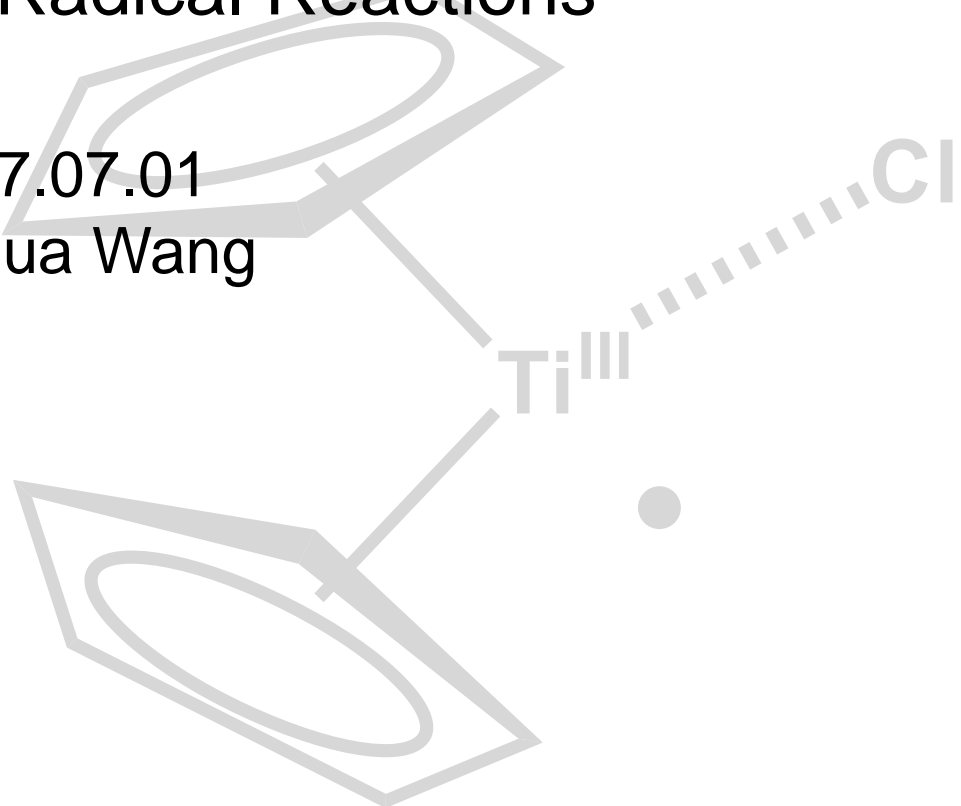


Nugent Reagent

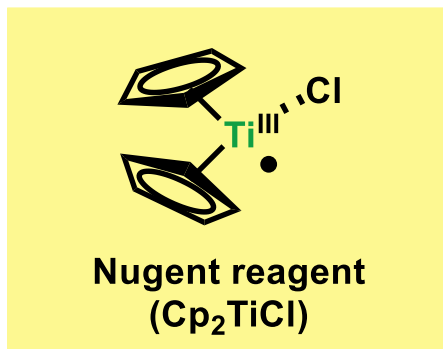
-Application for Radical Reactions-

2017.07.01

Yinghua Wang



Introduction of Nugent Reagent



- synthesized for the first time by Wilkinson in 1955
- a mild single-electron-transfer (SET) reagent
- **Ti**: 7th most abundant metal on Earth

Principal Transformations Mediated by Nugent Reagent are...

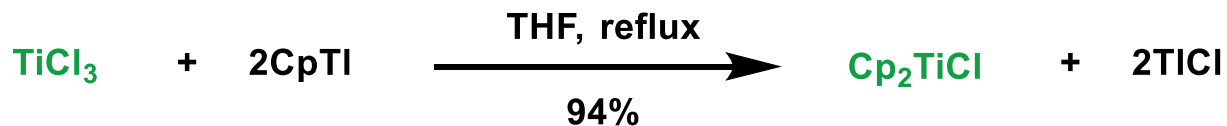
- **Radical ring-opening reactions** (such as epoxide)
- **Radical cascade Cyclizations**
- **Coupling reactions** (**homocoupling**, Pinacol coupling, McMurry coupling etc...)
- Umpolung reactions
- THF-ring formation reactions
- Hydrogen-atom transfer
- Barbier-type reactions
- Deoxygenation of alcohols (see; 131102_LS_Takahiro_KAWAMATA)
- Polymerization reactions

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1. Reductive Opening of Epoxide
2. Homocoupling Reaction
3. Application to Total Synthesis (main paper)

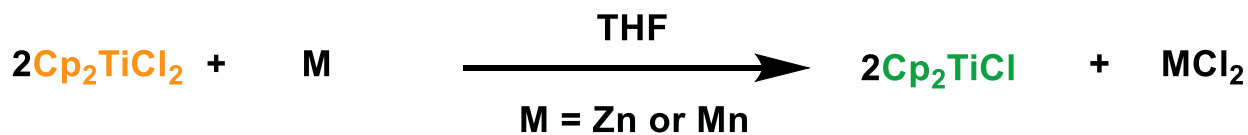
Preparation of Nugent Reagent

Isolation of Cp₂TiCl

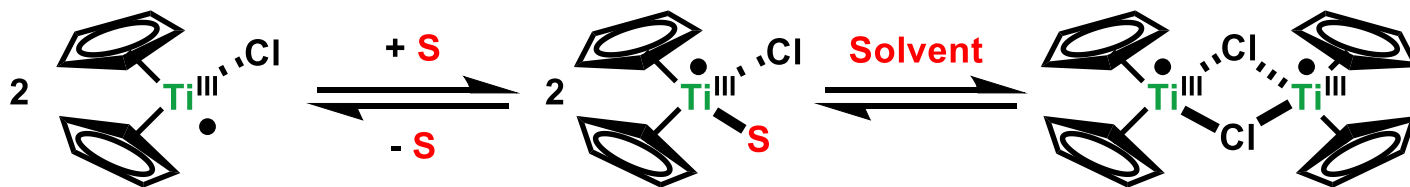


Manzer, L. *Inorg. Synth.* **1982**, 21, 84.

