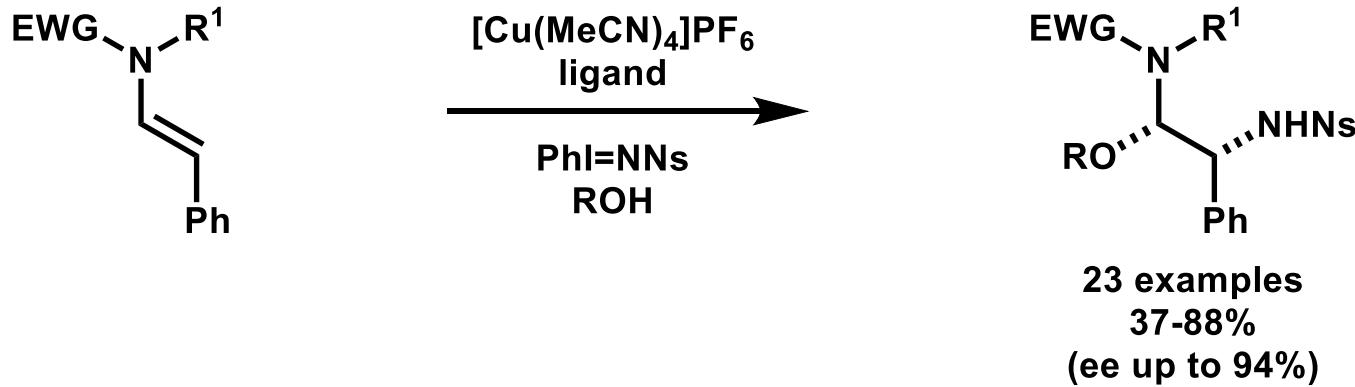


E/Z-selectivity of β -elimination

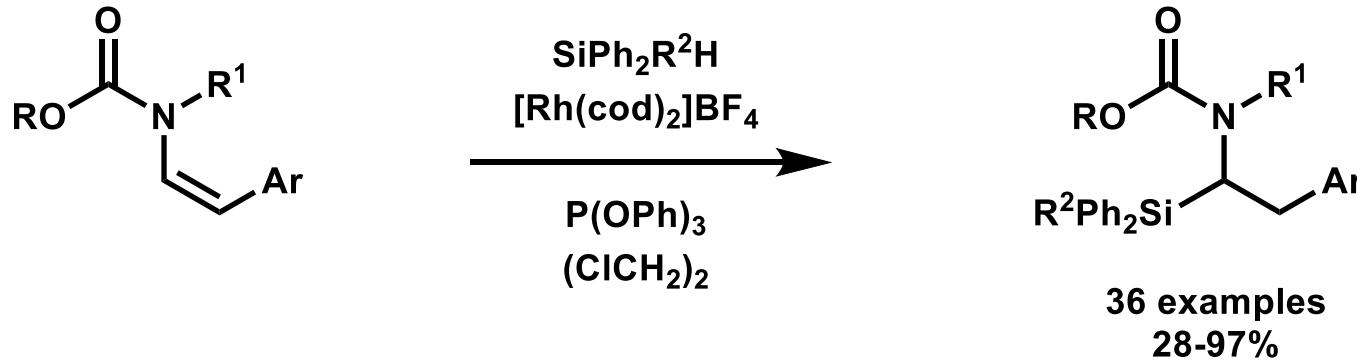


Utilization of enamide

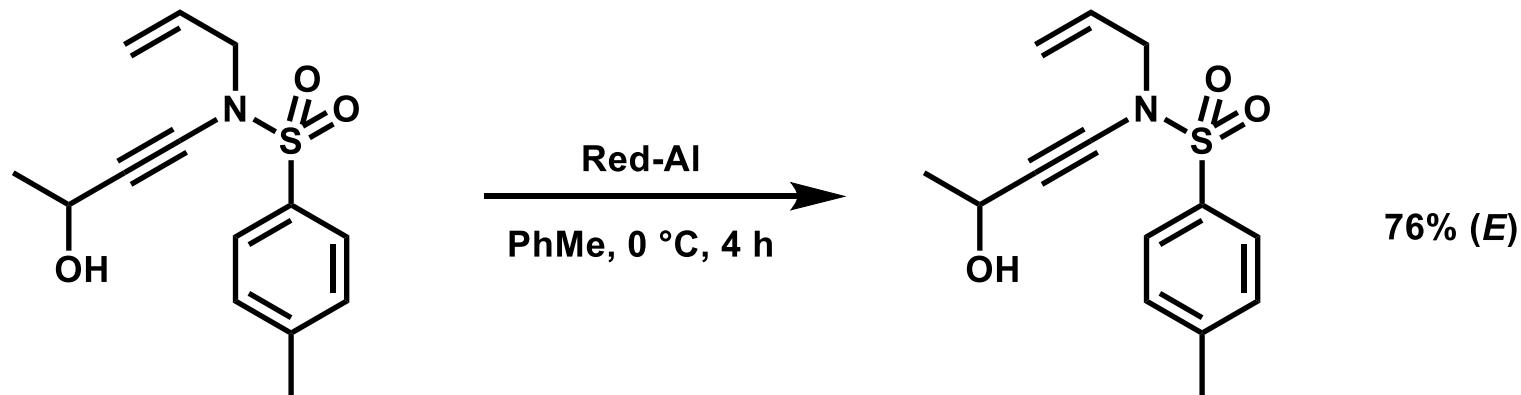
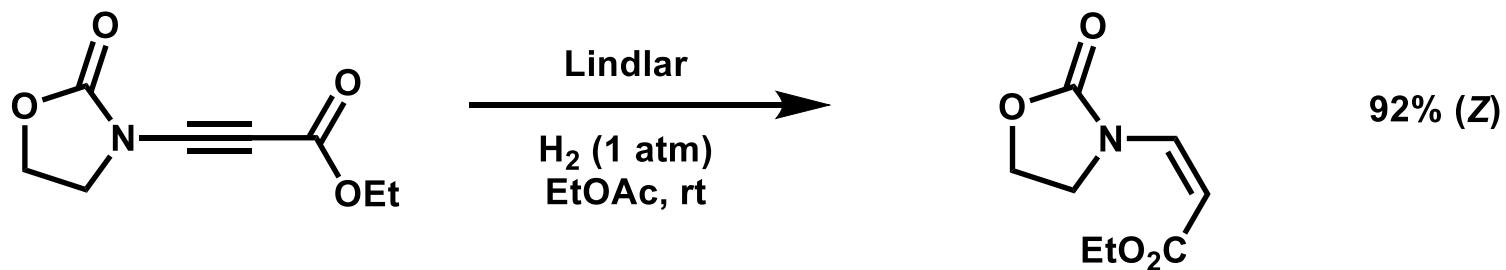
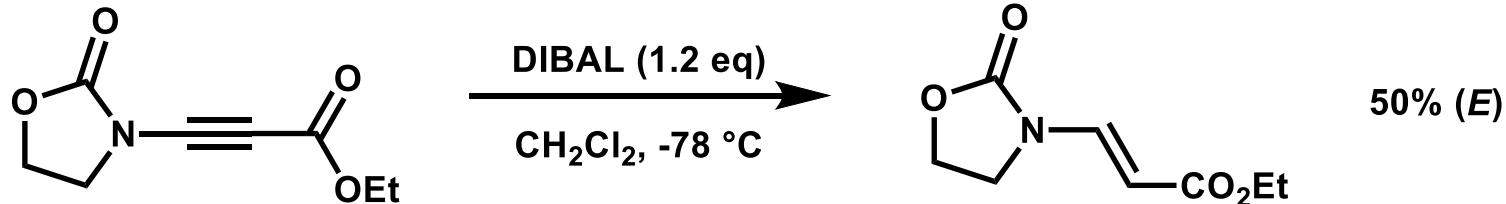
- Direct intermolecular difunctionalization



- Hydrosilylation of enamides using silanes



Other reduction methods of ynamide



Proposed mechanism for the trans addition

Scheme III. Proposed Mechanism for the Apparent Trans Addition