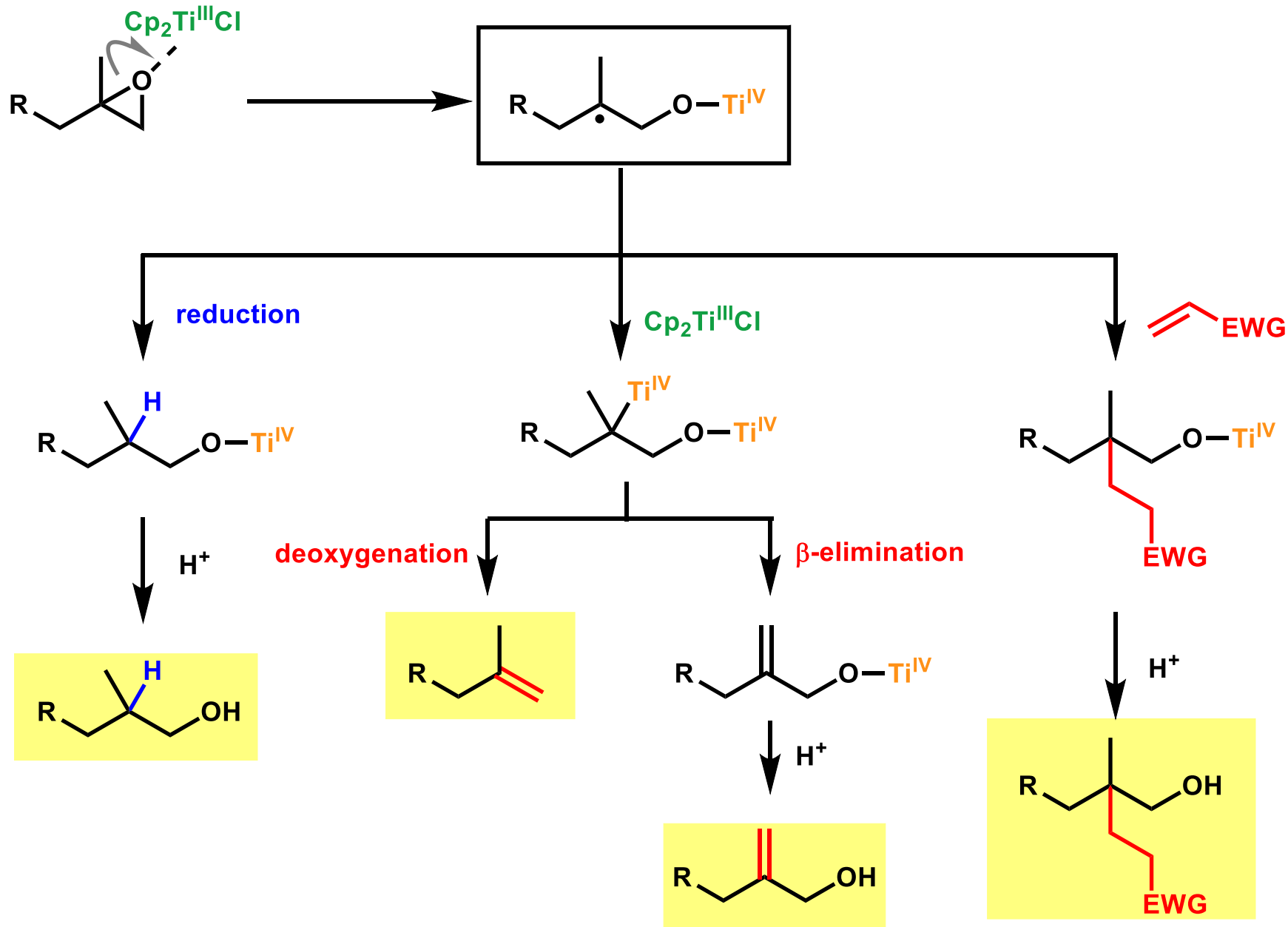
in situ preparation of Cp₂TiCl



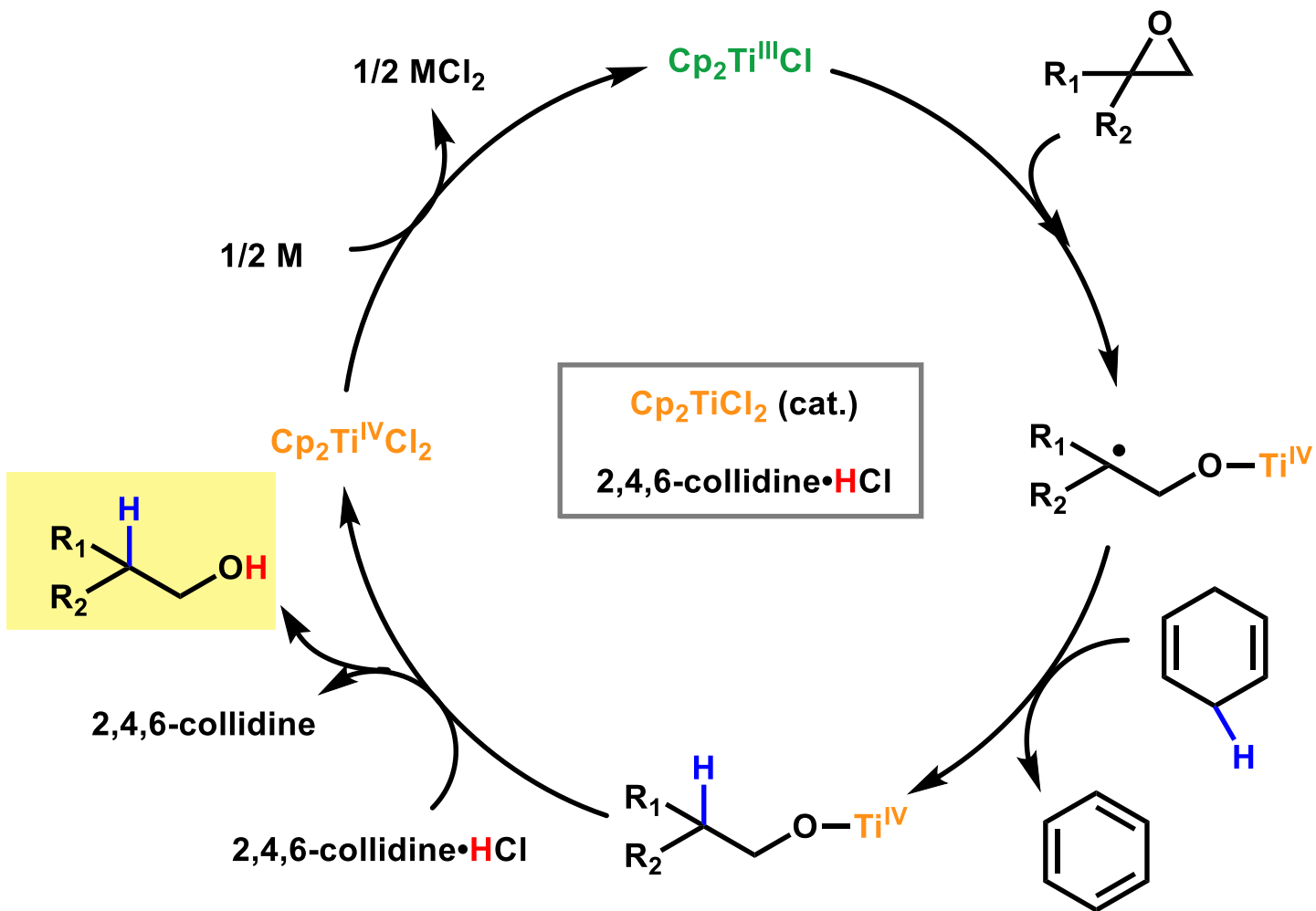
Equilibrium of species



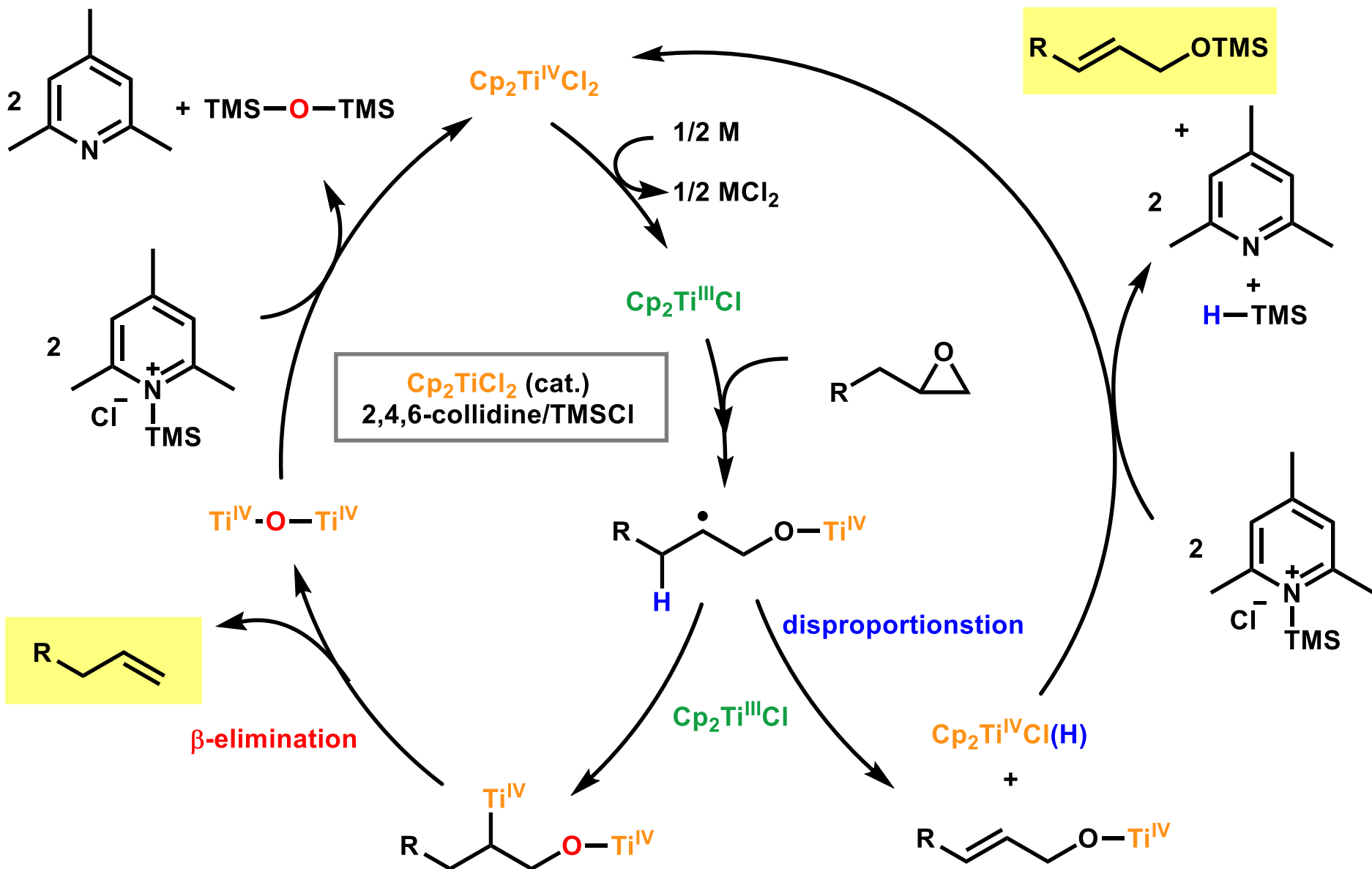
Radical Ring-Opening of Epoxides



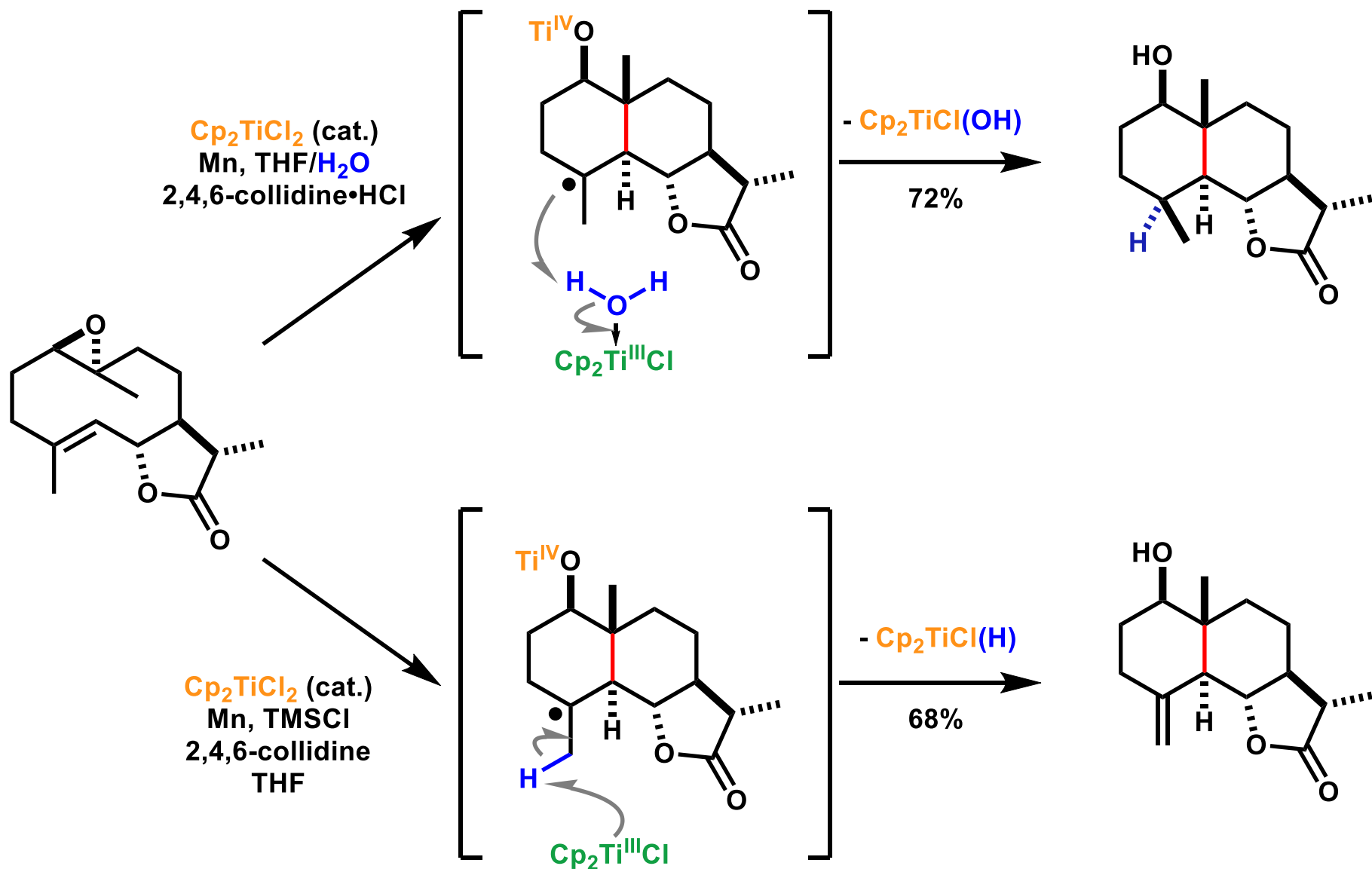
Catalytic Reductive Epoxide Opening Reaction



Catalytic Nonreductive Epoxide Opening Reaction

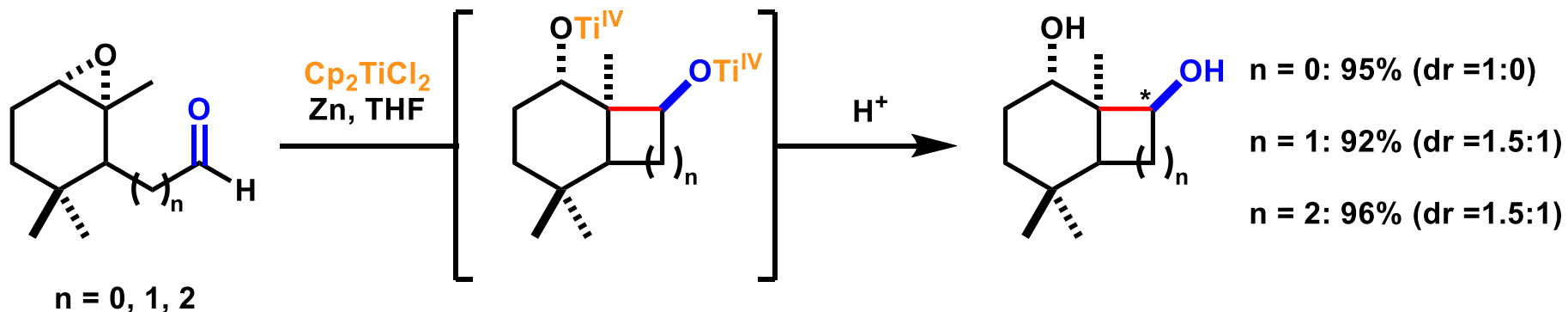


Application of Reductive/Nonreductive Catalytic Cycles ⁷

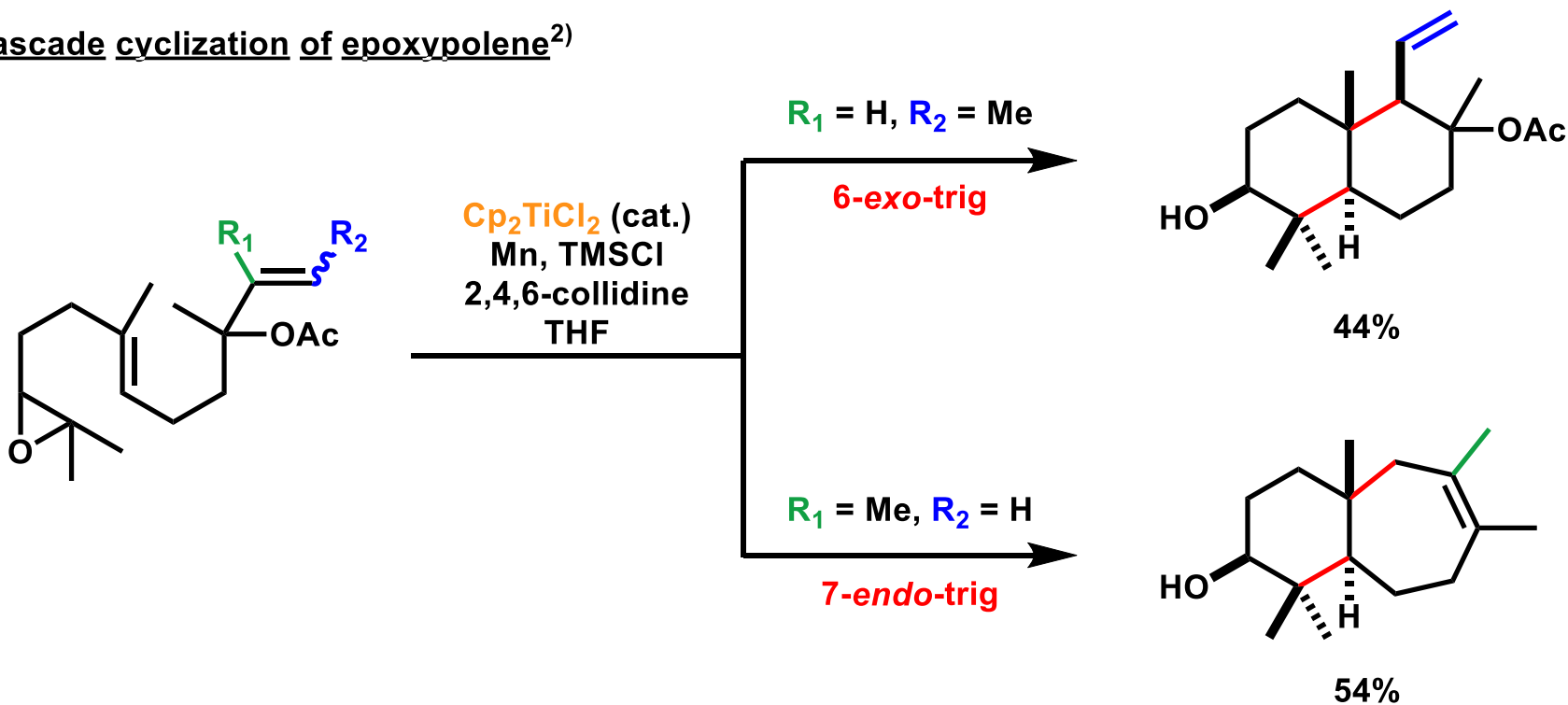


Examples of Radical Cyclization

Radical cyclization: Addition to carbonyl group¹⁾



Cascade cyclization of epoxyolene²⁾



1) Fernandez-Mateos, A.; Martin de la Nava, E.; Coca, G. P.; Silvo, A. R.; Gonzalez, R. R. *Org. Lett.* **1999**, *1*, 607. 2) Justica, J.; Jimenez, T.; Miguel, D.; Contreras-Montoya, R.; Chanboun, R.; Alvarez-Manzaneda, E.; Collado-Sanz, D.; Cardenas, D. J.; Cuerva, J. M. *Chem.-Eur. J.* **2013**, *19*, 14484.

Enantioselective Opening of Epoxide

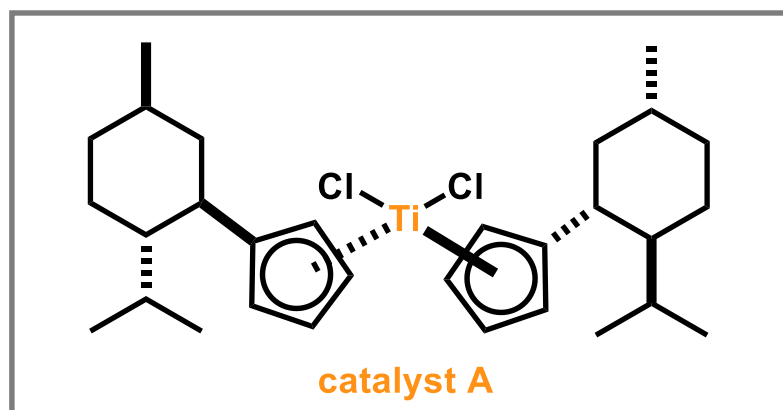
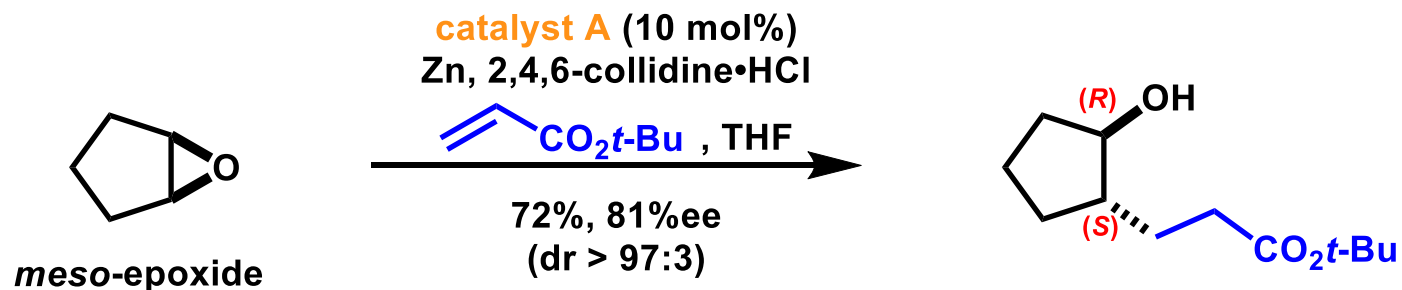
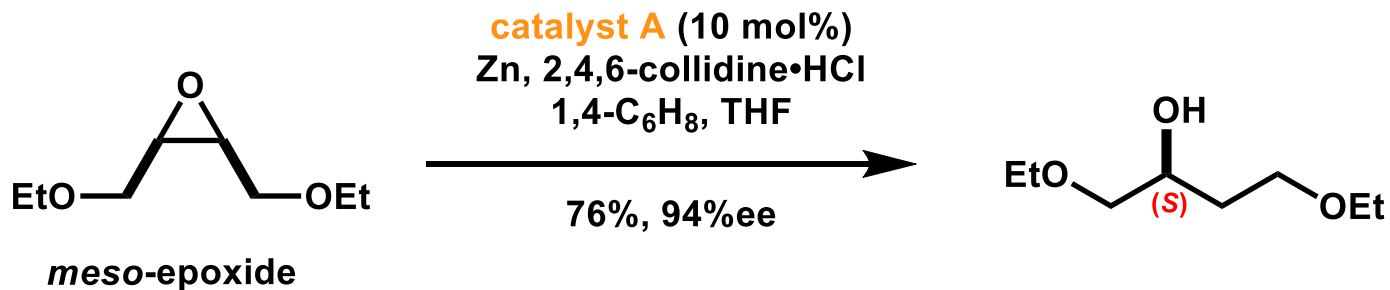
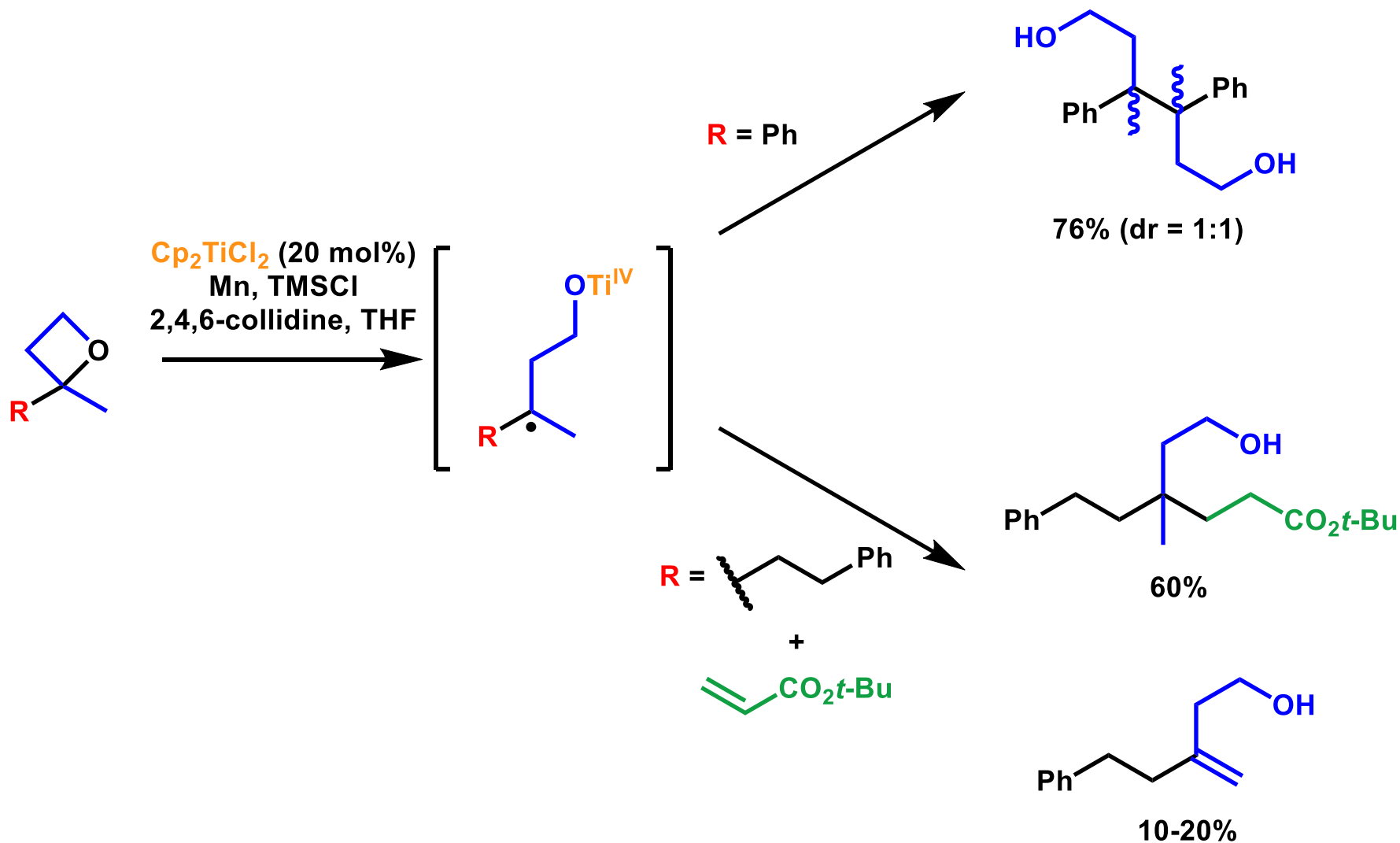


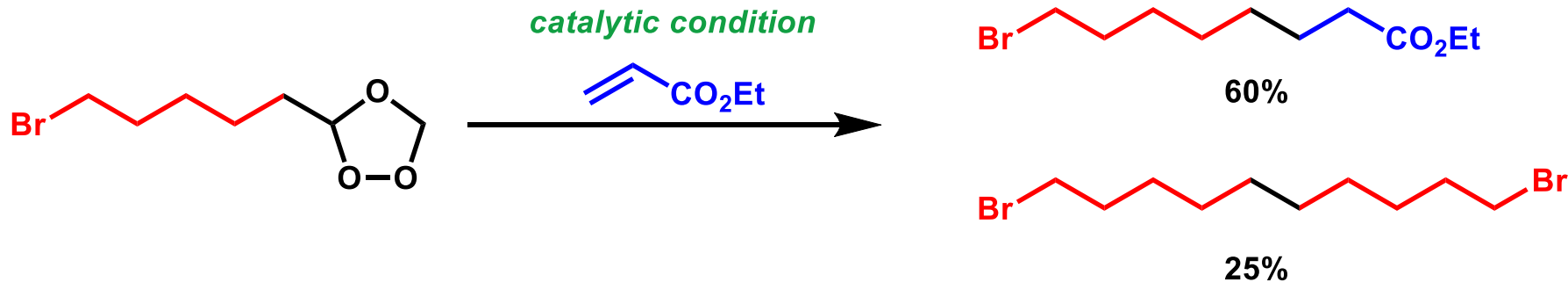
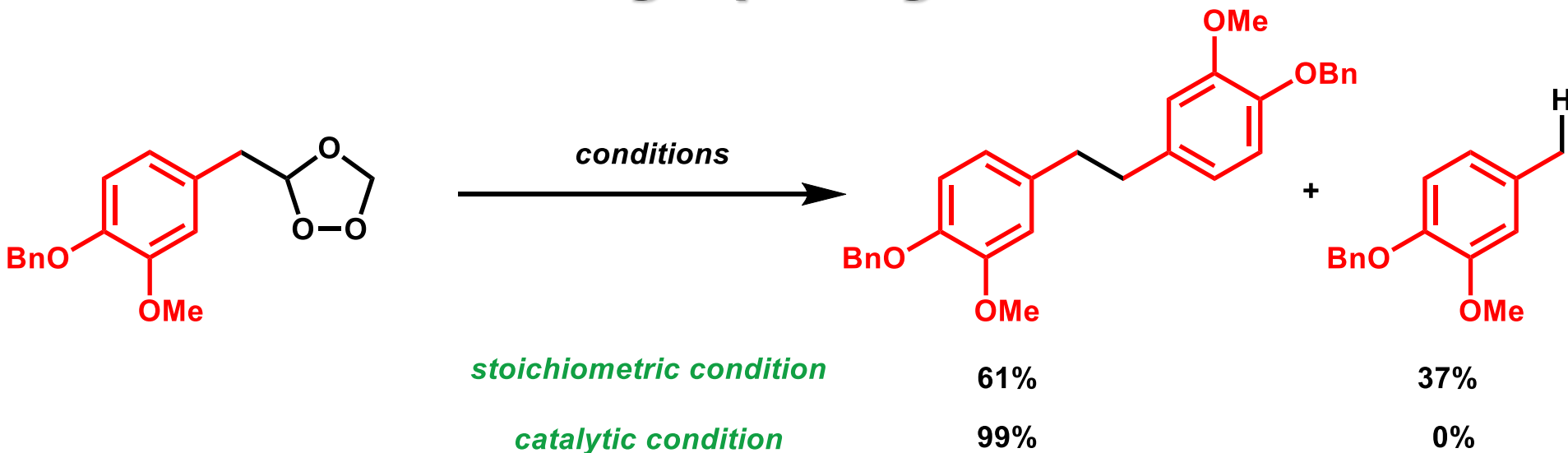
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1. Reductive Opening of Epoxide
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Radical Ring-Opening of Oxetanes

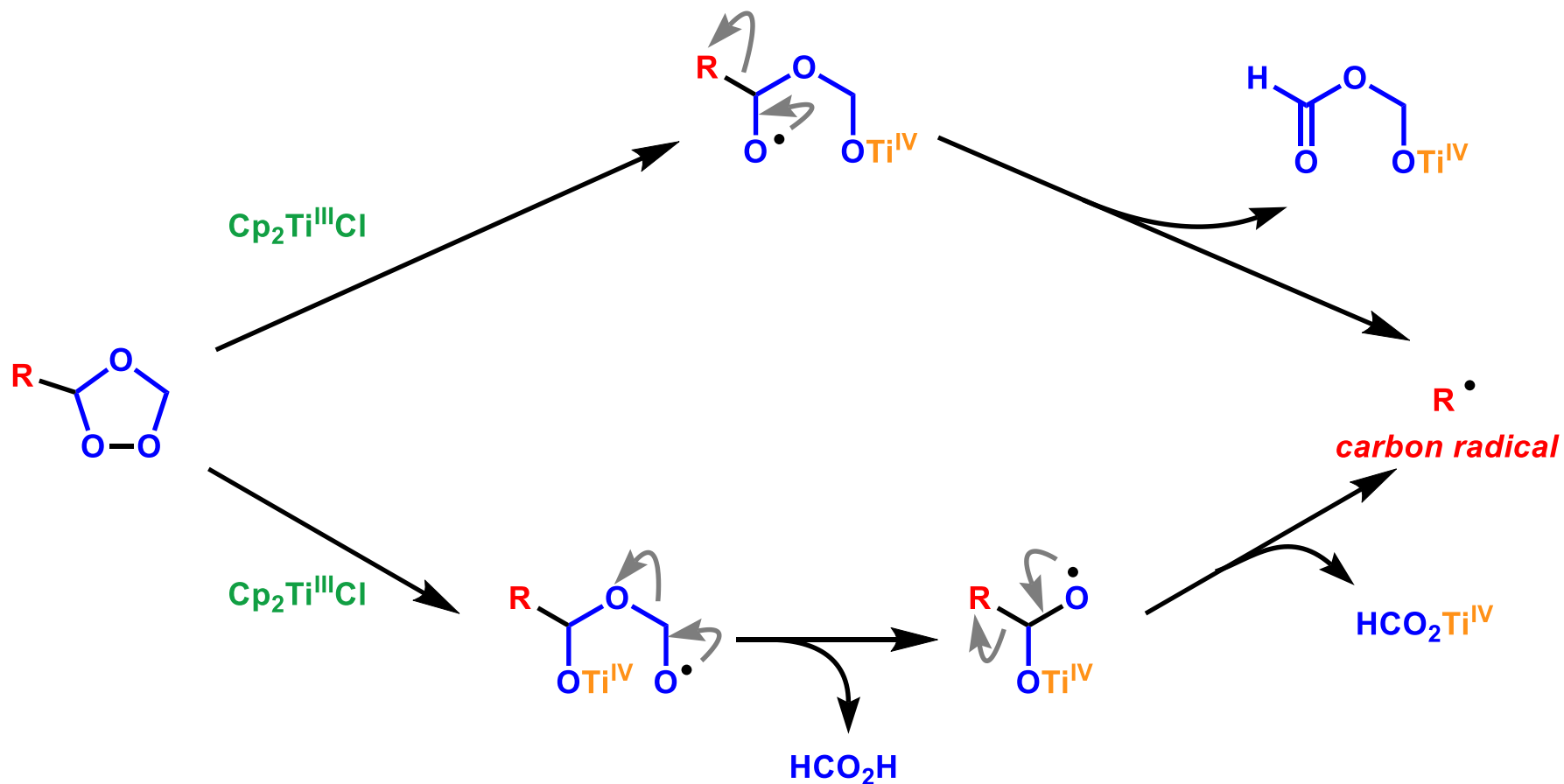


Radical Ring-Opening of Ozonide

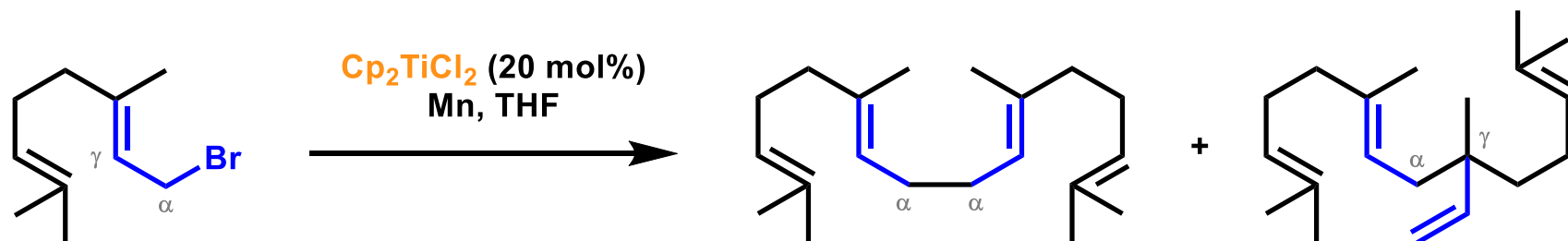


stoichiometric condition: Cp_2TiCl_2 (3 eq.), Mn, THF
catalytic condition: Cp_2TiCl_2 (10 mol%), Mn, TMSCl, 2,4,6-collidine, THF

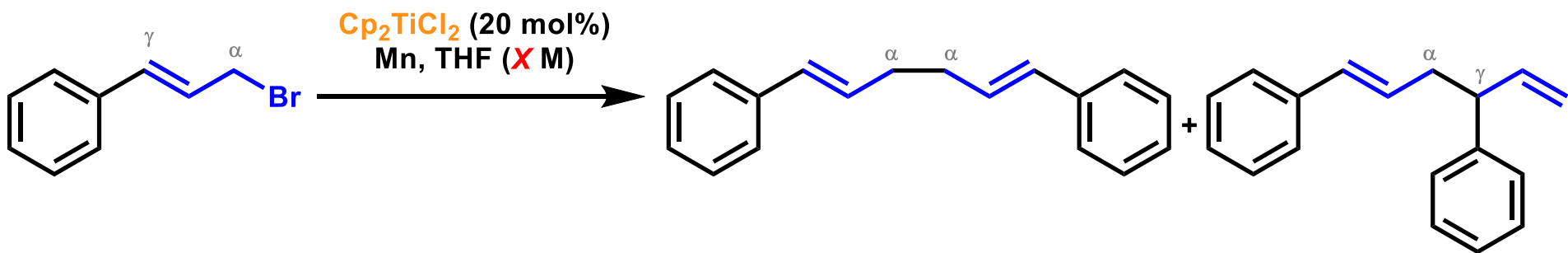
Proposed Mechanism of Homolytic Ozonide Opening



Homocoupling of Allylic Halides



89% ($\alpha\alpha:\alpha\gamma = 64:36$)

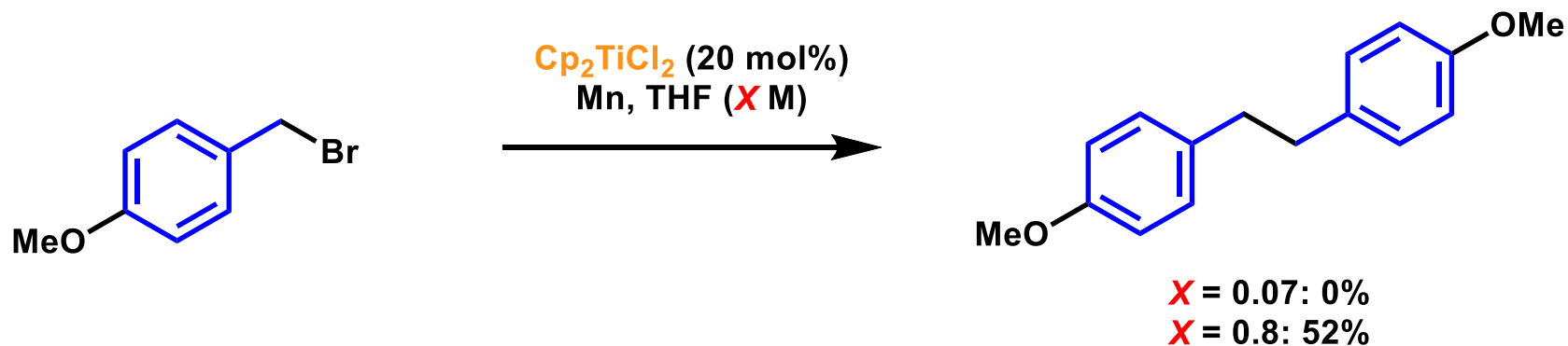


X = 0.07: 0%

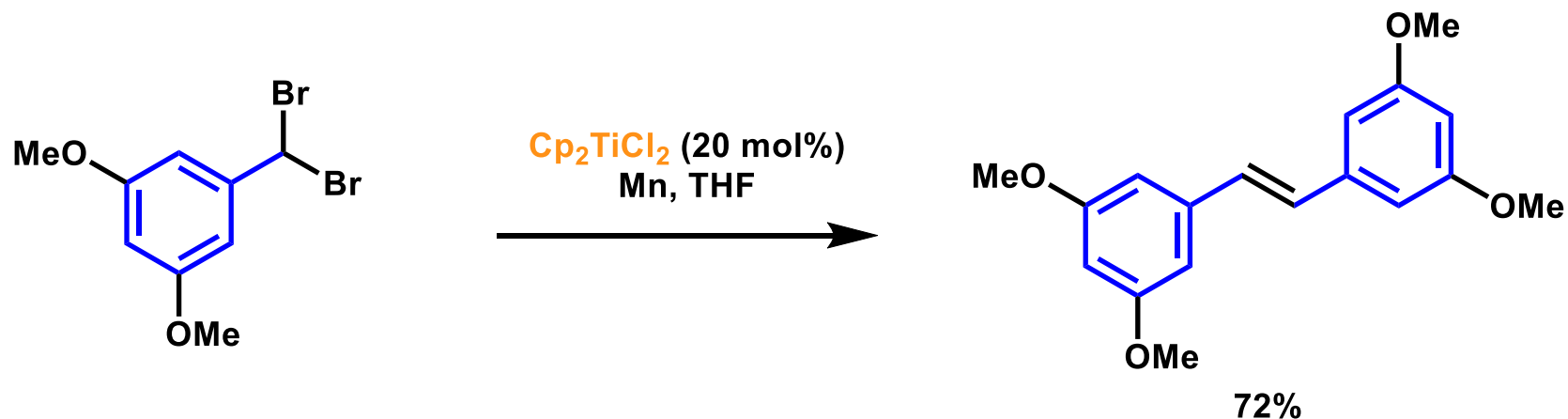
X = 0.8: 85% ($\alpha\alpha:\alpha\gamma = 45:55$)

Homocoupling of Benzylic Halides

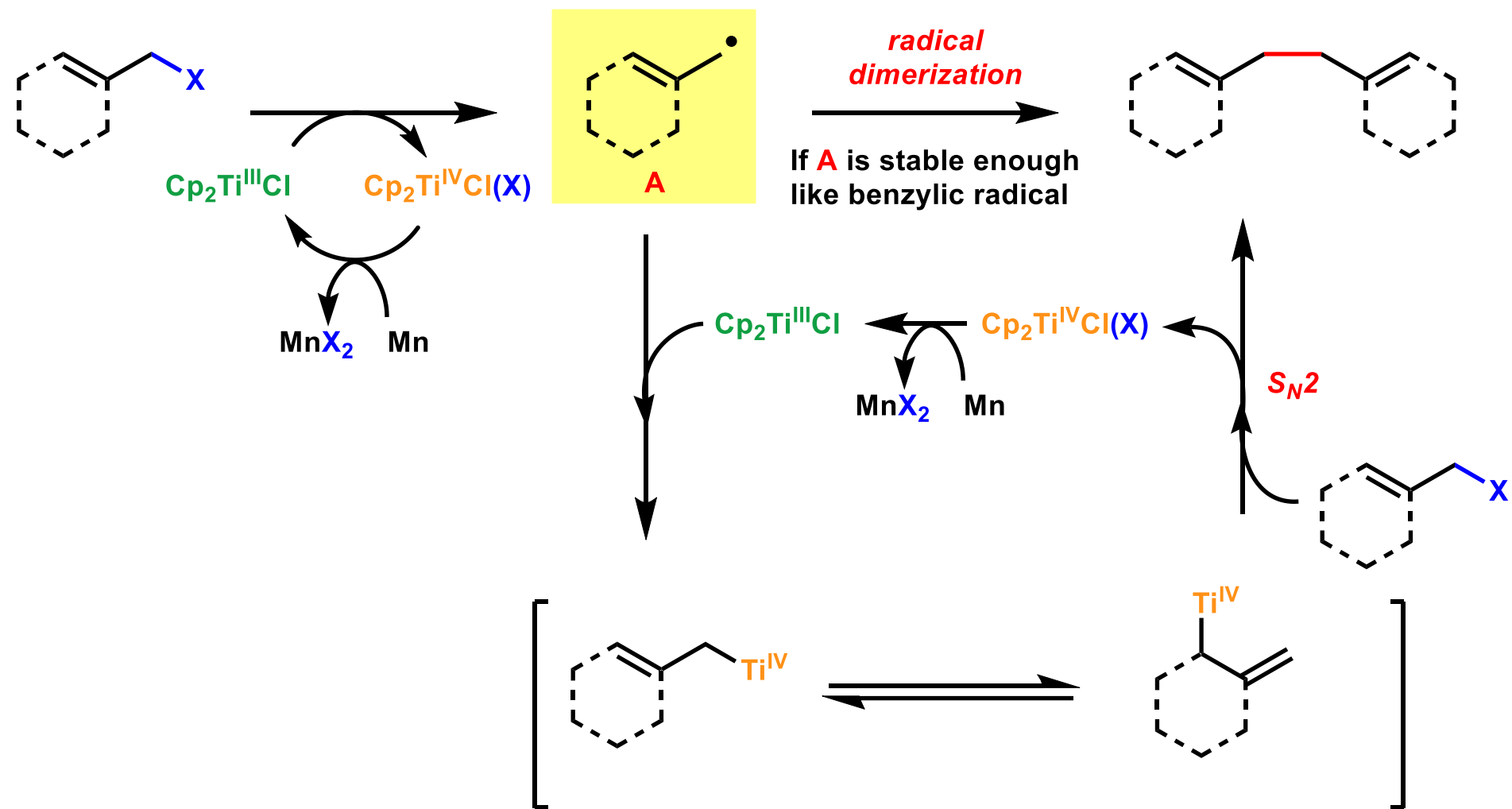
Homocoupling of Benzylic Bromide



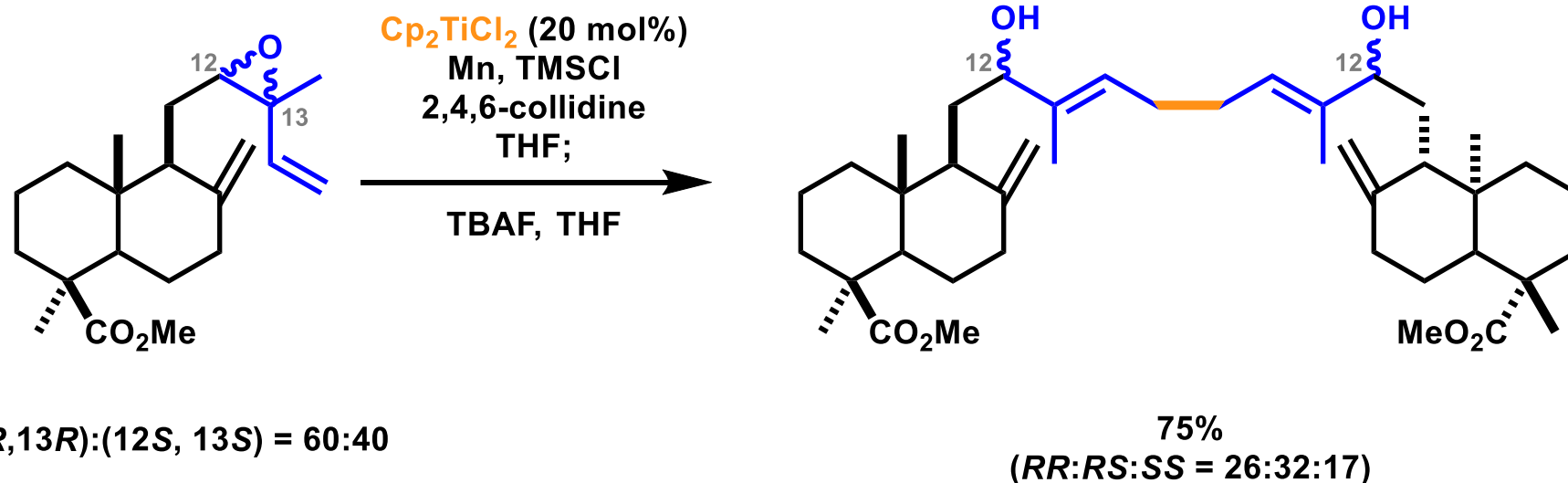
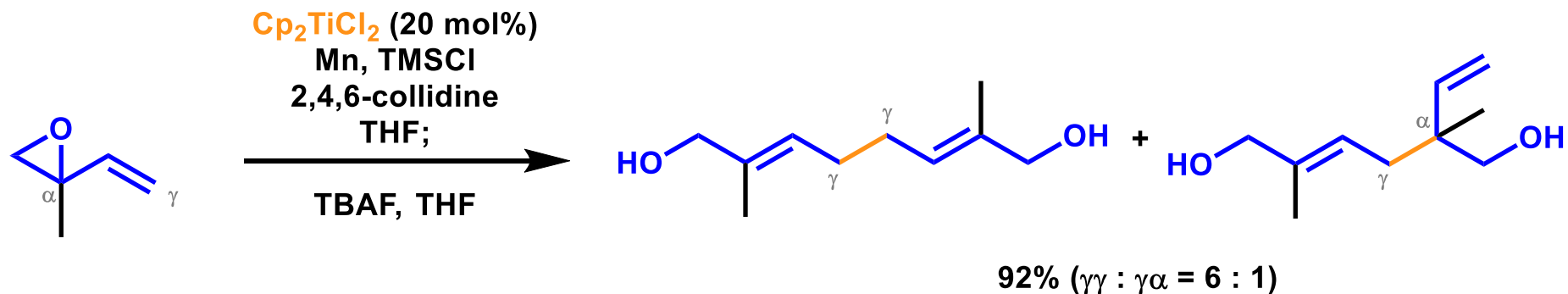
Homocoupling of Benzylic gem-Dibromide



Proposed Mechanism of Homocoupling of Allylic and Benzylic Halides



Homocoupling of Vinyl Epoxides



Proposed Mechanism of Homocoupling of Vinyl Epoxides

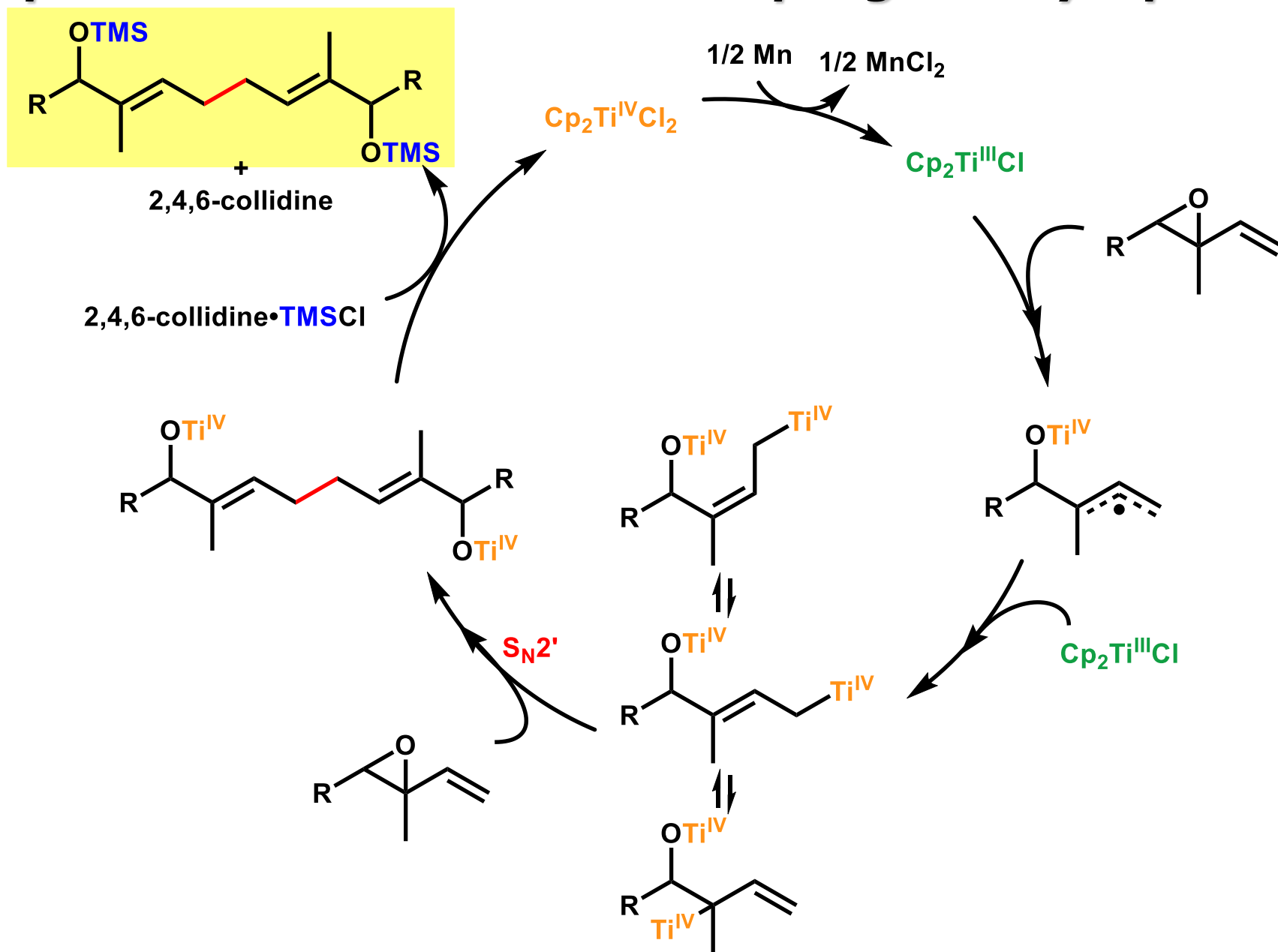
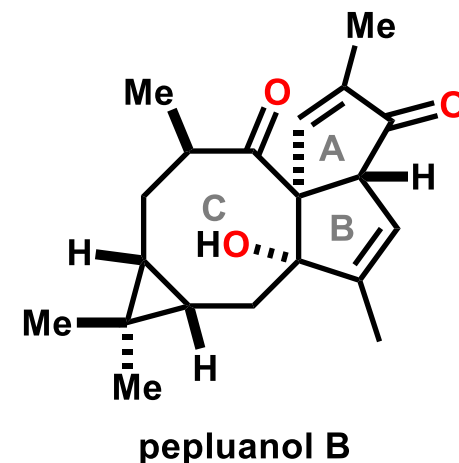
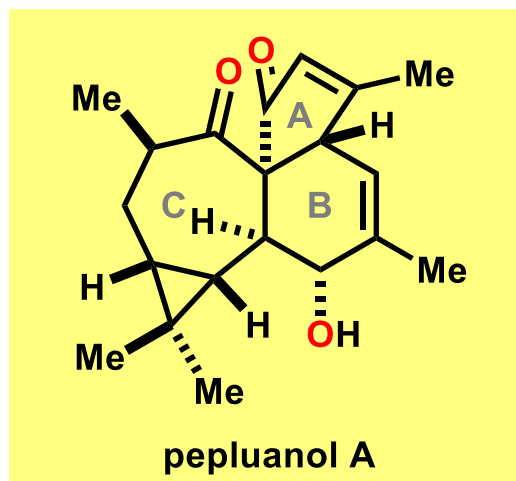
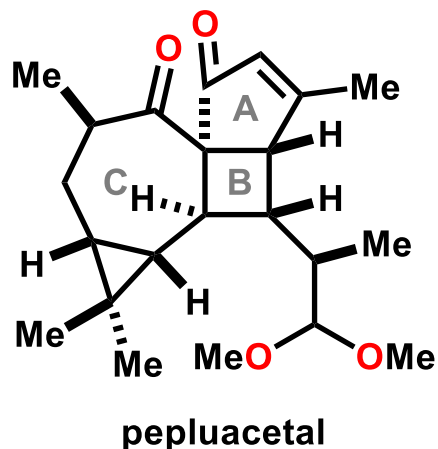


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1. Reductive Opening of Epoxide
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Introduction of Pepluanol A



Isolation:

from *Euphorbia peplus*

Wan, L.-S.; Nian, Y.; Ye, C.-J.; Shao, L.-D.; Peng, X.-R.; Geng, C.-A.; Zuo, Z.-L.; Li, X.-N.; Yang, J.; Zhou, M.; Qiu, M.-H. *Org. Lett.* **2016**, *18*, 2166.

Biological activity:

Kv1.3. potassium channel inhibitor

Structural feature:

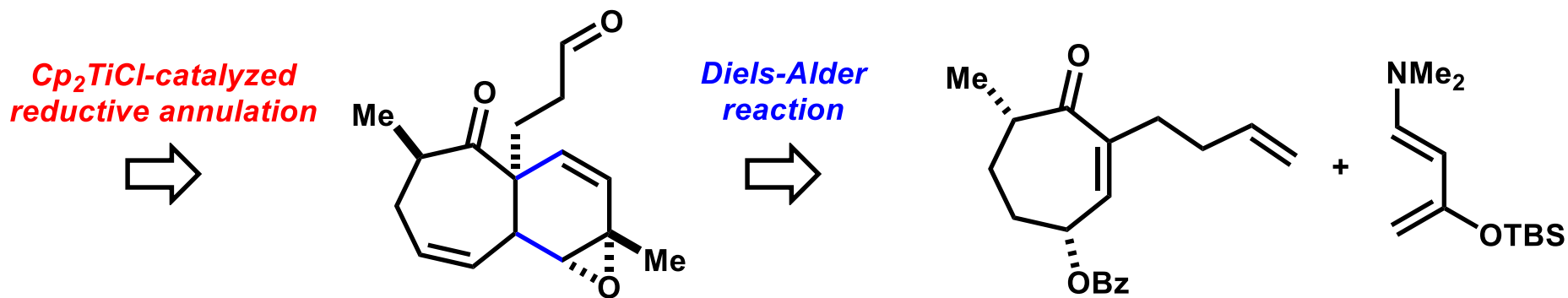
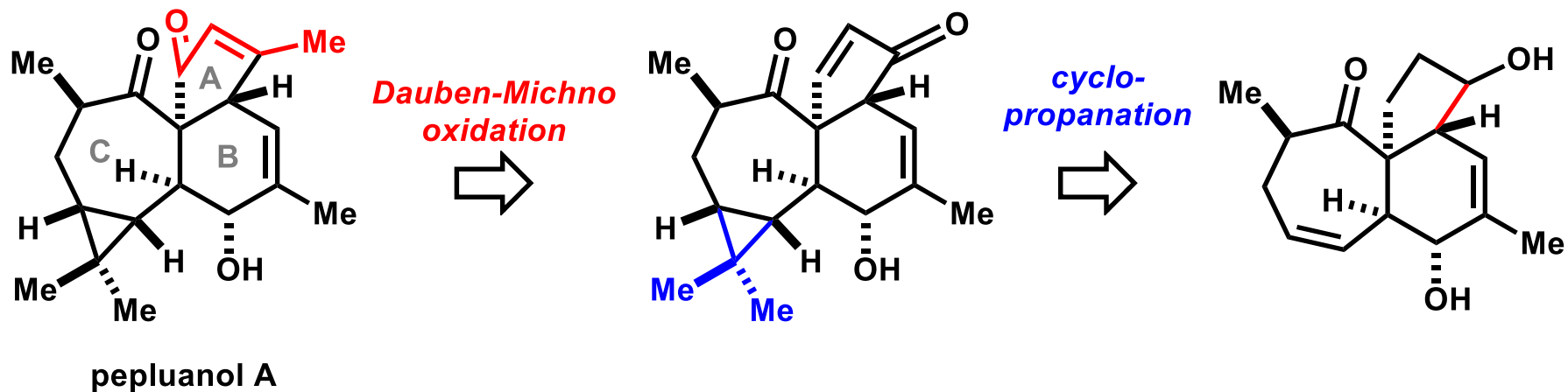
Euphorbia diterpenoid

novel 5/4/7/3, 5/6/7/3, and 5/5/8/3 fused-ring skeletons, including 6-8 stereogenic centers

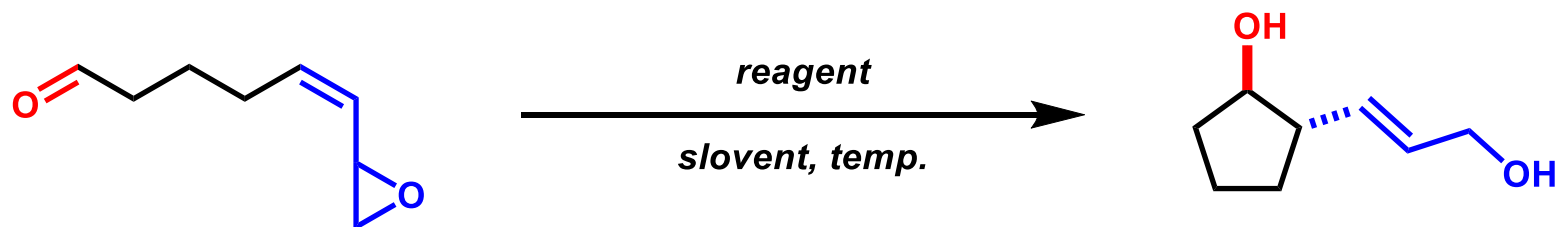
Total Synthesis:

Xuan, J.; Liu, Z.; Zhu, A.; Rao, P.; Yu, L.; Ding, H. *Angew. Chem. Int. Ed.* in press

Retrosynthetic Analysis of Pepluanol A

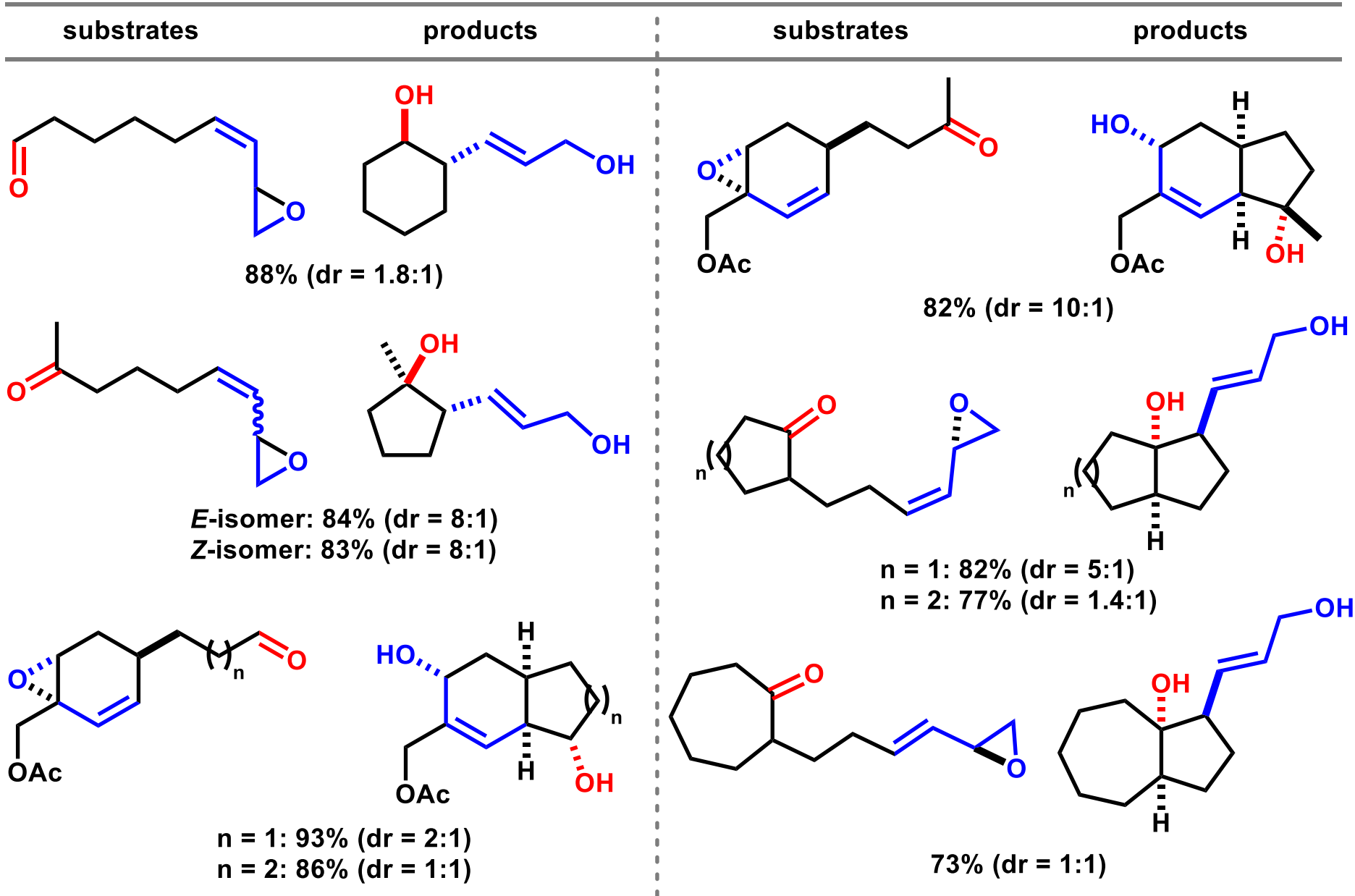


Reductive Annulation of Vinyl Epoxide-Aldehyde

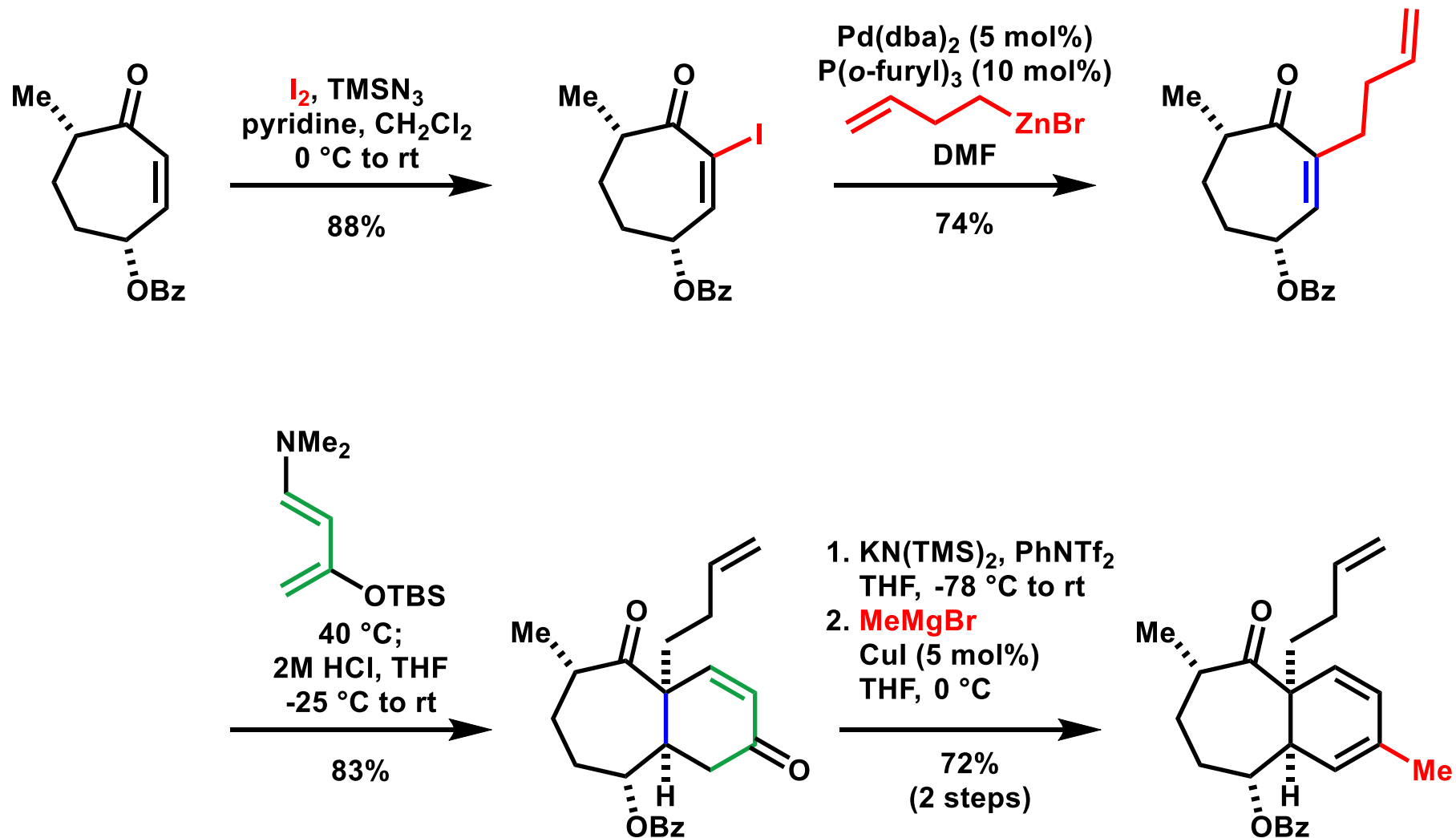


entry	reagent	solvent	temp.	yield (dr)
1	SmI_2 , HMPA	THF	25 °C	52% (>20:1)
2	<i>n</i> - Bu_3SnH , AIBN	benzene	80 °C	33% (1.3:1)
3	LiDBB	THF	0 °C	78% (1.6:1)
4	Cp_2TiCl_2 , Zn	THF	25 °C	88% (1.5:1)
5	Cp_2TiCl_2 (20 mol%), Zn 2,4,6-collidine•HCl	THF	25 °C	95% (1.5:1)

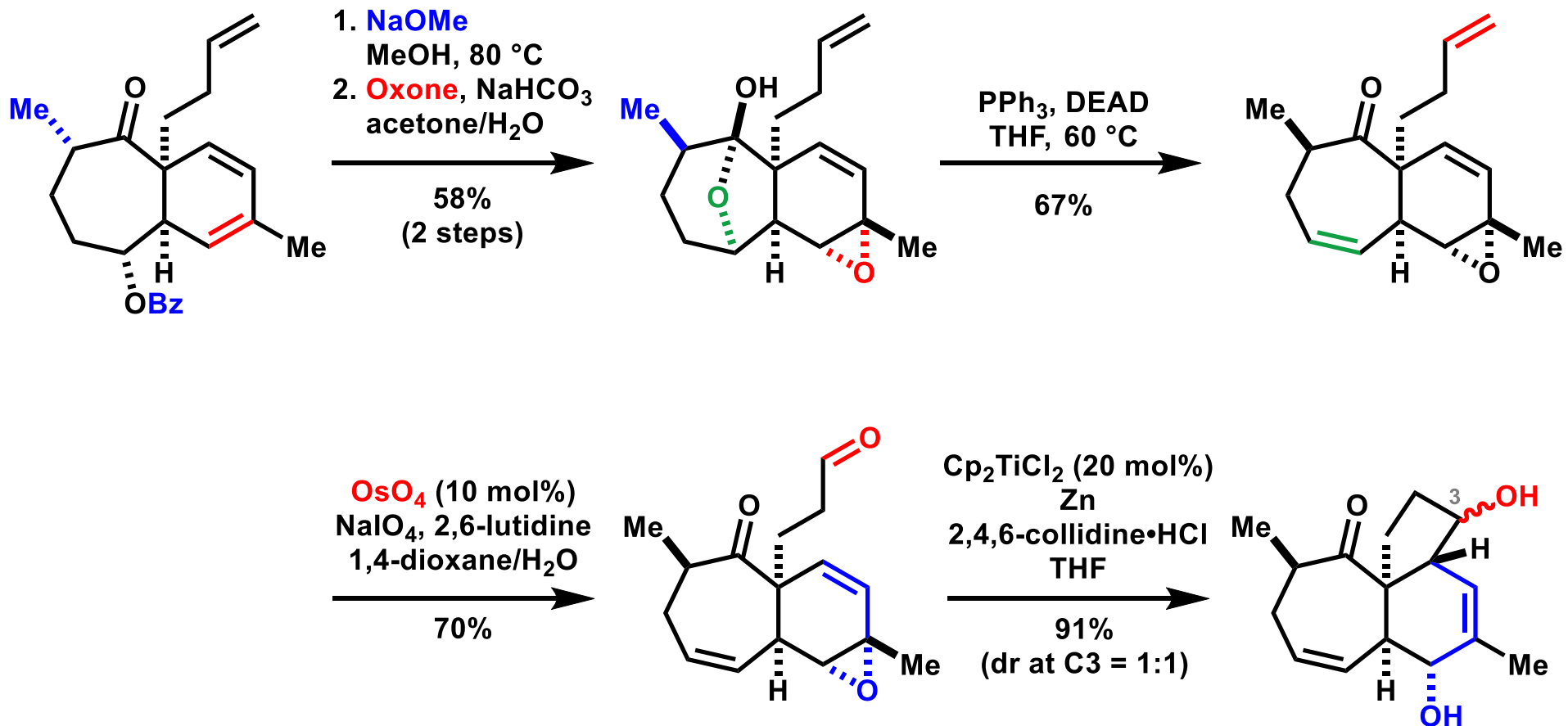
Substrate Scope



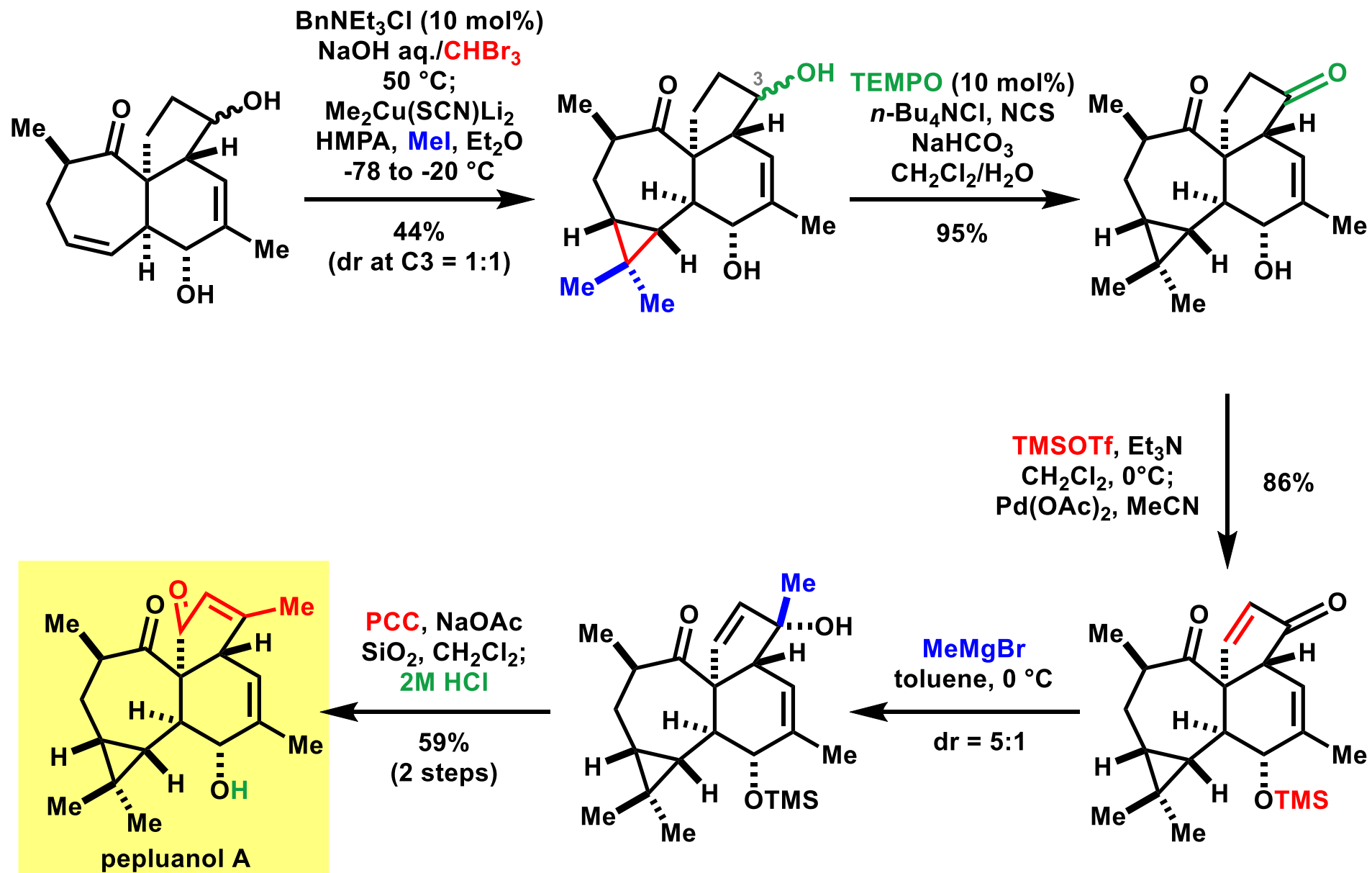
Total Synthesis of Pepluanol A (1)



Total Synthesis of Pepluanol A (2)



Total Synthesis of Pepluanol A (3)

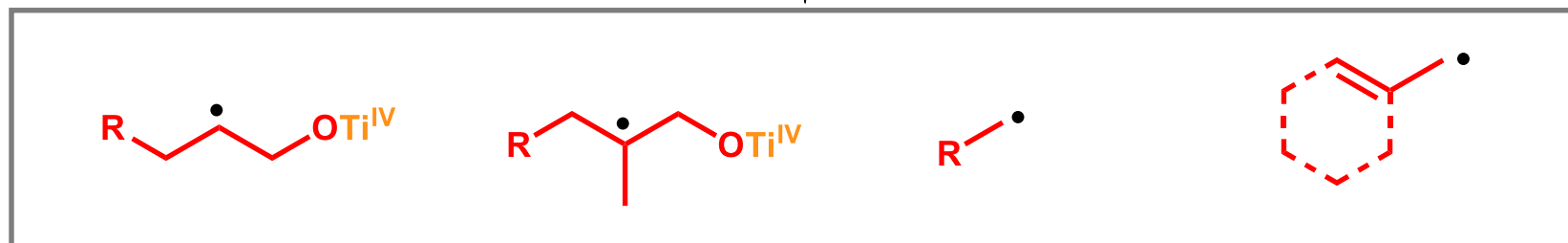


Summary

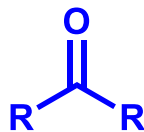


Cp_2TiCl_2

- mild condition
- cheap
- no toxicity



Radicals reactions



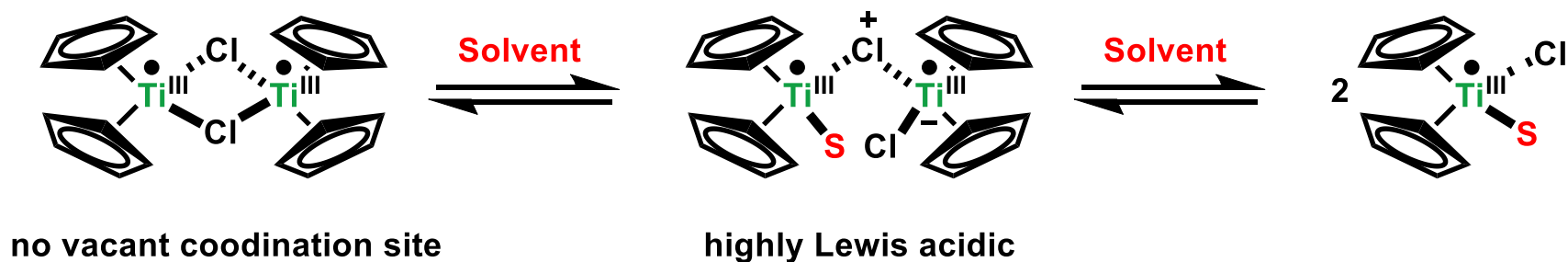
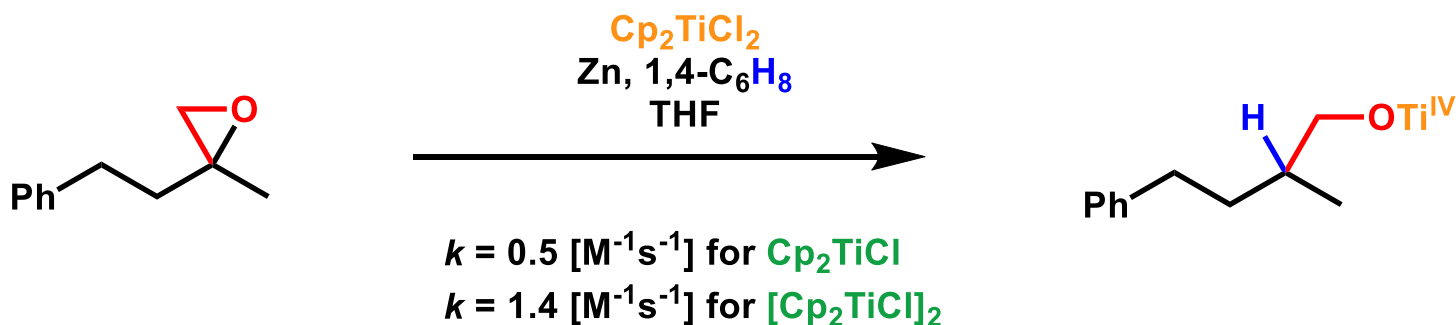
reduction

dimerization

Various Natural Products Syntheses!

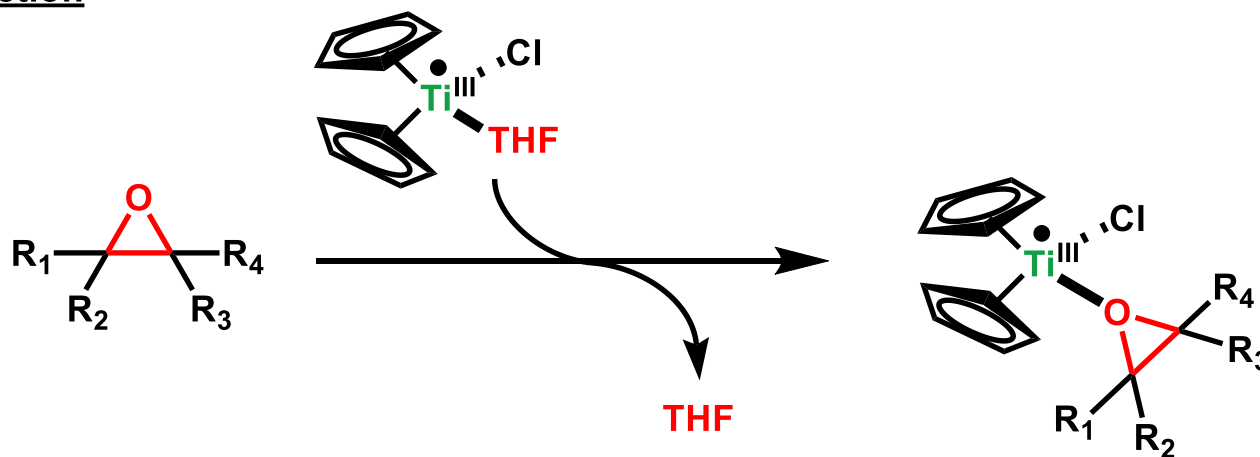
Appendix

A1. Reaction Mechanism of Epoxide Opening (1)



A2. Reaction Mechanism of Epoxide Opening (2)

Initiation of the reaction



Binding enthalpies of epoxides relative to those of THF-complex

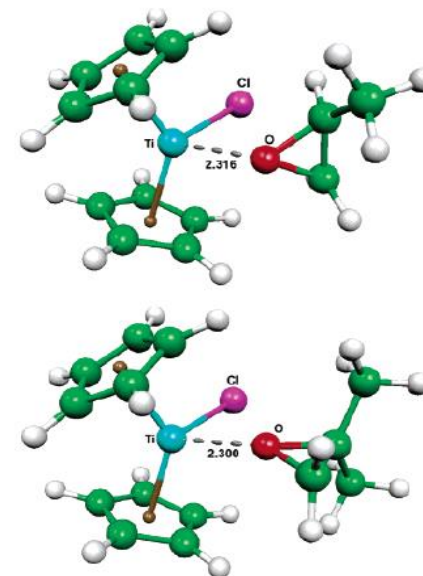
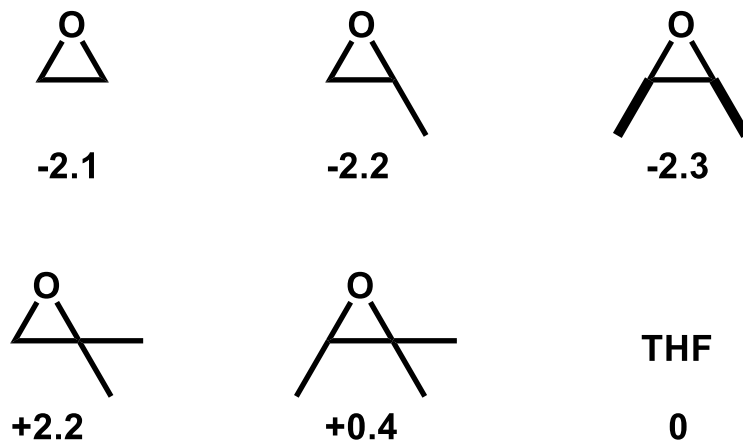

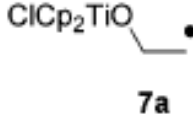

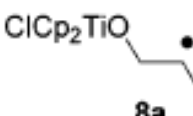
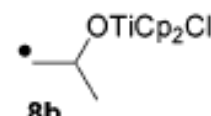
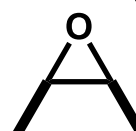
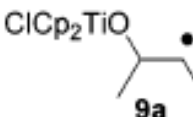

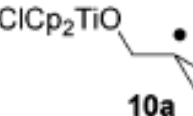
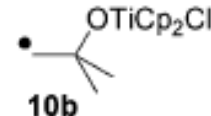
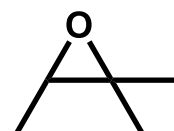
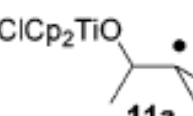
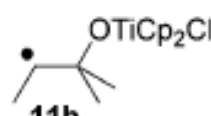


Figure 5. Structure (BP86/TZVP) of the 5a complexes of 8 (top) and 10 (bottom). Distances in angstrom.

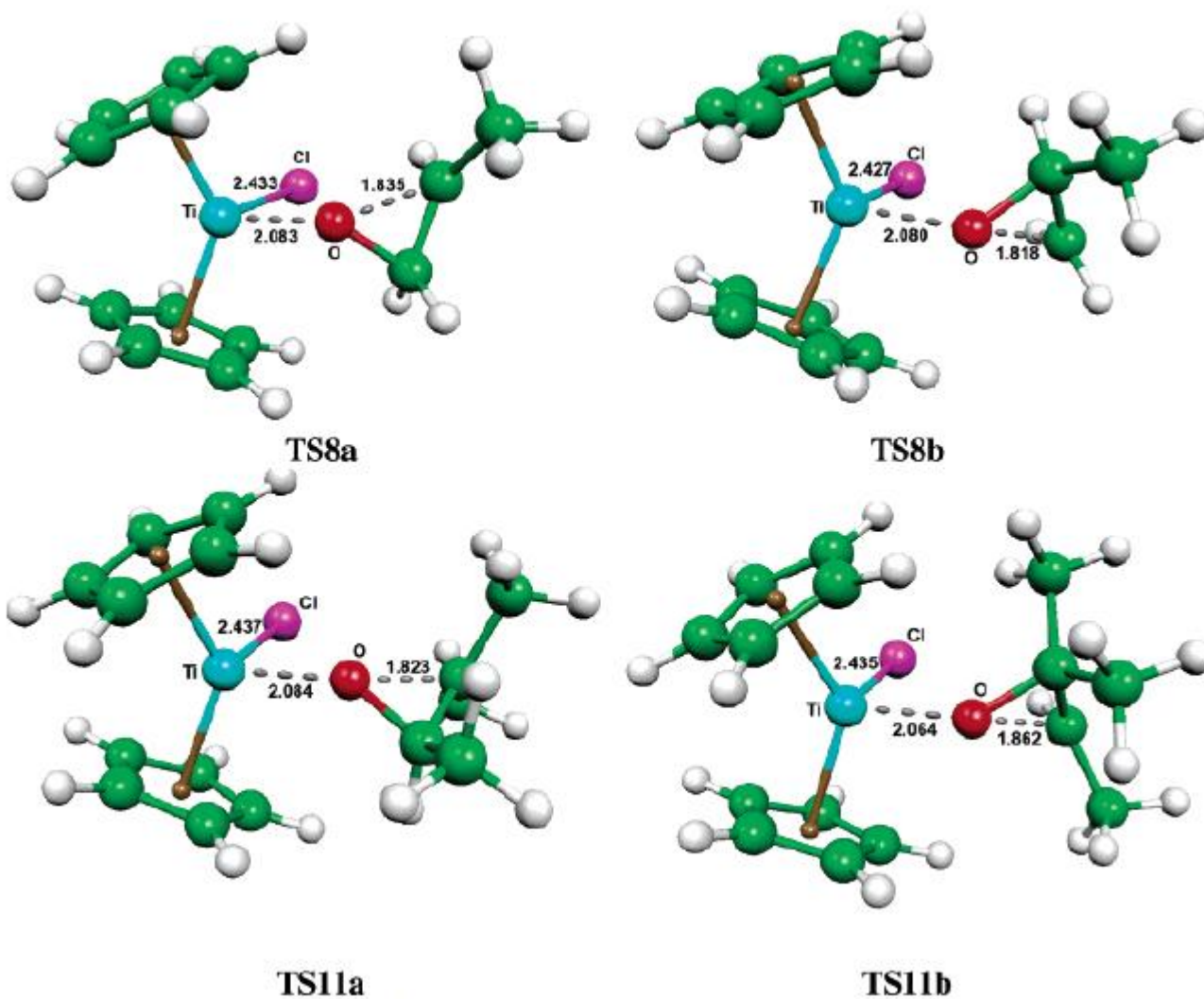
A3. Reaction Mechanism of Epoxide Opening (3)

Table 3. Activation and Reaction Energies of the Epoxide Complexes **5a*7–5a*11** in kcal mol⁻¹

	Product	ΔE	ΔE^\ddagger	Substrate	ΔE^\ddagger	ΔE	Product
	 7a	-4.1	8.7	5a*7	-	-	-
	 8a	-4.0	8.2	5a*8	9.4	-1.8	 8b
	 9a	-1.8	8.8	5a*9	-	-	-
	 10a	-8.5	7.0	5a*10	9.0	-1.3	 10b
	 11a	-4.9	8.7	5a*11	10.3	+0.7	 11b

by DFT calculation with the BP
functional and a TZVP basis set

A4. Reaction Mechanism of Epoxide Opening (4)



Transition structures of the opening of 8 and 11.

A5. Proposed Difference between Sm and Ti

